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AP42 Section: 11.1

Reference Number: 317

Title: Source Sampling For Particulate Emissions, Valley Asphalt Corp., Plant #17, Camp Dennison, OH,

Ramcon Environmental Corp., Memphis, TN,

June 6, 1988.

DEPARTMENT OF ENVIRONMENTAL MANAGEMENT

OFFICE OF AIR MANAGEMENT

OFFICE MEMORANDUM

To: Herm Carney
 From: Dave Cline
 Subject: Valley Asphalt Plant #17
 Source ID No. Not Assigned Yet

Date: 6/14/94
 Thru: E. Surla *6/20*
 D. Dubenetzky *6/21*

The subject company has submitted a report concerning particulate testing of their number 17 plant formerly located in Camp Dennison Ohio. The original test protocol and the actual test observation were completed by the Ohio EPA. The plant is relocating to Ripley County. The testing took place in Ohio on June 6, 1988 and was conducted by Ramcon Environmental Corporation. The purpose of the testing was to determine compliance with the Ohio EPA's operating permit conditions. I have reviewed this report and determined that the sampling procedures used and the results obtained are conditionally acceptable to this office relative to demonstrating initial compliance with the particulate NSPS standard. An initial site inspection by D.J. Knotts will be conducted on this plant to determine it's compliance status. Based upon the results of this inspection, a retest may be requested by the Air Compliance Section should conditions warrant it. A copy of the original test report is filed in the Compliance Data Section. The following is a summary of the test results:

PLANT #17

Maximum Permitted Rate:	350TPH
Average Rate During Test:	314TPH
Average Measured Emissions:	.014gr/dscf
Allowable Emissions:	.04gr/dscf
Average Measured Opacity:	N/A
Allowable Opacity:	20%
Type of Fuel:	No. 2 Fuel Oil

STATUS: CONDITIONAL (at 90% of maximum rated capacity)

cc: N. Landau
 D. Cline
 General File Ripley County

RAMCON

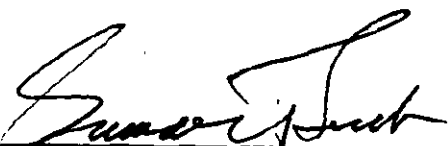
ENVIRONMENTAL CORPORATION

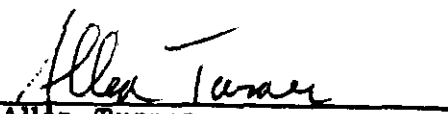
SOURCE SAMPLING
for
PARTICULATE EMISSIONS
VALLEY ASPHALT CORPORATION
CAMP DENNISON, OHIO

June 6, 1988

Plant 17


Paul Prottengeler
Valley Asphalt Corp.


G. Sumner Buck, III
President


Allen Turner
Team Leader

SOURCE TESTED: Valley Industrial Corp. *Plant 7*
 TESTING COMPANY: Sander
 DATE OF TEST: 8/6/88
 RUN NO: 1

Molecular Weight of Stack Gas, Dry Basis, Md
 SC02 = 4
 SO2 = 15.5
 Md = 29.28

Volume of Water Vapor Collected @ STD. COND., Vwv
 Vlc = 243
 Vw(std) = 11.43072

Volume of Dry Gas Collected @ STD. COND., Vm(std)
 Tm = 81.8823529 541.882353 121.7308
 Vm = 53.48 100.7308
 Pb = 29.98
 delta H = 2.16705882
 gamma = 0.984
 Vm(std) = 51.78

RUN 1

Moisture Content of Stack Gas, Bwa
 Bwa = 18.088
 1-Bwa = 81.912

Molecular Weight of Stack Gas, Ms
 Ms = 27.22

Area of Stack (enter diameter in inches), As
 L = 75.5
 W = 28
 diameter = 0
 No. of Stacks = 1
 As(L*W) = 14.6805556
 As(dia.) = 0

Static Pressure, Ps
 Static = 0.05
 Ps = 29.93

Stack Gas Velocity, Vs, fps
 Cp = 0.84
 sqrt(delta p) = 1.02100028 0.6736075
 Ts = 203.5
 Vs = 66.13

Stack Gas Flowrate, Qs, acfm
 As = 14.6805556
 Qs = 58248.10 3404897.98

Stack Gas Flowrate @ STD. COND., Dry Basis, Qstd.
 Qstd = 38149.55 2288973.07

Velocity at the Nozzle, Vn
 Dia. of Nozzle = 0.23 An 0.00028852
 Time of Run = 70
 Vn = 65.27

Isokinetic, Xi
 Xi = 98.70

Isokinetic measured from intermediate values
 Xi = 99.677552

Pollutant Mass Emission Rate, PMR, lbs./hr.
 Pb = 0.0471
 PMR = 4.5853

Grains per acf
 GC/acf = 0.0112

Grains per dscf
 GC/dscf = 0.0140
 Concentration, mg/dscf
 mg/dscf = 0.9053535

Particulate Emissions Rate
 Factor = n/a
 lb/MMBtu = *VALUE*

Theoretical Heat Input (THI)
 Qstd = 2288973.07
 Eds = n/a

THI(MMBtu/hr) = *VALUE*

POLYMER TESTER Polys. Analysis Corp.
 15111B COMPANY Bldg.
 15111B ST. 07678
 RUN NO. 2

Molecular Weight of Stack Gas, Dry Basis, M_d
 M_{CO_2} 4
 M_{O_2} 15.5 15.5
 M_d 29.26

Volume of Water Vapor Collected @ STD. COND., V_{wv}
 V_{ic} 200
 $V_{wv}(std)$ 13.17120

Volume of Dry Gas Collected @ STD. COND., $V_m(std)$
 T_m 97.9142857 557.914286 121.7306
 V_m 54.261 100.7306
 P_b 29.95
 $delle H$ 2.15342857
 $gamma$ 0.984
 $V_m(std)$ 51.03

RUN 2

Moisture Content of Stack Gas, g_{wa}
 R_{wa} 20.528
 $1-R_{wa}$ 79.472

Molecular Weight of Stack Gas, M_s
 M_s 28.95

Area of Stack (outer diameter in inches), A_s
 L 75.3
 W 20
 diameter 0
 No. of Stacks 1
 $A_s(L+W)$ 14.6805556
 $A_s(dia.)$ 0

Static Pressure, P_s
 Static 0.05
 P_s 29.95

Stack Gas Velocity, V_s , fpm
 C_p 0.04
 $sqrdelta p$ 1.02312627 0.6756879
 T_s 205.8
 V_s 66.72

Stack Gas Flowrate, Q_s , acfm
 A_s 14.6805556
 Q_s 58767.16 3526029.41

Stack Gas Flowrate @ STD. COND., Dry Basis, Q_{std}
 Q_{std} 37217.01 2233020.51

Velocity at the Nozzle, V_n
 Dia. of Nozzle 0.23 An 0.00028852
 Time of Run 70
 V_n 66.52

