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OBSERVER'S REPORT:

COMPLIANCE TEST

for

PARTICULATE

Facility: VALLEY ASPALT # 20
CAMP DENNISON, Ohio

Premise Number: 1431453448

Source: P901 ASPHALT BATCH PLANT

Conducted On: SEPT. 23 + 24, 1992

Test Firm: RAMCO ENVIRONMENTAL CORP.

Observed By: Trent Lewis, ETI;

Prepared By: Trent Lewis, ETI
LeRoy Gruber, AQE

Date Prepared: DECEMBER 1, 1992

DEPARTMENT OF ENVIRONMENTAL SERVICES
AIR QUALITY PROGRAMS
1632 CENTRAL PARKWAY
CINCINNATI, OHIO 45210

OEPA COPY
IP FILE COPY

APPENDIX K

OEPA STACK TEST REVIEW SUMMARY FORM

APPLICATION NUMBER 1431453448

FACILITY NAME Valley Asphalt #20

SOURCE DESCRIPTION (OR SCC CODE) SN: P901 - Asphalt Batch Plant

CONTROL EQUIPMENT Baghouse

DATE(S) OF TEST September 23 & 24, 1992

FINAL TEST REPORT RECEIVED ON October 20, 1992

POLLUTANT(S) TESTED Particulate

TEST METHOD U.S. EPA Test Method 5

TEST FIRM Ramcon

EMISSION RATES*:

[] (lb(s)/MMBtu)			
ACTUAL [X] (lb(s)/hr)	5.77	ALLOWABLE**	10.4

OPERATING RATES*:

DURING TEST**	245.5 tons	MAXIMUM**	300 tons
---------------	------------	-----------	----------

EMISSION FACTOR***

COMMENTS 5.77 lbs/hour is protated from 5.26 lbs per hour because test capacity was not met.

I HEREBY VERIFY THAT THE INFORMATION CONTAINED WITHIN THE STACK TEST REPORT HAS BEEN REVIEWED AND IT HAS BEEN DETERMINED THAT THE TEST PROCEDURES, ANALYSES AND CALCULATIONS ARE:

[X] AN ACCEPTABLE DEMONSTATION OF CONFORMANCE WITH THE APPROVED TESTING METHODOLOGY.

[] AN UNACCEPTABLE DEMONSTRATION OF CONFORMANCE WITH THE APPROVED TESTING METHODOLOGY.

12/1/92

PREPARED BY: Trent R. Lewis II, ET 1

DATE OF REVIEW

REVIEWED BY: LeRoy R. Gruber, AQE

* BASED ON 3 RUN AVERAGE
** SPECIFY APPLICABLE UNITS
*** SPECIFY UNITS OF MASS/INPUT

PN: 1431453448
SN: P901
PAGE: 1 of 2

On September 23 + 24, 1992 testing for particulate matter was conducted at Valley Asphalt Plant #20. This is a new source and is required to meet NSPS standards Subpart I (Particulate Matter is not to exceed .04 grain/dscf). Testing was conducted by Ramcon Environmental Inc., under the supervision of team leader Alan Turner. Two runs were conducted on the 23rd. The second run had a leak check rating of .027cfm and this is to be corrected with a correction factor. A third run had been attempted, however, a failure in the sampling equipment caused sampling to be discontinued. The fourth run of testing was conducted on 9-24-92.

Production levels and emission rates are shown in the following table.

<u>RUN #</u>	<u>TONS PER HOUR PRODUCTION**</u>	<u>GR./DSCF</u>	<u>LBS./HR. EMISSIONS</u>
1	257.6	.0369	8.89
2	237.2	.0159	3.52
4	241.7	.01496	3.37
total	245.5	.0226	5.26

(RUN 2 IS A CORRECTED VALUE CAUSED BY A FAILED LEAK CHECK)

** TONS OF PRODUCTION WAS SUPPLIED BY DATA PROVIDED BY VALLEY ASPHALT.

The tons per hour production averaged represents 82% of the rated capacity of 300 tons per hour. A standard 90% (or greater) of operation capacity is required during testing. If the sampling results are prorated to the minimum "worst case" standard of 90%, the average would be 5.773 lbs./hr.

Stack conditions averaged: carbon dioxide = 4.5%, oxygen = 14.8%, Nitrogen = 80.7%, DSCFM = 26708, and Percent moisture avg. = 12.25%.

Method 9 for visible emissions was also conducted and no violations were found. It should be noted that a particulate leak was observed at the vibratory screen, at a level of opacity ranging between 5-10%.

Quality assurance measures included: sampling train leak checks (sampling and pitot lines), proper documentation, nozzle diameter measurements, pitot tube/dry gas meter/thermocouple calibrations. Verification calculations were executed by D.O.E.S. staff, and found to be of sufficient accuracy.

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SN: P901
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In conclusion, the data received by this agency was of sufficient accuracy and precision to determine compliance. At 43% of allowable emission rate, testing at permit renewal is recommended, sooner if visible emissions of greater value than noted are observed.

EPA METHOD 5 STACK TEST ANALYSIS

PLANT NAME: VALLEY AIRPORT #10
 PRINTED: 8- APR # 14 1992
 SOURCE DESCRIPTION: ASHALT BATCH PLANT P201

REVIEWER: TPL
 TEST DATE: 8-23-92
 PRINTED: 11-27-1992

INPUT DATA:	RUN #1	RUN #2	RUN #3
METER VOLUME (ACF).....	45.775	39.925	44.1
WATER CORRECTION.....	1.02	1.02	1.02
BAROMETRIC PRESSURE (IN HG).....	30.1	30.1	30.1
ORIFICE PRESSURE (IN H2O).....	1.755	1.5425	1.553
METER TEMPERATURE (F).....	73.500	87.208	77.3735
VOLUME WATER (ML H2O).....	257.1	235	264.7
CO2 IN STACK GAS (%).....	4.5	4.5	4.5
CO IN STACK GAS (%).....	14.8	14.8	14.8
NO IN STACK GAS (%).....	80.7	80.7	80.7
O2 IN STACK GAS (%).....	0	0	0
ORIFICE COEFFICIENT.....	.84	.84	.84
AVERAGE SQUARE ROOT VELOCITY PRESS.....	.5855	.5501	.5551
STACK TEMPERATURE (F).....	225.75	222.71	220.68
STACK PRESSURE (IN HG).....	30.11	30.11	30.11
STACK DIAMETER (IN).....	60	60	60
STACK AREA (SQ FT).....	19.63496	19.63496	19.63496
PROBABILITY FACTOR (MULT).....	1.015	40.9	43.2
NOZZLE DIAMETER (IN).....	.315	.315	.315
NOZZLE AREA (SQ FT).....	0.00541	0.00541	0.00541
NOZZLE LENGTH (IN).....	50	50	50

TEST RESULTS:

METER VOLUME @STP (DSCF).....	46.1579	39.8635	44.5732
VOLUME OF WATER VAPOR @STP (CF).....	%12.101700	%12.002850	%12.459430
MOLE FRACTION OF WATER IN STACK GAS.....	0.207720	0.132315	0.218462
DRY MOLECULAR WEIGHT OF STACK GAS.....	29.3120	29.3120	29.3120
MOLECULAR WEIGHT OF STACK GAS.....	26.9603	26.0341	26.9408
STACK GAS VELOCITY (FT/SEC).....	33.0249	36.6130	36.5268
STACK GAS FLOW (DRY STD SCFH).....	%1883708.0	%1346314.0	%1577406.0
CARBONATE CONCENTRATION (PPM/DSCF).....	0.056909	0.015911	0.014955
SOOT INDEX RATIO.....	99.5075	93.1049	102.5670
PERCENTAGE OF EMISSION (1/HR).....	8.3809	0.5148	3.3699

REMARKS/TEST COMMENTS:

RUN 2 IS A CORRECTED SAMPLE GAS VOLUME REQUIRED BECAUSE OF A FAILED LEAK CHECK.

Source

Valley Asphalt #20

Premise No.

App 14-2291 P901

By

TRC

Date

11-27-92

DATA SHEET

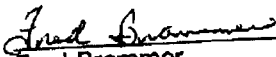
Examples

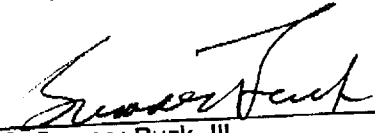
		41.545	39.925	44.1
38.847	Vm(dcf)	45.275	39.925	44.1
1.01	Y	1.02	1.02	1.02
27.93	Pbar(inHg)	30.1	30.1	30.1
2.35	DHP(H ₂ O)	1.756	1.5425	1.558
96°	Tm, °F	73.503	87.208	77.8935
78.5	Vlc(ml)	257.1	255	264.7
12.0	CO ₂ (percent)	4.5	4.5	4.5
08.7	Ø ₂	14.8	14.8	14.8
00.8	N	80.7	80.7	80.7
00.1	CO	0	0	0
.84	Op	.84	.84	.84
.4201	SQPD avg.	.5896	.5531	.5551
230	Ts(°F)	229.75	222.71	220.88
27.91	Ps(in, Hg. Pb+Ps)	30.11	30.11	30.11
21"	Stack diam	60"	60"	60"
110.6	Mn(mg.)	110.5	40.9	43.2
.25"	Dn(in)	.315	.315	.315
72	Theta(min)	60	60	60
9820	F' (Optional)			
4	H			
70	C			
10	S			
14	N			
6	O			
	Allowable GCV (Optional) BTU/lb.			

RAMCON

ENVIRONMENTAL CORPORATION

SOURCE SAMPLING
for
PARTICULATE EMISSIONS
VALLEY ASPHALT CORPORATION, PLANT #20
CAMP DENNISON, OHIO
September 23-24, 1992


Fred Bremmer
Valley Asphalt Corporation


G. Sumner Buck, III
President

RAMCON

ENVIRONMENTAL CORPORATION

October 5, 1992

Mr. Fred Bremmer
Valley Asphalt Corporation
11641 Mosteller Road
Cincinnati, OH 45421

Re: Particulate Emissions Test: Camp Dennison, Ohio - Plant #20

Dear Mr. Bremmer:

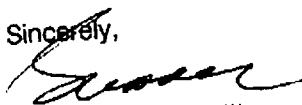
Enclosed you will find four copies of our report on the particulate emissions test we conducted at your plant. Based on our test results, the average grain loading of the three test runs do pass both EPA New Source Performance Standards and those set by the State of Ohio. Therefore, the plant is operating in compliance with Federal and State Standards.

