

Commonwealth of Pennsylvania
Environmental Resources
December 1, 1992

Subject: Stack Test Review

To: Data File
Waste Management of Pa., Inc.
Summit Township, Erie County

From: Richard St. Louis, Chief
Source Testing Section
Division of Source Testing and Monitoring
Bureau of Air Quality Control

Through: Acting Chief
Division of Source Testing and Monitoring

Waste Management of Pennsylvania, Inc., operates the Lake View Landfill (Erie County) gas extraction system and enclosed flare, consisting of gas extraction wells, gas collection header pipe, enclosed ground flare and condensate management structures. The enclosed ground flare (McGill Environment) is designed to operate at a minimum of 1500°F and to achieve a 99% (by weight) destruction efficiency for Non-Methane Volatile Organic Compounds (NMVOCs).

PA Air Quality Services simultaneously collected summa canister samples at the inlet and outlet of the flare, and analyzed the inlet samples for NMVOCs, methane (CH₄), nitrogen (N₂), carbon dioxide (CO₂), carbon monoxide (CO) and oxygen (O₂), and analyzed the outlet samples for NMVOCs, CH₄, N₂, CO₂, CO, O₂ and specific organic compounds (benzene, chloroform, 1, 2-dichloroethane, methyl chloride, methylene chloride, perchloroethylene, trichloroethylene, carbon tetrachloride, vinylidene chloride, and vinyl chloride) as part of a Method 25C test on July 29, 1992. Method 7A testing was also conducted at the outlet for NO_x emission rates. Flare inlet flow rates averaged 1,050 dry standard cubic feet per minute (DSCFM) and temperatures averaged 90°F. Flare outlet flow rates averaged 7,500 DSCFM (compared with 10,600 DSCFM determined by carbon balance) and 1,787°F. A total of three test runs were conducted for both Method 25C and Method 7A.

The test appears to be acceptable to the Department with the following comments:

1. Methylene chloride was the only NMVOC detected at the outlet for all three Method 25C runs. Trichloroethane was detected on Runs 1 and 2, but not on Run 3. All other NMVOC compounds were below the detectable limit.
2. Pressure drops measured by the pitot tube at the outlet were less than the accuracy of the manometer, resulting in unreliable volumetric flow determinations. An alternative method, based on carbon balance, was performed to obtain data in addition to the Method 2 data. Data in Tables 2, 3 and 4 use the calculated carbon balance data for emission rates.

Data File

Waste Management of Pac. Inc.

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December 1, 1992

cc: 25-922-003

Paul R. Shah, Mendville Office

EPA/MSL

Krishnan Hanumanthy, Central Office

Reading File - Stock Testing

ASB:ms

TABLE I

SUMMARY OF GAS CONDITIONS (July 29, 1982)

RUN NO.	SAMPLING LOCATION	GAS FLOW		TEMPERATURE °F	MOISTURE (%)	COMPOSITION (% BY VOLUME, DRY BASIS)			
		RATE DSCFM				CO ₂	O ₂	CH ₄	HC
VI-1	Inlet	1040		80	2.3	39.1	1.3	53.8	5.8
VO-1	Outlet	7640		1755	16.8	10.4	6.8	<0.1	83.0
		(10,800) ^a							<0.1
VI-2	Inlet	1050		80	2.4	40.1	0.8	54.9	4.2
VO-2	Outlet	7420		1820	15.5	10.4	6.4	<0.1	83.2
		(10,900) ^a							<0.1
VI-3	Inlet	1070		80	2.5	39.8	0.0	55.0	4.0
VI-3	Outlet	7360		1819	12.7	10.5	6.4	<0.1	83.1
		(10,200) ^a							<0.1
AVERAGE	Inlet	1050		80	2.4	39.7	1.0	54.7	4.7
	Outlet	7500		1787	15.0	10.4	6.2	<0.1	83.0
		(10,600) ^a							<0.1

^a Values in parentheses calculated by carbon balance method

TABLE II

SUMMARY OF NMVOC EMISSIONS AND REMOVAL EFFICIENCY

RUN NO.	LOCATION	TIME	CONCENTRATION	EMISSION RATE	DESTRUCTION
			ppm ^a	lb/hr ^b	EFFICIENCY
1	Inlet	1133 - 1233	440	7.1	96%
	Outlet		3.3 (0.4)	.06	99.2%
2	Inlet	1332 - 1432	460	7.5	96.7%
	Outlet		0.4 (0.5)	0.25	
3	Inlet	1518 - 1618	470	7.8	100.0%
	Outlet		2.0 (0.0)	0.00	
Average	Inlet		460	7.5	98.6%
	Outlet		4.9 (0.6)	0.10	

Note: ^a ppm = Parts per million by volume as heptane. Values in parentheses are blank-corrected.^b lb/hr = Pounds per hour as heptane. Values are based on calculated flow rates and blank-corrected concentrations.^c Efficiency = Based on calculated flow rates and no blank corrections.

TABLE III

SUMMARY OF HAP/CO ₂ EMISSIONS									
POLLUTANT	LOCATION	RUN NO. 1		RUN NO. 2		RUN NO. 3		AVERAGE	
		Concentration ppb	Emission Rate lb/h	Concentration ppb	Emission Rate lb/h	Concentration ppb	Emission Rate lb/h	Concentration ppb	Emission Rate lb/h
Methylene Chloride	Outlet	710	0.104	450	0.065	330	0.045	497	0.070
1,1,1-Trichloro- ethane	Outlet	5.5	0.0012	3.0	0.0009	<5.2	0.0011	<4.0	0.0011

Note: ppb = Parts per billion
 Less than sign indicates detection value limit used.

TABLE IV

SUMMARY OF NO _x EMISSIONS DATA (July 29, 1992)			
RUN NO.	TIME (24-h)	CONCENTRATION ppm	EMISSION RATE lb/h
LVL-0-1A	1545	21	1.5
LVL-0-1B	1700	23	1.7
LVL-0-1C	1715	27	2.0
LVL-0-1D	1730	21	1.5
Average		23	1.7
LVL-0-2A	1745	29	2.2
LVL-0-2B	1800	23	1.7
LVL-0-2C	1815	24	1.8
LVL-0-2D	1830	24	1.8
Average		25	1.9
LVL-0-3A	1845	26	1.9
LVL-0-3B	1900	17	1.3
LVL-0-3C	1915	19	1.2
LVL-0-3D	1930	21	1.5
Average		20	1.5
Overall Average		23	1.7



EMISSION TEST REPORT
PERFORMANCE EVALUATION
NORTH AMERICAN WASTE OF PENNSYLVANIA
ENCLOSED GAS FLARE
AT LAKE VIEW LANDFILL
ERIE, PENNSYLVANIA

July 1992

Prepared by: Brian L. Garls

JTN: 332190

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DEPARTMENT OF ENVIRONMENTAL PROTECTION
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1.0 Introduction

On July 29, 1992, IT Air Quality Services (ITAQSS) personnel conducted an atmospheric emission test program on the McGill enclosed flare system at Lake View Landfill in Erie, Pennsylvania. The purpose of the test program was to satisfy the requirements of the Commonwealth of Pennsylvania. Summa canister samples were collected simultaneously at the inlet and outlet of the flare. Inlet samples were analyzed for nonmethane volatile organic compounds (NMVOC), methane (CH_4), nitrogen (N_2), carbon dioxide (CO_2), carbon monoxide (CO), and oxygen (O_2). Outlet samples were analyzed for NMVOC, CH_4 , N_2 , CO_2 , CO , O_2 , and specific organic compounds (benzene, chloroform, 1,2-dichloroethane, methyl chloride, methylene chloride, perchloroethylene, trichloroethylene, carbon tetrachloride, vinylidene chloride, and vinyl chloride). Testing was also conducted to determine the presence of nitrogen oxide (NO_x) at the flare outlet.

Mr. Denny Luden of Lake View Landfill coordinated on-site activities.

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2.0 Summary of Results

Table 2-1 summarizes the inlet and outlet gas conditions observed during the test program. U.S. Environmental Protection Agency (EPA) Method 2* was used to collect flow rate data at both the inlet and outlet; EPA Method 2C* (standard pitot) could not be used at the inlet because the sampling port was too long. The velocity at the outlet stack, however, was very low. The pressure drop (ΔP) measured across the S-type pitot tube was at or near the level of detection (0.005 in. H_2O) on a 0- to 0.25-in. manometer, thereby increasing potential error in the methodology. Appropriate data were obtained during the sampling program to allow the use of an alternative method for low-flow determination. Procedures similar to those described in Method 2B* were used to determine the flare stack flow rate. Method 2B uses a carbon balance method to calculate the flow rate downstream of a combustion source. Appendix A presents the calculations for this determination.

Flare inlet flow rates averaged 1050 dry standard cubic feet per minute (dscfm) [1170 actual cubic feet per minute (acfm)]. Temperatures averaged 80°F; moisture content averaged 2.4 percent; and CO_2 , O_2 , and CH_4 contents averaged 39.7, 1.0, and 54.7 percent, respectively.

Flare outlet flow rates measured by Method 2 averaged 7500 dscfm (37,600 acfm) compared with 10,600 dscfm determined by carbon balance. Temperatures averaged 1787°F, moisture content averaged 15.0 percent, and CO_2 and O_2 contents averaged 10.4 and 6.2 percent, respectively. The outlet stack emission rates were calculated on the basis of the carbon balance flow-rate value; these results may be biased high. For consistency, the carbon balance flow rates were calculated using the inlet and outlet canister analysis data. Some differences existed between the normalized canister data and the Orsat data from the outlet O_2 and CO_2 . Use of the Orsat data would result in flow rates between the measured and calculated flow rates.

NMTVOC emissions and destruction efficiencies (DE) are summarized in Table 2-2. Concentrations are presented in parts per million (ppm) by volume and milligrams per dry normal cubic meter (mg/dNm^3) as heptane, and emission rates are presented in pounds per hour (lb/h). Inlet concentrations averaged 460 ppm, which corresponds to an emission rate of 7.5 lb/h. Outlet concentrations averaged 4.9 ppm, with a corresponding average emission rate of

* 40 CFR 60, Appendix A.

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Table 2-1
Summary of Gas Conditions (July 29, 1992)

Run No.	Sampling Location	Gas Flow Rate		Temperature (°F)	Moisture (%)	Composition (% by volume, dry basis)				
		acfm ^a	dschm ^b			CO ₂ ^c	C ₂ ^c	CH ₄	N ₂ ^c	CO
VI-1	Inlet	1,150	1,040	80	2.3	39.1	1.3	53.8	5.8	<0.3
VC-1	Outlet	38,400	7,640 ^d (10,800)	1755	18.8	10.4	5.8	<0.1	83.8	<0.1
VI-2	Inlet	1,170	1,050	80	2.4	40.1	0.8	54.9	4.2	<0.3
VC-2	Outlet	37,800	7,420 ^d (10,800)	1620	15.5	10.4	6.4	<0.1	83.2	<0.1
VI-3	Inlet	1,180	1,070	80	2.5	39.8	0.8	55.4	4.0	<0.3
VC-3	Outlet	38,600	7,360 ^d (10,200)	1819	12.7	10.5	6.4	<0.1	83.1	<0.1
Average	Inlet	1,170	1,050	80	2.4	39.7	1.0	54.7	4.7	<0.3
	Outlet	37,800	7,500 ^d (10,800)	1787	15.0	10.4	6.2	<0.1	83.4	<0.1

a acfm = Actual cubic feet per minute.

b dschm = Dry standard cubic feet per minute at 68°F and 29.92 in.Hg.

c Outlet CO₂ and C₂ measured by Orsat, N₂ determined by difference. Inlet composition determined from analyzer samples.

d Values in parentheses are flow rates calculated by carbon balance.

Table 2-2
Summary of NMVOC Emissions and Removal Efficiency
(July 29, 1992)

Run No.	Location	Time (24-h)	Concentration		Emission rate, lb/h ^c	Destruction efficiency, % ^d
			ppm ^a	mg/3Min ^b		
1	Inlet	1133-1233	440	1830	7.1	
	Outlet		3.3 (0.4)	13.7	0.6 (0.05)	92.3 (96.3)
2	Inlet	1332-1432	480	1813	7.5	
	Outlet		4.4 (1.5)	18.3	0.8 (0.17)	89.0 (97.7)
3	Inlet	1518-1618	470	1855	7.8	
	Outlet		2.0 (0.0)	8.3	0.3 (0.0)	95.8 (100.0)
Average	Inlet		460	1800	7.5	
	Outlet		4.6 (0.6)	13.4	0.6 (0.07)	92.7 (98.0)

a ppm = Parts per million by volume in heptane. Values in parentheses are blank-corrected.

b mg/3Min^b = Milligrams per dry normal cubic meter as heptane.

c lb/h = Pounds per hour as heptane. Values in parentheses are based on measured and blank-corrected concentrations.

d Efficiency = $\frac{\text{lb/h in} - \text{lb/h out}}{\text{lb/h in}} \times 100$. Based on calculated flow rates and on blank-corrected values in parentheses are based upon measured flow rates and blank-corrected outlet data.

0.6 lb/h based on the flow rates calculated from the carbon balance. NMVOC DE averaged 92.7 percent. The target efficiency was 98 percent.

Use of the Method 2 flow data to calculate emission rates would result in an average outlet emission rate of 0.38 lb/h, which corresponds to an average destruction efficiency of 94.9 percent. Also shown in Table 2-2 are the results after blank-correcting the outlet samples. An unused canister taken to the field and analyzed with the samples showed a concentration of 2.9 ppm, which is on the same level as the outlet samples, but insignificant with respect to the inlet samples. Using the blank-corrected outlet sample values and the measured volumetric flow rates yields an average DE of 99 percent. Calculation of the DE using the calculated flow rates and blank-corrected sample values results in an average DE of 98.6 percent.

Table 2-3 summarizes the emission data for the specific volatile organic compounds (VOC) that were detected in the samples. At the outlet, methylene chloride was the only compound detected above the detection level. Methylene chloride contamination is not unusual because it is a common laboratory solvent. Methylene chloride was detected in the blank, and it is possible that the sample values are biased high. The compounds 1,1,1-trichloroethane (in the samples), vinyl chloride, and tetrachloroethane (in the blank) were detected at or below the detection level of the method. Table 2-4 presents NO_x emission data. The average emission for NO_x was 1.7 lb/h. Emission rates for specific organics and NO_x are based on the flow rates calculated by carbon balance; results would be lower if the measured flow rates were used.

Calculations can be found in Appendix A, field data are in Appendix B, and laboratory data are in Appendix C.

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Table 2-4
Summary of NCC₂ Emissions Data
(July 29, 1992)

Run No.	Time (24-h)	Concentration, ppm	Emission rate, lb/h
LVL-O-1A	1645	21	1.5
LVL-O-1B	1700	23	1.7
LVL-O-1C	1715	27	2.0
LVL-O-1D	1730	21	1.5
Average		23	1.7
LVL-O-2A	1745	29	2.2
LVL-O-2B	1800	23	1.7
LVL-O-2C	1815	24	1.8
LVL-O-2D	1830	24	1.8
Average		25	1.9
LVL-O-3A	1845	26	1.9
LVL-O-3B	1900	17	1.3
LVL-O-3C	1915	17	1.2
LVL-O-3D	1930	21	1.6
Average		20	1.5
Overall average		23	1.7

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3.0 Quality Assurance

This section outlines the steps ITAQs takes to ensure that quality data are obtained from our testing procedures.

Routine standard reference method quality control (QC) procedures were followed throughout this test series. These included, but were not limited to, the following:

- Calibration of field sampling equipment. Sampling equipment was calibrated in accordance with the procedures of the "Quality Assurance Handbook for Air Pollution Measurement Systems, Volume III," EPA 600/4-72-027B, August 1977. The calibration data are summarized in Table 3-1. Calibration guidelines are described in more detail in Appendix E.
- Onsite dry gas meter and thermocouple digital indicator calibration checks (see Appendix B).
- Train configuration and calculation checks.
- Onsite quality assurance (QA) checks such as sampling train and pitot tube leak checks.
- Use of designated analytical equipment and sampling reagents.
- Laboratory analytical procedure QA checks.

Sampling equipment, reagents, and analytical procedures for this test series followed and met all necessary guidelines set forth for accurate test results. All sampling equipment was calibrated within the limits described in U.S. EPA Method 5.*

The QC procedures followed in the sample analyses in this test program included, but were not limited to, the following:

- Use of designated analytical equipment and experienced laboratory personnel.
- Internal and external audits to ensure accuracy in sampling and analysis.
- Reagent and field blanks to determine blank levels.

* 40 CFR 60, Appendix A, July 1991.

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Table 3-1
Field Equipment Calibration

Equipment	ID. No.	Calibrated Against	Allowable Error	Actual Error	Comments
Water box	FT-10	Wet test meter	± 0.02 and ± 0.15 (± 0.05 γ posttest)	0.004 0.12 0.017	Pretest (maximum deviation) Pretest (maximum deviation) Pre- versus posttest
Pistol tube	505 10000	Critical orifice	± 0.02 and ± 0.15	0.006 0.11	Checked on site prior to testing Checked on site prior to testing
Digital indicator	FT-10	Geometric spirit levels	$\pm$ $\pm$	$\pm$ $\pm$	Visually inspected on site Visually inspected on site
Stick thermometer	103 505	Millivolt signals	± 0.58	0.32 0.52	Pretest (maximum deviation) Checked on site prior to testing
Oreos analyzer	400	ASTM-35	1.52	0.28 0.08	Pretest (maximum deviation) Pretest (maximum deviation)
Impinger thermometer	1-1	Calibration gas	± 0.58 , by volume	0.02 0.22	CO ₂ O ₂
		ASTM-37	± 0.7	0.7	Pretest (maximum deviation)
Balance	402	Types S weights	± 0.5 g	0.0 g	Pretest (maximum deviation)
Barometer	401	MBS-creasable barometer	± 0.10 in. Hg (± 0.20 in. Hg, posttest)	0.01 in. Hg 0.03 in. Hg	Pretest Posttest
Dry gas thermometer	FT-10	ASTM-37	± 0.7	0.7 1.7	Pretest (maximum deviation) Posttest (maximum deviation)

^a See Appendix E.

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3.1 Nonmethane Organics

NMVOCs were not detected in the laboratory blank (<0.01 ppm), but were detected in the field blank at 2.9 ppm as heptane. This is significant when compared with the outlet samples; therefore, results were calculated using both measured and blank-corrected concentrations.

3.2 Specific VOCs

The field blank for TO-14 showed concentrations of vinyl chloride, methylene chloride, and tetrachloroethene at or below the detection limit.

3.3 Permanent Gases

Neither CH_4 , CO_2 , CO , N_2 , nor O_2 were detected in the field or laboratory blanks.

3.4 Nitrogen Oxides

NO_x was not detected in the reagent blank.

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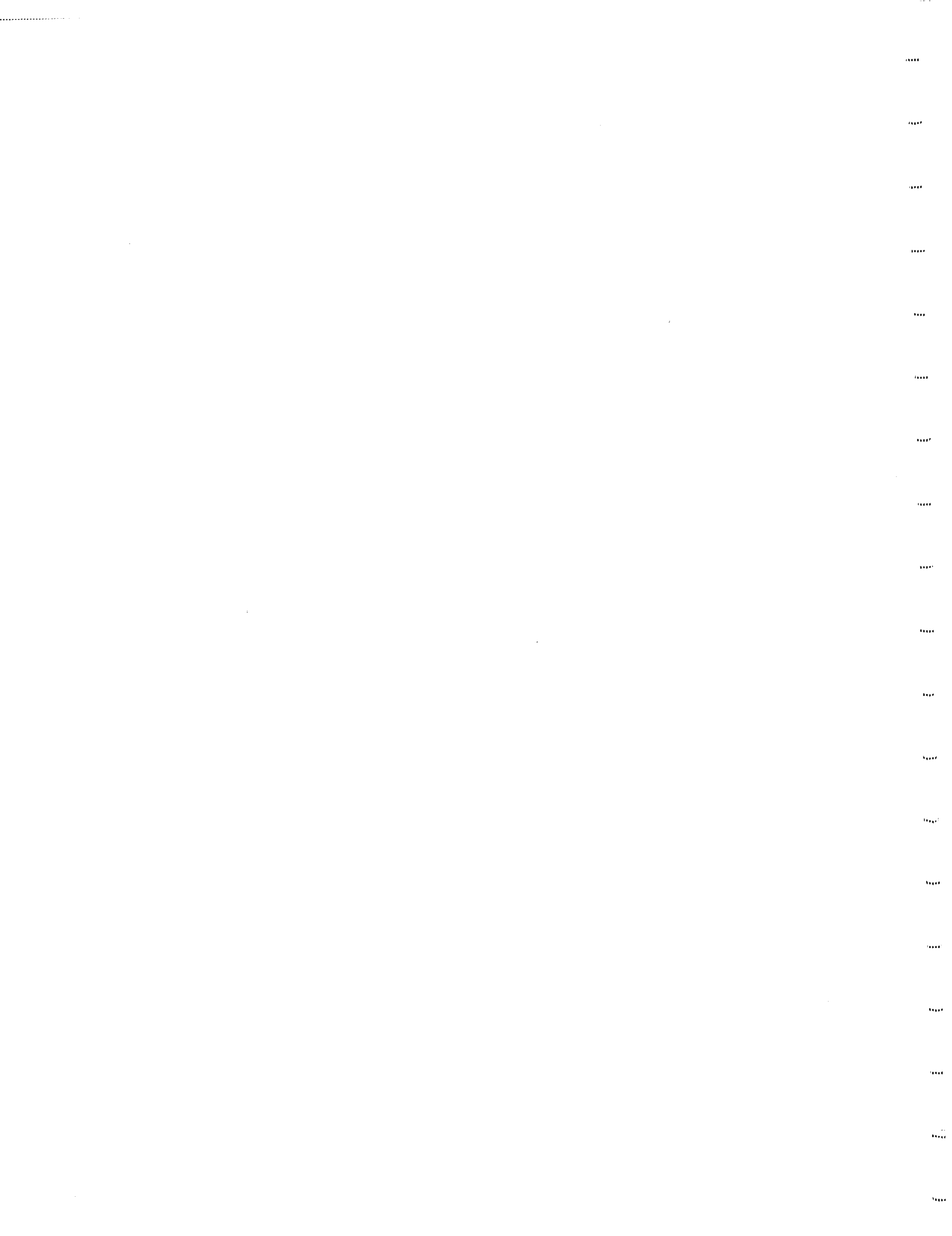
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4.0 Process Description

Lake View Landfill operates an enclosed ground flare to control emissions of CH_4 and NMVOC contained in the landfill gas collected at their disposal site in Erie, Pennsylvania. Landfill gas is combusted through a burner system equipped with a flame arrestor to prevent flashback in the event the gas flow is lost. Combustion air is controlled by two sets of automatic dampers located near the bottom of the flare assembly.



5.0 Sampling Locations and Test Methods

5.1 Sampling Locations

Sampling was conducted at the flare inlet and outlet as shown in Figures 5-1 and 5-2, respectively. At the inlet, two sampling ports were located approximately 9.2 duct diameters downstream and 4.4 duct diameters upstream from the nearest flow disturbances in the 11.5-in. inside-diameter (i.d.) duct. Eight points were used to traverse the duct for velocity measurements. At the outlet, two sampling ports were located approximately 3.1 stack diameters downstream and 0.5 stack diameter upstream from the nearest flow disturbances in the 9-ft 8-in.-i.d. stack. Sixteen points were used to traverse the duct for velocity measurements.

5.2 Test Methods

5.2.1 Canister Samples

Canister samples were collected simultaneously at the inlet and outlet of the flare. Each sample was collected in a passivated canister in accordance with procedures described in Draft U.S. EPA Method 25C.* The canister samples were then analyzed using Method 25C procedures, but the detection limit was not low enough to allow calculation of the destruction efficiency. The samples were subsequently analyzed using U.S. EPA Method TO-12,** which did achieve an acceptable detection limit. During TO-12 analysis, the sample is drawn through a cryogenic trap, capturing the NMVOC. The NMVOC sample is heated, driving off the captured compounds, which are then measured by the GC/FID. Both sets of samples were also analyzed for CH_4 , N_2 , O_2 , CO_2 , and CO by a gas chromatograph with a thermal conductivity detector (GC/TCD). It should be noted that the method of detection used in TO-12 is different from the method stated in 25C. In 25C, the NMVOC compounds are oxidized to CO_2 , reduced to methane, and measured by an FID. No oxidation/reduction process takes place in TO-12.

The outlet samples were also analyzed for specific organic compounds*** by procedures described in U.S. EPA Method TO-14.***

* 56 FR 24468, May 30, 1991.

** Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air, EPA-600/4-84-041.

*** Benzene, chloroform, 1,2-dichloroethane, methyl chloride, methylene chloride, perchloroethylene, trichloroethylene, carbon tetrachloride, vinylidene chloride, vinyl chloride.

