

RAMCON

ENVIRONMENTAL CORPORATION

Note: This is a reference cited in AP 42, *Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

The file name refers to the reference number, the AP42 chapter and section. The file name "ref02_c01s02.pdf" would mean the reference is from AP42 chapter 1 section 2. The reference may be from a previous version of the section and no longer cited. The primary source should always be checked.

SOURCE SAMPLING
for
PARTICULATE EMISSIONS
AJAX PAVING INDUSTRIES
DETROIT, MICHIGAN

Intester R. S.
Romulus
August 10, 1992



Mark Boden
Ajax Paving Industries



G. Sumner Buck, III
President



Mike Winkler
Team Leader

RAMCON

ENVIRONMENTAL CORPORATION

August 21, 1992

Mr. Mark Boden
Ajax Paving Industries
8744 S. Inkster Road
Romulus, MI 48174

Re: Particulate Emissions Test: Detroit, Michigan

Dear Mr. Boden:

Enclosed you will find four copies of our report on the particulate emissions test we conducted at your plant. Based on our test results, the average grain loading of the three test runs do pass both EPA New Source Performance Standards and those set by the State of Michigan. Therefore, the plant is operating in compliance with Federal and State Standards.

You will want to sign the report covers and send two copies to:

Mr. Mike Maillard
Michigan DNR
2211 E. Jefferson Avenue
Detroit, Michigan 48207

You will need to keep one copy of the report at the plant. We certainly have enjoyed working with you. Please let us know if we can be of further assistance.

Sincerely,



G. Sumner Buck, III
President

GSBIII:djb

Enclosures

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SECTION A:

- 1. INTRODUCTION**
- 2. TEST RESULTS**
- 3. TEST PROCEDURES**

SECTION A.

1. INTRODUCTION

On July 7, 1992 and on August 10, 1992 personnel from RAMCON Environmental Corporation conducted a source emissions test for particulate emissions compliance (one run done on 7/7/92; three runs on 8/10/92) at Ajax Paving Industries' Boeing drum mix asphalt plant located in Detroit, Michigan. RAMCON personnel conducting the test were Mike Winkler, Team Leader, and Chuck Hughes. Thomas South was responsible for the laboratory analysis including taring the beakers and filters and recording final data in the laboratory record books. Custody of the samples was limited to Mr. Winkler and Mr. South.

The purpose of the test was to determine if the rate of particulate emissions from this plant's baghouse is below or equal to the allowable N.S.P.S. emissions limit set by US EPA and the State of Michigan.

2. TEST RESULTS

Table I summarizes the test results. The grain loading limitation for EPA is .04 gr/dscf as specified in 39 FR 9314, March 8, 1974, 60.92 Standards for Particulate Matter (1), as amended. The allowable emissions for the State of Michigan are the same as those set by EPA.

Youn Tak Kwon of Wayne County Department of Public Health observed the testing conducted by RAMCON Environmental.

Permit (2)

SUMMARY OF TEST RESULTS

TABLE I

July 7, 1992 and August 10, 1992

<u>Test Run</u>	<u>Time</u>	<u>Grain Loading gr/DSCF</u>	<u>Isokinetic Variation</u>	<u>Emissions Lbs/Hr</u>
1	02:46 - 03:53	0.0074	111.1%	2.34
2	06:18 - 07:37	0.0137	138.2%	4.11
3	08:09 - 09:23	0.0118	128.7%	3.69
4	17:42 - 18:42	0.0095	109.8%	32.94
	Average:	0.0106	121.95%	10.77

On the basis of these test results, the average grain loading of the four test runs was below the .04 gr/DSCF allowable emissions limitation set by EPA and the State of Michigan. Therefore, the plant is operating in compliance with State and Federal Standards.

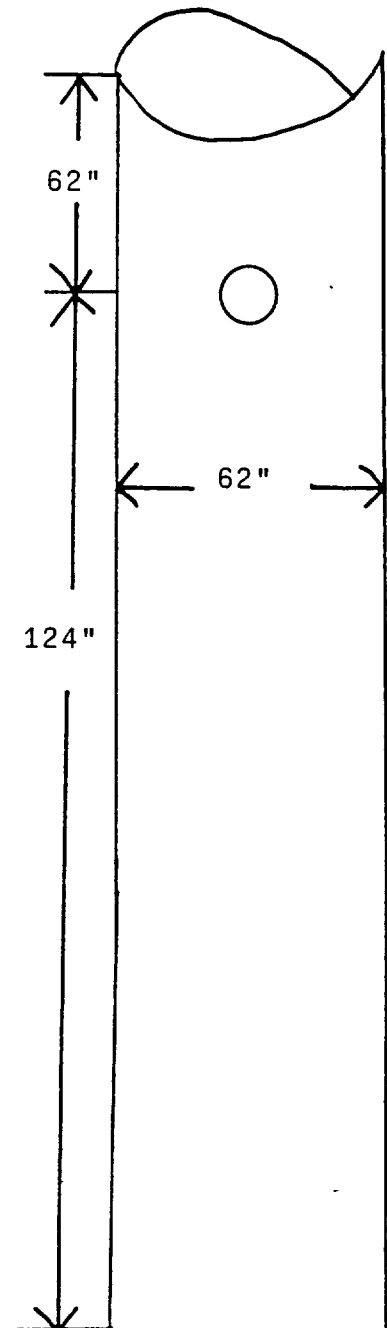
3. TEST PROCEDURES

(a) Method Used: Method 5 source sampling was conducted in accordance with requirements of the U.S. Environmental Protection Agency as set forth in 39 FR 9314, March 8, 1974, 60.93, as amended.

(b) Problems Encountered: No problems were encountered that affected testing.

(c) Sampling Site: The emissions test was conducted after a baghouse on a round stack with a diameter of 62". The sampling ports were placed 90° apart, 62" down (1.0 diameters upstream) from the top of the stack and 124" up (2.0 diameters downstream) from the last flow disturbance. Twenty four points were sampled, twelve through each traverse for 2.5 minutes each for a total testing time of 60 minutes.

<u>Points on a Diameter</u>	<u>Probe Mark</u>
1	1.3"
2	4.2"
3	7.3"
4	11.0"
5	15.5"
6	22.0"
7	40.0"
8	46.5"
9	51.0"
10	54.7"
11	57.8"
12	60.7"



SECTION B:
THE SOURCE

B. THE SOURCE

Ajax Paving Industries employs a Boeing drum mix asphalt plant which is used to manufacture hot mix asphalt for road pavement. The process consists of blending prescribed portions of cold feed materials (sand, gravel, screenings, chips, etc.) uniformly and adding sufficient hot asphalt oil to bind the mixture together. After the hot asphalt mix is manufactured at the plant, it is transported to the location where it is to be applied. The hot asphalt mix is spread evenly over the surface with a paver then compacted with a heavy roller to produce the final product.

The following is a general description of the plant's manufacturing process: The cold feed materials (aggregate) are dumped into separate bins which in turn feed a common continuous conveyor. The aggregate is dispensed from the bins in accordance with the desired formulation onto the cold feed system conveyor, to an inclined weigh conveyor, then to a rotating drum for continuous mixing and drying at approximately 300°F. When recycled asphalt mix is used, it is added halfway down the drum through a separate conveyor. The required amount of hot asphalt oil is then injected onto and mixed into the dried aggregate. The now newly formed hot asphalt mix is pulled to the top of a storage silo by a conveyor. The hot asphalt mix is then discharged from the storage silo through a slide gate into waiting dump trucks which transports the material to a final destination for spreading. The rated capacity of the plant will vary with each aggregate mix and moisture content with a 5% surface moisture removal.

The drum mixer uses a burner fired with Natural Gas to heat air to dry the aggregate, and the motion of the rotating drum to blend the aggregate. The air is drawn into the system via an exhaust fan. After passing through the gas burner and the mixing drum, the air passes through a baghouse. The exhaust gasses are drawn through the baghouse and discharged to the atmosphere through the stack. The design pressure drop across the tube sheet is 2-6 inches of water. The particulate matter, which is removed by the baghouse, is reinjected into the drum mixer.

COMPANY NAME AXAX MATERIALS COMPANY REP. Mark Boden PHONE (313) 398-7300
LOCATION OF FACILITY 88744 SINKOFF RD ORIGINAL START-UP DATE _____ DESIGNED CAPACITY _____
OEM _____ MODEL NO. _____

MODEL NO.

TYPE

AC TYPE 120/150

MODEL NO. _____												AC TYPE <u>120/150</u>	
TYPE _____													
1	2	3	4	5		6	7	8	9		10	11	12
Time (24 HR)	Fuel Use #Fuel Oil <input type="checkbox"/> Nat. Gas <input checked="" type="checkbox"/> Propane <input type="checkbox"/> Coal <input type="checkbox"/> other <input type="checkbox"/>	Burner Setting	Blower Pressure	Production Rate		Asphalt Cement %	Mix Temp. °F	Exhaust Gas Temp. °F	Venturi Scrubber <input checked="" type="checkbox"/> Baghouse		Ambient Temp. °F	Relative Humidity %	Exhaust Dampers Position
				Mix Aggregate TPH	RAP TPH				Pressure Drop In w.g.	Water Pressure psi			
<u>20:15</u>		<u>95</u>		<u>375</u>	<u>102</u>	<u>4.6</u>	<u>290°</u>	<u>355°</u>	<u>3</u>		<u>90°</u>		<u>60</u>
<u>20:30</u>		<u>95</u>		<u>375</u>	<u>103</u>	<u>4.6</u>	<u>295°</u>	<u>363°</u>	<u>3</u>		<u>90°</u>		<u>60</u>
<u>20:45</u>		<u>95</u>		<u>375</u>	<u>102</u>	<u>4.6</u>	<u>290°</u>	<u>354°</u>	<u>3</u>		<u>90°</u>		<u>60</u>
<u>21:00</u>		<u>95</u>		<u>375</u>	<u>102</u>	<u>4.6</u>	<u>295°</u>	<u>357°</u>	<u>3</u>		<u>90°</u>		<u>48</u>
<u>21:15</u>		<u>95</u>		<u>375</u>	<u>103</u>	<u>4.6</u>	<u>290°</u>	<u>355°</u>	<u>3</u>		<u>90°</u>		<u>48</u>
								</					

DATA SUMMARY ON STACK BEING TESTED

AGGREGATE

1. Name/type of mix Base Mix code 505
2. Name/type of 2nd mix (if used) _____
3. Type/temperature of Liquid Asphalt 120/150 / 333 °F
4. Sieve/Screening analysis: _____ % Passing;

1st mix / 2nd mix	1st mix / 2nd mix	1st mix / 2nd mix
1" _____	3/8" _____	# _____
3/4" _____	#200 _____	# _____
1/2" _____	# _____	# _____

CONTROL SYSTEM

Manufacturer _____

A. Baghouse:

1. Type of bags _____ # of bags _____ Sq. ft. of bags _____
2. Air to cloth ratio _____ Designed ACFM _____
3. Type of cleaning - pulse jet ☒ reverse air _____ plenum pulse _____ other _____
4. Cleaning cycle time _____ Interval between cleaning cycle _____
5. Pulse pressure on cleaning cycle 120 psi