You will want to sign the report covers and send two copies to:

Mr. Lee Gruber
Ohio EPA
Air Quality Bureau
1632 Central Parkway
Cincinnati, OH 45210

You will need to keep one copy of the report at the plant. We certainly have enjoyed working with you. Please let us know if we can be of further assistance.

Sincerely,



G. Sumner Buck, III
President

GSBIII:djb

Enclosures

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SECTION A:

- 1. INTRODUCTION**
- 2. TEST RESULTS**
- 3. TEST PROCEDURES**

SECTION A.

1. INTRODUCTION

On September 23-24, 1992 personnel from RAMCON Environmental Corporation conducted a source emissions test for particulate emissions compliance at Valley Asphalt Corporation's Barber-Greene batch mix asphalt plant located in Camp Dennison, Ohio. RAMCON personnel conducting the test were Allen Turner, Team Leader, and Chuch Hughes. Thomas South was responsible for the laboratory analysis including taring the beakers and filters and recording final data in the laboratory record books. Custody of the samples was limited to Mr. Turner and Mr. South.

The purpose of the test was to determine if the rate of particulate emissions from this plant's baghouse is below or equal to the allowable N.S.P.S. emissions limit set by US EPA and the State of Ohio.

2. TEST RESULTS

Table I summarizes the test results. The grain loading limitation for EPA is .04 gr/dscf as specified in 39 FR 9314, March 8, 1974, 60.92 Standards for Particulate Matter (1), as amended. The allowable emissions for the State of Ohio are the same as those set by EPA.

Mr. Trent Lewis of Ohio's Air Pollution Control Agency observed the testing conducted by RAMCON Environmental.

SUMMARY OF TEST RESULTS

TABLE I

September 23-24, 1992

<u>Test Run</u>	<u>Time</u>	<u>Grain Loading gr/DSCF</u>	<u>Isokinetic Variation</u>	<u>Emissions Lbs/Hr</u>
1	08:36 - 09:40	0.0369	99.4%	8.88
2	11:00 - 12:22	0.0153	96.6%	3.39
3	11:31 - 13:26	0.0149	101.7%	3.39
	Average:	0.0224		5.22

On the basis of these test results, the average grain loading of the three test runs was below the .04 gr/DSCF allowable emissions limitation set by EPA and the State of Ohio. Therefore, the plant is operating in compliance with State and Federal Standards.

3. TEST PROCEDURES

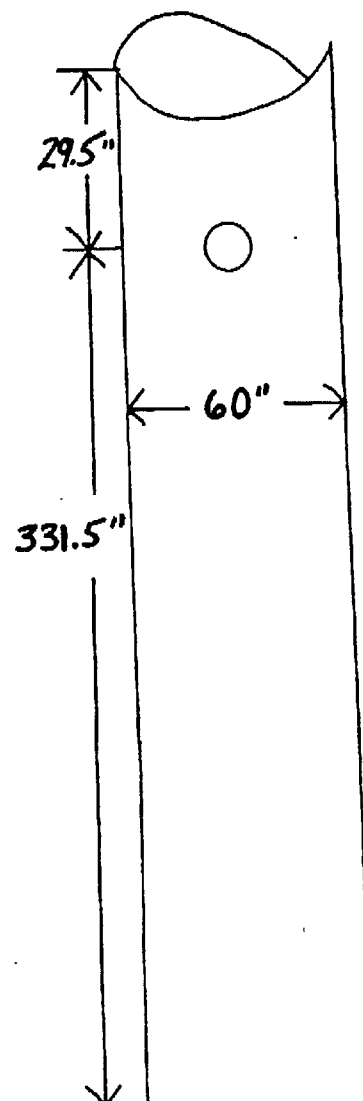
(a) Method Used: Method 5 source sampling was conducted in accordance with requirements of the U.S. Environmental Protection Agency as set forth in 39 FR 9314, March 8, 1974, 60.93, as amended.

(b) Problems Encountered: No problems were encountered that affected testing.

(c) Sampling Site: The emissions test was conducted after a baghouse on a round stack with a diameter of 60". The sampling ports were placed 90° apart, 29.5" down (0.5 diameters upstream) from the top of the stack and 331.5" up (5.5 diameters downstream) from the last flow disturbance. Twenty four points were sampled, twelve through each traverse for 2.5 minutes each for a total testing time of 60 minutes.

Points on a Diameter	Probe Mark
1	*7.26"
2	10.02"
3	13.08"
4	16.62"
5	21.00"
6	27.30"
7	44.70"
8	51.00"
9	55.38"
10	58.92"
11	61.98"
12	64.74"

*Measurements include a
6" standoff.



SECTION B:
THE SOURCE

B. THE SOURCE

Valley Asphalt Corporation (Plant #20) employs a Barber-Greene batch mix asphalt plant which is used to manufacture hot mix asphalt for road pavement. The process consists of blending prescribed portions of cold feed materials (sand, gravel, screenings, chips, etc.) uniformly and adding sufficient hot asphalt oil to bind the mixture together. After the hot asphalt mix is manufactured at the plant, it is transported to the location where it is to be applied. The hot asphalt mix is spread evenly over the surface with a paver then compacted with a heavy roller to produce the final product.

The following is a general description of the plant's manufacturing process: The cold feed materials (aggregate) are dumped into separate bins which in turn feed a common continuous conveyor. The aggregate is dispensed from the bins in accordance with the desired formulation onto the cold feed system conveyor, to an inclined weigh conveyor, then to a rotating drum for continuous mixing and drying at approximately 300°F. When recycled asphalt mix is used, it is added directly into the pugmill. The dried aggregate is pulled by a bucket elevator to the top of a gradation control unit which separates and stores the aggregate by size. The required amount of each aggregate is dispensed into a weigh-hopper and from there into a pugmill where the hot liquid asphalt pavement is mixed thoroughly with the aggregate. The hot asphalt mix is then discharged from the storage silo through a slide gate into waiting dump trucks which transports the material to a final destination for spreading. The rated capacity of the plant will vary with each aggregate mix and moisture content with a 5% surface moisture removal.

The mixer uses a burner fired with #2 fuel oil to heat air to dry the aggregate. The air is drawn into the system via an exhaust fan. After passing through the gas burner, the air passes through a baghouse. The baghouse is manufactured by Barber-Greene. The exhaust gas is drawn through the baghouse and discharged to the atmosphere through the stack. The design pressure drop across the tube sheet is 2 - 6 inches of water. The particulate matter, which is removed by the baghouse, is reinjected into the pugmill.

DATA ON FACILITY BEING STACK TESTED _____
 TODAY'S DATE: _____

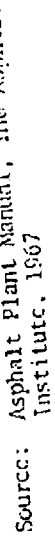
COMPANY NAME _____ COMPANY REP. _____ PHONE () _____
 LOCATION OF FACILITY _____ ORIGINAL START-UP DATE _____ DESIGNED CAPACITY _____
 OEM _____ MODEL NO. _____ TYPE _____ AC TYPE _____

1 Time (24 HR)	2 Fuel Use 2 Fuel Oil ☒ Nat. Gas ☐ Propane ☐ Coal ☐ Other ☐	3 Burner Setting	4 Blower Pressure	5 Production Rate		6 Asphalt Content %	7 Mix Temp. °F	8 Exhaust Gas Temp. °F	9 Venturi Scrubber Baghouse		10 Ambient Temp. °F	11 Relative Humidity %	12 Exhaust Dampers Position
				Mix Aggregate TPH	RAP TPH				Pressure Drop in w.g.	Water Pressure psi			
8:40		96%		258		4.5	323	290	5.3	N/A	60°		92
8:55		98%		260		4.44	318	280	5.6	N/A	60°		94
9:10		82%		250		4.34	320	290	5.2	N/A	60°		92
9:25		96%		258		4.46	309	285	5.2	N/A	65°		92
9:40		96%		262		4.5	302	285	6.1	N/A	65°		94
10:40		76%		257		4.43	315	285	5.4	N/A	65°		92
11:15		75%		220		4.51	310	270	5.8	N/A	65°		93
11:30		90%		243		4.43	306	280	6.2	N/A	65°		98
11:45		71%		230		4.38	321	290	6.0	N/A	65°		94
12:00		64%		237		4.40	307	270	5.8	N/A	65°		92
12:15		96%		256		4.49	303	290	6.0	N/A	65°		95
1:55		85%		249		4.44	304	290	5.8	N/A	70°		94
2:10		96%		254		4.45	296	290	5.7	N/A	70°		94

NOTE: check
small box in
column when
moisture
sample is
taken

[illegible]

