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# Research and Development

## COMBUSTION MODIFICATION CONTROLS FOR STATIONARY GAS TURBINE Volume II. Utility Unit Field Test

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COMBUSTION MODIFICATION CONTROLS FOR STATIONARY GAS TURBINES  
VOLUME II. UTILITY UNIT FIELD TEST

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## ABSTRACT

This test report describes the methods and results of an environmental assessment test program conducted at Houston Lighting and Power's T. H. Wharton Generating Station, Unit 52. The purpose of the test program was to measure changes in emissions as a result of applying  $\text{NO}_x$  controls. Emissions of trace elements, organic materials, sulfur species, and the criteria pollutants,  $\text{SO}_2$ ,  $\text{NO}_x$ , CO, and particulate matter, were measured. Comparisons of these emissions under normal operating conditions and controlled (for  $\text{NO}_x$ ) operating conditions were then made. Source operating data were also analyzed so that changes in operating parameters and efficiency could be assessed.

Unit 52 is a General Electric MS 7001C simple-cycle, single-shaft, heavy duty gas turbine rated at 70.8 MW nominal electrical output. This gas turbine may use either natural gas or distillate oil fuels. The test program was conducted using oil fuel.

Water injection was used for  $\text{NO}_x$  control. A water-to-fuel ratio of 0.42 resulted in a 58 percent reduction in  $\text{NO}_x$  from baseline levels. Changes in other emissions were within the limits of the analyses.

Operating efficiency decreased with water injection. The unit heat rate showed approximately 2 percent change in going from baseline to controlled (for  $\text{NO}_x$ ) operation.

The test program concludes that using water injection for  $\text{NO}_x$  control in this unit reduced  $\text{NO}_x$  and showed little effect on other emissions. Water injection implementation did reduce operating efficiency.

## ACKNOWLEDGEMENT

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## SECTION 1

### INTRODUCTION

This report is part of a series of test reports resulting from the experimental testing task of the "Environmental Assessment of Stationary Source NO<sub>x</sub> Control Technologies" Program (NO<sub>x</sub> EA), being performed under Environmental Protection Agency (EPA) contract 68-02-2160. The NO<sub>x</sub> EA is a 3-year program to: (1) identify the multimedia environmental impact of stationary combustion sources and combustion modification NO<sub>x</sub> controls; and (2) identify the most cost-effective environmentally-sound NO<sub>x</sub> controls for attaining and maintaining current and projected NO<sub>2</sub> air quality standards to the year 2000.

During the first year of the NO<sub>x</sub> EA a preliminary environmental assessment (Reference 1) concluded that emissions and operating data needed to perform adequate process engineering and environmental assessment activities were severely lacking in several key areas. Most noteworthy was the virtual absence of data on noncriteria flue gas emissions and liquid and solid effluents. In response to these identified data needs, seven field test programs were initiated. Source selection was based on a source/control priority listing developed in the preliminary environmental assessment. These test programs were designed to provide information on changes in emissions and operation due to NO<sub>x</sub> controls. The NO<sub>x</sub> EA Field Test Program is outlined in Table 1-1.

The test program documented in this report was conducted on Unit 52 of the T. H. Wharton Generating Station of the Houston Lighting and Power Company in Houston, Texas from April 21-24, 1978. Unit 52 was selected because its design is typical of large scale simple cycle utility gas turbines equipped with water injection and because of the possibility of collaborating with the engine manufacturer in detailed process evaluation tests. Unit 52 is a General Electric Model MS 7001C simple-cycle, single-shaft, heavy duty gas turbine rated at 70.8 MW nominal electrical

TABLE 1-1. NO<sub>x</sub> EA FIELD TEST PROGRAM

Source Category	Description	Test Points (Unit Operation)	Sampling Protocol	Test Collaborator	Status
Coal-fired Utility Boiler	Kingston #6; 180 MW tangentia; twin furnace, 12 burners/ furnace, 3 elevations; cyclone, 2 ESP's for particulate control	Baseline Biased Firing (2) B005 (2)	Continuous NO <sub>x</sub> , SO <sub>2</sub> , CO, CO <sub>2</sub> , O <sub>2</sub> Inlet to 1st ESP: -- SASS -- Method 5 -- Method 8 -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) Outlet of 1st ESP: -- SASS -- Method 5 -- Method 8 -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) Bottom ash Hopper ash (1st ESP, cyclone) Fuel Operating data	TVA	Complete, August 1977
Coal-fired Utility Boiler	Crist #7, 500 MW opposed wall fired; 24 burners, 3 elevations; ESP for particulate control	Baseline B005 (2)	Continuous NO <sub>x</sub> , CO CO <sub>2</sub> , O <sub>2</sub> ESP inlet -- SASS -- Method 5 -- Method 8 -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) ESP outlet -- SASS -- Method 5 -- Method 8 -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) Bottom ash ESP hopper ash Fuel Operating data Bioassay	Exxon	Complete, June 1978
Oil-fired Utility Boiler	Moss Landing #6, 740 MW opposed wall fired; 48 burners, 6 elevations	Baseline FGR FGR + OFA	Continuous NO <sub>x</sub> , CO, CO <sub>2</sub> , O <sub>2</sub> Flue gas -- SASS -- Method 5 -- Method 8 -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) Fuel Operating data Bioassay	New test start	Complete, September 1978

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TABLE 1-1. Continued

Source Category	Description	Test Points (Unit Operation)	Sampling Protocol	Test Collaborator	Status
Coal-fired Industrial Boiler	Traveling grate spreader stoker, 38 kg/s (300,000 lb/hr); ESP for particulate control; wet scrubber for SO <sub>2</sub> control	Baseline LEA + high OFA	Continuous NO <sub>x</sub> , CO, CO <sub>2</sub> , O <sub>2</sub> Boiler exit: -- SASS -- Method 5 -- Shell-Emerysville -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) ESP outlet: -- SASS -- Method 5 -- Shell-Emerysville -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) Bottom ash Cyclone hopper ash Fuel Operating data	KVB	Complete, October 1977
Coal-fired Industrial Boiler	Traveling grate spreader stoker, 25 kg/s (200,000 lb/hr) ESP for particulate	Baseline LEA + High OFA	Continuous NO <sub>x</sub> , CO, CO <sub>2</sub> , O <sub>2</sub> Boiler exit: -- SASS -- Method 5 -- Shell-Emerysville -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) ESP Outlet: -- SASS -- Method 5 -- Shell-Emerysville -- Gas grab (C <sub>1</sub> -C <sub>6</sub> HC) Bottom ash ESP hopper ash Fuel Operating data Bioassay	KVB	Complete, February 1978
Oil-fired Gas Turbine	T.H. Wharton Station, 60 MW GE MS 7001 C machine	Baseline water injection to meet proposed MSPS	Continuous NO <sub>x</sub> , CO, CO <sub>2</sub> , O <sub>2</sub> Exhaust gas: -- SASS -- Method 5 -- Method 8 Fuel Water Operating data	General Electric	Complete, April 1978

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TABLE 1-1. Concluded

Source Category	Description	Test Points (Unit Operation)	Sampling Protocol	Test Collaborator	Status
Oil-fired Residential Heating Unit	Blue Ray low NO <sub>x</sub> furnace, Medford, New York	Continuous Cycling	Continuous NO <sub>x</sub> , CO, CO <sub>2</sub> , O <sub>2</sub> Flue gas; -- SAS -- Method 5 -- Method 8 Fuel	New test start with EPA/IERL-RTP	Complete, November 1977

EE-073

output. Through the cooperation of the Houston Lighting and Power Company and the General Electric Company, this unit was made available for testing in the NO<sub>x</sub> EA Field Test Program.

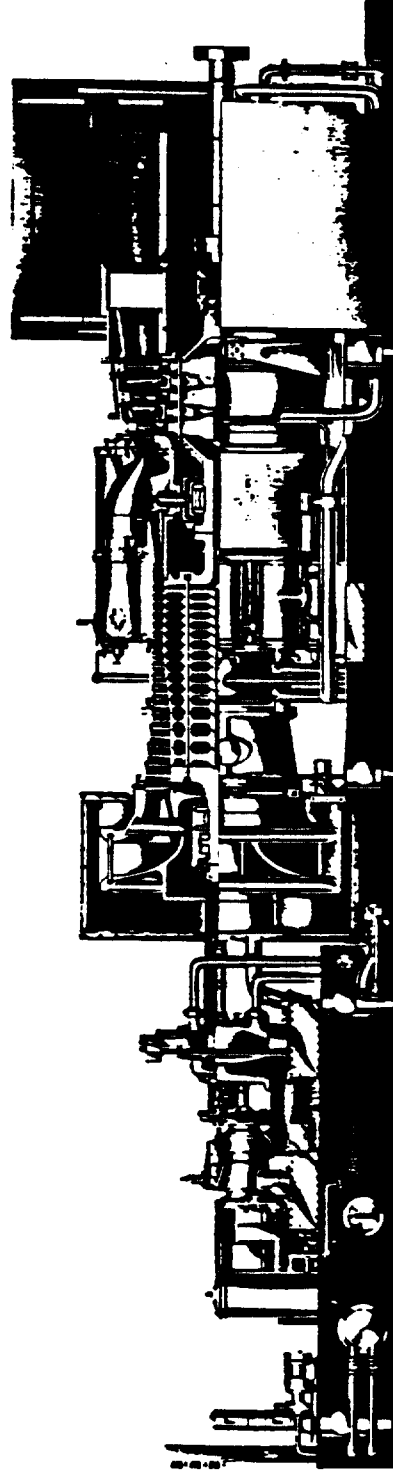
The test program at the T. H. Wharton Station consisted of a baseline (normal operation) test and a test with water injection being used for NO<sub>x</sub> control. The test program results will be used in both process analysis and source assessment modeling, conducted as part of the Environmental Assessment and Process Engineering Task of the NO<sub>x</sub> EA.

## SECTION 2

### PLANT DESCRIPTION

The field tests were conducted on Unit 52 of the T. H. Wharton Generating Station of the Houston Lighting and Power Company in Houston, Texas. Unit 52 is a General Electric Model MS 7001C (Figure 2-1) simple-cycle, single-shaft, heavy-duty stationary gas turbine rated at 70.8 MW nominal electrical output and is one of six such units at the Wharton Station. The Station also has eight GE 7001B combined cycle units, one Westinghouse 15 MW unit and two conventional steam boilers producing a total rated electrical output of 280 MW.

Unit 52 is fired with No. 2 distillate fuel oil with 0.11 percent sulfur by weight and approximately  $46.054 \times 10^3$  kJ/kg (19,800 Btu/lb) heat content. Table 2-1 lists the rated operating parameters of the unit. There is no flue gas cleaning equipment on a turbine of this type due to the clean fuel used and the unit's inherent efficient combustion. Unit 52 is, however, equipped with a water injection system used to control the formation of  $\text{NO}_x$  within the combustion chambers.  $\text{NO}_x$  formation is repressed when atomized water is injected directly into the primary zone of the combustor resulting in reduced flame temperatures. The degree of  $\text{NO}_x$  control is adjusted by altering the quantity of water injected -- the more water injected the greater the degree of control. The first test on Unit 52, a baseline test, was run with no water injected. The second test was run while 2.52 l/sec (40 gpm) water was being injected. This corresponds to a water to fuel mass ratio of approximately 0.42, a ratio sufficiently high to bring  $\text{NO}_x$  emissions to within 75 ppm at 15 percent  $\text{O}_2$  which is the level of the proposed New Source Performance Standards.



TC-7509A

Figure 2-1. Model series 7001 simple-cycle, single-shaft heavy-duty gas turbine.

TABLE 2-1. UNIT 52 RATED OPERATING PARAMETERS

Output power	70.8 MW
Overall pressure ratio	10.5
Heat rate	11.44 MJ/kWh (10,847 Btu/kWh)
Air flow	268 kg/sec (592 lb/sec)
Fuel flow	5.2 kg/sec (11.5 lb/sec)

### SECTION 3

#### SAMPLING AND ANALYSIS METHODOLOGIES

The sampling and analysis procedures used in the test program closely follow the procedures recommended in the IERL-RTP Level 1 Environmental Assessment Procedures Manual (Reference 2). The following subsections will contain notations of where the procedures differ significantly from the standard methods. Level 1 testing, according to EPA's phased environmental assessment approach, is for screening purposes. Through chemical and biological tests potential problem areas and needs for further analysis are identified. Furthermore, Level 1 testing provides the basis for setting priorities for discharge streams, components, and classes of materials for further consideration in an overall environmental assessment. Thus, the results of the sampling and analysis procedures used in Level 1 are semiquantitative, yielding an accuracy factor of  $\pm 2$  to 3.

All analyses for trace elements, organic species, particulates and sulfur species in the Method 5/8 and SASS trains and water samples were performed in the Acurex analytical laboratory. Commercial Testing and Engineering Company analyzed the fuels and the bioassay analyses were performed by Litton Bionetics, Incorporated.

#### 3.1 SAMPLING PROTOCOL

In order to effectively evaluate how emissions of compounds and pollutant species are affected by the use of water injection, all influent and effluent streams must be characterized during the baseline and water injection tests. The following streams on Unit 52 were sampled:

- Water feed (water injection system)
- Fuel feed
- Exhaust gas

Ambient air was not sampled. Descriptions of the specific sampling methods are given in the following paragraphs. Figure 3-1 shows the duct configuration and the location of the sampling ports.

### 3.1.1 Feed Streams

#### Water Feed

Samples of the demineralized feed water, from the water injection purification system were periodically sampled throughout the five hour duration of the  $\text{NO}_x$  control test. Samples were tapped off the inlet lines preceeding the combustor section and then composited into one integrated sample for each run.

#### Fuel Feed

Fuel oil samples were obtained for both tests. Samples were tapped off the fuel inlet lines, collected throughout the test period and finally composited into one integrated sample for each test. Sampling of the fuel feed commenced one hour into the test run, then approximately once for each 90 minute period throughout the test.

### 3.1.2 Flue Gas

The flue gas was monitored on a continuous basis during both test runs for  $\text{O}_2$ ,  $\text{CO}_2$ ,  $\text{NO}$ , total  $\text{NO}_x$ , total unburned hydrocarbons and  $\text{CO}$ . The continuous monitoring was provided by General Electric (GE) personnel and equipment. Table 3-1 lists the instrumentation used by General Electric. All sample lines were of Teflon construction and heated to 450 K ( $350^\circ\text{F}$ ) to assure the integrity of all sampled species. The sample flow was filtered to remove particulate matter and then split into two streams. One stream supplied the nitrogen oxides instrument and the total hydrocarbon monitor, while the other supplied the nondispersive infrared (NDIR) instruments and the paramagnetic oxygen analyzer. The latter stream was further conditioned in a saturator and refrigerated dryer before connecting to the  $\text{CO}_2$ ,  $\text{CO}$  and  $\text{O}_2$  analyzers. All monitors were frequently zeroed and calibrated with certified gases.

All continuous gaseous sampling was done through a single point probe located in the center of the exhaust duct approximately 1m (40 inches) upstream of the main row of sampling ports used for the SASS and Method 5/8 sampling.

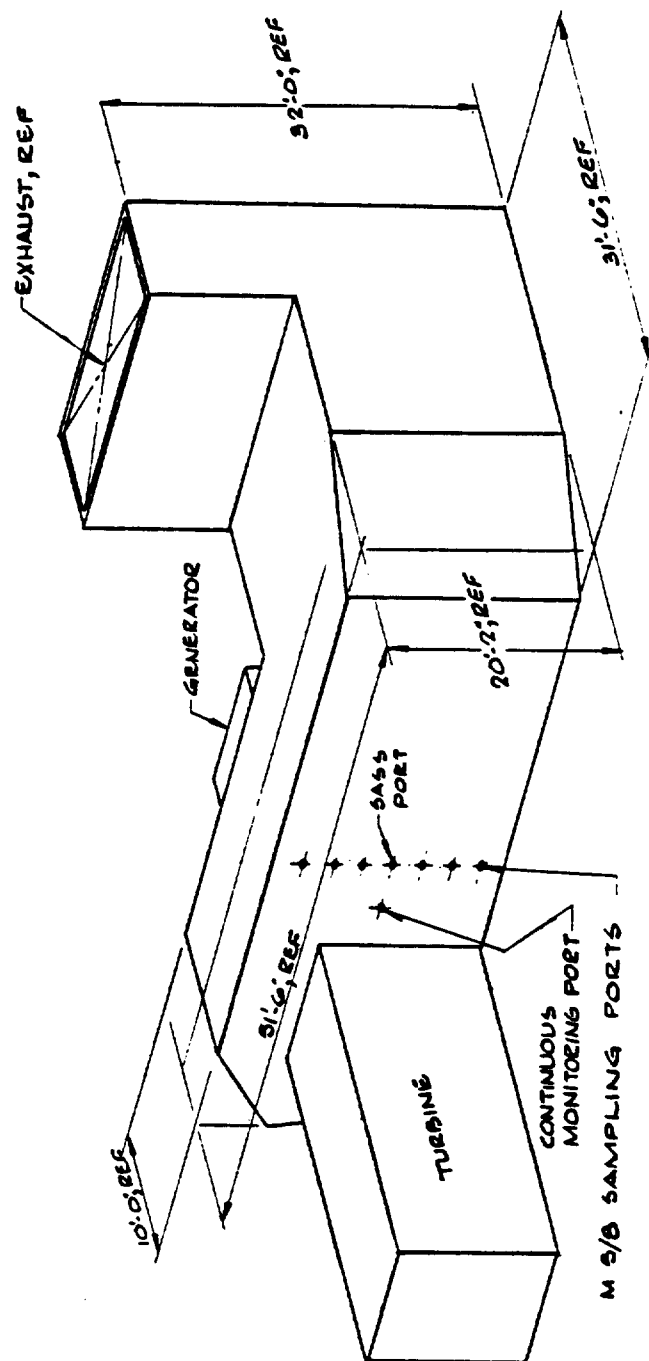


Figure 3-1. Exhaust duct configuration and sampling location.

TABLE 3-1. INSTRUMENTATION USED BY GENERAL ELECTRIC

Instrument	Technique	Measuring Range
Unburned hydrocarbons: Beckman Model 402	Flame ionization detector	5 ppm - full scale to 25% - full scale
NO and NO <sub>2</sub> : Beckman 955	Chemiluminescence	0 - 10 ppm 0 - 1,000 ppm 0 - 25 ppm 0 - 2,500 ppm 0 - 100 ppm 0 - 10,000 ppm 0 - 250 ppm
O <sub>2</sub> : Beckman Model F3	Paramagnetic	0 - 15% 13% - 18% 16% - 21% 0 - 25%
CO: Beckman 315B	Nondispersive infrared	0 - 50 ppm 0 - 200 ppm 0 - 500 ppm
CO <sub>2</sub> : Beckman 364	Nondispersive infrared	0 - 5% 0 - 10% 0 - 15%

### Particulate and Sulfur Species

Particulate and sulfur species in the flue gas were collected simultaneously with one sampling train -- a combined EPA Method 5 and Method 8 train. Such a system collects particulate samples on a filter heated to 394 K (250°F) in a conventional Method 5 arrangement. But rather than the conventional Method 5 water filled impinger train, the modified train employs a Method 8 impinger train containing isopropanol to remove SO<sub>3</sub> and hydrogen peroxide to remove SO<sub>2</sub>. Particulate sulfate (SO<sub>4</sub>) is also collected with this system. One run was completed for each test.

### C<sub>1</sub>-C<sub>6</sub> Hydrocarbons

Flue gas grab samples were collected in evacuated glass grab flasks. These samples were chromatographed onsite to determine C<sub>1</sub> to C<sub>6</sub> hydrocarbon compounds. A Carle Model 8500 portable gas chromatograph with a flame ionization detector was used for this analysis.

### Source Assessment Sampling System

A Source Assessment Sampling System (SASS) train was used to sample the gas turbine exhaust gas. The SASS train was conventional in every way except that cyclones were not used to classify the particulate by size. A single fiberglass mat filter was used to collect the small amounts of particulate produced. A special oil cooled probe was used to maintain the sample tube temperature at 394 K (250°F). This SASS train arrangement generates the following samples:

Particulate: filter 99.99 percent efficient for particulate greater than 0.2  $\mu$ m.

Vapor phase: 1) XAD-2 porous polymer resin sorbent cartridge  
2) Aqueous condensate  
3) Hydrogen peroxide impinger  
4) Ammonium persulfate-silver nitrate impingers

These samples were analyzed for trace elements and organic species to give both vapor and condensed phase composition.

A schematic of the SASS train is shown in Figure 3-2. The analysis protocol is given in Figure 3-3.

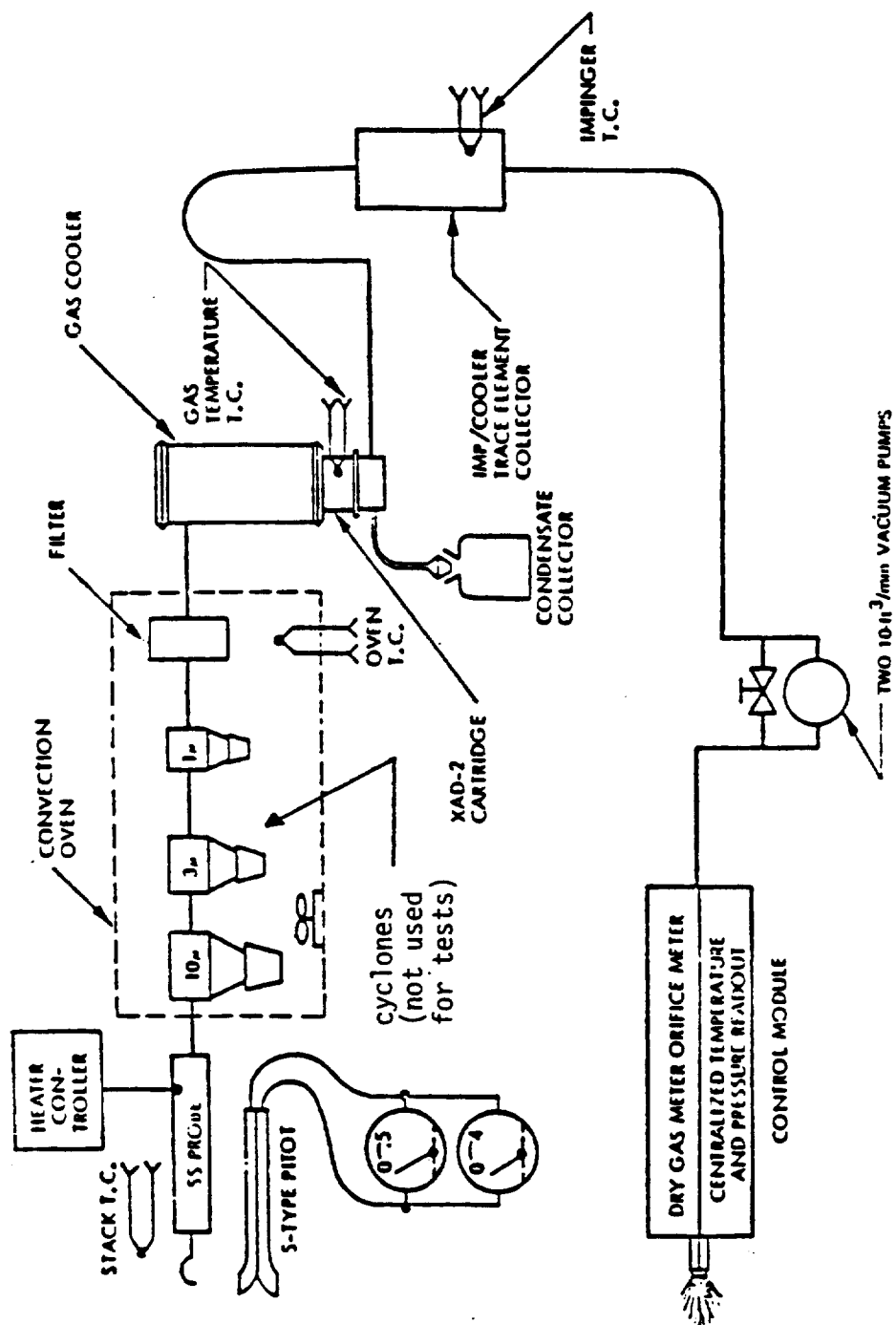


Figure 3-2. Source Assessment Sampling System (SASS) schematic.



### 3.2 ANALYSIS PROTOCOL

Table 3-2 lists the analyses performed on the samples collected during both tests. Descriptions of these analyses are given in the following paragraphs.

In addition, machine operating data were collected by General Electric personnel. Sufficient data were taken during both tests so that airflow rates and the operating condition of the machine could be established. A detailed field test report submitted by General Electric is contained in Appendix G. In summary, GE treats operating data, information on gaseous emissions, fuel composition, machine geometry and internal flow splits, using a data analysis program which calculates machine operational characteristics. Table 3-3 illustrates the kind of information that can be obtained. This program proved an excellent means of crosschecking actual flue gas measurements as well as checking calculation methods against each other.

#### 3.2.1 Inorganic Analysis

Trace element analyses for 23 selected trace elements were performed on the fuel, injected water, flyash, SASS XAD-2 and SASS impinger solutions. The procedure used to determine each trace element is outlined in Appendix H. Proximate and ultimate analyses were done on the fuel samples.

#### 3.2.2 Organic Analysis

Organic analyses were performed in accordance with EPA Level 1 protocol (Reference 2). These analyses included  $C_1$ - $C_6$  hydrocarbons in the flue gas, organic material condensed on the ash samples, and organic material caught in the XAD-2 sorbent trap and condensate trap.

#### 3.2.3 Bioassay

Bioassays were performed on the SASS train XAD-2 extract sample from the water injection test. Microbial mutagenesis and cytotoxicity assays were performed by Litton Bionetics, Incorporated.