B. Scrubber:

1. Type - Venturi _____ Wet Washer _____
Spray Booth _____ Other _____
2. Gallons per minute through system _____
3. Water source _____ (i.e., pond, lagoon, etc.)
4. Number of spray nozzles _____

Company Name Atax Materials Date 8-10-92

Company Representative Mark Boden

TODAY'S DATE: 7-7-92

COMPANY NAME ATAX MATERIALS COMPANY REP. _____ PHONE (313) 946 8080
LOCATION OF FACILITY Romulus MI ORIGINAL START-UP DATE _____ DESIGNED CAPACITY _____
OEM _____ MODEL NO. _____ TYPE _____ AC TYPE AC 5

[illegible]

DATA SUMMARY ON STACK BEING TESTED

AGGREGATE

1. Name/type of mix _____
2. Name/type of 2nd mix (if used) _____
3. Type/temperature of Liquid Asphalt _____ / _____ °F
4. Sieve/Screening analysis: _____ % Passing; _____ Moisture on Aggregate

1st mix / 2nd mix	1st mix / 2nd mix	1st mix / 2nd mix
1" _____ / _____	3/8" _____ / _____	# _____ / _____
3/4" _____ / _____	#200 _____ / _____	# _____ / _____
1/2" _____ / _____	# _____ / _____	# _____ / _____

CONTROL SYSTEM

Manufacturer: _____

- A. Baghouse:
1. Type of bags: 402 Virgin Nomex of bags 705 Sq. ft. of bags 17293
 2. Air to cloth ratio: 5.03/1 ~~Design~~ Actual ACFM 87,000
 3. Type of cleaning - pulse jet ☒ reverse air _____ plenum pulse _____ other _____
 4. Cleaning cycle time: _____ Interval between cleaning cycle: 10 sec
 5. Pulse pressure on cleaning cycle: 110 psi

- B. Scrubber:
1. Type - Venturi: _____ Wet Washer: _____
Spray Booth: _____ Other: _____
 2. Gallons per minute through system: _____
 3. Water source: _____ (i.e., pond, lagoon, etc.)
 4. Number of spray nozzles: _____

Company Name: Alax Materials Corporation Date: 07/07/92
Company Representative: Mark E. Boden

DATA SUMMARY ON STACK BEING TESTED

AGGREGATE

1. Name/type of mix _____
2. Name/type of 2nd mix (if used) _____
3. Type/temperature of Liquid Asphalt _____ / _____ °F
4. Sieve/Screening analysis: _____ % Passing; _____ Moisture on Aggregate

1st mix / 2nd mix	1st mix / 2nd mix	1st mix / 2nd mix
1" _____ / _____	3/8" _____ / _____	# _____ / _____
3/4" _____ / _____	#200 _____ / _____	# _____ / _____
1/2" _____ / _____	# _____ / _____	# _____ / _____

CONTROL SYSTEM

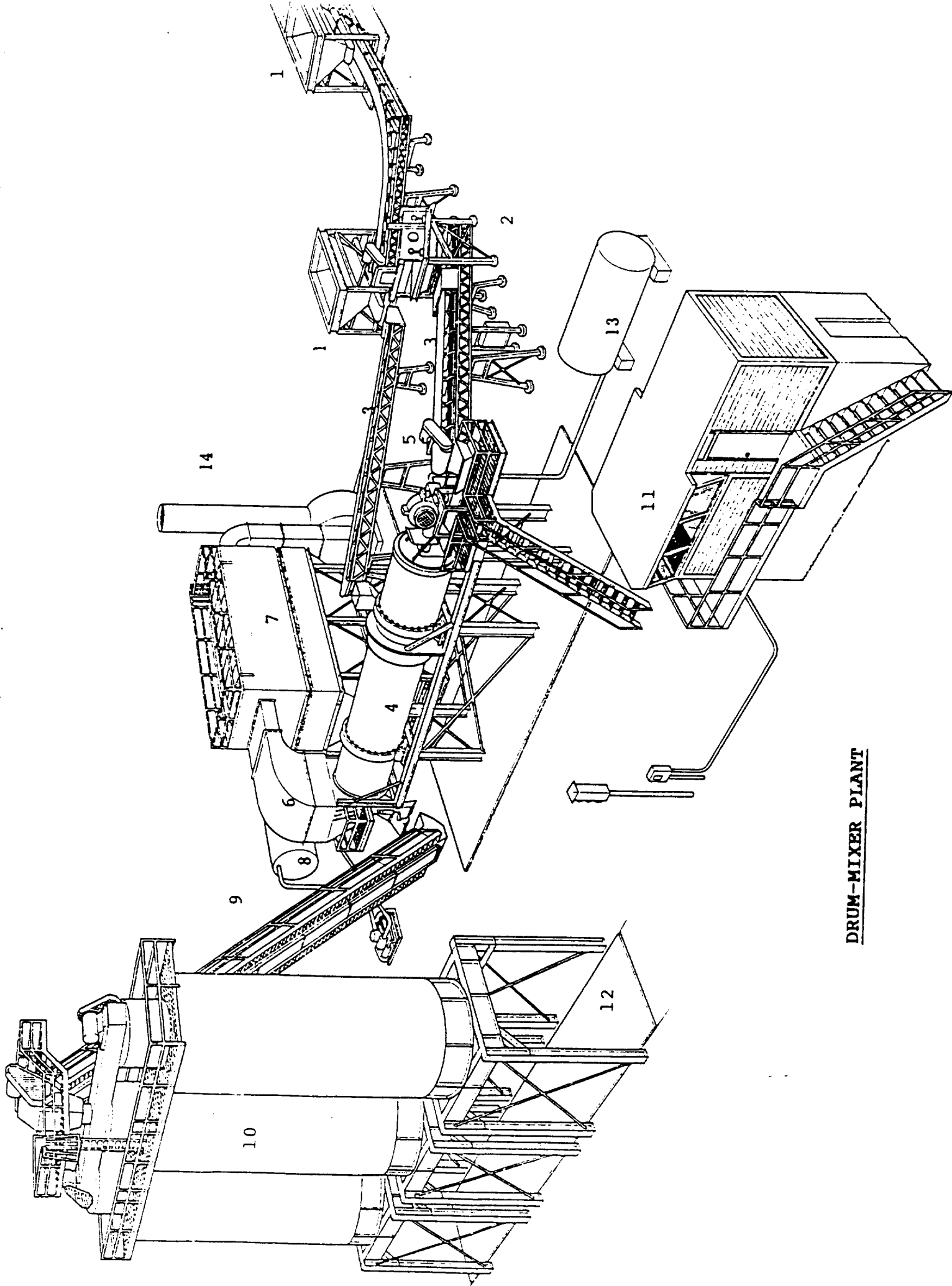
Manufacturer: _____

- A. Baghouse:
1. Type of bags: 1402 Virgin Nomex # of bags 705 Sq. ft. of bags 17293
 2. Air to cloth ratio: 5.03/1 ~~Designed~~ Actual ACFM 87,000
 3. Type of cleaning - pulse jet ☒ reverse air _____ plenum pulse _____ other _____
 4. Cleaning cycle time: _____ Interval between cleaning cycle: 10 sec
 5. Pulse pressure on cleaning cycle: 110 psi

B. Scrubber:

1. Type - Venturi: _____ Wet Washer: _____
Spray Booth: _____ Other: _____
2. Gallons per minute through system: _____
3. Water source: _____ (i.e., pond, lagoon, etc.)
4. Number of spray nozzles: _____

Company Name: Alax Materials Corporation Date: 07/07/92
Company Representative: Mark E. Boden



DRUM-MIXER PLANT

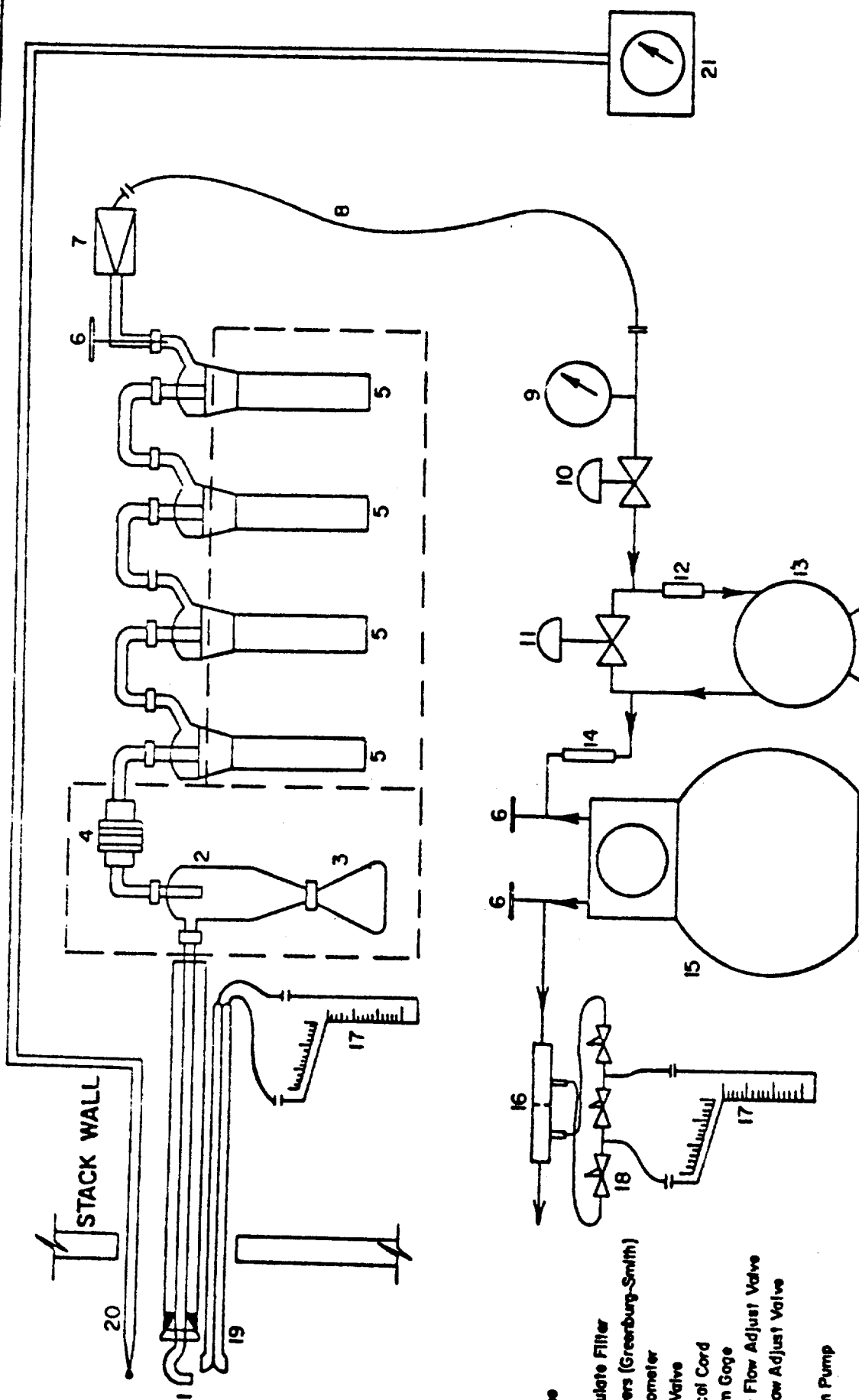
1. Aggregate bins: Virgin aggregate is fed individually into bins by type. It is metered onto a conveyor belt running under the bins to a shaker screen. The proportion to each aggregate type is determined by the job mix formula and pre-set to be metered out to meet these specifications.
2. Preliminary oversize screen: The aggregate is fed through a shaker screen where oversize rocks and foreign material is screened out of the mix.
3. Weigh conveyor belt: The aggregate is conveyed to the rotary drum dryer on a conveyor belt which weighs the material. The production rate is determined by this weight reading.
4. Rotary drum/dryer mixer: The aggregate is fed into the rotary drum dryer where it is tumbled by flighting into a veil in front of a flame which drives off the moisture. Further mixing is also accomplished in an outer shell of this drum. Hot liquid asphalt is injected in the outer shell of the drum where it is mixed with the aggregate.
5. Burner: The fuel fired burner is used to provide the flame which dries the aggregate.
6. Knock off baffling: A baffling plate is inserted in the "dirty" side plenum as a knock out for heavy particles in the air stream. These particles fall to the bottom of the baghouse.
7. Baghouse: The hot gases are pulled through the bags into the clean air plenum. The solid particulate matter is trapped on the dust coat buildup on the bags. A bag cleaning cycle consisting of jet burst of air from the inside (or clean air side) of the bags sends a large bubble of air down the inside of the bags shaking loose buildup on the bag surface. This particulate matter is collected at the bottom of the baghouse and reinjected into the drum mixer where it is used as part of the finished product.
8. Liquid asphalt storage: The liquid asphalt is stored in this heated tank until it is needed in the mixer. The amount of asphalt content and its temperature are pre-set for each different type job.
9. Conveyor to surge/storage bin: The finished product of aggregate mixed with liquid asphalt is conveyed to a surge bin.
10. Surge/Storage bin: The asphaltic cement is dumped into this surge bin and metered out to dump trucks which pull underneath a slide gate at the bottom of the bin.
11. Control/operators house: The entire plant operation is controlled from this operator's house.
12. Truck loading scale: As the trucks receive the asphalt from the storage/surge bin, they are weighed on the lading scale which tells the plant operator the amount of asphalt that is being trucked on each individual load.
13. Fuel storage.
14. Stack