ASPHALT BATCH MIX PLANT - AN EXPLODED VIEW



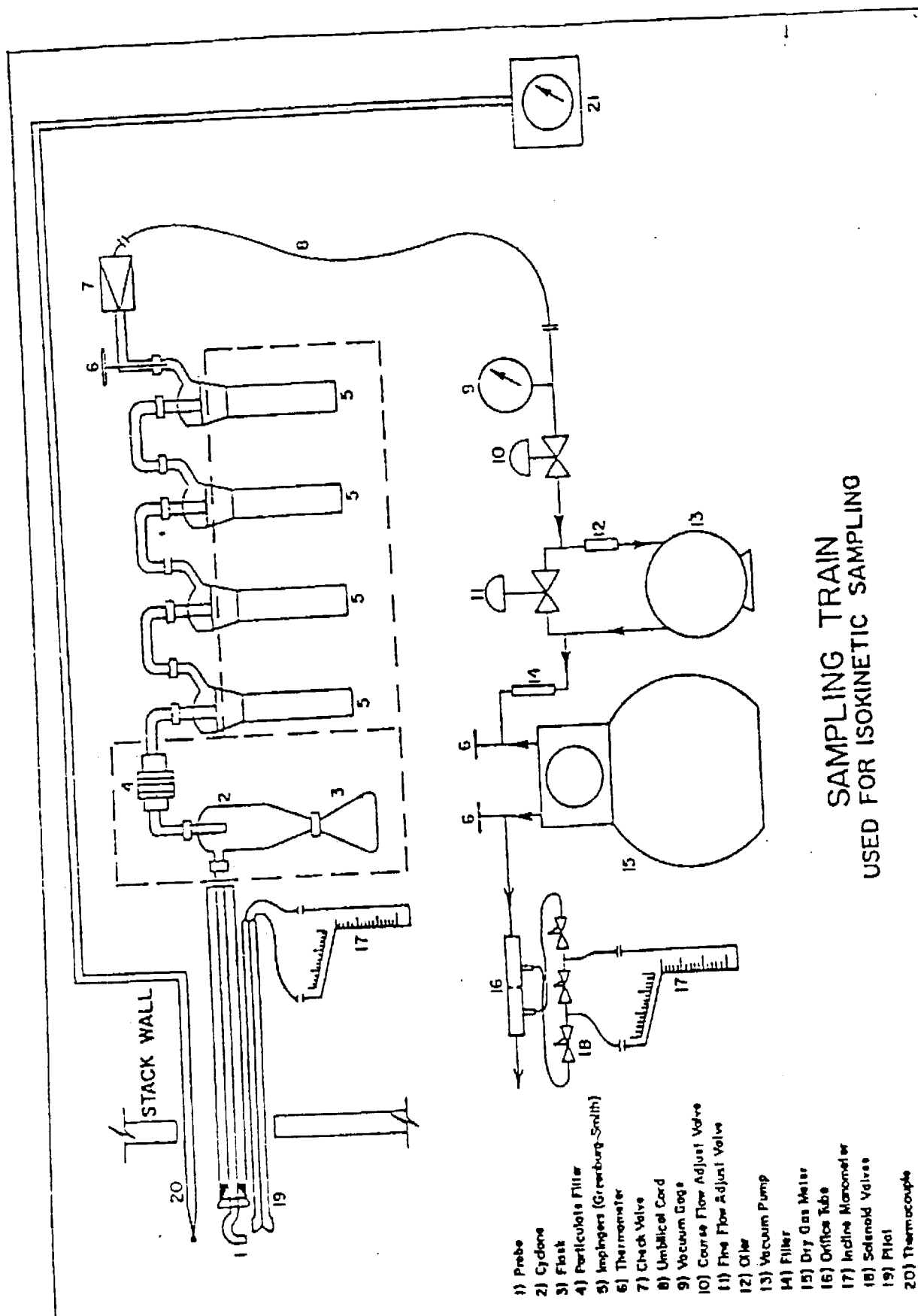
1. **Aggregate bins:** Virgin aggregate is fed individually into bins by type. It is metered onto a conveyor belt running under the bins to a shaker screen. The proportion to each aggregate type is determined by the job mix formula and pre-set to be metered out to meet these specifications.
2. **Preliminary oversize screen:** The aggregate is fed through a shaker screen where oversize rocks and foreign material is screened out of the mix.
3. **Weigh conveyor belt:** The aggregate is conveyed to the rotary drum dryer on a conveyor belt which weighs the material. The production rate is determined by this weight reading.
4. **Rotary drum/dryer mixer:** The aggregate is fed into the rotary drum dryer where it is tumbled by flighting into a veil in front of a flame which drives off the moisture. Further mixing is also accomplished in an outer shell of this drum. Hot liquid asphalt is injected in the outer shell of the drum where it is mixed with the aggregate.
5. **Burner:** The fuel fired burner is used to provide the flame which dries the aggregate.
6. **Knock off baffling:** A baffling plate is inserted in the "dirty" side plenum as a knock out for heavy particles in the air stream. These particles fall to the bottom of the baghouse.
7. **Baghouse:** The hot gases are pulled through the bags into the clean air plenum. The solid particulate matter is trapped on the dust coat buildup on the bags. A bag cleaning cycle consisting of jet burst of air from the inside (or clean air side) of the bags sends a large bubble of air down the inside of the bags shaking loose buildup on the bag surface. This particulate matter is collected at the bottom of the baghouse and reinjected into the drum mixer where it is used as part of the finished product.
8. **Liquid asphalt storage:** The liquid asphalt is stored in this heated tank until it is needed in the mixer. The amount of asphalt content and its temperature are pre-set for each different type job.
9. **Conveyor to surge/storage bin:** The finished product of aggregate mixed with liquid asphalt is conveyed to a surge bin.
10. **Surge/Storage bin:** The asphaltic cement is dumped into this surge bin and metered out to dump trucks which pull underneath a slide gate at the bottom of the bin.
11. **Control/operators house:** The entire plant operation is controlled from this operator's house.
12. **Truck loading scale:** As the trucks receive the asphalt from the storage/surge bin, they are weighed on the lading scale which tells the plant operator the amount of asphalt that is being trucked on each individual load.
13. **Fuel storage.**
14. **Stack**

SECTION C:
EQUIPMENT USED

EQUIPMENT USED

Equipment used on conducting the particulate emissions test was:

- A. The Lear Siegler PM-100 stack sampler with appropriate auxiliary equipment and glassware. The train was set up according to the schematic on the next page.
- B. An Airguide Instruments Model 211-B (uncorrected) aneroid barometer was used to check the barometric pressure.
- C. Weston dial thermometers are used to check meter temperatures. An Analogic Model 2572 Digital Thermocouple is used for stack temperatures.
- D. A Hays 621 Analyzer was used to measure the oxygen, carbon dioxide and carbon monoxide content of the stack gases. For non-combustion sources, a Bacharach Instrument Company Fyrite is used for the gas analysis.
- E. Filters are made by Schleicher and Schuell and are type 1-HV with a porosity of .03 microns.
- F. The acetone is reagent grade or ACS grade with a residue of $\leq .001$.



SECTION D:
LABORATORY PROCEDURES AND RESULTS

LABORATORY PROCEDURES FOR PARTICULATE SAMPLING

I. Field Preparation

A. FILTERS: Fiberglass 4" sampling filters are prepared as follows:

Filters are removed from their box and numbered on the back side with a felt pen. The numbering system is continuous from job to job. The filters are placed in a desiccator to dry for at least 24 hours. Clean plastic petri dishes, also numbered, top and bottom, are placed in the desiccator with the filters. After desiccation, the filters are removed, one at a time, and weighed on the Sartorius analytical balance then placed in the correspondingly numbered petri dish. Weights are then recorded in the lab record books. Three filters are used for each complete particulate source emissions test and there should be several extra filters included as spares.

B. SILICA GEL: Silica Gel used for the test is prepared as follows:

Approximately 200 g of silica gel is placed in a wide mouth "Mason" type jar and dried in an oven at 175°C for two hours. The open jars are removed and placed in a desiccator until cool for two hours and then tightly sealed. The jars are then numbered and weighed on the triple beam balance to the closest tenth of a gram. This weight is recorded for each sealed jar. The number of silica gel jars used is the same as the number of filters. Silica gel should be indicating type, 6-16 mesh.

II. Post - Testing Lab Analysis

A. FILTERS: The filters are returned to the lab in their sealed petri dishes. In the lab, the dishes are opened and placed into a desiccator for at least 24 hours. Then the filters are weighed continuously every six hours until a constant weight is achieved. All data is recorded on the laboratory forms that will be bound in the test report.

B. SILICA GEL: The silica gel used in the stack test is returned to the appropriate mason jar and sealed for transport to the laboratory where it is reweighed to a constant weight on a triple beam balance to the nearest tenth of a gram.

- C. **PROBE RINSINGS:** In all tests where a probe washout analysis is necessary, this is accomplished in accordance with procedures specified in "EPA Reference Method 5". These samples are returned to the lab in sealed mason jars for analysis. The front half of the filter holder is washed in accordance with the same procedures and included with the probe wash. Reagent or ACS grade acetone is used as the solvent. The backhalf of the filter holder is washed with deionized water into the impinger catch for appropriate analysis.
- D. **IMPINGER CATCH:** In some testing cases, the liquid collected in the impingers must be analyzed for solid content. This involves a similar procedure to the probe wash solids determination, except that the liquid is deionized water.
- E. **ACETONE:** A blank analysis of acetone is conducted from the one gallon glass container used in the field preparation. This acetone was used in the field for rinsing the probe, nozzle, and top half of the filter holder. A blank analysis is performed prior to testing on all new containers of acetone received from the manufacturer to insure that the quality of the acetone used will be exceed the .001% residual purity standard.

SPECIAL NOTE

When sampling sources high in moisture content, (such as asphalt plants) the filter paper sometimes sticks to the filter holder. When removing the filter, it may tear. In order to maintain control of any small pieces of filter paper which may be easily lost, they are washed with acetone into the probe washing. This makes the filter weight light (sometimes negative) and the probe wash correspondingly heavier. this laboratory procedure is taught by EPA in the "Quality Assurance for Source Emissions Workshop" at Research Triangle Park and is approved by EPA.

WEIGHING PROCEDURE - SARTORIUS ANALYTICAL BALANCE

The Sartorius balance is accurate to 0.1 mg and has a maximum capacity of 200 grams. The balance precision (standard deviation) is 0.05 mg. Before weighing an item, the balance should first be zeroed. This step should be taken before every series of weighings. To do this, the balance should have all weight adjustments at the "zero" position. The beam arrest lever (on the lower left hand side toward the rear of the balance) is then slowly pressed downward to the full release position. The lighted vernier scale on the front of the cabinet should align with the "zero" with the mark on the cabinet. If it is not so aligned, the adjustment knob on the right hand side (near the rear of the cabinet) should be turned carefully until the marks align. Now return the beam arrest to the horizontal arrest position. The balance is now "zeroed".

To weigh an item, it is first placed on the pan. And the sliding doors are closed to avoid air current disturbance. The weight adjustment knob on the right hand side must be at "zero". The beam arrest is then slowly turned upward. The lighted scale at the front of the cabinet will now indicate the weight of the item in grams. If the scale goes past the divided area, the item then exceeds 100 g weight (about 3 1/2 ounces) and it is necessary to arrest the balance (beam arrest lever) and move the lever for 100 g weight away from you. It is located on the left hand side of the cabinet near the front, and is the knob closest to the side of the cabinet. The balance will not weigh items greater than 200 grams in mass, and trying to do this might harm the balance. Remember, this is a delicate precision instrument.