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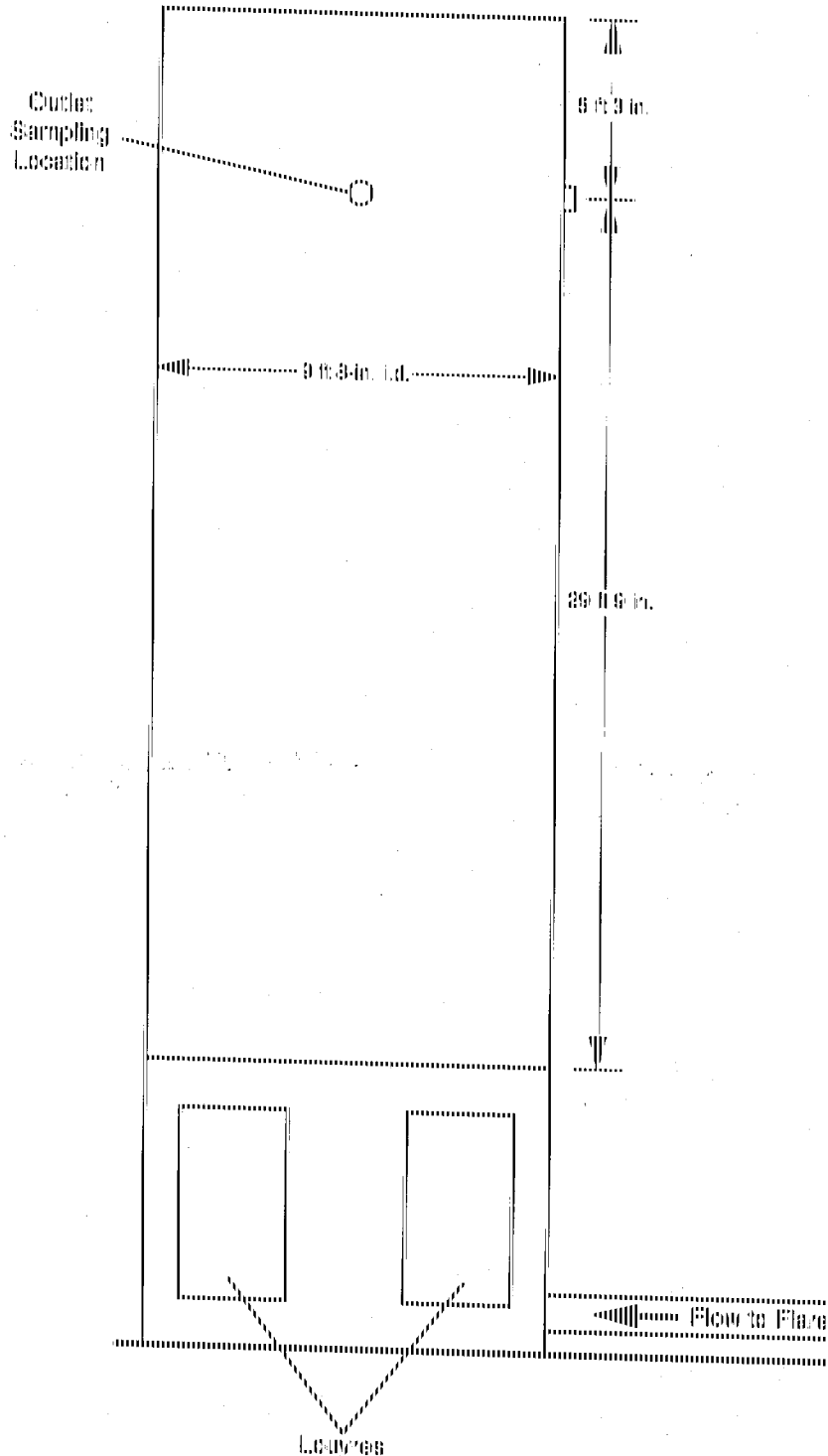
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DRAWING	BY	CHECKED BY	DATE	DRAWING NO.	DATE
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Figure 5-2.
Flare Outlet Sampling Location.

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Inlet and outlet samples were also analyzed for CH_4 , N_2 , O_2 , CO_2 , and CO by GC/TCD.

5.2.2 Nitrogen Oxides

The NO_x sampling was conducted at the outlet in accordance with U.S. EPA Method 7A.* Three tests were conducted, each consisting of four grab samples. The samples were collected in evacuated flasks containing a dilute sulfuric acid/hydrogen peroxide absorbing solution. The NO_x was measured by ion chromatography.

5.2.3 Velocity, Gas Temperature, Molecular Weight, and Moisture

At the inlet, flow rate and temperature measurements were obtained in accordance with U.S. EPA Method 2C* by use of a Type-S standard pitot tube, an inclined manometer, and a thermocouple and digital readout. Gas composition and molecular weight were determined by ASTM Method D1945.** The canister sample analyses described earlier (CH_4 , N_2 , O_2 , CO_2 , and CO by GC/TCD) were used to calculate molecular weight. Moisture determinations were made by use of a wet bulb/dry bulb thermometer and psychrometric chart.

At the outlet, velocity and temperature measurements were obtained in accordance with U.S. EPA Method 2 by use of a Type-S pitot tube, an inclined manometer, and a thermocouple and digital readout. During the sampling program, pressure drops across the pitot tube were extremely low (<0.01 in. H_2O). Such low pressure drops greatly decreased the accuracy of the method. Therefore, an alternative method was used to represent the flow data in addition to the data collected by Method 2 procedures. The outlet flow rate was determined by carbon balance in accordance with U.S. EPA Method 2B.

Levels of CO_2 and O_2 were measured in accordance with U.S. EPA Method 3,*** and stack gas moisture was measured in accordance with U.S. EPA Method 4,***

Appendix D contains more detailed descriptions of the sampling and analytical procedures used during stack gas testing.

* 40 CFR 60, Appendix A.

** Annual Book of ASTM Standards, Section 5, Volume 05.05.

*** 40 CFR 60, Appendix A.

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APPENDIX A

CALCULATIONS

Nomenclature and Abbreviations

A_{st}	=	Cross-sectional area of stack, sq. ft.
B_{wet}	=	Proportion by volume of water vapor in the gas stream, dimensionless
C_{NOx}	=	NO_x concentration, ppm
C_{sp}	=	Pitot tube coefficient, dimensionless
q_{CO}	=	Percent of carbon monoxide by volume, dry basis
q_{CO_2}	=	Percent of carbon dioxide by volume, dry basis
ΔH	=	Average pressure drop across the sampling meter flow orifice, inches of water (in. H ₂ O)
L_{le}	=	Maximum acceptable leakage rate for either a pretest leak check or for a leak check following a component change; equal to 0.030 cubic foot per minute or 4% of the average sampling rate, whichever is less
M_{drl}	=	Dry molecular weight, lb/lb-mole
M_{kg}	=	Molecular weight of stack gas (wet basis), lb/lb-mole
q_{N_2}	=	Percent of nitrogen by volume, dry basis
q_{O_2}	=	Percent of oxygen by volume, dry basis
ΔP	=	Velocity head of stack gas, inches of water (in. H ₂ O)
P	=	Flank pressure, in. Hg
P_{bar}	=	Barometric pressure, inches of mercury (in. Hg)
P_{mox}	=	Micrograms NO_x , μg
P_{st}	=	Absolute stack gas pressure, inches of mercury (in. Hg)
P_{std}	=	Gas pressure at standard conditions, inches of mercury (30.00 in. Hg)
Q_{ps}	=	Volumetric flow rate - wet basis at stack conditions, actual cubic feet per minute (acfm)
Q_{std}	=	Volumetric flow rate - dry basis at standard conditions, dry standard cubic feet per minute (dscfm)

Fluorocarbon and Chlorofluorocarbon (continued)

(continued)

- T = Flask temperature, °F
- T_{dm} = Average temperature of dry gas meter, °R
- T_s = Average temperature of stack gas, °R
- T_{std} = Temperature at standard conditions, (528°R)
- V_F = Flask volume, ml
- V_{lc} = Total volume of liquid collected in impingers and silica gel, ml
- V_m = Volume of dry gas sampled at meter conditions, cu. ft.
- V_{mstd} = Volume of dry gas sampled at standard conditions, cu. ft.
- V_{sacc} = N₂O sample volume, ml
- V_s = Average stack gas velocity at stack conditions, ft/s
- V_{wstd} = Volume of water vapor at standard conditions, scf
- K = Dry gas meter calibration factor, dimensionless
- Φ = Total sampling time, minutes

NOTE: Standard condition = 68°F and 29.92 in.Hg

Example Calculations For Pollutant Emissions

1. Volume of dry gas sampled corrected to standard conditions, ft³. Note: V_m must be corrected for leakage if any leakage rates exceed 1a.

$$V_{mstd} = 17.647 \times V_m \times \left[\frac{P_{bar} + \frac{AME}{13.6}}{P_m} \right]$$

2. Volume of water vapor at standard conditions, ft³.

$$V_{wstd} = 0.04707 \times W_e$$

3. Moisture content in stack gas, dimensionless.

$$B_{ws} = \frac{V_{wstd}}{V_{wstd} + V_{mstd}}$$

4. Dry molecular weight of stack gas, lb/lb-mole.

$$M_d = 0.44(\%CO_2) + 0.32(\%O_2) + 0.28(\%N_2 + \%CO)$$

5. Molecular weight of stack gas, lb/lb-mole.

$$M_s = M_d(1 - B_{ws}) + 18B_{ws}$$

6. Stack velocity at stack conditions, ft/s.

$$V_s = (85.49)(C_p)(\text{avg } \sqrt{\Delta P}) \sqrt{\frac{T_B}{(P_B)(M_s)}}$$

7. Stack gas volumetric flow rate at stack conditions, acfm.

$$Q_s = 60 \times V_s \times A_s$$

8. Dry stack gas volumetric flow rate at standard conditions, dscfm.

$$Q_{sstd} = (17.647)(Q_s)\left(\frac{P_B}{P_{std}}\right)(1 - B_{ws})$$

Example Calculations for Pollutant Emissions

UNITED FUEL CORRECTION FACTORS

$$17.647 = \left(\frac{\text{Total}}{\text{Fuel}} \right)$$

$$0.04707 = \left(\frac{\text{lb}^3}{\text{m}^3} \right)$$

$$0.44 = \text{molecular weight of CO}_2/100$$

$$0.32 = \text{molecular weight of O}_2/100$$

$$0.28 = \text{molecular weight of N}_2/100$$

$$18 = \text{molecular weight of water (H}_2\text{O), lb/lb-mole}$$

$$85.49 = \left[\frac{(\text{lb/lb-mole})(\text{in. Hg})}{(^{\circ}\text{F})(\text{in. H}_2\text{O})} \right]^{\frac{1}{4}} \times 49$$

NO_x CALCULATIONS

$$\text{Equation 1: } V_a = 17.647 \times (V_f - 25) \times \left[\left(\frac{P_b - P}{T + 460} \right)_i - \left(\frac{P_b - P}{T + 460} \right)_f \right]$$

Where:

- V_s = Sample volume, ml
- V_f = Flask volume, ml
- P_b = Barometric pressure, in.Hg
- P = Flask pressure, in.Hg
- T = Flask temperature, °F

Note: Subscripts f and i indicate final and initial, respectively.

$$\text{Equation 2: } C_{\text{NO}_x} = \frac{522.5 \times P_{\text{NO}_x}}{V_a}$$

Where:

- C_{NO_x} = NO_x concentration, ppm
- P_{NO_x} = micrograms NO_x, µg

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$$\text{Equation 3: } C_{\text{NO}_x}^* = \frac{6.243 \times 10^{-5} \times P_{\text{NO}_x}}{V_0}$$

Where:

$C_{\text{NO}_x}^*$ = NO_x concentration, lb/dscf

$$\text{Equation 4: } E_{\text{NO}_x} = C_{\text{NO}_x}^* \times Q_{\text{gas}} \times 60$$

Where:

E_{NO_x} = NO_x emission rate, lb/h

$Q_{\text{gas}}^{\text{std}}$ = Volumetric flow rate, dscfm

11/1/00



By E.C. 12 Date 3/11/61 Subject Molecular Weight of Talc Sheet No. 1 of 1
 Chkd. By J. H. 12 Date 11/2/61 Mr. T Proj. No. 332160

0.5cm. X 0.5cm.

Molecular Weight = $.32 \times 2002 + .27 \times 2002 + .16 \times 2004 + .25 \times 2002$

LWL - 1 - 1 = 227.85

LWL - 1 - 2 = 227.86

LWL - 1 - 3 = 227.75

Avg = 227.82

Total counter ^{Lab} results used.

VAPOR PRESSURES OF WATER AT SATURATION (From Red Book 1963)

Temp. Deg. F.	0	1	2	3	4	5	6	7	8	9
0	0.006	0.007	0.008	0.009	0.010	0.011	0.012	0.013	0.014	0.015
10	0.018	0.019	0.020	0.021	0.022	0.023	0.024	0.025	0.026	0.027
20	0.031	0.032	0.033	0.034	0.035	0.036	0.037	0.038	0.039	0.040
30	0.046	0.047	0.048	0.049	0.050	0.051	0.052	0.053	0.054	0.055
40	0.063	0.064	0.065	0.066	0.067	0.068	0.069	0.070	0.071	0.072
50	0.083	0.084	0.085	0.086	0.087	0.088	0.089	0.090	0.091	0.092
60	0.107	0.108	0.109	0.110	0.111	0.112	0.113	0.114	0.115	0.116
70	0.136	0.137	0.138	0.139	0.140	0.141	0.142	0.143	0.144	0.145
80	0.170	0.171	0.172	0.173	0.174	0.175	0.176	0.177	0.178	0.179
90	0.209	0.210	0.211	0.212	0.213	0.214	0.215	0.216	0.217	0.218
100	0.255	0.256	0.257	0.258	0.259	0.260	0.261	0.262	0.263	0.264
110	0.308	0.309	0.310	0.311	0.312	0.313	0.314	0.315	0.316	0.317
120	0.369	0.370	0.371	0.372	0.373	0.374	0.375	0.376	0.377	0.378
130	0.438	0.439	0.440	0.441	0.442	0.443	0.444	0.445	0.446	0.447
140	0.516	0.517	0.518	0.519	0.520	0.521	0.522	0.523	0.524	0.525
150	0.604	0.605	0.606	0.607	0.608	0.609	0.610	0.611	0.612	0.613
160	0.702	0.703	0.704	0.705	0.706	0.707	0.708	0.709	0.710	0.711
170	0.811	0.812	0.813	0.814	0.815	0.816	0.817	0.818	0.819	0.820
180	0.923	0.924	0.925	0.926	0.927	0.928	0.929	0.930	0.931	0.932
190	1.048	1.049	1.050	1.051	1.052	1.053	1.054	1.055	1.056	1.057
200	1.187	1.188	1.189	1.190	1.191	1.192	1.193	1.194	1.195	1.196
210	1.341	1.342	1.343	1.344	1.345	1.346	1.347	1.348	1.349	1.350
220	1.512	1.513	1.514	1.515	1.516	1.517	1.518	1.519	1.520	1.521
230	1.702	1.703	1.704	1.705	1.706	1.707	1.708	1.709	1.710	1.711
240	1.913	1.914	1.915	1.916	1.917	1.918	1.919	1.920	1.921	1.922
250	2.148	2.149	2.150	2.151	2.152	2.153	2.154	2.155	2.156	2.157
260	2.410	2.411	2.412	2.413	2.414	2.415	2.416	2.417	2.418	2.419
270	2.703	2.704	2.705	2.706	2.707	2.708	2.709	2.710	2.711	2.712
280	3.030	3.031	3.032	3.033	3.034	3.035	3.036	3.037	3.038	3.039
290	3.395	3.396	3.397	3.398	3.399	3.400	3.401	3.402	3.403	3.404
300	3.801	3.802	3.803	3.804	3.805	3.806	3.807	3.808	3.809	3.810
310	4.152	4.153	4.154	4.155	4.156	4.157	4.158	4.159	4.160	4.161
320	4.552	4.553	4.554	4.555	4.556	4.557	4.558	4.559	4.560	4.561
330	4.996	4.997	4.998	4.999	5.000	5.001	5.002	5.003	5.004	5.005
340	5.480	5.481	5.482	5.483	5.484	5.485	5.486	5.487	5.488	5.489
350	5.999	6.000	6.001	6.002	6.003	6.004	6.005	6.006	6.007	6.008
360	6.558	6.559	6.560	6.561	6.562	6.563	6.564	6.565	6.566	6.567
370	7.153	7.154	7.155	7.156	7.157	7.158	7.159	7.160	7.161	7.162
380	7.791	7.792	7.793	7.794	7.795	7.796	7.797	7.798	7.799	7.800
390	8.469	8.470	8.471	8.472	8.473	8.474	8.475	8.476	8.477	8.478
400	9.194	9.195	9.196	9.197	9.198	9.199	9.200	9.201	9.202	9.203
410	9.964	9.965	9.966	9.967	9.968	9.969	9.970	9.971	9.972	9.973
420	10.776	10.777	10.778	10.779	10.780	10.781	10.782	10.783	10.784	10.785
430	11.638	11.639	11.640	11.641	11.642	11.643	11.644	11.645	11.646	11.647
440	12.548	12.549	12.550	12.551	12.552	12.553	12.554	12.555	12.556	12.557
450	13.504	13.505	13.506	13.507	13.508	13.509	13.510	13.511	13.512	13.513
460	14.504	14.505	14.506	14.507	14.508	14.509	14.510	14.511	14.512	14.513
470	15.548	15.549	15.550	15.551	15.552	15.553	15.554	15.555	15.556	15.557
480	16.634	16.635	16.636	16.637	16.638	16.639	16.640	16.641	16.642	16.643
490	17.761	17.762	17.763	17.764	17.765	17.766	17.767	17.768	17.769	17.770
500	18.928	18.929	18.930	18.931	18.932	18.933	18.934	18.935	18.936	18.937
510	20.134	20.135	20.136	20.137	20.138	20.139	20.140	20.141	20.142	20.143
520	21.378	21.379	21.380	21.381	21.382	21.383	21.384	21.385	21.386	21.387
530	22.659	22.660	22.661	22.662	22.663	22.664	22.665	22.666	22.667	22.668
540	23.976	23.977	23.978	23.979	23.980	23.981	23.982	23.983	23.984	23.985
550	25.328	25.329	25.330	25.331	25.332	25.333	25.334	25.335	25.336	25.337
560	26.714	26.715	26.716	26.717	26.718	26.719	26.720	26.721	26.722	26.723
570	28.134	28.135	28.136	28.137	28.138	28.139	28.140	28.141	28.142	28.143
580	29.587	29.588	29.589	29.590	29.591	29.592	29.593	29.594	29.595	29.596
590	31.073	31.074	31.075	31.076	31.077	31.078	31.079	31.080	31.081	31.082
600	32.591	32.592	32.593	32.594	32.595	32.596	32.597	32.598	32.599	32.600
610	34.140	34.141	34.142	34.143	34.144	34.145	34.146	34.147	34.148	34.149
620	35.719	35.720	35.721	35.722	35.723	35.724	35.725	35.726	35.727	35.728
630	37.328	37.329	37.330	37.331	37.332	37.333	37.334	37.335	37.336	37.337
640	38.966	38.967	38.968	38.969	38.970	38.971	38.972	38.973	38.974	38.975
650	40.633	40.634	40.635	40.636	40.637	40.638	40.639	40.640	40.641	40.642
660	42.328	42.329	42.330	42.331	42.332	42.333	42.334	42.335	42.336	42.337
670	44.051	44.052	44.053	44.054	44.055	44.056	44.057	44.058	44.059	44.060
680	45.791	45.792	45.793	45.794	45.795	45.796	45.797	45.798	45.799	45.800
690	47.548	47.549	47.550	47.551	47.552	47.553	47.554	47.555	47.556	47.557
700	49.321	49.322	49.323	49.324	49.325	49.326	49.327	49.328	49.329	49.330
710	51.109	51.110	51.111	51.112	51.113	51.114	51.115	51.116	51.117	51.118
720	52.912	52.913	52.914	52.915	52.916	52.917	52.918	52.919	52.920	52.921
730	54.729	54.730	54.731	54.732	54.733	54.734	54.735	54.736	54.737	54.738
740	56.560	56.561	56.562	56.563	56.564	56.565	56.566	56.567	56.568	56.569
750	58.404	58.405	58.406	58.407	58.408	58.409	58.410	58.411	58.412	58.413
760	60.261	60.262	60.263	60.264	60.265	60.266	60.267	60.268	60.269	60.270
770	62.129	62.130	62.131	62.132	62.133	62.134	62.135	62.136	62.137	62.138
780	64.008	64.009	64.010	64.011	64.012	64.013	64.014	64.015	64.016	64.017
790	65.898	65.899	65.900	65.901	65.902	65.903	65.904	65.905	65.906	65.907
800	67.798	67.799	67.800	67.801	67.802	67.803	67.804	67.805	67.806	67.807
810	69.708	69.709	69.710	69.711	69.712	69.713	69.714	69.715	69.716	69.717
820	71.627	71.628	71.629	71.630	71.631	71.632	71.633	71.634	71.635	71.636
830	73.555	73.556	73.557	73.558	73.559	73.560	73.561	73.562	73.563	73.564
840	75.491	75.492	75.493	75.494	75.495	75.496	75.497	75.498	75.499	75.500
850	77.434	77.435	77.436	77.437	77.438	77.439	77.440	77.441	77.442	77.443
860	79.384	79.385	79.386	79.387	79.388	79.389	79.390	79.391	79.392	79.393
870	81.340	81.341	81.342	81.343	81.344	81.345	81.346	81.347	81.348	81.349
880	83.301	83.302	83.303	83.304	83.305	83.306	83.307	83.308	83.309	83.310
890	85.267	85.268	85.269	85.270	85.271	85.272	85.273	85.274	85.275	85.276
900	87.238	87.239	87.240	87.241	87.242	87.243	87.244	87.245	87.246	87.247
910	89.213	89.214	89.215	89.216	89.217	89.218	89.219	89.220	89.221	89.222
920	91.192	91.193	91.194	91.195	91.196	91.197	91.198	91.199	91.200	91.201
930	93.174	93.175	93.176	93.177	93.178	93.179	93.180	93.181	93.182	93.183
940	95.159	95.160	95.161	95.162	95.163	95.164	95.165	95.166	95.167	95.168
950	97.146	97.147	97.148	97.149	97.150	97.151	97.152	97.153	97.154	97.155
960	99.135	99.136	99.137	99.138	99.139	99.140	99.141	99.142	99.143	99.144
970	101.126	101.127	101.128	101.129	101.130	101.131	101.132	101.133	101.134	101.135
980	103.118	103.119	103.120	103.121	103.122	103.123	103.124	103.125	103.126	103.127
990	105.111	105.112	105.113	105.114	105.115	105.116	105.117	105.118	105.119	105.120
1000	107.105	107.106	107.107	107.108	107.109	107.110	107.111	107.112	107.113	107.114

Method for Determination of Vapor Pressure, Density, and Mass Content of Gases, Bulletin
 W.P. 50, American Petroleum Institute, New York, N.Y.

of Distillation
 (decimal)
 partial pressure (3) scale factor
 scale pressure

UP AND QUALITY SERVICES VELOCITY INTERMEDIATIONS

03/04/02 11:00

PLANT : Lake View Ranch
SAMPLE LOCATION : Photo Inlet
EQUIDISTANT : 1500-1500

DATE : 7/25/92
TEMPERATURE : 1300
CLOUDS : 100/100
PERCENT HUMIDITY : 2.3
WIND DIRECTION, deg. W : 109.87
NO. OF TRAVERSE PTH. : 8

BAROMETRIC PRESS., in. Hg : 29.96
SATUR. PRESS., in. Hg : 31.6
EFFECT WIND, Cp : 0.84
PERCENT COH : 97.0
PERCENT COH : 1.2

TRAVERSE PTH. NO.	VELOCITY HEAD, in. Hg	SEAKER TEMP., deg. W
1	0.18	78
2	0.19	79
3	0.22	80
4	0.21	79
5	0.17	78
6	0.19	78
7	0.19	78
8	0.18	78
	0.18	78

03/04/02 11:00

PIPE AND QUALITY SERVICES
VELOCITY DETERMINATION

dated 11/20/90

PLANT : Lake View Landfill
SAMPLE LOCATION : Flare Hole
HOLE NUMBER : 1571-1572

DATE : 7/29/92
TIME(SH-HR) : 1344
OPERATOR(S) : DODD
PERCENT MOISTURE : 2.2
WET WEIGHT, G/G, IN : 109.87
NO. OF TRAVELING PIPES : 8

BAROMETRIC PRESS., In. Hg : 29.9
STATION PRESS., In. Hg : 31
PIPETT TUBE, Gp : 0.84
PIPETT COB : 87.6
PIPETT CR : 1.0

TRAVELING PIP. NO.	VELOCITY HEAD, In. H ₂ O	STATION THICK., deg. F
1	0.17	80
2	0.19	80
3	0.24	80
4	0.23	80
5	0.21	70
6	0.21	70
7	0.21	70
8	0.18	80
	0.21	80

