

Note: This is a reference cited in AP 42, *Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

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Background Report Reference

AP-42 Section Number: 8.13

Background Chapter: 4

Reference Number: 8

Title: Compliance Test Report: Amoco Oil
Company

Southwestern Laboratories, Inc.

May 1989

reslan

Amoco Oil

R7-232
Rec'd
7/14/89



**SULFUR DIOXIDE
COMPLIANCE EMISSIONS TEST OF
THE SULFUR RECOVERY UNIT
AMOCO OIL COMPANY**

**AMOCO OIL COMPANY
TEXAS CITY, TEXAS**

May 4, 1989

**SwL Customer No. 2-0470-20
SwL Project No. 54-8905-253**

**This study was conducted by the
Environmental/Analytical Services Division
of Southwestern Laboratories, Inc.
Houston, Texas**



SOUTHWESTERN LABORATORIES



Materials, environmental and geotechnical engineering, nondestructive, metallurgical and analytical services

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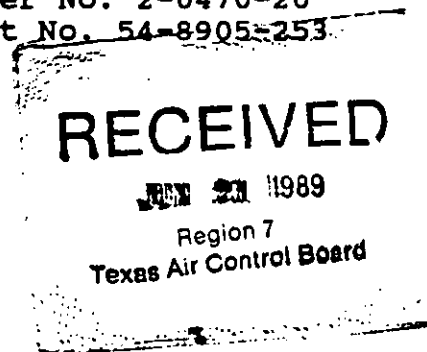
June 12, 1989

Re: SOx Emissions Sampling of SRU
Texas City, Texas
SwL Customer No. 2-0470-20
SwL Project No. 54-8905-253

AMOCO OIL COMPANY
P. O. Box 401
Texas City, Texas 77590

Attention: Mr. Watson Dupont

Gentlemen:



In accordance with our agreement, Southwestern Laboratories, Inc. hereby transmits our test report covering the testing performed at your Texas City, Texas facility. Field sampling was conducted May 4, 1989.

This report is for the exclusive use of AMOCO OIL COMPANY, and except for submission to regulatory agencies, the use of our name relative to the report must receive prior written approval.

It has been a pleasure working with you and your personnel at your Texas City, Texas facility once again. Please let us know if we may be of further service, or if you have any questions concerning this report.

Sincerely,

SOUTHWESTERN LABORATORIES, INC.

Phillip W. Yokley
Project Manager

Russell J. DiRaimo, P.E.
Manager
Environmental/Analytical Services

RJD:pm

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Texas Air Control Board

INTRODUCTION

Field sampling was conducted May 4, 1989 to determine sulfur dioxide (SO₂) emissions from the Sulfur Recovery Unit (SRU) stack at the Texas City, Texas facility of AMOCO OIL COMPANY (AMOCO). Sampling was performed by personnel of Southwestern Laboratories (SWL) Environmental/Analytical Services (EAS) Division. Personnel involved in the sampling program included: Mr. Watson Dupont and Ms. Sue Edrozo of AMOCO OIL COMPANY, and Messrs. Phillip Yokley, Robbie Daughtry, and Doug Belcher of Southwestern Laboratories, Inc. (SWL).

Sampling was conducted on the SRU in accordance with requirements contained in the Texas Air Control Board (TACB) Permit No. R-8811.

RESULTS

Results calculated in accordance with EPA and TACB procedures are shown in the Tables section of this report.

Pertinent test data are described below:

<u>Parameter</u>	<u>Average Emissions</u>	<u>Permit Allowable</u>	<u>% of • Allowable</u>
SO ₂	108 ppm (dry, O ₂ free)	250 ppm (dry, O ₂ free)	43.2
SO ₂	26.29 lb/hr	161.9 lb/hr	16.2
O ₂	2.1%	---	--
Flow	1,624,786 dscf/hr	---	--

Sulfur production was 406 long tons/day for the May 4, 1989 test period. Three trains were operating through one SCOT tail gas unit

PROCEDURES

Sampling equipment and procedures were in conformity, except where noted, with Reference Methods 1, 2, 3, 4 and 6 of the EPA (as described in 40 CFR 60, "Standards of Performance for New Stationary Sources", Appendix A - Reference Methods) and Chapter 6 of the TACB Compliance Sampling Manual (revised January, 1983). A pretest conference was held at AMOCO on April 18, 1989 to discuss test methodologies. TACB, AMOCO and SWL representatives were present at this meeting. All testing was conducted in accordance with agreements reached at this meeting.

Sample and Velocity Traverses - Method 1

The SRU exhaust stack was 7.5 feet internal diameter (ID). Two (2) four-inch flanged sample ports, 90 degrees apart, were present. Upstream distance from flow disturbance (top of stack) to sample ports (Distance A) was in excess of 2.0 stack diameters. Downstream distance from flow disturbance (top of entry duct) to sample ports (Distance B) was 81.25 feet (in excess of 8.0 stack diameters). It was determined that a

twelve (12) point traverse for velocity measurements (six points from each port) would be appropriate.

Determination of Stack Gas Velocity and Volumetric Flow Rate

- Method 2

Stack gas velocity was measured with an "S" type pitot tube constructed in accordance with "Proper Pitot Tube Sampling Nozzle Configuration", as specified in the Environmental Protection Agency, "Standards of Performance for New Stationary Sources - Revision to Reference Methods 1-8 (FR Thursday, August 18, 1977, Part II)". The pitot tube coefficient value was determined by wind tunnel calibration. Temperature measurements were determined by means of a calibrated digital thermometer with a Type "K" thermocouple. A ten (10) point velocity-stack temperature traverse was obtained for each pollutant test run. A ten (10) point traverse became necessary due to limited platform space and obstructions which prevented the use of a longer sample probe. Therefore, the last point on each traverse was unattainable. This criteria was discussed and approved at the Pretest Conference.

Leak checks of the pitot system were performed before and after the velocity tests. In addition, cyclonic flow checks were performed to ensure that the flow was parallel to the stack wall.

Gas Analysis and Molecular Weight Determination - Method 3

Integrated samples were taken during each SOx sample run and analyzed for CO₂ and O₂ by use of a standard Orsat analyzer. N₂ was determined by difference. Analysis was performed at the field site immediately after sampling. This data was utilized in stack gas molecular weight determinations used in flow rate calculations and also for oxygen correction purposes (SO₂ data) in accordance with requirements in 40 CFR 60, Subpart J.

Determination of Moisture Content in Stack Gases - Method 4

Moisture content of the stack gas was determined by gravimetric analysis of the impinger catch from each SOx sample run.

Determination of Sulfur Oxides in Stack Gas - EPA Method 6, and TACB Standard Method - Chapter 6

A sampling train similar to the one described in APTD-0581, complete with 8 foot long, stainless steel lined probe, was used to collect SOx samples (See Figure #2). The first impinger contained 100 ml of 80% isopropanol, the second and third impingers each contained 100 ml of 6% hydrogen peroxide (H₂O₂), and the fourth impinger contained a known weight of silica gel.

Three SO₂ compliance sample runs were obtained May 4, 1989. Each sample run was of thirty (30) minute duration with approximately 11.36 dry standard cubic feet (dscf) of stack gas sampled. Samples were obtained proportionally from the midpoint of the stack, at an approximate sample rate of 0.38 dscf/minute. At the conclusion of each sample run, an ambient air purge of approximately 2 cubic feet was performed, in order to sweep any residual SO₂ contained in the first impinger into the 6% H₂O₂ absorbing solution contained in the second and third impingers.

Prior to and following each sample run, a leak check of the sampling system was performed. After each sample run, all glassware was sealed to prevent contamination of the samples.

SAMPLE RECOVERY

Sulfur Dioxide

Analysis for SO₂ was performed in accordance with TACB procedures (as contained in Chapter 29, of the TACB's "Laboratory Methods for Determination of Air Pollutants," modified July 24, 1979), employing Alcoholic Barium Perchlorate as the titrant. Calculations and analytical data sheets are included with this report in Item IV, Field and Laboratory Data Sheets.

Oxygen

Stack gas O₂ determinations were made at the field site utilizing an Orsat analyzer.

CUSTODY OF SAMPLES

After completion of tests, each sample was placed in the custody of the technician responsible for analysis. It was his assigned responsibility to ensure that each sample was recorded and correctly analyzed. Analysis of samples was performed at the field site or in Southwestern Laboratories' facilities by Environmental/Analytical Services' personnel. It was the duty of the Department Manager and the Project Manager to answer any procedural queries from the Laboratory Technician. Final responsibility rested with the Department Manager.