SECTION C:
EQUIPMENT USED

V. EQUIPMENT USED

Equipment used on conducting the particulate emissions test was:

- A. The Lear Siegler PM-100 stack sampler with appropriate auxillary equipment and glassware. The train was set up according to the schematic on the nex page.
- B. An Airguide Instruments Model 211-B (uncorrected) aneroid barometer was used to check the barometric pressure.
- C. Weston dial thermometers are used to check meter temperatures. An Analogic Model 2572 Digital Thermocouple is used for stack temperatures.
- D. A Hays 621 Analyzer was used to measure the oxygen, carbon dioxide and carbon monoxide content of the stack gases. For non-combustion sources, A Bacharach Instrument Company Fyrite is used for the gas analysis.
- E. Filters are mady by Schleicher and Schuell and are type 1-HV with a porosity of .03 microns.
- F. The acetone is reagent grade or ACS grade with a residue of \leq .001.



**SAMPLING TRAIN
USED FOR ISOKINETIC SAMPLING**

- 1) Probe
- 2) Cyclone
- 3) Flask
- 4) Particulate Filter
- 5) Impingers (Greenburg-Smith)
- 6) Thermometer
- 7) Check Valve
- 8) Umbilical Cord
- 9) Vacuum Gage
- 10) Course Flow Adjust Valve
- 11) Fine Flow Adjust Valve
- 12) Oil
- 13) Vacuum Pump
- 14) Filter
- 15) Dry Gas Meter
- 16) Orifice Tube
- 17) Incline Manometer
- 18) Solenoid Valves
- 19) Pilot
- 20) Thermocouple
- 21) Pyrometer

SECTION D:
LABORATORY PROCEDURES AND RESULTS

LABORATORY PROCEDURES FOR PARTICULATE SAMPLING

I. Field Preparation

A. FILTERS: Fiberglass 4" sampling filters are prepared as follows:

Filters are removed from their box and numbered on the back side with a felt pen. The numbering system is continuous from job to job. The filters are placed in a desiccator to dry for at least 24 hours. Clean plastic petri dishes, also numbered, top and bottom, are placed in the desiccator with the filters. After desiccation, the filters are removed, one at a time, and weighed on the Sartorius analytical balance then placed in the correspondingly numbered petri dish. Weights are then recorded in the lab record books. Three filters are used for each complete particulate source emissions test and there should be several extra filters included as spares.

B. SILICA GEL: Silica Gel used for the test is prepared as follows:

Approximately 200 g of silica gel is placed in a wide mouth "Mason" type jar and dried in an oven at 175°C for two hours. The open jars are removed and placed in a desiccator until cool for two hours and then tightly sealed. The jars are then numbered and weighed on the triple beam balance to the closest tenth of a gram. This weight is recorded for each sealed jar. The number of silica gel jars used is the same as the number of filters. Silica gel should be indicating type, 6-16 mesh.

II. Post - Testing Lab Analysis

A. FILTERS: The filters are returned to the lab in their sealed petri dishes. In the lab, the dishes are opened and placed into a desiccator for at least 24 hours. Then the filters are weighed continuously every six hours until a constant weight is achieved. All data is recorded on the laboratory forms that will be bound in the test report.

B. SILICA GEL: The silica gel used in the stack test is returned to the appropriate mason jar and sealed for transport to the laboratory where it is reweighed to a constant weight on a triple beam balance to the nearest tenth of a gram.

- C. **PROBE RINSINGS:** In all tests where a probe washout analysis is necessary, this is accomplished in accordance with procedures specified in "EPA Reference Method 5". These samples are returned to the lab in sealed mason jars for analysis. The front half of the filter holder is washed in accordance with the same procedures and included with the probe wash. Reagent or ACS grade acetone is used as the solvent. The backhalf of the filter holder is washed with deionized water into the impinger catch for appropriate analysis.
- D. **IMPINGER CATCH:** In some testing cases, the liquid collected in the impingers must be analyzed for solid content. This involves a similar procedure to the probe wash solids determination, except that the liquid is deionized water.
- E. **ACETONE:** A blank analysis of acetone is conducted from the one gallon glass container used in the field preparation. This acetone was used in the field for rinsing the probe, nozzle, and top half of the filter holder. A blank analysis is performed prior to testing on all new containers of acetone received from the manufacturer to insure that the quality of the acetone used will be exceed the .001% residual purity standard.

SPECIAL NOTE

When sampling sources high in moisture content, (such as asphalt plants) the filter paper sometimes sticks to the filter holder. When removing the filter, it may tear. In order to maintain control of any small pieces of filter paper which may be easily lost, they are washed with acetone into the probe washing. This makes the filter weight light (sometimes negative) and the probe wash correspondingly heavier. this laboratory procedure is taught by EPA in the "Quality Assurance for Source Emissions Workshop" at Research Triangle Park and is approved by EPA.

WEIGHING PROCEDURE - SARTORIUS ANALYTICAL BALANCE

The Sartorius balance is accurate to 0.1 mg and has a maximum capacity of 200 grams. The balance precision (standard deviation) is 0.05 mg. Before weighing an item, the balance should first be zeroed. This step should be taken before every series of weighings. To do this, the balance should have all weight adjustments at the "zero" position. The beam arrest lever (on the lower left hand side toward the rear of the balance) is then slowly pressed downward to the full release position. The lighted vernier scale on the front of the cabinet should align with the "zero" with the mark on the cabinet. If it is not so aligned, the adjustment knob on the right hand side (near the rear of the cabinet) should be turned carefully until the marks align. Now return the beam arrest to the horizontal arrest position. The balance is now "zeroed".

To weigh an item, it is first placed on the pan. And the sliding doors are closed to avoid air current disturbance. The weight adjustment knob on the right hand side must be at "zero". The beam arrest is then slowly turned upward. The lighted scale at the front of the cabinet will now indicate the weight of the item in grams. If the scale goes past the divided area, the item then exceeds 100 g weight (about 3 1/2 ounces) and it is necessary to arrest the balance (beam arrest lever) and move the lever for 100 g weight away from you. It is located on the left hand side of the cabinet near the front, and is the knob closest to the side of the cabinet. The balance will not weigh items greater than 200 grams in mass, and trying to do this might harm the balance. Remember, this is a delicate precision instrument.

After the beam is arrested in either weight range, the procedure is the same. When the weight of the item in grams is found, "dial in" that amount with the two knobs on the left hand side (near the 100 g lever) color coded yellow and green. As you dial the weight, the digits will appear on the front of the cabinet. When the proper amount is dialed, carefully move the arrest lever down with a slow, steady turn of the wrist. The lighted dial will appear, and the right hand side knob (front of cabinet) is turned to align the mark with the lower of the two lighted scale divisions which the mark appears between. When these marks are aligned, the two lighted digits along with the two indicated on the right hand window on the cabinet front are fractional weight in grams (the decimal would appear before the lighted digits) and the whole number of grams weight is the amount "dialed in" on the left.

In general, be sure that the beam is in "arrest" position before placing weight on or taking weight off of the pan. Don't "dial in" weight unless the beam is arrested. The balance is sensitive to even a hand on the table near the balance, so be careful and painstaking in every movement while weighing.

SAMPLE ANALYTICAL DATA FORM

Company Name AJAXSample Location ROMULUS MIBlank Volume (V_a) 100 mlDate/Time wt. blank 8/15 11:00 ADate/Time wt. blank 8/17 3:00 ARelative Humidity in Lab 49 %Density of Acetone (ρ_a) .7857 mg/mlGross wt. 101.3842 gGross wt. 101.3842 gAve. Gross wt. 101.3842 gTare wt. 101.3840 gWeight of blank (m_{ab}) .0002 gAcetone blank residue concentration (C_a): $(C_a) = (m_{ab}) / (V_a) (\rho_a) = (.000003 \text{ mg/g})$ Acetone Blank Wt.: $W_a = C_a V_{aw} \rho_a = (.000003)(200)(.7857) = (.0005 \text{ g})$

	Run # 1	Run # 2	Run # 3
Acetone rinse volume (V_{aw}) ml	<u>200</u>	<u>200</u>	<u>200</u>
Date/Time of wt. <u>8/15 11:00 A</u> Gross wt. g	<u>101.1356</u>	<u>101.4372</u>	<u>101.6872</u>
Date/Time of wt. <u>8/17 3:00 A</u> Gross wt. g	<u>101.1355</u>	<u>101.4369</u>	<u>101.6870</u>
Average Gross wt. g	<u>101.1356</u>	<u>101.4371</u>	<u>101.6871</u>
Tare wt. g	<u>101.1176</u>	<u>101.3992</u>	<u>101.6562</u>
Less Acetone blank wt. (W_a) g	<u>.0005</u>	<u>.0005</u>	<u>.0005</u>
Weight of particulate in acetone rinse (m_a) g	<u>.0175</u>	<u>.0374</u>	<u>.0304</u>

Filter Numbers	#	JS 6175	JS 6176	JS 6171
Date/Time of wt. <u>8/15 11:00 A</u> Gross wt. g		<u>5836</u>	<u>5895</u>	<u>5838</u>
Date/Time of wt. <u>8/17 3:00 A</u> Gross wt. g		<u>5836</u>	<u>5895</u>	<u>5837</u>
Average Gross wt. g		<u>5836</u>	<u>5895</u>	<u>5838</u>
Tare wt. g		<u>5778</u>	<u>5764</u>	<u>5717</u>

Weight of particulate on filter (m_f) g	<u>.0058</u>	<u>.0131</u>	<u>.0121</u>
Weight of particulate in acetone rinse (m_a) g	<u>.0175</u>	<u>.0374</u>	<u>.0304</u>
Total weight of particulate (m_n) g	<u>.0233</u>	<u>.0505</u>	<u>.0425</u>

NOTE: In no case should a blank residue greater than 0.01 mg/g (or 0.001% of the blank weight) be subtracted from the sample weight.