R isokinetic, R_i
 R_i 99.70

R isokinetic measured from intermediate values
 R_i 100.07843

Pollutant Mass Emission Rate, PMR , lbs./hr.
 M_n 0.0454
 PMR 4.3757

Grains per acf
 gr/acf 0.0109

Grains per decf 0.0137
 Concentration, $mg/decf$ 0.88967819

Particulate Emissions Rate
 F factor n/a
 lb/PMR_{10} *VALUE

Theoretical Heat Input (THI)
 Q_{std} 2233020.51
 E_d n/a

THI (MMBtu/hr) *VALUE

SOURCE TESTED: Volcan Ashwell Corp.
 TESTER COMPANY: American
 DATE OF TEST: 5/6/98
 RUN NO.: 3

Molecular Weight of Stack Gas, Dry Basis, Mw
 SC02 = 43
 SO2 = 148
 Mw = 29.312

Volume of Water Vapor Collected @ STD. COND., Vwv
 Vw = 262
 Vw(std) = 12.37448

Volume of Dry Gas Collected @ STD. COND., Vm(std)
 Tm = 106.371429 566.371429 121.7308
 Vm = 50.4 100.7308
 Pb = 29.95
 delta H = 1.96265714
 gamma = 0.984
 Vm(std) = 46.67 RUN 3

Moisture Content of Stack Gas, Bws
 Bws = 20.698
 1-Bws = 79.118

Molecular Weight of Stack Gas, Ms
 Ms = 28.97

Area of Stack (enter diameter in inches), As
 L = 73.3
 W = 28
 diameter = 0
 No. of Stacks = 1
 As(L*W) = 146805556
 As(dia.) = 0

Static Pressure, Ps
 Static = 0.05
 Ps = 29.95

Stack Gas Velocity, Vs, fps
 Cp = 0.84
 sqrt(delta p) = 0.9811072 0.6756875
 Ts = 209.4
 Vs = 54.15

Stack Gas Flowrate, Qs, acfm
 As = 146805556
 Qs = 56507.02 3390420.1

Stack Gas Flowrate @ STD. COND., Dry Basis, Qstd
 Qstd = 35424.97 2125498.36

Velocity at the Nozzle, Vn
 Dia. of Nozzle = 0.23 An 0.00028852
 Time of Run = 70
 Vn = 61.45

S Isokinetic, Si
 Si = 95.79

S Isokinetic measured from intermediate values
 Si = 96.1571

Pollutant Mass Emission Rate, PMR, lbs./hr.
 Mn = 0.0386
 PMR = 3.8721

Grains per acf
 gr/acf = 0.0101

Grains per dscf
 gr/dscf = 0.0128
 Concentration, mg/dscf
 mg/dscf = 0.82710739

Particulate Emission Rate
 Factor = n/a
 lb/PMR10 = 0.0001

Theoretical Heat Input (THI)
 Qstd = 2125498.36
 Fgs = n/a

THI (PMR10/hci) = 0.0001

REVIEW SUMMARY

SOURCE TESTED: Valley Industrial Corp.
 TESTING COMPANY: Bannock
 DATE OF TEST: 8/2/88

Average Stack Gas Velocity, V_a , fpm	65.666
Average Stack Gas Flowrate, Q_a , acfm	57840.619
Average Stack Gas Flowrate @ STD. COND., Dry Basis, Qstd.	36930.511
Average R isokinetic, R_i	98.061585
Average R isokinetic measured from intermediate values	98.438027
Average Pollutant Mass Emission Rate, PMR, lbs./hr.	4.2777693
Average Grains per acf	0.0106987
Average Grains per dscf	0.013508
Average Concentration, mg/dscf	0.875439
Average Particulate Emissions Rate, lb/MMBtu	*VALUE!
Average Stack Temperature, $^{\circ}F$	206.23333
Average Theoretical Heat Input, MMBtu/hr	*VALUE!

RAMCON

ENVIRONMENTAL CORPORATION

June 20, 1988

Mr. Paul Prottengeier
Valley Asphalt
11641 Mosteller Road
Cincinnati, OH 45241

Re: Particulate Emissions Test - Camp Dennison, Ohio

Dear Mr. Prottengeier:

Enclosed you will find four copies of our report on the particulate emissions test we conducted on your plant. Based on our test results, your plant does pass both EPA New Source Performance Standards and those set by the State of Ohio. The average grain loading of the three test runs was below the allowable emissions standard set by EPA and the State of Ohio. Therefore, your plant is operating in compliance with State and Federal Standards.

You will want to sign the report covers and send two copies to:

Mr. Lee Gruber
Southwest Ohio A.P.C.
P. O. Box 972
Dayton, OH 45422

You will need to keep one copy of the report at the plant.

We certainly have enjoyed working with you and we look forward to serving you again in the future.

Sincerely,



G. Sumner Buck, III
President

GSBIII:kr

Enclosures

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I. INTRODUCTION

On June 6, 1988, personnel from RAMCON Environmental Corporation (REC) conducted a source emissions test for particulate emissions compliance at Valley Asphalt's Barber-Greene drum mix asphalt plant #17 located in Camp Dennison, Ohio. RAMCON personnel conducting the test were Allen Turner, Team Leader, and Murphy O'Neal. Kim Rea was responsible for the laboratory analysis including taring the beakers and filters and recording final data in the laboratory record books. Custody of the samples was limited to Mr. Turner and Ms. Rea.

The purpose of the test was to determine if the rate of particulate emissions from the plant's baghouse and the total contaminants by weight (grain loading) is below the allowable N.S.P.S. limits set by EPA and the State of Ohio.

II. TEST RESULTS

Table I summarizes the test results. The grain loading limitation for EPA is specified in 39 FR 9314, March 8, 1974, 60.92 Standards for Particulate Matter (1), as amended. The allowable N.S.P.S. particulate emissions for EPA and the State of Ohio is .04 gr/dscf.

TABLE I
SUMMARY OF TEST RESULTS
June 6, 1988

<u>Test Run</u>	<u>Time</u>	<u>Grain Loading</u>	<u>Isokinetic Variation</u>	<u>Actual Emissions</u>
1	07:51 to 09:04	0.0140 gr/DSCF	94.7%	4.8 lbs/hr
2	10:08 to 11:21	0.0140 gr/DSCF	99.6%	4.4 lbs/hr
3	11:50 to 13:03	0.0126 gr/DSCF	98.4%	3.8 lbs/hr
Average:		0.0135 gr/DSCF		4.3 lbs/hr

On the basis of these test results, the average grain loading of the three test runs was below the .04 gr/DSCF emissions limitation set by US EPA and the State of Ohio. Therefore, the plant is operating in compliance with State and Federal Standards.

III. TEST PROCEDURES

A. Method Used: The source sampling was conducted in accordance with requirements of the U.S. Environmental Protection Agency as set forth in 39 FR 9314, March 8, 1974, 60.93, as amended.