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PT AND QUALITY SERVICES WIRELESS TELECOMMUNICATION

04/04/00

PLANT : Radio View Landfill
SAMPLE LOCATION : Phase 1 Lot
RUN NUMBER : LEVEL-VR

DATE :	7/20/92	BAROMETRIC PRESS., in. Hg :	29.94
TEMPERATURE :	14.55	WIND SPEED, in. Hg :	21.0
COEFFICIENT :	100.00		
PERCENTAGE DISCREPANCY :	3.6	PERCENTAGE COR :	0.84
WIND DIRECTION, deg. Hg :	109.87	PERCENTAGE COR :	8.0
NO. OF TRAVERSE Pts :	8	PERCENTAGE COR :	0.8

TRAVERSE PTE. NO.	WIND SPEED, in. Hg	BLADE THICK, deg. Hg
1	0.15	82
2	0.18	80
3	0.24	78
4	0.22	78
5	0.19	79
6	0.22	79
7	0.23	79
8	0.20	79
	0.21	80

RT AIR QUALITY SURVEYING VELOCITY DETERMINATION

03/04/00

PLANT : Lake View Landfill
 SAMPLE LOCATION : Flare Inlet
 RUN NUMBER : LVL-144

DATE : 7/28/92
 TIME : 1024
 OPERATOR : JCM/DC
 PERCENTAGE RECOVERY : 2.3
 SCALES AREA, sq. in. : 103.67
 NO. OF TRAVERSES PER : 8

BAROMETRIC PRESS., in. Hg : 29.98
 STATIC PRESS., in. Hg : 29.9
 PIPOT TUBE, C/P : 0.64
 PERCENT C/P : 88.9
 PERCENT OF : 0.6

TRAVERSE P.T. NO.	VELOCITY READ, in. Hg	SCALE TRUMP, deg. F
1	0.17	80
2	0.21	80
3	0.24	79
4	0.29	80
5	0.20	79
6	0.23	80
7	0.23	79
8	0.23	80
	0.23	80

TEST AND QUALITY SERVICES VELOCITY DETERMINATIONS

PLACENT: SAMPLE LOCATION:	Lake View Landfill Place Inlet Run Nos	LEVEL-1-V1	LEVEL-1-V2	LEVEL-1-V3	LEVEL-1-V4	AVERAGE	
SPACE PRESSURE, kg Hg :		26.37	26.36	26.33	26.32	26.34	
SPACE TEMP., deg. F :		70	60	60	60	70	
MOLECULAR WEIGHT, DMET :		27.75"	27.76	27.57	27.75"	27.72	
MOLECULAR WEIGHT, STACHE :		27.62	27.64	27.55"	27.52	27.58"	
AVG. SIGHT VELOCITY HEAD :		0.44	0.45	0.45	0.46	0.44	
VELOCITY, ft/s :		26.64	26.74	27.60	27.74	26.94	
ACTUAL CUMUL FREQU MINUTE :		12.7	16.4	14.6	15.8	16.4	
DEF STANDARD CUMUL FREQU MINUTE :		102.2	105.4	105.2	104.3	105.3	

15/11/10
10/11/10

SEE PAGE 4-8

SEE FREQUENCY SUBSET calculations page 8-10

15/11/10
10/11/10



By EGG Date 10/4/91 Subject Oilfield flares by Chad Sheet No. 1 of 1
 Chd. By EGG Date 10/4/91 EGG Proj. No. 222192
 0.5cm. X 0.5cm.

$$Q_{std_{out}} = Q_{std_{in}} \frac{HC_{in} + CO_{2\ in} + CO_{in}}{HC_{out} + CO_{2\ out} + CO_{out} + 300}$$

- Where: $Q_{std_{out}}$ = volumetric flow rate of stack gas, dry standard cubic feet per minute, duct
- $Q_{std_{in}}$ = volumetric flow rate of inlet gas measured by EPA method 2, duct (average flow before restrictor)
- HC_{in} = concentration of hydrocarbons in inlet gas, ppm as methane (methane and non-HC converted to methane)
- $CO_{2\ in}$ = concentration of carbon dioxide in inlet gas, ppm
- CO_{in} = concentration of carbon monoxide in inlet gas, ppm
- HC_{out} = concentration of methane and non-HC (as methane) in stack gas, ppm
- $CO_{2\ out}$ = concentration of carbon dioxide in stack gas, ppm
- CO_{out} = concentration of carbon monoxide in stack gas, ppm
- 300 = estimated concentration of carbon dioxide in ambient air, ppm

Run No. LVL - 0 - 1

$$Q_{std_{out}} = (1033) \frac{591,000 + 391,000 + 0}{23.1 + 90,000 + 0 + 300} = 10,283 \text{ duct}$$

Run No. LVL - 0 - 2

$$Q_{std_{out}} = (1053) \frac{522,200 + 401,000 + 0}{30.8 + 92,000 + 0 + 300} = 10,942 \text{ duct}$$

Run No. LVL - 0 - 3

$$Q_{std_{out}} = (1068) \frac{552,200 + 398,000 + 0}{14 + 100,000 + 0 + 300} = 10,232 \text{ duct}$$

ICE AND QUALITY ABBREVIATIONS VELOCITY DETERMINATION

midland 10700

PLANT : Lake View Landfill
SAMPLE LOCATION : Phase Outlet
EQU NUMBER : LVL-C-V1

DATE :	7/29/92	BAROMETERIC PRESS., In. Hg :	29.94
TEMPERATURE :	10.09	BAROMETRIC PRESS., In. Hg :	29.94
OPERATOR :	DG/DC		
PERCENT MOISTURE :	16.8	TEMPERATURE, C :	10.4
WET AREA, Sq. IN. :	10568.82	PERCENT C/D :	5.8
NO. OF TRAVERSE PTH. :	16	PERCENT C/D :	

TRAVERSE PTH. NO.	VELOCITY HEAD, In. H ₂ O	STACHE HEAD, deg. F
1	0.009	1689
2	0.009	1679
3	0.007	1687
4	0.007	1709
5	0.007	1714
6	0.006	1718
7	0.009	1704
8	0.005	1676
9	0.006	1679
10	0.006	1681
11	0.006	1689
12	0.007	1702
13	0.007	1708
14	0.007	1685
15	0.005	1686
16	0.009	1674
0.0067		1688

7/11/92

ET ALER QUALITY SERVICES VELOCITY IDENTIFICATION

revised 11/10/00

PLANT : Lake View Landfill
SAMPLER LOCATION : Phase Outlet
RUN NUMBER : LVL-C-02

DATE : 7/29/98
TIME (24-HR) : 12:55
OPERATOR(S) : BCD/C
PERCENT MOISTURE : 10.1
STACK AREA, sq. ft. : 10668.32
NO. OF TELEVERSE PTS. : 16

BAROMETRIC PRESS., in. Hg : 29.9
STATIC PRESS., in. Hg : -0.073
PITOT TUBE, C/P : 0.84
PERCENT CO2 : 10.4
PERCENT CO : 6.1

TELEVERSE PRT. NO.	VELOCITY HEAD, in. H2O	STACK TEMPR., deg. F
1	0.003	1803
2	0.004	1811
3	0.004	1829
4	0.007	1846
5	0.007	1836
6	0.006	1820
7	0.004	1806
8	0.004	1799
9	0.005	1797
10	0.006	1806
11	0.007	1816
12	0.007	1820
13	0.008	1850
14	0.007	1831
15	0.005	1810
16	0.004	1804
	0.006	1813

8/11/98
ajf/ajf

TEMP AND QUALITY SERVICES VELOCITY DETERMINATION

calculated 11/00

PLACENT : Lake View Landfill
SAMPLE LOCATION : Phase Outlot
RECT NUMBER : LVL-0579

DATE : 7/20/02
TIME(SM-HEB) : 1438
CUMULATIVE : 10000
PERCENT DECOMPOSITION : 14.1
BLAKE AREA, sq. ft. : 10666.32
NO. OF THERMISTE PROB. : 16

BAROMETRIC PRESS., in. Hg : 29.99
STATIC PRESS., in. Hg : -0.070
WETBULB, C/F : 6.94
PERCENT COB : 10.5
PERCENT COB : 6.4

TEMPERATURE TPC, NO.	VELOCITY HEAD, in. H ₂ O	STATIC PRESS., in. H ₂ O
1	0.000	1800
2	0.005	1807
3	0.006	1821
4	0.006	1849
5	0.007	1862
6	0.005	1844
7	0.004	1810
8	0.005	1791
9	0.004	1766
10	0.004	1800
11	0.006	1827
12	0.007	1851
13	0.006	1846
14	0.005	1822
15	0.005	1814
16	0.004	1804
	0.005	1808

PIPE AND QUALITY SERVICES VELOCITY DETERMINATION

04/04/2004

PLANT : Lado View Landfill
SAMPLE LOCATION : Pipe Outlet
RUN NUMBER : 1571-0-04

DATE :	7/28/02	BAROMETRIC PRESS., In. Hg :	29.96
TEMPERATURE :	1636	STATIC PRESS., In. Hg :	-0.070
CORRECTOR :	DC/DC		
PERCENT MOISTURE :	12.7	PIPING TIME, Cp :	0.84
PEACH AREA, Sq. Ft. :	10588.82	PERCENT COE :	10.5
NO. OF TRAVERSE PTS. :	16	PERCENT CR :	6.4

TRAVERSE PNT. NO.	VELOCITY HEAD, In. H2O	PEACH TIDEP, dog. F
1	0.005	1604
2	0.006	1708
3	0.006	1603
4	0.006	1614
5	0.006	1620
6	0.005	1661
7	0.005	1622
8	0.005	1778
9	0.005	1756
10	0.005	1609
11	0.006	1616
12	0.006	1627
13	0.006	1646
14	0.006	1641
15	0.005	1619
16	0.004	1601
	0.005	1605

TEST AND QUALITY SERVICES VELOCITY DETERMINATION

955000 110.90

PLANT :	ECHO VIEW LUNDEN	SAMPLE LOCATION :	Place Outlet	Recon No:	LEVEL-C-571	LEVEL-C-572	LEVEL-C-573	LEVEL-C-574	AVERAGE	
		STACK TEMPERATURE, In Elg :			29.94	29.89	29.82	29.84	29.89	
		STACK TEMP., deg. F :			1692	1693	1692	1695	1694	
		MOLECULAR WEIGHT, DRY :			29.85	29.91	29.93	29.94	29.90	
		MOLECULAR WEIGHT, WET :			29.80	29.89	29.95	29.43	29.44	
		AVG. SIGHT VELOCITY HEAD :			0.07	0.07	0.07	0.07	0.09	
		VELOCITY, ft/s :			6.63	6.64	6.64	6.29	6.51	
		ACTUAL CUBIC FEET/MINUTE :			37913	38790	38724	36517	37400	
		DRY STANDARD CUBIC FEET/MINUTE :			7746	7582	7303	7403	7407	

9/21/90

validated 3/3/90

Moisture and Vmoist Calculations

Plant: Lake View Landfill

Date: 7/29/92

Run No.	Vmo cu-ft	W	Planar In. Hg	AME in. ELCO	Tm ° F	Vlc ml	Vmoist mol	Moisture %
LVL-O-1	22.501	0.978	29.93	2.01	84	91.9	21.491	16.76
LVL-O-2	22.851	0.978	29.94	2.01	89	84.3	21.622	15.51
LVL-O-3	31.111	0.978	29.94	2.01	100	88.6	28.662	12.65
							29.991	14.97

Handwritten signature
8/12/92



By B. G. G. L. A. Date 3/25/71 Subject Emission rate Sheet No. 1 of 1
Checked By 6/2/71 Date 6/2/71 Proj. No. 332-190

0.5cm. X 0.5cm.

Emission Rate Calc.

$$\text{Emission rate, lb/h} = \frac{(C_{\text{org}})(100)(Q_{\text{std}})(60)}{(375.3)(10^6)}$$

lb/h = pound per hour

C_{org} = concentration of nonmethane organic compounds, ppm by heptane

100 = molecular weight of heptane

Q_{std} = volumetric flow rate, dry standard cubic feet per minute

60 = conversion factor, minutes per hour

375.3 = molar volume, cubic feet/mole

10^6 = conversion factor

Total

$$\text{LVL-I-1} \quad \frac{(940)(100)(6,1040)(60)}{(375.3)(10^6)} = 7.13 \text{ lb/h}$$

$$\text{LVL-I-2} \quad \frac{(460)(100)(1050)(60)}{(375.3)(10^6)} = 7.52 \text{ lb/h}$$

$$\text{LVL-I-3} \quad \frac{(920)(100)(1070)(60)}{(375.3)(10^6)} = 7.73 \text{ lb/h}$$

Outlet

$$\text{LVL-O-1} \quad \frac{(3.3)(100)(10,700)(60)}{(375.3)(10^6)} = 0.55 \text{ lb/h}$$

$$\text{LVL-O-2} \quad \frac{(4.4)(100)(10,900)(60)}{(375.3)(10^6)} = 0.75 \text{ lb/h}$$

$$\text{LVL-O-3} \quad \frac{(2.0)(100)(10,200)(60)}{(375.3)(10^6)} = 0.32 \text{ lb/h}$$



INTERNATIONAL
TECHNOLOGY
CORPORATION



By W. J. Landrum Date 9/2/91 Subject Dechlorination Efficiency Sheet No. 1 of 1
Chkd. By W. J. Landrum Date 9/2/91 (D. G.) Proj. No. 232116
0.5 cm. X 0.5 cm.

$$\% \text{ D.E.} = \frac{(\text{Inlet, lb/h} - \text{Outlet, lb/h})}{\text{Inlet, lb/h}} \times 100$$

$$R_{\text{un}} \#1 \quad \frac{(7.13 - 0.55)}{7.13} \times 100 = 92.29 \%$$

$$R_{\text{un}} \#2 \quad \frac{(7.52 - 0.75)}{7.52} \times 100 = 90.03 \%$$

$$R_{\text{un}} \#3 \quad \frac{(7.83 - 0.32)}{7.83} \times 100 = 95.91 \%$$

$$Avg = 92.74 \%$$



INTERNATIONAL
TECHNOLOGY
CORPORATION



By Beck Date 10/2/92 Subject Exhaust Rate Calculating Sheet No. 1 of 1
 Chkd. By (12) Date 10/12/92 Manual File # BE Proj. No. 33333333
 0.5cm X 0.5cm

Exhaust emission rates remain the same.
 Output

$$LUL - 0 - 1 \quad \frac{(3.3)(100)(26.90)(60)}{(333.3)(10^6)} = 0.39 \text{ lb/lb}$$

$$LUL - 0 - 2 \quad \frac{(4.4)(100)(26.90)(60)}{(333.3)(10^6)} = 0.39 \text{ lb/lb}$$

$$LUL - 0 - 3 \quad \frac{(1.2)(100)(26.90)(60)}{(333.3)(10^6)} = 0.23 \text{ lb/lb}$$

Destruction Efficiency

$$Rem. \#1 \quad \frac{(7.13 - 0.39)}{7.13} \times 100 = 94.5\%$$

$$Rem. \#2 \quad \frac{(7.52 - 0.39)}{7.52} \times 100 = 93.2\%$$

$$Rem. \#3 \quad \frac{(7.83 - 0.23)}{7.83} \times 100 = 97.1\%$$

$$Avg = 94.9\%$$



By B. G. G. J. Date 10/12/91 Subject Recessed Efficiency Sheet No. 1 of 1
 Chkd. By (Signature) Date 10/12/91 Recessed Efficiency and Blank correction Proj. No. 212192

0.5cm. X 0.5cm.

Blank correction factor remove the noise.

Blank correction factor calculated in Appendix C.

Outlet Corrections		Outlet Efficiency being calculated
		Blank correction
LVL-0-1	$\frac{(0.4)(100)(7690)(60)}{(385.3)(10^4)} = 0.05$	0.06 b/h
LVL-0-2	$\frac{(1.5)(100)(7420)(60)}{(385.3)(10^4)} = 0.17$	0.25 b/h
LVL-0-3	$\frac{(0.0)(100)(7360)(60)}{(385.3)(10^4)} = 0.00$	0.00 b/h

Detection Efficiency		D. E. based on calculated rates
R ₀₀ ^{#1}	$\frac{(2.13 - 0.05)}{2.13} \times 100 = 99.3\%$	99.2%
R ₀₀ ^{#2}	$\frac{(2.52 - 0.17)}{2.52} \times 100 = 93.2\%$	96.7%
R ₀₀ ^{#3}	$\frac{(2.93 - 0.0)}{2.93} \times 100 = 100\%$	100.0%
Average = 99.0%		98.6%

Box Plotting Calculations

Project: Lake View Industrial
 Location: Pine Creek
 Date: 1/25/82

Sample Size: 100
 Sample Mean: 20.5
 Sample Std Dev: 2.5

Box	No.	Total Time	Total Volume	Frequency		Percent		Cumulative		Sample Value	P-Value	Q-Value	Q-Value	Box	Median	Box	Median
				in	out	in	out	in	out								
LVL-01A	1001	1001	1001	10	10	10	10	10	10	1001	0.10	1001	1001	1001	1001	1001	1001
	1002	1002	1002	10	10	10	10	10	10	1002	0.20	1002	1002	1002	1002	1002	1002
	1003	1003	1003	10	10	10	10	10	10	1003	0.30	1003	1003	1003	1003	1003	1003
	1004	1004	1004	10	10	10	10	10	10	1004	0.40	1004	1004	1004	1004	1004	1004
	1005	1005	1005	10	10	10	10	10	10	1005	0.50	1005	1005	1005	1005	1005	1005
LVL-01B	1006	1006	1006	10	10	10	10	10	10	1006	0.60	1006	1006	1006	1006	1006	1006
	1007	1007	1007	10	10	10	10	10	10	1007	0.70	1007	1007	1007	1007	1007	1007
	1008	1008	1008	10	10	10	10	10	10	1008	0.80	1008	1008	1008	1008	1008	1008
	1009	1009	1009	10	10	10	10	10	10	1009	0.90	1009	1009	1009	1009	1009	1009
	1010	1010	1010	10	10	10	10	10	10	1010	1.00	1010	1010	1010	1010	1010	1010
Average																	
LVL-02A	1011	1011	1011	10	10	10	10	10	10	1011	0.10	1011	1011	1011	1011	1011	1011
	1012	1012	1012	10	10	10	10	10	10	1012	0.20	1012	1012	1012	1012	1012	1012
	1013	1013	1013	10	10	10	10	10	10	1013	0.30	1013	1013	1013	1013	1013	1013
	1014	1014	1014	10	10	10	10	10	10	1014	0.40	1014	1014	1014	1014	1014	1014
	1015	1015	1015	10	10	10	10	10	10	1015	0.50	1015	1015	1015	1015	1015	1015
LVL-02B	1016	1016	1016	10	10	10	10	10	10	1016	0.60	1016	1016	1016	1016	1016	1016
	1017	1017	1017	10	10	10	10	10	10	1017	0.70	1017	1017	1017	1017	1017	1017
	1018	1018	1018	10	10	10	10	10	10	1018	0.80	1018	1018	1018	1018	1018	1018
	1019	1019	1019	10	10	10	10	10	10	1019	0.90	1019	1019	1019	1019	1019	1019
	1020	1020	1020	10	10	10	10	10	10	1020	1.00	1020	1020	1020	1020	1020	1020
Average																	
LVL-03A	1021	1021	1021	10	10	10	10	10	10	1021	0.10	1021	1021	1021	1021	1021	1021
	1022	1022	1022	10	10	10	10	10	10	1022	0.20	1022	1022	1022	1022	1022	1022
	1023	1023	1023	10	10	10	10	10	10	1023	0.30	1023	1023	1023	1023	1023	1023
	1024	1024	1024	10	10	10	10	10	10	1024	0.40	1024	1024	1024	1024	1024	1024
	1025	1025	1025	10	10	10	10	10	10	1025	0.50	1025	1025	1025	1025	1025	1025
LVL-03B	1026	1026	1026	10	10	10	10	10	10	1026	0.60	1026	1026	1026	1026	1026	1026
	1027	1027	1027	10	10	10	10	10	10	1027	0.70	1027	1027	1027	1027	1027	1027
	1028	1028	1028	10	10	10	10	10	10	1028	0.80	1028	1028	1028	1028	1028	1028
	1029	1029	1029	10	10	10	10	10	10	1029	0.90	1029	1029	1029	1029	1029	1029
	1030	1030	1030	10	10	10	10	10	10	1030	1.00	1030	1030	1030	1030	1030	1030
Average																	
Grand Average																	



By GC/MS Date 11/13/93 Subject Phenyl Nitro Concentration Street No. 1001
 Chd. By GC/MS Date 11/13/93 Calculated Proj. No. 232192
 O.Scm. & O.Scm.

$$\text{Conc, } \mu\text{g}/\mu\text{L} = \frac{\text{ppm (raw)}}{24.04}$$

Conc, $\mu\text{g}/\mu\text{L}$ = concentration of Nitro, milligrams per dec normal cubic meter
 ppm = parts per million, $\mu\text{g}/\mu\text{L}$

MW = molecular weight of specific noc or Nitro, g/g-molecule
 100 or higher

24.04 = molar volume, $\mu\text{L}/\text{g-molecule}$

Nitro concentration

Run No.		ppm	MW	$\mu\text{g}/\mu\text{L}$
Run No. 1	Exlet	440	100	18.30
	Outlet	3.3	100	0.14
Run No. 2	Exlet	460	100	19.13
	Outlet	4.4	100	0.18
Run No. 3	Exlet	470	100	19.55
	Outlet	2.0	100	0.08
Average	Exlet	457		19.04
	Outlet	3.2		0.13

By Henry J. Dyer, Jr. Date 4/2/73 Subject Specific Organic Compounds Sheet No. 1 of 1
 Chkd. By Delta Date 4/2/73 Calculation Proj. No. 222-2112
 O.5cm. X O.5cm.

$$\text{Conc. poly/mol No.}^3 = \frac{\text{ppb (ppm)}}{24,000}$$

Conc. poly/mol No. = Concentration, micrograms per decimol cubic meter

ppb = parts per billion, poly/mol No.

ppm = molecular weight, poly/mol No.