TABLE 3-2. SAMPLE ANALYSIS

Test/Analysis	Baseline no water injection	Low NO <sub>x</sub> with water injection
Fuel		
Proximate and ultimate	X	X
Trace elements	X	X
Water		
Trace elements		X
Organic material		X
SASS train - outlet		
Trace elements	X	X
Organic material	X	X
Method 5/8 -- outlet		
Particulate	X	X
Sulfur species	X	X
Flue gas		
O <sub>2</sub> , CO <sub>2</sub> , NO <sub>x</sub> , NO, CO	X	X
and total unburned hydrocarbons	X	X
C <sub>1</sub> -C <sub>6</sub> hydrocarbons	X	X

TABLE 3-3. GENERAL ELECTRIC CALCULATED OPERATING DATA<sup>a</sup>

Calculation Method	Measured	Calculated
Factory test flow	Fuel flow and composition, airflow during factory test, inlet guide vane position, ambient conditions	Machine airflow, O <sub>2</sub> , CO <sub>2</sub> , water in exhaust, turbine inlet temperature.
Choked flow	Compressor discharge pressure and temperature, first stage nozzle area, fuel flow and composition.	Machine airflow, O <sub>2</sub> , CO <sub>2</sub> , H <sub>2</sub> O, turbine inlet temperature.
Oxygen concentration	O <sub>2</sub> , fuel flow and composition	Machine airflow, CO <sub>2</sub> , H <sub>2</sub> O, turbine inlet temperature.
CO <sub>2</sub> concentration	CO <sub>2</sub> , fuel flow and composition	Machine airflow, O <sub>2</sub> , H <sub>2</sub> O, turbine inlet temperature.

<sup>a</sup>See Appendix G

## SECTION 4

### TEST PROGRAM RESULTS

Data from the test program provided information on unit operation, effluent gaseous composition, particulate emissions, trace element emissions, sulfur species emissions and organic material emissions.

#### 4.1 UNIT OPERATION

Unit 52 operated under steady-state conditions at rated continuous load and with operating parameters nominally the same for both the baseline (no water injection) and the NO<sub>x</sub> control (with water injection) test. The load for each test was approximately 62 MW electrical generator output. Table 4-1 lists the process operating conditions and parameters during each test. The only significant difference between Test 1 and Test 2 is that Test 2 had water injection while Test 1 did not. Since it is fairly easy to duplicate engine operating conditions in a gas turbine, one can be reasonably confident in comparing emissions from tests where only the one variable, water injection rate, was changed.

The operating variable readings were recorded on an hourly basis throughout the tests. The results shown in Table 4-1 are an average of those values. The actual data sheets can be found in Appendix G.

One of the most significant penalties resulting from the use of water injection for NO<sub>x</sub> control is the reduction in unit thermal efficiency or increased heat rate manifested as increased fuel consumption. As indicated in Table 4-1, the unit heat rate increased 2.4 percent with water injection at a water/fuel ratio equal to 0.42. This is because a portion of the fuel is required to vaporize the injected water. These effects on heat rate and fuel consumption are quite typical (Reference 3). Most users have reported heat rate penalties ranging from 2 to 5 percent, depending on the water to fuel ratio.

TABLE 4-1. UNIT 52 -- OPERATING CONDITIONS

	<u>Baseline</u>	<u>Injection</u>
Ambient barometric pressure - mm Hg (in. Hg)	755 (29.74)	756 (29.79)
Ambient temperature -- dry bulb -- K (°F)	295 (71.2)	301 (82.0)
Relative humidity	83.6	58.7
Compressor discharge pressure P <sub>CD</sub> -- kpa (psia)	915 (132.7)	901 (130.7)
Compressor discharge temperature T <sub>CD</sub> -- K (°F)	593 (607)	602 (624)
Speed (rpm)	3600	3600
Inlet guide vane angle (IGV degrees)	77	77
Load (MW)	61.9	61.5
Turbine exhaust temperature -- K (°F)	809 (997)	813 (1000)
Water injection rate -- liters/sec and (gpm)	0	2.52 (40)
Water/fuel ratio	0	0.42
Fuel temperature -- K (°F)	295.4 (71.7)	298 (76.4)
Fuel flow -- liters/sec and (gpm)	5.93 (94.0)	6.03 (95.6)
Atomizing air pressure -- kpa (psia)	1372 (199)	1372 (199)
Atomizing air temperature -- K (°F)	473 (392)	471 (387)
Combustion efficiency (%)	99.9	99.9
Exhaust flow -- m <sup>3</sup> /s (10 <sup>6</sup> SCFH)	205.4 (26.14)	200.9 (25.52)
Compressor inlet flow -- kg/s (lbm/sec)	253 (556.7)	255.7 (562.7)
Fuel/air ratio	0.0190	0.0196
Heat rate -- MJ/kWh (Btu/Kwh - based on LHV)	12.55 (11,892)	12.84 (12,173)

As noted in Section 3.2, GE personnel recorded operating data and monitored unit operation throughout the test program. In addition GE also evaluated recorded data using an in-house data analysis code. This program can be used to calculate inlet airflow and exhaust gas flow (in addition to other parameters -- see Table 3-3) using gaseous emissions data and other operating information. Four different calculational modes are possible, as outlined in Table 3-3. Calculated exhaust gas flowrates for each test, using the program, are listed in Table 4-2 for each of the calculation methods. Agreement among the methods is excellent (within one percent). Also shown in Table 4-2 are measured exhaust gas flowrates obtained by performing an EPA Method 5 velocity traverse across the exhaust duct. As indicated, measured rates are approximately 55 percent greater than calculated rates. This was not unexpected, though. The exhaust duct configuration was such that gas flow obstructions (i.e., bends) were very close to the sampling location, thus accurate velocity measurements were very difficult to obtain. In an attempt to equalize the effects of a poor sampling location, 42 sampling points were sampled. Nevertheless, measured gas flowrates were still unreasonably high due to the highly variable velocity readings.

Thus all exhaust flowrate values reported herein, including those noted in Table 4-1, are calculated values, averaged over the four possible calculational methods.

#### 4.2 FUEL ANALYSIS

Duplicate proximate and ultimate fuel analyses were performed by General Electric and Commercial Testing and Engineering (CT&E). General Electric's results are reported in Appendix G. CT&E's analysis is reported in Appendix A. Results from both analyses were very similar and typical of distillate fuel oil. In addition, a trace element analysis of the fuel oil was performed as part of the mass balance and reported in Concentration and mass flowrate units in Appendices B-E.

#### 4.3 EXHAUST GAS EMISSIONS

Exhaust emissions were tested for gaseous species, particulate emissions, sulfur species, trace elements and organic material emissions. Gaseous species were measured by General Electric personnel on a continuous basis throughout both tests. A combined EPA Method 5/8 train

TABLE 4-2. CALCULATED AND MEASURED EXHAUST GAS FLOWRATES --  $\text{m}^3/\text{s}$  ( $10^6$  SCFH)

Test No.	GE -- Calculated Values <sup>a</sup>				Average	Measured
	Factory Test Flow	Choked Flow	Oxygen Concentration	CO <sub>2</sub> Concentration		
1	203.8 (25.9244)	205.3 (26.1206)	207.2 (26.3644)	205.8 (26.1828)	205.4 (26.1481)	316.4 (40.2632)
2	199.3 (25.3136)	201.0 (25.5280)	202.5 (25.7149)	220.5 (27.9991)	200.9 (25.5188b)	311.4 (39.6247)

<sup>a</sup>See Appendix G for explanation of calculations

<sup>b</sup>Exhaust gas flowrate calculated by the CO<sub>2</sub> concentration scheme not included in average. CO<sub>2</sub> values believed to be affected by moisture in flue gas.

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was used to simultaneously sample particulates and sulfur species. A Source Assessment Sampling System (SASS) was used to collect samples for analysis of trace elements and organic material. This section presents the results of these analyses.

#### 4.3.1 Gaseous Emissions

Total  $\text{NO}_x$ ,  $\text{NO}$ ,  $\text{O}_2$ ,  $\text{CO}_2$ ,  $\text{CO}$  and total unburned hydrocarbons (UHC) were measured at a single point in the exhaust duct. Supporting tests conducted by General Electric, reported in Appendix G, have concluded that emissions of  $\text{NO}_x$  and  $\text{O}_2$  can be reliably and accurately measured from a single sampling point. However, species that are present only in very low concentrations, UHC for example (ppmv  $<2$ ), require a traverse of the duct when sampling.

Table 4-3 presents gaseous emissions data in a form summarized from the General Electric report in Appendix G. With regard to the proposed New Source Performance Standards (NSPS) for stationary gas turbines, there are two things of importance to note from this information. First, with water injection operating at a water/fuel weight ratio of 0.42,  $\text{NO}_x$  emissions were reduced by 58 percent from the baseline levels -- from 177.5 to 74.2 ppm at 15 percent  $\text{O}_2$  dry. This controlled level is within the NSPS proposed level of 75 ppm. The second item to note is that  $\text{SO}_2$  emissions are substantially below the proposed NSPS level of 150 ppmv at 15 percent  $\text{O}_2$ . The  $\text{SO}_2$  values for Unit 52 were calculated directly from the fuel sulfur content assuming 100 percent conversion. The calculated value for  $\text{SO}_2$  concentration, which assumes all fuel sulfur is converted to  $\text{SO}_2$ , is reasonably close to the measured total  $\text{SO}_x$  emission concentration (within 30 percent) as determined by the Method 8 analysis.

The results of the sulfur species analysis are shown in Table 4-4. The data show that the actual emission levels of sulfur species, as well as the  $\text{SO}_2/\text{SO}_3$  ratio, are not significantly affected by the use of water injection for  $\text{NO}_x$  control. Table 4-5 shows the results of a sulfur balance across the gas turbine. The quantity of sulfur recovered in the flue gas was approximately 70 percent of the inlet sulfur. Duplicate fuel oil sulfur analyses gave a sulfur content of approximately 0.11 percent, so inlet sulfur calculations should be correct. Consequently, the source of the inconsistency probably lies in the Method 8 sampling train and subsequent analysis.

TABLE 4-3. GASEOUS EMISSIONS RESULTS -- ppmv at 15 percent O<sub>2</sub> dry

	<u>Baseline</u>	<u>Water Injection</u>
NO <sub>x</sub>	175.5	74.2
CO	5.6	8.1
CO <sub>2</sub> (%)	4.1	4.3
SO <sub>2</sub> <sup>a</sup>	19.5	20.5
UHC <sup>b</sup>	2.3	3.5

<sup>a</sup>Calculated from fuel sulfur assuming  
100 percent conversion to SO<sub>2</sub>

<sup>b</sup>ppmv wet as CH<sub>4</sub>

TABLE 4-4. SULFUR SPECIES EMISSIONS

Test	Species	Emissions			
		ppmv dry	μg/m <sup>3</sup>	kg/min	μg/J
Baseline	SO <sub>2</sub>	11.7	3.12 x 10 <sup>4</sup>	0.385	0.029
	SO <sub>3</sub>	1.1	3.48 x 10 <sup>3</sup>	0.043	0.003
	SO <sub>4</sub>	1.2	4.61 x 10 <sup>3</sup>	0.057	0.004
Water Injection	SO <sub>2</sub>	12.7	3.37 x 10 <sup>4</sup>	0.407	0.030
	SO <sub>3</sub>	1.8	6.04 x 10 <sup>3</sup>	0.073	0.005
	SO <sub>4</sub> <sup>a</sup>	--	--	--	--

<sup>a</sup>Sample destroyed

TABLE 4-5. SULFUR BALANCE

	Baseline	Water Injection
Sulfur Input		
Fuel feedrate (kg/s)	4.85	4.96
Fuel sulfur content (% by wt.)	0.11	0.11
Total sulfur input (kg/s)	$5.33 \times 10^{-3}$	$5.50 \times 10^{-3}$
Sulfur Output		
SO <sub>2</sub> (kg/s)	$6.42 \times 10^{-3}$	$6.78 \times 10^{-3}$
SO <sub>3</sub> (kg/s)	$0.72 \times 10^{-3}$	$1.22 \times 10^{-3}$
SO <sub>4</sub> (kg/s)	$0.95 \times 10^{-3}$	
Total sulfur output (kg/s)	$3.83 \times 10^{-3}$	$3.83 \times 10^{-3}$
Sulfur recovery at outlet	72%	70%

An increase in emissions of unburned species due to lowered peak flame temperatures, is generally associated with the use of water injection for NO<sub>x</sub> control. During the NO<sub>x</sub> control test on Unit 52 average emissions of CO and UHC increased 54 and 52 percent respectively. While the increases seem significant, the actual emission concentrations for CO and UHC are still very low (<10 ppm) when water injection is being used.

Onsite analyses of C<sub>1</sub> to C<sub>6</sub> exhaust gas hydrocarbons were conducted for both the baseline and the water injection tests. The test results show that in the baseline test, C<sub>1</sub> to C<sub>6</sub> hydrocarbons were 6.5ppm at 15 percent O<sub>2</sub> wet, characterized as methane. In the water injection test, C<sub>1</sub> to C<sub>6</sub> hydrocarbon emissions were 1ppm at 15 percent O<sub>2</sub> wet, characterized as methane. These results are in general agreement with the total unburned hydrocarbon emissions measured by the continuous monitor.

#### 4.3.2 Particulate Emissions

Particulate emissions for Unit 52 are shown in Table 4-6. As expected from a gas turbine burning distillate fuel oil, particulate emissions were very low, on the order of 0.0037 to 0.0042 kg/s as measured by the EPA Method 5 train. However, correlation between particulate emission rates from the EPA Method 5 train and the SASS train is poor. SASS measurements are almost a factor of 10 lower. This is not surprising however, when one considers that a SASS train is run at a single point in the exhaust duct. In a duct such as that of Unit 52, where flow patterns are irregular due to the duct configuration, particulate matter can be highly stratified. Furthermore, since particulate matter generated in a gas turbine will be very small in size, it will have a greater tendency to stratify with a strong bias to high velocity regions. Since the SASS train is required to operate at a point of average velocity, away from the high velocity regions, particulate capture is expected to be considerably lower in the SASS train than in the Method 5 train, which fully traverses the duct cross section, as this gives representative results.

TABLE 4-6. PARTICULATE EMISSIONS

Test	Particulate Emissions		
	kg/s	μg/Joule	μg/DSCM
Method 5 -- Baseline	$4.2 \times 10^{-3}$	.019	572
Method 5 -- Water injection	$3.7 \times 10^{-3}$	.016	509
SASS -- Baseline	$0.45 \times 10^{-3}$	.002	63
SASS -- Water injection	$0.97 \times 10^{-3}$	.004	137

While according to the Method 5 measurements, particulate emissions dropped with water injection, the reduction was not significant. Water injection then appears to have little effect on particulate emissions. This is supported by data presented in Reference 3.

#### 4.3.3 Trace Element Characterization and Emissions

Fuel oil, injected water and flue gas samples were collected and analyzed for selected trace elements for the baseline and water injection tests. Grab samples were taken for the oil and water. The flue gas was sampled by using a SASS train. The detailed results of these analyses are presented in Appendices A through F.

The probe wash and the filter have been combined into one sample, as have the aqueous condensate and the first impinger. The XAD-2 cartridge was analyzed independently and the second and third impingers were combined into one sample as outlined in the Level 1 procedures manual (Reference 2).

#### Solid And Vapor Phase Trace Element Partitioning

The SASS train allows determining both solid phase and gas phase composition. Solid phase species are collected in the probe, cyclones, filter and interconnecting tubing, while the vapor phase species are collected in the organics module or the impinger portions of the SASS train. All SASS train components up to the filter are maintained at 394-478K (250 - 400°F). From there, the flue gas goes to the organics module, where it is cooled to approximately 293 K (68°F) and passed through a cross linked porous polymer resin (XAD-2) cartridge. From this section, two samples are generated: the condensate and the XAD-2 sorbent extract. From the organic module, the flue gas goes through an impinger train. The first impinger contains hydrogen peroxide and the second and third impingers contain silver nitrate-ammonium persulfate solutions. For trace element analysis, the organic module aqueous condensate sample is combined with the hydrogen peroxide impinger sample to form one sample for analysis. Thus three samples representing vapor phase composition are analyzed: the XAD-2, the aqueous condensate and hydrogen peroxide impinger solution, and the combined silver nitrate-ammonium persulfate impinger solution.

To determine whether a particular trace element was concentrated in the solid or vapor phase, trace element flowrates (kg/s) were compared. In order to partition the samples as to whether they were solid or vapor, the following partitioning criterion was used: trace elements were

considered to be preferentially concentrated in the vapor phase if their vapor phase concentrations were at least twice their solid phase concentrations. The partitioning results are shown in Table 4-7 for elements where sufficient data to determine partitioning were obtained.

#### Elemental Mass Balance

A trace element mass balance was performed across the gas turbine system using emissions flowrate data from Appendix C. Table 4-8 presents the results of the mass balance. In general and where sufficient data are available, the element mass balances are within the reliability of the Level 1 sampling and analysis procedures which are assumed to be quantitative within a factor of 2 to 3. Zinc and copper are somewhat outside of these boundaries but not significantly so. Iron, as measured at the outlet for both tests, far exceeds the amount entering the turbine as contained in the fuel oil and injected water. It is possible that the source of this excess iron is rust and scale coming loose from the internal gas turbine ductwork and being captured in the SASS train. The analysis of the fuel for iron was supported by a duplicate analysis performed by GE (Appendix G).

#### Effects Due to NO<sub>x</sub> Control

It appears that the use of water injection to control NO<sub>x</sub> emissions has an insignificant effect on trace element emissions. Outlet emissions of all trace elements analyzed remained within a factor of three when comparing the baseline and water injection emission flowrates. Also, water injection has an insignificant effect on trace element emissions with respect to solid/vapor phase partitioning. For those elements where sufficient data were available, the solid/vapor partitioning remained virtually the same.

#### 4.3.4 Organic Analyses

Organic analyses were performed on selected samples according to the EPA Level 1 protocol (Reference 2). Any differences from the Level 1 protocol will be noted in the following discussion. The analytical laboratory data are reported in Appendix F.

As recommended by Level 1 analysis procedures the samples were first extracted with methylene chloride in a Soxhlet apparatus. A Total Chromatographable Organic (TCO) and a gravimetric (GRAV) analysis were then performed on the sample extracts. This analysis separates each

TABLE 4-7. TRACE ELEMENT PARTITIONING -- SOLID PHASE/VAPOR PHASE

	<u>Baseline</u>	<u>Water Injection</u>
Arsenic	X	V
Barium	X	V
Beryllium	X	V
Cadmium	V	V
Chromium	V	EQ
Copper	V	V
Iron	V	V
Lead	V	EQ
Manganese	X	V
Mercury	V	X
Nickel	V	S
Thallium	X	V
Vanadium	V	V
Zinc	X	S

EQ -- Material partitioned equally between vapor and solid phase  
 S -- Material preferentially concentrated in solid phase  
 V -- Material preferentially concentrated in vapor phase  
 X -- Insufficient data

TABLE 4-8. TRACE ELEMENT MASS BALANCE -- OUTLET (g/min)/INLET(g/min)

	<u>Baseline</u>	<u>Water Injection</u>
Boron	<1	1.2
Cadmium	.62	--
Chromium	.24	>3
Cobalt	--	<1
Copper	>5	>7
Iron	>100	>100
Lead	.26	44
Mercury	.16	1.4
Nickel	>.7	>2
Selenium	<.1	--
Vanadium	>.3	>.6
Zinc	3.5	4.6

sample extract into two separate samples having definite boiling point ranges. The TCO fraction contains species with boiling points in a range from 373 K to 573 K. Those species with boiling points above 573 K are contained in the gravimetric sample.

An infrared spectrophotometric (IR) analysis was also performed on the total sample extracts. This aided in the identification of functional organic groups within the complex sample mixture. The organic material in the sample extract was not sufficient to warrant separation by liquid chromatography with further analyses of the fractions eluted. The total sample extracts were analyzed by gas chromatography-mass spectrometry (GCMS) for specific polycyclic organic molecules and priority pollutants.

C<sub>1</sub> to C<sub>6</sub> hydrocarbon compounds were analyzed onsite by gas chromatography. The same set of organic analyses was performed on the samples from the baseline and the water injection test. A discussion of the analytical results follows.

#### Total Chromatographable Organics (TCO) and Gravimetric Analyses (GRAV) of Organic Extracts

Total Chromatographable Organics (TCO) and Gravimetric Analyses (GRAV) were performed on the XAD-2 resin extracts from the baseline and water injection tests. The extract samples were combined with the organic portion of the sorbent module condensate. The results from these analyses are shown in Table 4-9. Three conclusions can be drawn from these results. First, virtually all of the organics in the flue gas can be found in compounds with the boiling point ranging from 373 K to 573 K. Second, there is little effect on the distribution of compounds, with regard to the boiling point, between the baseline and water injection test. Third, the use of water injection has a very small effect on the total amount of organics in the sample extract, decreasing the quantity by approximately 6 percent.

#### Infrared Spectra of Total Extracts

The results of the infrared spectral analyses done on the total XAD-2 sample extracts for the baseline and water injection test are shown in Tables 4-10 and 4-11 respectively. Comparison of the wave numbers and assignments (as well as the spectra themselves) indicate that the organics were almost identical for both tests. Both spectra indicate that the

TABLE 4-9. TCO AND GRAV ANALYSES RESULTS OF THE XAD-2 EXTRACT

Test	Sample Type	Gravimetric Result (mg)	TCO Result (mg)	Total Organics in Extract (mg)	Total Organics Concentration (mg/m <sup>3</sup> )
Baseline	XAD-2 extract	0.6	27.5	28.1	1.3 <sup>a</sup>
Water injection	XAD-2 extract	<0.1	26.3	26.3	1.1 <sup>b</sup>

<sup>a</sup>Based on sample volume = 20.9 m<sup>3</sup>

<sup>b</sup>Based on sample volume = 23.2 m<sup>3</sup>

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principal constituents were an ester or a carboxylic acid and an alcohol. Unsaturated compounds and/or aromatic groups were also present although the intensity of the bands suggest that they may not be part of the principal constituents.

#### Gas Chromatography-Mass Spectrometry (GCMS) Analysis of Total Extracts

Liquid column separation and low resolution mass spectrometry were not performed on the sample extracts because an insufficient sample volume remained after concentration. However, the sample extracts were analyzed by GCMS. Specific compounds were identified and quantified with this technique.

Tables 4-12 and 4-13 show the GCMS results for the baseline and water injection tests. Table 4-14 lists the specific compounds which were analyzed with the GCMS.

#### 4.3.5 Bioassay

Mutagenicity and cytotoxic evaluations were performed on the XAD-2 extract sample from the water injection test. Results from the Ames salmonella/microsome plate test show the sample nonmutagenic. Results of the cytotoxicity assay indicate the extract has low toxicity to WI-38 human cells and that the viability index showed an EC50 value would be obtained at approximately 152 liters gas/ml. Complete results and supporting data are located in Appendix I.

#### 4.3.6 Conclusions

The use of water injection for NO<sub>x</sub> control on Unit 52 appears to have little effect on organic emissions. Neither the total organics, as reported in the TCO and GRAV analyses, nor the species and classes of organics changed significantly from the baseline test to the water injection test.

TABLE 4-10. INFRARED SPECTRA ANALYSIS RESULTS -- XAD-2 EXTRACT,  
BASELINE TEST

Wave Number (cm <sup>-1</sup> )	Intensity <sup>a</sup>	Assignment	Comments
3400-3500	S	O-H	Broad peak
2960, 2920, 2850	S	C-H	Aliphatic
1720-1700	S	C=O	Carbonyl possibly ester
1600	W	C=C	Unsaturated, aromatic
1450	M	-CH <sub>3</sub> bending	Methyl groups
1370	M		Possibly methyl
1260	S	-O- bending	Ether/ester
1070-1090	M	Unassigned	Broad peak
800	W	Unassigned	Possibly aromatic Substitution bands
710	W	Unassigned	Possibly aromatic Substitution bands

<sup>a</sup>Intensity: S - strong, M - medium, W - weak

TABLE 4-11. INFRARED SPECTRA ANALYSIS RESULTS -- XAD-2 EXTRACT,  
WATER INJECTION TEST

Wave Number (cm <sup>-1</sup> )	Intensity <sup>a</sup>	Assignment	Comments
3400-3500	S	O-H	Broad peak
2960, 2920, 2850	S	C-H	Aliphatic
1690-1720	S	C=O	Carbonyl broad
1600	W	C=C	Unsaturated, aromatic
1450, 1460	M	-CH <sub>3</sub>	Methyl band
1380	M		Possibly methyl
1260	S	-O-	Ether/ester
1070-1100	S	Unassigned	Broad peak
800	M	Unassigned	Possibly aromatic Substitution bands
710, 700	M	Unassigned	Possibly aromatic Substitution bands

<sup>a</sup>Intensity: S - strong, M - medium, W - weak

TABLE 4-12. GCMS RESULTS -- XAD-2 EXTRACT -- BASELINE TEST

Compound	Concentration ( $\mu\text{g}/\text{m}^3$ ) <sup>b</sup>
Bis(2-ethylhexyl)phthalate <sup>a</sup>	1.0
Other phthalates	1.0
Phenanthrene/anthracene <sup>a</sup>	0.5
Diphenyl ether	0.5
Phenol	1.0

<sup>a</sup>Confirmed by comparison with standard

<sup>b</sup>Based on sample volume = 20.9 m<sup>3</sup>

TABLE 4-13. GCMS RESULTS -- XAD-2 EXTRACT -- WATER INJECTION TEST

Compound	Concentration ( $\mu\text{g}/\text{m}^3$ ) <sup>b</sup>
Bis(2-ethylhexyl)phthalate <sup>a</sup>	1.0
Other phthalates	1.0
Phenanthrene/anthracene	1.0
Fluoranthene	0.5
Pyrene	0.5
Terphenyl	5.0
Diphenylcyclohexane (2 isomers)	10.0
Phenol	1.0
Naphthalene	1.0

<sup>a</sup>Based on sample volume = 23.2 m<sup>3</sup>

TABLE 4-14. COMPOUNDS SCREENED FOR IN GCMS ANALYSIS OF XAD-2 EXTRACTS

Compound	Representative m/e Values	Compound	Representative m/e Values
7,12 dimethyl benz (a) anthracene	256	Methylnaphthalene	42,127
Dibenz (a,h) anthracene*	278	Biphenyl	154
Benzo (c) phenanthrene	228	Phthalic Anhydride	148,104
3-methyl cholanthrene	268	Nitronaphthalene	173,115
Benzo (a) pyrene*	252	Dibenzothiophene	184,139
Dibenzo (a,h) pyrene	302	Alkanes	57,71
Dibenzo (a,i) pyrene	302	Decalin	67,138
Dibenzo (c,g) carbozole	267	Anthracene*	178
Fluoranthene*	202	Phenanthrene*	178
Pyrene*	202		
Anthanthrene	276		
Benz (a) anthracene*	228		
Benzo (g,h,i) perylene*	276		
Benzo (e) pyrene	252		
Perylene	252		
Naphthalene	128		
Acenaphthylene	152		
Acenaphthene	154		
Fluorene	166		
Diphenyl ether	170		
Dibenzofuran	168		
Fluorenone	180,152		
Naphthoquinone	158,130		
Xanthone	196,168		
Xanthene	182,181		
Quinoline	129		
Phenol	94,65		

\*Contained in standard mixture.