Remarks: _____

Signature of Analyst Thomas Smith

Signature of Reviewer _____

SAMPLE ANALYTICAL DATA FORM

Company Name ATAXSample Location Panopolis, MIBlank Volume (V_a) 100 mlDate/Time wt. blank 7/13 8:00 ADate/Time wt. blank 7/13 2:00 PRelative Humidity in Lab 48 %Density of Acetone (ρ_a) .7857 mg/mlGross wt. 169.4030 gGross wt. 169.4025 gAve. Gross wt. 169.4028 gTare wt. 169.4026 gWeight of blank (m_{ab}) .0002 gAcetone blank residue concentration (C_a): $(C_a) = (m_{ab}) / (V_a) (\rho_a) = (.000003 \text{ mg/g})$ Acetone Blank Wt.: $W_a = C_a V_{aw} \rho_a = (.000003)(200)(.7857) = (.0005 \text{ g})$

	Run # 1	Run #	Run #
Acetone rinse volume (V_{aw}) ml	<u>200</u>		
Date/Time of wt. <u>7/13 8:00 A</u> Gross wt. g	<u>98.2795</u>		
Date/Time of wt. <u>7/13 2:00 P</u> Gross wt. g	<u>98.2793</u>		
Average Gross wt. g	<u>98.2794</u>		
Tare wt. g	<u>98.2550</u>		
Less Acetone blank wt. (W_a) g	<u>.0005</u>		
Weight of particulate in acetone rinse (m_a) g	<u>.0239</u>		

Filter Numbers	#		
Date/Time of wt. <u>7/13 8:00 A</u> Gross wt. g	<u>T56274</u>		
Date/Time of wt. <u>7/13 2:00 P</u> Gross wt. g	<u>.5678</u>		
Average Gross wt. g	<u>.5673</u>		
Tare wt. g	<u>.5676</u>		
	<u>.5631</u>		

Weight of particulate on filter (m_f) g	<u>.0045</u>		
Weight of particulate in acetone rinse (m_a) g	<u>.0239</u>		
Total weight of particulate (m_n) g	<u>.0284</u>		

NOTE: In no case should a blank residue greater than 0.01 mg/g (or 0.001% of the blank weight) be subtracted from the sample weight.

Remarks: _____

Signature of Analyst James South

Signature of Reviewer _____

Qjos
Company Name

8-10-92
Date

REFERENCE METHOD 3: GAS ANALYSIS BY FYRITE

<u>FUEL</u>	<u>F_o FACTORS</u>
WOOD	1.0540
BARK	1.0830
ANTHRACITE	1.0699
BITUMINOUS	1.1398
LIGNITE	1.0761
OIL	1.3465
GAS	1.7489
PROPANE	1.5095
BUTANE	1.4791

$$O_2\% = 20.9 - [F_o \times CO_2\%]$$

RUN #1: _____ = 20.9 - [_____ x _____]

RUN #2: _____ = 20.9 - [_____ x _____]

RUN #3: _____ = 20.9 - [_____ x _____]

RUN 1:	CO _{2x} <u>5.8</u>	CO _{2x} <u>5.8</u>	CO _{2x} <u>5.8</u>	AVG. <u>5.8</u>
	O _{2x} <u>14.0</u>	O _{2x} <u>14.0</u>	O _{2x} <u>14.0</u>	AVG. <u>14.0</u>
	N _{2x} _____	N _{2x} _____	N _{2x} _____	AVG. _____
RUN 2:	CO _{2x} <u>5.4</u>	CO _{2x} <u>5.0</u>	CO _{2x} <u>5.0</u>	AVG. <u>5.0</u>
	O _{2x} <u>14.5</u>	O _{2x} <u>14.5</u>	O _{2x} <u>14.5</u>	AVG. <u>14.5</u>
	N _{2x} _____	N _{2x} _____	N _{2x} _____	AVG. _____
RUN 3:	CO _{2x} <u>5.0</u>	CO _{2x} <u>5.0</u>	CO _{2x} <u>5.0</u>	AVG. <u>5.0</u>
	O _{2x} <u>14.0</u>	O _{2x} <u>14.0</u>	O _{2x} <u>14.0</u>	AVG. <u>14.0</u>
	N _{2x} _____	N _{2x} _____	N _{2x} _____	AVG. _____

Apex Paving
Company Name

7/7
Date

REFERENCE METHOD 3: GAS ANALYSIS BY FYRITE

FUEL

F_o FACTORS

WOOD	1.0540
BARK	1.0830
ANTHRACITE	1.0699
BITUMINOUS	1.1398
LIGNITE	1.0761
OIL	1.3465
GAS	1.7489
PROPANE	1.5095
BUTANE	1.4791

$$O_2\% = 20.9 - [F_o \times CO_2\%]$$

RUN #1: _____ = 20.9 - [_____ x _____]

RUN #2: _____ = 20.9 - [_____ x _____]

RUN #3: _____ = 20.9 - [_____ x _____]

RUN 1: CO_{2x} _____ CO_{2x} _____ CO_{2x} _____ AVG. _____

O_{2x} _____ O_{2x} _____ O_{2x} _____ AVG. _____

N_{2x} _____ N_{2x} _____ N_{2x} _____ AVG. _____

RUN 2: CO_{2x} _____ CO_{2x} _____ CO_{2x} _____ AVG. _____

O_{2x} _____ O_{2x} _____ O_{2x} _____ AVG. _____

N_{2x} _____ N_{2x} _____ N_{2x} _____ AVG. _____

RUN 3: CO_{2x} _____ CO_{2x} _____ CO_{2x} _____ AVG. _____

O_{2x} _____ O_{2x} _____ O_{2x} _____ AVG. _____

N_{2x} _____ N_{2x} _____ N_{2x} _____ AVG. _____

SECTION E:
CALCULATIONS

**AJAX PAVING INDUSTRIES
DETROIT, MI**

SUMMARY OF TEST DATA

SAMPLING TRAIN DATA

8-10-92 8-10-92 8-10-92
RUN #1 RUN #2 RUN #3

	start	02:46	06:18	08:09
	finish	03:53	07:37	09:23
1. Sampling time, minutes	Θ	60.0	60.0	60.0
2. Sampling nozzle diameter, in.	D_n	.2750	.2750	.2750
3. Sampling nozzle cross-sect. area, ft ²	A_n	.000412	.000412	.000412
4. Isokinetic variation	I	111.1	138.2	128.7
5. Sample gas volume - meter cond., cf.	V_m	51.027	60.800	58.924
6. Average meter temperature, °R	T_m	562	567	566
7. Avg. orifice pressure drop, in. H ₂ O	dH	1.96	2.05	2.07
8. Total particulate collected, mg.	M_n	23.30	50.50	42.50

VELOCITY TRAVERSE DATA

9. Stack area, ft ²	A	21.00	21.00	21.00
10. Absolute stack gas pressure, in. Hg.	P_s	30.00	30.00	30.00
11. Barometric pressure, in. Hg.	P_{bar}	30.00	30.00	30.00
12. Avg. absolute stack temperature, R°	T_s	739	754	756
13. Average $-\sqrt{v_e l} \cdot \bar{h} \bar{e} \bar{a} \bar{d}$, ($C_p = .84$)	$-\sqrt{dP}$	0.78	0.77	0.81
14. Average stack gas velocity, ft./sec.	V_s	53.78	54.01	57.00

STACK MOISTURE CONTENT

15. Total water collected by train, ml.	V_{ic}	325.00	441.00	444.00
16. Moisture in stack gas, %	B_{ws}	24.10	26.67	27.44

EMISSIONS DATA

17. Stack gas flow rate, dscf/hr.(000's)	Q_{sd}	2210	2102	2189
18. Stack gas flow rate, cfm	acfm	67763	68053	71820
19. Particulate concentration, gr/dscf	C_s	0.0074	0.0137	0.0118
20. Particulate concentration, lb/hr	E	2.34	4.11	3.69
21. Particulate concentration, lb/mBtu	E'	0.00000	0.00000	0.00000

ORSAT DATA

22. Percent CO ₂ by volume	CO ₂	5.80	5.00	5.00
23. Percent O ₂ by volume	O ₂	14.00	14.50	14.00
24. Percent CO by volume	CO	.00	.00	.00
25. Percent N ₂ by volume	N ₂	80.20	80.50	81.00

Dry Gas Volume

$$V_{m(std)} = V_m \left[\frac{T_{(std)}}{T_m} \right] \left[\frac{P_{bar} + \frac{dH}{13.6}}{P_{(std)}} \right] = 17.64 \frac{^{\circ}R}{in.Hg} Y V_m \left[\frac{P_{bar} + \frac{dH}{13.6}}{T_m} \right]$$

Where:

- $V_{m(std)}$ = Dry Gas Volume through meter at standard conditions, cu. ft.
 V_m = Dry Gas Volume measured by meter, cu. ft.
 P_{bar} = Barometric pressure at orifice meter, in. Hg.
 P_{std} = Standard absolute pressure, (29.92 in. Hg.).
 T_m = Absolute temperature at meter $^{\circ}R$.
 T_{std} = Standard absolute temperature (528 $^{\circ}R$).
 dH = Average pressure drop across orifice meter, in. H₂O.
 Y = Dry gas meter calibration factor.
13.6 = Inches water per inches Hg.

RUN 1:

$$V_{m(std)} = (17.64)(.998)(51.027) \left[\frac{(30.00) + \frac{1.96}{13.6}}{562} \right] = 48.183 \text{ dscf}$$

RUN 2:

$$V_{m(std)} = (17.64)(.998)(60.800) \left[\frac{(30.00) + \frac{2.05}{13.6}}{567} \right] = 56.918 \text{ dscf}$$

RUN 3:

$$V_{m(std)} = (17.64)(.998)(58.924) \left[\frac{(30.00) + \frac{2.07}{13.6}}{566} \right] = 55.262 \text{ dscf}$$

Total Contaminants by Weight: GRAIN LOADING

Particulate concentration C'_s gr./dscf.

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{M_n}{V_{m(\text{std})}} \right]$$

Where:

C'_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr./dscf.

M_n = Total amount of particulate matter collected, mg.

$V_{m(\text{std})}$ = Dry gas volume through meter at standard conditions, cu. ft.