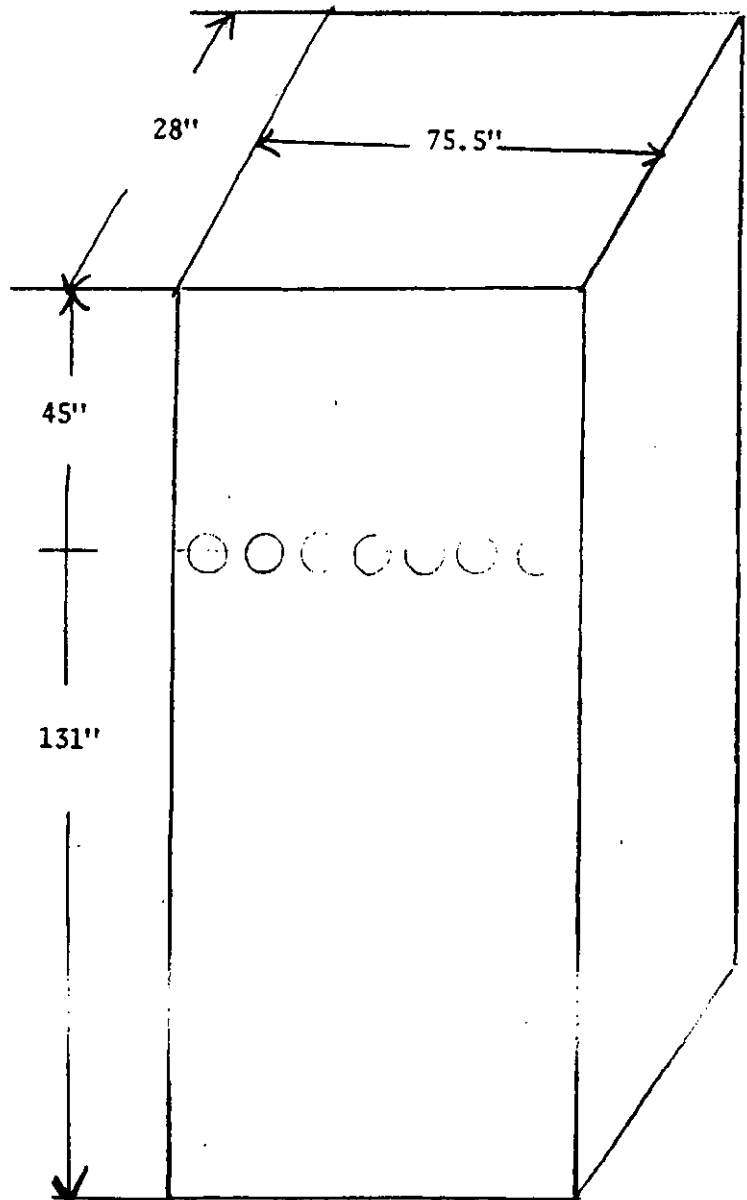
B. Problems Encountered: No problems were encountered that affected testing.

(3)

C. Sampling Site: The emissions test was conducted after a baghouse on a rectangular stack measuring 28" x 75.5" with an equivalent diameter of 40.9". Seven sampling ports were placed 45" down (1.1 diameters upstream) from the top of the stack and 131" up (3.2 diameters downstream) from the last flow disturbance. Thirty-five points were sampled, seven through each port for two minutes each for a total test time of seventy minutes per test run.

<u>Points on a Diameter</u>	<u>Probe Mark</u>
1	*7.8"
2	13.4"
3	19.0"
4	24.6"
5	30.2"

*Measurements include a
5.0" standoff.



IV. THE SOURCE

IV. THE SOURCE

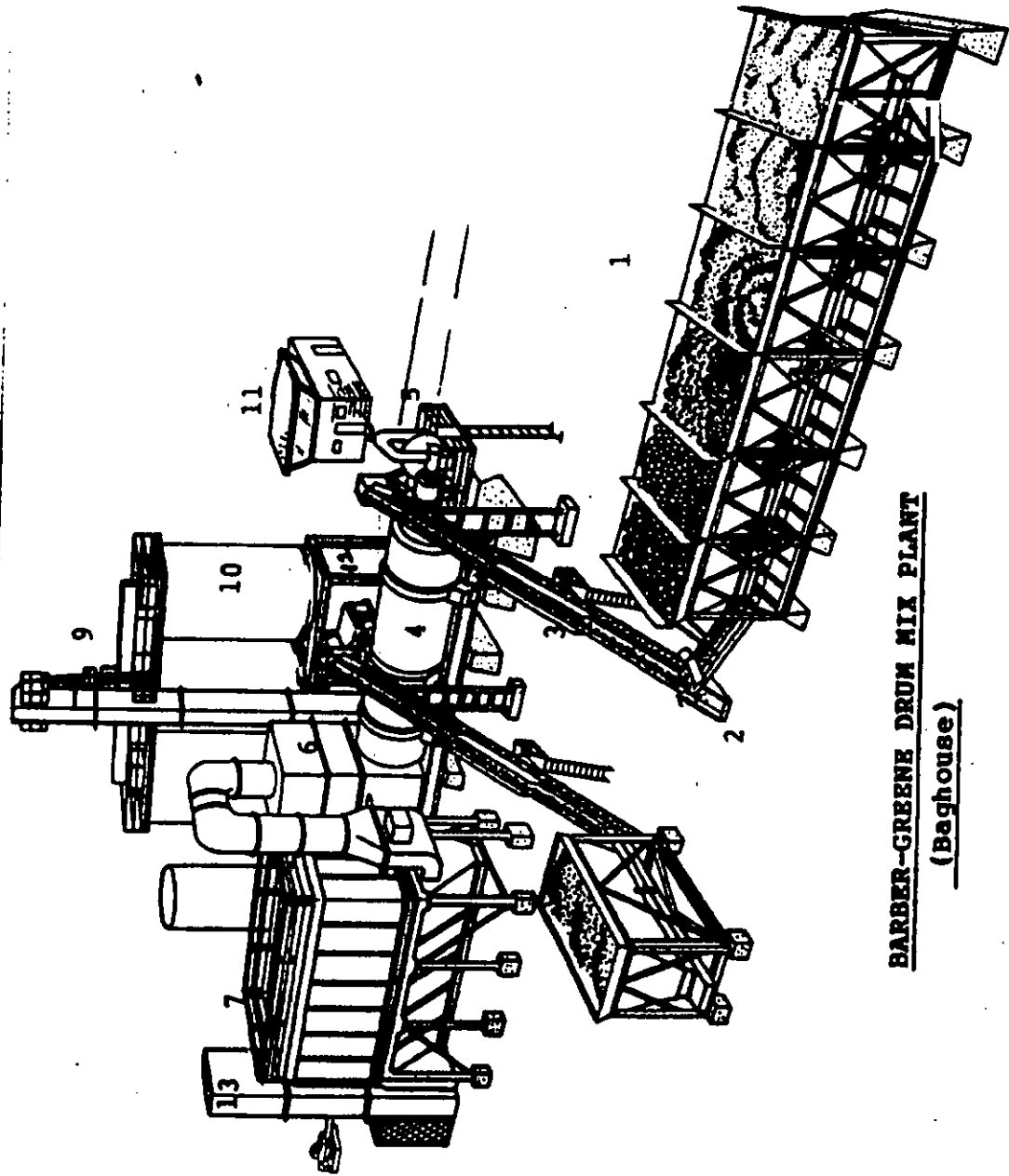
Valley Asphalt Corp. employs a Barber-Greene drum mix asphalt plant which is used to manufacture hot mix asphalt for road pavement. The process consists of blending prescribed portions of cold feed materials (sand, gravel, screenings, chips, etc.) uniformly and adding sufficient hot asphalt oil to bind the mixture together. After the hot asphalt mix is manufactured at the plant, it is transported to the location where it is to be applied. The hot asphalt mix is spread evenly over the surface with a paver and then compacted with a heavy roller to produce the final product.