After the beam is arrested in either weight range, the procedure is the same. When the weight of the item in grams is found, "dial in" that amount with the two knobs on the left hand side (near the 100 g lever) color coded yellow and green. As you dial the weight, the digits will appear on the front of the cabinet. When the proper amount is dialed, carefully move the arrest lever down with a slow, steady turn of the wrist. The lighted dial will appear, and the right hand side knob (front of cabinet) is turned to align the mark with the lower of the two lighted scale divisions which the mark appears between. When these marks are aligned, the two lighted digits along with the two indicated on the right hand window on the cabinet front are fractional weight in grams (the decimal would appear before the lighted digits) and the whole number of grams weight is the amount "dialed in" on the left.

In general, be sure that the beam is in "arrest" position before placing weight on or taking weight off of the pan. Don't "dial in" weight unless the beam is arrested. The balance is sensitive to even a hand on the table near the balance, so be careful and painstaking in every movement while weighing.

SAMPLE ANALYTICAL DATA FORM

Company Name VALLEY PLANT #20

Sample Location _____

Blank Volume (V_a) 100 mlDate/Time wt. blank 10/2 8:00 ADate/Time wt. blank 10/2 3:00 PRelative Humidity in Lab 50 %Density of Acetone (ρ_a) .7857 mg/mlGross wt. 98.6774 gGross wt. 98.6774 gAve. Gross wt. 98.6774 gTare wt. 98.6773 gWeight of blank (m_{ab}) .0001 gAcetone blank residue concentration (C_a): $(C_a) = (m_{ab}) / (V_a) (\rho_a) = (.000001 \text{ mg/g})$ Acetone Blank Wt.: $W_a = C_a V_{aw} \rho_a = (.000001)(300)(.7857) = (.0002 \text{ g})$

	Run # 1	Run # 2	Run # 3
Acetone rinse volume (V_{aw}) ml	<u>300</u>	<u>300</u>	<u>300</u>
Date/Time of wt. <u>10/2 8:00 A</u> Gross wt. g	<u>99.5440</u>	<u>99.9700</u>	<u>99.8194</u>
Date/Time of wt. <u>10/2 3:00 P</u> Gross wt. g	<u>99.5440</u>	<u>99.9701</u>	<u>99.8194</u>
Average Gross wt. g	<u>99.5440</u>	<u>99.9701</u>	<u>99.8194</u>
Tare wt. g	<u>99.4589</u>	<u>99.9331</u>	<u>99.7910</u>
Less Acetone blank wt. (W_a) g	<u>.0002</u>	<u>.0002</u>	<u>.0002</u>
Weight of particulate in acetone rinse (m_a) g	<u>.0849</u>	<u>.0368</u>	<u>.0282</u>

Filter Numbers	#	TS6404	TS6512	TS6513
Date/Time of wt. <u>10/2 8:00 A</u> Gross wt. g		<u>.5817</u>	<u>.5625</u>	<u>.5706</u>
Date/Time of wt. <u>10/2 3:00 P</u> Gross wt. g		<u>.5815</u>	<u>.5626</u>	<u>.5706</u>
Average Gross wt. g		<u>.5816</u>	<u>.5626</u>	<u>.5706</u>
Tare wt. g		<u>.5560</u>	<u>.5585</u>	<u>.5556</u>

Weight of particulate on filter (m_f) g	<u>.0256</u>	<u>.0041</u>	<u>.0150</u>
Weight of particulate in acetone rinse (m_a) g	<u>.0849</u>	<u>.0368</u>	<u>.0282</u>
Total weight of particulate (m_n) g	<u>.1105</u>	<u>.0409</u>	<u>.0432</u>

NOTE: In no case should a blank residue greater than 0.01 mg/g (or 0.001% of the blank weight) be subtracted from the sample weight.

Remarks: _____

Signature of Analyst Thomas SmithSignature of Reviewer [Signature]

Valley # 20
Company Name

9-23-92
Date

REFERENCE METHOD 3: GAS ANALYSIS BY FYRITE

FUEL

F_o FACTORS

WOOD	1.0540
BARK	1.0830
ANTHRACITE	1.0699
BITUMINOUS	1.1398
LIGNITE	1.0761
OIL	1.3465
GAS	1.7489
PROPANE	1.5095
BUTANE	1.4791

$$O_2\% = 20.9 - [F_o \times CO_2\%]$$

RUN #1: = 20.9 - [x]

RUN #2: = 20.9 - [x]

RUN #3: = 20.9 - [x]

RUN 1:	CO _{2x}	<u>4.5</u>	CO _{2x}	<u>4.5</u>	CO _{2x}	<u>4.5</u>	AVG.	<u>4.5</u>
	O _{2x}	<u>14.8</u>	O _{2x}	<u>14.8</u>	O _{2x}	<u>14.8</u>	AVG.	<u>14.8</u>
	N _{2x}	<u> </u>	N _{2x}	<u> </u>	N _{2x}	<u> </u>	AVG.	<u>80.7</u>
RUN 2:	CO _{2x}	<u>4.5</u>	CO _{2x}	<u>4.5</u>	CO _{2x}	<u>4.5</u>	AVG.	<u>4.5</u>
	O _{2x}	<u>14.8</u>	O _{2x}	<u>14.8</u>	O _{2x}	<u>14.8</u>	AVG.	<u>14.8</u>
	N _{2x}	<u> </u>	N _{2x}	<u> </u>	N _{2x}	<u> </u>	AVG.	<u>80.7</u>
RUN 3:	CO _{2x}	<u>4.5</u>	CO _{2x}	<u>4.5</u>	CO _{2x}	<u>4.5</u>	AVG.	<u>4.5</u>
	O _{2x}	<u>14.8</u>	O _{2x}	<u>14.8</u>	O _{2x}	<u>14.8</u>	AVG.	<u>14.8</u>
	N _{2x}	<u> </u>	N _{2x}	<u> </u>	N _{2x}	<u> </u>	AVG.	<u>80.7</u>

OHIO BACK-HALF ANALYSIS

EVAPORATION

Plant Location VALLEY PLANT #20
 Sample Location OHIO
 Blank Volume (V_{wb}) 200 mL

Date 10/5/92

Relative Humidity in lab 52 %

Density of Water (ρ_w) = 1.0

Date/Time wt. blank 10/7 8:00 A

Gross wt. 175.2107 g

Date/Time wt. blank 10/8 1:30 P

Gross wt. 175.2105 g

Ave. Gross wt. 175.2106 g

Tare wt. 175.2104 g

Weight of blank (m_{wb}) .0002 g

Water blank residue concentration (C_w) $(C_w) = (M_{wb}) / (V_{wb}) (\rho_w) = (.000001 \text{ mg/g})$

Weight of blank residue in impinger rinse: $W_w = C_w V_w$

Impinger rinse volume (V_w)

mL

Date/Time of wt 10/7 8:30 A

Gross wt g

Date/Time of wt 10/8 1:30 P

Gross wt g

Average Gross wt g

Tare wt g

Less Water blank wt (W_w) g

Wt of particulate in impinger rinse (m_w) g

Run #1	Run #2	Run #3
460	460	450
177.4861	168.3940	126.4400
177.4861	168.3944	126.4402
177.4861	168.3942	126.4401
177.4761	168.3836	126.4270
.0005	.0005	.0005
.0095	.0101	.0126

Remarks: _____

Signature of Analyst

Thomas Smith

Signature of Reviewer

[Signature]

SECTION E:
CALCULATIONS

**VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO**

SUMMARY OF TEST DATA

SAMPLING TRAIN DATA

		09-23-92 RUN #1	09-23-92 RUN #2	09-24-92 RUN #3
	start	08:36	11:00	11:31
	finish	09:40	12:22	13:26
1. Sampling time, minutes	θ	60.0	60.0	60.0
2. Sampling nozzle diameter, in.	D_n	.3150	.3150	.3150
3. Sampling nozzle cross-sect. area, ft ²	A_n	.000541	.000541	.000541
4. Isokinetic variation	I	99.4	96.6	101.7
5. Sample gas volume - meter cond., cf.	V_m	45.275	41.545	44.100
6. Average meter temperature, °R	T_m	534	547	538
7. Avg. orifice pressure drop, in. H ₂ O	dH	1.76	1.54	1.56
8. Total particulate collected, mg.	M_n	110.50	40.90	43.20

VELOCITY TRAVERSE DATA

9. Stack area, ft ²	A	19.63	19.63	19.63
10. Absolute stack gas pressure, in. Hg.	P_s	30.10	30.10	30.10
11. Barometric pressure, in. Hg.	P_{bar}	30.10	30.10	30.10
12. Avg. absolute stack temperature, R°	T_s	689	683	681
13. Average $-\sqrt{V_{eff}} \cdot \bar{H}_{eff}$, (C _p = .84)	$-\sqrt{dP}$	0.59	0.55	0.56
14. Average stack gas velocity, ft./sec.	V_s	39.04	36.36	36.91

STACK MOISTURE CONTENT

15. Total water collected by train, ml.	V_{ic}	257.10	253.00	264.10
16. Moisture in stack gas, %	B_{ws}	20.79	22.37	21.77

EMISSIONS DATA

17. Stack gas flow rate, dscf/hr.(000's)	Q_{sd}	1684	1551	1591
18. Stack gas flow rate, cfm	acfm	45981	42825	43473
19. Particulate concentration, gr/dscf	C_s	0.0369	0.0153	0.0149
20. Particulate concentration, lb/hr	E	8.88	3.39	3.39
21. Particulate concentration, lb/mBtu	E'	0.00000	0.00000	0.00000

ORSAT DATA

22. Percent CO ₂ by volume	CO ₂	4.50	4.50	4.50
23. Percent O ₂ by volume	O ₂	14.80	14.80	14.80
24. Percent CO by volume	CO	.00	.00	.00
25. Percent N ₂ by volume	N ₂	80.70	80.70	80.70

Format: summaryR3

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Dry Gas Volume

$$V_{m(std)} = V_m \left[\frac{T_{(std)}}{T_m} \right] \left[\frac{P_{bar} + \frac{dH}{13.6}}{P_{(std)}} \right] = 17.64 \frac{^{\circ}R}{in.Hg} Y V_m \left[\frac{P_{bar} + \frac{dH}{13.6}}{T_m} \right]$$

Where:

- $V_{m(std)}$ = Dry Gas Volume through meter at standard conditions, cu. ft.
 V_m = Dry Gas Volume measured by meter, cu. ft.
 P_{bar} = Barometric pressure at orifice meter, in. Hg.
 P_{std} = Standard absolute pressure, (29.92 in. Hg.).
 T_m = Absolute temperature at meter $^{\circ}R$.
 T_{std} = Standard absolute temperature (528 $^{\circ}R$).
 dH = Average pressure drop across orifice meter, in. H_2O .
 Y = Dry gas meter calibration factor.
 13.6 = Inches water per inches Hg.