Run 1:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{23.30}{48.183} \right] = 0.0074 \text{ gr./dscf.}$$

Run 2:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{50.50}{56.918} \right] = 0.0137 \text{ gr./dscf.}$$

Run 3:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{42.50}{55.262} \right] = 0.0118 \text{ gr./dscf.}$$

$$M_d = 0.44(\%CO_2) + 0.32(\%O_2) + 0.28(\%CO + \%N_2)$$

Where:

- M_d = Dry molecular weight, lb./lb.-mole.
 $\%CO_2$ = Percent carbon dioxide by volume (dry basis).
 $\%O_2$ = Percent oxygen by volume (dry basis).
 $\%N_2$ = Percent nitrogen by volume (dry basis).
 $\%CO$ = Percent carbon monoxide by volume (dry basis).
0.264 = Ratio of O_2 to N_2 in air, v/v.
0.28 = Molecular weight of N_2 or CO, divided by 100.
0.32 = Molecular weight of O_2 divided by 100.
0.44 = Molecular weight of CO_2 divided by 100.

Run 1:

$$M_d = 0.44(5.80\%) + 0.32(14.00\%) + 0.28(.00\% + 80.20\%) = 29.49 \frac{lb}{lb-mole}$$

Run 2:

$$M_d = 0.44(5.00\%) + 0.32(14.50\%) + 0.28(.00\% + 80.50\%) = 29.38 \frac{lb}{lb-mole}$$

Run 3:

$$M_d = 0.44(5.00\%) + 0.32(14.00\%) + 0.28(.00\% + 81.00\%) = 29.36 \frac{lb}{lb-mole}$$

$$V_{wc_{std}} = \left[V_f - V_i \right] \left[\frac{P_w R T_{(std)}}{M_w P_{(std)}} \right] = 0.04707 \left[V_f - V_i \right]$$

$$V_{wsg_{std}} = \left[W_f - W_i \right] \left[\frac{R T_{(std)}}{M_w P_{(std)}} \right] = 0.04715 \left[W_f - W_i \right]$$

Where:

0.04707 = Conversion factor, ft.³/ml.

0.04715 = Conversion factor, ft.³/g.

$V_{wc_{std}}$ = Volume of water vapor condensed (standard conditions), scf.

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel (standard conditions), ml.

$V_f - V_i$ = Final volume of impinger contents less initial volume, ml.

$W_f - W_i$ = Final weight of silica gel less initial weight, g.

P_w = Density of water, 0.002201 lb/ml.

R = Ideal gas constant, 21.85 in.Hg. (cu.ft./lb.-mole)(°R).

M_w = Molecular weight of water vapor, 18.0 lb/lb-mole.

T_{std} = Absolute temperature at standard conditions, 528°R.

P_{std} = Absolute pressure at standard conditions, 29.92 inches Hg.

Run 1:

$$\begin{aligned} V_{wc(std)} &= (0.04707) (313.0) = 14.7 \text{ cu.ft} \\ V_{wsg(std)} &= (0.04715) (12.0) = 0.6 \text{ cu.ft} \end{aligned}$$

Run 2:

$$\begin{aligned} V_{wc(std)} &= (0.04707) (430.0) = 20.2 \text{ cu.ft} \\ V_{wsg(std)} &= (0.04715) (11.0) = 0.5 \text{ cu.ft} \end{aligned}$$

Run 3:

$$\begin{aligned} V_{wc(std)} &= (0.04707) (435.0) = 20.5 \text{ cu.ft} \\ V_{wsg(std)} &= (0.04715) (9.0) = 0.4 \text{ cu.ft} \end{aligned}$$

Moisture Content of Stack Gases

$$B_{ws} = \frac{V_{wc_{std}} + V_{wsg_{std}}}{V_{wc_{std}} + V_{wsg_{std}} + V_{m_{std}}} \times 100$$

Where:

B_{ws} = Proportion of water vapor, by volume, in the gas stream.

V_m = Dry gas volume measured by dry gas meter, (dcf).

$V_{wc_{std}}$ = Volume of water vapor condensed corrected to standard conditions (scf).

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel corrected to standard conditions (scf).

Run 1:

$$B_{ws} = \frac{14.7 + 0.6}{14.7 + 0.6 + 48.183} \times 100 = 24.10 \%$$

Run 2:

$$B_{ws} = \frac{20.2 + 0.5}{20.2 + 0.5 + 56.918} \times 100 = 26.67 \%$$

Run 3:

$$B_{ws} = \frac{20.5 + 0.4}{20.5 + 0.4 + 55.262} \times 100 = 27.44 \%$$

Molecular Weight of Stack Gases

$$M_s = M_d (1 - B_{ws}) + 18 (B_{ws})$$

Where:

M_s = Molecular weight of stack gas, wet basis, (lb./lb.-mole).

M_d = Molecular weight of stack gas, dry basis, (lb./lb.-mole).

Run 1:

$$M_s = 29.49 (1 - 24.10) + 18 (24.10) = 26.72 \text{ (lb./lb.-mole)}$$

Run 2:

$$M_s = 29.38 (1 - 26.67) + 18 (26.67) = 26.34 \text{ (lb./lb.-mole)}$$

Run 3:

$$M_s = 29.36 (1 - 27.44) + 18 (27.44) = 26.24 \text{ (lb./lb.-mole)}$$

$$V_s = K_p C_p \left[-\sqrt{dP} \right] \text{ avg. } -\sqrt{\frac{T_{s(\text{avg.})}}{P_s M_s}}$$

Where:

- V_s = Average velocity of gas stream in stack, ft./sec.
 K_p = 85.49 ft/sec $\left[\frac{(\text{g/g-mole}) - (\text{mm Hg})}{(^{\circ}\text{K})(\text{mm H}_2\text{O})} \right]^{1/2}$
 C_p = Pitot tube coefficient, (dimensionless).
 dP = Velocity head of stack gas, in. H_2O .
 P_{bar} = Barometric pressure at measurement site, (in. Hg).
 P_g = Stack static pressure, (in. Hg).
 P_s = Absolute stack gas pressure, (in. Hg) = $P_{\text{bar}} + P_g$
 P_{std} = Standard absolute pressure, (29.92 in. Hg).
 t_s = Stack temperature, ($^{\circ}\text{f}$).
 T_s = Absolute stack temperature, ($^{\circ}\text{R}$). = $460 + t_s$.
 M_s = Molecular weight of stack gas, wet basis, (lb/lb-mole).

Run 1:

$$V = (85.49) (.84) (0.78) -\sqrt{\frac{739}{(30.00)(26.72)}} = 53.78 \text{ ft/sec.}$$

Run 2:

$$V = (85.49) (.84) (0.77) -\sqrt{\frac{754}{(30.00)(26.34)}} = 54.01 \text{ ft/sec.}$$

Run 3:

$$V = (85.49) (.84) (0.81) -\sqrt{\frac{756}{(30.00)(26.24)}} = 57.00 \text{ ft/sec.}$$

$$Q_{sd} = 3600 \left[1 - B_{wc} \right] V_s A \left[\frac{T_{std}}{T_{stk}} \right] \left[\frac{P_s}{P_{std}} \right]$$

Where:

- Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, (dscf/hr).
 A = Cross sectional area of stack, (ft.²).
 3600 = Conversion factor, (sec./hr.).
 t_s = Stack temperature, (°f).
 T_s = Absolute stack temperature, (°R).
 T_{std} = Standard absolute temperature, (528°R).
 P_{bar} = Barometric pressure at measurement site, (in.Hg.).
 P_g = Stack static pressure, (in.Hg.).
 P_s = Absolute stack gas pressure, (in.Hg.); = $P_{bar} + P_g$
 P_{std} = Standard absolute pressure, (29.92 in.Hg.).

Run 1:

$$Q_{sd} = 3600(1 - .2410)(53.78)(21.00) \left[\frac{528}{739} \right] \left[\frac{30.00}{29.92} \right] = 2210718.9 \frac{\text{dscf}}{\text{hr}}$$

Run 2:

$$Q_{sd} = 3600(1 - .2667)(54.01)(21.00) \left[\frac{528}{754} \right] \left[\frac{30.00}{29.92} \right] = 2102325.2 \frac{\text{dscf}}{\text{hr}}$$

Run 3:

$$Q_{sd} = 3600(1 - .2744)(57.00)(21.00) \left[\frac{528}{756} \right] \left[\frac{30.00}{29.92} \right] = 2189604.7 \frac{\text{dscf}}{\text{hr}}$$

Emissions Rate from Stack

$$E = \frac{(C_s)(Q_{sd})}{7000 \text{ gr./lb.}} = \text{lb. / hr.}$$

Where:

E = Emissions rate, lb/hr.

C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr/dscf.

Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, dscf/hr.

Run 1:

$$E = \frac{(0.0074)(2210718.9)}{7000} = 2.34 \text{ lb. / hr.}$$

Run 2:

$$E = \frac{(0.0137)(2102325.2)}{7000} = 4.11 \text{ lb. / hr.}$$

Run 3:

$$E = \frac{(0.0118)(2189604.7)}{7000} = 3.69 \text{ lb. / hr.}$$

$$I = 100 T_s \left[\frac{0.002669 V_{ic} + \frac{(V_m / T_m) (P_{bar} + dH / 13.6)}{60 \theta V_s P_s A_n}}{1} \right]$$

Where:

- I = Percent isokinetic sampling.
 100 = Conversion to percent.
 T_s = Absolute average stack gas temperature, $^{\circ}R$.
 0.002669 = Conversion factor, $Hg - ft^3/ml - ^{\circ}R$.
 V_{ic} = Ttl vol of liquid collected in impingers and silica gel, ml.
 T_m = Absolute average dry gas meter temperature, $^{\circ}R$.
 P_{bar} = Barometric pressure at sampling site, (in. Hg).
 dH = Av pressure differential across the oriface meter, (in. H_2O).
 13.6 = Specific gravity of mercury.
 60 = Conversion seconds to minutes.
 θ = Total sampling time, minutes.
 V_s = Stack gas velocity, ft./sec.
 P_s = Absolute stack gas pressure, in. Hg.
 A_n = Cross sectional area of nozzle, ft^2 .