The following is a general description of the plant's manufacturing process: The cold feed materials (aggregate) are dumped into four separate bins which in turn feed a common continuous conveyor. The aggregate is dispensed from the bins in accordance with the desired formulation onto the cold feed system conveyor to an inclined weigh conveyor then to a rotating drum for continuous mixing and drying at approximately 300°F. The required amount of hot asphalt oil is then injected onto and mixed into the dried aggregate. The now newly formed hot asphalt mix is pulled to the top of a storage silo by conveyor. The hot asphalt mix is then discharged from the storage silo through a slide gate into waiting dump trucks, which transport the material to a final destination for spreading. The rated capacity of the plant will vary with each aggregate mix and moisture content with a 5% surface moisture removal.

The drum mixer uses a burner fired with #2 burner fuel to heat air to dry the aggregate, and the motion of the rotating drum to blend the aggregate and hot asphalt oil thoroughly. The air is drawn into the system via an exhaust fan. After passing through the burner and the mixing drum, the exhaust gasses pass through a baghouse. The baghouse is manufactured by Barber-Greene. The exhaust gasses are drawn through the baghouse and discharged to the atmosphere through a stack. The design pressure drop across the tube-sheet is 1 - 6 inches of water. The particulate matter, which is removed by the baghouse, is reinjected into the drum mixer.

(5)

1. **Aggregate bins:** Virgin aggregate is fed individually into each of four bins by type. It is metered onto a conveyor belt running under the bins to a shaker screen. The proportion of each aggregate type is determined by the job mix formula and pre-set to be metered out to meet these specifications.
2. **Preliminary oversize screen:** The aggregate is fed through a shaker screen where oversize rocks and foreign material is screened out of the mix.
3. **Weigh conveyor belt:** The aggregate is conveyed to the rotary drum dryer on a conveyor belt which weighs the material. The production rate is determined by this weight reading.
4. **Rotary drum dryer/mixer:** The aggregate is fed into the rotary drum dryer where it is tumbled by flinging into a veil in front of a flame which drives off the moisture. Further mixing is also accomplished in this drum. Hot liquid asphalt is injected into the drum where it is blended with the aggregate. Mixing continues in a coater located at the end of the drum.
5. **Burner:** The fuel fired burner is used to provide the flame which dries the aggregate.
6. **Knock off baffling:** A baffling plate is inserted in the "dirty" side plenum as a knock out for heavy particles in the air stream. These particles fall to the bottom of the baghouse.
7. **Baghouse:** The hot gases are pulled through the bags into the clean air plenum. The solid particulate matter is trapped on the dust coat buildup on the bags. A bag cleaning cycle consisting of jet burst of air from the inside (or clean air side) of the bags sends a large bubble of air down the inside of the bags shaking loose buildup on the bag surface. This particulate matter is collected at the bottom of the baghouse and reinjected into the drum mixer where it is used as part of the finished product.
8. **Liquid asphalt storage:** The liquid asphalt is stored in this heated tank until it is needed in the mixer. The amount of asphalt content and its temperature are pre-set for each different type job.
9. **Conveyor to surge/storage bin:** The finished product of aggregate mixed with liquid asphalt is conveyed to a surge bin.
10. **Surge/Storage bin:** The asphaltic cement is dumped into this surge bin and metered out to dump trucks which pull underneath a slide gate at the bottom of the bin.
11. **Control/operators house:** The entire plant operation is controlled from this operator's house.
12. **Truck loading scale:** As the trucks receive the asphalt from the storage/surge bin they are weighed on the loading scale which tells the plant operator the amount of asphalt that is being trucked on each individual load.
13. **Fuel Storage**
14. **Stack**



BARBER-GREENE DRUM MIX PLANT
(Baghouse)

(7)

DATA SUMMARY

Plant

1. Manufacturer of plant BARBER GREENE
2. Designed maximum operating capacity 352 TPH @ 3 % moisture.
3. Actual operation rate _____ TPH @ 3 % moisture.
4. Startup date _____.
5. Type of fuel used in dryer #2 BURNER FUEL.
6. Quantity of fuel consumption _____.

Aggregate

7. Name/type of mix 846 TYPE II
8. Percent asphalt in mix 4.7 %.
9. Temperature of asphalt 290°.
10. Sieve/Screening analysis: % Passing;
1" 100 3/8" 62 #50 _____
3/4" 94 #4 47 #100 _____
1/2" 70 #8 41 #200 _____

Baghouse

11. Manufacturer _____.
12. No. of bags 800. Type of bags 14 oz. BETA NORMA
13. Air to cloth ratio _____. Designed ACFM 54,000.
14. Square feet of bags _____.
15. Type of cleaning; pulse jet _____, reverse air _____,
plenum pulse YES, other _____.
16. Cleaning cycle time _____.
17. Interval between cleaning cycle _____.
18. Pressure drop across baghouse _____ psi.
19. Pulse pressure on cleaning cycle _____ psi.

COMPANY NAME _____ DATE _____

COMPANY REPRESENTATIVE _____

(8)
PLANT DATA

COMPANY NAME Valley Asphalt Corp.
 COMPANY REP. FRED BRUNNER DATE 6/6/88 PHONE # 513-831-5676
 DATA SOURCE _____
 PLANT LOCATION Camp Dennison Ohio
 PLANT MFG. Bar Bar Drum PLANT MODEL # DM66X365 PLANT TYPE DRUM MIXER
 MIX SPECIFICATION # _____ OIL SPECIFICATION # 2 BURNER FUEL