QUALITY ASSURANCE

Southwestern Laboratories maintains a strict quality assurance program. Included in this program are: equipment calibration checks such as the meter box gamma check, performed after each field use; balance calibration checks; analytical calibration checks; and calculation checks. Also included in SWL's Quality Assurance program is its participation in the EPA Inter-Laboratory Source Survey for Methods 3, 5, 6 and 7.

SwL personnel also performed a quality assurance verification of the SO₂ analysis employing EPA prepared standards (from SwL's library of standards).

DISCUSSION

All sampling was coordinated with AMOCO personnel to insure that the plant was operating at stable conditions.

Equipment setup and all compliance testing was performed on May 4, 1989. Compliance testing consisted of three (3) thirty minute SOx runs, three (3) thirty minute integrated Tedlar bag samples taken for O₂ analysis, and three (3) velocity-stack temperature traverses.

The limited clearance on the platform caused the test crew to utilize an eight (8) foot probe for velocity measurements, which made the last traverse point on each diameter inaccessible. SwL believes that the data is unbiased due to the low pressure differentials (ΔP) encountered (0.05 to 0.10) and the adequate upstream and downstream distance from flow disturbances (greater than 2.0 and 8.0 respectively). This criteria was discussed and agreed upon at the Pretest Conference. Otherwise, all testing was completed without further incident.

SO₂ emissions were reported two ways. First, to comply with 40 CFR 60, Subpart J (TACB Permit Special Provision

2) the SO₂ emissions were calculated on a ppm, dry basis, corrected to zero % oxygen basis. The SO₂ emissions were also quantitated on a lb/hr emission rate basis, in accordance with the TACB Permit Maximum Allowable Emissions Table.

SUMMARY

The average SO₂ emissions for the three (3) compliance sample runs were 108.0 ppm (dry basis, O₂ free) and 26.3 lb/hr., which represents 43.2% and 16.2% of the stated allowable emissions (250 ppm, dry basis, O₂ free and 161.9 lb/hr). The unit produced 406 long tons per day of sulfur product, with three (3) trains going through one (1) SCOT tail gas unit (operational data provided by AMOCO).

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Texas Air Control Board

TABLE NO. 1
SOURCE IDENTIFICATION

Plant Name: AMOCO OIL COMPANY Phone: (713) 945-1158

Address: P. O. Box 401

City: Texas City State: Texas Zip: 77590

Plant Manager: Mr. L. J. Sandheinrich

Plant Representative: Mr. Watson Dupont

Process or Function: Sulfur Recovery Unit

Source Identification: SRU

Stack or Duct Description: Vertical, circular

Stack Height: Platform Height 95' Stack Diameter: 7' 6"

Stack Temperature: 512°F Percent Moisture: 10.7

Sampling Facilities: Two (2) four-inch flanged ports

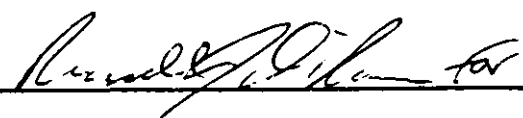
Process (Batch or Continuous): Continuous

Normal Operating Capacity: _____

Operational Status (Sampling Period): 406 long tons/day (3 trains going
through 1 scot tail gas unit)

Sampling Parameters: Sulfur Dioxide, Flow, Oxygen

Date Tested: May 4, 1989

Signed: 

Phillip Yokley
SOUTHWESTERN LABORATORIES, INC.

SOUTHWESTERN LABORATORIES

TABLE NO. 2

SUMMARY OF SAMPLING RESULTS - Sulfur Dioxide

AMOCO OIL COMPANY
TEXAS CITY, TEXAS

SWL Project No. 54-8905-253

Run No.	Date 1989	Time Period	Net Time (mins.)	Vm (Std) dscf	SO ₂ lb/hr	% of * Allowable	**	
							Oxygen Adjusted SO ₂ ppm	% of * Allowable
1	5-4	1206-1236	30	11.366	19.83	12.2	83.27	33.3
2	5-4	1308-1338	30	11.367	29.11	18.0	119.54	47.8
3	5-4	1437-1507	30	11.346	29.93	18.5	121.33	48.5

* SO₂ allowables as per Permit R-8811 are 161.9 lb/hr and 250 ppm, dry and O₂-free

** dry basis, oxygen free

TABLE NO. 3

SUMMARY OF SAMPLING RESULTS - Orsat Analysis

AMOCO OIL COMPANY
TEXAS CITY, TEXAS

SRU
SWL Project No. 54-8905-253

Date 1989	Time Period	Run No.	Orsat Results	
			$\frac{\%O_2}{}$	$\frac{\%CO}{}$
5-4	1206-1236	1	2.2	NA
5-4	1308-1338	2	2.2	NA
5-4	1437-1507	3	2.0	NA

TABLE NO. 4

SUMMARY OF SAMPLING RESULTS - Flow Data (EPA Parameters)

AMOCO OIL COMPANY
TEXAS CITY, TEXAS

BRU

SWL Project No. 54-8905-253

Run No.	Date 1989	Time Period	Ts Stack Temp. °F	Vs Stack Velocity ft/sec	Percent Moisture	ACFM Stack Gas Flow	Qsd Dry Stack Gas Flow dscf/hr
SO ₂ Compliance Test							
1	5-4	1206-1236	508	21.17	12.5	56,113.5	1,599,306
2	5-4	1308-1338	515	21.12	9.7	55,987.3	1,635,818
2	5-4	1437-1507	512	21.15	9.9	56,067.6	1,639,235

FIGURE NO. 1
AMOCO OIL COMPANY
Texas City, Texas
Sulfur Recovery Unit
SWL Project No. 54-8905-253