24,000 = molar volume, poly/mol No.

		ppb	ppm	poly/mol No.
Run No. 1	Methylphenyl Chloride	77.6	85.3	8.576
	1,1,1-Trichloroethane	35.3	13.3	3.6
Run No. 2	Methylphenyl Chloride	65.6	85.3	15.91
	1,1,1-Trichloroethane	31.6	13.3	2.1
Run No. 3	Methylphenyl Chloride	33.6	85.3	10.67
	1,1,1-Trichloroethane	65.2	13.3	6.29
Average	Methylphenyl Chloride	44.7		17.36
	1,1,1-Trichloroethane	64.8		6.27



By Alfred G. G. G. G. Subject Estimation Rate Scale Sec Sheet No. 1 of 1
 Chkd. By W. G. G. G. Date 10/2/61 Spec. Estimation Rate Proj. No. AL321912
 W.G.G. 0.5 cm. X 0.5 cm.

$$E, lb/h = \frac{(ppb)(mw)(Q_{std})(60)}{(373.3)(10^6)}$$

$E, lb/h$ = Estimation rate, pound per hour

ppb = parts per billion volume

mw = molecular weight, lb/mole

Q_{std} = volumetric flow rate, dry standard cubic feet per minute

60 = minutes per hour

373.3 = molar volume, cubic feet per lb-mole

10^6 = conversion factor

Methylacetylene chloride (mw = 93)

Run No. 1 $E, lb/h = \frac{(210)(10,000)(93)(60)}{(373.3)(10^6)} = 0.101$

Run No. 2 $E, lb/h = \frac{(600)(10,000)(93)(60)}{(373.3)(10^6)} = 0.065$

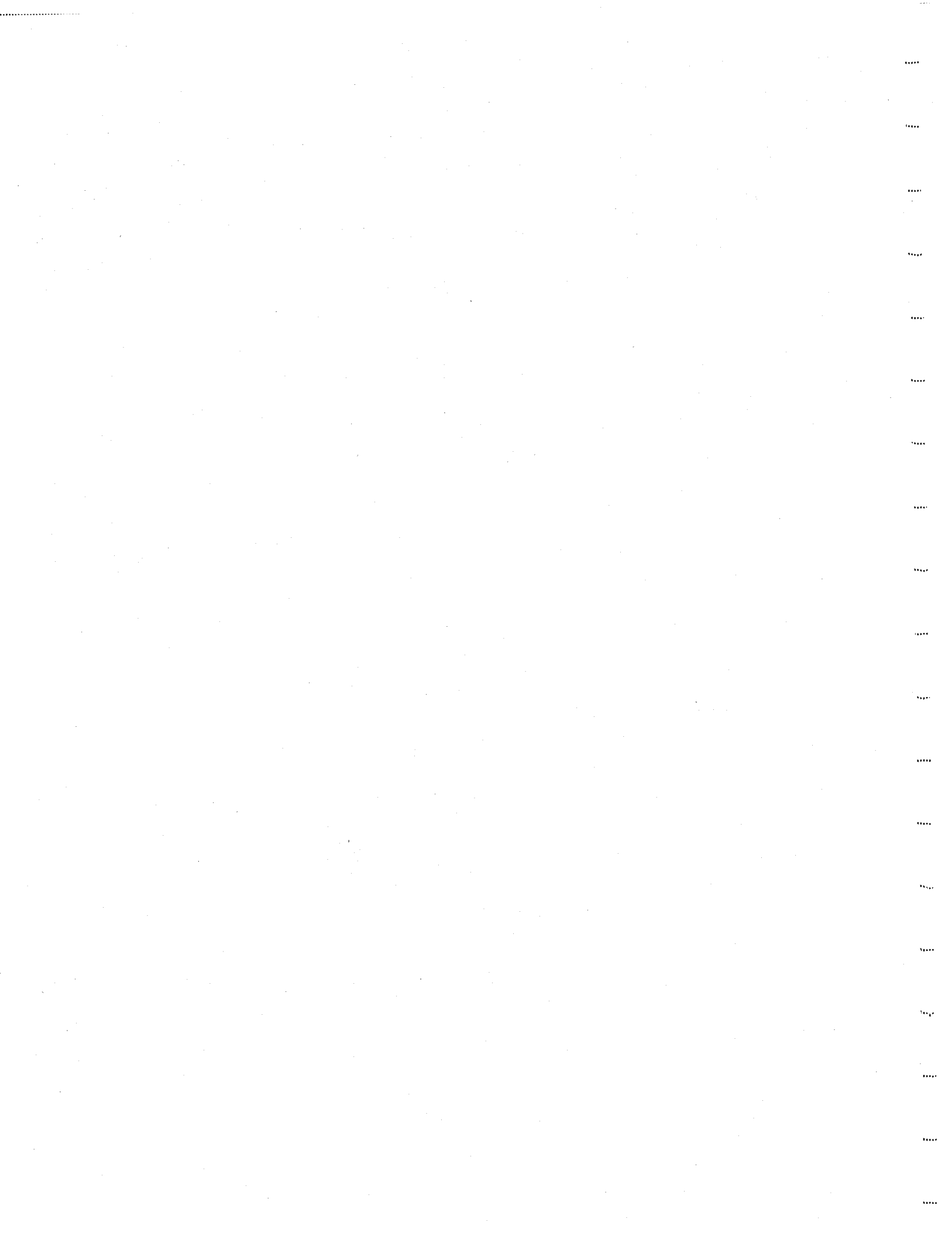
Run No. 3 $E, lb/h = \frac{(930)(10,000)(93)(60)}{(373.3)(10^6)} = 0.045$

1,1,1-Trichloroethane (mw = 133)

Run No. 1 $E, lb/h = \frac{(5.5)(10,000)(133)(60)}{(373.3)(10^6)} = 0.0012$

Run No. 2 $E, lb/h = \frac{(3.8)(10,000)(133)(60)}{(373.3)(10^6)} = 0.0009$

Run No. 3 $E, lb/h = \frac{(65.2)(10,000)(133)(60)}{(373.3)(10^6)} = 0.0011$

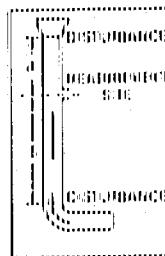
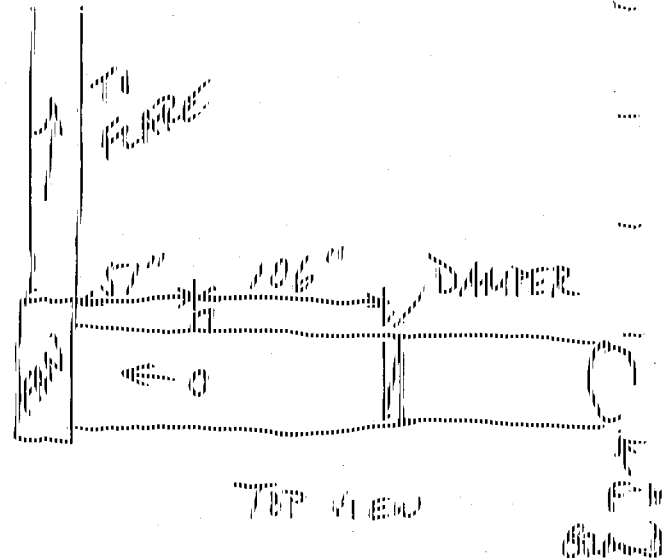


APPENDIX B

FIELD DATA

TRAVERSE POINT LOCATION FOR CIRCULAR DUCTS

Plant Lake View Landfill
 Date 2/28/92
 Sampling location Flare Inlet
 Inside of far wall to outside of nipple 17 1/2"
 Inside of near wall to outside of nipple (nipple length):
6"
 Stack inside diameter, inches 11 1/2"
 Distance downstream from flow disturbance (Distance B):
106" inches / diameter = 9.2 dd
 Distance upstream from flow disturbance (Distance A):
57" inches / diameter = 4.9 dd
 Calculated by (Signature)



SCHEMATIC OF SAMPLING LOCATION

TRAVERSE POINT NUMBER	FRACTION OF STACK I.D.	STACK I.D.	PRODUCT OF COLUMNS 2 AND 3 (TO NEAREST 1/10 INCH)	NIPPLE LENGTH	TRAVERSE POINT LOCATION FROM OUTSIDE OF NIPPLE (CHOICE OF COLUMNS 4 & 5)
1	1/2	11 1/2		6	17 1/2
2	1/2	11 1/2		6	17 1/2
3	1/2	11 1/2		6	17 1/2
4	1/2	11 1/2		6	17 1/2
5	1/2	11 1/2		6	17 1/2
6	1/2	11 1/2		6	17 1/2
7	1/2	11 1/2		6	17 1/2
8	1/2	11 1/2		6	17 1/2
9	1/2	11 1/2		6	17 1/2
10	1/2	11 1/2		6	17 1/2

GAS VELOCITY AND VOLUMETRIC FLOW RATE

Plant and City Lakeview, NM Date 7/19/92
 Sampling Location Flare Stack Clock Time 1100
 Run No. 1-VL-2-V1 Operator DO
 Barometric Pressure, in. Hg 29.94 Static Pressure, in. H₂O -21.6
 Moisture, % 3.33 Molecular wt., Dry 28.96 Pitot Tube, Cp 1.54
 Stack Dimension, in. Diameter or Side 1 11 1/2 Side 2 —

7 500 63 70 F
 70 101 63 70 F

std = 6.2 m³

FIELD DATA

CALCULATIONS

INVERSE POINT NUMBER	VELOCITY HEAD (in. H ₂ O)	STACK TEMP., °F
1	0.0000	—
2	0.0000	—
3	0.0000	—
4	0.0000	—
5	0.0000	—
6	0.0000	—
7	0.0000	—
8	0.0000	—
9	0.0000	—
10	0.0000	—
11	0.0000	—
12	0.0000	—
13	0.0000	—
14	0.0000	—
15	0.0000	—
16	0.0000	—
17	0.0000	—
18	0.0000	—
19	0.0000	—
20	0.0000	—
21	0.0000	—
22	0.0000	—
23	0.0000	—
24	0.0000	—
25	0.0000	—
26	0.0000	—
27	0.0000	—
28	0.0000	—
29	0.0000	—
30	0.0000	—

$$\begin{aligned}
 v_1 &= 40.8 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_2 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_3 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_4 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_5 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_6 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_7 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_8 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_9 &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{10} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{11} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{12} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{13} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{14} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{15} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{16} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{17} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{18} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{19} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{20} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{21} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{22} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{23} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{24} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{25} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{26} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{27} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{28} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{29} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100} \\
 v_{30} &= 140.0 \times (1 - \frac{0.0}{100}) \times 10 \times \frac{0.0}{100}
 \end{aligned}$$

GAS VELOCITY AND VOLUMETRIC FLOW RATE

Plant and City Lakeview, Louisiana Date 7/25/72
Sampling Location Shelf Clock Time 12:00
Run No. 111-3-112 Operator BO
Barometric Pressure, in. Hg 29.90 Static Pressure, in. H₂O -21.0
Moisture, % 1.25 Molecular wt., Dry 44.0 Pitot Tube, Cp .04
Stack Dimension, in. Diameter or Side 1 11 1/2 Side 2

WB 69 F
WB 77 F

$\rho_d = \log_{10} \frac{1}{1 - \rho^2}$

2000

FIELD DATA

CONCLUSIONS

FRAME #	VELOCITY	STACK
POINT	HEAD	HEAD
NUMBER	TOP 1. TO 2.0	TOP 1. TO 2.0
1	1.0	1.0
2	1.0	1.0
3	1.0	1.0
4	1.0	1.0
5	1.0	1.0
6	1.0	1.0
7	1.0	1.0
8	1.0	1.0
9	1.0	1.0
10	1.0	1.0
11	1.0	1.0
12	1.0	1.0
13	1.0	1.0
14	1.0	1.0
15	1.0	1.0
16	1.0	1.0
17	1.0	1.0
18	1.0	1.0
19	1.0	1.0
20	1.0	1.0
21	1.0	1.0
22	1.0	1.0
23	1.0	1.0
24	1.0	1.0
25	1.0	1.0
26	1.0	1.0
27	1.0	1.0
28	1.0	1.0
29	1.0	1.0
30	1.0	1.0
31	1.0	1.0
32	1.0	1.0
33	1.0	1.0
34	1.0	1.0
35	1.0	1.0
36	1.0	1.0
37	1.0	1.0
38	1.0	1.0
39	1.0	1.0
40	1.0	1.0
41	1.0	1.0
42	1.0	1.0
43	1.0	1.0
44	1.0	1.0
45	1.0	1.0
46	1.0	1.0
47	1.0	1.0
48	1.0	1.0
49	1.0	1.0
50	1.0	1.0
51	1.0	1.0
52	1.0	1.0
53	1.0	1.0
54	1.0	1.0
55	1.0	1.0
56	1.0	1.0
57	1.0	1.0
58	1.0	1.0
59	1.0	1.0
60	1.0	1.0
61	1.0	1.0
62	1.0	1.0
63	1.0	1.0
64	1.0	1.0
65	1.0	1.0
66	1.0	1.0
67	1.0	1.0
68	1.0	1.0
69	1.0	1.0
70	1.0	1.0
71	1.0	1.0
72	1.0	1.0
73	1.0	1.0
74	1.0	1.0
75	1.0	1.0
76	1.0	1.0
77	1.0	1.0
78	1.0	1.0
79	1.0	1.0
80	1.0	1.0
81	1.0	1.0
82	1.0	1.0
83	1.0	1.0
84	1.0	1.0
85	1.0	1.0
86	1.0	1.0
87	1.0	1.0
88	1.0	1.0
89	1.0	1.0
90	1.0	1.0
91	1.0	1.0
92	1.0	1.0
93	1.0	1.0
94	1.0	1.0
95	1.0	1.0
96	1.0	1.0
97	1.0	1.0
98	1.0	1.0
99	1.0	1.0
100	1.0	1.0

$$\begin{aligned} u_3 &= 0.1 + 0.1 \left(1 - \frac{0.1}{0.2} \right) + 0.1 \left(\frac{0.1}{0.2} \right)^2 \\ u_5 &= \left(\frac{0.1 \cdot 0.1 \cdot 0.1 \cdot 0.1}{0.2 \cdot 0.2 \cdot 0.2 \cdot 0.2} \right) + 0.1 \left(\frac{0.1 \cdot 0.1 \cdot 0.1}{0.2 \cdot 0.2 \cdot 0.2} \right) + 0.1 \left(\frac{0.1 \cdot 0.1 \cdot 0.1}{0.2 \cdot 0.2 \cdot 0.2} \right) \\ u_6 &= \frac{0.1 \cdot 0.1 \cdot 0.1 \cdot 0.1 \cdot 0.1 \cdot 0.1}{0.2 \cdot 0.2 \cdot 0.2 \cdot 0.2 \cdot 0.2 \cdot 0.2} \quad [2, 7, 10, 4] \end{aligned}$$

$$T_6 = 30^\circ\text{C} \quad \text{or} \quad 86^\circ\text{F} \quad \text{IR} = (25 \pm 460)$$

$$p_5 + p_6 + \frac{5 \cdot p_7}{192} = (2^{10} + 2^9 + 2^8) = \frac{2^{11} + 2^{10}}{192}$$

p. 113 6 11.19

$\sqrt{13} = 3, 605' 2$.

$$V_1 = 05.49 \text{ m}^2 \text{ s}^{-1} \sqrt{\frac{1}{2} \rho} \sqrt{\frac{1}{2} \frac{1}{\rho} \frac{1}{V_1^2}}$$

$V_1 = 0.49 \times 10^{-3} \text{ m}^3$ $\rho = 0.932 \text{ g/cm}^3$ $\sqrt{\frac{340}{26.56 \times 10^{-3} \times 0.932}}$

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2. 2. 6. 9

$$A_1 = 0.92 \text{ m}^2$$

Q. Now, I have a question for you. Did you see the man who was shot on the 10th of April, 1968, at the same place where you saw the man who was shot on the 4th of April, 1968?

$$C_6 = \frac{2.6 \times 10^4}{4.9 \times 10^4} = 0.52 \quad \text{at } 60^\circ \text{C.}$$

$\zeta_6 = \frac{q^2 + q^{-2}}{2}$ and $\zeta_9 = \frac{q^3 + q^{-3}}{3}$.

$$Q_{6,642} = Q_6 + 12,642 \times \frac{P_6}{P_6} \times (1 - \frac{91,000}{100})$$

$$Q_{6,112} = \frac{q(q+1)q!}{(q+1)q!} = 12,600 = \frac{2 \cdot 2 \cdot 3 \cdot 5 \cdot 7}{5 \cdot 7 \cdot 7} = 12 = \frac{2 \cdot 2 \cdot 3}{3 \cdot 3} = 4$$

06-09-2018

CONCLUSIONS

1059 *Ab. 1059*

May 13, 1932

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6/10/72

Plant and City Lakeview, New Mexico Date 7/29/53
Sampling Location Spot 1 Clock Time 1453
Run No. LM-3-13 Operator DB
Barometric Pressure, in. Hg 29.97 Static Pressure, in. H₂O -2.1
Moisture, % 2.62 Molecular wt., Dry 27.9 Pitot Tube, Cp 1.1
Stack Dimension, in. Diameter or Side 1 11.75 Side 2 _____

WB 734F
DB 100F

WILLIAM S. BRYAN

REFERENCES

[illegible]

[illegible]

checked
by
1/16/94

GAS VELOCITY AND VOLUMETRIC FLOW RATE

plant and City

Sampling Location

Ref. No.

Barometric Pressure in Hg. 30.00

Moisture, %

Moisture, % 24.5 Molecular wt., 179

Stack Dimension, in. Diameter or Side Length, in.

Date _____

Clock Time

Operator

Static Pressure, in. H₂O

Figure 1. *Phragmites* and *Spartina* coverages in the marshes of the Sacramento-San Joaquin River Delta, California, 1990-1999.

Pilot Tube, CP

Side 23

100

PLA = Polyurethane

608 70
29 77

15 JUL 1964

[illegible]

CONCLUSIONS

$$U_2 = U_0 \left(1 + \frac{U_0^2}{2} \right) + 15 \left(\frac{U_0^2}{100} \right) \quad (2.94)$$

$$U_1 = \left(\frac{U_0^2}{100} \right) + 15 \left(\frac{U_0^2}{100} \right) + 15 \left(\frac{U_0^2}{100} \right) \quad (2.95)$$

6. අනුමත 2.7.32.

U. S. GEOLOGICAL SURVEY

[illegible]

$\rho_1 = 7.973 \pm 0.003$

$\sqrt{\frac{10}{11}} = 0.9464$

$$M_{\text{eff}} = 0.546 \times 0.5 \times \sqrt{\frac{1}{1.5}} \times \sqrt{\frac{1}{1.5} + \frac{0.1}{1.5}}$$

$$y = 0.0549 \times (G)^{0.687} + 0.0001(G)^{0.687} - 1 \quad \frac{5.49}{246.32} = 0.0223$$

7-94904 2774

$\mu_{\text{eff}} = 0.79 \pm 0.02$

$$C_{\text{eff}} = C_1 + C_2 + C_3 + \dots + \frac{C_n}{n}$$

U. S. P. 2,740,000

10-11-57

$$P = 0.011,607 + \frac{P}{9} + (1 - \frac{0.7}{9})$$

$$11.607 \div 3.46 = 3.3546242774566474$$

(2) 0-6(1)(ii), (iii), (iv)

... ..

10939

03/24/92

Checked
10/11/06

Method 25 Field Data

Plant <u>Lakeview Landfill</u>	City <u>Greene, Pa</u>	Date <u>7/29/92</u>
Run No. <u>LV-1-1</u>	Sample Location <u>Stack</u>	
Operator <u>BB</u>	Tank	
Tank No. <u>04/10</u>	Volume, liters <u>6</u>	
Trap No. <u>01</u>	Gasometer No. <u>713</u>	
Sampling Rate <u>100</u> cc/min.	Filter Type <u>50</u>	in stack out of stack

Sample Train Data

	Barometric Pressure, mm Hg	Ambient Temperature, °F °C	Sample Tank Gauge Vacuum, mm Hg
Pre-Test	<u>29.9</u>	<u>85</u> °	<u>22.1</u>
Post-Test	<u>29.9</u>	<u>87</u> °	<u>22</u>

Front-Half - Leak Check Data

Time	Initial Vacuum, mm Hg	Time	Final Vacuum, mm Hg	ΔP, mm Hg	Leak Rate,* cc/min

Sampling Data

Time		Gauge Vacuum, mm Hg	Temperatures, °F	
Clock	Sampling, min		Probe	Filter Box
<u>1153</u>	<u>0</u>	<u>22.1</u>		
	<u>10</u>	<u>22.1</u>		
	<u>20</u>	<u>22.1</u>		
	<u>30</u>	<u>22.1</u>		
	<u>40</u>	<u>22.1</u>		
	<u>50</u>	<u>22.1</u>		
<u>1233</u>	<u>60</u>	<u>22.1</u>		

* Leak Rate, cc/min = $\frac{\Delta P, \text{ mm} \times \text{Volume, cm}^3}{P_2, \text{ mm} \times \text{Time, min}}$

Leak rate must be <0.01 cc sampling rate
(0.0 cc/min leak rate at sampling rate of 60 cc/min)
(volume = 100 cm³ for less than 4 ft probe)

Method 25 Field Data

Plant Lake View Coal Ref City Grass, PA Date 2/29/82
 Run No. 002-3-112 Sample Location Feed
 Operator DS Tank 6
 Tank No. 09282 Volume, liters 6
 Trap No. NA Barometer No. 98
 Sampling Rate 100 cc/min. Filter Type Standard B in stack
 out of stack

Sample Train Data

	Barometric Pressure, mm Hg	Ambient Temperature, °F °C	Sample Tank Gauge Vacuum, mm Hg
Pre-Test	<u>29.6</u>	<u>66</u> ° <u> </u> °C	<u>22.1</u>
Post-Test	<u>29.6</u>	<u>61</u> ° <u> </u> °C	<u>26</u>

Front-Half Leak Check Data

Time	Initial Vacuum, mm Hg	Time	Final Vacuum, mm Hg	ΔP, mm Hg	Leak Rate,* cc/min

Sampling Data

Time		Gauge Vacuum, mm Hg	Temperatures, °F	
Clock	Sampling, min		Probe	Filter Box
<u>1332</u>	<u>0</u>	<u>22.1</u>		
	<u>10</u>	<u>22.1</u>		
	<u>20</u>	<u>22.1</u>		
	<u>30</u>	<u>22.1</u>	<u>NA</u>	<u>NA</u>
	<u>40</u>	<u>22.1</u>		
	<u>50</u>	<u>22.1</u>		
<u>1432</u>	<u>0</u>	<u>22.1</u>		
	<u>10</u>	<u>22.1</u>		
	<u>20</u>	<u>22.1</u>		
	<u>30</u>	<u>22.1</u>		
	<u>40</u>	<u>22.1</u>		
	<u>50</u>	<u>22.1</u>		

* Leak Rate, cc/min = $\frac{\Delta P, \text{ mm} \times \text{Volume, cm}^3}{Pb, \text{ mm} \times \text{Time, min}}$

Leak rate must be <0.01 x sampling rate
 (0.0 cc/min leak rate at sampling rate of 60 cc/min)
 (volume = 100 cm³ for less than 4 ft probe)

Method 25 Field Data

Plant Lake View, Calif City Fontana Date 7/29/92
 Run No. CU-3-3 Sample Location 360
 Operator DO Tank 6
 Tank No. 1036 Volume, liters 6
 Trap No. NA Gasometer No. 48
 Sampling Rate 100 cc/min Filter Type 5Mced33 in stack
 out of stack

Sample Train Data

	Barometric Pressure, mm Hg	Ambient Temperature, °F °C	Sample Tank Gauge Vacuum, mm Hg
Pre-Test	<u>761</u>	<u>91</u> °F °C	<u>722</u>
Post-Test	<u>760</u>	<u>91</u> °F °C	<u>722</u>

Front-Half - Leak Check Data

Time	Initial Vacuum, mm Hg	Time	Final Vacuum, mm Hg	ΔP , mm Hg	Leak Rate,* cc/min

Sampling Data

Clock	Time Sampling, min	Gauge Vacuum, mm Hg	Temperatures, °F	
			Probe	Filter Box
<u>157</u>	<u>0</u>	<u>722</u>		
	<u>10</u>	<u>722</u>		
	<u>20</u>	<u>722</u>		
	<u>30</u>	<u>722</u>		
	<u>40</u>	<u>722</u>		
	<u>50</u>	<u>722</u>		
<u>161</u>	<u>60</u>	<u>722</u>		

* Leak Rate, cc/min = $\frac{\Delta P, \text{ mm} \times \text{Volume, cm}^3}{P_b, \text{ mm} \times \text{Time, min}}$

Leak rate must be <0.04 x sampling rate
 (0.4 cc/min leak rate at sampling rate of 60 cc/min)
 (volume = 100 cm³ for less than 4 ft probe)



DRY MOLECULAR WEIGHT DETERMINATION

PLANT Lake Mead Mill
 DATE 7/11/12 TEST NO 100-1
 SAMPLING TIME (24 hr clock) 1405
 SAMPLING LOCATION 1st bag
 SAMPLE TYPE (BAG, INTEGRATED, CONTINUOUS)
 ANALYTICAL METHOD Direct
 AMBIENT TEMPERATURE 70°F
 OPERATOR [Signature]
 ORSAT LEAK CHECKED ✓

COMMENTS:

GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) H ₂ is 2.016
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	39.4	38.9	37.6	37.6			39.5	4/100	
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)	37.2	0.0	37.4	0.0			0.0	32/100	
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)								25/100	
H ₂ (NET IS 100 MINUS ACTUAL CO READING)								25/100	
TOTAL									

DRY MOLECULAR WEIGHT DETERMINATION

PLANT Lakeview Landfill
 DATE 7/19/92 TEST NO. CVL-1-1
 SAMPLING TIME (24 hr clock) _____
 SAMPLING LOCATION 1405
 SAMPLE TYPE (GAS, LIQUID, CONTINUOUS) Gas
 ANALYTICAL METHOD GC
 AMBIENT TEMPERATURE 68°F
 OPERATOR [Signature]
 ORSAT LEAK CHECKED ✓

COMMENTS:

GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) M _g is basis
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	39.5	39.5	39.5	39.5			39.5	4/100	
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)	39.2	0.7	39.4	0.9			0.9	32/100	
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)								28/100	
N ₂ (NET IS 100 MINUS ACTUAL CO READING)								28/100	
TOTAL									



DRY MOLECULAR WEIGHT DETERMINATION

PLANT Lakeview Laundry
 DATE 7/27/92 TEST NO. CL-1-3
 COMMENTS:
 SAMPLING TIME (AM/PM) 1:45 PM
 SAMPLING LOCATION 1st Flr
 SAMPLE TYPE (GAS, INTEGRATED, CONTINUOUS) Int. Gas
 ANALYTICAL METHOD Direct
 AMBIENT TEMPERATURE 75
 OPERATOR [Signature]
 ORSAT LEAK CHECKED ✓

GAS	RUN	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) #4, 15 to mole
		ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂		39.5	14.5	39.4	39.4			38.4	10/100	
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)		39.4	0.9	39.3	0.9			0.9	12/100	
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)									25/100	
N ₂ (NET IS 100 MINUS ACTUAL CO READING)									25/100	
TOTAL										

TRAVERSE POINT LOCATION FOR CIRCULAR DUCTS

[illegible]