RUN 1:

$$V_{m(std)} = (17.64)(1.020)(45.275) \left[\frac{(30.10) + \frac{1.76}{13.6}}{534} \right] = 46.115 \text{ dscf}$$

RUN 2:

$$V_{m(std)} = (17.64)(1.020)(41.545) \left[\frac{(30.10) + \frac{1.54}{13.6}}{547} \right] = 41.288 \text{ dscf}$$

RUN 3:

$$V_{m(std)} = (17.64)(1.020)(44.100) \left[\frac{(30.10) + \frac{1.56}{13.6}}{538} \right] = 44.563 \text{ dscf}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Total Contaminants by Weight: GRAIN LOADING

Particulate concentration C'_s gr./dscf.

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{M_n}{V_{m(\text{std})}} \right]$$

Where:

C'_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr./dscf.

M_n = Total amount of particulate matter collected, mg.

$V_{m(\text{std})}$ = Dry gas volume through meter at standard conditions, cu. ft.

Run 1:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{110.50}{46.115} \right] = 0.0369 \text{ gr./dscf.}$$

Run 2:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{40.90}{41.288} \right] = 0.0153 \text{ gr./dscf.}$$

Run 3:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{43.20}{44.563} \right] = 0.0149 \text{ gr./dscf.}$$

VALLEY ASPHALT CORP./PLANT #20 -BACKHALF
CAMP DENNISON, OHIO

Total Contaminants by Weight: GRAIN LOADING

Particulate concentration C'_s gr./dscf.

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{M_n}{V_{m(\text{std})}} \right]$$

Where:

C'_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr./dscf.

M_n = Total amount of particulate matter collected, mg.

$V_{m(\text{std})}$ = Dry gas volume through meter at standard conditions, cu. ft.

Run 1:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{9.50}{46.115} \right] = 0.0032 \text{ gr./dscf.}$$

Run 2:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{10.10}{41.288} \right] = 0.0038 \text{ gr./dscf.}$$

Run 3:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{12.60}{44.563} \right] = 0.0044 \text{ gr./dscf.}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Dry Molecular Weight

$$M_d = 0.44(\%CO_2) + 0.32(\%O_2) + 0.28(\%CO + \%N_2)$$

Where:

M_d = Dry molecular weight, lb./lb.-mole.

$\%CO_2$ = Percent carbon dioxide by volume (dry basis).

$\%O_2$ = Percent oxygen by volume (dry basis).

$\%N_2$ = Percent nitrogen by volume (dry basis).

$\%CO$ = Percent carbon monoxide by volume (dry basis).

0.264 = Ratio of O_2 to N_2 in air, v/v.

0.28 = Molecular weight of N_2 or CO, divided by 100.

0.32 = Molecular weight of O_2 divided by 100.

0.44 = Molecular weight of CO_2 divided by 100.

Run 1:

$$M_d = 0.44(4.50\%) + 0.32(14.80\%) + 0.28(.00\% + 80.70\%) = 29.31 \frac{\text{lb}}{\text{lb-mole}}$$

Run 2:

$$M_d = 0.44(4.50\%) + 0.32(14.80\%) + 0.28(.00\% + 80.70\%) = 29.31 \frac{\text{lb}}{\text{lb-mole}}$$

Run 3:

$$M_d = 0.44(4.50\%) + 0.32(14.80\%) + 0.28(.00\% + 80.70\%) = 29.31 \frac{\text{lb}}{\text{lb-mole}}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Water Vapor Condensed

$$V_{wc_{std}} = \left[V_f - V_i \right] \left[\frac{P_w R T_{(std)}}{M_w P_{(std)}} \right] = 0.04707 \left[V_f - V_i \right]$$

$$V_{wsg_{std}} = \left[W_f - W_i \right] \left[\frac{R T_{(std)}}{M_w P_{(std)}} \right] = 0.04715 \left[W_f - W_i \right]$$

Where:

0.04707 = Conversion factor, ft.³/ml.

0.04715 = Conversion factor, ft.³/g.

$V_{wc_{std}}$ = Volume of water vapor condensed (standard conditions), scf.

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel (standard conditions), ml.

$V_f - V_i$ = Final volume of impinger contents less initial volume, ml.

$W_f - W_i$ = Final weight of silica gel less initial weight, g.

P_w = Density of water, 0.002201 lb/ml.

R = Ideal gas constant, 21.85 in.Hg. (cu.ft./lb.-mole)(°R).

M_w = Molecular weight of water vapor, 18.0 lb/lb-mole.

T_{std} = Absolute temperature at standard conditions, 528°R.

P_{std} = Absolute pressure at standard conditions, 29.92 inches Hg.

Run 1:

$$\begin{aligned} V_{wc(std)} &= (0.04707) (250.0) = 11.8 \text{ cu.ft} \\ V_{wsg(std)} &= (0.04715) (7.1) = 0.3 \text{ cu.ft} \end{aligned}$$

Run 2:

$$\begin{aligned} V_{wc(std)} &= (0.04707) (250.0) = 11.8 \text{ cu.ft} \\ V_{wsg(std)} &= (0.04715) (3.0) = 0.1 \text{ cu.ft} \end{aligned}$$

Run 3:

$$\begin{aligned} V_{wc(std)} &= (0.04707) (260.0) = 12.2 \text{ cu.ft} \\ V_{wsg(std)} &= (0.04715) (4.1) = 0.2 \text{ cu.ft} \end{aligned}$$

Format: vaporR3

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Moisture Content of Stack Gases

$$B_{ws} = \frac{V_{wc_{std}} + V_{wsg_{std}}}{V_{wc_{std}} + V_{wsg_{std}} + V_{m_{std}}} \times 100$$

Where:

B_{ws} = Proportion of water vapor, by volume, in the gas stream.

V_m = Dry gas volume measured by dry gas meter, (dcf).

$V_{wc_{std}}$ = Volume of water vapor condensed corrected to standard conditions (scf).

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel corrected to standard conditions (scf).

Run 1:

$$B_{ws} = \frac{11.8 + 0.3}{11.8 + 0.3 + 46.115} \times 100 = 20.79 \%$$

Run 2:

$$B_{ws} = \frac{11.8 + 0.1}{11.8 + 0.1 + 41.288} \times 100 = 22.37 \%$$

Run 3:

$$B_{ws} = \frac{12.2 + 0.2}{12.2 + 0.2 + 44.563} \times 100 = 21.77 \%$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Molecular Weight of Stack Gases

$$M_s = M_d (1 - B_{ws}) + 18 (B_{ws})$$

Where:

M_s = Molecular weight of stack gas, wet basis, (lb./lb.-mole).

M_d = Molecular weight of stack gas, dry basis, (lb./lb.-mole).

Run 1:

$$M_s = 29.31 (1 - 20.79) + 18 (20.79) = 26.96 \text{ (lb./lb.-mole)}$$

Run 2:

$$M_s = 29.31 (1 - 22.37) + 18 (22.37) = 26.78 \text{ (lb./lb.-mole)}$$

Run 3:

$$M_s = 29.31 (1 - 21.77) + 18 (21.77) = 26.85 \text{ (lb./lb.-mole)}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Stack Gas Velocity

$$V_s = K_p C_p \left[-\sqrt{dP} \right] \text{ avg. } \sqrt{\frac{T_s(\text{avg.})}{P_s M_s}}$$

Where:

- V_s = Average velocity of gas stream in stack, ft./sec.
 K_p = 85.49 ft/sec $\left[\frac{\text{g/g-mole} \cdot (\text{mm Hg})}{(^{\circ}\text{K})(\text{mm H}_2\text{O})} \right]^{1/2}$
 C_p = Pitot tube coefficient, (dimensionless).
 dP = Velocity head of stack gas, in. H_2O .
 P_{bar} = Barometric pressure at measurement site, (in. Hg).
 P_g = Stack static pressure, (in. Hg).
 P_s = Absolute stack gas pressure, (in. Hg) = $P_{\text{bar}} + P_g$
 P_{std} = Standard absolute pressure, (29.92 in. Hg).
 t_s = Stack temperature, ($^{\circ}\text{f}$).
 T_s = Absolute stack temperature, ($^{\circ}\text{R}$). = $460 + t_s$.
 M_s = Molecular weight of stack gas, wet basis, (lb/lb-mole).

Run 1:

$$V = (85.49) (.84) (0.59) \sqrt{\frac{689}{(30.10)(26.96)}} = 39.04 \text{ ft/sec.}$$

Run 2:

$$V = (85.49) (.84) (0.55) \sqrt{\frac{683}{(30.10)(26.78)}} = 36.36 \text{ ft/sec.}$$

Run 3:

$$V = (85.49) (.84) (0.56) \sqrt{\frac{681}{(30.10)(26.85)}} = 36.91 \text{ ft/sec.}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Stack Gas Flow Rate

$$Q_{sd} = 3600 \left[1 - B_{wc} \right] V_s A \left[\frac{T_{std}}{T_{stk}} \right] \left[\frac{P_s}{P_{std}} \right]$$

Where:

- Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, (dscf/hr).
- A = Cross sectional area of stack, (ft.²).
- 3600 = Conversion factor, (sec./hr.).
- t_s = Stack temperature, (°f).
- T_s = Absolute stack temperature, (°R).
- T_{std} = Standard absolute temperature, (528°R).
- P_{bar} = Barometric pressure at measurement site, (in.Hg.).
- P_g = Stack static pressure, (in.Hg.).
- P_s = Absolute stack gas pressure, (in.Hg.); = $P_{bar} + P_g$
- P_{std} = Standard absolute pressure, (29.92 in.Hg.).