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## SECTION 5

### SUMMARY

Exhaust emissions sampling and analysis was performed on Houston Lighting and Power's Unit 52 at the T. H. Wharton Generating Station in Houston, Texas from April 21-24, 1978. Unit 52 is a General Electric Model MS 70001C simple-cycle, single-shaft, heavy-duty gas turbine rated at 70.8 MW nominal electrical output and fired with distillate oil fuel. The unit is equipped with a water injection system for controlling  $\text{NO}_x$  emissions. The purpose of the tests was to determine the effectiveness of water injection in reducing  $\text{NO}_x$  and to assess the effects that water injection has on emissions other than  $\text{NO}_x$ . In addition, operating parameters were recorded so that effects on turbine operation due to water injection could also be observed.

One baseline test (without water injection) and one water injection test, under nominally similar operating conditions, were performed. A summary of the results is presented below.

#### Unit Operation

The use of water injection did not appear to have any significant impact on unit operations other than an increase in heat rate of approximately 2 percent. This results from some of the fuel heat content being used to vaporize the water. A water/fuel weight ratio of 0.42 was used to reduce  $\text{NO}_x$  emissions to a level just below the proposed New Source Performance Standard (NSPS) of 75 ppm for stationary gas turbines. A higher or lower water/fuel ratio would respectively raise or lower the resulting heat rate. No other significant operational effects were observed as a result of water injection.

#### Emissions

Exhaust emissions were measured for changes resulting from the use of water injection for  $\text{NO}_x$  control. Gaseous, particulate, sulfur species, trace element and organic species emissions were evaluated.

Total  $\text{NO}_x$ ,  $\text{NO}$ ,  $\text{CO}$ ,  $\text{O}_2$ ,  $\text{CO}_2$  and total unburned hydrocarbons were measured by continuous monitoring. With water injection operating at a water/fuel weight ratio of 0.42,  $\text{NO}_x$  emissions were reduced 58 percent from the baseline levels -- from 177.5 to 74.2 ppmv  $\text{NO}_x$  at 15 percent  $\text{O}_2$  dry.  $\text{SO}_2$  emissions are wholly determined by the fuel sulfur content and are not affected by water injection. Changes in emissions of  $\text{CO}$  and total unburned hydrocarbons were within the limits of the analyses. Particulate emissions were very low due to the clean fuel and efficient combustion and did not change significantly with the use of water injection.

It appears that the use of water injection to control  $\text{NO}_x$  emissions has an insignificant effect on trace element emissions. Furthermore, water injection was found to have little effect on trace element emissions with respect to solid/vapor phase partitioning.

Comparisons of organic species emissions between the baseline and water injection test indicate that water injection has little effect on these emissions. Neither the total organics nor the species and classes of organics changed significantly.

Bioassay tests on the XAD-2 extract from the water injection test showed the exhaust gas to be nonmutagenic and of low toxicity as determined by the Ames Salmonella/microsome plate test and the WI-38 cytotoxicity test respectively.

## REFERENCES

## REFERENCES

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APPENDIX A  
FUEL ANALYSIS

PART III - Section No. 1

Table No. 1 - Proximate Analysis  
Concentration in Wt. %

<u>Parameter</u>	<u>Baseline</u>	<u>Water Injection</u>
Moisture	≤0.01	≤0.01
Volatile Matter	-----	-----
Ash	0.01	≤0.005
Carbon	-----	-----
Sulfur	0.10	0.12
BTU (per lb.)	19849	19751
Specific Gravity at 60°F	0.833	0.831

Table No. 2 - Ultimate Analysis  
Concentration in Wt. %

<u>Parameter</u>	<u>Baseline</u>	<u>Water Injection</u>
Moisture	≤ 0.01	≤ 0.01
Ash	0.01	0.005
Carbon	81.98	84.91
Hydrogen	13.16	13.25
Nitrogen	0.11	0.17
Chlorine	0.14	0.13
Sulfur	0.10	0.12
Oxygen (by diff.)	<u>4.5</u>	<u>1.42</u>
TOTAL	100.00	100.00

APPENDIX B  
TRACE ELEMENT CONCENTRATIONS -- ppm

Symbols appearing in the tables:

DSCM	Dry Standard Cubic Meter
ESP	Electrostatic Precipitator
kg	Kilogram
MCG	Microgram
min	Minute
ppm	Part per million by weight
<	Less than
*	Sample not analyzed for the particular element/or ionic specie
N	Sample not analyzed
-	Concentration in the sample less than the concentration in the blank

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (PPH) - FUEL

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .800	< .800
ARSENIC	< 1.00	< 1.00
BARIUM	< .700	< .700
BERYLLIUM	< .300-01	< .300-01
BISMUTH	< .400	< .400
BORON	69.0	53.0
CADMIUM	.900	< .600-01
CHROMIUM	2.90	< .100+00
COBALT	< .200-01	< .500-01
COPPER	< .400	< .400
IRON	< .300-01	< .300-01
LEAD	13.0	2.20
MANGANESE	< .100+00	< .100+00
MERCURY	.700	.600
MOLYBDENUM	< 1.00	< 1.00
NICKEL	< .100-01	< .100-01
SELENIUM	3.00	< 2.00
TELLURIUM	.600	< .800
THALLIUM	< 2.00	< 2.00
TIN	< 3.00	< 3.00
TITANIUM	< 7.00	< 7.00
URANIUM	< 1.00	< 1.00
VANADIUM	< 5.00	< 5.00
ZINC	8.70	7.10
ZIRCONIUM	< 15.0	< 17.0

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (PPM) - WATER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	N .000	< .700-02
ARSENIC	N .000	< .200-01
BARIUM	N .000	< .400-02
BERYLLIUM	N .000	< .200-03
BISMUTH	N .000	< .200-02
BORON	N .000	< 3.00
CADMIUM	N .000	< .400-03
CHROMIUM	N .000	< .600-03
COBALT	N .000	< .100-03
COPPER	N .000	< .300-02
IRON	N .000	< .300-03
LEAD	N .000	< .190
MANGANESE	N .000	< .100-03
MERCURY	N .000	< .880-02
MOLYBDENUM	N .000	< .160
NICKEL	N .000	< .100-03
SELENIUM	N .000	< .200-01
TELLURIUM	N .000	< .600-02
THALLIUM	N .000	< .900-02
TIN	N .000	< .200-01
TITANIUM	N .000	< .600-01
URANIUM	N .000	< .000
VANADIUM	N .000	< .200-01
ZINC	N .000	< .700-01
ZIRCONIUM	N .000	< .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (PPH) - UNIT OUTLET  
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< 10.0	< 10.0
ARSENIC	30.0	40.0
BARIUM	< 7.00	38.0
BERYLLIUM	.600	2.90
BISMUTH	< 2.00	< 2.00
BORON	< .400+04	< .300+04
CADMIUM	8.50	7.10
CHROMIUM	260.	210.
COBALT	.500	6.60
COPPER	160.	77.0
IRON	21.0	190.
LEAD	150.	880.
MANGANESE	< .100+00	.400
MERCURY	< .900	< .900
MOLYBDENUM	< 9.00	< 7.00
NICKEL	3.00	35.0
SELENIUM	< 30.0	< 20.0
TELLURIUM	< 10.0	< 8.00
THALLIUM	< 20.0	50.0
TIN	< 300.	.280+04
TITANIUM	< 60.0	< 50.0
URANIUM	* .000	* .000
VANADIUM	57.0	45.0
ZINC	.540+05	.630+05
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (PPH) - UNIT OUTLET  
XAD-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .700	< .800
ARSENIC	< 2.00	< 2.00
BARIUM	< .600	< .600
BERYLLIUM	< .200-01	< .200-01
BISMUTH	< .300	< .400
BORON	< 370.	< 360.
CADMIUM	< .400-01	< .400-01
CHROMIUM	< .500-01	< .600-01
COBALT	< .800-02	< .800-02
COPPER	2.60	6.60
IRON	11.0	15.0
LEAD	15.0	2.40
MANGANESE	< .800-02	< .800-02
MERCURY	.440	4.30
MOLYBDENUM	< .900	< .900
NICKEL	< .800-02	< .800-02
SELENIUM	< 2.00	< 2.00
TELLURIUM	< .600	< .600
THALLIUM	< 2.00	< 2.00
TIN	< 2.00	< 2.00
TITANIUM	< 5.00	< 6.00
URANIUM	* .000	* .000
VANADIUM	4.00	10.0
ZINC	< .600	4.00
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (PPM) - UNIT OUTLET  
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .600-02	< .500-02
ARSENIC	< .300-01	< .300-01
BARIUM	< .500-02	< .400-02
BERYLLIUM	< .200-03	< .200-03
BISMUTH	< .400-02	< .500-02
BORON	< 3.00	< 3.00
CADMIUM	.200	.370-02
CHROMIUM	.200	.630-01
COBALT	< .100-03	< .200-03
COPPER	.390	.370
IRON	.210	.210
LEAD	.300-01	.160-01
MANGANESE	< .700-04	< .100-03
MERCURY	< .700-03	< .100-02
MOLYBDENUM	.150-01	< .600-02
NICKEL	.240-02	.210-02
SELENIUM	< .100-01	< .100-01
TELLURIUM	< .600-02	< .600-02
THALLIUM	< .500-02	< .600-02
TIN	< .200-01	< .200-01
TITANIUM	< .400-01	< .500-01
URANIUM	* .000	* .000
VANADIUM	< .700-02	< .100-01
ZINC	.560	.530
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (PPM) - UNIT OUTLET  
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTHONY	< .500-02	< .600-02
ARSENIC	< .200-01	< .300-01
BARIUM	* .000	* .000
BERYLLIUM	* .000	* .000
BISMUTH	* .000	* .000
BORON	* .000	* .000
CADMIUM	* .000	* .000
CHROMIUM	* .000	* .000
COBALT	* .000	* .000
COPPER	* .000	* .000
IRON	* .000	* .000
LEAD	* .000	* .000
MANGANESE	* .000	* .000
MERCURY	* .500-02	* .500-02
MOLYBDENUM	* .000	* .000
NICKEL	* .000	* .000
SELENIUM	* .000	* .000
TELLURIUM	* .000	* .000
THALLIUM	* .000	* .000
TIN	* .000	* .000
TITANIUM	* .000	* .000
URANIUM	* .000	* .000
VANADIUM	* .000	* .000
ZINC	* .000	* .000
ZIRCONIUM	* .000	* .000

APPENDIX C  
TRACE ELEMENT FLOWRATES -- kg/min

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (KG/MIN) - FUEL

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .235-03	< .236-03
ARSENIC	< .294-03	< .296-03
BARIUM	< .206-03	< .207-03
BERYLLIUM	< .883-05	< .887-05
BISMUTH	< .118-03	< .118-03
BORON	.203-01	.157-01
CADMIUM	.265-03	.177-04
CHROMIUM	.853-03	.296-04
COBALT	.588-05	.148-04
COPPER	< .118-03	< .118-03
IRON	< .883-05	< .887-05
LEAD	.382-02	.650-03
MANGANESE	< .294-04	< .296-04
MERCURY	.206-03	.177-03
MOLYBDENUM	< .294-03	< .296-03
NICKEL	< .294-05	< .296-05
SELENIUM	.883-03	.591-03
TELLURIUM	< .235-03	< .236-03
THALLIUM	< .588-03	< .591-03
TIN	< .883-03	< .887-03
TITANIUM	< .206-02	< .207-02
URANIUM	< .294-03	< .296-03
VANADIUM	< .147-02	< .148-02
ZINC	.256-02	.210-02
ZIRCONIUM	< .441-02	< .502-02

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (KG/MIN) - WATER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	N .000	< .103-03
ARSENIC	N .000	< .295-05
BARIUM	N .000	< .591-06
BERYLLIUM	N .000	< .295-07
BISMUTH	N .000	< .295-06
BORON	N .000	< .443-03
CADMIUM	N .000	< .591-07
CHROMIUM	N .000	< .886-07
COBALT	N .000	< .148-07
COPPER	N .000	< .443-06
IRON	N .000	< .443-07
LEAD	N .000	.281-04 †
MANGANESE	N .000	< .148-07
MERCURY	N .000	.130-05
MOLYBDENUM	N .000	.236-04 †
NICKEL	N .000	< .148-07
SELENIUM	N .000	< .295-05
TELLURIUM	N .000	< .886-06
THALLIUM	N .000	< .133-05
TIN	N .000	< .295-05
TITANIUM	N .000	< .886-05
URANIUM	N .000	* .000
VANADIUM	N .000	< .295-05
ZINC	N .000	.115-04
ZIRCONIUM	N .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET  
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .165-05	< .143-05
ARSENIC	.494-05	.571-05
BARIUM	< .115-05	.542-05
BERYLLIUM	.989-07	.414-06
BISMUTH	< .330-06	.285-06
BORON	< .659-03	< .428-03
CADMIUM	.140-05	.101-05
CHROMIUM	.429-04	.300-04
COBALT	.624-07	.942-06
COPPER	.264-04	.110-04
IRON	.346-05	.271-04
LEAD	.247-04	.126-03
MANGANESE	< .165-07	.571-07
MERCURY	< .148-06	.128-06
MOLYBDENUM	< .148-05	< .999-06
NICKEL	.494-06	.500-05
SELENIUM	.494-05	.285-05
TELLURIUM	< .165-05	.114-05
THALLIUM	< .330-05	.428-05
TIN	< .494-04	.400-03
TITANIUM	.989-05	.714-05
URANIUM	.000	.000
VANADIUM	.939-05	.642-05
ZINC	.890-02	.899-02
ZIRCONIUM	.000	.000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET  
XAU-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	<	.459-04
ARSENIC	<	.115-03
BARIUM	<	.344-04
BERYLLIUM	<	.115-05
BISMUTH	<	.229-04
BORON	<	.206-01
CADMIUM	<	.229-05
CHROMIUM	<	.344-05
COBALT	<	.459-06
COPPER	<	.378-03
IRON		.860-03
LEAD		.138-03
MANGANESE	<	.459-06
MERCURY	<	.246-03
MOLYBDENUM	<	.516-04
NICKEL	<	.459-06
SELENIUM	<	.115-03
TELLURIUM	<	.344-04
THALLIUM	<	.115-03
TIN	<	.115-03
TITANIUM	<	.344-03
URANIUM	*	.000
VANADIUM	<	.573-03
ZINC	*	.229-03
ZIRCONIUM	*	.000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET  
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .478-05	< .456-05
ARSENIC	< .239-04	< .274-04
BARIUM	< .398-05	< .365-05
BERYLLIUM	< .159-06	< .182-06
BISMUTH	< .319-05	< .456-05
BORON	< .239-02	< .274-02
CADMIUM	< .159-03	< .337-05
CHROMIUM	< .159-03	< .575-04
COBALT	< .797-07	< .182-06
COPPER	< .311-03	< .357-03
IRON	< .167-03	< .192-03
LEAD	< .239-04	< .146-04
MANGANESE	< .558-07	< .912-07
MERCURY	< .558-06	< .912-06
MOLYBDENUM	< .120-04	< .547-05
NICKEL	< .191-05	< .192-05
SELENIUM	< .797-05	< .912-05
TELLURIUM	< .478-05	< .547-05
THALLIUM	< .398-05	< .547-05
TIN	< .159-04	< .182-04
TITANIUM	< .319-04	< .456-04
URANIUM	* .000	* .000
VANADIUM	< .558-05	< .912-05
ZINC	* .446-03	* .483-03
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET  
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION		
	BASELINE, NO WATER	WATER INJECTION	
ANTIMONY	< .537-05	< .494-05	
ARSENIC	< .215-04	< .247-04	
BARIUM	* .000	* .000	
BERYLLIUM	* .000	* .000	
BISMUTH	* .000	* .000	
BORON	* .000	* .000	
CADMIUM	* .000	* .000	
CHROMIUM	* .000	* .000	
COBALT	* .000	* .000	
COPPER	* .000	* .000	
IRON	* .000	* .000	
LEAD	* .000	* .000	
MANGANESE	* .000	* .000	
MERCURY	* .537-05	< .247-05	
MOLYBDENUM	* .000	* .000	
NICKEL	* .000	* .000	
SELENIUM	* .000	* .000	
TELLURIUM	* .000	* .000	
THALLIUM	* .000	* .000	
TIN	* .000	* .000	
TITANIUM	* .000	* .000	
URANIUM	* .000	* .000	
VANADIUM	* .000	* .000	
ZINC	* .000	* .000	
ZIRCONIUM	* .000	* .000	

APPENDIX D  
TRACE ELEMENT FLOWRATES -- MCG/Joule

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - FUEL

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .174-04	< .175-04
ARSENIC	< .218-04	< .219-04
BARIUM	< .152-04	< .153-04
BERYLLIUM	< .653-06	< .656-06
BISMUTH	< .871-05	< .875-05
BORON	.150-02	.116-02
CADMIUM	.196-04	.131-05
CHROMIUM	.631-04	.219-05
COBALT	.435-06	.109-05
COPPER	< .871-05	< .875-05
IRON	< .653-06	< .656-06
LEAD	< .283-03	< .481-04
MANGANESE	< .218-05	< .219-05
MERCURY	.152-04	.131-04
MOLYBDENUM	< .218-04	< .219-04
NICKEL	< .218-06	< .219-06
SELENIUM	.653-04	.437-04
TELLURIUM	< .174-04	< .175-04
THALLIUM	< .435-04	< .437-04
TIN	< .653-04	< .656-04
TITANIUM	< .152-03	< .153-03
URANIUM	< .218-04	< .219-04
VANADIUM	< .109-03	< .109-03
ZINC	.189-03	.155-03
ZIRCONIUM	< .327-03	< .372-03

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - WATER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	N .000	< .765-07
ARSENIC	N .000	< .219-06
BARIUM	N .000	< .437-07
BERYLLIUM	N .000	< .219-08
BISMUTH	N .000	< .219-07
BORON	N .000	< .328-04
CADMIUM	N .000	< .437-08
CHROMIUM	N .000	< .656-08
COBALT	N .000	< .109-08
COPPER	N .000	< .328-07
IRON	N .000	< .328-08
LEAD	N .000	< .208-05
MANGANESE	N .000	< .109-08
MERCURY	N .000	.962-07
MOLYBDENUM	N .000	.175-05
NICKEL	N .000	< .109-08
SELENIUM	N .000	< .219-06
TELLURIUM	N .000	< .656-07
THALLIUM	N .000	< .984-07
TIN	N .000	< .219-06
TITANIUM	N .000	< .656-06
URANIUM	N .000	* .000
VANADIUM	N .000	< .219-06
ZINC	N .000	.853-06
ZIRCONIUM	N .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET  
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .122-06	< .106-06
ARSENIC	< .366-06	.423-06
BARIUM	< .855-07	.402-06
BERYLLIUM	< .732-08	.307-07
BISMUTH	< .244-07	.211-07
BORON	< .488-04	< .317-04
CAUDIUM	.104-06	.751-07
CHROMIUM	.317-05	.222-05
COBALT	.610-08	.698-07
COPPER	.195-05	.814-06
IRON	.256-06	.201-05
LEAD	.183-05	.930-05
MANGANESE	< .122-08	.423-08
MERCURY	< .110-07	.951-08
MOLYBDENUM	< .110-06	.740-07
NICKEL	.366-07	.370-06
SELENIUM	< .366-06	.211-06
TELLURIUM	< .122-06	.846-07
THALLIUM	< .244-06	.317-06
TIN	< .366-05	.296-04
TITANIUM	< .732-06	< .529-06
URANIUM	* .000	* .000
VANADIUM	.696-06	.476-06
ZINC	.659-03	.666-03
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET  
XAD-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	<	<
ARSENIC	<	<
BARIUM	<	<
BERYLLIUM	<	<
BISMUTH	<	<
BORON	<	<
CADMIUM	<	<
CHROMIUM	<	<
COBALT	<	<
COPPER	<	<
IRON	<	<
LEAD	<	<
MANGANESE	<	<
MERCURY	<	<
MOLYBDENUM	<	<
NICKEL	<	<
SELENIUM	<	<
TELLURIUM	<	<
THALLIUM	<	<
TIN	<	<
TITANIUM	<	<
URANIUM	*	*
VANADIUM	<	<
ZINC	*	*
ZIRCONIUM	*	*

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET  
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .354-06	< .338-06
ARSENIC	< .177-05	< .203-05
BARIUM	< .295-06	< .270-06
BERYLLIUM	< .118-07	< .135-07
BISMUTH	< .236-06	< .338-06
BORON	< .177-03	< .203-03
CADMIUM	< .118-04	< .250-06
CHROMIUM	< .118-04	< .426-05
COBALT	< .590-08	< .135-07
COPPER	< .230-04	< .250-04
IRON	< .124-04	< .142-04
LEAD	< .177-05	< .108-05
MANGANESE	< .413-08	< .675-08
MERCURY	< .413-07	< .675-07
MOLYBDENUM	< .885-06	< .403-06
NICKEL	< .142-06	< .142-06
SELENIUM	< .590-06	< .675-06
TELLURIUM	< .354-06	< .405-06
THALLIUM	< .295-06	< .405-06
TIN	< .118-05	< .135-05
TITANIUM	< .236-05	< .338-05
URANIUM	* .000	* .000
VANADIUM	< .413-06	< .675-06
ZINC	* .330-04	* .358-04
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET  
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .398-06	< .366-06
ARSENIC	< .159-05	< .183-05
BARIUM	* .000	* .000
BERYLLIUM	* .000	* .000
BISMUTH	* .000	* .000
BORON	* .000	* .000
CADMIUM	* .000	* .000
CHROMIUM	* .000	* .000
COBALT	* .000	* .000
COPPER	* .000	* .000
IRON	* .000	* .000
LEAD	* .000	* .000
MANGANESE	* .000	* .000
MERCURY	* .398-06	< .183-06
MOLYBDENUM	* .000	* .000
NICKEL	* .000	* .000
SELENIUM	* .000	* .000
TELLURIUM	* .000	* .000
THALLIUM	* .000	* .000
TIN	* .000	* .000
TITANIUM	* .000	* .000
URANIUM	* .000	* .000
VANADIUM	* .000	* .000
ZINC	* .000	* .000
ZIRCONIUM	* .000	* .000

APPENDIX E  
TRACE ELEMENT CONCENTRATION -- MCG/DSCM

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET  
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .133	< .118
ARSENIC	< .400	< .473
BARIUM	< .934-01	< .449
BERYLLIUM	< .800-02	< .343-01
BISMUTH	< .267-01	< .236-01
BORON	< 53.4	< 35.4
CADMIUM	< .113	< .839-01
CHROMIUM	< 3.47	< 2.48
COBALT	< .667-02	< .780-01
COPPER	< 2.13	< .910
IRON	< .280	< 2.25
LEAD	< 2.00	< 10.4
MANGANESE	< .133-02	< .473-02
MERCURY	< .120-01	< .106-01
MOLYBDENUM	< .120	< .627-01
NICKEL	< .400-01	< .414
SELENIUM	< .400	< .236
TELLURIUM	< .133	< .945-01
THALLIUM	< .267	< .354
TIN	< 4.00	< 33.1
TITANIUM	< .800	< .591
URANIUM	* .000	* .000
VANADIUM	< .760	< .532
ZINC	< 720.	< 744.
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET  
XAO-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< 3.64	< 3.80
ARSENIC	< 10.4	< 9.49
BARIUM	< 3.12	< 2.85
BERYLLIUM	< .104	< .949-01
BISMUTH	< 1.56	< 1.90
BORON	< .193+04	< .171+04
CADMIUM	< .208	< .190
CHROMIUM	< .260	< .285
COBALT	< .417-01	< .380-01
COPPER	14.6	31.3
IRON	57.3	71.2
LEAD	78.1	11.4
MANGANESE	< .417-01	< .380-01
MERCURY	2.29	20.4
MOLYBDENUM	< 4.69	< 4.27
NICKEL	< .417-01	< .380-01
SELENIUM	< 10.4	< 9.49
TELLURIUM	< 3.12	< 2.85
THALLIUM	< 10.4	< 9.49
TIN	< 10.4	< 9.49
TITANIUM	< 26.0	< 28.5
URANIUM	* .000	* .000
VANADIUM	20.6	47.5
ZINC	< 3.12	19.0
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET  
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .387	< .378
ARSENIC	< 1.93	< 2.27
BARIUM	.322	< .302
BERYLLIUM	< .129-01	< .151-01
BISMUTH	< .258	< .378
BORON	< 193.	< 227.
CADMIUM	12.9	.279
CHROMIUM	12.9	4.76
CODALY	.645-02	< .151-01
COPPER	25.1	27.9
IRON	13.5	15.9
LEAD	1.93	1.21
MANGANESE	< .451-02	< .755-02
MERCURY	< .451-01	< .755-01
MOLYBDENUM	.967	< .453
NICKEL	.155	.159
SELENIUM	.645	.755
TELLURIUM	.387	.453
THALLIUM	.322	.453
TIN	1.29	1.51
TITANIUM	2.58	3.78
URANIUM	* .000	* .000
VANADIUM	< .451	< .755
ZINC	36.1	40.0
ZIRCONIUM	* .000	* .000

GAS TURBINE  
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET  
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION		
	BASELINE, NO WATER	WATER INJECTION	
ANTIMONY	< .435	< .409	
ARSENIC	< 1.74	< 2.05	
BARIUM	* .000	* .000	
BERYLLIUM	* .000	* .000	
BISMUTH	* .000	* .000	
BORON	* .000	* .000	
CADMIUM	* .000	* .000	
CHROMIUM	* .000	* .000	
COBALT	* .000	* .000	
COPPER	* .000	* .000	
IRON	* .000	* .000	
LEAD	* .000	* .000	
MANGANESE	* .000	* .000	
MERCURY	* .435	< .205	
MOLYBDENUM	* .000	* .000	
NICKEL	* .000	* .000	
SELENIUM	* .000	* .000	
TELLURIUM	* .000	* .000	
THALLIUM	* .000	* .000	
TIN	* .000	* .000	
TITANIUM	* .000	* .000	
URANIUM	* .000	* .000	
VANADIUM	* .000	* .000	
ZINC	* .000	* .000	
ZIRCONIUM	* .000	* .000	

**APPENDIX F**  
**ORGANIC ANALYSIS RESULTS**

TABLE F-1. IR ANALYSIS REPORT

Contractor Acurex  
 Sample Site Houston Lighting + Power Sample Acquisition Date 4-22-78  
 Type of Source Organic extracts from XAD-2 resin, SASS Train.  
 Test Number \_\_\_\_\_ Sample ID Number HLP-1-X  
 Sample Description Baseline  
 Responsible Analyst Harriette Atkins Date Analyzed 8-21-78  
 Calculations and Report Reviewed By M. Beth Heffernan Report Date 9-26-78  
 Instrument Perkin-Elmer M. 297 Sample Cell Type smear on NaCl plate  
 Utilized Max/Min Signal Intensity Values \_\_\_\_\_  
 Observations Obviously, carboxylic acid and/or ester  
and alcohol are major component

Wave Number ( $\text{cm}^{-1}$ )	Intensity	Assignment	Comments
3400-3500	S	O-H	broad peak
2960, 2920,			
- 2850	S	C-H	aliphatic
1720-1700	S	C=O	carbonyl possibly ester
1600	W	C=C	unsaturated aromatic
1450	M	-CH <sub>3</sub> bending	methyl groups
1370	M		possibly methyl
1260	S	-O- bending	ether/ester
1070-1050	M	unassigned	broad peak
800	W	unassigned	possibly aromatic
710	W	unassigned	substitution bands

TABLE F-2. IR ANALYSIS REPORT

Contractor Acurex  
 Sample Site Houston L+P unit #52 Sample Acquisition Date 4-24-78  
 Type of Source XAD-2 resin, SASS train turbine outlet  
 Test Number \_\_\_\_\_ Sample ID Number HLP-2-X  
 Sample Description condensed organic extracts from XAD-2 resin  
Water addition.  
 Responsible Analyst Harriette Atkins Date Analyzed 8-21-78  
 Calculations and Report Reviewed By M. Beth Heffernin Report Date \_\_\_\_\_  
 Instrument Parker-Elmer M. 297 Sample Cell Type Succin or NaCl plate  
 Utilized Max/Min Signal Intensity Values \_\_\_\_\_  
 Observations obviously, carboxylic acid and ester (or kester)  
principal constituent.