Time 24 Hour	Fuel Oil _____ Nat. Gas _____ Propane _____ Coal _____	Burner Setting	Aggregate TPH	Recycle TPH	Liquid Asphalt TPH	Mix Temp. OF	Venturi Baghouse Pressure Drop Inches Water
7:45 am	#2 Fuel	50%	286	0	14.8	300	4 1/2
8:00 am	"	45%	281	"	14.4	290	5 1/2
8:15 am	"	45%	280	"	14.3	290	6
8:30 am	"	45%	283	"	14.4	285	5 1/2
8:45 am	"	40%	284	"	14.4	290	5 1/2
9:00 am	"	45%	283	"	14.4	285	6
10:05	"	50%	281	"	14.3	300	5 1/2
10:20	"	45%	284	"	14.8	290	6
10:35	"	55%	301	"	15.5	285	6
10:50	"	55%	302	"	15.5	290	6 1/2
11:05	"	50%	310	"	15.6	290	6 1/2
11:20	"	55%	307	"	15.5	290	6
11:43	"	52%	315	"	16.1	290	6 1/2
12:02	"	50%	310	"	16.0	290	6 1/2
12:17	"	52%	315	"	16.1	290	6
12:32	"	52%	310	"	15.9	295	6 1/2
12:47	"	50%	308	"	15.9	300	6 1/2
1:02	"	52%	320	"	16.3	300	6 1/2
			5360				
			298		15.4		

V. EQUIPMENT USED

V. EQUIPMENT USED

Equipment used on conducting the particulate emissions test was:

- A. The Lear Siegler PM-100 stack sampler with appropriate auxillary equipment and glassware. The train was set up according to the schematic on the nex page.
- B. An Airguide Instruments Model 211-B (uncorrected) aneroid barometer was used to check the barometric pressure.
- C. Weston dial thermometers are used to check meter temperatures. An Analogic Model 2572 Digital Thermocouple is used for stack temperatures.
- D. A Hays 621 Analyzer was used to measure the oxygen, carbon dioxide and carbon monoxide content of the stack gases. For non-combustion sources, A Bacharach Instrument Company Fyrite is used for the gas analysis.
- E. Filters are mady by Schleicher and Schuell and are type 1-HV with a porosity of .03 microns.
- F. The acetone is reagent grade or ACS grade with a residue of $\leq .001$.

VI. LABORATORY PROCEDURES & RESULTS

LABORATORY PROCEDURES FOR PARTICULATE SAMPLING**I. Field Preparation**

- A. **FILTERS:** Fiberglass 4" sampling filters are prepared as follows:

Filters are removed from their box and numbered on the back side with a felt pen. The numbering system is continuous from job to job. The filters are placed in a dessicator to dry for at least 24 hours. Clean plastic petri dishes, also numbered, top and bottom, are placed in the dessicator with the filters. After dessication, the filters are removed one at a time and weighed on the Sartorius analytical balance, then placed in the correspondingly numbered petri dish. Weights are then recorded in the lab record book. Three filters are used for each complete particulate source emissions test and there should be several extra filters included as spares.

- B. **SILICA GEL:** Silica Gel used for the test is prepared as follows:

Approximately 200 g of silica gel is placed in a wide mouth "Mason" type jar and dried in an oven (175°C for two hours). The open jars are removed and placed in a dessicator until cool (2 hours) and then tightly sealed. The jars are then numbered and weighed on the triple beam balance to the closest tenth of a gram, and this weight is recorded for each sealed jar. The number of silica gel jars used is the same as the number of filters. Silica gel should be indicating type, 6-16 mesh.

II. Post-Testing Lab Analysis

- A. **FILTERS:** The filters are returned to the lab in their sealed petri dishes. In the lab, the dishes are opened and placed into a dessicator for at least 24 hours. Then, the filters are weighed continuously every 6 hours until a constant weight is achieved. All data is recorded on the laboratory forms that will be bound in the test report.

Alternately, the test team may opt to oven dry the filters at 220°F for two to three hours, weigh the sample, and use this weight as a final weight.

- B. **SILICA GEL:** The silica gel used in the stack test is returned to the appropriate mason jar and sealed for transport to the laboratory where it is reweighed to a constant weight on a triple-beam balance to the nearest tenth of a gram.

- C. **PROBE RINSINGS:** In all tests, where a probe washout analysis is necessary, this is accomplished in accordance with procedures specified in "EPA Reference Method 5". These samples are returned in sealed mason jars to the laboratory for analysis. The front half of the filter holder is washed in accordance with the same procedures and included with the probe wash. Reagent or ACS grade acetone is used as the solvent. The backhalf of the filter holder is washed with deionized water into the impinger catch for appropriate analysis.
- D. **IMPINGER CATCH:** In some testing cases, the liquid collected in the impingers must be analyzed for solid content. This involves a similar procedure to the probe wash solids determination, except that the liquid is deionized water.
- E. **ACETONE:** Conduct a blank analysis of acetone from the one gallon glass container. This acetone will be used in the field for rinsing the probe, nozzle, and top half of the filter holder. Performing such a blank analysis prior to testing will insure that the quality of the acetone to be used will not exceed the .001% residual purity standard.

SPECIAL NOTE

When sampling sources high in moisture content, (such as asphalt plants) the filter paper sometimes sticks to the filter holder. When removing the filter, it may tear. In order to maintain control of any small pieces of filter paper which may be easily lost, they are washed with acetone into the probe washing. This makes the filter weight light (sometimes negative) and the probe wash correspondingly heavier. The net weight is the same and no particulate is lost. This laboratory procedure is taught by EPA in the Quality Assurance for Source Emissions Workshop at Research Triangle Park and is approved by EPA.

WEIGHING PROCEDURE - SARTORIUS ANALYTICAL BALANCE

The Sartorius balance is accurate to 0.1 mg and has a maximum capacity of 200 grams. The balance precision (standard deviation) is 0.05 mg. Before weighing an item, the balance should first be zeroed. This step should be taken before every series of weighings. To do this, the balance should have all weight adjustments at "zero" position. The beam arrest lever (on the lower left hand side toward the rear of the balance) is then slowly pressed downward to full release position. The lighted vernier scale on the front of the cabinet should align the "zero" with the mark on the cabinet. If it is not so aligned, the adjustment knob on the right hand side (near the rear of the cabinet) should be turned carefully until the marks align. Now return the beam arrest to horizontal arrest position. The balance is now "zeroed".

To weigh an item, it is first placed on the pan. And the sliding doors are closed to avoid air current disturbance. The weight adjustment knob on the right hand side must be at "zero". The beam arrest is then slowly turned upward. The lighted scale at the front of the cabinet will now indicate the weight of the item in grams. If the scale goes past the divided area, the item then exceeds 100 g weight (about 3-1/2 ounces) and it is necessary to arrest the balance (beam arrest lever) and move the lever for 100 g weight away from you. It is located on the left hand side of the cabinet near the front, and is the knob closest to the side of the cabinet. The balance will not weigh items greater than 200 grams in mass, and trying to do this might harm the balance. Remember -- this is a delicate precision instrument.

After the beam is arrested, in either weight range, the procedure is the same. When the weight of the item in grams is found, "dial in" that amount with the two knobs on the left hand side (near the 100 g lever) color coded yellow and green. As you dial the weight, the digits will appear on the front of the cabinet. When the proper amount is dialed, carefully move the arrest lever down with a slow, steady turn of the wrist. The lighted dial will appear, and the right hand side knob (front of cabinet) is turned to align the mark with the lower of the two lighted scale divisions which the mark appears between. When these marks are aligned, the two lighted digits along with the two indicated on the right hand window on the cabinet front are the fractional weight in grams (the decimal would appear before the lighted digits) and the whole number of grams weight is the amount "dialed in" on the left.