1. **CONSTITUTIONAL PRINCIPLES**
2. **GOVERNMENT STRUCTURE**
3. **LEGISLATION**
4. **JUDICIARY**
5. **EXECUTIVE**
6. **LOCAL GOVERNMENT**
7. **EDUCATION**
8. **HEALTH**
9. **INDUSTRY**
10. **AGRICULTURE**
11. **FOREIGN RELATIONS**
12. **DEFENCE**
13. **SCIENCE AND TECHNOLOGY**
14. **ENVIRONMENT**
15. **SPORTS**
16. **ARTS AND CULTURE**
17. **RELIGION**
18. **SOCIAL WELFARE**
19. **LABOUR**
20. **FINANCE**
21. **INDUSTRIAL RELATIONS**
22. **CO-OPERATION**
23. **PLANNING**
24. **WELFARE**
25. **INDUSTRIAL RELATIONS**
26. **CO-OPERATION**
27. **PLANNING**
28. **WELFARE**
29. **INDUSTRIAL RELATIONS**
30. **CO-OPERATION**
31. **PLANNING**
32. **WELFARE**
33. **INDUSTRIAL RELATIONS**
34. **CO-OPERATION**
35. **PLANNING**
36. **WELFARE**
37. **INDUSTRIAL RELATIONS**
38. **CO-OPERATION**
39. **PLANNING**
40. **WELFARE**
41. **INDUSTRIAL RELATIONS**
42. **CO-OPERATION**
43. **PLANNING**
44. **WELFARE**
45. **INDUSTRIAL RELATIONS**
46. **CO-OPERATION**
47. **PLANNING**
48. **WELFARE**
49. **INDUSTRIAL RELATIONS**
50. **CO-OPERATION**
51. **PLANNING**
52. **WELFARE**
53. **INDUSTRIAL RELATIONS**
54. **CO-OPERATION**
55. **PLANNING**
56. **WELFARE**
57. **INDUSTRIAL RELATIONS**
58. **CO-OPERATION**
59. **PLANNING**
60. **WELFARE**
61. **INDUSTRIAL RELATIONS**
62. **CO-OPERATION**
63. **PLANNING**
64. **WELFARE**
65. **INDUSTRIAL RELATIONS**
66. **CO-OPERATION**
67. **PLANNING**
68. **WELFARE**
69. **INDUSTRIAL RELATIONS**
70. **CO-OPERATION**
71. **PLANNING**
72. **WELFARE**
73. **INDUSTRIAL RELATIONS**
74. **CO-OPERATION**
75. **PLANNING**
76. **WELFARE**
77. **INDUSTRIAL RELATIONS**
78. **CO-OPERATION**
79. **PLANNING**
80. **WELFARE**
81. **INDUSTRIAL RELATIONS**
82. **CO-OPERATION**
83. **PLANNING**
84. **WELFARE**
85. **INDUSTRIAL RELATIONS**
86. **CO-OPERATION**
87. **PLANNING**
88. **WELFARE**
89. **INDUSTRIAL RELATIONS**
90. **CO-OPERATION**
91. **PLANNING**
92. **WELFARE**
93. **INDUSTRIAL RELATIONS**
94. **CO-OPERATION**
95. **PLANNING**
96. **WELFARE**
97. **INDUSTRIAL RELATIONS**
98. **CO-OPERATION**
99. **PLANNING**
100. **WELFARE**

13.13

GAS VELOCITY AND VOLUMETRIC FLOW RATE

Plant and City Lakeview, New Mexico Date 7/29/92
 Sampling Location Flare outlet Clock Time 1009
 Run No. LVL-0-VI Operator SGP/BC
 Barometric Pressure, in. Hg 29.54 Static Pressure, in. H₂O -0.27
 Moisture, % 16.2 Molecular wt., Dry 29.85 Pitot Tube, Cp 1.99
 Stack Dimension, in. Diameter or Side 1 11.6 Side 2 _____

FIELD DATA

CALCULATIONS

TRaverse Point Number	VELOCITY HEAD (in. H ₂ O)	Static Pressure (in. H ₂ O)
1	0.15	0.15
2	0.15	0.15
3	0.15	0.15
4	0.15	0.15
5	0.15	0.15
6	0.15	0.15
7	0.15	0.15
8	0.15	0.15
9	0.15	0.15
10	0.15	0.15
11	0.15	0.15
12	0.15	0.15
13	0.15	0.15
14	0.15	0.15
15	0.15	0.15
16	0.15	0.15
17	0.15	0.15
18	0.15	0.15
19	0.15	0.15
20	0.15	0.15
21	0.15	0.15
22	0.15	0.15
23	0.15	0.15
24	0.15	0.15
25	0.15	0.15
26	0.15	0.15
27	0.15	0.15
28	0.15	0.15
29	0.15	0.15
30	0.15	0.15
31	0.15	0.15
32	0.15	0.15
33	0.15	0.15
34	0.15	0.15
35	0.15	0.15
36	0.15	0.15
37	0.15	0.15
38	0.15	0.15
39	0.15	0.15
40	0.15	0.15
41	0.15	0.15
42	0.15	0.15
43	0.15	0.15
44	0.15	0.15
45	0.15	0.15
46	0.15	0.15
47	0.15	0.15
48	0.15	0.15
49	0.15	0.15
50	0.15	0.15
51	0.15	0.15
52	0.15	0.15
53	0.15	0.15
54	0.15	0.15
55	0.15	0.15
56	0.15	0.15
57	0.15	0.15
58	0.15	0.15
59	0.15	0.15
60	0.15	0.15
61	0.15	0.15
62	0.15	0.15
63	0.15	0.15
64	0.15	0.15
65	0.15	0.15
66	0.15	0.15
67	0.15	0.15
68	0.15	0.15
69	0.15	0.15
70	0.15	0.15
71	0.15	0.15
72	0.15	0.15
73	0.15	0.15
74	0.15	0.15
75	0.15	0.15
76	0.15	0.15
77	0.15	0.15
78	0.15	0.15
79	0.15	0.15
80	0.15	0.15
81	0.15	0.15
82	0.15	0.15
83	0.15	0.15
84	0.15	0.15
85	0.15	0.15
86	0.15	0.15
87	0.15	0.15
88	0.15	0.15
89	0.15	0.15
90	0.15	0.15
91	0.15	0.15
92	0.15	0.15
93	0.15	0.15
94	0.15	0.15
95	0.15	0.15
96	0.15	0.15
97	0.15	0.15
98	0.15	0.15
99	0.15	0.15
100	0.15	0.15

$$\begin{aligned}
 P_s &= P_b + (1 - \frac{P_b}{P_s}) \cdot 10 \cdot \frac{P_b}{P_s} \\
 P_s &= 1 + (1 - \frac{1}{100}) \cdot 10 \cdot \frac{1}{100} \\
 P_s &= 1.09 \\
 P_s &= 1.09 \cdot (1 + 0.0001) \\
 P_s &= 1.0901 \\
 P_s &= P_b + \frac{P_b}{100} \cdot 1 \\
 P_s &= 1.0901 \\
 \frac{P_s}{P_b} &= 1.0901 \\
 V_1 &= 85.49 \cdot \sqrt{\frac{P_s}{P_b}} \cdot \sqrt{\frac{1}{P_s} \cdot \frac{P_b}{P_s}} \\
 V_1 &= 85.49 \cdot \sqrt{1.0901} \cdot \sqrt{\frac{1}{1.0901}} \\
 V_1 &= 85.49 \cdot 1.0441 \cdot 0.9577 \\
 V_1 &= 85.49 \cdot 1.0 \\
 V_1 &= 85.49 \text{ ft/s} \\
 V_1 &= 192 \text{ ft/min} \\
 Q_1 &= V_1 \cdot A_1 \cdot \frac{60}{1} \\
 Q_1 &= 85.49 \cdot 1.0 \cdot 60 \\
 Q_1 &= 5129.4 \text{ acfm} \\
 Q_{std} &= Q_1 \cdot 27.68 \cdot \frac{P_s}{P_b} \cdot (1 - \frac{P_b}{P_s}) \\
 Q_{std} &= 5129.4 \cdot 27.68 \cdot 1.0901 \cdot (1 - \frac{1}{100}) \\
 Q_{std} &= 38000 \text{ acfm}
 \end{aligned}$$

GAS VELOCITY AND VOLUMETRIC FLOW RATE

Plant and City Rocky Mountain Date 7/29/92
 Sampling Location Challenger Clock Time 1255
 Run No. LV-0-V2 Operator BOB
 Barometric Pressure, in. Hg 29.70 Static Pressure, in. H₂O -0.73
 Moisture, % 16.14 Molecular wt., Dry 29.90 Pitot Tube, Co 1.84
 Stack Dimension, in. Diameter or Side 1 11.6 Side 2

FIELD DATA

TRaverse POINT NUMBER	VELOCITY MPS (ft./s. in. H ₂ O)	STACK TEMP., °F
1	10.0	100
2	10.0	100
3	10.0	100
4	10.0	100
5	10.0	100
6	10.0	100
7	10.0	100
8	10.0	100
9	10.0	100
10	10.0	100
11	10.0	100
12	10.0	100
13	10.0	100
14	10.0	100
15	10.0	100
16	10.0	100
17	10.0	100
18	10.0	100
19	10.0	100
20	10.0	100
21	10.0	100
22	10.0	100
23	10.0	100
24	10.0	100
25	10.0	100
26	10.0	100
27	10.0	100
28	10.0	100
29	10.0	100
30	10.0	100
31	10.0	100
32	10.0	100
33	10.0	100
34	10.0	100
35	10.0	100
36	10.0	100
37	10.0	100
38	10.0	100
39	10.0	100
40	10.0	100
41	10.0	100
42	10.0	100
43	10.0	100
44	10.0	100
45	10.0	100
46	10.0	100
47	10.0	100
48	10.0	100
49	10.0	100
50	10.0	100
51	10.0	100
52	10.0	100
53	10.0	100
54	10.0	100
55	10.0	100
56	10.0	100
57	10.0	100
58	10.0	100
59	10.0	100
60	10.0	100
61	10.0	100
62	10.0	100
63	10.0	100
64	10.0	100
65	10.0	100
66	10.0	100
67	10.0	100
68	10.0	100
69	10.0	100
70	10.0	100
71	10.0	100
72	10.0	100
73	10.0	100
74	10.0	100
75	10.0	100
76	10.0	100
77	10.0	100
78	10.0	100
79	10.0	100
80	10.0	100
81	10.0	100
82	10.0	100
83	10.0	100
84	10.0	100
85	10.0	100
86	10.0	100
87	10.0	100
88	10.0	100
89	10.0	100
90	10.0	100
91	10.0	100
92	10.0	100
93	10.0	100
94	10.0	100
95	10.0	100
96	10.0	100
97	10.0	100
98	10.0	100
99	10.0	100
100	10.0	100

CALCULATIONS

$$\begin{aligned}
 P_s &= P_b \left(1 - \frac{H_2O}{13.7} \right) + 10 \left(\frac{H_2O}{13.7} \right) \\
 P_s &= (\quad) + (\quad) = 10 \left(\frac{H_2O}{13.7} \right) \\
 P_s &= \quad \\
 T_s &= T_b + \frac{P_s - P_b}{1.5} \\
 P_s &= P_b + \frac{P_s - P_b}{1.5} \\
 P_s &= 10.0 \\
 V_{AP} &= \quad \\
 U_s &= 0.5 \times C_p \times \sqrt{\frac{P_s - P_b}{P_s}} \\
 U_s &= 0.5 \times (\quad) \times (\quad) \\
 U_s &= 10.0 \\
 A_s &= 10.0 \\
 Q_s &= U_s \times A_s \times \frac{60}{1} \\
 Q_s &= 10.0 \times 10.0 \times 60 \\
 Q_s &= 6000 \\
 Q_{std} &= Q_s \times \frac{P}{P_b} \times \left(\frac{T_b}{T} \right) \\
 Q_{std} &= 6000 \times \frac{P}{P_b} \times \left(\frac{T_b}{T} \right) \\
 Q_{std} &= 6000
 \end{aligned}$$

GAS VELOCITY AND VOLUMETRIC FLOW RATE

Plant and City Lakeview, La. Date 7/25/92
Sampling Location Chlor Clock Time 1435
Run No. 111-0-13 Operator SPC
Barometric Pressure, in. Hg 29.93 Static Pressure, in. H₂O -0.29
Moisture, % 18.54 Molecular wt., Dry 39.34 Pitot Tube, Cp 68
Stack Dimension, in. Diameter or Side 1 116 Side 2

FILED DATE

CHICAGO, 1985.

[illegible]

$$\mu_5 = \mu_4 + (1 + \frac{0.20}{100}) + 10 (\frac{0.20}{100})$$

$$U_i = \left(\begin{array}{c} 0 \\ 0 \\ 0 \\ \vdots \\ 0 \end{array} \right) + \frac{\dots\dots\dots}{10^6} \rightarrow \text{LB } \left(\frac{\dots\dots\dots}{10^6} \right)$$

11

$$V_1 = 100 \text{ m}^3 \quad \text{and} \quad V_2 = 100 \text{ m}^3 \quad \text{and} \quad V_3 = (100 + 100)$$

$$P_1 = P_b + \frac{S_{\text{eff}}}{A} = (0.78 + \frac{6.9}{15.6}) = 1.44$$

P_1 at $t = 0$ is 100. (40)

√³ 3

$$W_{\text{eff}} = 0.5 \cdot C_p + \frac{\sqrt{G^2 + G_0^2}}{G} \left(\frac{1}{5} - \frac{1}{5} \right)$$

$$y_{\text{eff}} = (25.6 \pm 0.1) \times 10^{-3} \text{ m} \quad (1)$$

134

183

Q. Now, if you were to take the 1950 census, would you be able to tell me how many of those 100,000 people were Negroes?

() " " " 60

1. "Before"

$$0_{\text{field}} = 0.1 \pm 0.002 + \frac{p}{q} \left(1 - \frac{0.002}{10} \right)$$

U.S. DEPT. OF JUSTICE

08/06/00 08:00

GAS VELOCITY AND VOLUMETRIC FLOW RATE

Plant and City Lake View Landfill Date 7/29/92
 Sampling Location Stack Clock Time 1635
 Run No. LVL-0-114 Operator DEH
 Barometric Pressure, in.Hg 29.95 Static Pressure, in.H₂O -0.079
 Moisture, % 11.6% Molecular Wt., Dry 28.86 Pitot Tube, Cp 1.25
 Stack Dimension, in. Diameter or Side 1 11 1/2 Side 2

FIELD DATA

CALCULATIONS

TRAVERSE POINT NUMBER	VELOCITY HEAD (in. H ₂ O)	STACK TEMP., °F
1	0.00	100
2	0.00	100
3	0.00	100
4	0.00	100
5	0.00	100
6	0.00	100
7	0.00	100
8	0.00	100
9	0.00	100
10	0.00	100
11	0.00	100
12	0.00	100
13	0.00	100
14	0.00	100
15	0.00	100
16	0.00	100
17	0.00	100
18	0.00	100
19	0.00	100
20	0.00	100
21	0.00	100
22	0.00	100
23	0.00	100
24	0.00	100
25	0.00	100
26	0.00	100
27	0.00	100
28	0.00	100
29	0.00	100
30	0.00	100
31	0.00	100
32	0.00	100
33	0.00	100
34	0.00	100
35	0.00	100
36	0.00	100
37	0.00	100
38	0.00	100
39	0.00	100
40	0.00	100
41	0.00	100
42	0.00	100
43	0.00	100
44	0.00	100
45	0.00	100
46	0.00	100
47	0.00	100
48	0.00	100
49	0.00	100
50	0.00	100
51	0.00	100
52	0.00	100
53	0.00	100
54	0.00	100
55	0.00	100
56	0.00	100
57	0.00	100
58	0.00	100
59	0.00	100
60	0.00	100
61	0.00	100
62	0.00	100
63	0.00	100
64	0.00	100
65	0.00	100
66	0.00	100
67	0.00	100
68	0.00	100
69	0.00	100
70	0.00	100
71	0.00	100
72	0.00	100
73	0.00	100
74	0.00	100
75	0.00	100
76	0.00	100
77	0.00	100
78	0.00	100
79	0.00	100
80	0.00	100
81	0.00	100
82	0.00	100
83	0.00	100
84	0.00	100
85	0.00	100
86	0.00	100
87	0.00	100
88	0.00	100
89	0.00	100
90	0.00	100
91	0.00	100
92	0.00	100
93	0.00	100
94	0.00	100
95	0.00	100
96	0.00	100
97	0.00	100
98	0.00	100
99	0.00	100
100	0.00	100

$$\begin{aligned}
 P_1 &= P_0 + (2.046 \times 10^{-4}) \times \left(\frac{V_1^2}{C_p} \right) \times 10^6 \left(\frac{P_0}{P_1} \right) \\
 P_1 &= (29.95) + (2.046 \times 10^{-4}) \times \left(\frac{100^2}{1.25} \right) \times 10^6 \left(\frac{29.95}{P_1} \right) \\
 P_1 &= 29.95 + 16.37 \left(\frac{100^2}{P_1} \right) \\
 P_1^2 &= 29.95 P_1 + 1637 \\
 P_1^2 - 29.95 P_1 - 1637 &= 0 \\
 P_1 &= \frac{29.95 \pm \sqrt{29.95^2 + 4 \times 1637}}{2} \\
 P_1 &= \frac{29.95 \pm \sqrt{897.0025 + 26192}}{2} \\
 P_1 &= \frac{29.95 \pm \sqrt{27089.0025}}{2} \\
 P_1 &= \frac{29.95 \pm 164.61}{2} \\
 P_1 &= \frac{194.56}{2} \\
 P_1 &= 97.28 \text{ in.Hg} \\
 V_1 &= 100 \text{ ft/min} \\
 Q_1 &= V_1 \times A_1 \times \frac{60}{n} \\
 Q_1 &= 100 \times \left(\frac{\pi \times 11.5^2}{4} \right) \times \frac{60}{60} \\
 Q_1 &= 100 \times 102.67 \times 1 \\
 Q_1 &= 10267 \text{ ft}^3/\text{min} \\
 Q_{std} &= Q_1 \times \left(\frac{P_1}{P_{std}} \right) \times \left(\frac{T_{std}}{T_1} \right) \\
 Q_{std} &= 10267 \times \left(\frac{97.28}{29.95} \right) \times \left(\frac{520}{540} \right) \\
 Q_{std} &= 10267 \times 3.248 \times 0.963 \\
 Q_{std} &= 31300 \text{ ft}^3/\text{min}
 \end{aligned}$$

Method 25 Field Data

Plant ALB. RICHMOND WASTE MGT. CO. City ERIE PA Date 7-23-92
 Run No. 111-0-1 Sample Location FLARE OFFICE
 Operator D. CARROLL Tank _____
 Tank No. 04391 Volume, Liters 6
 Trap No. N/A Meter No. 5-5620
 Sampling Rate 1.20 cc/min. Filter Type _____ in stack
ENTERED 11 out of stack

Sample Train Data

	Barometric Pressure, mm Hg	Ambient Temperature, °F °C	Sample Tank Gauge Vacuum, mm Hg
Pre-Test	<u>760</u>	<u>83</u>	<u>722</u>
Post-Test	<u>760</u>	<u>122</u>	<u>248</u>

Front-Half - Leak Check Data

Time	Initial Vacuum, mm Hg	Time	Final Vacuum, mm Hg	ΔP, mm Hg	Leak Rate,* cc/min
<u>1120</u>	<u>722</u>	<u>1131</u>	<u>222</u>	<u>0</u>	<u>0.00</u>

Sampling Data

Clock	Time Sampling, min	Gauge Vacuum, mm Hg	Temperatures, °F	
			Probe	Filter Box
<u>1123</u>	<u>0</u>	<u>722</u>	<u>N/A</u>	<u>N/A</u>
<u>1143</u>	<u>10</u>	<u>606</u>		
<u>1153</u>	<u>20</u>	<u>510</u>		
<u>1203</u>	<u>30</u>	<u>408</u>		
<u>1213</u>	<u>40</u>	<u>316</u>		
<u>1223</u>	<u>50</u>	<u>269</u>		
<u>1233</u>	<u>60</u>	<u>249</u>		

$$* \text{Leak Rate, cc/min} = \frac{\Delta P, \text{ mm} \times \text{Volume, cm}^3}{Pb, \text{ mm} \times \text{Time, min}}$$

Leak rate must be <0.01 cc/min (0.6 cc/min leak rate at sampling rate of 60 cc/min)
 (volume = 100 cm³ for less than 4 ft probe)

Method 25 Field Data

Plant/Client Amoco Refinery of PA City PA Date 2/29/2002
 Run No. 111-2-2 Sample Location Flare Header
 Operator D. CARROLL Tank _____
 Tank No. 04326 Volume, Liters 6
 Trap No. 8/2 Rotameter No. 5-520
 Sampling Rate 1.05 cc/min. Filter Type _____ in stack
52522-1.1 out of stack

Sample Train Data

	Barometric Pressure, mm Hg	Ambient Temperature, °F °C	Sample Tank Gauge Vacuum, mm Hg
Pre-Test	<u>760</u>	<u>96</u> °	<u>723</u>
Post-Test	<u>760</u>	<u>102</u> °	<u>116</u>

Front-Hab - Leak Check Data

Time	Initial Vacuum, mm Hg	Time	Final Vacuum, mm Hg	ΔP, mm Hg	Leak Rate,* cc/min
<u>1325</u>	<u>723</u>	<u>1326</u>	<u>721</u>	<u>2</u>	<u>2.50</u>

Sampling Data

Time		Gauge Vacuum, mm Hg	Temperatures, °F	
Clock	Sampling, min		Probe	Filter Bpx
<u>1332</u>	<u>0</u>	<u>721</u>	<u>474</u>	<u>474</u>
<u>1342</u>	<u>10</u>	<u>622</u>		
<u>1352</u>	<u>20</u>	<u>496</u>		
<u>1402</u>	<u>30</u>	<u>385</u>		
<u>1412</u>	<u>40</u>	<u>279</u>		
<u>1422</u>	<u>50</u>	<u>190</u>		
<u>1432</u>	<u>60</u>	<u>116</u>		

* Leak Rate, cc/min = $\frac{\Delta P, \text{ mm Hg} \times \text{Volume, cm}^3}{P_b, \text{ mm Hg} \times \text{Time, min}}$

Leak rate must be <0.01 at sampling rate
 (0.8 cc/min leak rate at sampling rate of 80 cc/min)
 (volume = 100 cm³ for less than 4 ft probe)

Method 25 Field Data

Plant NORTH AMERICAN ROPE MFG CO City ELIZABETH Date 7-27-72
 Run No. 411-2-3 Sample Location FLARE COLLECT
 Operator D. CARROLL Tank _____
 Tank No. 2443 Volume, Liters 2.6
 Trap No. N/A Flowmeter No. 5-500
 Sampling Rate 100 cc/min. Filter Type _____ in stack
SINTERED G.S. out of stack

Sample Train Data

	Barometric Pressure, mm Hg	Ambient Temperature, °F °C	Sample Tank Gauge Vacuum, mm Hg
Pre-Test	<u>761</u>	<u>118</u>	<u>722</u>
Post-Test	<u>762</u>	<u>126</u>	<u>722</u>

Front-Hat* Leak Check Data

Time	Initial Vacuum, mm Hg	Time	Final Vacuum, mm Hg	ΔP, mm Hg	Leak Rate,* cc/min
<u>1505</u>	<u>722</u>	<u>1506</u>	<u>722</u>	<u>0</u>	<u>0.00</u>

Sampling Data

Clock	Time Sampling, min	Gauge Vacuum, mm Hg	Temperatures, °F	
			Probe	Filter Box
<u>1518</u>	<u>0</u>	<u>722</u>	<u>N/A</u>	<u>N/A</u>
<u>1528</u>	<u>10</u>	<u>628</u>		
<u>1538</u>	<u>20</u>	<u>520</u>		
<u>1548</u>	<u>30</u>	<u>394</u>		
<u>1558</u>	<u>40</u>	<u>285</u>		
<u>1608</u>	<u>50</u>	<u>191</u>		
<u>1618</u>	<u>60</u>	<u>122</u>		

* Leak Rate, cc/min = $\frac{\Delta P, \text{ mm} \times \text{Volume, cm}^3}{P_0, \text{ mm} \times \text{Time, min}}$

Leak rate must be <0.01 x sampling rate
 (0.8 cc/min leak rate at sampling rate of 80 cc/min)
 (volume = 100 cm³ for less than 4 ft probe)

MOISTURE
SAMPLE RECOVERY AND INTEGRITY SHEET

Plant Lake View Ranch Sample date 7/29/92
 Sample location 1st, 2nd, and 3rd Recovery date 7/29/92
 Run number 272-0-191 Recovered by Chad

MOISTURE

Balance # 422

	1st, 2nd, and 3rd	4th silica gel
Final volume	<u>256</u> ml	Final weight <u>245.9</u> g
Initial volume	<u>200</u> ml	Initial weight <u>200.0</u> g
Net Gain	<u>56</u> ml	Net gain <u>45.9</u> g

Total moisture 91.9 g

MOISTURE
SAMPLE RECOVERY AND INTEGRITY SHEET

Plant Lake Washington Sample date 7/23/97
Sample location Washelli Recovery date 7/23/97
Run number 27-8-62 Recovered by 1120

MOISTURE

Impingers	1st, 2nd, and 3rd	4th silica gel
Final volume	<u>237</u> ml	Final weight <u>222.3</u> g
Initial volume	<u>200</u> ml	Initial weight <u>200.0</u> g
Net Gain	<u>37</u> ml	Net gain <u>22.3</u> g

Total moisture 22.3 g

MOISTURE
SAMPLE RECOVERY AND INTEGRITY SHEET

Plant Lake Mead, Nevada Sample date 2/23/97
 Sample location Elmer, NV Recovery date 2/23/97
 Run number 2750-01 Recovered by 1250