Wave Number (cm <sup>-1</sup> )	Intensity	Assignment	Comments
3400-3500	S	O-H	broad peak
2960 2920			
2850	S	C-H	aliphatic
1690-1720	S	C=O	carbonyl, broad
1600	W	C=C	unsaturation, aromatic
1450, 1460	m	-CH <sub>3</sub>	methyl bend
1380	m		possibly methyl
1260	S	-O-	ether/ester
1070-1100	S	unassigned	broad peak
800	m	unassigned	aromatic substitution
710, 700cm	m	unassigned	

04/27/79 16:44:00  
 SAMPLE: MLP-1-X 2  
 RANGE: 0 1.1200  
 MLS. CONC. EXTRACT 7/12/78  
 LABEL: N 432  
 0.1.0 BASE: U 20. 3

COL: C42779 035

11712.

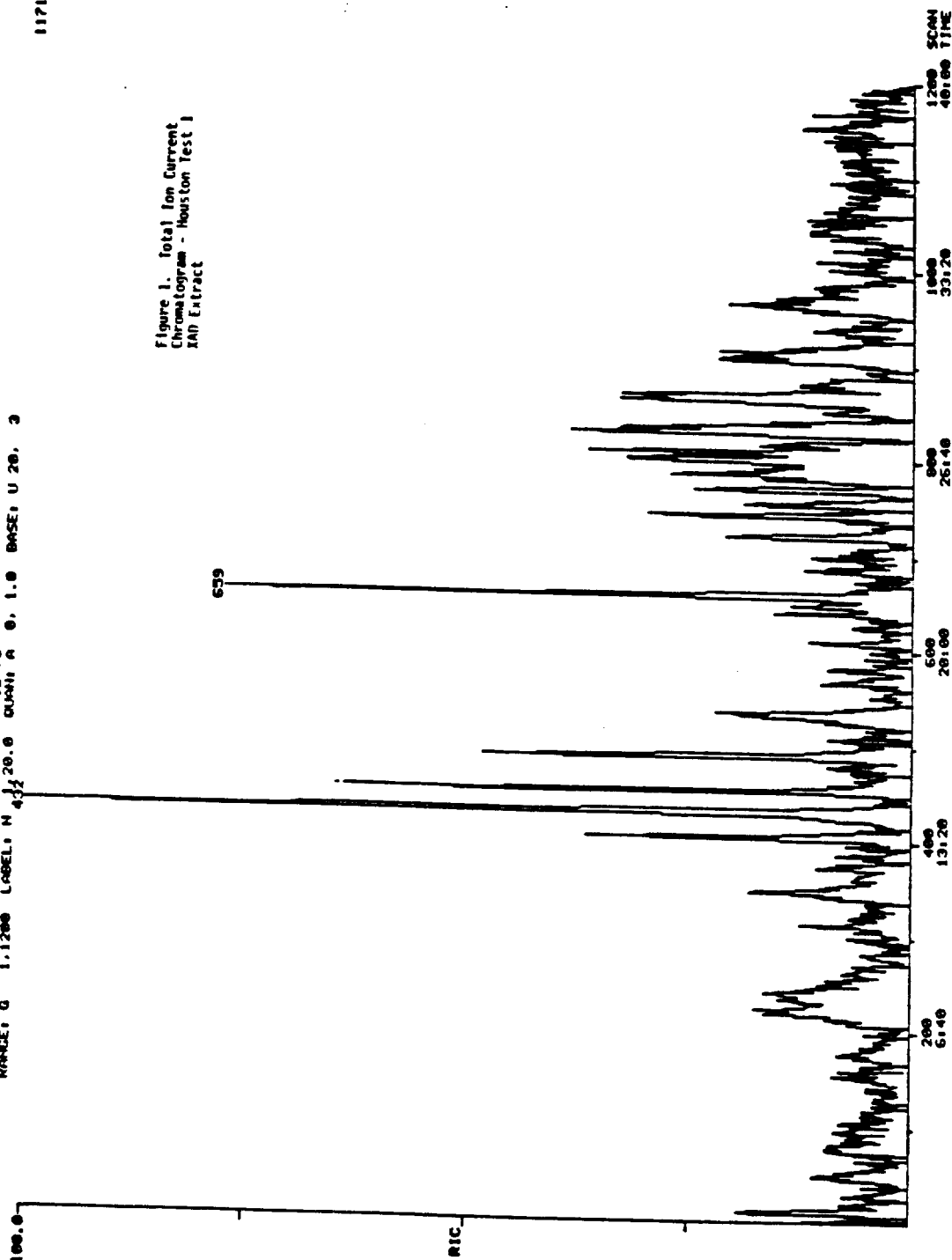


Figure F-1. Total ion current chromatogram -- Houston test 1 XAD extract.

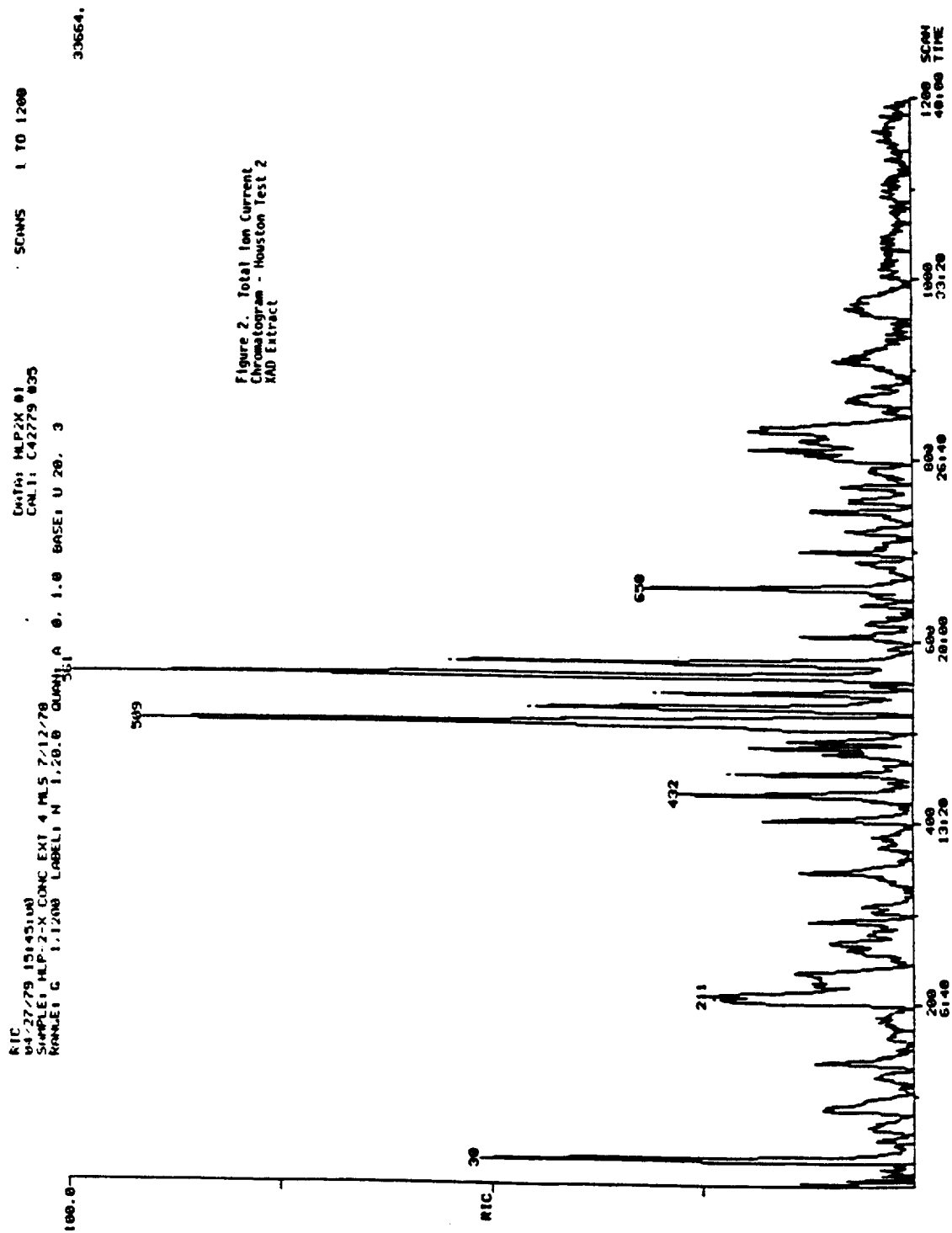


Figure F-2. Total ion current chromatogram -- Houston test 2 XAD extract.

APPENDIX G  
GENERAL ELECTRIC TEST RESULTS

# GENERAL ELECTRIC

## GAS TURBINE PRODUCTS DIVISION

GENERAL ELECTRIC COMPANY, ONE RIVER ROAD, SCHENECTADY, N. Y., U.S.A. 12345  
Phone (518) 374-2211, Telex 145354

July 13, 1978

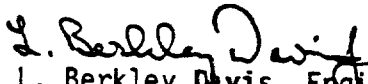
Mr. Brent Higginbotham  
Acurex Aerotherm  
485 Clyde Avenue  
Mountain View, CA. 94042

Dear Brent:

Five copies of General Electric's report on the Acurex-GE joint test at Houston Lighting and Power are attached. These should provide sufficient data for your analysis, but don't hesitate to contact us if you have further questions.

Working with the Acurex test team was a pleasant experience. Perhaps there will be other opportunities for such tests.

Sincerely yours,

  
L. Berkley Davis, Engineer  
Combustion Development-LGT  
Bldg. 53 - Rm. 322

Attachments

LBD:rhb

cc: Nancy Fitzroy, 500-224  
M. B. Hilt, 53-322

MS7001C FIELD TEST RESULTS  
UNIT 52, HOUSTON LIGHTING AND POWER  
APRIL 1978

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During the recent field test of a MS7001C gas turbine, Unit 52, at Houston Lighting and Power, personnel from General Electric's Gas Turbine Division were responsible for measuring gaseous emissions and assessing turbine operation.

This report details the results of these tests and fulfills General Electric's reporting requirements to Acurex under contract RB68439A.

1. RESULTS

Some nineteen test points were run over a period from April 21 to 24, 1978. As indicated, these were spread over three days, with the first day

<u>DATE</u>	<u>TEST</u>
APRIL 21	PRELIMINARY TEST, VELOCITY TRAVERSE
APRIL 22	BASE LOAD, NO WATER INJECTION
APRIL 24	BASE LOAD, WATER INJECTION

devoted to a preliminary test to establish the velocity profile in the exhaust duct.

At each test point, gaseous emissions ( $O_2$ ,  $CO_2$ ,  $NO_x$ , NO, unburned hydrocarbons (UHC), and CO were measured at a single point in the exhaust duct.

Machine operating data, sufficient to establish the airflow rate and operating state of the machine, were also recorded.

The subsequent discussion in part III of this report will address the quality of results expected from single point sampling, as compared to those from a traverse.

Tables (1-2) list the data points for each of the three days. Data from a point are input to a data analysis program called FIRCAL9.

This program utilizes machine performance data, gaseous emissions, fuel composition, and machine geometry and internal flow splits to predict machine operational characteristics. For example, compressor inlet airflow is calculated using four methods; refer to Table (3) for an output sheet from FIRCAL9.

Each column of results (e.g., FT. TEST FLOW) makes use of certain of the data to calculate machine airflow, turbine inlet temperature and exhaust composition. This is illustrated in the table given below.

<u>COLUMN</u>	<u>MEASURED</u>	<u>CALCULATED</u>
FT. TEST FLOW	fuel flow and composition, airflow during factory test, inlet guide lane position, ambient conditions	machine airflow, O <sub>2</sub> CO <sub>2</sub> , water in exhaust, turbine inlet temperature.
CHKD. FLOW	compressor discharge pressure and temperature, first stage nozzle area, fuel flow and composition.	machine airflow/turbine inlet temp., O <sub>2</sub> , CO <sub>2</sub> , H <sub>2</sub> O
OXYGEN CONC.	O <sub>2</sub> , fuel flow and composi- tion	machine airflow, CO <sub>2</sub> , H <sub>2</sub> O, turbine inlet temperature
CO <sub>2</sub> CONC.	CO <sub>2</sub> , fuel flow and composi- tion	machine airflow, turbine inlet temperature, O <sub>2</sub> , H <sub>2</sub> O

Turbine temperatures are proprietary to General Electric and are not included in the results.

Results tables for each test point are interpreted as described in Table (3). The complete sets of results are given in Tables (3-17).

## II. DISCUSSION

The core of the test results is contained in the test points 6-11 and 14-18. The first set is at base load with no water injection, while the second is at base load with 40 gpm water injection. The main points to note are the  $\text{NO}_x$ ,  $\text{NO}_x(\text{ISO})$ ,  $\text{O}_2$ , and airflow rates.

Emissions of nitrogen oxides from a gas turbine are strongly affected by ambient humidity. This is taken into account using the relationship<sup>\*</sup>

$$\text{NO}_x(\text{ISO}) = \text{NO}_x(\text{MEASURED}) e^{23.2(\text{H}-.0063)}$$

While measured  $\text{NO}_x$  varied considerably during the course of the dry test (pts. 6-11) in response to a cold front moving through,  $\text{NO}_x(\text{ISO})$  changes about  $\pm 1$  percent. A similar statement applies to results from the water injection test (pts. 14-18).

---

<sup>\*</sup> The constant 23.2 appearing in the exponential is derived from General Electric data. It gives numbers 2.7 percent higher than the EPA constant .. of 19.0.

### NO<sub>x</sub> EMISSIONS

OPERATING POINT	MEAN	MEAN
	NO <sub>x</sub> (ISO) ppmv	O <sub>2</sub> (% by Vol., dry)
BASE, DRY	159.1 $\pm$ 1.0%	15.30 $\begin{matrix} +1.7\% \\ -3.5\% \end{matrix}$
BASE, 40 gpm	67.7 $\pm$ 1.1%	15.12 $\pm$ 1%

The variation<sup>\*</sup> in O<sub>2</sub> measurements during the dry test is somewhat larger than that in NO<sub>x</sub>(ISO). This is apparently an outright measurement error in one point (pt. 8) that was taken during the period of severe weather.

Airflow in a gas turbine is directly affected by ambient variations and as the numbers in the tables vary by some  $\pm$  2.5 percent with time. The mean levels (for, say, FT. TEST FLOW) are representative of the expected performance of the MS7001C axial compressor.

A comparison of the compressor inlet airflow values from each of the four calculation schemes (CHKD. FLOW, etc.) reveals excellent agreement ( $\pm$  1.0%) between airflow from FT. TEST FLOW, CHKD. FLOW, and OXYGEN CONC. Note that pt. 8 is an exception ( $\pm$  2.2%).

The airflow calculated from measured CO<sub>2</sub> (CO<sub>2</sub> CONC.) shows relatively poor consistency and agreement with the other values. This is particularly apparent in the water injection test, where the airflows predicted from measured CO<sub>2</sub> concentration are approximately ten percent higher than expected.

---

<sup>\*</sup>This variation is calculated as a deficit from 21 percent atmospheric O<sub>2</sub>.

The large amounts of water vapor in the exhaust obviously affect the instrument.

The  $\text{SO}_2$  values in the tables are calculated assuming one hundred percent conversion of fuel sulfur. This is consistent with General Electric's experience.

Two fuel analyses are given in Table (18). The variation in measured hydrogen is typical of that found from repeated measurements of distillate fuels. The ash content is, in both cases, higher than expected for distillates.

Table (19) gives a breakdown of the ash, with the primary constituent being an oxide of zinc. The hydrogen values in Table (19) are determined using the Galbraith method and are not as accurate as those in Table (18).

### III. SUPPORTING TESTS

Subsequent to the Acurex tests, considerable effort was put into obtaining very accurate hydrocarbon measurements. In preparation for these tests, the stainless sampling probe was washed with acetone and methylene chloride, passivated with nitric acid, and washed again with methylene chloride.

In addition, an abbreviated stack traverse (18 pts.) was made.

There are two aspects of this portion of the testing of interest to Acurex: the hydrocarbon readings themselves, and the variation of  $\text{NO}_x$  and  $\text{O}_2$  over the cross section of the exit duct.

Figures (1-3) give the results taken from a base load point with 40 gpm of water being injected. Each figure shows a definite profile across the stack, with a 3  $\text{NO}_x$  variation of  $\pm 5.9\%$  ( $\pm 3.3$  ppmv). The point customarily used

for single point sampling is just above position D-4. These data indicate that the NO<sub>x</sub> is approximately 2.2% lower than the mean value at that point. Thus, the readings reported in the previous section would be within two percent of the true mean.

Oxygen readings have a  $\pm 3\sigma$  variation of  $\pm 4.5\%$  (referenced to  $21 - \overline{O}_2$ ), and verify the trends observed in NO<sub>x</sub>. Unburned hydrocarbons are quite low ( $< 2$  ppmv); they show a very wide spread ( $3\sigma$  is  $\pm 79$  percent).

Other points presented in the tables give supporting data. For example, points 12 and 19 are at base load, dry; they were taken just before and just after the water injection test. Note that NO<sub>x</sub>(ISO) values are in excellent agreement with those previously obtained.

Machine performance and emissions measurements taken from Unit 52 at Houston Lighting and Power are both self-consistent and in agreement with data acquired from other machines of this class.

Close agreement between airflows from different methods of calculation give considerable confidence in the airflow levels. Any discrepancy between these values and that obtained during a stack traverse should be carefully reviewed.

Single point sampling should give adequate results for gaseous emissions such as NO<sub>x</sub> and O<sub>2</sub>. Measurement of species present at very low concentrations requires a traverse.

TABLE 1

To Acurex

GENERAL ELECTRIC CO. SCHENECTADY, N.Y.  
**GAS TURBINE TEST DATA**

(12)

APPARATUS HOUSTON UNIT 52

TEST \_\_\_\_\_

CONDITIONS \_\_\_\_\_

DATA \_\_\_\_\_

POINT \_\_\_\_\_

SHEET \_\_\_\_\_

DATE 4-21-78BAR. 4-22-78

BAR. TIME \_\_\_\_\_

OBSERVER \_\_\_\_\_

DATE	4-21-78	4-21-78	4-21-78	4-21-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78			
TIME	17:35	18:45	19:20	19:45	15:17	17:08	18:00	20:00	21:00	22:00	23:00				
TEST	1	2	3	4	5	6	7	8	9	10	11				
P <sub>cd</sub>	130.5		132.4		119.2	131.0	131.9	134.2	132.0	132.2	131.9				
T <sub>cd</sub>	613.0		605	602	590	612	609	605	607	608	609				
Speed	3600		3600	3600	3600	3600	3600	3600	3600	3600	3600				
V <sub>cf</sub>	7.4														
161	77°	77°	77°	77°	77°	77°	77°	77°	77°	77°	77°				
162	59.3	59.0	60.0	60.0	60.0	61.0	62.0	61.8	62.0	63.0	62.0				
1-2															
3-4															
5-6															
7-8															
9-10															
11-12															
163	978	975	971	969		1002	999	995	996	999	999				
164	0		0		0	0	0	0	0	0	0				
165	0		0												
1-2															
3-4															
5-6															
7-8															
9-10															
166	72°F		66°F			73°F	72°F	73°F	70°F	72°F	72°F				
167	1 min		1 min			1:04.5	1:04.1	2:06.5	2:08.3	2:07.5	2:07.9				
168	91 gpm		92 gpm			100	100	200	200	200	200				
169	196		199		180	198	198	200	200	199	198				
170	390		390		390	395	390	395	390	392	390				
171															
172															
173															
174															
175															
176															
177															
178															
179															
180															

TABLE 2

To Acurex

GENERAL ELECTRIC CO. SCHENECTADY, N.Y.  
**GAS TURBINE TEST DATA**

(12)

APPARATUS HOUSTON UNIT 52

TEST \_\_\_\_\_

CONDITIONS \_\_\_\_\_

DATA \_\_\_\_\_

POINT \_\_\_\_\_

SHEET \_\_\_\_\_

DATE 4-24-78BAR. 0 0 0

BAR. TIME \_\_\_\_\_

OBSERVER MATSONE - RUSSELL

DATE	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78						
TIME	13:10	14:10	15:00	16:00	17:00	18:00	19:00							
POINT	12	13	14	15	16	17	18	19						
P <sub>02</sub>	129.5	130.2	130.5	130.1	129.9	130.9	132.2	131.8						
T <sub>02</sub>	619	624	627	626	628	624	616	612						
S <sub>02</sub>	3600	3600	3600	3600	3600	3600	3600	3600						
V <sub>02</sub>														
161	77°	77°	77°	77°	77°	77°	77°	77°						
162	60.0	60.0	61.0	61.0	61.0	61.5	63.0	62.0						
1-2														
3-4														
5-6														
7-8														
9-10														
11-12														
Ave	1001	1001	1001	1001	1001	1001	996	999						
Grav	0	20	40	40	40	40	40	0						
Temp														
1-2														
3-4														
5-6														
7-8														
9-10														
Temp	80°	80°	81°	80°	77°	74°	70°	70°						
Time	2:11.2	2:28.5	2:06.3	2:06.0	2:06.6	2:05.5	2:03.6	2:05.3						
Cal	200	200	200	200	200	200	200	200						
Temp	195	198	199	198	198	199	200	198						
Temp	395	-	385	385	390	385	390	395						
T <sub>1</sub>														
T <sub>2</sub>														
T <sub>3</sub>														
Temp														
Temp														

04/22/78 12:52HOURS  
FIRCALY REVISED 03/31/78  
MONISTON LIGHTING & POWER  
MS7961C APRIL 1978

G-12

TABLE 4

06/29/78 9.19HOURS EIRCALY REVISED 03/30/78 HOUSTON LIGHTING & POWER M57001C APRIL 1976					
TEST PT	3D-61NW	6	OUTPUT	61.00	ML
DATE	780422		NOX	131.5	PPMV (NET)
TIME	1657	HOURS	NOX (ISO)	158.0	PPMV (NET)
BAROMETER	29.740	INS. HG.	NO	126.1	PPMV (NET)
T (DRY)	74.4	DEG. F.	NO (ISO)	151.5	PPMV (NET)
T (WET)	69.1	DEG. F.	O2	15.43	VOL. (DRY)
REL. HUM.	77.0	%	CO2	4.04	VOL. (DRY)
ABS. HUM.	0.0142	LB5/LB	CO	5.4	PPMV (DRY)
CO	165.5	PPM	UNE	2.7	PPMV (NET) CH4
CO2	612.0	DEG. F.	H2O	0.	GR
I.G.V.	77.0	DEG. F.	FUEL	10.7e3	PPM
COND. EFF.	99.90	%	WATER/FUE	0.	LB5/LB
			FUE	0.008	% BY WT.
FT. TEST (CALC. DAY 1) 202					
WEX (DRY) MSCF/H	FLON	FLON	CONC.	CONC.	
WA	25.7250	25.6775	26.2540	26.0525	
WA (ISO)	563.7	562.0	575.1	570.0	
	569.0	568.9	501.4	546.4	
H2O % OF COND.	0.	0.	0.	0.	
F/A	0.0191	0.0190	0.0187	0.0189	
NOX	435.4	437.6	443.9	440.5	
NOX (ISO)	523.0	525.8	533.3	529.3	
NOX 15% O2 DRY	174.7	176.1	180.6	179.3	
NOX PPM (ISO)	1.57	1.57	1.57	1.56	
NOX (ISO) PPM (ISO)	0.69	0.69	0.70	0.69	
EI NOX PPTP FUEL	11.2	11.3	11.5	11.4	
EI NOX (ISO)	13.5	13.6	13.8	13.7	
URG. NOX 100% YEL	3.1	3.1	3.0	3.0	
O2 CONC %VOL DRY	15.40	15.35	15.43	15.39	
CO2 % VOL DRY	4.14	4.12	4.10	4.09	
CO AT 15% O2 DRY	5.8	5.6	5.4	5.8	
EI CO PPTP FUEL	0.28	0.28	0.29	0.29	
SO2 PPM DRY	19.4	19.5	19.7	19.4	
DRY TO NET VOL	1.062	1.062	1.061	1.062	
MMW (EXHAUST)	26.74	26.74	26.74	26.74	

TABLE 5

05/29/76 9.19HOURS

FISCAL 9 REVISED 03/30/76

HOUSTON LIGHTING & POWER  
M57001C APRIL 1976

TEST PT	40-02100 7)	OUTPUT	62.00	NO
DATE	760422	NOX	130.3	PPMV (NET)
TIME	1402 HOURS	NOX (ISO)	127.2	PPMV (NET)
BAROMETER	29.780 INS. HG.	NO	124.4	PPMV (NET)
T (DRY)	71.2 DEG. F.	NO (ISO)	130.4	PPMV (NET)
T (WET)	68.5 DEG. F.	O2	15.44	VOL. (DRY)
REL. HUM.	67.5 %	CO2	4.04	VOL. (DRY)
ABS. HUM.	0.0144 LBS/LB	CO	5.5	PPMV (DRY)
CDP	140.5 PSI	UNC	2.7	PPMV (NET) CWS
CDT	604.0 DEG. F.	H2O	0.	GPM
I.G.V.	77.5 DEG. F.	FUEL	10.851	PDS
COMB. EFF.	94.40 %	WATER/FUEL	0.	LBS/LB
		FUEL	0.006	W BT.