Run 1:

$$I = (100)(739) \left[\frac{(0.002669)(325.0) + \frac{51.027}{562} \left[30.00 + \frac{1.96}{13.6} \right]}{60 (60.0) (53.78) (30.00) (.000412)} \right] = 111.1\%$$

Run 2:

$$I = (100)(754) \left[\frac{(0.002669)(441.0) + \frac{60.800}{567} \left[30.00 + \frac{2.05}{13.6} \right]}{60 (60.0) (54.01) (30.00) (.000412)} \right] = 138.2\%$$

Run 3:

$$I = (100)(756) \left[\frac{(0.002669)(444.0) + \frac{58.924}{566} \left[30.00 + \frac{2.07}{13.6} \right]}{60 (60.0) (57.00) (30.00) (.000412)} \right] = 128.7\%$$

SUMMARY OF TEST DATA

7-7-92

RUN # 4

SAMPLING TRAIN DATA

	start	17:42
	finish	18:42
1. Sampling time, minutes	Θ	60.0
2. Sampling nozzle diameter, in.	D_n	.2740
3. Sampling nozzle cross-sect. area, ft ²	A_n	.000409
4. Isokinetic variation	I	109.8
5. Sample gas volume - meter cond., cf.	V_m	48.974
6. Average meter temperature, °R	T_m	560
7. Avg. orifice pressure drop, in. H ₂ O	dH	1.85
8. Total particulate collected, mg.	M_n	28.40

VELOCITY TRAVERSE DATA

9. Stack area, ft ²	A	21.00
10. Absolute stack gas pressure, in. Hg.	P_s	30.07
11. Barometric pressure, in. Hg.	P_{bar}	30.07
12. Avg. absolute stack temperature, R°	T_s	747
13. Average $-\sqrt{v_e l} \cdot \bar{h} \bar{e} \bar{a} \bar{d}$, ($C_p = .84$)	$-\sqrt{dP}$	0.85
14. Average stack gas velocity, ft./sec.	V_s	60.59

STACK MOISTURE CONTENT

15. Total water collected by train, ml.	V_{ic}	495.00
16. Moisture in stack gas, %	B_{ws}	33.49

EMISSIONS DATA

17. Stack gas flow rate, dscf/hr.(000's)	Q_{sd}	2164
18. Stack gas flow rate, cfm	acfm	76343
19. Particulate concentration, gr/dscf	C_s	0.0095
20. Particulate concentration, lb/hr	E	2.94
21. Particulate concentration, lb/mBtu	E'	0.00000

ORSAT DATA

22. Percent CO ₂ by volume	CO ₂	.00
23. Percent O ₂ by volume	O ₂	21.00
24. Percent CO by volume	CO	.00
25. Percent N ₂ by volume	N ₂	79.00

$$V_{m(std)} = V_m \left[\frac{T_{(std)}}{T_m} \right] \left[\frac{P_{bar} + \frac{dH}{13.6}}{P_{(std)}} \right] = 17.64 \frac{o_R}{in.Hg} Y V_m \left[\frac{P_{bar} + \frac{dH}{13.6}}{T_m} \right]$$

Where:

$V_{m(std)}$ = Dry Gas Volume through meter at standard conditions, cu. ft.

V_m = Dry Gas Volume measured by meter, cu. ft.

P_{bar} = Barometric pressure at orifice meter, in. Hg.

P_{std} = Standard absolute pressure, (29.92 in. Hg.).

T_m = Absolute temperature at meter $^{\circ}R$.

T_{std} = Standard absolute temperature (528 $^{\circ}R$).

dH = Average pressure drop across orifice meter, in. H_2O .

Y = Dry gas meter calibration factor.

13.6 = Inches water per inches Hg.

RUN 4 :

$$V_{m(std)} = (17.64)(.993)(48.974) \left[\frac{(30.07) + \frac{1.85}{13.6}}{560} \right] = 46.272 \text{ dscf}$$

Total Contaminants by Weight: GRAIN LOADING

Particulate concentration C'_s gr./dscf.

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{M_n}{V_{m(\text{std})}} \right]$$

Where:

C'_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr./dscf.

M_n = Total amount of particulate matter collected, mg.

$V_{m(\text{std})}$ = Dry gas volume through meter at standard conditions, cu. ft.

Run 4 :

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{28.40}{46.272} \right] = 0.0095 \text{ gr./dscf.}$$

$$M_d = 0.44(\%CO_2) + 0.32(\%O_2) + 0.28(\%CO + \%N_2)$$

Where:

M_d = Dry molecular weight, lb./lb.-mole.

$\%CO_2$ = Percent carbon dioxide by volume (dry basis).

$\%O_2$ = Percent oxygen by volume (dry basis).

$\%N_2$ = Percent nitrogen by volume (dry basis).

$\%CO$ = Percent carbon monoxide by volume (dry basis).

0.264 = Ratio of O_2 to N_2 in air, v/v.

0.28 = Molecular weight of N_2 or CO, divided by 100.

0.32 = Molecular weight of O_2 divided by 100.

0.44 = Molecular weight of CO_2 divided by 100.

Run 4:

$$M_d = 0.44(.00\%) + 0.32(21.00\%) + 0.28(.00\% + 79.00\%) = 28.84 \frac{\text{lb}}{\text{lb-mole}}$$

Water Vapor Condensed

$$V_{wc_{std}} = \left[V_f - V_i \right] \left[\frac{P_w R T_{(std)}}{M_w P_{(std)}} \right] = 0.04707 \left[V_f - V_i \right]$$

$$V_{wsg_{std}} = \left[W_f - W_i \right] \left[\frac{R T_{(std)}}{M_w P_{(std)}} \right] = 0.04715 \left[W_f - W_i \right]$$

Where:

0.04707 = Conversion factor, ft.³/ml.

0.04715 = Conversion factor, ft.³/g.

$V_{wc_{std}}$ = Volume of water vapor condensed (standard conditions), scf.

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel (standard conditions), ml.

$V_f - V_i$ = Final volume of impinger contents less initial volume, ml.

$W_f - W_i$ = Final weight of silica gel less initial weight, g.

P_w = Density of water, 0.002201 lb/ml.

R = Ideal gas constant, 21.85 in.Hg. (cu.ft./lb.-mole)(°R).

M_w = Molecular weight of water vapor, 18.0 lb/lb-mole.

T_{std} = Absolute temperature at standard conditions, 528°R.

P_{std} = Absolute pressure at standard conditions, 29.92 inches Hg.

Run 4 :

$$V_{wc(std)} = (0.04707) (475.0) = 22.4 \text{ cu.ft}$$

$$V_{wsg(std)} = (0.04715) (20.0) = 0.9 \text{ cu.ft}$$

Moisture Content of Stack Gases

$$B_{ws} = \frac{V_{wc_{std}} + V_{wsg_{std}}}{V_{wc_{std}} + V_{wsg_{std}} + V_{m_{std}}} \times 100$$

Where:

B_{ws} = Proportion of water vapor, by volume, in the gas stream.

V_m = Dry gas volume measured by dry gas meter, (dcf).

$V_{wc_{std}}$ = Volume of water vapor condensed corrected to standard conditions (scf).

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel corrected to standard conditions (scf).

Run 4 :

$$B_{ws} = \frac{22.4 + 0.9}{22.4 + 0.9 + 46.272} \times 100 = 33.49 \%$$

$$M_s = M_d (1 - B_{ws}) + 18 (B_{ws})$$

Where:

M_s = Molecular weight of stack gas, wet basis, (lb./lb.-mole).

M_d = Molecular weight of stack gas, dry basis, (lb./lb.-mole).

Run 4 :

$$M_s = 28.84 (1 - 33.49) + 18 (33.49) = 25.21 \text{ (lb./lb.-mole)}$$

$$V_s = K_p C_p \left[-\sqrt{dP} \right]_{\text{avg.}} \sqrt{\frac{T_{s(\text{avg.})}}{P_s M_s}}$$

Where:

- V_s = Average velocity of gas stream in stack, ft./sec.
 K_p = 85.49 ft/sec $\left[\frac{(\text{g/g-mole})-(\text{mm Hg})}{(^{\circ}\text{K})(\text{mm H}_2\text{O})} \right]^{1/2}$
 C_p = Pitot tube coefficient, (dimensionless).
 dP = Velocity head of stack gas, in. H_2O .
 P_{bar} = Barometric pressure at measurement site, (in. Hg).
 P_g = Stack static pressure, (in. Hg).
 P_s = Absolute stack gas pressure, (in. Hg) = $P_{\text{bar}} + P_g$
 P_{std} = Standard absolute pressure, (29.92 in. Hg).
 t_s = Stack temperature, ($^{\circ}\text{f}$).
 T_s = Absolute stack temperature, ($^{\circ}\text{R}$). = $460 + t_s$.
 M_s = Molecular weight of stack gas, wet basis, (lb/lb-mole).

Run 4 :

$$V = (85.49) (.84) (0.85) \sqrt{\frac{747}{(30.07)(25.21)}} = 60.59 \text{ ft/sec.}$$

Stack Gas Flow Rate

$$Q_{sd} = 3600 \left[1 - B_{wc} \right] V_s A \left[\frac{T_{std}}{T_{stk}} \right] \left[\frac{P_s}{P_{std}} \right]$$

Where:

- Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, (dscf/hr).
 A = Cross sectional area of stack, (ft.²).
 3600 = Conversion factor, (sec./hr.).
 t_s = Stack temperature, (°f).
 T_s = Absolute stack temperature, (°R).
 T_{std} = Standard absolute temperature, (528°R).
 P_{bar} = Barometric pressure at measurement site, (in.Hg.).
 P_g = Stack static pressure, (in.Hg.).
 P_s = Absolute stack gas pressure, (in.Hg.); = $P_{bar} + P_g$
 P_{std} = Standard absolute pressure, (29.92 in.Hg.).

Run 4 :

$$Q_{sd} = 3600 (1 - .3349) (60.59) (21.00) \left[\frac{528}{747} \right] \left[\frac{30.07}{29.92} \right] = 2164187.4 \frac{\text{dscf}}{\text{hr}}$$

AJAX PAVING INDUSTRIES
DETROIT, MI

Emissions Rate from Stack

$$E = \frac{(C_s) (Q_{sd})}{7000 \text{ gr./lb.}} = \text{lb. / hr.}$$

Where:

E = Emissions rate, lb/hr.

C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr/dscf.

Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, dscf/hr.

Run 4 :

$$E = \frac{(0.0095) (2164187.4)}{7000} = 2.94 \text{ lb. / hr.}$$

Isokinetic Variation

$$I = 100 T_s \left[\frac{0.002669 V_{ic} + \frac{(V_m / T_m) (P_{bar} + dH / 13.6)}{60 \theta V_s P_s A_n}}{1} \right]$$

Where:

- I = Percent isokinetic sampling.
- 100 = Conversion to percent.
- T_s = Absolute average stack gas temperature, $^{\circ}R$.
- 0.002669 = Conversion factor, $Hg - ft^3/ml - ^{\circ}R$.
- V_{ic} = Ttl vol of liquid collected in impingers and silica gel, ml.
- T_m = Absolute average dry gas meter temperature, $^{\circ}R$.
- P_{bar} = Barometric pressure at sampling site, (in. Hg).
- dH = Av pressure differential across the oriface meter, (in. H_2O).
- 13.6 = Specific gravity of mercury.
- 60 = Conversion seconds to minutes.
- θ = Total sampling time, minutes.
- V_s = Stack gas velocity, ft./sec.
- P_s = Absolute stack gas pressure, in. Hg.
- A_n = Cross sectional area of nozzle, ft^2 .