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Texas Air Control Board

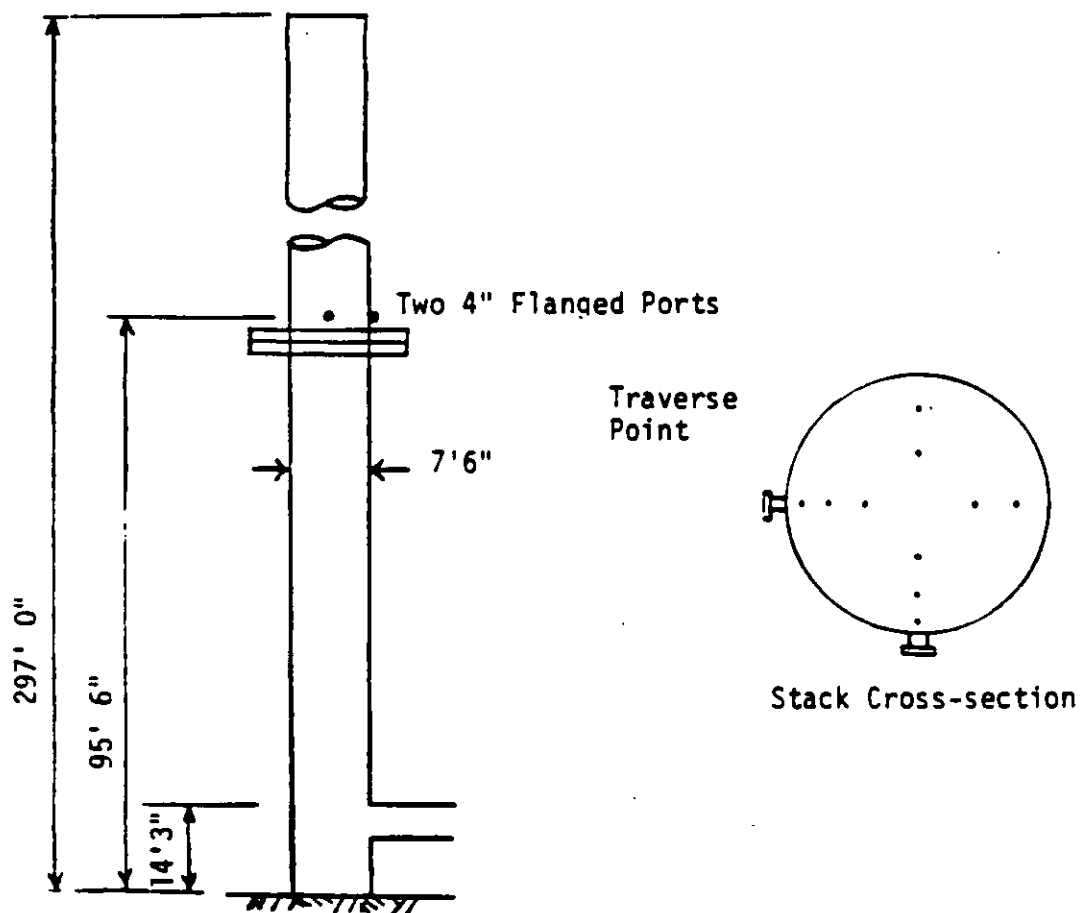
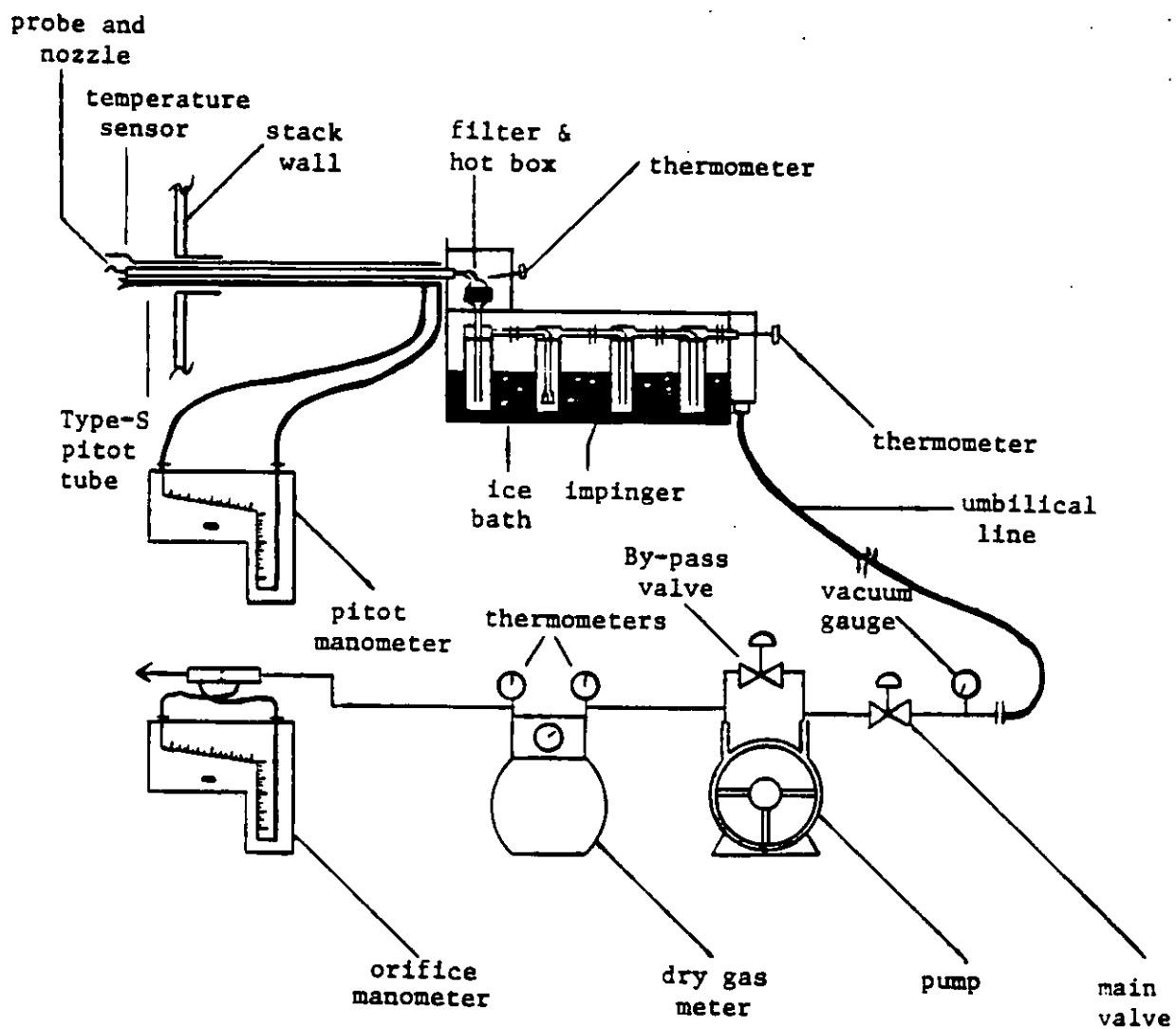


FIGURE # 2

SOUTHWESTERN LABORATORIES
ENVIRONMENTAL ENGINEERING SERVICES DIVISION

SOURCE SAMPLING TRAIN



E.P.A. Method 5 and 6 Source Sampling Train

FIELD & LABORATORY DATA

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Texas Air Control Board

TEXAS CO. TX.

Time:

Run: 122

[illegible]

Ambient Temp. °F

(1) 11 1/2 (2) 11 1/2

Schematic of Stack Cross Section

- 15 -

AMOCO OIL COMPANY -- SULFUR RECOVERY UNIT

Southwestern Laboratories
Environmental Analytical Services Division
Houston, Texas

E.P.A. Method 5 Analytical Data

Run No.	Samp Time (min)	Net Vol. (cu ft)	Avg delta H (in. H2O)	Avg Meter Temp. (F)	Avg Stack Temp. (F)	Avg sq rt delta P (in. H2O)	Meter P Pm (in. Hg)	Stack P Ps (in. Hg)	DGMCF	Cp	As (sq ft)
1	30	11.676	0.5	99	508	0.2895	29.95	29.79	1.0296	0.787	44.179
2	30	11.886	0.5	109	515	0.2895	29.95	29.79	1.0296	0.787	44.179
3	30	11.822	0.5	107	512	0.2902	29.95	29.79	1.0296	0.787	44.179

106

Tot. H2O

Run No.	Collect (mL)	Orsat Data %CO2	%O2	%H2
1	34.6	4.9	2.2	92.9
2	25.9	5	2.2	92.8
3	26.5	5	2	93

E.P.A. Method 5 Flow Calculations

Run No.	Vm Std. dscf	% H2O	NMSG g/g mole	Vs ft/sec	ACFM	Qsd dscf/hr
1	11.366	12.53	27.509	21.17	56,113.49	1,599,306.0
2	11.367	9.69	27.833	21.12	55,987.32	1,635,817.5
3	11.346	9.98	27.802	21.15	56,067.56	1,639,234.5

11.367
9.71

128

SOUTHWESTERN LABORATORIES
SOURCE SAMPLING FIELD DATA

PLANT <u>Amoco oil CO</u>	STACK <u>SRU</u>
DATE <u>5-4-89</u>	OPER. <u>ms</u>
STACK DIA. (exit) _____	STACK DIA. (port) _____
PROBE NO. <u>8-B 83</u>	PTCF <u>787</u>
NOZZLE AREA <u>N/A</u>	FILTER NO. <u>By Pass</u>
METER NO. <u>3</u>	ΔH @ <u>1.7899</u>
PROCESS DATA <u>5.210.547/5.210.548</u>	STACK MOIST. _____
SAMPLE NO. <u>1</u>	STACK HEIGHT _____
STACK PRESS. " <u>H₂O</u> - <u>16</u>	BAR. PRESS. " <u>Hg</u> <u>29.96</u>
DGMCf <u>1.0296</u>	NOZZLE NO. <u>N/A</u>
SILICA GEL NO. <u>29</u>	INITIAL LEAK RATE <u>0.010</u>
BOX NO. <u>E</u>	FINAL LEAK RATE <u>0.012</u>
NOM. REF. PT. <u>N/A</u>	PROBE LINER <u>8'SS</u>

[illegible]

$$V_m(\text{Std.}) = \frac{11.676 \text{ ft}^3}{(\text{Net Vol.})} \times \frac{1.0294}{(\text{DGIMCF})} \times \frac{528}{559} \times \frac{\text{Std. T(R)}}{\text{AMT (R)}} \times \frac{29.95 \text{ Pm}}{29.92 \text{ Std. P.}} = \frac{11.346}{\text{dscf}}$$

Ambient Temp. _____ °F Avg. Meter Temp. 559 °R Meter Pressure (Pm) = $\frac{29.91}{13.6} + \frac{.50}{13.6} = \frac{29.95}{13.6}$ (In. Hg.)

Stack Pressure (Ps) = $\frac{29.91}{13.6} + \frac{-1.6}{13.6} = \frac{29.79}{13.6}$ (in Hg.)