	FT. 1441	CHOC.	DAY 02	CO2
WEX (DRY) NSCF/HR	FLCH	FLCH	CHOC.	CHOC.
NS	25.9441	26.0559	26.4706	26.2149
MA (ISO)	571.0	571.0	571.0	571.0
	571.0	571.0	571.0	571.0

H2O % OF COMB.	0.	0.	0.	0.
F/A	0.0190	0.0190	0.0117	0.0189
NOX	432.3	437.6	443.7	439.5
NOX (ISO)	525.2	528.2	530.3	531.3
NOX 15% O2 DRY	177.2	177.8	180.5	178.9
NOX BASED TO ISO	0.27	0.27	0.27	0.27
NOX (ISO) PPM	0.69	0.69	0.70	0.69
EI NOX PPM FUEL	11.2	11.2	11.4	11.3
EI NOX (ISO)	13.2	13.5	13.8	13.6
ORG. NOX 100% YEL	3.1	3.1	3.1	3.0
O2 CONC MVOL. DRY	15.33	15.35	15.44	15.39
CO2 % VOL DRY	4.13	4.11	4.10	4.04
CO AT 15% O2 DRY	5.8	5.8	5.9	5.9
EI CO PPM FUEL	0.24	0.29	0.24	0.24
SO2 PPMV DRY	19.6	19.5	19.2	19.4
DRY TO WET VOL	1.062	1.062	1.062	1.062
MMW (EXHAUST)	28.74	28.74	28.73	28.73

TABLE 6

06/29/78 9.19HOURS					
EIRCAL9 REVISED 04/30/78					
HOUSTON LIGHTING & POWER					
MS7001C APRIL 1978					
TEST PT	5D-62MA 8		OUTPUT	62.00	MA
DATE	780422		MLX	138.4	PPMV (NET)
TIME	2030	MSUN	NO (ISO)	157.5	PPMV (NET)
BAROMETER	29.800	INS.HG.	NO	131.7	PPMV (NET)
T (DRY)	67.0	DEG.F.	NO (ISO)	149.4	PPMV (NET)
T (WET)	64.4	DEG.F.	O2	15.47	VOL. (DRY)
REL.HUM.	83.4	%	CO2	4.19	VOL. (DRY)
ABS.HUM.	0.0119	LB/LB	CO	4.4	PPMV (DRY)
CDP	14.0	PSIA	CO2	1.0	PPMV (NET) CH4
CDT	605.0	DEG.F.	H2O	0.	GPM
1.6.V.	77.0	DEG.F.	FUEL	11.002	PPS
COND.EFF.	99.00	%	WATER/FUEL	0.	LB/LB
			FLW	0.000	LB/WT.
TEST TEST					
		FLOW	FLOW	CONC.	CONC.
WEA (DRY) MSCF/HR		26.2866	26.5736	27.0390	25.9918
PPS		57.4	57.4	540.7	548.2
WAT (ISO)		507.0	595.9	500.1	563.0
% OF COMP.					
H2O		0.	0.	0.	0.
F74		0.0191	0.0180	0.0180	0.0194
NOX	PPM	466.3	471.2	474.1	461.2
NOX (ISO)	PPM	530.5	536.4	545.4	525.1
NOX 15% O2 DRY		175.5	177.7	180.7	174.0
NOX PPM 15% O2		0.60	0.61	0.61	0.59
NOX (ISO) PPM 15% O2		0.66	0.67	0.70	0.67
EI NOX PPM FUEL		11.8	11.9	12.1	11.6
EI NOX (ISO)		13.4	13.5	13.8	13.3
URG NOX 100% YEL		3.1	3.1	3.0	3.1
O2 CONC %VOL DRY		15.35	15.30	15.47	15.25
CO2 %VOL DRY		4.14	4.10	4.13	4.19
CO AT 15% O2 DRY		5.2	5.2	5.3	5.1
EI CO PPM FUEL		0.25	0.26	0.26	0.25
SO2 PPM DRY		19.7	19.4	19.1	19.9
DRY TO WET VOL		1.058	1.050	1.057	1.054
MMa (EXHAUST)		28.76	28.71	28.78	28.76

TABLE 7

06/29/78 9.19HOURS

FIRCAL9 REVISIO 03/30/78

HOUSTON LIGHTING & POWER  
MS7001C APRIL 1976

TEST PT	6D-62MA 9	OUTPUT	62.00	MA
DATE	780422	NOX	129.6	PPMV(NET)
TIME	2101 HOURS	NOX(ISO)	155.7	PPMV(NET)
BAROMETER	29.620 INS.HG.	NO	123.1	PPMV(NET)
T (DRY)	71.0 DEG.F.	NO(ISO)	147.9	PPMV(NET)
T (WET)	68.0 DEG.F.	O2	15.31	VOL (DRY)
REL.HUM.	86.1 %	CO2	4.11	VOL (DRY)
ABS.HUM.	0.0142 LBS/LB	CO	5.1	PPMV(DRY)
ODP	146.5 PSIA	UNE	2.1	PPMV(NET) GM4
CD1	607.0 DEG.F.	H2O	0.	GPI
I.G.V.	71.0 DEG.F.	FUEL	10.622	PPS
		WATER/FUEL	0.	LBS/LB
COND.EFF.	94.50 %	FRI.	0.006	% BY WT.

	FT.1857	CH.12	DAYGL	CO2
	FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/HR	25.8254	26.0407	25.8406	26.0602
WA	PPS	565.9	571.6	571.0
WA(ISO)	PPS	584.6	584.9	584.9

H2O	% OF COMB.	0.	0.	0.	0.
W/F		0.0191	0.0189	0.0191	0.0190
NOX	PPM	434.7	434.9	434.9	434.5
NOX(ISO)	PPM	517.5	512.6	517.1	522.1
NOX	15% O2 DRY	174.6	176.1	174.6	175.9
NOX	PPM(T) 15% O2	0.57	0.57	0.57	0.57
NOX(ISO)	PPM(T)	0.67	0.68	0.67	0.68
EI NOX	PPM(T) FUEL	11.1	11.2	11.1	11.2
EI NOX(ISO)		13.4	13.4	13.3	13.4
ORG.NOX	100% YLL	3.1	3.1	3.1	3.1
O2	CONC %VOL DRY	15.32	15.37	15.31	15.36
CO2	% VOL DRY	4.15	4.11	4.14	4.11
CO	AT 15% O2 DRY	5.4	5.5	5.4	5.5
EI CO	PPM(T) FUEL	0.27	0.27	0.27	0.27
SO2	PPMV DRY	19.6	19.5	19.7	19.5
DRY TO WET	VOL	1.062	1.062	1.062	1.062
MMW(EXHAUST)		28.74	28.74	28.74	28.74

TABLE 8

06/29/78 9.19HOURS  
FIRCA19 REVISED 04/30/78  
HOUSTON LIGHTING & POWER  
MS7001C APRIL 1976

TEST PT	70-63MW 10	OUTPUT	63.00	MW
DATE	780422	NOX	131.6	PPHMV(WET)
TIME	2200 HOURS	COX(ISO)	158.8	PPHMV(WET)
BAROMETER	29.720 INS.HG.	NO	124.5	PPHMV(WET)
T (DRY)	70.0 DEG.F.	NO(ISO)	150.9	PPHMV(WET)
T (WET)	68.0 DEG.F.	O2	15.36	VOL (DRY)
REL.HUM.	90.5 %	CO2	4.06	VOL (DRY)
ABS.HUM.	0.0144 LBS/LB	CO	5.3	PPHMV(WET)
CDP	146.7 PSIA	HC	2.9	PPHMV(WET) CH4
CDI	608.0 DEG.F.	H2O	0.	GPM
I.G.V.	77.0 DEG.F.	FUEL	10.890	PDS
COMP.EFF.	99.90 %	WATER/FUEL	9.	LBS/LB
		FLM	0.000	% BY WT.

	FT. TEST	CH2O	OXYGEN	CO2
WEX(DRY) MSCF/H	FLOW	FLOW	CONC.	CONC.
WA	PDS 25.9677	26.0787	46.2552	26.6873
WA(ISO)	PDS 549.1	571.5	574.8	584.5
	PDS 289.6	592.1	595.5	605.6

H2O % OF CONC.	0.	0.	0.	0.
P/A	0.0191	0.0191	0.0189	0.0186
NOX PPM	439.8	441.6	446.2	451.5
NOX(ISO) PPM	530.8	533.0	536.0	544.9
NCA 15% O2 DRY	177.8	178.5	179.4	182.4
NOX DEF TO 15% O2	0.57	0.57	0.58	0.58
NOX(ISO) PPMETS	0.66	0.66	0.69	0.71
EI NOX PPM FUEL	11.2	11.3	11.3	11.5
EI NOX(ISO)	13.5	13.6	13.7	13.9
UPG-NOX 100% YLD	3.1	3.1	3.1	3.0
O2 CONC %VOL DRY	15.31	15.33	15.36	15.46
CO2 % VOL DRY	4.15	4.13	4.11	4.04
CO AT 15% O2 DRY	5.6	5.6	5.7	5.8
EI CO PPM FUEL	0.28	0.28	0.28	0.28
SO2 PPMV DRY	19.7	19.6	19.2	19.2
DRY TO WET VOL	1.062	1.062	1.062	1.061
MMW(EXHAUST)	28.74	28.74	28.73	28.73

TABLE 9

06/29/78 9.19HOURS  
 FIRCALS REVISED 03/30/78  
 HOUSTON LIGHTING & POWER  
 MS7001C APRIL 1976

TEST PT	80-62MW	11	OUTPL1	62.00	MW
DATE	780422		NOX	130.7	PPMV(WET)
TIME	2255	HOURS	NOX(ISO)	155.3	PPMV(WET)
BAROMETER	29.760	INS.HG.	NO	125.1	PPMV(WET)
T (DRY)	73.8	DEG.F.	NO(ISO)	149.2	PPMV(WET)
T (WET)	68.5	DEG.F.	O2	15.40	VOL (DRY)
REL.HUM.	78.8	%	CO2	4.12	VOL (DRY)
ABS.HUM.	0.0136	LBS/LB	CO	5.8	PPMV(DRY)
CDP	144.5	PSIA	UHC	2.0	PPMV(WET) CH4
CDT	609.0	DEG.F.	H2O	0.	GPM
1.G.V.	77.0	DEG.F.	FUEL	10.856	PPS
COND.EFF.	94.40	%	WATER/FUE	0.	LBS/LB
			FBI	0.008	% BY WT.
		FT. TEST	CH4	DRYGL	CO2
WEX(DRY) MSCF/HR	FLGA	25.7564	FLGA	CONC.	CONC.
KA	PPS	566.9	570.5	576.7	571.2
KA(ISO)	PPS	569.0	565.4	501.4	596.2
H2O % OF COMP.	0.	0.	0.	0.	0.
F/A	0.0192	0.0190	0.0188	0.0190	
NOX	PPM	433.4	437.6	442.5	438.4
NOX(ISO)	PPM	517.0	527.0	527.6	522.7
NOX 15% O2 DRY		173.6	175.3	177.7	175.5
NOX PPTD 15% O2		0.56	0.57	0.58	0.57
NOX(ISO) PPTD		0.67	0.66	0.69	0.66
E1 NOX PPTD FUEL		11.1	11.2	11.3	11.2
E1 NOX(ISO)		13.2	13.4	13.5	13.4
ORG.NOX 100% YLD		3.1	3.1	3.0	3.1
O2 CONC 15% O2 DRY		15.26	15.34	15.40	15.35
CO2 % VOL DRY		4.17	4.12	4.08	4.12
CO AT 15% O2 DRY		5.8	5.9	6.0	5.9
E1 CO PPTD FUEL		0.29	0.29	0.24	0.29
SUK	PPMV DRY	19.8	19.6	19.3	19.5
DRY TO WET VOL		1.062	1.061	1.061	1.061
MAW(EXHAUST)		26.75	26.75	26.74	26.74

TABLE 10

06/29/76 10.37HOURS EIRCALY REVISID 03/30/76					
HOUSTON LIGHTING & POWER MS7001C APRIL 1976					
TEST PT	90-60111 12	OUTPUT	80.00	MA	
DATE	780424	NOX	135.8	PPV (NET)	
TIME	1305 HOURS	NOX (ISO)	152.7	PPV (NET)	
BAROMETER	29.850 IN. HG.	HC	130.1	PPV (NET)	
T (DRY)	83.0 DEG. F.	NO (ISO)	152.1	PPV (NET)	
T (WET)	70.2 DEG. F.	O2	15.46	VOL. (DRY)	
REL. HUM.	53.5 %	CO2	3.44	VOL. (DRY)	
ABS. HUM.	0.0130 LBS/LB	CO	5.5	PPV (DRY)	
ODP	144.1	HC	4.1	PPV (NET) CH4	
CDT	619.0 DEG. F.	H2O	0.	GH	
I.G.V.	77.0 DEG. F.	FUEL	10.546	PHS	
COND. EFF.	99.40 %	FUEL/FUEL	0.	LBS/LB	
		ISO	0.008	% BY WT.	
PT. TEST					
NOX (DRY) MSCF/HR	FLOW	FLOW	CONC.	CONC.	
	25.3430	25.7001	25.8058	26.505-	
NOX	PPV	554.5	562.2	565.5	579.5
NOX (ISO)	PPV	585.0	597.7	601.7	610.1
H2O % OF COND.	0.	0.	0.	0.	
F/A	0.0190	0.0108	0.0180	0.0182	
NOX	PPV	441.9	447.5	450.7	461.4
NOX (ISO)	PPV	519.5	529.4	540.1	539.2
NOX	15% O2 DRY	157.5	161.0	157.1	160.3
NOX PPV (15% O2)	157.5	161.0	157.1	160.3	
NOX (ISO) PPV (15% O2)	0.64	0.70	0.70	0.72	
EI NOX PUTE FUEL	11.6	11.6	11.4	12.2	
EI NOX (ISO)	13.1	13.1	13.0	14.2	
URG. NOX 100% YEL	5.1	5.0	5.1	5.9	
O2 CONC. VOL. DRY	15.46	15.43	15.46	15.54	
CO2	3.44	3.44	3.44	3.44	
CO AT 15% O2 DRY	5.9	5.9	5.9	6.1	
EI CO PUTE FUEL	0.24	0.24	0.24	0.30	
SO2	PPV DRY	19.3	19.1	18.7	
DRY TO WET VOL	1.000	1.004	1.005	1.008	
MMW (EXHAUST)	26.76	26.76	26.76	26.76	

TABLE 11

06/29/78 10.37 HOURS  
 FISCAL 9 REVISED 04/30/78  
 HOUSTON LIGHTING & POWER  
 MS7001C APRIL 1978

TEST PT	100-60MW-20GPH	13	OUTPUT	60.00	M <sub>W</sub>
DATE	780424		NO <sub>2</sub>	89.7	PPMV (WET)
TIME	1400	HOURS	NO <sub>X</sub> (ISO)	106.6	PPMV (WET)
BAROMETER	29.830	INS. HG.	NO	84.0	PPMV (WET)
T (DRY)	84.5	DEG. F.	NO (ISO)	99.8	PPMV (WET)
T (WET)	71.5	DEG. F.	O <sub>2</sub>	15.28	VOL. (DRY)
REL. HUM.	53.6	%	CO <sub>2</sub>	3.85	VOL. (DRY)
ABS. HUM.	6.0137	LBS/LB	CO	6.1	PPMV (DRY)
ODP	146.1	PSIA	NO <sub>2</sub>	2.3	PPMV (WET) 1.5%
ODT	624.0	DEG. F.	N <sub>2</sub> O	20.0	GM
I.G.V.	77.0	DEG. F.	FUEL	10.768	PPS
LOMB. EFF.	94.50	%	WATER/FUEL	0.26	LBS/LB
			FIR	0.005	2 LY WT.

	IT-TEST	CFR-1	DAY-1	CL-2
WEX (DRY) MSCF/HR	FLDN	FLDY	CONC.	CONC.
WA	25.2109	25.6333	45.7423	27.8290
WA (ISO)	552.2	561.3	563.8	608.7
	569.4	599.3	602.0	649.8

H <sub>2</sub> O % OF COM.	0.50	0.51	0.49	0.46
F/A	0.0145	0.0142	0.0141	0.0177
NO <sub>X</sub>	253.2	254.6	255.6	322.4
NO <sub>X</sub> (ISO)	348.5	354.6	355.4	383.0
NO <sub>X</sub> 15% O <sub>2</sub> DRY	117.5	119.3	119.8	129.0
NO <sub>X</sub> PPTP FUEL	0.46	0.46	0.47	0.52
NO <sub>X</sub> (ISO) PPTP FUEL	0.46	0.46	0.47	0.50
EI NO <sub>X</sub> PPTP FUEL	7.6	7.7	7.7	8.3
EI NO <sub>X</sub> (ISO)	9.6	9.7	9.7	9.9
URG. NO <sub>X</sub> 100% YLR	3.1	3.1	3.1	2.8
O <sub>2</sub> CONC % VOL DRY	15.17	15.27	15.29	15.72
CO <sub>2</sub> % VOL DRY	4.25	4.16	4.16	3.85
CO AT 15% O <sub>2</sub> DRY	6.3	6.4	6.4	6.9
EI CO PPTP FUEL	0.31	0.32	0.32	0.34
SO <sub>2</sub> PPTP DRY	20.1	19.8	19.7	18.2
DRY TO WET VOL	1.071	1.070	1.070	1.066
MMW (EXHAUST)	28.65	28.65	28.66	28.67

TABLE 12

06/29/78 10.37HOURS  
 EIRCAL9 REVISED 04/30/78  
 HOUSTON LIGHTING & POWER  
 MS/001C APRIL 1976

TEST PT	110-61MW-40GPM	14	OUTPUT	61.00	HA
DATE	780424		NOX	57.5	PPMV(WET)
TIME	1501	HOURS	CO2(ISO)	67.8	PPMV(WET)
BAROMETER	29.810	INS.HG.	NO	53.4	PPMV(WET)
T (DRY)	85.0	DEG.F.	NO(ISO)	62.9	PPMV(WET)
T (WET)	71.2	DEG.F.	O2	15.14	VOL (DRY)
REL.HUM.	51.4	%	CO2	4.00	VOL (DRY)
ABS.HUM.	0.0133	LBS/LB	CO	7.4	PPMV(DRY)
CDP	145.1	PSIA	H2O	4.3	PPMV(WET, CO2)
CD1	627.0	DEG.F.	H2O	40.0	GPM
I.G.V.	77.0	DEG.F.	FUEL	10.455	PPS
COMB.EFF.	94.90	%	WATER/FUEL	0.21	LBS/LB
			FEI	0.008	% BY WT.

	FT. TEST	CHL.	DAYG.	CO2
	FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/HK	25.1664	25.5431	25.6357	27.3753
WA	PPS	551.2	554.3	599.1
WA(ISO)	PPS	589.5	598.2	640.7

H2O % OF COMB.	1.01	0.99	0.99	0.93
F/A	0.0199	0.0196	0.0195	0.0183
NOX PPM	185.4	192.1	197.8	205.1
NOX(ISO) PPM	223.2	226.4	247.1	241.7
NOX 15% O2 DRY	73.5	74.6	74.4	79.7
NOX PPM/D, 15% O2	0.24	0.25	0.22	0.26
NOX(ISO) PPM/D	0.29	0.29	0.24	0.31
EI NOX PTP FUEL	4.8	4.9	4.9	5.2
EI NOX(ISO)	5.7	5.7	5.8	6.1
URG.NOX 100% YLI	3.1	3.1	3.1	2.9
O2 CONC %VOL.DRY	15.03	15.12	15.14	15.51
CO2 % VOL DRY	4.35	4.29	4.27	4.00
CO AT 15% O2 DRY	7.9	8.0	8.0	8.6
EI CO PTP FUEL	0.40	0.40	0.41	0.43
SO2 PPMV DRY	20.6	20.3	20.2	18.9
DRY TO WET VOL	1.080	1.074	1.074	1.075
MMW(EXHAUST)	28.56	28.56	28.56	28.59

TABLE 13

06/29/78 10.37 HOURS

FIRCAIR REVISED 04/30/78

HOUSTON LIGHTING & POWER  
MS7001C APRIL 1976

TEST PT	120-61AW-40GPM	15	OUTPUT	61.00	MA
DATE	780424		FLX	57.8	PPMV (NET)
TIME	1600	HOURS	NOX (ISO)	57.4	PPMV (NET)
BAROMETER	29.790	INS. HG.	NO	53.7	PPMV (NET)
T (DRY)	83.5	DEG. F.	NO (ISO)	63.0	PPMV (NET)
T (WET)	70.5	DEG. F.	O2	15.13	VOL. (DRY)
REL. HUM.	53.1	%	CO2	3.92	VOL. (DRY)
ABS. HUM.	0.0131	LBS/LB	CO	8.1	PPMV (DRY)
CHP	144.7	PSIA	UHC	2.4	PPMV (NET) CH4
CDT	626.0	DEG. F.	H2O	40.0	GPI
LEG. V.	27.0	DEG. F.	FUEL	10.961	PDS
COMB. EFF.	99.90	%	WATER/FUEL	0.21	LBS/LB
			FLX	0.005	% BY WT.
FT. 1.57					
WEX (DRY) MSCF/HF	FLC	FLC	CONC.	CONC.	
WA	PDS	PDS	25.4344	25.6521	28.0028
WA (ISO)	PDS	PDS	552.8	552.9	552.0
			554.6	554.6	553.2
H2O % OF COMB.					
F/A	T.01	1.00	0.99	0.91	
NOX	0.0199	0.0197	0.0195	0.0179	
NOX (ISO)	PPH	197.2	193.8	210.6	
NOX 15% O2 DRY	PPH	223.4	225.5	247.3	
NOX PPM-TJ WET	PPH	73.6	74.1	81.3	
NOX (ISO) PPM-TJ	PPH	0.29	0.29	0.27	
EI NOX PPTP FULL	PPH	4.8	4.9	0.32	
EI NOX (ISO)	PPH	5.7	5.7	5.3	
URG. NOX 100% YLD	PPH	3.1	3.1	2.9	
O2 CONC % VOL. DRY	PPH	15.03	15.00	15.13	
CO2 % VOL. DRY	PPH	4.35	4.37	4.27	
CO AT 15% O2 DRY	PPH	8.1	8.2	8.3	
EI CO PPTP FUEL	PPH	0.41	0.41	0.42	
CO2 PPMV DRY	PPH	20.6	20.5	20.3	
DRY TO WET VOL	PPH	1.079	1.074	1.076	
MMW (EXHAUST)	PPH	26.56	26.57	26.54	

TABLE 14

06/29/76 10.37HOURS

FIRCALLY REVISED 04/30/78

HOUSTON LIGHTING & POWER  
MS7U01C APRIL 1976

TEST PT	130-61ML-40GPH 16	OUTPUT	61.00	MA
DATE	780424	NOX	57.6	PPMV(WET)
TIME	1700 HOURS	NOX(ISO)	67.6	PPMV(WET)
BAROMETER	29.780 IN.HG.	NO	53.6	PPMV(WET)
T (DRY)	83.3 DEG.F.	NO(ISO)	62.9	PPMV(WET)
T (WET)	70.5 DEG.F.	O2	15.14	VOL. (DRY)
REL.HUM.	53.6 %	CO2	3.93	VOL. (DRY)
ABS.HUM.	0.0132 LBS/LB	CO	7.8	PPMV(DRY)
CO	166.5 PPM	ULE	6.1	PPMV(WET, CH6)
CDT	628.0 DEG.F.	H2O	40.0	GPH
I.G.V.	77.0 DEG.F.	FUEL	10.944	PPS
COMB.EFF.	94.90 %	WATER/FUEL	0.21	LBS/LB
		FE	0.008	% BY WT.