Run 1:

$$Q_{sd} = 3600(1 - .2079)(39.04)(19.63) \left[\frac{528}{689} \right] \left[\frac{30.10}{29.92} \right] = 1684737.5 \frac{dscf}{hr}$$

Run 2:

$$Q_{sd} = 3600(1 - .2237)(36.36)(19.63) \left[\frac{528}{683} \right] \left[\frac{30.10}{29.92} \right] = 1551295.0 \frac{dscf}{hr}$$

Run 3:

$$Q_{sd} = 3600(1 - .2177)(36.91)(19.63) \left[\frac{528}{681} \right] \left[\frac{30.10}{29.92} \right] = 1591592.5 \frac{dscf}{hr}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Emissions Rate from Stack

$$E = \frac{(C_s) (Q_{sd})}{7000 \text{ gr./lb.}} = \text{lb. / hr.}$$

Where:

E = Emissions rate, lb/hr.

C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr/dscf.

Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, dscf/hr.

Run 1:

$$E = \frac{(0.0369) (1684737.5)}{7000} = 8.88 \text{ lb. / hr.}$$

Run 2:

$$E = \frac{(0.0153) (1551295.0)}{7000} = 3.39 \text{ lb. / hr.}$$

Run 3:

$$E = \frac{(0.0149) (1591592.5)}{7000} = 3.39 \text{ lb. / hr.}$$

VALLEY ASPHALT CORP./PLANT #20 -BACKHALF
CAMP DENNISON, OHIO

Emissions Rate from Stack

$$E = \frac{(C_s) (Q_{sd})}{7000 \text{ gr./lb.}} = \text{lb. / hr.}$$

Where:

E = Emissions rate, lb/hr.

C_s = Concentration of particulate matter in stack gas, dry basis,
corrected to standard conditions, gr/dscf.

Q_{sd} = Dry volumetric stack gas flow rate corrected to
standard conditions, dscf/hr.

Run 1:

$$E = \frac{(0.0032) (1684737.5)}{7000} = 0.77 \text{ lb. / hr.}$$

Run 2:

$$E = \frac{(0.0038) (1551295.0)}{7000} = 0.84 \text{ lb. / hr.}$$

Run 3:

$$E = \frac{(0.0044) (1591592.5)}{7000} = 1.00 \text{ lb. / hr.}$$

VALLEY ASPHALT CORP. - PLANT #20
CAMP DENNISON, OHIO

Isokinetic Variation

$$I = 100 T_s \left[\frac{0.002669 V_{ic} + \frac{(V_m / T_m) (P_{bar} + dH / 13.6)}{60 \theta V_s P_s A_n}}{1} \right]$$

Where:

- I = Percent isokinetic sampling.
- 100 = Conversion to percent.
- T_s = Absolute average stack gas temperature, $^{\circ}R$.
- 0.002669 = Conversion factor, $Hg - ft^3/ml - ^{\circ}R$.
- V_{ic} = Ttl vol of liquid collected in impingers and silica gel, ml.
- T_m = Absolute average dry gas meter temperature, $^{\circ}R$.
- P_{bar} = Barometric pressure at sampling site, (in. Hg).
- dH = Av pressure differential across the oriface meter, (in. H_2O).
- 13.6 = Specific gravity of mercury.
- 60 = Conversion seconds to minutes.
- θ = Total sampling time, minutes.
- V_s = Stack gas velocity, ft./sec.
- P_s = Absolute stack gas pressure, in. Hg.
- A_n = Cross sectional area of nozzle, ft^2 .

Run 1:

$$I = (100)(689) \left[\frac{(0.002669)(257.1) + \frac{45.275}{534} \left[30.10 + \frac{1.76}{13.6} \right]}{60 (60.0) (39.04) (30.10) (.000541)} \right] = 99.4\%$$

Run 2:

$$I = (100)(683) \left[\frac{(0.002669)(253.0) + \frac{41.545}{547} \left[30.10 + \frac{1.54}{13.6} \right]}{60 (60.0) (36.36) (30.10) (.000541)} \right] = 96.6\%$$

Run 3:

$$I = (100)(681) \left[\frac{(0.002669)(264.1) + \frac{44.100}{538} \left[30.10 + \frac{1.56}{13.6} \right]}{60 (60.0) (36.91) (30.10) (.000541)} \right] = 101.7\%$$

SECTION F:
FIELD DATA

RAMCON ENVIRONMENTAL CORPORATION

Plant 1000g 20
 Location Ohio
 Operator John
 Date 9-23-92
 Run No. 1
 Sample Box No. 1
 Meter Box No. C-282
 Meter H @ 1.505
 C Factor 1.02
 Pilot Tube Coefficient Cp 94

60
0.332

Ambient Temperature 60
 Barometric Pressure 30.10 450 605.7
 Assumed Moisture, % 18 200 574.6
 Probe Length, m(ft) 6 ft 250 7.1
 Nozzle Identification No. 9005411
 Avg. Calibrated Nozzle Dia. (in.) 3/5/315/315
 Probe Heater Setting 4
 Leak Rate, m³/min. (ccm) 205 at 1210
 Probe Liner Material polyester fiber
 Static Pressure, mm Hg (in. Hg) 4.01
 Filter No. TS-6707

Schematic of Stack Cross Section

TRAV. PT.	SAMPLING TIME (θ)min.	VACUUM in. Hg	STACK TEMP (°F)	VELOCITY HEAD (P _g)	PRESSURE DIFF. ORF. in. H ₂ O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F	FILTER HOLDER TEMP °F	GAS TEMP LOG CONDENSER OR LAST IMPINGER °F	
1	8:36	2	220	30	1.51	976.9	62	50	245	55
2	8:41	2	219	22	1.15	979.6	76	50	250	53
3	8:43-30	2	220	40	2.0	981.5	84	50	255	52
4	8:46	4	234	43	2.1	984.5	86	50	258	50
5	8:48-30	6	234	45	2.2	986.8	90	52	255	50
6	8:51	6	235	40	2.0	988.3	90	52	255	50
7	8:53-30	8	234	45	2.2	989.6	90	52	242	50
8	8:56	8	233	45	2.2	991.9	90	54	243	52
9	8:58-30	5	232	40	2.0	994.1	90	54	244	59
10	9:01	3	230	35	1.7	996.3	90	54	240	59
11	9:03-30	3	230	35	1.7	998.3	90	54	240	57
12	9:06	2	230	30	1.5	1000.0	94	56	240	56
13	9:10-30	2	230	20	0.98	1001.9	82	60	238	56

11/11/92

RAMCON

emissions test log sheet, cont.

DATE 9-23-92 LOCATION CAN TEST NO. 1

TRAVERSE POINT	SAMPLING TIME (min)	VACUUM mm Hg (in. Hg)	STACK TEMP. (°F)	VELOCITY FEET/SEC (ft/min)	ORIF. DIFF. PRESSURE INCHES H ₂ O	GAS VOLUME SCFM (13)	GAS SAMPLE TEMP. (°F)		SAMPLE BOX TEMP. (°F)	IMPINGER TEMP. (°F)
2	9:15-30	2	215	20	98	1003.1	92	60	242	50
3	9:18	2	225	20	98	1004.3	94	60	244	50
4	9:20-30	2	229	21	104	1006.0	96	60	246	50
5	9:23	2	231	23	111	1007.3	94	60	245	49
6	9:25-30	2	232	25	125	1008.8	96	60	243	48
7	9:28	2	233	25	20	1010.2	98	60	241	48
8	9:30-30	2	234	40	20	1012.7	98	60	238	47
9	9:33	2	234	45	22	1015.2	98	60	232	46
10	9:35-30	2	231	50	25	1017.2	100	60	225	46
11	9:38	2	229	50	25	1019.4	100	60	217	46
12	9:40-30	2	228	50	25	1021.675	100	60	217	46
							91.841	56.166		

RAMCON ENVIRONMENTAL CORPORATION

Ambient Temperature 60
 Barometric Pressure 30.10 IN.
 Assumed Moisture, % 19
 Probe Length, m(ft) 6.8
 Nozzle Identification No. 000541
 Avg. Calibrated Nozzle Dia., (in.) 3.15/3.15/3.15
 Probe Heater Setting 4
 Leak Rate, m³/min, (cfm) 0.27 at 15.26
 Probe Liner Material Stainless Steel
 Static Pressure, mm Hg (in. Hg) 1.01
 Filter No. TS-6513

Plant Valley 4.9
 Location Copper River
 Operator John
 Date 9-23-92
 Run No. 2
 Sample Box No. 2
 Meter Box No. 2
 Meter H. 1.505
 C Factor 1.02
 Pitot Tube Coefficient 0.84

Schematic of Stack Pipe Section

TRAV. PT.	SAMPLING TIME (min.)	VACUUM (in. Hg)	STACK TEMP (°F)	VELOCITY HEAD (PP)	PRESSURE DIFF. (in. Hg)	GAS SAMPLE VOLUME (ft ³)	GAS SAMPLE TEMP. AT DRY GAS METER (°F)	FILTER HOLDER TEMP (°F)	GAS TEMP LNG CONDENSER OR LAST IMPIRIMER (°F)
1	11:00-20	2	225	20	98	223	88	233	60
2	11:20-30	2	225	20	98	251	92	263	57
3	11:30-40	2	225	20	98	269	100	259	58
4	11:40-50	2	225	20	98	281	100	257	61
5	11:50-00	2	225	20	98	297	98	249	58
6	11:30-30	2	225	20	98	313	98	247	58
7	11:33-40	2	225	20	98	337	100	247	58
8	11:35-50	2	225	20	98	347	98	247	58
9	11:38-00	2	225	20	98	357	98	248	58
10	11:40-10	2	225	20	98	371	100	248	58
11	11:43-20	2	225	20	98	40.8	100	245	58
12	11:45-30	2	225	20	98	43.0	100	245	61
13	11:48-40	2	225	20	98	44.3	100	245	61