Thermometer ID Dont #37440
TC-83

Barometer ID B

CRSAT Run 1
12:06-12:36

SOUTHWESTERN LABORATORIES
SOURCE SAMPLING
ANALYTICAL DATA

Plant: AMOCO OIL CO
TEXAS Cty TX.

Box E (A) Run No. 1 Date: 5-4-89

PARTICULATES: Filter No. BYPASS

Wt. Filter - Final _____

Wt. Filter - Tare _____

Wt. on Filter _____

Acetone Wash Particulate ✓

Weight Partial Particulate _____

(IMPP) _____ g.

1st Impinger _____

2nd & 3rd Impinger _____

Impinger Catch-Total _____

(IMPI) _____ g.

Weight Total Particulate _____

(GPC) _____ g.

MOISTURE: ^{80% H₂O} Flask 1 ^{6% H₂O} Flask 2 ^{6% H₂O} Flask 3

Impinger Wash H₂O
Final Weight 153.8 184.9 173.6

Initial Weight 148.1 167.2 167.7

Net Increase 5.7 17.7 5.9

Total Water Collected (TWW) 34.6 ml

Silica Gel No. 2129

Final Weight 236.0

Initial Weight 175.5 230.7

Net Increase (TG) 5.3 g.

Vm(Std) 11.366 dscf

$$\% \text{ Moisture} = \frac{(0.04707 \times \text{TWW}) \times 100}{V_m(\text{Std}) + (0.04707 \times \text{TWW})} =$$

$$\frac{162.8622}{12.994622} = 12.57$$

ORSAT:

Time _____

CO₂ _____

O₂ _____

CO _____

MOLECULAR WEIGHT: (MWSG) g/g mol

$$\text{H}_2\text{O} : .125 \times 18 = 2.25$$

$$\text{CO}_2 : .875 \times .049 \times 44 = 1.886$$

$$\text{O}_2 : .875 \times .022 \times 32 = 0.616$$

$$\text{CO} : \text{---} \times \text{---} \times 28 = \text{---}$$

$$\text{N}_2 : .875 \times .929 \times 28 = 22.76$$

*Converted to ml by dividing by the density of water (1 g/ml)

mwsG 27.51

SOUTHWESTERN LABORATORIES
SOURCE SAMPLING FIELD DATA

PLANT	<u>Amoco D.110</u>		STACK	<u>SRU</u>	
DATE	<u>5-4-82</u>	OPER.	<u>JED</u>	SAMPLE NO.	<u>2</u> STACK HEIGHT
STACK DIA. (exit)	<u>90"</u>	STACK DIA. (port)		STACK PRESS. "H ₂ O	<u>1.6</u> BAR. PRESS. "Hg <u>29.91</u>
PROBE NO.	<u>8-R</u>	PTCF	<u>1787</u>	DGMCF	<u>1-0296</u> NOZZLE NO. <u>N/A</u>
NOZZLE AREA	<u>N/A</u>	FILTER NO.	<u>BYPASS</u>	SILICA GEL NO.	<u>24</u> INITIAL LEAK RATE <u>0.10</u>
METER NO.	<u>3</u>	ΔH @	<u>1.7899</u>	BOX NO.	<u>E</u> FINAL LEAK RATE <u>0.01</u>
PROCESS DATA	<u>50x1000/40</u>		ASSUM. MOIST.	NOM. REF. PT.	<u>N/A</u> PROBE LINER <u>3/55</u>

[illegible]

$$V_m(\text{Std.}) = \frac{11.896 \text{ ft}^3}{(\text{Net Vol.})} \times \frac{1.0296}{(\text{DGIMCF})} \times \frac{528}{560} \times \frac{\text{Std. T(R)}}{\text{AMT (R)}} \times \frac{29.95 \text{ Pm}}{29.92 \text{ Std. P}} = \underline{11.367 \text{ dscf}}$$

Ambient Temp. _____ °F Avg. Meter Temp. _____ °R Meter Pressure (Pm) = $\frac{29.21 + .50}{13.6} = \underline{2.195}$
(In. Hg.)

$$\text{Stack Pressure (Ps)} = \frac{29.91 + 71.6}{13.6} = \underline{29.72} \text{ (in Hg.)}$$

Run 2 ORSAT
1308 - 1338

Thermometer ID DONE #37440
TC-8.3

Barometer ID 3

Plant: Amoco Oil Co.
Texas City, TX.

- 20 -

PLANT <u>Amoco Oil Co</u>	STACK <u>SRU</u>
DATE <u>5-4-82</u>	SAMPLE NO. <u>3</u>
OPER. <u>JED</u>	STACK HEIGHT
STACK DIA. (exit) <u>90"</u>	STACK PRESS. " <u>H₂O</u> <u>7.6</u>
STACK DIA. (port)	BAR. PRESS. " <u>Hg</u> <u>22.91</u>
PROBE NO. <u>B-B</u>	DGMCF <u>1.0296</u>
PTCF <u>.737</u>	NOZZLE NO. <u>N/A</u>
NOZZLE AREA <u>N/A</u>	SILICA GEL NO. <u>25</u>
FILTER NO. <u>BYPASS</u>	INITIAL LEAK RATE <u>6.012</u>
METER NO. <u>3</u>	BOX NO. <u>F</u>
ΔH @ <u>1.7829</u>	FINAL LEAK RATE <u>6.012</u>
PROCESS DATA <u>Subtotal Flow</u>	NOM. REF. PT. <u>N/A</u>
ASSUM. MOIST.	PROBE LINER <u>8'SS</u>

$$V_m(\text{Std.}) = \frac{1.822}{(\text{Net Vol.})} \times \frac{1.0296}{(\text{DG MCF})} \times \frac{528}{569} \times \frac{\text{Std. T(R)}}{\text{AMT (R)}} \times \frac{29.95 \text{ Pm}}{29.92 \text{ Std. P}} = \frac{11.346}{\text{dscf}}$$

Ambient Temp. _____ °F Avg. Meter Temp. $\frac{56.6}{22.2}$ °F OR Meter Pressure (Pm) = $\frac{29.91 + .50}{13.6} = \frac{29.95}{\text{(In. Hg.)}}$

Stack Pressure (Ps) = $\frac{29.91 + -1.6}{13.6} = \frac{29.79}{\text{(in Hg.)}}$

Row 3 ORSAT
1437-1507

SOUTHWESTERN LABORATORIES
SOURCE SAMPLING
ANALYTICAL DATA

Plant: Amoco Oil Co.
Texas City, TX

Box E (C) Run No. 3 Date: 5-4-89

PARTICULATES: Filter No. Bypass

Wt. Filter - Final _____

Wt. Filter - Tare _____

Wt. on Filter _____

Acetone Wash Particulate _____

Weight Partial Particulate _____

(IMPP) _____ g.

1st Impinger _____

2nd & 3rd Impinger _____

Impinger Catch-Total _____

(IMPI) _____ g.

Weight Total Particulate _____

(GPC) _____ g.

MOISTURE: ^{80% Isop} Flask 1 ^{62H₂O₂} Flask 2 Flask 3

Impinger Wash H₂O _____

Final Weight 147.4 289.4

Initial Weight 147.4 269.1

Net Increase — 20.3

Silica Gel No. 25

Final Weight 235.4

Initial Weight 229.2

Net Increase (TG) 6.2 g.