Run 4 :

$$I = (100)(747) \left[\frac{(0.002669)(495.0) + \frac{48.974}{560} \left[30.07 + \frac{1.85}{13.6} \right]}{60 (60.0) (60.59) (30.07) (.000409)} \right] = 109.8\%$$

SECTION F:
FIELD DATA

RAMCON ENVIRONMENTAL CORPORATION

Plant Oxy Ambient Temperature 90

Location Debit m Barometric Pressure 30.00 FINAL 513 SILICA GEL WEIGHT 524

Operator DOB Assumed Moisture, % 25 INITIAL 200 512

Date 8-10-92 Probe Length, m(ft) 68.3 DIFFERENCE 313 12

Run No. 1 (2) Nozzle Identification No. 0004124

Sample Box No. 1 Avg. Calibrated Nozzle Dia., (in.) 278/27/225

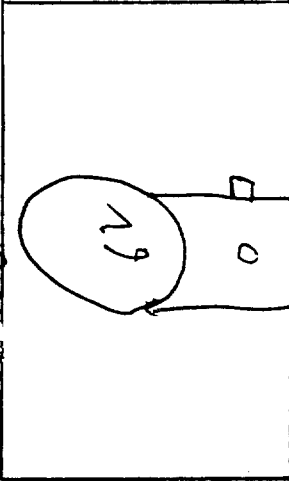
Meter Box No. C-282 Probe Heater Setting 4

Meter H @ 158 Leak Rate, m³/min. (cfm) 0.05 at 6 in. H₂O

C Factor 998 Probe Liner Material PTFE

Pitot Tube Coefficient Cp 94 Static Pressure, mm Hg (in. Hg) 75-6175

Filter No. 75-6175



Schematic of Stack Cross Section

TRAV. PT. NO.	SAMPLING TIME (θ) min.	VACUUM in. Hg	STACK TEMP (Ts) °F	VELOCITY HEAD (Ps) in H2O	PRESSURE DIFF. ORF. MTR in H2O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F		FILTER HOLDER TEMP °F	GAS TEMP LVG CONDENSER OR LAST IMPINGER °F
							Inlet	Outlet		
A 1	2:46 2:48-30	3	290	.52	1.3	254.1 256.0	102	80	260	58
A 2	2:51	3	293	.80	2.0	257.4	108	80	265	58
A 3	2:53-30	3	285	.85	.22	259.7	112	84	265	60
A 4	2:56	3	289	.85	.22	262.4	112	84	260	60
A 5	2:58-30	3	290	.85	2.2	263.9	120	88	260	59
A 6	3:01	3	298	.80	.20	266.2	120	88	260	59
A 7	3:03-30	3	292	.75	1.9	268.2	120	88	260	60
A 8	3:06	3	290	.80	.20	270.2	122	88	260	60
A 9	3:08-30	3	290	.90	2.3	272.4	122	88	255	60
A 10	3:11	3	293	.95	.24	274.8	130	88	255	60
A 11	3:13-30	4	292	.93	2.4	277.5	130	90	260	60
A 12	3:16	4	290	.93	.24	279.5	130	90	260	60
B 1	3:23 3:25-30	4	290	.95	2.4	281.4	130	90	260	60

174V

RAMCON ENVIRONMENTAL CORPORATION

Plant Ogish

2.56

Location Detroit

Operator 9000-June

Date 8-10-92

Run No. 2

Sample Box No. 6-282

Meter Box No. 1.58

Meter H @ 998

C Factor 94

Pitot Tube Coefficient Cp 94

Ambient Temperature 70

Barometric Pressure 30.0

Assumed Moisture, % 75

Probe Length, m(ft) 6.75

Nozzle Identification No. 0004124

Avg. Calibrated Nozzle Dia., (in.) 27/32/175

Probe Heater Setting 4

Leak Rate, m³/min. (cfm) .005

Probe Liner Material PTFE

Static Pressure, mm Hg (in. Hg) 25-6.26

Filter No. 25-6.26

MPH/HR VOLUME	SILICA GEL WEIGHT.
FINAL	531
INITIAL	520
DIFFERENCE	11

Schematic of Stack Cross Section

72 min test

TRAV. PT NO.	SAMPLING TIME (θ) min.	VACUUM in. Hg	STACK TEMP (Ts) °F	VELOCITY HEAD (Ps) in H2O	PRESSURE DIFF. ORF. MTR in H2O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F		FILTER HOLDER TEMP °F	GAS TEMP LVG CONDENSER OR LAST IMPINGER °F
							Inlet	Outlet		
A 1	6:18.21	2	294	.88	2.2	313.4	110	85	250	60
A 2	6:24	2	295	.90	2.3	318.6	112	85	250	60
A 3	6:27	2	295	.84	2.2	321.0	114	85	250	60
A 4	6:30	2	295	.1.1	2.8	324.1	120	85	250	60
A 5	6:33	2	298	.90	2.3	326.7	120	90	254	60
A 6	6:36	2	298	.85	2.2	329.2	120	90	255	60
A 7	6:39	2	287	.68	1.7	331.5	122	90	249	60
A 8	6:42	3	288	.70	1.8	334.2	122	90	249	58
A 9	6:45	3	289	.67	1.7	337.3	122	90	250	58
A 10	6:48	2	289	.60	1.5	338.5	126	90	260	62
A 11	6:51	2	291	.45	1.2	341.0	126	90	260	62
A 12	6:54	2	291	.45	1.2	342.8	126	90	260	62
B 1	7:04	2	291	.80	2.0	344.2	126	90	260	62

RAMCON

emissions test log sheet, cont.

DATE 8-10-92 LOCATION

TEST NO.

2

[illegible]

RAMCON ENVIRONMENTAL CORPORATION

Plant Cjex 256

Location Debit NJ

Operator Goodman

Date 8-10-92

Run No. 3

Sample Box No. 1

Meter Box No. C-282

Meter H @ 1.58

C Factor 998

Pitot Tube Coefficient Cp 94

Ambient Temperature 70

Barometric Pressure 30.0

Assumed Moisture, % 35

Probe Length, m(ft) 2.00

Nozzle Identification No. 0004124

Avg. Calibrated Nozzle Dia., (in.) 25/175/275

Probe Heater Setting 4

Leak Rate, m³/min. (cfm) 0.04

Probe Liner Material Aluminum

Static Pressure, mm Hg (in. Hg) 25.6171

Filter No. 25-6171

Schematic of Stack Cross Section

TRAV. PT NO.	SAMPLING TIME (h:min.)	VACUUM in. Hg	STACK TEMP (Ts) °F	VELOCITY HEAD (Ps) in H2O	PRESSURE DIFF. ORF. MTR in H2O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F		FILTER HOLDER TEMP °F	GAS TEMP LVG CONDENSER OR LAST IMPINGER °F
							Inlet	Outlet		
A 1	8:09	3	298	.55	1.4	324.8	102	90	250	60
A 2	8:15	3	299	.55	1.4	326.9	110	90	250	60
A 3	8:18	3	300	.65	1.7	381.0	112	90	250	60
A 4	8:21	3	300	.68	1.7	383.0	112	90	250	60
A 5	8:24	3	294	.55	1.4	385.2	120	90	250	60
A 6	8:27	3	290	.75	1.9	387.5	120	90	250	60
A 7	8:30	3	290	.85	2.2	390.6	120	90	250	60
A 8	8:33	3	291	.95	2.4	393.9	125	90	250	60
A 9	8:36	3	293	1.1	2.8	398.6	125	90	250	60
A 10	8:39	3	294	1.1	2.8	398.4	125	90	250	60
A 11	8:42	3	294	1.80	2.0	402.6	125	90	250	60
A 12	8:45	3	294	1.80	2.0	405.3	125	90	247	60
B 1	8:48	3	294	.85	2.2	406.8	125	90	247	60

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8-10-92 LOCATION

DATE:

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SECTION G:

CALIBRATION

METER BOX CALIBRATION DATA AND CALCULATION FORM

(English units)

Date 2-13-92

Meter box number C-272

Barometric pressure, $P_b = 29.98$ in. Hg Calibrated by W. J. J. J.

Orifice manometer setting (ΔH), in. H ₂ O	Gas volume		Temperature				Time (Θ), min	Y _i	$\Delta H@_i$ in. H ₂ O
	Wet test meter (V _w), ft ³	Dry gas meter (V _d), ft ³	Wet test meter (t _w), °F	Dry gas meter					
				Inlet (t _{d_i}), °F	Outlet (t _{d_o}), °F	Avg ^a (t _d), °F			
0.5	5								
1.0	10	77.953 89.423	68	92 102	62 62	79.5	1201	.974	1.58
1.5	10								
2.0	10	77.120 77.489	68	92 106	62 64	81	1250	.983	1.70
3.0	10	81.541 86.826	68	95 108	62 64	82.5	997	.991	1.62
4.0	10								
Avg								.983	1.63

ΔH , in. H_2O	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)}$	$\Delta H @_i = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460) \Theta}{V_w} \right]^2$
0.5	0.0368		
1.0	0.0737		
1.5	0.110		
2.0	0.147		
3.0	0.221		
4.0	0.294		

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d .

Quality Assurance Handbook M4-2.3A (front side)

POSTTEST DRY GAS METER CALIBRATION DATA FORM (English units)

Test number Date 8-2-72 Meter box number C-282 Plant
 Barometric pressure, $P_b = 30.00$ in. Hg Dry gas meter number Pretest Y

Orifice manometer setting, (ΔH), in. H_2O	Gas volume		Temperature				Time (θ), min	Vacuum setting, in. Hg	Y_i	Y_i $V_w P_b (t_d + 460)$ $V_d (P_b + \frac{\Delta H}{13.6})(t_w + 460)$
	Wet test meter (V_w), ft^3	Dry gas meter (V_d), ft^3	Wet test meter (t_w), $^{\circ}F$	Dry gas meter		Average (t_d), $^{\circ}F$				
				Inlet (t_{d_i}), $^{\circ}F$	Outlet (t_{d_o}), $^{\circ}F$					
5.277 1.0	10	23.80 23.36	74	107 114	80 90	95	8.56		984	1.59
10.372 2.0	10	24.60 24.33	74	107 124	80 92	97.5	12.15		1.00	1.60
10.2 3.0	10	27.38 27.78	74	107 120	78 90	95.5	9.78		1.01	1.56
									$Y = 998$	0.48 1.58

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d where

V_w = Gas volume passing through the wet test meter, ft^3 .

V_d = Gas volume passing through the dry gas meter, ft^3 .

t_w = Temperature of the gas in the wet test meter, $^{\circ}F$.

t_{d_i} = Temperature of the inlet gas of the dry gas meter, $^{\circ}F$.

t_{d_o} = Temperature of the outlet gas of the dry gas meter, $^{\circ}F$.

t_d = Average temperature of the gas in the dry gas meter, obtained by the average of t_{d_i} and t_{d_o} , $^{\circ}F$.

ΔH = Pressure differential across orifice, in. H_2O .

Y_i = Ratio of accuracy of wet test meter to dry gas meter for each run.

Y = Average ratio of accuracy of wet test meter to dry gas meter for all three runs; tolerance = pretest $Y \pm 0.05Y$.

P_b = Barometric pressure, in. Hg.

θ = Time of calibration run, min.

POSTTEST DRY GAS METER CALIBRATION DATA FORM (English units)

Test number _____ Date 7-24-92 Meter box number C-282 Plant _____
 Barometric pressure, $P_b = 30.17$ in. Hg Dry gas meter number _____ Pretest Y _____

Orifice manometer setting, (ΔH), in. H_2O	Gas volume		Temperature			Time (θ), min	Vacuum setting, in. Hg	Y_i	$V_w P_b (t_d + 460)$ $V_d (P_b + \frac{\Delta H}{13.6})(t_w + 460)$
	Wet test meter (V_w), ft^3	Dry gas meter (V_d), ft^3	Wet test meter (t_w), $^{\circ}F$	Inlet (t_{d_i}), $^{\circ}F$	Dry gas meter Outlet (t_{d_o}), $^{\circ}F$				
30	10	70.726	74	98	82 90	9.92	5	1.01	1.60
20	10	710.892	74	106	78 78	15.85	5	1.02	1.61
10	10	724.191	74	194	78 78	9.57	5	1.00	1.59
								$Y = 1.01$	$\Delta H = 1.60$

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d where

V_w = Gas volume passing through the wet test meter, ft^3 .