In general, be sure that the beam is in "arrest" position before placing weight on or taking weight off of the pan. Don't "dial in" weight unless the beam is arrested. The balance is sensitive to even a hand on the table near the balance, so be careful and painstaking in every movement while weighing.

SAMPLE ANALYTICAL DATA FORM

Plant Location Valley Asphalt #17 Relative humidity in lab 49.45%Sample Location hot mix asphalt plant Density of Acetone (ρ_a) .7853 mg/mlBlank volume (V_a) 200 mlDate/Time wt. blank 6/13/88Gross wt. 93.1045 mgDate/Time wt. blank 6/14/88Gross wt. 93.1044 mgAve. Gross wt. 93.1045 mgTare wt. 93.1039 mgWeight of blank (m_{ab}) .0006 mgAcetone blank residue concentration (C_a) (C_a) = (m_{ab}) / (V_a) (ρ_a) = (.0006 mg) / (200 ml) (.7853 mg/ml) = (.00036 mg/g)Weight of residue in acetone wash: $W_a = C_a V_{aw}$ $\rho_a = (.00036)(200)(.7853) = (.0006)$ Acetone rinse volume (V_{aw}) mlDate/Time of wt 6/13/88 9:10 AM gross wt gDate/Time of wt 6/14/88 3:45 PM gross wt g

Average Gross wt g

Tare wt g

Less acetone blank wt (W_a) gWt of particulate in acetone rinse (m_a) g

Run # 1	Run # 2	Run # 3
<u>200</u>	<u>200</u>	<u>200</u>
<u>159.6320</u>	<u>129.2057</u>	<u>154.4571</u>
<u>159.6315</u>	<u>129.2053</u>	<u>154.4570</u>
<u>159.6318</u>	<u>129.2055</u>	<u>154.4571</u>
<u>159.6102</u>	<u>129.1810</u>	<u>154.4426</u>
<u>.0006</u>	<u>.0006</u>	<u>.0006</u>
<u>.0210</u>	<u>.0239</u>	<u>.0139</u>

Filter Numbers #

Date/Time of wt 6/13/88 9:10 AM gross wt gDate/Time of wt 6/14/88 3:45 PM gross wt g

Average Gross wt g

Tare wt g

SG-2512	SG-2508	SG-2509
<u>.5528</u>	<u>.5484</u>	<u>.5509</u>
<u>.5530</u>	<u>.5481</u>	<u>.5506</u>
<u>.5529</u>	<u>.5483</u>	<u>.5507</u>
<u>.5268</u>	<u>.5268</u>	<u>.5260</u>

Weight of particulate on filters(s) (m_f) g

Weight of particulate in acetone rinse g

Total weight of particulate (m_T) g

<u>.0261</u>	<u>.0215</u>	<u>.0247</u>
<u>.0210</u>	<u>.0239</u>	<u>.0139</u>
<u>.0471</u>	<u>.0454</u>	<u>.0386</u>

Note: In no case should a blank residue greater than 0.01

mg/g (or 0.001% of the blank weight) be subtracted from

the sample weight.

Remarks

Signature of analyst

Kim Rea

Signature of reviewer

Jack

Back-Half

Plant Location Valley #17

Relative humidity in lab _____ %

Sample Location _____

Density of Acetone (ρ_a) .78 mg/mlBlank volume (V_a) _____ ml

Date/Time wt. blank _____

Gross wt. _____ mg

Date/Time wt. blank _____

Gross wt. _____ mg

Ave. Gross wt. _____ mg

Tare wt. _____ mg

Weight of blank (m_{ab}) _____ mgAcetone blank residue concentration (C_a) (C_a) = (M_{ab}) / (V_a) (ρ_a) = (_____) mg/gWeight of residue in acetone wash: $W_a = C_a V_{aw}$ $\rho_a = (\quad) (\quad) (\quad) = (\quad)$

	Run # 1	Run # 2	Run # 3
Acetone rinse volume (V_{aw}) ml	<u>230</u>	<u>265</u>	<u>250</u>
Date/Time of wt _____ Gross wt g	<u>138.5378</u>	<u>134.2465</u>	<u>159.6242</u>
Date/Time of wt _____ Gross wt g	<u>138.5377</u>	<u>134.2460</u>	<u>159.6240</u>
Average Gross wt g	<u>138.5378</u>	<u>134.2463</u>	<u>159.6241</u>
<u>Water</u> Tare wt g	<u>138.5377</u>	<u>134.2029</u>	<u>159.5984</u>
Less acetone blank wt (W_a) g	<u>—</u>	<u>—</u>	<u>—</u>
Wt of particulate in acetone rinse (m_a) g	<u>.0181</u>	<u>.0434</u>	<u>.0249</u>

Filter Numbers	#		
Date/Time of wt _____ Gross wt g			
Date/Time of wt _____ Gross wt g			
Average Gross wt g			
Tare wt g			

Weight of particulate on filters(s) (m_f) g			
Weight of particulate in <u>Water</u> acetone rinse g	<u>.0181</u>	<u>.0434</u>	<u>.0249</u>
Total weight of particulate (m_p) g			

Note: In no case should a blank residue greater than 0.01

mg/g (or 0.001% of the blank weight) be subtracted from the sample weight.

Remarks _____

Signature of analyst

Kim Pea

Signature of reviewer _____

VII. CALCULATIONS

VALLEY ASPHALT #17
CAMP DENNISON, OHIO
6/6/88

(15)

SUMMARY OF TEST DATA

RUN #1 RUN #2 RUN #3

SAMPLING TRAIN DATA

	start	07:51	10:08	11:50
	finish	09:04	11:21	13:03
1. Sampling time, minutes	θ	70.0	70.0	70.0
2. Sampling nozzle diameter, in.	D_n	.2300	.2300	.2300
3. Sampling nozzle cross-sect. area, ft^2	A_n	.000289	.000289	.000289
4. Isokinetic variation	I	94.7	99.6	98.4
5. Sample gas volume - meter cond., cf.	V_m	54.458	54.261	52.400
6. Average meter temperature, $^{\circ}R$	T_m	543	561	571
7. Avg. orifice pressure drop, in. H_2O	dH	2.38	2.16	1.96
8. Total particulate collected, mg.	M_n	47.10	45.40	38.60

VELOCITY TRAVERSE DATA

9. Stack area, ft^2	A	14.70	14.70	14.70
10. Absolute stack gas pressure, in. Hg.	P_s	29.95	29.95	29.95
11. Barometric pressure, in. Hg.	P_{bar}	29.95	29.95	29.95
12. Avg. absolute stack temperature, R°	T_s	664	666	669
13. Average $-\sqrt{vel. head}$, ($C_p = .83$)	$-\sqrt{dP}$	1.08	1.02	0.98
14. Average stack gas velocity, ft./sec.	V_s	69.15	65.79	63.30