Total Water Collected (TWW) 26.5 ml

Vm(Std) 11.366 dscf

$$\% \text{ Moisture} = \frac{(0.04707 \times \text{TWW}) \times 100}{V_m(\text{Std}) + (0.04707 \times \text{TWW})} =$$

$$\frac{124.7355}{12.613355} = 9.9\%$$

ORSAT:

Time _____

CO₂ _____

O₂ _____

CO _____

MOLECULAR WEIGHT: (MWSG) g/g mole

$$\text{H}_2\text{O} \cdot 099 \times 18 = 1.782$$

$$\text{CO}_2 \cdot 901 \times .050 \times 44 = 1.9822$$

$$\text{O}_2 \cdot 901 \times .020 \times 32 = .5766$$

$$\text{CO} - - \times - \times 28 = -$$

$$\text{N}_2 \cdot 901 \times .030 \times 28 = 23.462$$

*Converted to ml by dividing by the density of water (1 g/ml)

mwsG 27.80

Materials, environmental and geotechnical engineering, nondestructive, metallurgical and analytical services

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PROJECT: Amoco Oil Co. TEXAS City, TX. SRU 203 PROJECT NO.: _____

CALCD. BY: _____ DATE: _____ CHKD BY: JED. DATE: 5-5-89

$$\frac{N}{Ba(ClO_4)_2} = \frac{1 \text{ ml} \times 0.10180 \text{ N}}{11.45 \text{ ml}} = 0.00889 \text{ N}$$

RUN / Amp.	Vf:	Aliquot	mls Titrant	Bik Corr. Titrant
1/1	96	5	4.00	4.00
"	"	5	4.00	
1/2	118	5	9.20	9.20
"	"	5	9.20	
1/3	105	10	0.70	0.70
"	"	10	0.70	
2/1	88	5	0.40	0.40
"	"	5	0.40	
2/2:3	214	5	7.50	7.53
"	"	5	7.55	
3/1	90	5	0.15	0.15
"	"	5	0.15	
3/2:3	220	5	7.50	7.50
"	"	5	7.50	
80% Soap Bik.		5	4.05	4.05
"		5	4.05	
6% H ₂ O ₂ Bik.		5	4.05	4.05
"		5	4.05	
CPA QA	0588-6091	5	4.75	4.75
"	"	5	4.75	
3/2:3 spk.		4 + 1 spk.	17.55	17.55
"		4 + 1 spk.	17.55	
spk = 0.10180 N H ₂ SO ₄				

AMOCO OIL COMPANY, TEXAS CITY, TEXAS -- SULFUR RECOVERY UNIT

Southwestern Laboratories
Environmental Analytical Services Division
Houston, Texas

SO2 Lab Data						
Run No.	Titrant Volume (mLs)	Imp. Final V (mLs)	Aliq Vol (mLs)	Normality (N)	Dil Fact	
1A	9.20	118	5.0	0.00889	1	
1B	0.70	105	10.0	0.00889	1	
2	7.53	214	5.0	0.00889	1	
3	7.50	220	5.0	0.00889	1	

Flow Inputs			
Vm Std. (dscf)	% H2O	% O2	Qad (dscf/hr)
11.366	12.53	2.2	1,599,306.0
11.366	12.53	2.2	1,599,306.0
11.367	9.69	2.2	1,635,817.5
11.346	9.90	2	1,639,234.5

MS

E.P.A. Method 6 Sulfur Dioxide (SO2) Calculations					
Run No.	Conc. (lb/dscf)	ppm (dry)	ppm (wet)	ppm (O2 free)	lb/hr
1A	1.199E-05	72.06	63.03	80.54	19.18
1B	4.059E-07	2.44	2.13	2.73	0.65
2	1.780E-05	106.95	96.59	119.54	29.11
3	1.826E-05	109.72	98.86	121.33	29.93

SOUTHWESTERN LABORATORIES, INC.
ORSAT FIELD DATA SHEET

PLANT: Amoco oil co. SOURCE NAME: SRU Stack
TEXAS city, TX. DATE: 5-4-89

RUN NO: <u>1</u>		TIME SAMPLED: <u>1206-1236</u>		TIME ANALYZED: <u>1545-1605</u>	
$\frac{100}{100_2}$	$\frac{5.0}{100_2} = \frac{5.0}{100_2}$	$\frac{100}{100_2}$	$\frac{4.8}{100_2} = \frac{4.8}{100_2}$	$\frac{100}{100_2}$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2$	$\frac{7.2}{100_2} = \frac{7.2}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{7.0}{100_2} = \frac{7.0}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2}$

RUN NO: <u>2</u>		TIME SAMPLED: <u>1308-1338</u>		TIME ANALYZED: <u>1608-1632</u>	
$\frac{100}{100_2}$	$\frac{5.0}{100_2} = \frac{5.0}{100_2}$	$\frac{100}{100_2}$	$\frac{5.0}{100_2} = \frac{5.0}{100_2}$	$\frac{100}{100_2}$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2$	$\frac{7.2}{100_2} = \frac{7.2}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{7.2}{100_2} = \frac{7.2}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2}$

RUN NO: <u>3</u>		TIME SAMPLED: <u>1437-1507</u>		TIME ANALYZED: <u>1635-1658</u>	
$\frac{100}{100_2}$	$\frac{5.0}{100_2} = \frac{5.0}{100_2}$	$\frac{100}{100_2}$	$\frac{5.0}{100_2} = \frac{5.0}{100_2}$	$\frac{100}{100_2}$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2$	$\frac{7.0}{100_2} = \frac{7.0}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{7.0}{100_2} = \frac{7.0}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2}$

RUN NO: _____		TIME SAMPLED: _____		TIME ANALYZED: _____	
$\frac{100}{100_2}$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2}$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2}$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2$	$\frac{100}{100_2}$
$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2} = \frac{100}{100_2}$	$\frac{100}{100_2} + O_2 + CO$	$\frac{100}{100_2}$

RECEIVED

JUN 21 1989

Region 7
Texas Air Control Board

DEFINITIONS

Standard Conditions: 68° and 29.92 inches of mercury

Stack Conditions: Stack temperature, pressure and moisture

NOMENCLATURE

ACFM	Volumetric stack gas flow in cubic feet per minute at stack conditions
AMT	Average temperature at meter in degrees Rankin
AN	Area of nozzle in square feet
As	Area of stack in square feet
C	Total pollutant concentration in grains per dry standard cubic feet
CFM	Cubic feet per minute
CO ₂	Carbon Dioxide
CO	Carbon Monoxide
Cp	Pitot tube correction factor or (PTCF)
Cs	Partial pollutant concentration in grains or grams per dry standard cubic foot (total less impinger catch)
De	Equivalent stack diameter of rectangular stack

$$(De = \frac{2LW}{L+W})$$

DGMCF	Dry gas meter correction factor
DI	Deionized water
dscf	dry standard cubic feet
EA	Excess Air (expressed as percent)
°F	Temperature in degrees Fahrenheit
GPC	Grams of particulate caught (total)

NOMENCLATURE (Cont'd)

g	Grams
gr	Grains
Hg	Mercury
H ₂ O	Water
I	Isokinetics as percent
IMPI	Grams of particulate caught in impinger
IMPP	Grams of particulate caught before impinger (total less impinger catch)
MWSG	Molecular weight of stack gas in grams/gram-mole (g/g-mole) or pounds/pound-mole (lb/lb-mole)
N ₂	Nitrogen
O ₂	Oxygen
Pb	Barometric pressure in inches of mercury
Pm	Meter pressure in inches of mercury
ppm	Parts per million (Volume/Volume or mass/mass)
P _r	Barometric pressure of reference barometer
Ps	Absolute pressure in stack in inches of mercury
PMR	True pollutant mass rate in pounds per hour
PMRs	Pollutant mass rate for the "front half" in pounds per hour (total less impinger catch)
Qsd	Dry volumetric stack gas flow rate corrected to standard conditions in dscf/hr
°R	Temperature in degrees Rankin (equivalent to °F + 460°)
Std.P	Pressure at standard conditions (29.92 inches of mercury)

NOMENCLATURE (Cont'd)

Std.T	Temperature at standard conditions (528 °R)
TG	Total weight of water collected in silica gel, in grams
T _r	Temperature of reference thermometer
Ts	Stack gas temperature in degrees Rankin
T _t	Temperature of test thermometer
TWW	Total water wash volume collected in impingers and silica gel, in milliliters (ml) NOTE: Density of H ₂ O equals 1 g/ml
Vm (Std)	Total gas sampled converted to standard conditions, dry basis, in cubic feet
Vs	Stack gas velocity in feet per second
ΔH	Velocity head orifice reading in inches of water
ΔP	Stack gas velocity head in inches of water
θ	Sample time in minutes



SOUTHWESTERN LABORATORIES

ENVIRONMENTAL SERVICES, HOUSTON, TEXAS
CONTROL UNIT CALIBRATION

BOX NO. 000 #3

DATE 1-16-89

DRY GAS METER NO. _____

Barometric Pressure, P_b 30.24 in. Hg

Orifice Manometer setting ΔH , in. H ₂ O	Gas Volume wet test meter V_w , ft ³	Gas Volume dry gas meter V_d , ft ³	Temperature				Time min.- sec.	Time θ min.	γ	$\Delta H\theta$
			Wet Test Meter t_w , °F	Dry Gas Meter						
				Inlet t_{di} , °F	Outlet t_{do} , °F	Average t_d , °F				
0.25	3	362.535 365.469	74 74	72 86	76 80	78.5	10.21½	10.36	1.0304	1.6550
0.50	5	366.455 371.738	74 74	86 104	82 90	90.5	12.46	12.77	1.0330	1.7710
1.0	5	372.451 377.565	74 74	100 116	92 100	102	9.19½	9.325	1.0266	1.8500
2.0	10	378.595 388.988	74 74	116 132	98 108	113.5	13.26½	13.442	1.0283	1.8836
Average									1.0296	1.7899