Valley 20

RAMCON & emissions test log sheet. CONT. DATE 9-23-92 LOCATION Greenwood TEST NO. 2

RAMCON & emissions test log sheet. CONT. DATE 9-23-92 LOCATION Greenwood TEST NO. 2

[illegible]

RAVCON ENVIRONMENTAL CORPORATION

Plant Valley

Location Quincy

Operator W. J. Smith

Date 9-24-92

Run No. 3

Sample Box No. 280

Meter Box No. 1508

Meter H. 100

C Factor 100

Pitot Tube Coefficient 0.99

Ambient Temperature 70

Barometric Pressure 29.8

Assumed Moisture, % 10

Probe Length, m(ft) 6.0

Nozzle Identification No. 0005411

Avg. Calibrated Nozzle Dia., (in.) 3/16/315

Probe Heater Setting 4

Leak Rate, m³/min. (cfm) 0.01 at 4 m³

Probe/Liner Material 0.000000

Static Pressure, mm Hg (in. Hg) 7.01

Filter No. TS-512

Schematic of Stack Cross Section

TRAV. TIME	SAMPLING TIME (8) min.	VACUUM in. Hg	STACK GROSS AREA (sq. ft)	VELOCITY HEAD (ft)	VELOCITY (ft/min)	PRESSURE DIFF. (in. H ₂ O)	GAS SAMPLE VOLUME (ft ³)	GAS SAMPLE TEMP. AT DRY GAS METER °F	FILTER HOLDER TEMP °F	GAS TEMP LVG CONDENSER OR LAST IMPIINGER °F
1	10:31	2	216	116	128	1.78	225	72	223	60
2	10:36	2	218	118	130	1.88	243	80	222	60
3	10:38-30	2	216	130	135	1.5	280	92	223	60
4	10:41	2	209	127	133	1.3	241	86	234	60
5	10:43-30	3	219	132	136	1.6	262	90	232	60
6	10:46	3	226	135	137	1.7	276	92	231	52
7	10:48-30	3	231	140	140	2.0	299	92	229	52
8	10:51	3	232	145	146	2.2	316	94	227	52
9	11:53-30	3	228	143	143	2.1	298	94	228	52
10	12:28-30	4	220	145	145	2.2	287	90	225	49
11	12:31	3	223	143	143	2.1	293	90	251	49
12	12:33-30	3	209	132	132	1.7	263	96	237	50
13	12:34-30	3	216	125	125	1.7	249	90	240	50

Valley

TEST NO. 3

[illegible]

SECTION G:
CALIBRATION

POSTTEST DRY GAS METER CALIBRATION DATA FORM (English units)

Test number Date 9-25-92 Meter box number C-282 Plant
 Barometric pressure, $P_b = 29.70$ in. Hg Dry gas meter number 147810 Pretest Y

Orifice manometer setting, (ΔH), in. H_2O	Gas volume		Temperature				Vacuum setting, in. Hg	Y_i	$V_w P_b (t_d + 460)$ $V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)$
	Wet test meter (V_w), ft^3	Dry gas meter (V_d), ft^3	Wet test meter (t_w), $^{\circ}F$	Inlet (t_{d_i}), $^{\circ}F$	Outlet (t_{d_o}), $^{\circ}F$	Average (t_d), $^{\circ}F$			
5.198	18.5	16.639 17.857	72	90.4 73	74	54.3	1	.99	1.550
10.276	10	17.426 18.702	72	94.1 74	75	54.9	1	1.0	1.553
10.12	10	16.507 17.167	72	100.1 71	73	548.8	1	1.01	1.505
								$Y = 1.0$	1.536

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d where

V_w = Gas volume passing through the wet test meter, ft^3 .

V_d = Gas volume passing through the dry gas meter, ft^3 .

t_w = Temperature of the gas in the wet test meter, $^{\circ}F$.

t_{d_i} = Temperature of the inlet gas of the dry gas meter, $^{\circ}F$.

t_{d_o} = Temperature of the outlet gas of the dry gas meter, $^{\circ}F$.

t_d = Average temperature of the gas in the dry gas meter, obtained by the average of t_{d_i} and t_{d_o} , $^{\circ}F$.

ΔH = Pressure differential across orifice, in. H_2O .

Y_i = Ratio of accuracy of wet test meter to dry gas meter for each run.

Y = Average ratio of accuracy of wet test meter to dry gas meter for all three runs;
 tolerance = pretest $Y \pm 0.05Y$.

P_b = Barometric pressure, in. Hg.

θ = Time of calibration run, min.

Quality Assurance Handbook M4-2.4A

POSTTEST DRY GAS METER CALIBRATION DATA FORM (English units)

Test number 9-18-92 Date 7-18-92 Meter box number 2-282 Plant Pretest Y
 Barometric pressure, $P_b = 31.7$ in. Hg Dry gas meter number Pretest Y

Orifice manometer setting, (ΔH), in. H_2O	Gas volume		Temperature		Time (θ), min	Vacuum setting, in. Hg	Y_i	$V_w P_b (t_d + 460)$ $V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)$
	Wet test meter (V_w), ft ³	Dry gas meter (V_d), ft ³	Wet test meter (t_w), °F	Dry gas meter Inlet (t _d), °F	Outlet (t _d), °F			
1	10-8	10-8	77	77	77	1	1.03	1.03
2	10	10	77	77	77	1	1.03	1.03
3	10	10	77	77	77	1	1.03	1.03
							$Y = 1.03$	

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d where

V_w = Gas volume passing through the wet test meter, ft³.

V_d = Gas volume passing through the dry gas meter, ft³.

t_w = Temperature of the gas in the wet test meter, °F.

t_{d_i} = Temperature of the inlet gas of the dry gas meter, °F.

t_{d_o} = Temperature of the outlet gas of the dry gas meter, °F.

t_d = Average temperature of the gas in the dry gas meter, obtained by the average of t_{d_i} and t_{d_o} , °F.

ΔH = Pressure differential across orifice, in. H_2O .

Y_i = Ratio of accuracy of wet test meter to dry gas meter for each run.

Y = Average ratio of accuracy of wet test meter to dry gas meter for all three runs;

tolerance = pretest $Y \pm 0.05Y$.

P_b = Barometric pressure, in. Hg.

θ = Time of calibration run, min.

Quality Assurance Handbook M4-2.4A

METER BOX CALIBRATION DATA AND CALCULATION FORM

(English units)

Date 2-13-92

Meter box number C-282

Barometric pressure, $P_b = 29.98$ in. Hg

Calibrated by Bob - Jim

Orifice manometer setting (ΔH), in. H ₂ O	Gas volume		Wet test meter (t_w), °F	Temperature			Time (θ), min	Y_i	ΔH_{i-1} in. H ₂ O
	Wet test meter (V_w), ft ³	Dry gas meter (V_d), ft ³		Inlet (t_{di}), °F	Outlet (t_{do}), °F	Avg ^a (t_d), °F			
0.5	5								
10.47 1.0	10	77.953 88.423	68	92 102	62 62	79.5	1201	.974	1.58
1.5	10								
369 2.0	10	77.120 77.489	68	92 106	62 64	81	1250	.983	1.20
285 3.0	10	76.541 76.826	68	96 108	64 64	82.5	997	.991	1.62
4.0	10								
Avg								.983	1.63

ΔH , in. H ₂ O	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)}$	$\Delta H_{i-1} = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460) \theta}{V_w} \right]^2$
0.5	0.0368		
1.0	0.0737		
1.5	0.110		
2.0	0.147		
3.0	0.221		
4.0	0.294		

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d .

Quality Assurance Handbook M4-2.3A (front side)

RAMCON ENVIRONMENTAL CORPORATION

Lear Siegler Stack Sampler

Nozzle Diameter Calibration

Date _____ Signature _____

Nozzle No.	Average Diameter	Nozzle No.	Average Diameter
1	_____	7	_____
2	_____	8	_____
3	_____	9	_____
4	_____	10	_____
5	_____	11	_____
6	_____	12	_____

Pitot Tube Calibration (S Type)

Pitot Tube Identification No. 6-4 Date 6-9-91

Calibrated by: S. Buck

"A" SIDE CALIBRATION

Run No.	Δp std cm H ₂ O (in. H ₂ O)	Δp (s) cm H ₂ O (in. H ₂ O)	C_p (s)	DEVIATION $C_p(s) - \bar{C}_p(A)$
1	2.2	3.2	.829	.001
2	1.5	2.6	.832	.002
3	1.1	1.6	.829	.001
$\bar{C}_p$ (SIDE A)			.830	