	FT. TEST	CHCL.	DRY G-1	CO2
WEX(DRY) MSCF/HK	FLOW	FLOW	CONC.	CONC.
MA	25.2463	25.4036	45.6020	27.8356
MA(ISO)	552.9	556.2	501.0	608.8
	589.6	593.2	578.3	649.3

H2O % OF COMB.	1.001	1.001	0.997	0.991
F/A	0.0196	0.0197	0.0195	0.0180
NOX	PPM	190.1	191.2	172.7
NOX(ISO)	PPM	223.3	224.6	246.3
NOX 15% O2 DRY	74.6	74.1	74.7	80.3
NOX PERMIT 15% O2	0.25	0.25	0.25	0.27
NOX(ISO) PPMETJ	0.25	0.25	0.25	0.32
EI NOX PPTP FUEL	4.6	4.6	4.6	5.3
EI NOX(ISO)	5.7	5.7	5.7	6.2
ORG.NOX 10L% YLL	3.1	3.1	3.1	2.9
O2 CONC %VOL DRY	15.13	15.09	15.14	15.61
CO2 % VOL DRY	4.33	4.31	4.27	3.93
CO AT 15% O2 DRY	8.0	7.9	8.0	8.7
EI CO PPTP FUEL	0.40	0.40	0.40	0.44
SO2 PPMV DRY	20.3	20.4	20.2	18.6
DRY TO WET VOL	1.074	1.074	1.076	1.074
MMW(EXHAUST)	28.56	28.56	28.54	28.60

TABLE 15

06/29/78 10.37HOURS  
FIRCAIR REVISED 03/30/78HOUSTON LIGHTING & POWER  
MS7001C APRIL 1978

TEST PT	140-51.5MW-40GPM	17	OUTPUT	61.50	HP
DATE	780424		NOX	55.8	PPMV(WET)
TIME	1800	10.37	NOX(ISO)	56.4	PPMV(WET)
BAROMETER	29.760	INS.HG.	NO	51.8	PPMV(WET)
T (DRY)	81.2	DEG.F.	NO(ISO)	62.1	PPMV(WET)
T (WET)	71.0	DEG.F.	O2	15.10	VOL (DRY)
REL.HUM.	61.2	%	CO2	3.87	VOL (DRY)
ABS.HUM.	0.0141	LBS/LB	CI	8.0	PPMV(WET)
CO2	14.5	PPM	CO	4.1	PPMV(WET)
CDT	624.0	DEG.F.	H2O	40.0	GPM
LOG.V.	77.0	DEG.F.	FUEL	11.054	PPS
COMP.EFF.	94.80	%	WATER/FUEL	6.26	LBS/LB
			FIH	0.006	% BY WT.

	FT. LBS	CHLOR.	DAYLIGHT	CO2
WEX(DRY) MSCF/HK	FLOW	FLOW	CONC.	CONC.
WEX	25.3440	25.5269	25.6873	26.5513
WEX(ISO)	554.4	554.4	553.3	624.9
	557.8	553.6	557.6	663.1

H2O % OF COMP.	1.00	0.95	0.99	0.89
F/A	0.0139	0.0145	0.0156	0.0177
NOX	165.4	164.5	147.0	207.4
NOX(ISO)	222.0	223.5	244.1	248.6
NOX 15% DRY	73.5	73.0	73.5	81.3
NOX PPM T. ISO T.	0.25	0.25	0.25	0.26
NOX(ISO) PPM T.	0.28	0.29	0.29	0.32
EI NOX PPM FUEL	4.7	4.7	4.7	5.2
EI NOX(ISO)	5.0	5.1	5.7	6.2
NOX NOX 100% YEL	3.1	3.1	3.1	2.8
O2 CONC SVOL DRY	15.10	15.00	15.10	15.69
CO2	4.34	4.33	4.32	3.87
CO AT 15% O2 DRY	8.2	8.1	8.2	9.1
EI CO PPM FUEL	0.41	0.41	0.41	0.46
SO2 PPM DRY	20.4	20.5	20.4	18.3
SO2 TO WET VOL	1.081	1.081	1.080	1.074
MMH(EXHAUST)	26.55	26.55	26.51	26.54

TABLE 16

06/29/78 10.37HOURS

RECALCULATED 04/30/78

HOUSTON LIGHTING & POWER  
MS7001C APRIL 1978

TEST PT	150-6311-40GPM 18	OUTPUT	63.00	M.
DATE	780424	NOX	96.1	PPMV (NET)
TIME	1800 HOURS	NOX (ISO)	68.5	PPMV (NET)
BAROMETER	29.780 INS. HG.	NO	51.9	PPMV (NET)
T (DRY)	77.0 DEG. F.	NO (ISO)	63.4	PPMV (NET)
T (WET)	70.8 DEG. F.	O2	15.07	VOL. (DRY)
REL. HUM.	74.2 %	CO2	3.95	VOL. (DRY)
ABS. HUM.	0.0149 LBS/LI	CO	7.9	PPMV (DRY)
CDP	142.8 PSIA	UNIC	2.8	PPMV (NET) CH4
CDT	616.0 DEG. F.	H2O	40.0	GPM
I.G.V.	77.0 DEG. F.	FUEL	11.244	PPS
COMB. EFF.	99.90 %	WATER/FUEL	0.69	LBS/LB
		FIN.	0.008	% BY WT.

	FT. TEST	CH4	OXYGEN	CO2
	FLOW	FLOW	CONC.	CONC.
NOX (DRY) MSCF/HK	25.5647	25.7321	25.9913	26.2305
NO	561.0	566.4	570.5	616.7
NO (ISO)	389.0	593.3	579.6	650.1

H2O % OF COMB.	0.99	0.98	0.97	0.90
F/A	0.0200	0.0199	0.0197	0.0182
NOX	188.1	184.2	191.0	206.6
NOX (ISO) PPM	229.6	231.2	233.4	252.4
NOX 15% O2 DRY	74.9	74.3	75.0	81.1
NOX PPM (NET) (DRY)	0.24	0.24	0.24	0.26
NOX (ISO) PPM (NET)	0.29	0.29	0.29	0.32
EI NOX PPTP FULL	4.6	4.7	4.7	5.1
EI NOX (ISO)	5.7	5.7	5.8	6.2
UNQ. NOX 100% YLD	3.2	3.2	3.1	2.9
O2 CONC % VOL. DRY	15.06	15.01	15.07	15.54
CO AT 15% O2 DRY	8.0	7.9	8.0	8.7
EI CO PPTP FULL	0.40	0.40	0.41	0.44
SO2 PPMV DRY	20.5	20.7	20.5	18.9
DRY TO WET VOL	1.063	1.062	1.062	1.077
MMW (EXHAUST)	28.54	28.54	28.50	28.57

TABLE 17

06/29/78 10.37HOURS  
 EIRCALY REVISED 03/30/78  
 HOUSTON LIGHTING & POWER  
 MS7001C APRIL 1978

TEST PT	16D-62MW 19	OUTPUT	62.00	MA
DATE	7/10/24	NOX	130.8	PPMV(WET)
TIME	2000 HOURS	NOX(ISO)	158.1	PPMV(WET)
BAROMETER	29.820 INS.HG.	NO	123.6	PPMV(WET)
T (DRY)	73.5 DEG.F.	NO(ISO)	149.8	PPMV(WET)
T (WET)	69.3 DEG.F.	O2	15.35	VOL. (DRY)
REL.HUM.	81.4 %	CO2	3.82	VOL. (DRY)
ABS.HUM.	0.0145 LBS/LB	CO	5.7	PPMV (DRY)
CDP	146.4 PSI4	UNE	1.2	PPMV (WET) CH4
CDT	612.0 DEG.F.	H2O	0.	GPM
I.G.V.	77.0 DEG.F.	FUEL	10.832	PPS
COND.EFF.	94.90 %	WATER/FUEL	0.	LBS/LE
		FEI	0.006	% BY WT.

	FT. TEST	CH4	DAY 1	CO2
WEX(DRY) MSCF/HK	FLOW	FLOW	CONC.	CONC.
MA	25.8436	26.0124	26.0477	26.0618
MA(ISO)	549.6	570.1	570.1	614.5
	549.6	593.4	574.1	639.6

H2O % OF COND.	0.	0.	0.	0.
F/A	0.0191	0.0190	0.0190	0.0170
NOX	433.1	437.5	438.4	471.4
NOX(ISO)	526.7	530.0	530.4	570.6
NOX 15% O2 DRY	177.2	178.4	178.5	191.4
NOX DRY 15% O2	0.57	0.57	0.57	0.61
NOX(ISO) PPTP	0.59	0.59	0.59	0.74
EI NOX PPTP FUEL	11.2	11.2	11.2	12.1
EI NOX(ISO)	13.5	13.6	13.6	15.6
ORG. NOX 15% O2 YLP	3.1	3.1	3.1	2.8
O2 CONC %VOL DRY	15.31	15.34	15.35	15.76
CO AT 15% O2 DRY	4.15	4.17	4.17	3.82
CO AT 15% O2 DRY	6.0	6.0	6.0	6.5
EI CO PPTP FULL	0.29	0.30	0.30	0.32
SO2 PPMV DRY	19.7	19.5	19.5	18.1
DRY TO WET VOL	1.063	1.062	1.062	1.060
MMK(EXHAUST)	28.73	28.73	28.73	28.73

TABLE 18  
MATERIALS AND PROCESSES LABORATORY  
SCHENECTADY, NEW YORK  
ANALYSIS REQUEST-REPORT

RECEIVED 5/24/78 L.O. NO. 78C-1068  
 REPORTED 5/31/78 SPECTRUM NO. \_\_\_\_\_  
 REQUESTED BY R.A. Hiskanin BLDG. 262 RM. 105 EXT. \_\_\_\_\_  
 PROJECT NO. \_\_\_\_\_  
 SHOP ORDER 5910-440-300-114 DEPT. \_\_\_\_\_ CC. \_\_\_\_\_  
 DESCRIPTION OF MATERIAL Gas Turb. Lab  
Fuel Type #2 (4-26) HLP  
SN 209

Analysis Requested	X-Ray		Spectro		Probe	Organic	Thermal	Gas	Part.	
	Met	AA	emiss	diff					qual	quant
	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>

Met: MMW, ash, H X-ray: S Evendale: H McBride: Spec. Grav. & Visc. @ 100°F  
% water 210°F, C residue, Distillation curve,  
Galbraith: C/H ratio aniline pt.

RESULTS	MMW (BTU/lb)	19,730
pow ash		32
% N		0.008
% S		0.11
% H		13.55
% H <sub>2</sub> O		<0.02
Sp. Gr. @ 150°F		0.820
Sp. Gr. @ 210°F		0.782
Viscosity @ 100°F		2.75
Viscosity @ 210°		1.07
Aniline Point °F		156
Carbon Residue (10° Bottoms)		0.18
Distillation	°F	°F
Initial Point	326	50% Distilled 482
10% Distilled	396	60% " 502
20% "	422	70% " 526
30% "	440	80% " 550
40% "	464	90% " 576

Inquiries should be directed to: Dick Northrup/ Bob McBride Ext. 5-2113

# TABLE 18 CONCLUDED

RECEIVED 5/24/78 LAB. NO. 78C-1069  
 PORTED 5/21/78 SPECTRUM NO. \_\_\_\_\_  
 REQUESTED BY R.A. Miskarin BLDG. 262 RM. 105 EXT. \_\_\_\_\_  
 PROJECT NO. \_\_\_\_\_  
 SHOP ORDER 5818-440-300-113 DEPT. \_\_\_\_\_ CC. \_\_\_\_\_

DESCRIPTION OF MATERIAL Gas Turb. Lab  
Fuel Type 42 (4-22) HL&P  
SN 288

Analysis Requested	Wet	AA	X-Ray		Spectro		Probe	Organic	Thermal	Gas	Part. Study	Other
			emiss	diff	qual	quant						
	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>

Wet: HHV, ash, N X-ray: S Evendale: H McBride: Spec. Grav. & Visc. @ 100°F

% water 210°F, C residue, Distillation curve.

Galbrath: C/H ratio aniline pt.

RESULTS	HHV (BTU/lb)	19.665
	ppm ash	23
	SN	0.009
	XS	0.11
	XH	13.40
	XH <sub>2</sub> O	<0.02
	Sp. Gr. @ 100°F	0.823
	Sp. Gr. @ 210°F	0.782
	Viscosity @ 100°F	2.59
	Viscosity @ 210°F	1.08
	Aniline Point	155°F
	Carbon Residue (10% bottoms)	0.29
	Distillation	°F
	Initial Point	324
	10% Distilled	401
	20% "	428
	30% "	448
	40% "	466
	50% Distilled	487
	60% "	508
	70% "	528
	80% "	552
	90% "	550

Inquiries should be directed to: Dick Wertheff / Bob McBride Ext. 5-2113

TABLE 19  
MATERIALS AND PROCESSES LABORATORY  
SCHENECTADY, NEW YORK  
ANALYSIS REQUEST-REPORT

LAB. NO. 75C-1068  
SPECTRUM NO. BF1  
E.A. Hisklin BLDG. 267 RM. 105 EXT.  
DEPT. CC.  
SECTION OF MATERIAL Gas Turb. Lab  
Fuel Type #2 (4-26) H&P  
SM 289

Analysis Requested	X-Ray		Spectro		Probe	Organic	Thermal	Gas	Part. Study	Other
	Wet	AA	emiss	diff						
	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>

Galbraith: C/H ratio

RESULTS	approx. % in ash
Zn	major >10
Pb	0.3
Sn	0.2
Si	0.2
Fe	0.2
Cu	0.1
Ca	0.04
Mg	0.02
V	0.01
Mo	0.01
Al	0.009
B	0.008
Mn	0.004
Ni	0.003
C	85.07
H	13.23

Inquiries should be directed to: B.H. Kint Ext. 5-2117

# TABLE 19 CONCLUDED

RECEIVED 5/24/78 LAB. NO. 79C-1069  
 REPORTED 6/7/78 SPECTRUM NO. 8F1  
 REQUESTED BY R.A. Miskarin BLDG. 262 RM. 105 EXT.   
 PROJECT NO.   
 SHOP ORDER 5818-440-300-113 DEPT.  CC.   
 DESCRIPTION OF MATERIAL Gas Turb. Lab  
Fuel Type #2 (4-22) HL&P  
SN 288

Analysis Requested	Wet		X-Ray		Spectro		Probe	Organic	Thermal	Gas	Part.	
		AA	emiss	diff	qual	quant					Study	Other
	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>

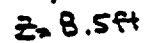
Galbraith: C/H ratio

RESULTS approx. % in ash

Zn	major	>10
Pb		0.2
Fe		0.2
Sn		0.1
Cu		0.1
V		0.05
Si		0.04
Ca		0.03
Ni		0.02
Mn		0.01
Mo		0.006
Al		0.005
Mn		0.002
C		84.83
H		13.07

Inquiries should be directed to: BH Kilt Ext. 5-2113

ALL QUANTITIES ARE  
IN PPM BY VOLUME



!

DATA ARE MEAN.  
 READINGS AT EACH SAMPLE  
 POINT. STANDARD  
 DEVIATIONS FOR THESE  
 READINGS ARE LESS  
 THAN 0.01.

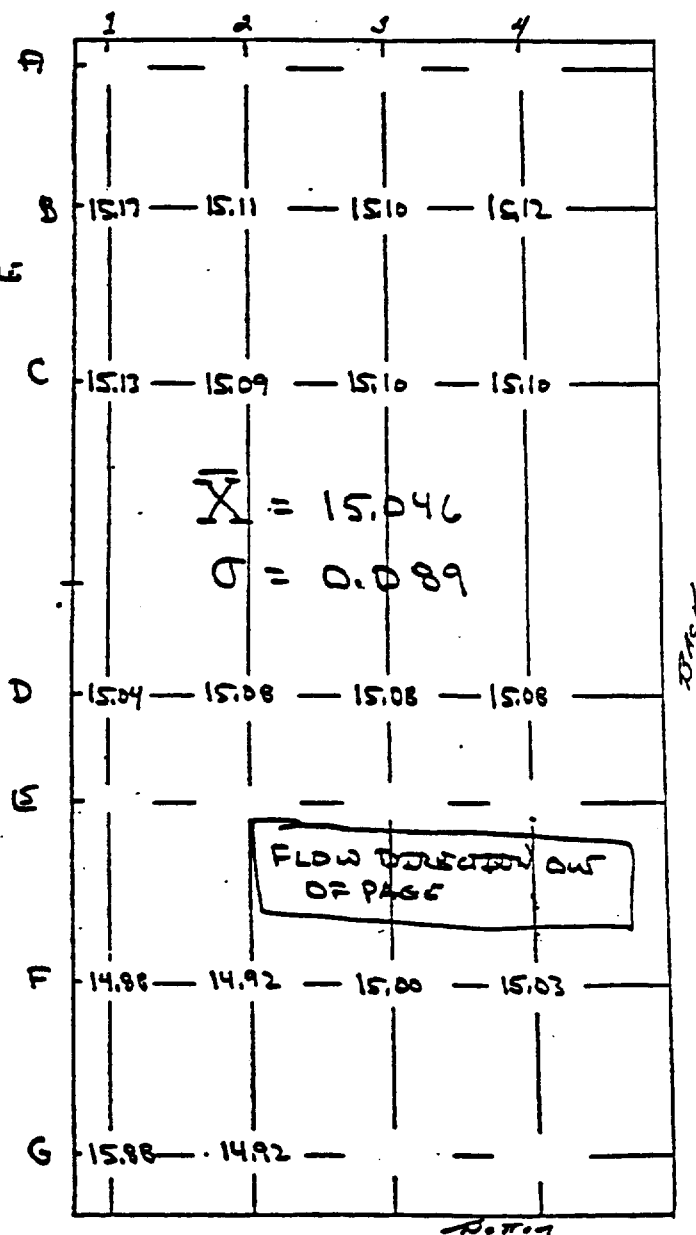


Figure 2. M57001C oxygen concentration in exhaust at base load with water injection.

FACING  
AGAINST AIR  
FLOW

NO<sub>x</sub> DATA  
As P/sms.  
PPMV

THIS PORT  
is 40° up Stream

$\bar{x} = 57.19$   
 $s = 1.113$   
 $\frac{s}{\bar{x}} = .0195$   
2.452  
40 Gpm  
160 ft

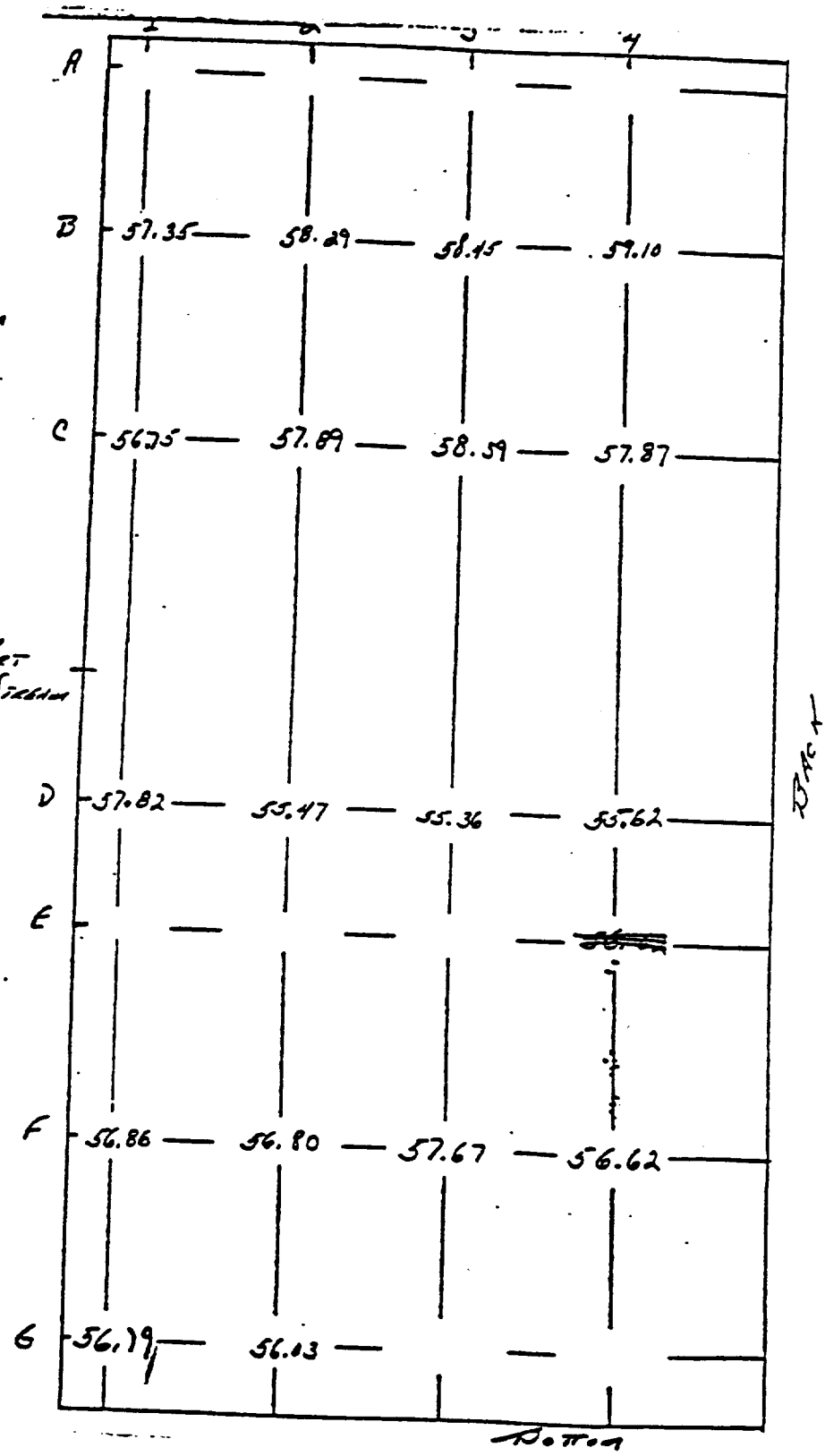
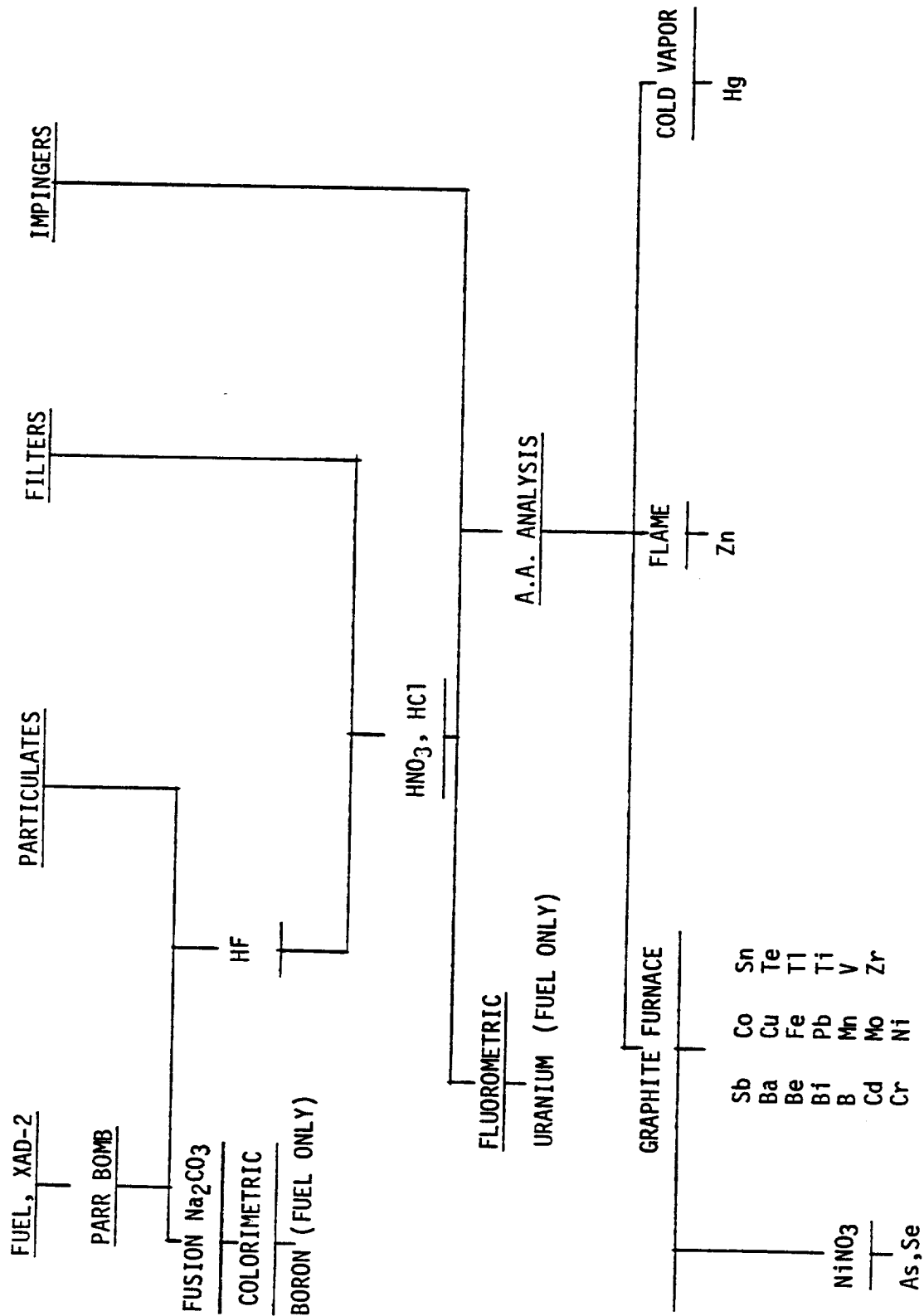


Figure 3.