Calculations

ΔH	$\frac{\Delta H}{13.6}$	γ	$\Delta H\theta$
		$\frac{V_w P_b (t_d + 460)}{V_d \left(P_b + \frac{\Delta H}{13.6} \right) (t_w + 460)}$	$\frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460)\theta}{V_w} \right]^2$
0.25	0.018	$\frac{3.00 \cdot 30.24 \cdot 538.5}{2.934 \cdot 30.26 \cdot 534}$	$\frac{0.0317 \cdot 0.25}{30.24 \cdot 538.5} \left[\frac{534 \cdot 10.36}{3.00} \right]^2$
0.50	0.037	$\frac{5.00 \cdot 30.24 \cdot 550.5}{4.983 \cdot 30.28 \cdot 534}$	$\frac{0.0317 \cdot 0.50}{30.24 \cdot 550.5} \left[\frac{534 \cdot 12.77}{5.00} \right]^2$
1.0	0.074	$\frac{10.00 \cdot 30.24 \cdot 562}{9.114 \cdot 30.31 \cdot 534}$	$\frac{0.0317 \cdot 1.00}{30.24 \cdot 562} \left[\frac{534 \cdot 9.325}{5.00} \right]^2$
2.0	0.147	$\frac{10.00 \cdot 30.24 \cdot 573.5}{10.393 \cdot 30.39 \cdot 534}$	$\frac{0.0317 \cdot 2.00}{30.24 \cdot 573.5} \left[\frac{534 \cdot 13.442}{10.00} \right]^2$

Is each $\Delta H\theta$ within ± 0.15 of $\Delta H\theta$
☐ yes, then $\Delta H\theta$ is valid.
☐ No, then repair and recalibrate.

Is each γ within ± 0.02 of $\bar{\gamma}$
☐ yes, then $\bar{\gamma}$ is valid.
☐ No, then repair and recalibrate

SOUTHWESTERN LABORATORIES

Robert D. [Signature]



SOUTHWESTERN LABORATORIES

ENVIRONMENTAL SERVICES, HOUSTON, TEXAS

GAMMA CALIBRATION SHEET

PROJECT Amoco oil co. Texas city LAB NO.: 54-2305-253 DATE 5-5-89

BOX NO. 3

CURRENT γ 1.0296

AVERAGE ΔH .50

MAXIMUM VACUUM 4 1/2

ACCEPTABLE γ LIMITS

WET TEST GAS VOLUME: 1.00 FT^3

.9781 TO 1.0811

BAROMETRIC PRESSURE: 29.96 IN. Hg

DRY GAS METER VOLUME V_d FT ³	WET TEST METER t_w	TEMPERATURE			γ
		DRY GAS METER			
		INLET t_{d1} °F	OUTLET t_{d2} °F	AVERAGE t_d °F	
280.757 281.732	73	76 80	76 76	77	1.0320
281.732 282.709	73	80 84	76 78	79.5	1.0346
282.709 283.688	73	84 88	78 78	82	1.0373
			AVERAGE γ		1.0346

Ready
to read

$$\frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{H}{13.6}) (t_w + 460)} = \gamma$$

$$\frac{1.00 \cdot 29.96 \cdot 537}{.975 \cdot 30.00 \cdot 533} = 1.0320$$

$$\frac{1.00 \cdot 29.96 \cdot 539.5}{.977 \cdot 30.00 \cdot 533} = 1.0346$$

$$\frac{1.00 \cdot 29.96 \cdot 542}{.979 \cdot 30.00 \cdot 533} = 1.0373$$

$$\frac{1.00 \cdot 29.96 \cdot 542}{.979 \cdot 30.00 \cdot 533} = 1.0373$$

SOUTHWESTERN LABORATORIES

Robert D. Doughty

EES 01/81-19/801



99 REINERMAN ST. • HOUSTON, TEXAS 77007 • 713-861-3816

August 9, 1988

Southwestern Laboratories
P.O. Box 8768
Houston, TX 77249
Attn: R. Daughtery

Dear Sir:

This is to certify that your 150 Precision Wet Test Meter, Serial No. AB-3, has been calibrated with an American 10 Ft. Bell Prover, Serial No. 3157. It is traceable to the Bureau of Standards.

Reference No. 106870, PI-TAPE.

Test results are as follows:

<u>IN-TEST</u>		<u>OUT-TEST</u>	
<u>Rate of Flow</u>	<u>% Error</u>	<u>Rate of Flow</u>	<u>% Error</u>
75 CFH	1.5%-	75 CFH	0.0%±

Sincerely,

CARL POE CO., INC.

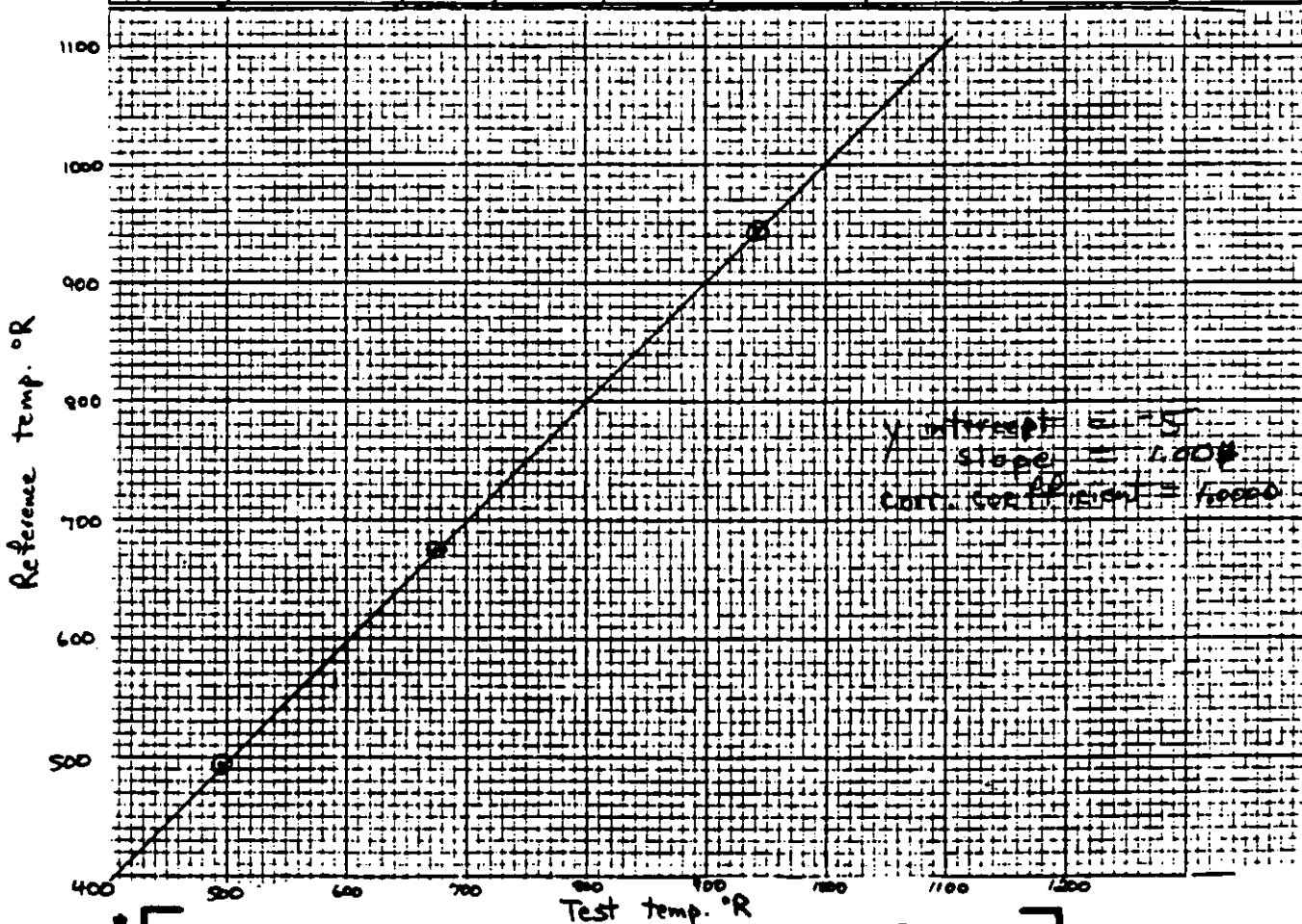
Carl W. Poe

CNP:mv

SOUTHWESTERN LABORATORIES
TEMPERATURE SENSOR CALIBRATION DATA

DATE: 9-23-82 THERMOCOUPLE NO.: 8-3
 AMBIENT TEMPERATURE: 77 °F BAROMETRIC PRESSURE: 30.02 In. Hg
 CALIBRATOR: M. Jones REFERENCE: Mercury-in-glass ASTM 3C