MOISTURE

Impingers	1st, 2nd, and 3rd	4th silica gel
Final volume	<u>240</u> ml	Final weight <u>246.8</u> g
Initial volume	<u>200</u> ml	Initial weight <u>200.0</u> g
Net Gain	<u>40</u> ml	Net gain <u>46.8</u> g
Total moisture		<u>46.8</u> g



DRY MOLECULAR WEIGHT DETERMINATION

PLANT Lakeview Laundry
 DATE 7/27/12 COMMENTS Oxyd #423
 TEST NO CVL-0-1

SAMPLING TIME (MM:SS) 00:00
 SAMPLING LOCATION 00000
 SAMPLE TYPE (PAG, INTEGRATED, CONTINUOUS) W kg
 ANALYTICAL METHOD 0.2
 AMBIENT TEMPERATURE 0.5
 OPERATOR [Signature]
 ORSAT LEAK CHECKED ✓

GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) #4 is 15 mole
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	10.3	10.3	10.5	10.5	10.4	10.4	10.4	4/100	
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)	14.0	35.7	14.3	5.8	14.3	5.9	5.8	32/100	
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)								28/100	
N ₂ (NET IS 100 MINUS ACTUAL CO READING)								28/100	
TOTAL									

DRY MOLECULAR WEIGHT DETERMINATION

PLANT Lake View Landfill COMMENTS: _____

DATE 7/24/92 TEST NO. LV-0-7

SAMPLING TIME (AM/PM) 0800

SAMPLING LOCATION 0000

SAMPLE TYPE (GAS, INTEGRATED, CONTINUOUS) at top

ANALYTICAL METHOD Orsat

AMBIENT TEMPERATURE 81°F

OPERATOR [Signature]

ORSAT LEAK CHECKED _____

GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) #g lb lb-mole
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	10.5	10.5	10.4	10.4			10.4	44/100	
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)	17.0	5.9	16.9	6.4			6.4	32/100	
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)								28/100	
N ₂ (NET IS 100 MINUS ACTUAL CO READING)								28/100	
TOTAL									

DRY MOLECULAR WEIGHT DETERMINATION

PLANT Lake View Landfill
 DATE 7/24/92 10000 LVL-0-3
 COMMENTS:
 SAMPLING TIME (min) 0.000
 SAMPLING LOCATION 0.000
 SAMPLE TYPE (GAS, INTEGRATED, CONTINUOUS) Int Gas
 ANALYTICAL METHOD GC/MS
 AMBIENT TEMPERATURE 70°F
 OPERATOR MO
 ORSAT LEAK CHECKED ✓

GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) #2 to 10 mole
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	10.5	10.5	10.5	10.5	10.5	10.5	10.5	44/100	
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL O ₂ READING)	16.9	6.4	17.0	6.5	16.9	6.4	16.9	32/100	
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)								28/100	
MP ₂ (NET IS 100 MINUS ACTUAL CO READING)								28/100	
TOTAL									

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THE UNIVERSITY OF CHICAGO

1982

Figure 6

Figure 1

RESEARCH

THE UNIVERSITY OF CHICAGO

[illegible]

SECRET

FLN NO.	SAMPLE TIME (Z+M)	FLASK AND VALVE NOS	VOLUME OF FLASK AND VALVE, ml	FLASK TEMP. (ACTUAL), °F	INITIAL FLASK PRESSURE		FLASK TEMP. (FURN), °F	FINAL FLASK PRESSURE	
					LEG 1 in. Hg	LEG 2 in. Hg		LEG 1 in. Hg	LEG 2 in. Hg
LVL-A-0-15	1645	MM	2049.2	87	13.6	13.6	73	1.6	1.5
"	1700	MM	1966.3	87	13.5	13.5	74	0.6	0.5
"	1715	MM	2057.6	87	13.5	13.5	74	1.6	1.5
"	1720	MM	2029.4	89	13.5	13.5	75	1.8	1.7
LVL-A-0-20	1745	106	1992.9	88	13.6	13.6	73	1.6	1.5
"	1800	5K8	1959.6	88	13.4	13.4	74	0.7	0.6
"	1815	2K8	1932.5	89	13.5	13.5	74	1.8	1.7
"	1830	22	2064.2	90	13.5	13.5	73	2.3	2.2
LVL-A-0-30	1845	6C	2044.9	89	13.1	13.1	74	1.7	1.6
"	1900	1R	2052.0	89	13.6	13.6	73	0.5	0.4
"	1915	2R	2094.8	90	13.6	13.6	74	1.8	1.7
30	1930	MM	2051.0	90	13.5	13.5	74	0.6	0.5
		7C							

FIELD AUDIT REPORT: DRY GAS METER
BY CRITICAL ORIFICE

DATE: 7-28-92 CLIENT: WASTE MGT
BAROMETRIC PRESSURE (P_{bar}): 30.232 in.Hg METER BOX NO. 157-10
ORIFICE NO. 3 PRETEST Y: 1.78 AND 2.21 in.H₂O
ORIFICE K FACTOR: 5.548 x 10⁻⁶ AUDITOR: D. C. M. M. M.

Orifice manometer reading ΔH , in. H ₂ O	Dry gas meter reading V_i/V_f , ft ³	Temperatures					Duration of run θ min.
		Ambient		Dry gas meter			
		T_{af}/T_{af} , °F	Average T_a , °F	Inlet T_{if}/T_{if} , °F	Outlet T_{of}/T_{of} , °F	Average T_m , °F	
2.55	379.80	80	80	86	83	86	16 52.45
	370.00	80		80	86		60

Dry gas meter V_m , ft ³	$V_{m, std}$, ft ³	$V_{m, act}$, ft ³	Audit, Y	Y devia- tion, %	Audit ΔH , in.H ₂ O	ΔH Devia- tion, in.H ₂ O
<u>15.20</u>	<u>17.122</u>	<u>16.640</u>	<u>.986</u>	<u>0.82</u>	<u>1.90</u>	<u>0.11</u>

$$V_{m, std} = \frac{17.647(V_m)(P_{bar} + \Delta H/13.6)}{(T_m + 460)} = \frac{29.06 \times 30.232 \times 19.827}{596.546} = 17.122 \text{ ft}^3$$

$$V_{m, act} = \frac{1203(\theta)(K)(P_{bar})}{(T_a + 460)} = \frac{1203(16)(5.548 \times 10^{-6})(30.232)}{540} = 16.640 \text{ ft}^3$$

$$\text{Audit } Y = \frac{V_{m, act}}{V_{m, std}} = .986 \quad Y \text{ deviation} = \frac{\text{Audit } Y - \text{Pretest } Y}{\text{Pretest } Y} \times 100 = .53 \%$$

$$\text{Audit } \Delta H = (0.0317)(\Delta H)(P_{bar})(T_m + 460) \left[\frac{1}{V(V_m)(P_{bar} + \Delta H/13.6)} \right]^2 = 1.90 \text{ in.H}_2\text{O}$$

Audit Y must be in the range, pretest $\pm 0.05 Y$.
Audit ΔH must be in the range pretest ± 0.15 inches H₂O.

THERMOCOUPLE DIGITAL INDICATOR
AUDIT DATA SHEET

Date 7-28-92 Indicator No. ET-10 Operator D. C. Miller

Test Point No.	Millivolt signal*	Equivalent temperature, °F ^a	Digital indicator temperature reading, °F	Difference, %
1		1600	1591	0.44
2		1700	1690	0.46
3		1800	1789	0.44
4		1900	1888	0.50

Percent difference must be less than or equal to 0.5%.

Percent difference:

$$\frac{(\text{Equivalent temperature } ^\circ\text{R} - \text{Digital indicator temperature reading } ^\circ\text{R})(100\%)}{(\text{Equivalent temperature } ^\circ\text{R})}$$

Where $^\circ\text{R} = ^\circ\text{F} + 460^\circ\text{F}$

* These values are to be obtained from the calibration data sheet for the calibration device.

APPENDIX C
LABORATORY DATA



ANALYTICAL SERVICES

CERTIFICATE OF ANALYSIS

ITT Corporation

Date: September 30, 1992

Attn: Brian Garls

P.O. Number 332150

This is the Certificate of Analysis for the following samples:

Client Project ID: North American Waste Management
Date Received: July 31, 1992
Work Order: XC2-07-175
Number of Samples: 20
Sample Type: Air

I. Introduction

Twenty samples arrived at ITAS Cincinnati on July 31, 1992. The samples were collected on July 29, 1992 and were labeled as follows:

Canister # EVL-C-1 Out	Flask # EVL-N-C-1A	Flask # EVL-N-C-2B
Canister # EVL-C-2 Out	Flask # EVL-N-C-1B	Flask # EVL-N-C-3A
Canister # EVL-C-3 Out	Flask # EVL-N-C-1C	Flask # EVL-N-C-3B
Canister # EVL-I-1 Inlet	Flask # EVL-N-C-1D	Flask # EVL-N-C-3C
Canister # EVL-I-2 Inlet	Flask # EVL-N-C-2A	Flask # EVL-N-C-3D
Canister # EVL-I-3 Inlet	Flask # EVL-N-C-2B	Flask # EVL-N-Blank
Canister # EVL-Blank	Flask # EVL-N-C-2C	

II. Analytical Results/Methodology

The analytical results for this report are presented by analytical test. The data will include sample identification information, the analytical results, and the appropriate detection limits.

The analyses requested included and methods used are listed on the following page.

Reviewed and Approved by:

Laurel Tomason
Project Manager
307175

American Council of Independent Laboratories
International Association of Environmental Testing Laboratories
American Association for Laboratory Accreditation

II. Analytical Results/Methodology (cont.)

- " Oxygen, Nitrogen, Carbon Monoxide, Methane,
Carbon Dioxide and H₂O by EPA Method TO-12
by GC/FID
- " Volatile Organics by Gas Chromatography/Mass
Spectrometry; EPA Method TO-14
- " HOC; EPA Method TA

III. Quality Control

Immediately following the analytical data for the samples can be found the QA/QC information that pertains to these samples. The purpose of this information is to demonstrate that the data enclosed is scientifically valid and defensible. This QA/QC data is used to assess the laboratory's performance during the analysis of the samples it accompanies. All quantitations were performed within the calibrated range of the analytical instrument.

IV. Data Report Qualifiers

Following are descriptions of data report qualifiers which may have been used in this analytical report.

U The analyte was not detected in the sample or extract. The value reported with the "U" is the detection limit for that compound in that sample.

VALUE The result is a value equal to or greater than the detection limit for that compound.

J Indicates an estimated value. This flag is used when mass spectral data indicates the presence of the compound, but the result is less than the specified detection limit.

B This flag is used whenever the analyte is found in the blank as well as in the sample.

E This flag indicates that the quantity of this compound detected in this sample is above the linear range of the instrument. Results are probably lower than actual.

V. Comments

The samples were diluted in helium to accommodate the HOC analysis. Therefore, the data for volatile organics may be biased high.

The report is being reviewed to include ITA's additional substructure data for HOC by EPA Method TO-12 GC/FID. EPA Method ASTM D3416 and EPA Method 25 did not have adequate detection limits.

Client: EAGS NAME
Work Order: K2-07-175
20717502

IT ANALYTICAL SERVICES
CINCINNATI, OH

=====

Analytical Results, Total ug

Client Sample ID	Lab No.	NCr 1st Run	NCr 2nd Run	NCr Average
Flask # EVL-N-C-1A	08	60	65	63
Flask # EVL-N-C-1B	09	74	68	71
Flask # EVL-N-C-1C	10	83	81	82
Flask # EVL-N-C-1D	11	60	61	60
Flask # EVL-N-C-2A	12	88	84	86
Flask # EVL-N-C-2B	13	71	71	71
Flask # EVL-N-C-2C	14	65	69	67
Flask # EVL-N-C-2D	15	70	67	68
Flask # EVL-N-C-3A	16	77	76	77
Flask # EVL-N-C-3B	17	56	60	58
Flask # EVL-N-C-3C	18	50	51	51
Flask # EVL-N-C-3D	19	71	69	70
Flask # EVL-N-Blank	20	ND	ND	ND

Detection Limit = 26 Total ug

ND = Not detected at or above the reported detection limit

Client: ITRQS NAVM
Work Order: R2-07-175
20117560

IT ANALYTICAL SERVICES
CHROMENATE, CH

Volatile Organics

Client ID:

Carloter # LVL-0-1 Cat

Date Analyzed: 8/24/92,

8/26/92

Dilution Factors: 27

Lab No.:

R2-01-175-01

Compound	ppb v/v	Detection Limit
Vinyl Chloride	ND	5.4
Chloroethane	ND	5.4
1,1-Dichloroethene	ND	5.4
Methylene Chloride	710 (1)	240
Chloroform	ND	5.4
Benzene	ND	5.4
1,2-Dichloroethane	ND	5.4
1,1,1-Trichloroethane	5.5	5.4
Carbon Tetrachloride	ND	5.4
Trichloroethene	ND	5.4
Tetrachloroethene	ND	5.4

(1) Dilution Factor = 1214

RE ANALYTICAL SERVICES
CONFIDENTIAL, CFI

Client ID:	Canister # LVL-C-2 Out	Date Analyzed: 8/28/92
Lab No.:	XC-01-175-02	Dilution Factor: 20

(3.) Dilution Factor = 1000

(C-6)

Client: ECHOS NASH
Work Order: K2-01-175
20717542

IT ANALYTICAL SERVICES
CINCINNATI, OH

=====

Volatile Organics

Client ID: Canister # LVL-0-3 Date: Date Analyzed: 8/24/92
Lab No.: K2-01-175-03 Dilution Factor: 26

Compound	ppb v/v	Detection Limit
Vinyl Chloride	ND	5.2
Chloroethane	ND	5.2
1,1-Dichloroethane	ND	5.2
Methylene Chloride	330 (1)	240
Chloroform	ND	5.2
Benzene	ND	5.2
1,2-Dichloroethane	ND	5.2
1,1,1-Trichloroethane	ND	5.2
Carbon Tetrachloride	ND	5.2
Trichloroethene	ND	5.2
Tetrachloroethene	ND	5.2

(1) Dilution Factor = 1178

ND = Not detected at or above the reported detection limit

Client: ITAPS NAME
Work Order: K2-07-175
REV17543

IT ANALYTICAL SERVICES
CINCINNATI, OH

Volatile Organics

Client ID:

Cartridge # EVL-Blank

Date Analyzed: 8/26/92

Dilution Factors: 10

Lab No.:

K2-01-175-07

Compound	ppb v/v	Detection Limit
Vinyl Chloride	1.3 J	2
Chloroethane	ND	2
1,1-Dichloroethane	ND	2
Methylene Chloride	2.7	2
Chloroform	ND	2
Benzene	ND	2
1,2-Dichloroethane	ND	2
1,1,1-Trichloroethane	ND	2
Carbon Tetrachloride	ND	2
Trichloroethene	ND	2
Tetrachloroethene	1.2 J	2

ND = Not detected at or above the reported detection limit

Client: ITAGS RM004
Work Order: 32-01-175
20717544

ET ANALYTICAL SERVICES
CONCINNATI, OH

=====

Volatile Organics

Client ID: Method Blank: Date Analyzed: 8/24/92
Lab No.: ABLEVL Dilution Factors: 1

Compound	ppb v/v	Detection Limit
Vinyl Chloride	ND	0.2
Chloromethane	ND	0.2
1,1-Dichloroethane	ND	0.2
Methylene Chloride	ND	0.2
Chloroform	ND	0.2
Benzene	ND	0.2
1,2-Dichloroethane	ND	0.2
1,1,1-Trichloroethane	ND	0.2
Carbon Tetrachloride	ND	0.2
Trichloroethene	ND	0.2
Tetrachloroethene	ND	0.2

ND = Not detected at or above the reported detection limit

Client: ITAPS BANGOR
Work Order: 82-07-175
20/1/75-05

IT ANALYTICAL SERVICES
CINCINNATI, OH

=====

Volatile Organics

Client ID: Method Blank Date Analyzed: 8/28/92
Lab No.: REC304 Dilution Factor: 1

Compound	ppb w/v	Detection Limit
Vinyl Chloride	ND	0.2
Chloromethane	ND	0.2
1,1-Dichloroethane	ND	0.2
Methylene Chloride	ND	0.2
Chloroform	ND	0.2
Benzene	ND	0.2
1,2-Dichloroethane	ND	0.2
1,1,1-Trichloroethane	ND	0.2
Carbon Tetrachloride	ND	0.2
Trichloroethane	ND	0.2
Tetrachloroethane	ND	0.2

ND = Not detected at or above the reported detection limit

Client: IT&QS NAWA
 Work Order: X2-07-175
 20717503

IT ANALYTICAL SERVICES
 CINCINNATI, OH

Analytical Results, 0 (1)

Client Sample ID	Canister # EVL-Q-1 Out	Canister # EVL-Q-2 Out	Canister # EVL-Q-3 Out	Detection Limit
Lab No.	01	02	03	
Analyte				
Oxygen	11.2	11	9.6	0.1
Nitrogen	79.8	79.8	80.4	0.1
Carbon Monoxide	ND	ND	ND	0.1
Methane	ND	ND	ND	0.1
Carbon Dioxide	9.0	9.2	10	0.1

Client Sample ID	Canister # EVL-I-1 Inlet	Canister # EVL-I-2 Inlet	Canister # EVL-I-3 Inlet	Detection Limit
Lab No.	04	05	06	
Analyte				
Oxygen	1.3	0.8	0.8	0.3
Nitrogen	51.8	4.2	4.0	0.3
Carbon Monoxide	ND	ND	ND	0.3
Methane	53.8	54.9	55.4	0.3
Carbon Dioxide	39.1	40.1	39.8	0.3

Client Sample ID	Canister # EVL-Blank	Detection Limit
Lab No.	07	
Analyte		
Oxygen	ND	0.1
Nitrogen	ND	0.1
Carbon Monoxide	ND	0.1
Methane	ND	0.1
Carbon Dioxide	ND	0.1

ND = Not detected at or above the reported detection limit

(1) Results normalized to 1000

Client: ETOGS NAWM
Work Order: 22-01-175
20717550

IT ANALYTICAL SERVICES
CHICAGO, IL

Analytical Results, ppbv

Client Sample ID	Lab No.	RMOC *	Detection Limit
Canister # LVL-Q-1 Out	01	3100	27
Canister # LVL-Q-2 Out	02	4400	24
Canister # LVL-Q-3 Out	03	2000	26
Canister # LVL-E-1 Inlet	04	460000	1200
	04 DUP	420000	1200
Canister # LVL-E-2 Inlet	05	460000	1100
Canister # LVL-E-3 Inlet	06	470000	1300
Canister # LVL-Blank	07	1100	10
Method Blank		ND	10

ND = Not detected at or above the reported detection limit

* Referenced to Heptane



By Teacher Date 10/19/74 Subject Calculation of Blackwater Sheet No. 1 of 1
 Chd. By MF Date 10/19/74 Proj. No. 22342
 O.5cm. X O.5cm.

DATE - 3 June 2009 @ 1700 pph

How often I have enjoyed the company of 5030 9000

[illegible]

Letter to Book Editor Regarding my article

$$\left(15^{\circ}47' \pm \frac{26.0}{14.7}\right) + 25^{\circ}1' = 15^{\circ}47'.2$$

$$\text{Billets en circulation} = \frac{\text{Billets en circulation}}{\text{Billets en circulation}} = \frac{154722}{58346} = 2.650$$

Belong to the ^{group} Arthropoda = Arthropoda = Arthropoda

Bentuk Corrected Output Correctness			
	Amount	Bentuk	Corrected
	Output	Output	Correctness
Bentuk 1	3300	2000	1000
Bentuk 2	4000	2000	1500
Bentuk 3	3000	2000	0

Handwritten musical notation on a five-line staff, featuring various notes, rests, and a key signature of one sharp (F#).

Dilution Record of IT-AQS-DNA-WM With ME AT 15 PSI

DATE:

07/30/92 275°K

ANALYST:

MEWM

INTERFERENCE

PRESSURE (mm Hg)

750

Work order # 322-CT-175

Pressure Dilution Table

(54)

(31)

	Sample	Sample	P. Initial	P. Initial	P. Final	P. Final	Dilution
			Initial	Initial	Final	Final	Factor
For Data Entry	03A	04391/2.1	-6.45	573.2	13.6	1,952.0	3.030
	03A	04335/2.6	-6.45	567.2	13.2	1,933.6	3.043
	03A	04163/2.9	-6.39	591.0	13.6	1,932.0	3.036
	04A	04170/2.5	-6.30	604.3	13.6	1,932.0	3.038
	05A	04742/2.6	-6.60	624.2	16.4	1,999.1	3.532
	06A	11961/2.7	-6.75	640.0	16.2	1,614.6	3.561
Blank	07A	03912/2.3	-29.60		13.4		1.000

FIELD BLANK

4644

03 8/12/92

Dilution Factor Table

Interference Data Location: 00001, 00002, 00003

APPENDIX D

SAMPLING AND ANALYTICAL PROCEDURES

DETERMINATION OF SELECTED VOLATILE ORGANIC COMPOUNDS AND TOTAL GASEOUS NONMETHANE ORGANIC EMISSIONS

Sampling and analysis for selected volatile organic compounds (VOC) and total gaseous nonmethane organic compounds (NMOC) are conducted in accordance with sampling procedures described in U.S. Environmental Protection Agency (EPA) Method TO-14* and proposed U.S. EPA Method 25C.** Since both methodologies required evacuated canisters as the collection media, one sample is obtained and analyzed by the respective procedures of each method for the required pollutants.

Sampling Apparatus

A schematic of the sampling apparatus used at the landfill gas flare inlet is shown in Figure D-1. Figure D-2 depicts the sampling apparatus used at the landfill gas flare outlet. The outlet sampling train differs from the inlet by the addition of a moisture knock-out jar, which is used to remove possible condensation produced by cooling the exhaust gases from the exit of the landfill gas flare. The sampling trains used in these tests meet design specifications established by the U.S. EPA. Assembled by IT personnel, the train consists of the following:

Gas Collection Tank - 316 stainless steel tank with premeasured volume of approximately 6.0 liters. Tank is fitted with a valve and the interior surfaces are SUMMATM passivated.

Sampling Regulator System - A mercury manometer to monitor changes in tank pressure during sampling, stainless steel flow shutoff valve to start and stop sample flow, 2- μ m sintered stainless steel in-line filter, and calibrated preset stainless steel flow meter to maintain constant flow rate.

* Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air. EPA-600/4-84-041. April 1984.

** 56 FR 24468, May 30, 1991.

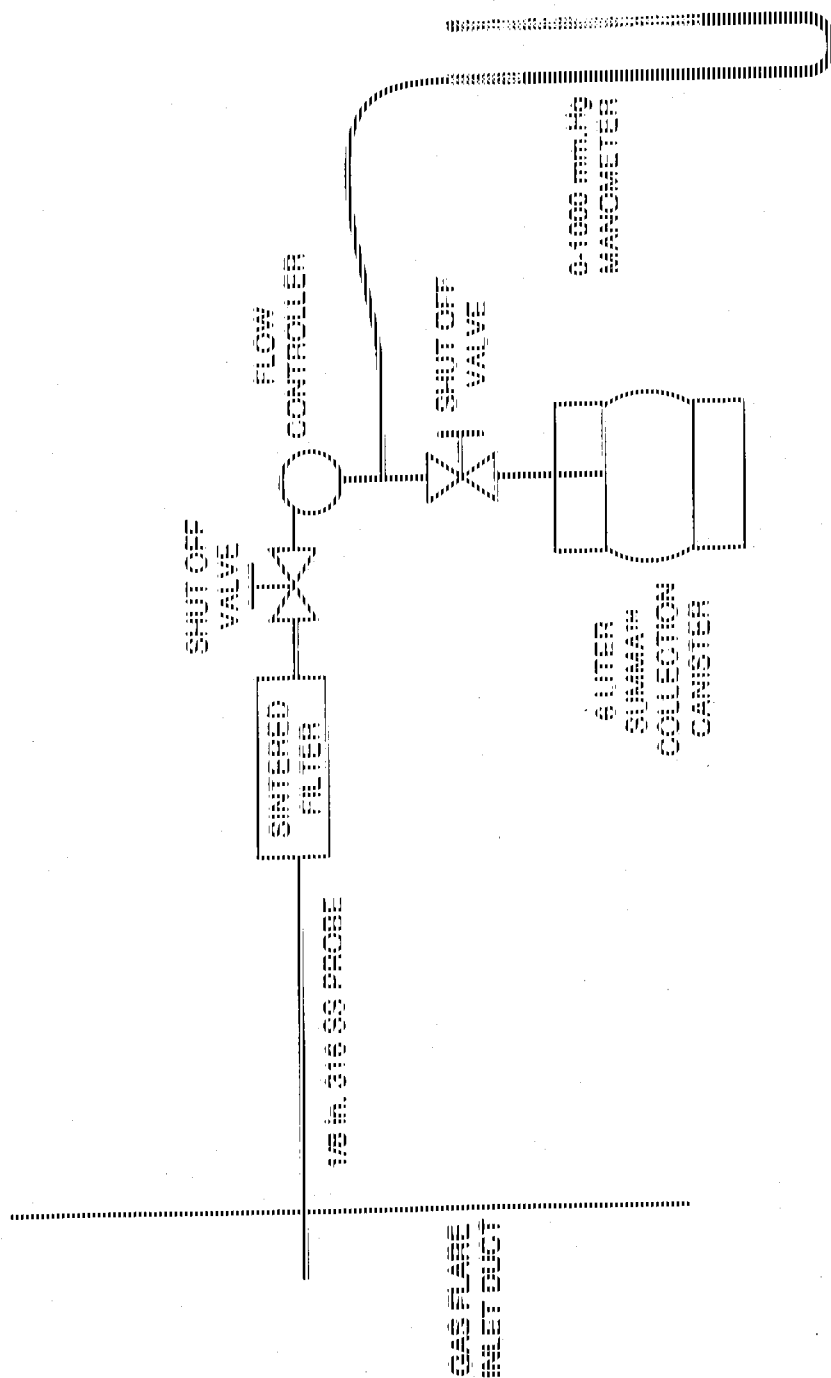


Figure D-1. VOC/MO sampling train - here installed.