APPENDIX H  
ANALYTICAL PROCEDURES

# ELEMENTAL ANALYSIS FLOWSHEET



APPENDIX I  
BIOASSAY RESULTS

LBI ASSAY NO. 3986

LBI SAFETY NO. 3643

MUTAGENICITY EVALUATION OF  
OIL FIRED GAS TURBINE NO. 1 SASS TRAIN  
XAD-2 EXTRACT (IN METHYLENE CHLORIDE)

IN THE  
AMES SALMONELLA/MICROSOME  
PLATE TEST

FINAL REPORT

SUBMITTED TO:

ACUREX CORPORATION  
485 CLYDE AVENUE  
MOUNTAIN VIEW, CA.  
94042

SUBMITTED BY:

LITTON BIONETICS, INC.  
5516 NICHOLSON LANE  
KENSINGTON, MARYLAND 20795

LBI PROJECT NO. 20962

REPORT DATE: JUNE 1979



BIONETICS

## PREFACE

This report contains a summary of the data compiled during the evaluation of the test compound. The report is organized to present the results in a concise and easily interpretable manner. The first part contains items I-IX. Items I-IV provide sponsor and compound identification information, type of assay, and the protocol reference number. All protocol references indicate a standard procedure described in the Litton Bionetics, Inc. "Screening Program for the Identification of Potential Mutagens and Carcinogens." Item V provides the initiation and completion dates for the study, and Item VI provides identification of supervisory personnel. Item VII identifies the tables and/or figures containing the data used by the study director in interpreting the test results. The interpretation itself is in Item VIII. Item IX provides the conclusion and evaluation.

The second part of the report, entitled PROTOCOL, describes the materials and procedures employed in conducting the assay. This part of the report also contains evaluation criteria used by the study director, and any appendices. The evaluation criteria are included to acquaint the sponsor with the methods used to develop and analyze the test results.

All test and control results presented in this report are supported by fully documented raw data which are permanently maintained in the files of the Department of Genetics and Cell Biology or in the archives of Litton Bionetics, Inc., 6616 Nicholson Lane, Kensington Maryland, 20795.



LITTON BIONETICS

- I. SPONSOR: Acurex Corporation
- II. MATERIAL (TEST COMPOUND): LBI ASSAY NUMBER 3986
  - A. Identification: Oil Fired Gas Turbine No. 1 Sass Train XAD-2 Extract (in Methylene Chloride)\*
  - B. Date Received: February 23, 1979
  - C. Physical Description: Clear yellow liquid
- III. TYPE OF ASSAY: Ames Salmonella/microsome Mutagenesis Assay
- IV. PROTOCOL NUMBER: 401
- V. STUDY DATES:
  - A. Initiation: May 18, 1979
  - B. Completion: June 8, 1979
- VI. SUPERVISORY PERSONNEL:
  - A. Study Director: D.R. Jagannath, Ph.D.
  - B. Laboratory Supervisor: Sibyl Goode
- VII. RESULTS:

The results of this assay are presented in Tables 1 and 2.
- VIII. INTERPRETATION OF RESULTS:

The test compound was examined for mutagenic activity in a series of in vitro microbial assays employing Salmonella indicator organisms. The compound was tested directly and in the presence of liver microsomal enzyme preparations from Aroclor-induced rats.

The compound was tested at four concentrations according to the IERL-RTP procedures Manual: Level I (1977). The compound was tested for its mutagenic activity as well as for its toxicity at 0.01 mg, 0.1 mg, 1.0 mg and 10 mg per plate.

The toxicity results presented in Table 1 indicate that the test compound was not toxic at the doses employed in these studies both in the presence and absence of metabolic activation systems.

\*See Sample Preparation and Handling.

VIII. INTERPRETATION OF RESULTS (continued):

The mutagenicity results presented in Table 2 indicate that the test compound did not induce any genetic activity in any of the test organisms employed in these assays. The results of tests conducted on the test compound in the presence of a rat liver activation system were also negative. The test with TA-1537 was repeated in nonactivation and activation assays because of high solvent values in the initial test. The repeat tests were also negative.

IX. CONCLUSIONS:

The test compound, Oil Fired Gas Turbine No. 1 Sass train XAD-2 extract (in methylene chloride) did not demonstrate genetic activity or toxicity in any of the assays conducted in this evaluation and was considered as not mutagenic under these test conditions.

Submitted by:

Study Director

*D.R. Jagannathan*

6.11.79

D.R. Jagannathan, Ph.D.  
Section Chief  
Submammalian Genetics  
Department of Genetics  
and Cell Biology

Reviewed by:

*David J. Brustock*

6/11/79

David J. Brustock, Ph.D.  
Director  
Department of Genetics  
and Cell Biology



BIONETICS

# V. RESULTS

TABLE I

A. NAME OR CODE DESIGNATION OF THE TEST COMPOUND: OIL FIRED GAS TURBINE NO1 SASS TRAIN XAD-2 EXTRACT (IN METHYLENE CHLORIDE)  
 B. SOLVENT: CP50  
 C. TEST INITIATION DATE: 05/10/79 09/30/79  
 D. TEST COMPLETION DATE: 06/08/79  
 E. 5-9 LUNG CHOC24  
 NOTE: CONCENTRATIONS ARE GIVEN IN MILLIGRAMS (MG) PER PLATE.

TEST	SPECIES	TISSUE	INDICATOR ORGANISMS (POPULATION/10 <sup>6</sup> )									
			TA-1515		TA-1517		TA-98		TA-100			
			1	2	1	2	1	2	1	2	1	2
MINIACTIVATION												
SOLVENT CONTROL	---	---	393	264	167	298	298	1853				
POSITIVE CONTROL**	---	---	315	159	91	298	1649					
TEST COMPOUND												
0.010000 MG	---	---	341	280	153	221	440					
0.100000 MG	---	---	362	246	194	280	391					
1.000000 MG	---	---	346	282	125	305	353					
10.000000 MG	---	---	326	267	177	354	353					
ACTIVATION												
SOLVENT CONTROL	RAT	LIVER	378	311	149	351	400					
POSITIVE CONTROL***	RAT	LIVER	551	191	101	312	331					
TEST COMPOUND												
0.010000 MG	RAT	LIVER	437	285	153	388	389					
0.100000 MG	RAT	LIVER	380	299	158	341	348					
1.000000 MG	RAT	LIVER	383	297	186	402	350					
10.000000 MG	RAT	LIVER	196	270	168	378	346					
***												
TA-1515	SODIUM AZIDE							1A-1535	2-ANTHRAMINE	2.5		
TA-1517	9-AMINACRIDINE							1A-1537	2-ANTHRAMINE	2.5		
TA-98	2-HEXACHLORINE							1A-98	2-ANTHRAMINE	2.5		
TA-100	SODIUM AZIDE							1A-100	2-ANTHRAMINE	2.5		
SOLVENT	5C HL/PLATE											

# V. RESULTS

TABLE 2

A. NAME OR CODE DESIGNATION OF THE TEST COMPOUND: DIE LIRED GAS TURBINE NO1 SASS TRAIN XAD-2 EXTRACT (IN METHYLENE CHLORIDE)  
 B. SOLVENT: CPMO  
 C. TEST INITIATION DATE: 05/18/79  
 D. TEST COMPLETION DATE: 06/08/79  
 E. S-9 LC1#:

NOTE: CONCENTRATIONS ARE GIVEN IN MILLIGRAMS (MG) PER PLATE

## INDICATOR ORGANISMS (REVERTANTS/PLATE)

TEST	SPECIES	TISSUE	IA-1535						IA-1537						IA-98						IA-100					
			1	2	1	2	1	2	1	2	1	2	1	2	1	2	1	2	1	2	1	2	1	2	1	2

### NONACTIVATION

SOLVENT CONTROL  
 POSITIVE CONTROL \*\*

TEST COMPOUND

0.010000 MG  
 0.100000 MG  
 1.000000 MG  
 10.000000 MG

### ACTIVATION

SOLVENT CONTROL  
 POSITIVE CONTROL \*\*\*

TEST COMPOUND

0.010000 MG  
 0.100000 MG  
 1.000000 MG  
 10.000000 MG

\*\*

IA-1535 SODIUM AZIDE  
 IA-1537 2-ANTHRACENE  
 IA-98 2-NITROFLUORENE  
 IA-100 SODIUM AZIDE  
 SOLVENT 50 UG/PLATE

1 UG/PLATE  
 50 UG/PLATE  
 10 UG/PLATE  
 1 UG/PLATE

\*\*\*

IA-1535 2-ANTHRACENE 2.5 UG/PLATE  
 IA-1537 2-ANTHRACENE 2.5 UG/PLATE  
 IA-98 2-ANTHRACENE 2.5 UG/PLATE  
 IA-100 2-ANTHRACENE 2.5 UG/PLATE

C. INDICATES CONTAMINATION

### SAMPLE PREPARATION AND HANDLING

The test material was received as a solution in 1.2 ml of methylene chloride and was stored at 4°C until solvent exchanged. The entire sample was exchanged into DMSO by first adding 2 ml DMSO and reducing the volume to 2 ml under a stream of nitrogen in a warm water bath (33°C). Then 0.5 ml DMSO was added and the solution evaporated again to 2 ml. This last process was repeated once more, leaving the sample in a final volume of 2.0 ml. This sample was stored at 4°C until use in the cytotoxicity assay. Since the original test sample represented 307 ft<sup>3</sup> of exhaust gas, the solvent exchanged sample corresponded to 153.5 ft<sup>3</sup> gas/ml or 4346.5 L gas/ml.

A solvent exchanged DMSO blank was also prepared by the above procedure, starting with 1.2 ml methylene chloride (same volume as the original test material). Since the test material did not exhibit any mutagenic or toxic effect on the indicator organism in these assays, solvent exchanged DMSO blank was not tested separately.



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PROTOCOL NO. 401

AMES SALMONELLA/MICROSOME PLATE ASSAY

1. OBJECTIVE

The objective of this study is to evaluate a test material for mutagenic activity in a bacterial assay with and without a mammalian S9 activation system.

2. RATIONALE

The Salmonella typhimurium strains used at LBI are all histidine auxotrophs by virtue of mutations in the histidine operon. When these histidine-dependent cells are grown in a minimal media petri plate containing a trace of histidine, only those cells that revert to histidine independence (his<sup>+</sup>) are able to form colonies. The trace amount of histidine allows all the plated bacteria to undergo a few divisions; this growth is essential for mutagenesis to occur. The his<sup>+</sup> revertants are easily scored as colonies against the slight background growth. The spontaneous mutation frequency of each strain is relatively constant; but when a mutagen is added to the agar, the mutation frequency is increased 2- to 100-fold. Cells which grow to form colonies on the minimal media petri plates are therefore assumed to have reverted, either spontaneously or by the action of a test substance to his- genotype.

3. MATERIALS

A. Indicator Microorganisms

The Salmonella typhimurium strains used in this assay were obtained from Dr. Bruce Ames, University of California at Berkeley.<sup>1-3</sup> The following 5 strains are routinely used.

Strain Designation	Gene Affected	Additional Mutations			Mutation Type Detected
		Repair	LPS	R Factor	
TA-1535	<u>his</u> G	Δ <u>uvr</u> B	<u>rfa</u>	-	Base-pair substitution
TA-1537	<u>his</u> C	Δ <u>uvr</u> B	<u>rfa</u>	-	Frameshift
TA-98	<u>his</u> D	Δ <u>uvr</u> B	<u>rfa</u>	pKM101	Frameshift
TA-100	<u>his</u> G	Δ <u>uvr</u> B	<u>rfa</u>	pKM101	Base-pair substitution



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All the above strains have, mutation in the histidine operon, mutation (rfa<sup>-</sup>) that leads to defective lipopolysaccharide coat, a deletion that covers genes involved in the synthesis of vitamin biotin (bio<sup>-</sup>) and in the repair of ultraviolet (uv) - induced DNA damage (uvrB<sup>-</sup>). The rfa<sup>-</sup> mutation makes the strains more permeable to many large molecules. The uvrB<sup>-</sup> mutation decreases repair of some types of chemically or physically damaged DNA and thereby enhances the strain's sensitivity to some mutagenic agents. The resistant transfer factor plasmid (R factor) pKM101, in TA-98 and TA-100 is believed to cause an increase in error-prone DNA repair that leads to many more mutations for a given dose of most mutagens<sup>5</sup>. In addition, plasmid pKM101 confers resistance to the antibiotic ampicillin, which is a convenient marker to detect the presence of plasmid in the cells.

All indicator strains are kept at 4°C on minimal medium plates supplemented with a trace of biotin and an excess of histidine. The plates with plasmid-carrying strains contain in addition ampicillin (25 µg/ml), to ensure stable maintenance of plasmid pKM101. New stock culture plates are made every two months from the frozen master cultures or from single colony reisolates that were checked for their genotypic characteristics (his, rfa, uvrB, bio) and for the presence of plasmid. For each experiment, an inoculum from the stock culture plates is grown overnight at 37°C in nutrient broth (Oxoid CM67) and used.

### B. Media

The bacterial strains were cultured in Oxoid Media #2 (nutrient Broth). The selective medium was Vogel Bonner Medium E with 2% glucose. The overlay agar will consist of 0.6% purified agar with 0.6 mM histidine, 0.05 mM biotin and 0.1M NaCl according to the methods of Ames et al.<sup>6</sup>

### C. Activation System

#### (1) S9 Homogenate

A 9,000 x g supernatant prepared from Sprague-Dawley adult male rat liver induced by Aroclor 1254 (described by Ames et al.<sup>6</sup>) was purchased from Bionetics Laboratory Products, Litton Bionetics, Inc. and used in this assay.

#### (2) S9 Mix

Components	Concentration per Milliliter S9 Mix
NADP (sodium salt)	1 µmoles
D-glucose-6-phosphate	3 µmoles
MgCl <sub>2</sub>	3 µmoles
KCl	33 µmoles
Sodium phosphate buffer pH 7.4	100 µmoles
Organ homogenate from rat liver (S9 fraction)	100 µliters



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#### 4. EXPERIMENTAL DESIGN

##### A. Dosage Selection

The tests are run at four concentrations according to the EPA Level I Manual. The recommended doses are 0.01, 0.1, 1.0 and 10 mg per plate. Both mutagenicity testing and toxicity testing are performed using these four doses.

##### B. Mutagenicity Testing

The procedure used is based on the paper published by Ames et al.<sup>6</sup> and is performed as follows:

###### (1) Nonactivation Assay

To a Sterile 13 x 100 mm test tube placed in a 43°C water bath the following is added in order:

- (a) 2.00 ml of 0.6% agar containing 0.05 mM histidine and 0.05 mM biotin.
- (b) 0.05 ml of a solution of the test chemical to give approximate dose.
- (c) 0.1 ml - 0.2 ml of indicator organism/s.
- (d) 0.50 ml of 0.01M phosphate buffer, pH 7.4.

This mixture is swirled gently and then poured into minimal agar plates (see 3B, Media). After the top agar has set, the plates are incubated at 37°C for approximately 2 days. The number of his<sup>+</sup> revertant colonies growing in the plates is counted and recorded.

###### (2) Activation Assay

The activation assay is run concurrently with the nonactivation assay. The only difference is the addition of 0.5 ml of S9 mix (see 3C:2, Activation System) to the tubes in place of 0.5 ml of phosphate buffer which is added in nonactivation assays. All other details are similar to the procedure for nonactivation assays.

A detailed flow diagram for the plate incorporation assay is provided in Figure 1.

#### C. Control Compounds

A negative control consisting of the solvent used for the test material is performed in all cases. For negative controls, step 'b' of Nonactivation Assays is replaced by 0.05 ml of the solvent. The negative controls are employed for each indicator strain and is performed in the absence and presence of S9 mix. The solvent used to prepare the stock solution of the test material is given in the Results section of this report. All dilutions of the test material made using this solvent.

Specific positive control compounds known to revert each strain are also used in the assays. The concentrations and specificities of these compounds to specific strains are given in the following table.

Assay	Chemical	Solvent	Concentration per Plate ( $\mu$ g)	<u>Salmonella</u> <u>Strains</u>
Nonactivation	Sodium azide	Water	1	TA-1538, TA-100
	2-Nitrofluorene (NF)	Dimethyl- sulfoxide	10	
	9-aminoacridine (9AA)	Ethanol	50	TA-1537
Activation	2-anthramine (ANTH)	Dimethyl- sulfoxide	2.5	For all strains

#### D. Toxicity Test

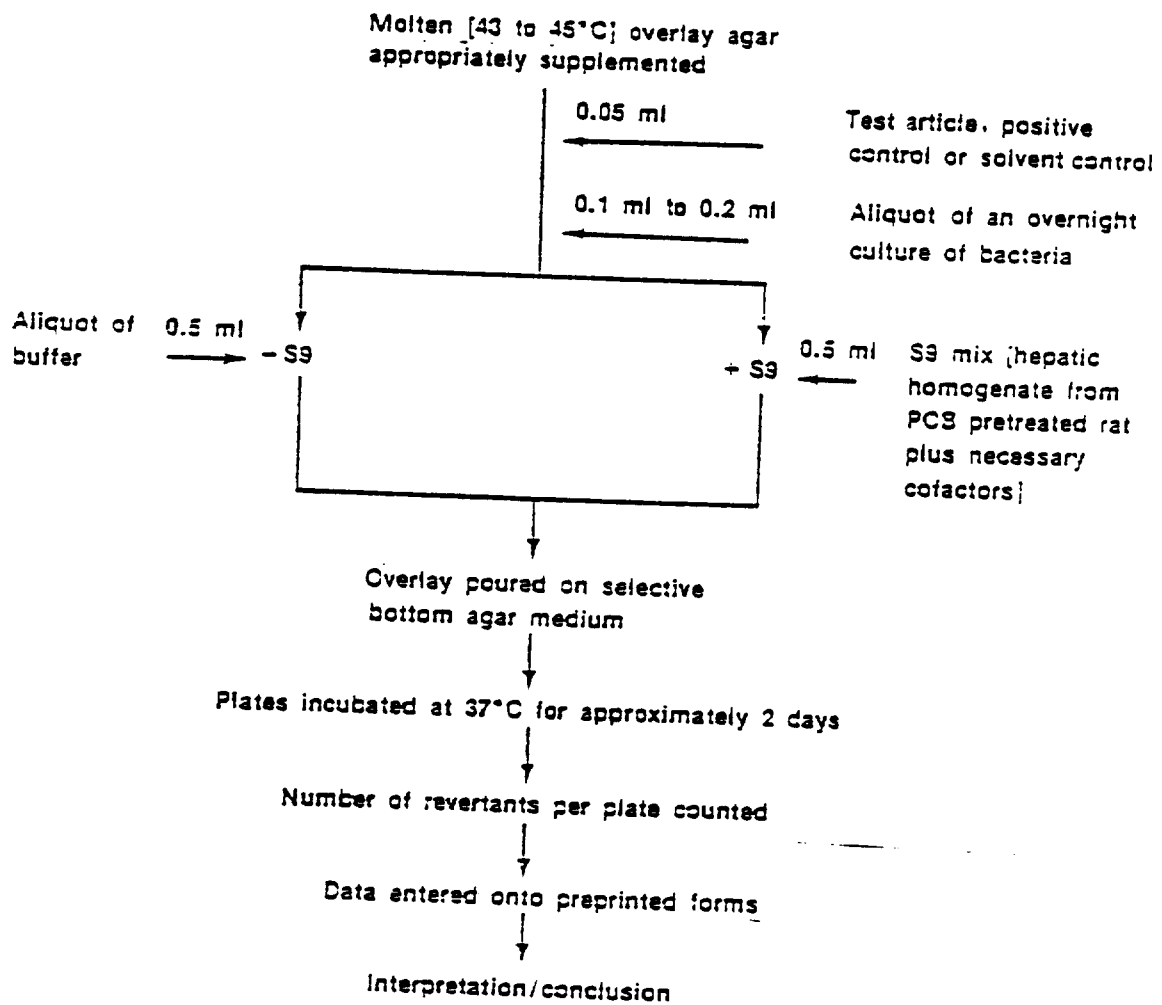
To a sterile 13 x 100 mm test tube placed in a 43°C water bath the following is added in order:

- (a) 2.0 ml of 0.6% agar containing 0.05 mM histidine and 0.05 mM biotin.
- (b) 0.05 ml of a solution of the test chemical to give approximate dose.
- (c) 0.1-0.2 ml of indicator cells (approximately 200 cells from an overnight culture appropriately dilute)
- (d) 0.50 ml of 0.01M phosphate buffer, pH 7.4 (for nonactivation assays) or S9 mix (see 3c:2) (for activation assays)

This mixture is swirled gently and then poured over the surface of nutrient agar plates. After the top agar has set, the plates are incubated at 37°C for 2 days. The number of colonies growing on the plates is counted and recorded.

FIGURE 1

REVERSE MUTATION ASSAY  
[Agar Incorporation Method]



## 5. EVALUATION CRITERIA

Statistical methods are not currently used, and evaluation is based on the criteria included in this protocol.

Plate test data consists of direct revertant colony counts obtained from a set of selective agar plates seeded with populations of mutant cells suspended in a semisolid overlay. Because the test material and the cells are incubated in the overlay for approximately 2 days and a few cell divisions occur during the incubation period, the test is semiquantitative in nature. Although these features of the assay reduce the quantitation of results, they provide certain advantages not contained in a quantitative suspension test:

- . The small number of cell divisions permits potential mutagens to act on replication DNA, which is often more sensitive than nonreplicating DNA.
- . The combined incubation of the test article and the cells in the overlay permits constant exposure of the indicator cells for approximately 2 days.

### A. Surviving Populations

Plate test procedures do not permit exact quantitation of the number of cells surviving chemical treatment. At low concentrations of the test material, the surviving population on the treatment plates is essentially the same as that on the negative control plate. At high concentrations, the surviving population is usually reduced by some fraction. Our protocol will normally employ several doses ranging over two or three log concentrations, the highest of these doses being selected to show slight toxicity as determined by subjective criteria.

### B. Dose-Response Phenomena

The demonstration of dose-related increases in mutant counts is an important criterion in establishing mutagenicity. A factor that might modify dose-response results for a mutagen would be the selection of doses that are too low (usually mutagenicity and toxicity are related). If the highest dose is far lower than a toxic concentration, no increases may be observed over the dose range selected. Conversely, if the lowest dose employed is highly cytotoxic, the test material may kill any mutants that are induced, and the test material will not appear to be mutagenic.



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### C. Control Tests

Positive and negative control assays will be conducted with each experiment and will consist of direct-acting mutagens for nonactivation assays and mutagens that require metabolic biotransformation in activation assays. Negative controls will consist of the test material solvent in the overlay agar together with the other essential components. The negative control plate for each strain will give a reference point to which the test data will be compared. The positive control assay will be conducted to demonstrate that the test systems are functional with known mutagens.

### D. Evaluation Criteria for Ames Assay

Because the procedures to be used to evaluate the mutagenicity of the test material are semiquantitative, the criteria to be used to determine positive effects are inherently subjective and are based primarily on a historical data base. Most data sets will be evaluated using the following criteria.

#### (1) Strains TA-1535, TA-1537

If the solvent control value is within the normal range, a test material that produces a positive dose response over three concentrations with the highest increase equal to three times the solvent control value will be considered to be mutagenic.

#### (2) Strains TA-98 and TA-100

If the solvent control value is within the normal range, a test material that produces a positive dose response over three concentrations with the highest increase equal to twice the solvent control value for TA-98 and TA-100 will be considered to be mutagenic.

#### (3) Pattern

Because TA-1535 and TA-100 are both derived from the same parental strain (G-46) and because TA-1538 and TA-98 are both derived from the same parental strain (DS052), to some extent there is a built-in redundancy in the microbial assay. In general, the two strains of a set respond to the same mutagen and such a pattern is sought. Generally, if a strain responds to a mutagen in nonactivation tests, it will do so in activation tests.



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#### (4) Reproducibility

If a test material produces a response in a single test that cannot be reproduced in additional runs, the initial positive test data lose significance.

The preceding criteria are not absolute, and other extenuating factors may enter into a final evaluation decision. However, these criteria will be applied to the majority of situations and are presented to aid those individuals not familiar with this procedure. As the data base is increased, the criteria for evaluation can be more firmly established.

#### E. Relation between Mutagenicity and Carcinogenicity

It must be emphasized that the Ames Salmonella/Microsome Plate Assay is not a definitive test for chemical carcinogens. It is recognized, however, that correlative and functional relations have been demonstrated between these two endpoints. The results of comparative tests on 300 chemicals by McCann et al. show an extremely good correlation between results of microbial mutagenesis tests and in vivo rodent carcinogenesis assays.

All evaluations and interpretation of the data to be presented in the final report will be based only on the demonstration, or lack, of mutagenic activity.