Reference Point Source	Reference Thermometer Temperature			Thermocouple Temperature		Temperature Difference* %
	°C	°F	°R	°F	°R	
Ice water	0	32	492	35	495	0.6
Boiling water	101	213.8	673.8	217	677	0.5
Boiling oil	249.7	481.5	941.5	483	943	0.2



$$* \left[\frac{(\text{ref. temp., } ^\circ\text{F} + 460) - (\text{test thermom. temp., } ^\circ\text{F} + 460)}{\text{ref. temp., } ^\circ\text{F} + 460} \right] 100 \leq 1.5\% \text{ absolute value}$$

THERMOMETER AND BAROMETER CALIBRATION

Project: Amper 0.1 cc By: Robert D. Dwyer

Project No: 54-8955-253 Date: 5-5-89

<u>Thermometer</u>	<u>Thermocouple</u>	<u>Temperature °R (T_t)</u>	<u>Reference °R (T_r)*</u>
Dwyer 37440	TC-83	532	531.6

*Mercury-in-Glass Thermometer

$$\% \text{ Difference} = \frac{T_t - T_r}{T_r} \times 100$$

$$\frac{532 - 531.6}{531.6} \times 100 = .075 \leq 1.58 : \text{Reading is valid.}$$

Aneroid °Hg (P_b)

Reference °Hg (P_r)*

Barometer

29.96

29.99

*Mercurial Barometer

$$\text{Difference} = P_b - P_r$$

$$29.99 - 29.96 = 0.02 \leq 0.1 \text{ °Hg} : \text{Reading is valid}$$



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Materials, environmental and geotechnical engineering, nondestructive, metallurgical and analytical services

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PROJECT: Pitot Tube Calibration PROJECT NO.: _____

CALCD. BY: _____ DATE: _____ CHKD BY: _____ DATE: _____

PITOT TUBE IDENTIFICATION NUMBER: 8-B DATE: 12-21-88
CALIBRATED BY: JRO 24" x 70" Pitot Tube Cp STD = 0.99

"A" SIDE CALIBRATION				
RUN NO.	ΔP_{std} cm H ₂ O (in. H ₂ O)	$\Delta P(u)$ cm H ₂ O (in. H ₂ O)	$C_p(u)$	DEVIATION $C_p(u) - \bar{C}_p(A)$
1	.31	.49	.787	0
2	.31	.49	.787	0
3	.31	.49	.787	0
$\bar{C}_p$ (SIDE A)			.787	

"B" SIDE CALIBRATION				
RUN NO.	ΔP_{std} cm H ₂ O (in. H ₂ O)	$\Delta P(u)$ cm H ₂ O (in. H ₂ O)	$C_p(u)$	DEVIATION $C_p(u) - \bar{C}_p(B)$
1	.31	.49	.787	0
2	.31	.49	.787	0
3	.31	.49	.787	0
$\bar{C}_p$ (SIDE B)			.787	

$$\text{AVERAGE DEVIATION} = \sigma(A \text{ OR } B) = \frac{\sum_{i=1}^3 |C_p(u) - \bar{C}_p(A \text{ OR } B)|}{3} \leftarrow \text{MUST BE } < 0.01$$

$$|\bar{C}_p(\text{SIDE A}) - \bar{C}_p(\text{SIDE B})| \leftarrow \text{MUST BE } < 0.01$$

$$C_{Ps} = C_{pSTD} \sqrt{\frac{A_{pSTD}}{A_{Ps}}}$$

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PROJECT: Ba (ClO₄)₂ Standardization PROJECT NO.: _____
 CALCD. BY: TRD/pj DATE: 4-28-89 CHKD BY: _____ DATE: _____

mls. Titrant

Titration 1 : 11.40

2 : 11.50

11.45 mls Titrant Avg.

$$\begin{aligned} \underline{N \text{ Ba (ClO}_4)_2} &= \frac{1 \text{ ml} \times 0.10180 \text{ N}}{11.45 \text{ mls.}} \\ &= 0.00889 \text{ N} \end{aligned}$$

41 mls. ΔI. H₂O

1 ml 0.10180 N H₂SO₄

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PROJECT: Amoco Oil Company PROJECT NO.: 8705-253
CALCD. BY: py DATE: 5-5-89 CHKD BY: _____ DATE: _____

SO_x TITRATION - EPA Q.A.

<u>Ref. Sample #</u>	<u>Volume</u>	<u>Aliquot</u>	<u>Titrant</u>	<u>Blank Corr. Avg</u>
0588-6091	5 ml / 100 ml	5 ml	4.75 ml	4.75 ml
"	"	"	"	"
6% H ₂ O ₂	N.A.	5 ml	< 0.05 ml	< 0.05 ml
"	"	"	"	"

Eq. 6-2:

$$C_{SO_2} = \frac{32.03 (V_t - V_{t0}) N (Vol. soln / Vol. al.)}{V_{mstd.}}$$

$$C_{0588-6091} = \frac{32.03 (4.75 ml) (0.00889 N) (100/5)}{0.021}$$

$$= 1288.1 \text{ mg/dscm}$$

$$\text{EPA Ref. Value} = 1296.0 \text{ mg/dscm}$$

$$\% \Delta \text{ff. From EPA Value} = \frac{1288.1 - 1296.0}{1296.0} \times 100$$

$$= -0.6\% \checkmark$$



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PROJECT: Amoco O:1 - SR4 PROJECT NO.: 905-253CALCD. BY: PM DATE: 5-5-89 CHKD BY: _____ DATE: _____

SO₄ ANALYSIS - SPIKE RECOVERY

$$C_{\text{spike}} = \frac{(1 \text{ ml})(0.10180 \text{ N})}{5 \text{ ml}} = 0.02036 \text{ N} \quad \checkmark$$

$$C_{\text{sample}} = \frac{(7.50 \text{ ml})(0.00889 \text{ N})(4 \text{ ml})}{5 \text{ ml} \times 5 \text{ ml}} = 0.01067 \text{ N} \quad \checkmark$$

$$C_{\text{spk + spl.}} = \frac{(17.55 \text{ ml})(0.00889 \text{ N})}{5 \text{ ml}} = 0.03120 \text{ N} \quad \checkmark$$

$$\% \text{ Recovery} = \frac{0.03120 \text{ N} - 0.01067 \text{ N}}{0.02036 \text{ N}} = 100.8 \% \quad *$$

* TACB Allowable Limits : $90 \% \leq X \leq 110 \%$.

SOUTHWESTERN LABORATORIES
SOURCE SAMPLING CALCULATIONS

Run No: 1
Plant: Amoco Oil - SZ4
Texas City, TX
Date: 5-5-89

Hand Calculation Q.A. Check
of Computer Data.