V_d = Gas volume passing through the dry gas meter, ft^3 .

t_w = Temperature of the gas in the wet test meter, $^{\circ}F$.

t_{d_i} = Temperature of the inlet gas of the dry gas meter, $^{\circ}F$.

t_{d_o} = Temperature of the outlet gas of the dry gas meter, $^{\circ}F$.

t_d = Average temperature of the gas in the dry gas meter, obtained by the average of t_{d_i} and t_{d_o} , $^{\circ}F$.

ΔH = Pressure differential across orifice, in. H_2O .

Y_i = Ratio of accuracy of wet test meter to dry gas meter for each run.

Y = Average ratio of accuracy of wet test meter to dry gas meter for all three runs;
 tolerance = pretest $Y \pm 0.05Y$.

P_b = Barometric pressure, in. Hg.

θ = Time of calibration run, min.

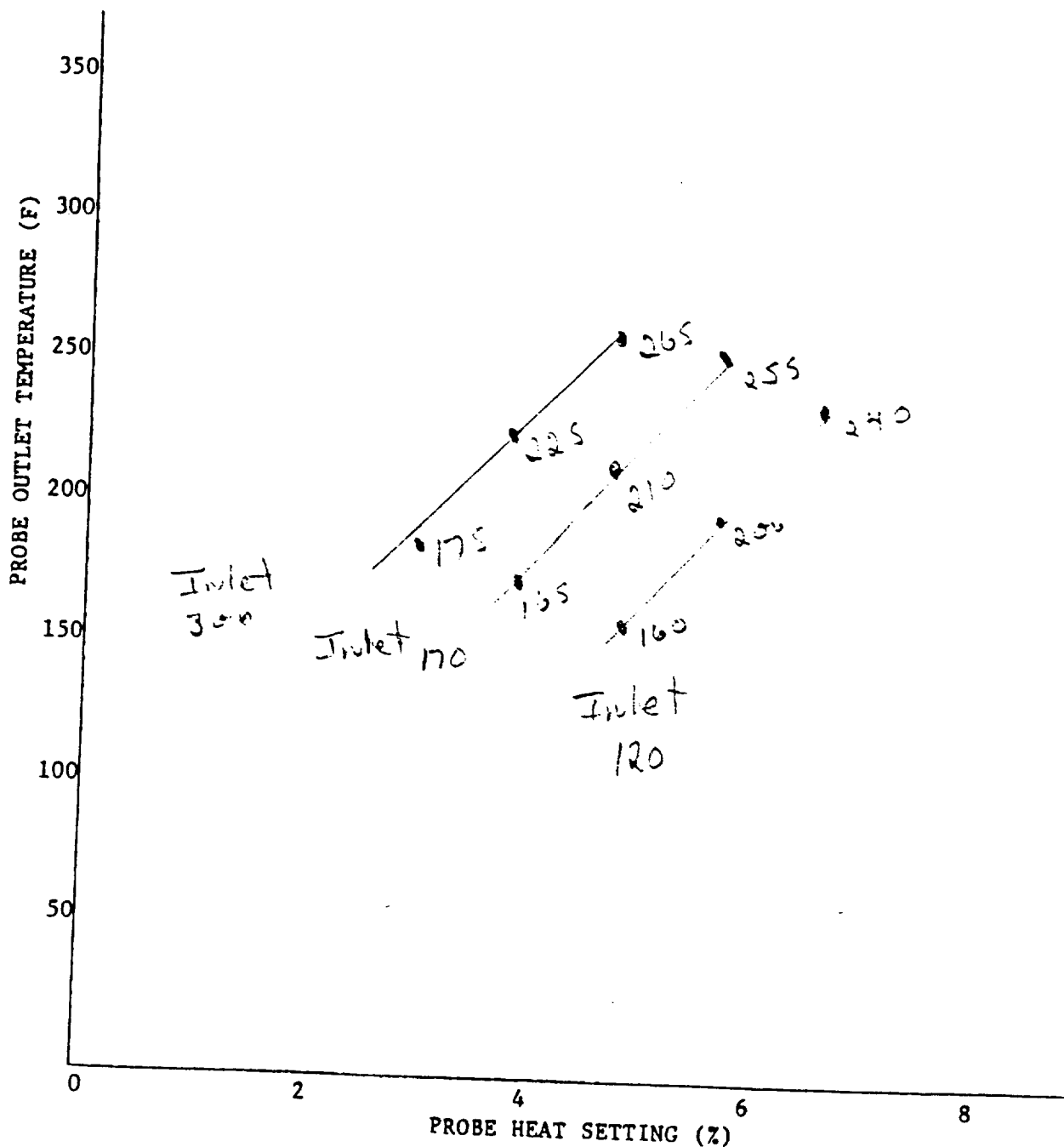
RAMCON

Lear Siegler Stack Sampler

Heating Probe Calibration

Probe No. 62 Probe Length 6'
 Date of Calibration 4-6-89 Signature S. Turner
 Name of Company to be tested _____

Note: 3 ft. probe - 5 min. warmup
 6 ft. probe - 15 min. warmup
 10 ft. probe - 30 min. warmup
 Calibration flow rate = .75 CFM



Date 4-8-91 Thermocouple number 6^{FT}
 Ambient temperature 81 °F Barometric pressure 29.96 in. Hg
 Calibrator Tennant Reference: mercury-in-glass ✓
 other _____

Reference point number ^a	Source ^b (specify)	Reference Thermometer Temperature, °C	Thermocouple Potentiometer Temperature, °C	Temperature Difference, °C
Boiling H ₂ O A	Boiling H ₂ O	212 °F	212 °F	0
Boiling Oil B	Boiling oil	417 °F	417 °F	0
Ice H ₂ O C	Ice water	32 °F	32 °F	0
Ambient Temp D	Ambient Temp.			

^aEvery 30°C (50°F) for each reference point.

^bType of calibration system used.

^c
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%$$

Figure 2.5 stack temperature sensor calibration data form.

RAMCON ENVIRONMENTAL CORPORATION

EPA QA MANUAL VOL. III
Section No. 3.4.2
Revision No. 0
Date January 15, 1980
Page 17 of 22

Date 4-8-91 Thermocouple number 6^F NEW
Ambient temperature 68 °C Barometric pressure 29.96 in. Hg
Calibrator 5 Turner Reference: mercury-in-glass
other

Reference point number ^a	Source ^b (specify)	Reference Thermometer Temperature, °C	Thermocouple Potentiometer Temperature, °C	Temperature Difference, °C
HOT H ₂ O		145 °F	146 °F	-.69
BOILING OIL		428 °F	428 °F	0
ICE WATER		32 °C	32 °F	0
AMBIENT TEMP		68 °F	68 °F	0

^aEvery 30°C (50°F) for each reference point.

^bType of calibration system used.

^c
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%$$

Figure 2.5 stack temperature sensor calibration data form.

STACK TEMPERATURE SENSOR CALIBRATION DATA FORM

Date 4-3-91 Thermocouple number 62
 Ambient temperature 20 °C Barometric pressure 29.88 in. Hg
 Calibrator Tuner Reference: mercury-in-glass ✓
 other _____

Reference point number	Source ^a (specify)	Reference thermometer temperature, °C	Thermocouple potentiometer temperature, °C	Temperature difference, ^b %
A	oil	381	381°	0
B	water boiling	212	212°	0
C	Ice water	32	32°	0
D	Ambient			

^aType of calibration system used.

^b
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%.$$

A

SECTION H:
RAMCON PERSONNEL

Name: **Mr. G. Sumner Buck, III**
Title: **President**

Qualifications: Mr. Buck is a graduate of the University of Mississippi with graduate studies at Memphis State University and State Technical Institute of Memphis. He is a graduate of the EPA 450 "Source Sampling for Particulate Pollutant's" course and the 474 "Continuous Emissions Monitoring" courses outlined by EPA at Research Triangle Park, N.C. He has been directly involved in conducting and supervising air emission testing for over 15 years. He has personally conducted over 400 air emission tests. He currently sponsors and directs visual emission certification schools for US EPA Reference Method 9.

Project Duties: Mr. Buck will be responsible for the overall supervision of the project. He provided supervision for the project preparation, testing schedules for each team on-site, and overall organization between the testing crews and facility.

Name: **Mr. Joe Sewell**
Title: **Vice President**

Qualifications: Mr. Sewell is currently serving as the Vice President of RAMCON Environmental Corporation. Mr. Sewell is a graduate of Christian Brothers University in Memphis, Tennessee where he obtained a Bachelor of Science degree in Chemical Engineering. He has conducted and supervised air emissions testing projects ranging a broad spectrum of facility process categories. His accomplishments include the development of the instrumental branch of emissions testing utilizing continuous emission monitors and gas chromatography. Mr. Sewell performs a major role in the upgrading of testing capabilities and professional quality that RAMCON Environmental Corporation offers.

Project Duties: Mr. Sewell provides staff engineering and project administration to ensure the integrity of the requested services. He served as the primary contact person for RAMCON Environmental Corporation handling all correspondence between the facility personnel involved in the project and respective state agency representative(s). He provides project leadership to RAMCON Environmental Corporation field supervisors and managers involved in the test project.

Name: **Mr. Ray Jenkins**
Title: **Source Sampling Director**

Qualifications: Mr. Jenkins is serving as the Source Sampling Director for RAMCON Environmental Corporation. He was promoted to this leadership position after gaining a significant amount of experience in conducting and providing field supervision of a variety of air testing projects. Mr. Jenkins has personally conducted and/or supervised all of the prevalent EPA approved procedures with expertise in the instrumental analyzer procedures. He graduated from Memphis State University where he obtained a Bachelor of Science degree in Biology. He is also currently certified to conduct US EPA Reference Method 9 for the visual determination of emission opacity.

Project Duties: Mr. Jenkins provided project leadership to the Team Leader Field Technicians. He ensured that the test crew(s) involved in the test project were properly informed to his respective duties and responsibilities during the testing process. Mr. Jenkins also serves as the Quality Assurance/Quality Control Coordinator and provided guidance in QA/QC to each Team Leader with regard to sample integrity.

Name: **Mr. Thomas South**
Title: **Laboratory Technician**

Qualifications: Mr. South is currently serving as Laboratory Technician. He is proficient in conducting many analysis procedures such as front and back-half particulate analysis, titrations, extractions, etc. He received an Associate degree in Chemical Engineering from State Technical Institute in Memphis, Tennessee.

Project Duties: Mr. South conducts the laboratory analysis on the particulate samples. He is also responsible for accepting the remaining field samples from the Field Sample Bank Manager and performing inspection as to integrity. He documents the transfer on the chain of custody forms and distributed the subcontracted samples to the respective laboratories.