"B" SIDE CALIBRATION

Run No.	Δp std cm H ₂ O (in. H ₂ O)	Δp (s) cm H ₂ O (in. H ₂ O)	C_p (s)	DEVIATION $C_p(s) - \bar{C}_p(B)$
1	2.2	3.2	.829	.001
2	1.5	2.6	.832	.002
3	1.1	1.6	.829	.001
$\bar{C}_p$ (SIDE B)			.830	

$$\text{AVERAGE DEVIATION} = \sigma(A \text{ OR } B) = \frac{\sum |C_p(s) - \bar{C}_p(A \text{ OR } B)|}{3} \quad + \text{MUST BE} \leq 0.01$$

$$|\bar{C}_p(\text{SIDE A}) - \bar{C}_p(\text{SIDE B})| + \text{MUST BE} \leq 0.01$$

$$C_p(s) = C_p(std) \sqrt{\frac{\Delta p \text{ std}}{\Delta p s}}$$

RAMCON

Lear Siegler Stack Sampler

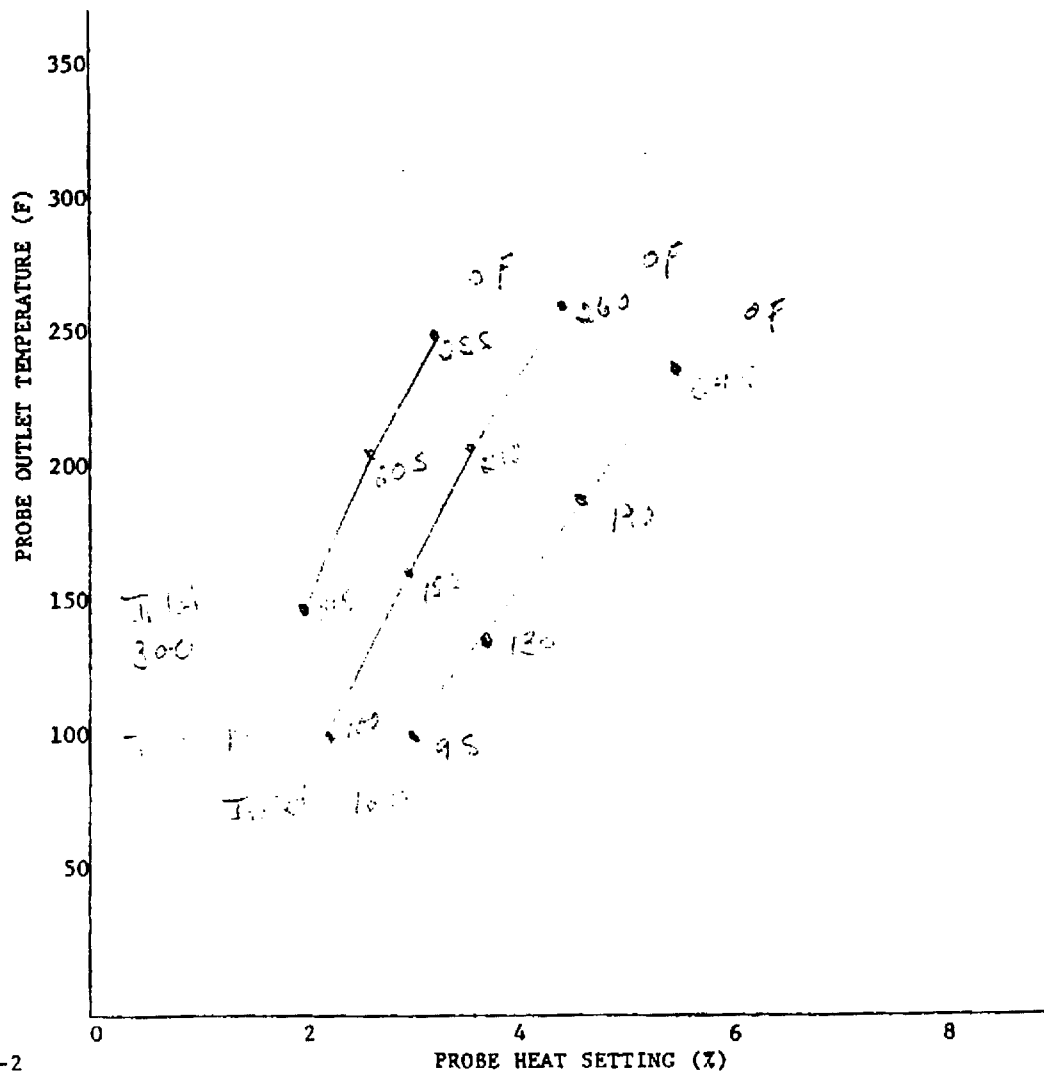
Heating Probe Calibration

Probe No. 641 Probe Length 6'

Date of Calibration 5-7-90 Signature Sam Cunnery

Name of Company to be tested _____

Note: 3 ft. probe - 5 min. warmup
6 ft. probe - 15 min. warmup
10 ft. probe - 30 min. warmup
Calibration flow rate = .75 CFM



STACK TEMPERATURE SENSOR CALIBRATION DATA FORM

Date 5-5-90 Thermocouple number 64
 Ambient temperature 20 °C Barometric pressure 29.88 in. Hg
 Calibrator Sturges Reference: mercury-in-glass ✓
 other _____

Reference point number	Source ^a (specify)	Reference thermometer temperature, °C	Thermocouple potentiometer temperature, °C	Temperature difference, % ^b
A	Ice Bath	32	32	0
B	Boiling water	212	211	.005
C	Boiling oil	381	378	.008
C	2:50:4			

^aType of calibration system used.

^b
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%$$

4A

TYPE S PITOT TUBE INSPECTION DATA FORM

Pitot tube assembly level? ☒ yes ☐ no

Pitot tube openings damaged? ☐ yes (explain below) ☒ no

$\alpha_1 = 1.3^\circ$ ($<10^\circ$), $\alpha_2 = 0.8^\circ$ ($<10^\circ$), $\beta_1 = 0.5^\circ$ ($<5^\circ$),
 $\beta_2 = 1.8^\circ$ ($<5^\circ$)

$\gamma = 2.9^\circ$, $\theta = 1.7^\circ$, $A = .97$ cm (in.)

$z = A \sin \gamma = .05$ cm (in.); <0.32 cm ($<1/8$ in.),

$w = A \sin \theta = .03$ cm (in.); $<.08$ cm ($<1/32$ in.)

$P_A = .48$ cm (in.) $P_D = .49$ cm (in.)

$D_t = .38$ cm (in.)

Comments: _____

Calibration required? ☐ yes ☒ no

SECTION H:

RAMCON PERSONNEL

4B

TYPE S PITOT TUBE INSPECTION DATA FORM

Pitot tube assembly level? ☒ yes ☐ no

Pitot tube openings damaged? ☐ yes (explain below) ☐ no

$\alpha_1 = 2.3^\circ$ ($<10^\circ$), $\alpha_2 = 5^\circ$ ($<10^\circ$), $\beta_1 = 1.8^\circ$ ($<5^\circ$),
 $\beta_2 = 1.8^\circ$ ($<5^\circ$)

$\gamma = 3.2^\circ$, $\theta = 1.0^\circ$, $A = .98$ cm (in.)

$z = A \sin \gamma = .05$ cm (in.); <0.32 cm ($<1/8$ in.),

$w = A \sin \theta = .02$ cm (in.); $<.08$ cm ($<1/32$ in.)

$P_A = .49$ cm (in.) $P_B = .49$ cm (in.)

$D_t = .38$ cm (in.)

Comments: _____

Calibration required? ☐ yes ☒ no

Name: Mr. G. Sumner Buck, III
Title: President

Qualifications: Mr. Buck is a graduate of the University of Mississippi with graduate studies at Memphis State University and State Technical Institute of Memphis. He is a graduate of the EPA 450 "Source Sampling for Particulate Pollutant's" course and the 474 "Continuous Emissions Monitoring" courses outlined by EPA at Research Triangle Park, N.C. He has been directly involved in conducting and supervising air emission testing for over 15 years. He has personally conducted over 400 air emission tests. He currently sponsors and directs visual emission certification schools for US EPA Reference Method 9.

Project Duties: Mr. Buck will be responsible for the overall supervision of the project. He provided supervision for the project preparation, testing schedules for each team on-site, and overall organization between the testing crews and facility.

Name: Mr. Joe Sewell
Title: Vice President

Qualifications: Mr. Sewell is currently serving as the Vice President of RAMCON Environmental Corporation. Mr. Sewell is a graduate of Christian Brothers University in Memphis, Tennessee where he obtained a Bachelor of Science degree in Chemical Engineering. He has conducted and supervised air emissions testing projects ranging a broad spectrum of facility process categories. His accomplishments include the development of the instrumental branch of emissions testing utilizing continuous emission monitors and gas chromatography. Mr. Sewell performs a major role in the upgrading of testing capabilities and professional quality that RAMCON Environmental Corporation offers.

Project Duties: Mr. Sewell provides staff engineering and project administration to ensure the integrity of the requested services. He served as the primary contact person for RAMCON Environmental Corporation handling all correspondence between the facility personnel involved in the project and respective state agency

representative(s). He provides project leadership to RAMCON Environmental Corporation field supervisors and managers involved in the test project.

Name: Mr. Ray Jenkins
Title: Source Sampling Director

Qualifications: Mr. Jenkins is serving as the Source Sampling Director for RAMCON Environmental Corporation. He was promoted to this leadership position after gaining a significant amount of experience in conducting and providing field supervision of a variety of air testing projects. Mr. Jenkins has personally conducted and/or supervised all of the prevalent EPA approved procedures with expertise in the instrumental analyzer procedures. He graduated from Memphis State University where he obtained a Bachelor of Science degree in Biology. He is also currently certified to conduct US EPA Reference Method 9 for the visual determination of emission opacity.

Project Duties: Mr. Jenkins provided project leadership to the Team Leader Field Technicians. He ensured that the test crew(s) involved in the test project were properly informed to his respective duties and responsibilities during the testing process. Mr. Jenkins also serves as the Quality Assurance/Quality Control Coordinator and provided guidance in QA/QC to each Team Leader with regard to sample integrity.

Name: Mr. Thomas South
Title: Laboratory Technician

Qualifications: Mr. South is currently serving as Laboratory Technician. He is proficient in conducting many analysis procedures such as front and back-half particulate analysis, titrations, extractions, etc. He received an Associate degree in Chemical Engineering from State Technical Institute in Memphis, Tennessee.

Project Duties: Mr. South conducts the laboratory analysis on the particulate samples. He is also responsible for accepting the remaining field samples from the Field Sample Bank Manager and performing inspection as to integrity. He

documents the transfer on the chain of custody forms and distributed the subcontracted samples to the respective laboratories.

Name: Allen Turner
Title: Team Leader

Qualifications: Mr. Turner has been employed with RAMCON Environmental Corporation for five years. Altogether, he has sampled approximately 300 stacks of all types. Mr. Turner became qualified for a Team Leader in 1988 and has served as such since that time. He is a current V.E. reader and continues his studies at State Technical Institute in Memphis, Tennessee. Mr. Turner has extensive experience in EPA Methods 1-9.

Project Duties: Mr. Turner is responsible for isokinetic sampling procedures, including but not limited to, Method 5 for particulate, multi-metals, PAH, calibration and cleaning of necessary equipment for his testing. His duties on-site include assembling the sample train, leak checking the system, operation of the train and recording the test data on the field data forms.