SULFUR DIOXIDE

Impinger No. 2:

$$(C_{SO_2})_2 = \frac{7.061 \times 10^{-5} (V_t - V_{tb})}{V_m (std)} \times N \times \frac{V_{soln.}}{V_a} = \frac{7.061 \times 10^{-5} \times 9.20 \times 0.00889 M \times 118}{11.366 \times 5}$$
$$= 1.199 \times 10^{-5} \text{ lb/dscf}$$

Impinger No. 3:

$$(C_{SO_2})_3 = \frac{7.061 \times 10^{-5} (V_t - V_{tb})}{V_m (std)} \times N \times \frac{V_{soln.}}{V_a} = \frac{7.061 \times 10^{-5} \times 0.70 \times 0.00889 \times 105}{11.366 \times 10}$$
$$= 4.059 \times 10^{-7} \text{ lb/dscf}$$

TOTAL CONCENTRATION OF SO₂

$$C_{SO_2} = (C_{SO_2})_2 + (C_{SO_2})_3 = 1.199 \times 10^{-5} + 4.059 \times 10^{-7} = 1.240 \times 10^{-5} \text{ lb/dscf}$$

$$\text{ppm SO}_2 = \frac{C_{SO_2} \times 1.6014 \times 10^{10}}{2664.78} = \frac{1.240 \times 10^{-5} \times 1.6014 \times 10^{10}}{2664.78} = 74.49 \text{ ppm (dry)}$$

$$\text{ppm SO}_2 (\text{wet}) = \text{ppm SO}_2 (\text{dry}) \times \left(1 - \frac{\%H_2O}{100}\right) = 74.49$$

$$\text{ppm SO}_2 (\text{dry-O}_2 \text{ free}) = \text{ppm SO}_2 \times \left(\frac{20.9}{20.9 - \%O_2}\right) = 74.49 \times \left(\frac{20.9}{20.9 - 2.2}\right) = 83.25 \text{ ppm (dry, O}_2 \text{ free)}$$

$$\text{lb/hr SO}_2 = C_{SO_2} \times Q_{sd} = 1.240 \times 10^{-5} \times 1,599,306.0 = 19.83 \text{ lb/hr}$$

RUSSELL J. DIRAIMO, P.E.,

Manager
Environmental Services
Houston

Formal Education:

B.S. Civil and Environmental Engineering - University of
Rhode Island - 1977

Additional Technical Education:

Texas A & M University- Cause and Prevention of Grain
Elevator Fires and Explosions - 1978
Kansas State University - International Symposium on Grain
Dust - 1979
Houston Building Owners and Managers Association, Inc. -
"Controlling Exposure to Asbestos in the Office
Environment" - 1985
The National Asbestos Training Center - University of
Kansas "Practices and Procedures for Asbestos Control"
- 1986
Industrial Hygiene & Safety Technology, Inc. - "Hazardous
Waste Site Operations and Emergency Response", 29 CFR
1910.120. 40 hour training course, August, 1987.
Texas A & M Extension Course - "Asbestos Hazardous
Emergency Response Act", 40 CFR 763 Subpart D, April,
1988. Certified Inspector; Certified Management
Planner.

Professional Engineering Registration:

Texas

Certified Texas Air Control Board Visible Emissions Evaluator:

1978 to present

Awards and Honors:

Graduated High Distinction
Tau Beta Pi Honor Society
Phi Kappa Phi Honor Society

Professional Affiliations:

National Society of Professional Engineers (NSPE)
Texas Society of Professional Engineers (TSPE)
Air Pollution Control Association (APCA)
Source Evaluation Society
Water and Wastewater Analyst Association
Texas Hazardous Waste Management Society
National Asbestos Council

Since joining SwL in 1978, Russell has gained engineering experience with a concentration in civil and environmental applications.

As Manager of the Environmental Services Division, Mr. DiRaimo supervises the field and laboratory personnel which provide services including air pollution testing, personnel monitoring, water and wastewater analysis, hazardous waste characterizations, site contamination studies, and gas chromatography analysis. Mr. DiRaimo's responsibilities include personnel assignments, scheduling, data interpretation, calculations and report preparation.

Mr. DiRaimo has supervised environmental testing project for numerous municipal, industrial and petrochemical facilities including the Norco, Louisiana and Deer Park Texas facilities of Shell Oil Company, the Beaumont Specialty Chemicals Plant of Mobil Chemical Company, the Texas City, Texas facility of Amoco Oil Company, and the Houston, Texas facility of Hatheway Patterson Corporation.

PHILLIP W. YOKLEY, Environmental Scientist
Environmental Services
Houston

Formal Education:

B.S. - Environmental Health, East Tennessee State
University - 1981

Additional Technical Education:

Asbestos Abatement Training Program, the University of
Texas, Arlington, Texas - May, 1987.
Asbestos Technique Workshop, American Industrial Hygiene
Association, Houston, Texas - April, 1987.
Identification of Asbestos Utilizing Polarized Light
Microscopy, McCrone Research Institute, Chicago,
Illinois - 1986.
In-Stack Opacity Monitor Audit Procedures, Environmental
Protection Agency Regional Office, Annapolis, Maryland
Texas A & M Extension Course - "Asbestos Hazardous
Emergency Response Act", 40 CFR 763 Subpart D, April,
1988. Certified Inspector; Certified Management
Planner.

Certified Texas Air Control Board Visual Emissions Evaluator:
Since 1986

Professional Affiliations:

Air Pollution Control Association (APCA)
Source Evaluation Society

Since graduation, Phillip Yokley has gained an extensive background in the field of environmental monitoring, sampling and analysis. He began his career with the United States Department of Energy working in an industrial hygiene/safety capacity and subsequently held a position with coal-gasification wastewater treatment pilot facility. Most recently Mr. Yokley worked as an air pollution analyst for the Air Pollution Control Bureau in Pittsburgh, Pennsylvania.

Mr. Yokley joined SwL in 1985 as an environmental technician for the Environmental Services Division. He has subsequently been promoted to Senior Environmental Technician and Field Supervisor. His responsibilities include supervising field testing operations, equipment calibration, and the utilization of analytical instrumentation such as total organic carbon and total organic halogen analyzers necessary for the analysis of environmental samples. In addition, he is responsible for the identification and quantitation of asbestos content present in bulk samples as well as the determination of fibrous concentrations present in airborne samples.

JESSE ROBERT DAUGHTRY, Environmental Technician
Environmental Services
Houston

Additional Technical Education:

Texas Water Commission Wastewater Operator, Class D
Certification, January 21, 1986.

Texas A&M University System, Wastewater Analysis, February
6, 1986.

Robbie Daughtry is a member of SwL's air pollution team and has participated and served as Field Supervisor in source testing for parameters such as particulates, sulfur oxides, volatile organics, nitrogen oxides, hydrogen sulfide and metals, Methodology includes those published by EPA and the Texas Air Control Board. Robbie is also a certified Visible Emissions Evaluator.

Robbie has also assisted in the preparation of samples for gas chromatographic analysis, and the sampling and analysis of water and wastewater samples.



L.J. Sandheinrich
Refinery Manager

June 19, 1989

DH

Letter
covering the
sampling & data

Amoco Oil Company

Post Office Box 401
Texas City, Texas 77592
409-945-1011

8.13 BR R-4 #8

RECEIVED

JUN 21 1989

Region 7
Texas Air Control Board

CERTIFIED NO.: P 882 515 693

Mr. Herbert W. Williams
Regional Supervisor
Texas Air Control Board, Region 7
5555 West Loop South, Suite 300
Bellaire, Texas 77401

R7-232

Amoco Oil Company, Texas City Refinery, Permit R-8811
Performance Test - Sulfur Recovery Unit Incinerator Stack

Dear Mr. Williams:

As required by Special Provision 4E of Permit R-8811, Amoco Oil Company, Texas City refinery submits the results from stack testing performed on the Sulfur Recovery Unit stack on May 4, 1989. This test was required to demonstrate operations with the IFP tail gas treater shut down and the gas routed to the SCOT tail gas treater.

The data indicate that the SRU incinerator emitted SO₂ at a rate of 26.29 lb/hr during the test. This rate is well below the allowable rate of 161.9 lb/hr reflected in permit R-8811. Measured SO₂ concentrations averaged 108 ppm, compared to the permit limit concentration of 250 ppm.

Sulfur production was 406 long tons/day for the May 4, 1989 test period. To compensate for the low sulfur production rate, the total production was routed through a single SCOT tail gas unit. Since both SCOT tail gas units are identical, this loading is the same as running two tail gas units while producing 812 long tons/day of sulfur. Therefore, the SO₂ concentration measured at these test conditions is equivalent to what would be measured at a production rate of 812 long tons/day through both SCOT tail gas units. (406 long tons/day per SCOT unit). The SO₂ emission rate would be twice the measured rate, or 52.58 lb/hr, which is below the allowable rate of 161.9 lb/hr reflected in permit R-8811.

8.13

Mr. H. W. Williams
June 19, 1989
Page 2

If you have any questions concerning this test, please contact Ms. R. S. Edrozo, Engineer, Environmental Control, at (409) 945-1156.

Very truly yours,

A handwritten signature in cursive script, appearing to read "L. J. Sandheinrich".

L. J. Sandheinrich

Enclosures

Dr. E. R. Ibert
Director of Environmental Control Services
Galveston County Health District
P. O. Box 939
La Marque, Texas 77568

Mr. Eli Bell
Executive Director
Texas Air Control Board
6330 Highway 290 East
Austin, Texas 78723