STACK MOISTURE CONTENT

15. Total water collected by train, ml.	V_{ic}	243.00	280.00	262.00
16. Moisture in stack gas, %	B_{ws}	18.04	20.91	20.75

EMISSIONS DATA

17. Stack gas flow rate, dscf/hr. (000's)	Q_{sd}	2387	2185	2097
18. Stack gas flow rate, cfm	$acfm$	60990	58027	55831
19. Particulate concentration, gr/dscf	C_s	0.0140	0.0140	0.0126
20. Particulate concentration, lb/hr	E	4.77	4.37	3.78
21. Particulate concentration, lb/mBtu	E'	0.00000	0.00000	0.00000

ORSAT DATA

22. Percent CO_2 by volume	CO_2	4.00	4.00	4.50
23. Percent O_2 by volume	O_2	15.50	15.50	14.80
24. Percent CO by volume	CO	.00	.00	.00
25. Percent N_2 by volume	N_2	80.50	80.50	80.70

Format: summaryR3

NAME: VALLEY ASPHALT #17
 LOCATION: CAMP DENNISON, OHIO (16)
 TEST DATE: 6\6\88

Dry Gas Volume

$$V_{m(std)} = V_m \left[\frac{T_{(std)}}{T_m} \right] \left[\frac{P_{bar} + \frac{dH}{13.6}}{P_{(std)}} \right] = 17.64 \frac{^{\circ}R}{in.Hg} Y V_m \left[\frac{P_{bar} + \frac{dH}{13.6}}{T_m} \right]$$

Where:

- $V_{m(std)}$ = Dry Gas Volume through meter at standard conditions, cu. ft.
 V_m = Dry Gas Volume measured by meter, cu. ft.
 P_{bar} = Barometric pressure at orifice meter, in. Hg.
 P_{std} = Standard absolute pressure, (29.92 in. Hg.).
 T_m = Absolute temperature at meter $^{\circ}R$.
 T_{std} = Standard absolute temperature (528 $^{\circ}R$).
 dH = Average pressure drop across orifice meter, in. H₂O.
 Y = Dry gas meter calibration factor.
 13.6 = Inches water per inches Hg.

RUN 1:

$$V_{m(std)} = (17.64) (.972) (54.458) \left[\frac{(29.95) + \frac{2.38}{13.6}}{543} \right] = 51.803 \text{ dscf}$$

RUN 2:

$$V_{m(std)} = (17.64) (.972) (54.261) \left[\frac{(29.95) + \frac{2.16}{13.6}}{561} \right] = 49.933 \text{ dscf}$$

RUN 3:

$$V_{m(std)} = (17.64) (.972) (52.400) \left[\frac{(29.95) + \frac{1.96}{13.6}}{571} \right] = 47.352 \text{ dscf}$$

NAME: VALLEY ASPHALT #17
LOCATION: CAMP DENNISON, OHIO
TEST DATE: 6/6/88

(17)
Total Contaminants by Weight: GRAIN LOADING

Particulate concentration C'_s gr./dscf.

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{M_n}{V_{m(\text{std})}} \right]$$

Where:

C'_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, gr./dscf.

M_n = Total amount of particulate matter collected, mg.

$V_{m(\text{std})}$ = Dry gas volume through meter at standard conditions, cu. ft.

Run 1:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{47.10}{51.803} \right] = 0.0140 \text{ gr./dscf.}$$

Run 2:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{45.40}{49.933} \right] = 0.0140 \text{ gr./dscf.}$$

Run 3:

$$C'_s = \left[0.0154 \frac{\text{gr}}{\text{mg}} \right] \left[\frac{38.60}{47.352} \right] = 0.0126 \text{ gr./dscf.}$$

NAME: VALLEY ASPHALT #17
LOCATION: CAMP DENNISON, OHIO (18)
TEST DATE: 6\6\88

Dry Molecular Weight

$$M_d = 0.44(\%CO_2) + 0.32(\%O_2) + 0.28(\%CO + \%N_2)$$

Where:

- M_d = Dry molecular weight, lb./lb.-mole.
 $\%CO_2$ = Percent carbon dioxide by volume (dry basis).
 $\%O_2$ = Percent oxygen by volume (dry basis).
 $\%N_2$ = Percent nitrogen by volume (dry basis).
 $\%CO$ = Percent carbon monoxide by volume (dry basis).
0.264 = Ratio of O_2 to N_2 in air, v/v.
0.28 = Molecular weight of N_2 or CO, divided by 100.
0.32 = Molecular weight of O_2 divided by 100.
0.44 = Molecular weight of CO_2 divided by 100.

Run 1:

$$M_d = 0.44(4.00\%) + 0.32(15.50\%) + 0.28(.00\% + 80.50\%) = 29.26 \frac{\text{lb}}{\text{lb-mole}}$$

Run 2:

$$M_d = 0.44(4.00\%) + 0.32(15.50\%) + 0.28(.00\% + 80.50\%) = 29.26 \frac{\text{lb}}{\text{lb-mole}}$$

Run 3:

$$M_d = 0.44(4.50\%) + 0.32(14.80\%) + 0.28(.00\% + 80.70\%) = 29.31 \frac{\text{lb}}{\text{lb-mole}}$$

Format: mdR3

NAME: VALLEY ASPHALT #17
 LOCATION: CAMP DENNISON, OHIO
 TEST DATE: 6\6\88

(19)

Water Vapor Condensed

$$V_{wc_std} = \left[V_f - V_i \right] \left[\frac{P_w R T_{(std)}}{M_w P_{(std)}} \right] = 0.04707 \left[V_f - V_i \right]$$

$$V_{wsg_std} = \left[W_f - W_i \right] \left[\frac{R T_{(std)}}{M_w P_{(std)}} \right] = 0.04715 \left[W_f - W_i \right]$$

Where:

0.04707 = Conversion factor, ft.³/ml.

0.04715 = Conversion factor, ft.³/g.

V_{wc_std} = Volume of water vapor condensed (standard conditions), scf.

V_{wsg_std} = Volume of water vapor collected in silica gel (standard conditions), ml.

$V_f - V_i$ = Final volume of impinger contents less initial volume, ml.

$W_f - W_i$ = Final weight of silica gel less initial weight, g.

P_w = Density of water, 0.002201 lb/ml.

R = Ideal gas constant, 21.85 in.Hg. (cu.ft./lb.-mole) (°R).

M_w = Molecular weight of water vapor, 18.0 lb/lb-mole.

T_{std} = Absolute temperature at standard conditions, 528°R.

P_{std} = Absolute pressure at standard conditions, 29.92 inches Hg.