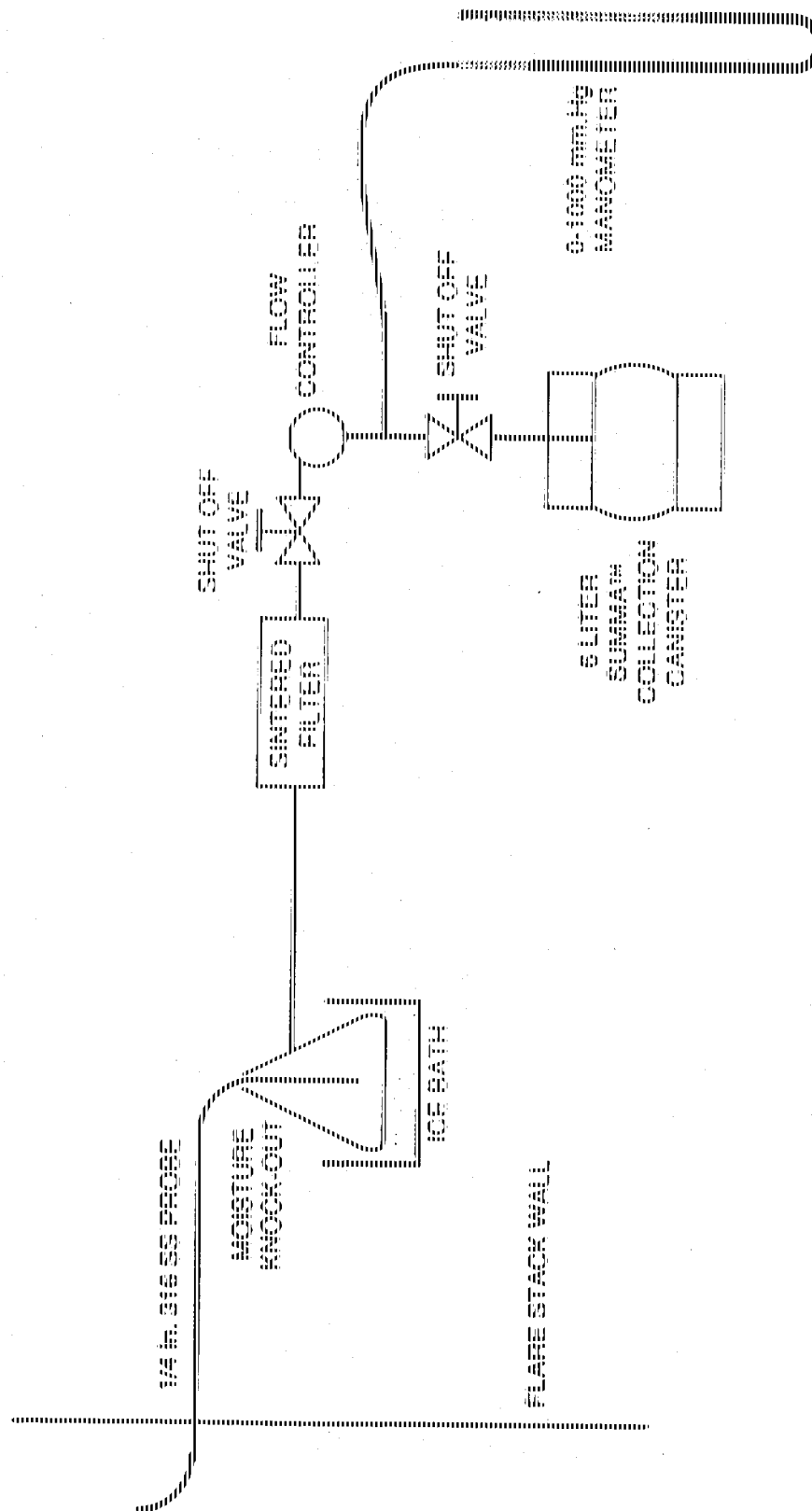


Figure D-2. VOC/NVOC sampling train - flare outlet.

Pressure Gauge - U-tube mercury manometer with a range of 0 to 1000 millimeters of mercury (mmHg) to measure tank pressure to the nearest 1 mmHg.

Vacuum Pump - Single-stage vacuum pump to evacuate sample tanks to an absolute pressure of 10 mmHg.

Sampling Procedures

Prior to sampling, the collection tank is evacuated to 10 mmHg absolute pressure and the sampling train is allowed to sit for a predetermined period of time. An acceptable leak check is considered to be no change in pressure during this period. This procedure is conducted in the laboratory before transport to the sampling site. At the sampling site (prior to sampling) the internal pressure is checked and recorded to verify no leakage occurs during transport. Sampling proceeds if the tank pressure is within 10 mmHg of absolute pressure. If the canister fails this check, it is either discarded or reevacuated and observed to determine the movement within a predetermined period time. If at this time no change is observed, the canister is considered acceptable for use. The sampling trains are assembled as shown in Figures D-1 and D-2. The outlet sampling train is leak-checked from the probe inlet to the shutoff valve on the collection tank. This is conducted following procedures described in U.S. EPA Method 25.* The ambient temperature and barometric pressure are recorded.

After the leak check, the probe is inserted into the gas stream, the valve connected to the collection tank is closed, and the sampling train is purged with sample gas using the vacuum pump. At least two sample probe volumes are evacuated from the system at a flow rate similar to that of sampling. Once the purge is complete, sampling is begun by closing the vacuum pump valve and opening the collection tank valve. The flow is maintained at a constant rate by the preset flow controller. Manometer readings are recorded at 10-minute intervals throughout the sampling period.

After completion of the test, the collection tank pressure and temperature are measured and the train is disassembled and transported to the laboratory for the appropriate analyses.

* 40 CFR 60, Appendix A, July 1991.

Analytical Procedure for Selected VOCs

Because the collection tanks are sampled down to near ambient pressures, the tanks are pressurized with humidified helium in the laboratory to facilitate sample travel from the canister as a result of positive pressure release. A dilution factor is calculated for each canister and is reported on the analytical data sheets presented in Appendix C. Analyses are performed for the pollutants listed in Table D-1 by removing an aliquot from the collection tank and injecting it into the gas chromatograph/mass spectroscopy (GC/MS) sample loop. Sample components are separated by GC utilizing fused silica capillary columns. Mass analysis is performed on each peak and ion counts are recorded for quantitation by GC/MS full scan.

TABLE D-1. VOC COMPOUNDS

Benzene	Perchloroethylene
Chloroform	Trichloroethylene
1,2-Dichloroethane	Carbon tetrachloride
Methyl chloride	Vinylidene chloride
Methylene chloride	Vinyl chloride

Each SUMMATM polished collection is cleaned and batch blank-checked before sampling. The cleaning cycle consists of placing the tanks on a manifold system within a 150°C heated oven, then evacuating and pressurizing the tanks. The evacuation/pressurization cycles are conducted 25 times and then the canisters are removed under vacuum. One tank from each batch is blank-checked and the batch is considered clean if all target compounds are less than 0.2 ppb and there are no unknown peaks not already present in the direct analysis of the ultrapure air.

The analytical system is tuned daily with 4-bromofluorobenzene to meet the key ion and ion abundance criteria of TO-14.

An initial minimum three-point calibration is conducted; check standards are run every 12 hours.

Sample identities are confirmed by Triple Plot mass spectra, which simultaneously display the enhanced target compound spectra, the unenhanced target compound spectra, and the library spectra of the target compound.

Analytical Procedures for Permanent Gases

Prior to analysis, the gas chromatograph/thermal conductivity detector (GC/TCD) is calibrated with at least three concentrations in the expected range of each compound of interest and the linearity of the detector is verified. An aliquot of the sample is extracted from the canister and injected into the GC. Each sample is analyzed in duplicate.

Analytical Procedures for NMVOCs (TO-12*)

Analytical System Leak Check

The analytical system is pressurized to approximately 50 psig with zero air. If after 3 hours the pressure does not drop more than 2 psig, the system is considered leak-tight.

System Calibration

The calibration curve is determined by injecting five or more gas standards on the range to be used. Each standard is analyzed three times. The relative standard deviation for the three analyses should be less than 3 percent. Linearity is expected and should be checked. If nonlinearity is observed, the problem is identified and corrected.

Sample Analysis

The NMVOC sample container is attached to the sample manifold. A vacuum reservoir draws the sample through the system where the NMVOCs are trapped in the cryogenic cooler. The system flow is then reversed and the trap is heated to 90°C. The NMVOCs volatilize and are directed to the gas chromatograph/flame ionization detector (GC/FID) for measurement. The GC/FID is calibrated using heptane standards; therefore, all results are stated as heptane.

* Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air, EPA-600/4-94-041, June 1997.

DETERMINATION OF NITROGEN OXIDE EMISSIONS

This test program follows the sampling procedures described in EPA Method 7A.*

Sampling Apparatus

The nitrogen oxide (NO_x) sampling train used in these tests meets design specifications established by the Federal EPA. Assembled by ITACS personnel, it consists of the following:

Probe - Borosilicate glass tubing with a heating system capable of maintaining a gas temperature of approximately 250 °F at the exit end to prevent water condensation. A glass-wool plug is placed in the end of the probe to remove particulate matter, as shown in Figure D-3.

Collection Flask - Two-liter borosilicate, round-bottom flask, with short neck and 24/40 standard taper opening, protected against implosion or breakage.

Flask Valve - Two-way stopcock connected to a 24/10 standard taper joint.

Manifold - Two- and three-way Teflon stopcock and connected 24/10 standard taper joints.

Temperature Gauge - Dial-type thermometer capable of measuring 2 °F intervals from 25 ° to 125 °F.

Vacuum Line - Tubing capable of withstanding a vacuum of 3 in.Hg absolute pressure, with "T" connection and T-bore stopcock.

Vacuum Gauge - U-tube manometer, 36 in.Hg with 0.1-in. divisions.

Pump - Capable of evacuating the collection flask to a pressure equal to or less than 3 in.Hg absolute.

Barometer - Aneroid type to measure atmospheric pressures to ± 0.1 in.Hg.

* 40 CFR 60, Appendix A, July 1990.

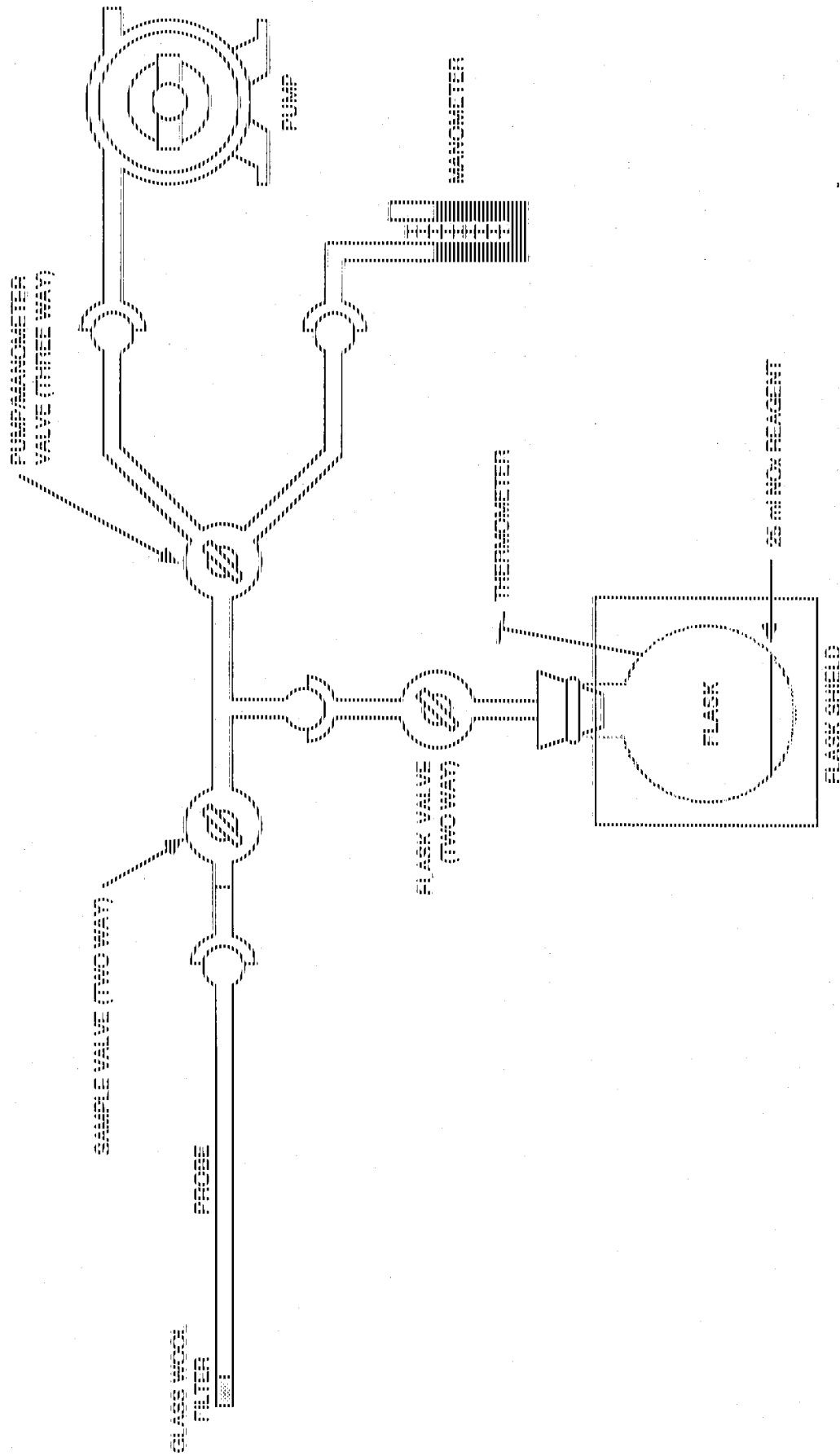


Figure D-3. Method 7 sampling train.

Sampling Procedure

Following the selection of the sampling site and minimum number of traverse points, the stack pressure, temperature, moisture, and range of velocity head are measured according to procedures described in EPA Methods 1 through 4.*

One sampling point, representing the average flow rate, is chosen. The sampling flask is charged with 25 mL of absorbing solution (made by mixing 1 liter of distilled water, 2.8 mL of concentrated H_2SO_4 , and 6 mL of 3 percent H_2O_2). A portion of the reagent is retained for use in preparing the calibration standards.

With the sample valve closed, the flask valve open, and the pump/manometer valve open to pump, the flask is evacuated to less than 3 in.Hg absolute pressure. Flask vacuum is determined by moving the pump/manometer valve to manometer. Leakage is checked by observing the manometer for pressure fluctuation. Any variation greater than 0.4 in.Hg over a 1-minute period is corrected before sampling.

With the flask closed, the sample valve open, and the pump/manometer valve turned to pump, the sampling apparatus is purged for 5 minutes. The probe heater setting is adjusted to prevent any visible condensation. With the sample valve closed and the pump/manometer valve open to manometer, an initial flask vacuum is determined by opening the flask valve. The sample valve is then slowly opened, allowing stack gas to enter the flask. The vacuum is allowed to drop to 3 in.Hg (approximately 15 seconds), and then the sample flask valves are closed. The flask is shaken for 5 minutes to ensure contact between the sample and absorbing solution.

Sample Recovery Procedure

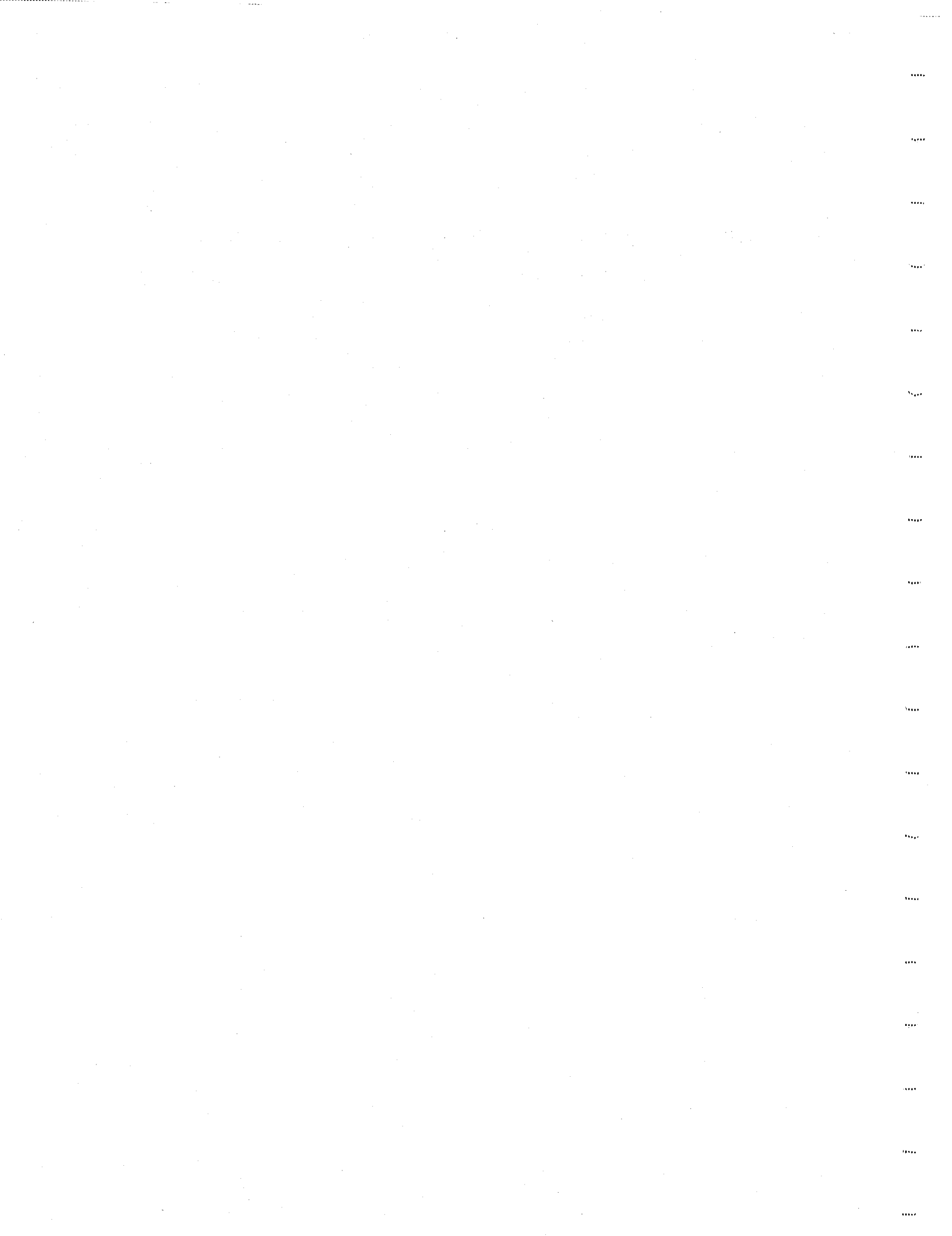
The collection flasks are transported to the laboratory for sample recovery. The flasks are set aside for a minimum of 16 hours. At the end of the period, the flasks are shaken for 2 minutes and the final flask pressure and temperatures are measured. All pertinent data are recorded on the recovery data sheets.

* 40 CFR 60, Appendix A, July 1990.

The contents of the flask are measured volumetrically and transferred to a polyethylene container. The flask is then rinsed with two separate 5-mL portions of deionized, distilled water, and the solution is added to the same container.

Analytical Procedure

The volume of absorbing solution is recorded and diluted to 50 mL with deionized, distilled water. A 5-mL aliquot of this solution is pipetted to a volumetric flask and diluted to 50 mL with deionized water. The samples from the second dilution are analyzed by ion chromatography. A calibration curve of $\mu\text{g NO}_3^-$ versus peak area response is prepared using standards of known concentration.



APPENDIX E

CALIBRATION PROCEDURES AND RESULTS

CALIBRATION PROCEDURES AND RESULTS

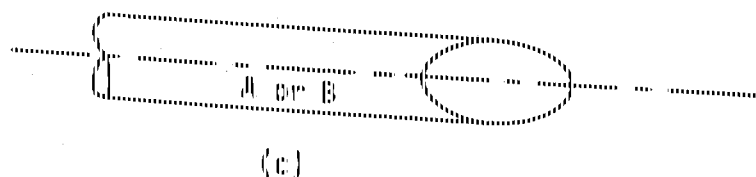
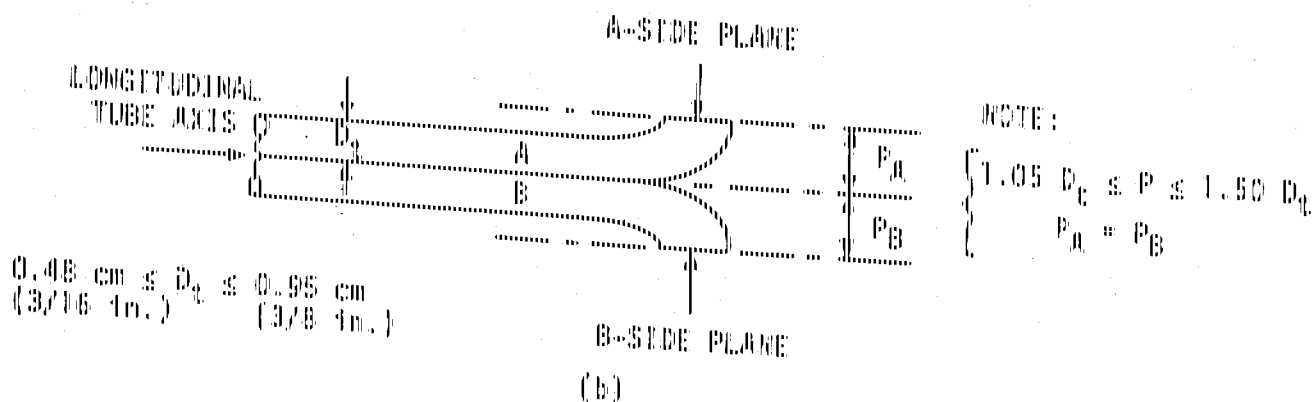
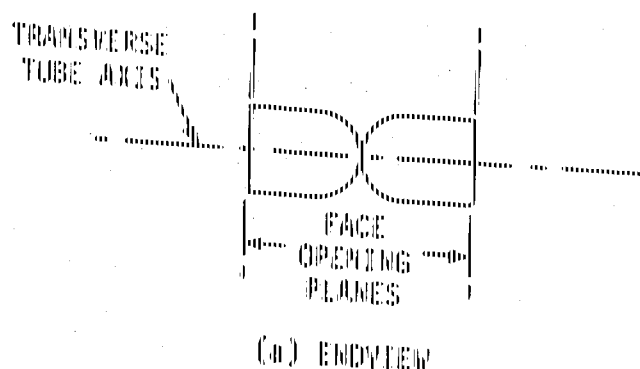
All of the equipment used is calibrated in accordance with the procedures outlined in the Quality Assurance Handbook for Air Pollution Measurement Systems, Volume III.^{*} The following pages describe these procedures and include the data sheets.

^{*}EPA 600/4-77-027b.