## REFERENCES

1. J. McCann, E. Choi, E. Yamasaki, and B.N. Ames. Detection of carcinogens as mutagens in the Salmonella/microsome test: Assay of 300 chemicals. Proc. Nat. Acad. Sci. USA 72, 5135-5139 (1975).
2. B.N. Ames, E.G. Gurney, J.A. Miller, and H. Bartsch. Carcinogens as frameshift mutagens: Metabolites and derivatives of 2-acetylaminofluorene and other aromatic amine carcinogens. Proc. Nat. Acad. Sci. USA 69, 3128-3132 (1972).
3. B.N. Ames, F.D. Lee, and W.E. Durston. An improved bacterial test system for the detection and classification of mutagens and carcinogens. Proc. Nat. Acad. Sci. USA 70, 782-786 (1973).
4. B.N. Ames, W.E. Durston, E. Yamasaki, and F.D. Lee. Carcinogens are mutagens: A simple test system combining liver homogenates for activation and bacteria for detection. Proc. Nat. Acad. Sci. USA 70, 2231-2235 (1973).
5. J. McCann, N.E. Springarn, J. Kobori, and B.N. Ames. Detection of carcinogens as mutagens: Bacterial tester strains with R factor plasmids. Proc. Nat. Acad. Sci. USA 72, 979-983 (1975).
6. B.N. Ames, J. McCann, and E. Yamasaki. Methods for detecting carcinogens and mutagens with the Salmonella/mammalian-microsome mutagenicity test. Mutation Res. 31, 347-364 (1975).
7. H.J. Vogel and D.M. Bonner. Acetylornithinase of E. coli: partial purification and some properties. J. Biol. Chem., 213, 97-106 (1955).



LIFTONETICS

LBI ASSAY NO. 3986

LBI SAFETY NO. 3643

CYTOTOXIC EVALUATION OF  
OIL FIRED GAS TURBINE NO. 1  
SASS TRAIN XAD-2 EXTRACT

IN THE  
WI-38 HUMAN CELL  
CYTOTOXICITY ASSAY

FINAL REPORT

SUBMITTED TO:

ACUREX CORPORATION  
485 CLYDE AVE.  
MOUNTAIN VIEW, CA 94042

SUBMITTED BY:

LITTON BIONETICS, INC.  
5516 NICHOLSON LANE  
KENSINGTON, MARYLAND 20795

LBI PROJECT NO. 20993

REPORT DATE: JUNE, 1979



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## PREFACE

This report contains a summary of the data compiled during the evaluation of the test compound. The report is organized to present the results in a concise and easily interpretable manner. The first part contains items I-IX. Items I-IV provide sponsor and compound identification information, type of assay, and the protocol reference number. All protocol references indicate a standard procedure described in the Litton Bionetics, Inc. "Screening Program for the Identification of Potential Mutagens and Carcinogens." Item V provides the initiation and completion dates for the study, and Item VI provides identification of supervisory personnel. Item VII identifies the tables and figures containing the data used by the study director in interpreting the test results. The interpretation itself is in Item VIII. Item IX provides the conclusion and evaluation.

The second part of the report, entitled PROTOCOL, describes the materials and procedures employed in conducting the assay. This part of the report also contains evaluation criteria used by the study director, and any appendices. The evaluation criteria are included to acquaint the sponsor with the methods used to develop and analyze the test results.

All test and control results presented in this report are supported by fully documented raw data which are permanently maintained in the files of the Department of Genetics and Cell Biology or in the archives of Litton Bionetics, Inc., 5516 Nicholson Lane, Kensington Maryland, 20795.

Copies of raw data will be supplied to the sponsor upon request.



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- I. SPONSOR: ACUREX CORPORATION
- II. MATERIAL (TEST COMPOUND): LBI ASSAY NUMBER 3986
  - A. Identification: Oil Fired Gas Turbine No. 1, SASS Train XAD-2 Extract
  - B. Date Received: February 23, 1979
  - C. Physical Description: Light yellow solution in DMSO
- III. TYPE OF ASSAY: WI-38 Human Cell Cytotoxicity Assay
- IV. PROTOCOL NUMBER: Special Protocol
- V. STUDY DATES:
  - A. Initiation: May 29, 1979
  - B. Completion: June 5, 1979
- VI. SUPERVISORY PERSONNEL:
  - A. Study Director: Brian C. Myhr, Ph.D.
  - B. Laboratory Supervisor: Robert Young
- VII. RESULTS:

The data are presented in Table 1 on page 3 and in Figures 1 and 2 on pages 4 and 5.
- VIII. INTERPRETATION OF RESULTS:

The methylene chloride extract of the test sample on XAD-2 resin, after solvent exchange into DMSO, appeared to remain soluble in the culture medium at the highest assayed concentration of 20  $\mu$ l/ml. Higher concentrations could not be tested because of the introduction of greater than 2% organic solvent by volume. As shown in Table 1, 1% DMSO reduced the viability index, total protein, and total ATP to about 70-80% of the untreated negative control; 2% DMSO reduced these parameters even further to about 40-65%. The corresponding concentrations of solvent exchanged DMSO were somewhat less toxic to these assay parameters, showing that residual methylene chloride does not contribute to the solvent toxicity. Because of the solvent toxicity, the effect of the test material was measured relative to the assay parameters obtained for the appropriate solvent exchanged DMSO negative control.



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VIII. INTERPRETATION OF RESULTS (continued):

The most responsive assay parameter appeared to be the viability index, although the protein and ATP contents started to decrease similarly at the highest dose of 20  $\mu$ l/ml. The percent viability and ATP per  $10^6$  cells parameters gave no indication of any toxicity. A 50% reduction was not achieved for any assay parameter, but the curve for the viability index (Figure 1) indicated that an EC50 would occur near 35  $\mu$ l/ml. In terms of the volume of exhaust gas represented by the DMSO test solution (4346.5 L gas/ml), this EC50 corresponds to 152.1 L gas/ml. Therefore, on the basis of the viability index and expectations for the ATP and protein parameters, the test material appears to yield EC50 values in the low toxicity region (100 L/ml to 1000 L/ml).

IX. CONCLUSIONS:

The test material, SASS train XAD-2 Extract, Oil Fired Gas Turbine No. 1, is evaluated as having low toxicity to WI-38 human cells. The viability index indicated an EC50 value would be obtained near 152 L gas/ml, and the ATP and protein contents were decreasing in the same toxicity range.

Submitted by:

Study Director

Brian Myhr 6/11/79  
Brian Myhr, Ph.D. date  
Section Chief  
Mammalian Genetics  
Department of Genetics  
and Cell Biology

Reviewed by:

Robert J. Brusick (for) 6/11/79  
David J. Brusick, Ph.D. date  
Director  
Department of Genetics  
and Cell Biology



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### WI-38 HUMAN CELL CYTOTOXICITY ASSAY

**Test Material Identity:** Oil Fired Gas Turbine No. 1,  
SASS Train XAD-2 Extract

Passage number: 28

Vehicle: DMSO/growth medium

TEST RESULTS											
Sample	Concentration <sup>††</sup> μl/ml	Average Values per Culture Flask				ATP Per 10 <sup>6</sup> Cells 10 <sup>6</sup> fg	Expressed as Percent of Negative Control (NC) <sup>***</sup>				
		Viable Cells 10 <sup>6</sup> Units	Total Cells 10 <sup>6</sup> Units	Cellular Protein μg	ATP 10 <sup>6</sup> fg		Viability %	Viability Index	Protein	ATP	ATP Per 10 <sup>6</sup> Cells
NEGATIVE CONTROL ---											
		.602	.610	145	50.4	82.6	98.7%	100.0	100.0	100.0	100.0
1% DMSO BLANK S.E. <sup>**</sup>		.448	.449	101	43.3	96.4	99.8	101.1	69.7	85.9	116.7
1% DMSO†		.446	.446	98	38.9	87.2	100.0	101.3	67.6	77.2	105.6
2% DMSO BLANK S.E. <sup>**</sup>		.463	.481	80	33.1	68.8	96.3	97.6	55.2	65.7	83.3
2% DMSO		.397	.406	55	27.3	67.2	97.8	99.1	37.9	54.2	81.4
TEST 0.5		.467	.473	100	43.3	91.5	98.7	98.9	99.0	100.0	94.9
TEST 2.0		.453	.454	102	39.0	85.9	99.8	100.0	101.0	90.1	89.1
TEST 5.0		.420	.425	91	39.4	92.7	98.8	99.0	90.1	91.0	96.2
TEST 10.0		.411	.420	79	38.9	92.6	97.9	98.1	78.2	89.8	96.1
TEST 20.0		.325	.333	61	27.5	82.6	97.6	101.3	76.3	83.1	120.1
†† If change in culture medium: None observed											
** 1.2 ml methylene chloride solvent exchanged to 2.0 ml DMSO											
*** or appropriate DMSO blank											
							>20.0	>20.0	>20.0	>20.0	>20.0
							(35)				

<sup>a</sup>Determined from data plots in Figures 1 and 2

**Toxicity**  
**Classification: Low toxicity**

FIGURE 1

EC50 DETERMINATION FOR  
PERCENT VIABILITY (O) AND VIABILITY INDEX (●)

OIL FIRED GAS TURBINE NO. 1

SASS TRAIN XAD-2 EXTRACT

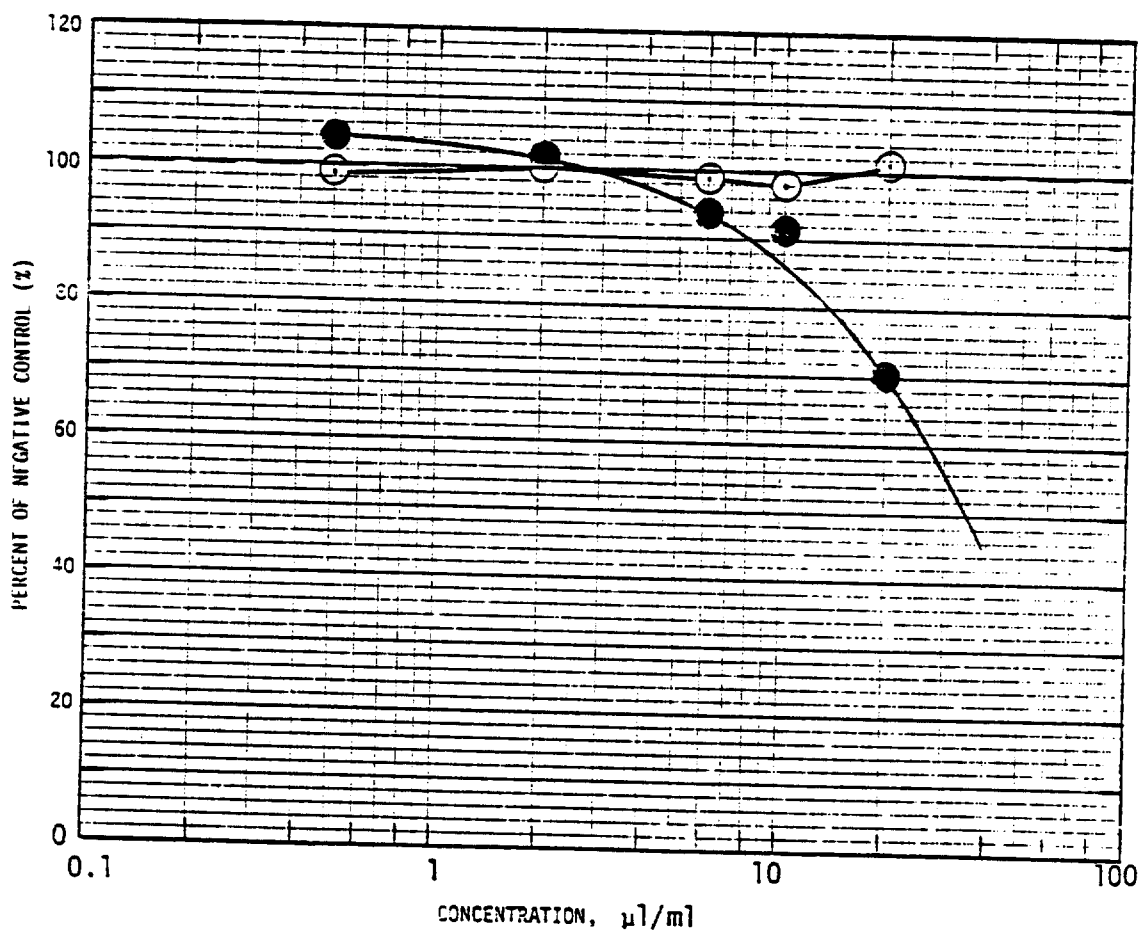
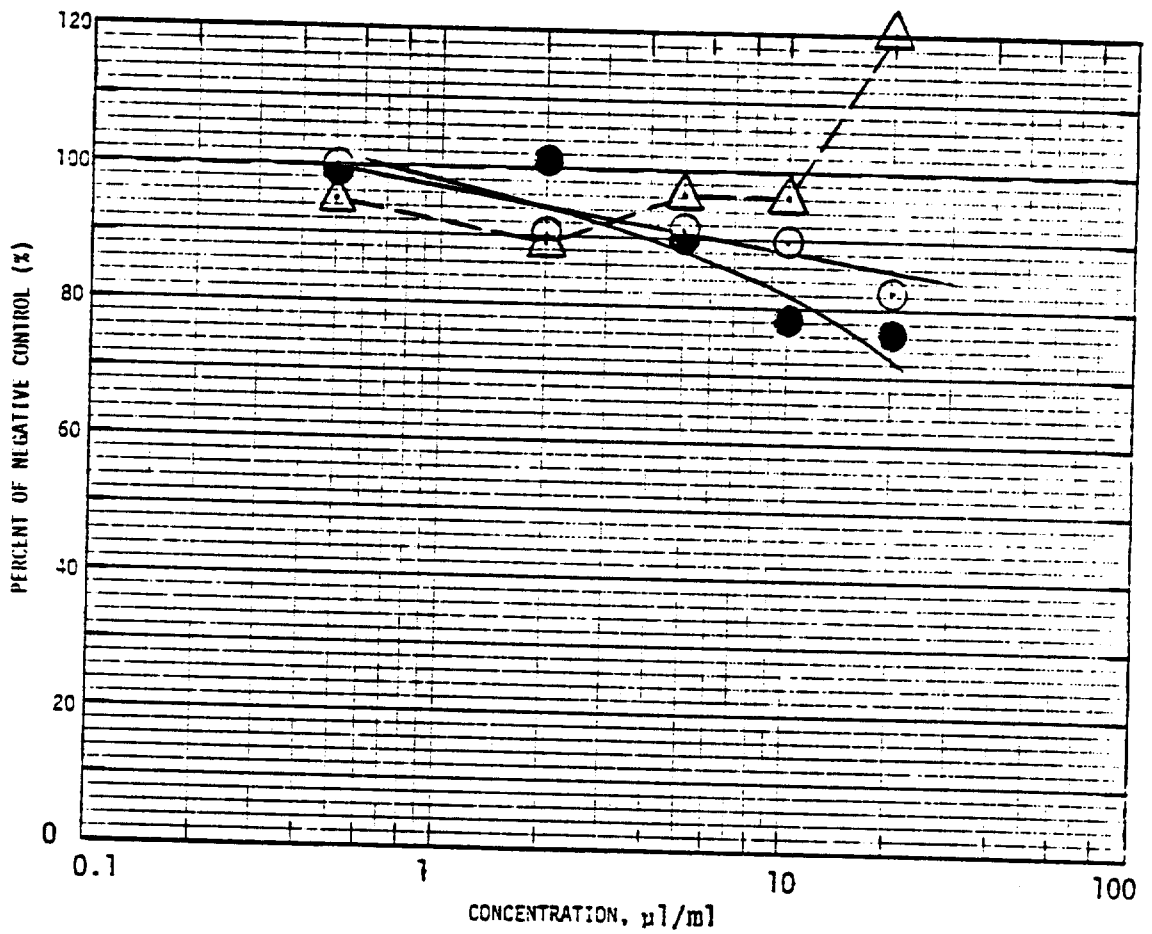


FIGURE 2

EC50 DETERMINATION FOR  
PROTEIN (●), ATP (○), and ATP/10<sup>6</sup> CELLS (Δ)

OIL FIRED GAS TURBINE NO. 1

SASS TRAIN XAD-2 EXTRACT



## ASSAY PROTOCOL

### 1. OBJECTIVE

The objective of this assay is to determine the concentrations of test material that reduces by 50% the number of viable cells, the cellular protein, and the ATP content after a 20 hour exposure. These concentrations are referred to as the EC50 values for each measured parameter.

### 2. MATERIALS

#### A. Indicator Cells

The indicator cells used for this study were WI-38 human embryonic lung fibroblasts obtained from Flow Laboratories, Inc., Rockville, Maryland. The cells were supplied as confluent monolayers at passage numbers 23 or 24 in Eagle's Minimum Essential Medium. This test system is specified by the Environmental Protection Agency's Level 1 Environmental Assessment Program.<sup>1</sup>

#### B. Medium and Cell Maintenance

The cells were maintained and treated in Basal Medium Eagle (BME) supplemented with 10% fetal bovine serum, 2mM L-glutamine, 100 units/ml penicillin, 100 µg/ml streptomycin, and 1.0 µg/ml amphotericin B (Fungizone). Subcultures were prepared twice weekly at a 1:2 split ratio using 0.25% trypsin. Cultures were discarded after the 35th subculture (passage).

#### C. Negative Controls

Five sets of negative control cultures, each in triplicate, were carried through the same experimental time period as the treated cells. One set was an untreated negative control consisting of cultures exposed only to BME culture medium. Two sets were solvent controls containing 1% and 2% of the solvent-exchanged DMSO blank, prepared as described below. In addition, two solvent control sets containing 1% and 2% pure DMSO were assayed in order to determine whether residual methylene chloride in the solvent-exchanged blank was contributing to solvent toxicity. The average viability, ATP content, and protein content of the solvent-exchanged negative controls provided the reference points for determining the effects of different concentrations of the test material on the assay parameters. The 2% solvent-exchanged control was the reference for the highest assayed concentration (20 µl/ml) and the 1% solvent-exchanged control was the reference for the remaining test concentrations.



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## 2. MATERIALS (Continued)

### D. Sample Preparation and Handling

The test material was received as a solution in 1.2 ml of methylene chloride and was stored at 4°C until solvent exchanged. The entire sample was exchanged into DMSO by first adding 2 ml DMSO and reducing the volume to 2 ml under a stream of nitrogen in a warm water bath (33°C). Then 0.5 ml DMSO was added and the solution evaporated again to 2 ml. This last process was repeated once more, leaving the sample in a final volume of 2.0 ml. This sample was stored at 4°C until use in the cytotoxicity assay. Since the original test sample represented 307 ft<sup>3</sup> of exhaust gas, the solvent exchanged sample corresponded to 153.5 ft<sup>3</sup> gas/ml or 4346.5 L gas/ml.

A solvent exchanged DMSO blank was also prepared by the above procedure, starting with 1.2 ml methylene chloride (same volume as the original test material).

## 3. EXPERIMENTAL DESIGN

### A. Dose Selection

The solvent exchanged sample was tested from 20 µl/ml to 0.5 µl/ml in five dose steps. The 20 µl/ml treatment was the maximum dose because of the introduction of 2% DMSO in the growth medium. All other concentrations were obtained by 1:100 dilutions of the test sample and dilutions thereof (using DMSO) into the growth medium to give a 1% final concentration of solvent.

### B. Culture Preparation

Stock cultures were subcultured into 100-mm plastic culture dishes 24 to 72 hours prior to use. This procedure provides a population of actively growing, sub-confluent cells to initiate the assay.

The cells were then suspended in BME culture medium by treatment with 0.25% trypsin for 3-5 minutes and the cell number determined by hemocytometer. A series of 25-cm<sup>2</sup> culture flasks were seeded with  $20 \times 10^4$  cells and 4 ml culture medium per flask. The cultures were incubated overnight at 37°C in a humidified atmosphere containing 5% CO<sub>2</sub> to allow attachment of the cells and resumption of growth.



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### 3. EXPERIMENTAL DESIGN (Continued)

#### C. Treatment

The medium was aspirated from the cultures and 4 ml of BME culture medium containing the test sample was applied. Three cultures were exposed to each test concentration and solvent exchanged DMSO blank. The flasks were then placed on a rocker platform in a 37°C incubator with a humidified atmosphere containing 5% CO<sub>2</sub>. The flasks were slowly rocked for a 20 hour exposure period. Any color changes in the culture medium caused by the test material were noted and the pH determined in additional treated flasks.

#### D. Cell Viability Assay

At the end of the treatment period, the medium containing unattached cells was decanted into a centrifuge tube on ice. The cell monolayer was washed with 1 ml 0.05% trypsin/versene and this wash combined with the decanted media. The attached cells were then removed with 2 ml of 0.25% trypsin at 37°C and the suspended cells combined with the decantate. The cells from each flask were thereby resuspended in 7 ml volumes for subsequent analysis.

A 1.0 ml aliquot was removed for cell count and viability determination. The aliquot was combined with 0.2 ml or 0.5 ml of 0.4% trypan blue and counted by hemocytometer about 5 to 15 minutes later. Between 60 and 154 cells were counted per flask and the number of live (colorless) and dead (blue) cells were recorded.

#### E. ATP Assay

ATP was immediately analyzed by extraction of a 0.1 ml cell suspension sample with 0.9 ml of 90% DMSO. After 2 minutes at room temperature, 5.0 ml cold MOPS buffer (0.01 M morpholinopropane sulfonic acid) at pH 7.4 was added and the extract was vortexed and placed on ice. Aliquots of 10 µl were injected into a cuvette containing a luciferin-luciferase reaction mixture in a DuPont Model 760 Luminescence Biometer. The Biometer was calibrated with standard ATP solutions to provide a direct read-out of the ATP content. Each test sample was assayed three times to demonstrate consistent readings.

#### F. Lowry Protein Assay<sup>2</sup>

A 3.0 ml aliquot of the cell suspension was taken for protein analysis by the Lowry method. The aliquot was centrifuged at 365 x g for 10 minutes, the medium decanted, and the cell pellet resuspended in 3 ml PBS. After two additional centrifugation washes with PBS, the pellet was resuspended in 1.5 ml of PBS and frozen at -20°C or analyzed immediately. A 1.0 ml aliquot was used for the Lowry assay.



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3. EXPERIMENTAL DESIGN (Continued)

At the end of the color development period, the tubes were centrifuged to remove any particulate test material prior to making absorbance readings at 750 nm. Lowry protein standard curves were constructed with bovine serum albumin for each assay.

4. REFERENCES

<sup>1</sup>IERL-RTP Procedures Manual: Level I. EPA-600/7-77-043, April 1977.

<sup>2</sup>Lowry, O.H., Rosebrough, N.J., Farr, A.L., and Randall, R.J.: Protein Measurement with Folin Phenol Reagent. J.Bio.Chem., 193:265-275, 1951.



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## ASSAY ACCEPTANCE CRITERIA

The assay will be considered acceptable for evaluation of the test results if the following criteria are met:

1. The passage level of the cells (number of subcultures) prior to use in the assay does not exceed 35.
2. The percent viability of the WI-38 cells used to initiate the assay is 95% or greater.
3. At least  $1.5 \times 10^5$  cells are seeded per flask. The untreated negative control cultures must increase in cell number by at least 2-fold over the 20 hour treatment period.
4. A sufficient number of data points (for five test concentrations or less) are available to clearly locate the EC50 of the most sensitive test parameter within a toxicity region as defined under Evaluation Criteria.
5. The data points critical to the location of the EC50 for the most sensitive parameter are the averages of at least two treated cultures.
6. If all the test parameters yield EC50 values greater than 1000  $\mu\text{g/ml}$  or 600  $\mu\text{l/ml}$ , the plotted curves for any parameter will not exceed 120% of the negative control.



## ASSAY EVALUATION CRITERIA

The EC50 value represents the concentration of test material that reduces an assay parameter to 50% of the negative control value. EC50 values are determined graphically by fitting a curve by eye through the data points associated with each test parameter plotted as a function of the logarithm of the applied concentration. Each point normally represents the average of three culture flasks for each treatment. Statistical analysis is unnecessary in most cases for evaluation.

The evaluation of the test material is based upon determinations of the EC50 values for five parameters: percent viability (ratio of viable cells to total cells x 100% for each treatment), viability index (ratio of viable cells for each treatment to viable cells in the negative control x 100%), cellular protein, total ATP content, and ATP per  $10^6$  cells. Except for the ATP content, these parameters are specified in the EPA Procedures Manual.<sup>1</sup> The ATP content will generally be a more sensitive parameter than ATP per  $10^6$  cells because any cell loss due to treatment will increase the latter parameter. ATP released into the growth medium by disrupted cells contributes to the ATP measurement.

The toxicity of the test material is evaluated as high, moderate, low or nondetectable according to the ranges of EC50 values defined in the following table. The actual concentration of extract at the EC50 is converted to the equivalent volume of exhaust gas per milliliter of culture medium prior to the evaluation. The assay parameter yielding the lowest EC50 will classify the test material.

Toxicity*	EC50 Values
High	EC50 < 10 L gas/ml
Moderate	EC50 range of 10-100 L gas/ml
Low	EC50 range of 100-1000 L gas/ml
Nondetectable	EC50 > 1000 L gas/ml

\*Formulated by Litton Bionetics, Inc., under contract to the Environmental Protection Agency, Contract No. 68-02-2681.



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<b>TECHNICAL REPORT DATA</b> <i>(Please read instructions on the reverse before completing)</i>		
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16. ABSTRACT <b>The report gives methods and results of an environmental assessment test program at Houston Lighting and Power's T. H. Wharton Generating Station, Unit 52. The aim of the program was to measure emissions changes resulting from applying NOx controls. Emissions of trace elements, organic materials, sulfur species, SO2, NOx, CO, and particulate matter were measured. These emissions, under normal and controlled (for NOx) operating conditions, were compared. Source operating data were also analyzed so that changes in operating parameters and efficiency could be assessed. Unit 52 is a General Electric MS 7001C simple-cycle, single-shaft, heavy duty gas turbine, rated at 70.8 MW nominal electrical output. This gas turbine may use either natural gas or distillate oil fuels. The test program was conducted using oil fuel. Water injection was used for NOx control. A water-to-fuel ratio of 0.42 reduced NOx by 58% from baseline levels. The unit heat rate showed about 2% change in going from baseline to controlled (for NOx) operation. Test results indicate that using water injection for NOx control in this unit reduced NOx and showed little effect on other emissions. Water injection reduced operating efficiency.</b>		
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