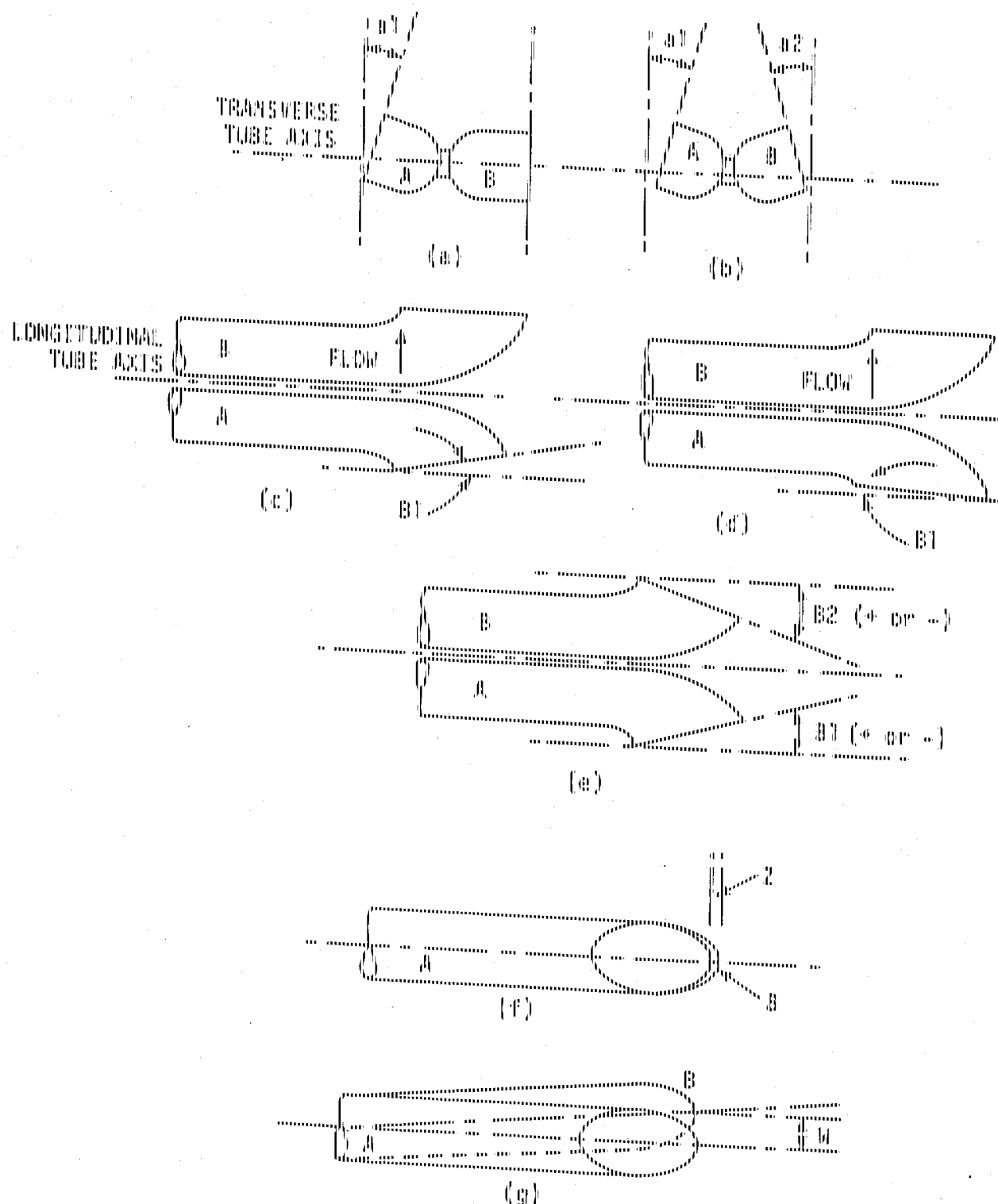
Pitot Tube Calibration

Each pitot tube used in sampling is constructed by ITACS and meets all requirements of EPA Method 2, Section 4.1.* Therefore, a baseline coefficient of 0.84 is assigned to each pitot tube. The following pages show the alignment requirements of Method 2 and the Pitot Tube Inspection Data Sheet(s) for each pitot tube used during the test program.

*40 CFR 60, Appendix A, July 1989.



Properly constructed Type S pitot tubes shown in: (a) end view, face opening planes perpendicular to transverse axis; (b) top view, face opening planes parallel to longitudinal axis; (c) side view, both legs of equal length and centerlines coincident when viewed from both sides. Baseline coefficient values of 0.84 may be assigned to pitot tubes constructed this way.



Types of face-opening misalignment that can result from field use or improper construction of Type S pitot tubes. These will not affect C_p as long as a_1 and a_2 are $<10^\circ$, B_1 and B_2 are $<5^\circ$, z is <0.32 (1/8 in.), and w is <0.08 cm (1/32 in.).



PITOT TUBE INSPECTION DATA SHEET

Pitot Tube No. 505-3' Date 12-9-91 Inspector T. KERRICK

α_1 degrees	α_2 degrees	β_1 degrees	β_2 degrees
<u>2</u>	<u>3</u>	<u>2</u>	<u>3</u>
$<10^\circ$	$<10^\circ$	$<5^\circ$	$<5^\circ$

D_1 inches	P inches	$1.05 D_1$ inches	$1.50 D_1$ inches
<u>.378</u>	<u>.276</u>	<u>.397</u>	<u>.567</u>
$0.185 \leq D_1 < 0.380$	.	.	.

γ degrees	ϕ degrees	$P_{\sin(\gamma)}$ inches	$P_{\sin(\phi)}$ inches
<u>2</u>	<u>1</u>	<u>.035</u>	<u>.017</u>
.	.	<0.125	<0.03125

P_1 inches	P_2 inches	$ P_1 - P_2 $ inches	Meet specifications
<u>.485</u>	<u>.491</u>	<u>.006</u>	<u>✓</u>
$1.05 D_1 < P_1 < 1.50 D_1$	$1.05 D_1 < P_2 < 1.50 D_1$	≤ 0.010	

Lower line in each table is limits for meeting specifications.

Checked by Doug Galt Date 12/11/91

PITOT TUBE INSPECTION DATA SHEET

Pitot Tube No. F-223 Date 12/20/79 Inspector J. V. Long

α_1 degrees	α_2 degrees	β_1 degrees	β_2 degrees
<u>1°</u>	<u>3°</u>	<u>2°</u>	<u>4°</u>
$<10^\circ$	$<10^\circ$	$<5^\circ$	$<5^\circ$

D_1 inches	P inches	$1.05 D_1$ inches	$1.50 D_1$ inches
<u>.359</u>	<u>1.091</u>	<u>.376</u>	<u>.539</u>
$0.185 \leq P_1 < 0.380$	-	-	-

γ degrees	ϕ degrees	$P \sin(\gamma)$ inches	$P \sin(\phi)$ inches
<u>2°</u>	<u>1°</u>	<u>.035</u>	<u>.017</u>
-	-	<0.125	<0.03125

P_1 inches	P_2 inches	$ P_1 - P_2 $ inches	Meet specifications
<u>.520</u>	<u>.501</u>	<u>.019</u>	<u>/</u>
$1.05 D_1 < P_1 < 1.50 D_1$	$1.05 D_1 < P_2 < 1.50 D_1$	≤ 0.010	

Lower line in each table is limits for meeting specifications.

Checked by Don G. P. Date 12/20/79

Dry Gas Meter and Orifice Meter

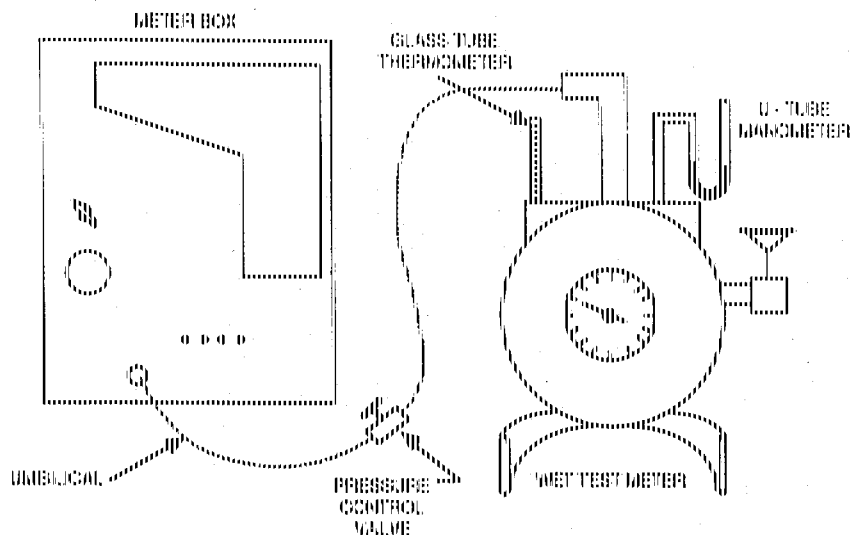
The following page shows the Calibration Setup used for the initial and post-test calibration. A wet-test meter with a 2-cubic-feet-per-minute capacity and ± 1 percent accuracy is used. The pump is run for approximately 15 minutes at an orifice manometer setting of 0.5 in. H₂O to heat up the pump and wet the interior surface of the wet-test meter. The information in the following example Calibration Data Sheet is gathered for the initial calibration; the ratio of accuracy of the wet-test meter to the dry-test meter and the $\Delta H @$ are then calculated.

Post-Test Meter Calibration Check

A post-test meter calibration check is made on each meter box used during the test to check its accuracy against the last calibration check. This post-test calibration must be within ± 5 percent of the initial calibration. The initial calibration is performed as described in APTD-C576. The post-test calibration is performed by the same method. Three calibration runs are made by using the average orifice setting obtained during each test run and setting the vacuum at the maximum value obtained during each test run. The post-test calibration check indicated that all three runs for each meter box were within the ± 5 percent range allowed by EPA Method 5.*

The Particulate Sampling Meter Box Initial Calibration and Post-Test Calibration data sheets are included in the following pages.

* 40 CFR 60, Appendix A, July 1990.



Title: E-2
Date: 3/23/92

Calibration setup.

DATE: _____ METER BOX NO. _____

BAROMETRIC PRESSURE, $P_b =$ _____ P_b Hg

DRY GAS METER NO. _____

ORIFICE MANOMETER SETTING ΔH in. H ₂ O	GAS VOLUME WET TEST METER V_w H ₂	GAS VOLUME DRY GAS METER V_d H ₂	WET TEST METER T_w °F	DRY GAS METER			TIME (s)	Y	d-qp
				INLET T_{in} °F	OUTLET T_{out} °F	AVERAGE T_{avg} °F			
0.5	5								
1.0	5								
1.5	10								
2.0	10								
3.0	10								
4.0	10								

AVERAGE _____

ΔH	$\frac{\Delta H}{13.6}$	Y $\frac{V_w P_b (T_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (T_w + 460)}$	d-qp	
			$\frac{0.0017 \Delta H}{P_b (T_d + 460)}$	$\frac{(T_w + 460) (Q)}{V_w}$
0.5	0.0368			
1.0	0.0737			
1.5	0.110			
2.0	0.147			
3.0	0.221			
4.0	0.294			

Y = Ratio of accuracy of wet test meter to dry test meter. Tolerance ± 0.01
d-qp = Coefficient of pressure differential that gives 0.75 cfm of air at 70° F and 20.02 inches of mercury (at P_b). Tolerance ± 0.15

Calibration data sheet.



METER BOX INITIAL CALIBRATION

Date: 11/19/91 Meter Box No.: FT-16
Barometric Pressure (ft): 29.68 Calibration: BUC

Run No.	Office			Wet test			Dry gas			Temperatures			Wet test			Dry gas meter			Average			Vacuum			Duration		
	meter			meter			meter			Inlet			Inlet			Outlet			Top			in. Hg			min. sec		
	Set	Vol	Wt	Set	Vol	Wt	Set	Vol	Wt	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F	°F
1	0.50	5.000	71	0.50	5.000	71	0.50	5.000	71	77	77	75	77	77	75	74	74	74	77.3	77.3	77.3	10	10	10	13	13	23
2	1.00	10.000	71	1.00	10.000	71	1.00	10.000	71	79	79	74	79	79	74	76	76	76	77.3	77.3	77.3	10	10	10	13	13	46
3	1.50	10.000	71	1.50	10.000	71	1.50	10.000	71	81	81	76	81	81	76	78	78	78	79.3	79.3	79.3	10	10	10	13	13	40
4	2.00	10.000	71	2.00	10.000	71	2.00	10.000	71	83	83	76	83	83	76	77	77	77	80.0	80.0	80.0	10	10	10	13	13	26
5	3.00	10.000	71	3.00	10.000	71	3.00	10.000	71	85	85	77	85	85	77	77	77	77	80.5	80.5	80.5	10	10	10	13	13	51
6	4.00	10.000	71	4.00	10.000	71	4.00	10.000	71	85	85	78	85	85	78	78	78	78	81.5	81.5	81.5	10	10	10	13	13	23

$$Y = \frac{(V_{\text{wet}} - V_{\text{dry}})(T_{\text{dry}} - 459)}{(V_{\text{dry}} - V_{\text{wet}})(T_{\text{wet}} - 459)}$$

$$\Delta H_2 = \frac{(0.0017 \Delta H_1) \times (P_{\text{bar}} - 14.7)}{(P_{\text{bar}} - 14.7)}$$

Run No.	Y	ΔH ₂
1	0.951	2.09
2	0.979	1.97
3	0.976	2.05
4	0.979	2.02
5	0.976	1.97
6	0.976	1.95

Pretest average	0.976	2.01
-----------------	-------	------

Difference = 0.004 0.12

* Y must not deviate by more than 10.0%
ΔH₂ must not deviate by more than 10.16 ΔH₁

Date checked by: *[Signature]*
Date: 11/19/91



WATER BOX POSTTEST SUMMARY

DATE: 9/27/02
TIME: 11:00 AM
LOCATION: 20102
TESTER: J. P. [illegible]
SPECIAL INSTRUCTIONS: [illegible]
PREPARED BY: [illegible]
CHECKED BY: [illegible]

Box	Box No.	Box Vol	Box Wt	Box Ht	Box Dia	Box Area	Box Perim	Box Vol	Box Wt	Box Ht	Box Dia	Box Area	Box Perim
1	201	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0
2	202	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0
3	203	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0	10.0

WATER BOX POSTTEST SUMMARY

DATE: 9/27/02
TIME: 11:00 AM
LOCATION: 20102
TESTER: J. P. [illegible]
SPECIAL INSTRUCTIONS: [illegible]
PREPARED BY: [illegible]
CHECKED BY: [illegible]

Box No. 201
Box Vol 10.0
Box Wt 10.0
Box Ht 10.0
Box Dia 10.0
Box Area 10.0
Box Perim 10.0

Box No. 202
Box Vol 10.0
Box Wt 10.0
Box Ht 10.0
Box Dia 10.0
Box Area 10.0
Box Perim 10.0

Box No. 203
Box Vol 10.0
Box Wt 10.0
Box Ht 10.0
Box Dia 10.0
Box Area 10.0
Box Perim 10.0

DATE: 9/27/02
TIME: 11:00 AM
LOCATION: 20102
TESTER: J. P. [illegible]
SPECIAL INSTRUCTIONS: [illegible]
PREPARED BY: [illegible]
CHECKED BY: [illegible]

Stack Thermocouples

Each thermocouple is calibrated by comparing it with an ASTM-3F thermometer at approximately 32 °F, ambient temperature, 100 °F, and 500 °F. The thermocouple read within 1.5 percent of the reference thermometer throughout the entire range when expressed in degrees Rankine. If the stack gas is saturated with moisture, the thermocouple is calibrated at 10 ° intervals between 70 ° and 180 °F using a water bath. The thermocouple agreed within ± 2 °F of the reference thermometer over the entire range. The thermocouples may be checked at ambient temperature at the test site to verify the calibration. Calibration data are included in the following Thermocouple Calibration Data Sheet(s).

THERMOCOUPLE CALIBRATION DATA SHEET

Date: 12-10-91 Thermocouple No: 101 3rd

Calibrator: K. MALLIKARJUN Reference: A.S.T.M.

Range: _____

Reference point no.	Source ¹	Reference thermometer temperature °F	Thermocouple temperature °F	Difference % ²
1	2	67	66	.19 %
2	1	35	35	0 %
3	3	213	213	0 %
4	4	460	459	.11 %

- ¹ Source: 1) Ice bath
2) Ambient
3) Water bath
4) Oil bath

² Percent difference.

$$\frac{\text{Reference Temp. } ^\circ\text{F} - \text{Thermocouple Temp. } ^\circ\text{F}}{(\text{Reference temp. } ^\circ\text{F})} \times 100\%$$

where $^\circ\text{F} = ^\circ\text{C} + 460$

Each percent difference must be less than or equal to 1.5%.

Checked by Donny C. Calkins Date 12/12/91

THERMOCOUPLE CALIBRATION DATA SHEET

Date: 12-11-91 Thermocouple No: 645-16^{FT}

Calibrator: J. Williams Reference: A.S.T.M.

Range: _____

Reference point no.	Source*	Reference thermometer temperature °F	Thermocouple temperature °F	Difference %**
1	2	71	71	0 %
2	1	36	36	0 %
3	3	212	212	0 %
4	4	448	448	0 %

- * Source: 1) Ice bath
2) Ambient
3) Water bath
4) Oil bath

** Percent difference.

$$\frac{\text{Reference temp. } ^\circ\text{F} - \text{Thermocouple temp. } ^\circ\text{F}}{(\text{Reference temp. } ^\circ\text{F})} \times 100\%$$

where $^{\circ}\text{R} = ^{\circ}\text{F} + 460$

Each percent difference must be less than or equal to 1.5%.

Checked by Gregory G. Bell Date 12/11/91

Digital Indicators for Thermocouple Readout

A digital indicator is calibrated by feeding a series of millivolt signals to the input and comparing the indicator reading with the reading the signal should have generated. Error did not exceed 0.5 percent when the temperatures were expressed in degrees Rankine. Calibration data are included in the following Thermocouple Digital Indicator Calibration Data Sheet(s).

DIGITAL INDICATOR CALIBRATION DATA SHEET

DATE 11/19/91 INDICATOR: ET-10

OPERATOR: B. J. Gosses

Test Point Number	Equivalent Temperature, °F T_e	Digital Indicator Temperature, °F T_{di}	Difference, % %
1	0	0	0
2	100	99	.18%
3	200	201	.15%
4	300	299	.13%
5	400	398	.23%
6	500	499	.10%
7	1000	999	.02%
8	1300	1296	.23%
9	1600	1525	.24%
10	1900	1873	.30%

*PERCENT DIFFERENCE MUST BE LESS THAN OR EQUAL TO 0.5%

$$\% \text{ DIFFERENCE} = \frac{(T_e \text{ °F} - T_{di} \text{ °F})(100)}{T_e \text{ °F}}$$

Where, °R = °F + 460

Checked By gm Date 11-19-91

Dry Gas Thermocouples and Impinger Thermocouples

The dry gas thermocouples are calibrated by comparing them with an ASTM-3F thermometer at approximately 32 °F, ambient temperature, and a higher temperature between approximately 100° and 200 °F. The thermocouples agreed within 5 °F of the reference thermometer. The impinger thermocouples are checked in a similar manner at approximately 32 °F and ambient temperature, and they agreed within 2 °F. The thermocouples may be checked at ambient temperature prior to the test series to verify calibration. Calibration data are included in the following Dry Gas Thermometer and Impinger Thermocouple Calibration Data Sheet(s).

DRY GAS THERMOCOUPLE CALIBRATION DATA SHEET

Date: 11/12/91 Thermocouple No: ET-10

Calibrator: BST Reference: 1574-3E

Inlet

Reference point no.	Source [*]	Reference thermometer temperature °F	Thermocouple temperature °F	Difference °F ^{**}
1	1	73	74	1
2	2	38	40	2
3	3	147	150	3

Outlet

Reference point no.	Source [*]	Reference thermometer temperature °F	Thermocouple temperature °F	Difference °F ^{**}
1	1	73	72	1
2	2	38	38	0
3	3	147	148	1

* Source: 1) Ambient
2) Ice bath
3) Water bath

** Difference must be less than 5°F at both points.

Checked by Jm

Date 12/3/91

IMPINGER THERMOCOUPLE CALIBRATION DATA SHEET

Date: 12/14/91 Thermocouple No: 3-4
 Calibrator: M. L. Latta Reference: ASTM

Reference point no.	Source [*]	Reference thermometer temperature °F	Thermocouple temperature °F	Difference °F ^{***}
1	1	68	68	0
2	2	34	34	0

* Sources: 1) Ambient
 2) Ice bath

*** Difference must be less than 2°F at both points.

Checked by Doug Latta Date 12/13/91

Trip Balance

The trip balance is calibrated by comparing it with Class-S standard weights, and it agreed within 0.5 g. Calibration data are shown in the following Trip Balance Calibration Data Sheet(s).

TRIP BALANCE CALIBRATION DATA SHEET

Balance No.	Date	Calibrator	Mass determined for					
			5 g	Error	50 g	Error	100 g	Error
422	12/10/14	M. Kett	56	0	506	0	1006	0
421	12/10/14	M. Kett	56	0	506	0	1006	0
420	12/10/14	M. Kett	56	.02	506	.03	1006	.05
418	12/10/14	M. Kett	56	.2	506	.3	1006	.2
199	12/10/14	M. Kett	56	.3	506	.2	1006	0
198	12/10/14	M. Kett	56	.1	506	.1	1006	0

Error must not exceed 0.5 grams at each point.

Checked by M. Kett Date 12/10/14

Barometer

The field barometer is calibrated to within 0.1 in.Hg of an NBS-traceable mercury-in-glass barometer before the test series. It is checked against the reference barometer after each test series to determine if it reads within 0.2 in.Hg. The barometer read within the allowable limits each time. Calibration data are included in the following Barometer Calibration Log(s).

BAROMETER CALIBRATION LOG

PRETEST

BAROMETER NO.	415	412	401	431	460	408	444
Client	Jeffrey's	Alcoro	Alcoro	Alcoro	Alcoro	Alcoro	Alcoro
Project No.		332220	332220	332220	332220	332220	332220
BAROMETER READING	29.58	29.66	29.66	29.60	29.32	29.54	29.47
REFERENCE BAROMETER READING	29.58	29.67	29.67	29.59	29.37	29.53	29.47
DIFFERENCE*	0.00	0.02	0.01	.01	.05	.01	.04
DATE	7-13-92	7-20-92	7-20-92	7-24-92	7-27-92	7-28-92	7-31-92
CALIBRATOR	DL	DL	DL	DL	DL	DL	DL

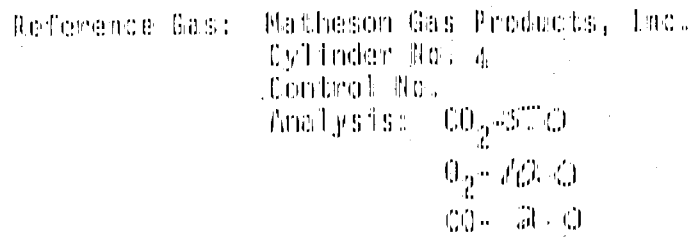
POST-TEST

BAROMETER READING	29.66			29.36	29.45	29.50	
REFERENCE BAROMETER READING	29.67			29.39	29.50	29.57	
DIFFERENCE**	.01			0.03	.05	.07	
DATE	7-28-92			8-1-92	8-10-92	8-19-92	
CALIBRATOR	J. Nease			DL	DL	DL	

*Barometer is adjusted so that difference does not exceed 0.05 in. Hg.
 **Barometer is not adjusted. If difference exceeds 0.10 in. Hg, inform project manager immediately.

Orsat Analyzer

The Orsat analyzer is calibrated before the test series by determining the percentages of carbon dioxide and oxygen in a calibration gas containing known percentages of each. The analyzer read within 0.5 percent of the known values. Calibration data are shown in the following Orsat Calibration Data Sheets.



Orsat No.: 423 Gas: CO₂
4.5 5.0 5.5

E.25

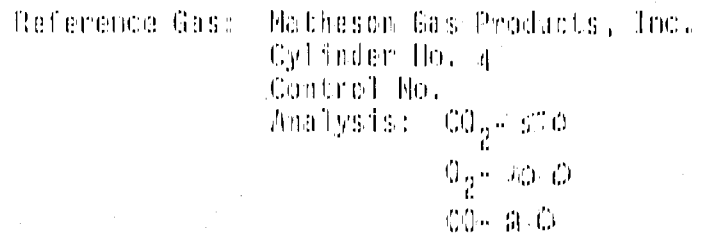


Reference Gas: Matheson Gas Products, Inc.
Cylinder No. 4
Control No.
Analysis: CO_2 5% O
 O_2 10.0
 CO 2.0

ORSAT CALIBRATION DATA SHEET

Orsat No.: 923 Gas: O_2

Calibrator	Date	PN	Value determined	2.5	10.0	10.5
<i>T. G. Hoff</i>	<i>5/1/92</i>	<i>3000</i>	<i>9.8</i>			
<i>A. K. Hoff</i>	<i>5/1/92</i>		<i>9.7</i>			
<i>D. C. Hoff</i>	<i>6/1/92</i>	<i>3000</i>	<i>9.8</i>			
<i>T. N. Hoff</i>	<i>6/1/92</i>	<i>3000</i>	<i>9.8</i>			
<i>J. H. Hoff</i>	<i>6/1/92</i>	<i>3000</i>	<i>9.8</i>			
<i>G. H. Hoff</i>	<i>6/1/92</i>	<i>3000</i>	<i>9.8</i>			
<i>D. C. Hoff</i>	<i>7/2/92</i>	<i>3000</i>	<i>9.8</i>			
<i>J. H. Hoff</i>	<i>8/1/92</i>	<i>3000</i>	<i>9.9</i>			
<i>D. C. Hoff</i>	<i>8/1/92</i>	<i>3000</i>	<i>9.9</i>			
<i>D. C. Hoff</i>	<i>8/2/92</i>	<i>3000</i>	<i>9.9</i>			

Orsat No.: 423 Gas: CO