Run 1:

$$V_{wc(std)} = (0.04707) (230.0) = 10.8 \text{ cu.ft}$$

$$V_{wsg(std)} = (0.04715) (13.0) = 0.6 \text{ cu.ft}$$

Run 2:

$$V_{wc(std)} = (0.04707) (265.0) = 12.5 \text{ cu.ft}$$

$$V_{wsg(std)} = (0.04715) (15.0) = 0.7 \text{ cu.ft}$$

Run 3:

$$V_{wc(std)} = (0.04707) (250.0) = 11.8 \text{ cu.ft}$$

$$V_{wsg(std)} = (0.04715) (12.0) = 0.6 \text{ cu.ft}$$

Format: vaporR3

NAME: VALLEY ASPHALT #17
LOCATION: CAMP DENNISON, OHIO
TEST DATE: 6/6/88

(20)

Moisture Content of Stack Gases

$$B_{ws} = \frac{V_{wc_{std}} + V_{wsg_{std}}}{V_{wc_{std}} + V_{wsg_{std}} + V_{m_{std}}} \times 100$$

Where:

B_{ws} = Proportion of water vapor, by volume, in the gas stream.

V_m = Dry gas volume measured by dry gas meter, (dcf).

$V_{wc_{std}}$ = Volume of water vapor condensed corrected to standard conditions (scf).

$V_{wsg_{std}}$ = Volume of water vapor collected in silica gel corrected to standard conditions (scf).

Run 1:

$$B_{ws} = \frac{10.8 + 0.6}{10.8 + 0.6 + 51.803} \times 100 = 18.04 \%$$

Run 2:

$$B_{ws} = \frac{12.5 + 0.7}{12.5 + 0.7 + 49.933} \times 100 = 20.91 \%$$

Run 3:

$$B_{ws} = \frac{11.8 + 0.6}{11.8 + 0.6 + 47.352} \times 100 = 20.75 \%$$

NAME: VALLEY ASPHALT #17
LOCATION: CAMP DENNISON, OHIO
TEST DATE: 6/6/88

(21)

Molecular Weight of Stack Gases

$$M_s = M_d (1 - B_{ws}) + 18 (B_{ws})$$

Where:

M_s = Molecular weight of stack gas, wet basis, (lb./lb.-mole).

M_d = Molecular weight of stack gas, dry basis, (lb./lb.-mole).

Run 1:

$$M_s = 29.26 (1 - 18.04) + 18 (18.04) = 27.23 \text{ (lb./lb.-mole)}$$

Run 2:

$$M_s = 29.26 (1 - 20.91) + 18 (20.91) = 26.91 \text{ (lb./lb.-mole)}$$

Run 3:

$$M_s = 29.31 (1 - 20.75) + 18 (20.75) = 26.96 \text{ (lb./lb.-mole)}$$

NAME: VALLEY ASPHALT #17
 LOCATION: CAMP DENNISON, OHIO
 TEST DATE: 6\6\88

(22)

Stack Gas Velocity

$$V_s = K_p C_p \left[-\sqrt{dP} \right] \text{ avg. } \sqrt{\frac{T_s(\text{avg.})}{P_s M_s}}$$

Where:

- V_s = Average velocity of gas stream in stack, ft./sec.
- K_p = 85.49 ft/sec $\left[\frac{(\text{g/g-mole}) - (\text{mm Hg})}{(^{\circ}\text{K}) (\text{mm H}_2\text{O})} \right]^{1/2}$
- C_p = Pitot tube coefficient, (dimensionless).
- dP = Velocity head of stack gas, in. H_2O .
- P_{bar} = Barometric pressure at measurement site, (in. Hg).
- P_g = Stack static pressure, (in. Hg).
- P_s = Absolute stack gas pressure, (in. Hg) = $P_{\text{bar}} + P_g$
- P_{std} = Standard absolute pressure, (29.92 in. Hg).
- t_s = Stack temperature, ($^{\circ}\text{f}$).
- T_s = Absolute stack temperature, ($^{\circ}\text{R}$). = $460 + t_s$.
- M_s = Molecular weight of stack gas, wet basis, (lb/lb-mole).

Run 1:

$$V = (85.49) (.83) (1.08) \sqrt{\frac{664}{(29.95)(27.23)}} = 69.15 \text{ ft/sec.}$$

Run 2:

$$V = (85.49) (.83) (1.02) \sqrt{\frac{666}{(29.95)(26.91)}} = 65.79 \text{ ft/sec.}$$

Run 3:

$$V = (85.49) (.83) (0.98) \sqrt{\frac{669}{(29.95)(26.96)}} = 63.30 \text{ ft/sec.}$$

NAME: VALLEY ASPHALT #17
 LOCATION: CAMP DENNISON, OHIO
 TEST DATE: 6/6/88

(23)

Stack Gas Flow Rate

$$Q_{sd} = 3600 \left[1 - B_{wc} \right] V_s A \left[\frac{T_{std}}{T_{stk}} \right] \left[\frac{P_s}{P_{std}} \right]$$

Where:

- Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, (dscf/hr).
- A = Cross sectional area of stack, (ft.²).
- 3600 = Conversion factor, (sec./hr.).
- t_s = Stack temperature, (°f).
- T_s = Absolute stack temperature, (°R).
- T_{std} = Standard absolute temperature, (528°R).
- P_{bar} = Barometric pressure at measurement site, (in.Hg.).
- P_g = Stack static pressure, (in.Hg.).
- P_s = Absolute stack gas pressure, (in.Hg.); = $P_{bar} + P_g$
- P_{std} = Standard absolute pressure, (29.92 in.Hg.).

Run 1:

$$Q_{sd} = 3600(1 - .1804)(69.15)(14.70) \left[\frac{528}{664} \right] \left[\frac{29.95}{29.92} \right] = 2387344.3 \frac{\text{dscf}}{\text{hr}}$$

Run 2:

$$Q_{sd} = 3600(1 - .2091)(65.79)(14.70) \left[\frac{528}{666} \right] \left[\frac{29.95}{29.92} \right] = 2185225.3 \frac{\text{dscf}}{\text{hr}}$$

Run 3:

$$Q_{sd} = 3600(1 - .2075)(63.30)(14.70) \left[\frac{528}{669} \right] \left[\frac{29.95}{29.92} \right] = 2097325.6 \frac{\text{dscf}}{\text{hr}}$$

NAME: VALLEY ASPHALT #17
LOCATION: CAMP DENNISON, OHIO (24)
TEST DATE: 6/6/88

Emissions Rate from Stack

$$E = \frac{(C_s) (Q_{sd})}{7000 \text{ gr./lb.}} = \text{lb. / hr.}$$

Where:

E = Emissions rate, lb./hr.

C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions (gr/dscf).

Q_{sd} = Dry volumetric stack gas flow rate corrected to standard conditions, (dscf/hr).

Run 1:

$$E = \frac{(0.0140) (2387344.3)}{7000} = 4.77 \text{ lb. / hr.}$$

Run 2:

$$E = \frac{(0.0140) (2185225.3)}{7000} = 4.37 \text{ lb. / hr.}$$

Run 3:

$$E = \frac{(0.0126) (2097325.6)}{7000} = 3.78 \text{ lb. / hr.}$$

NAME: VALLEY ASPHALT #17
 LOCATION: CAMP DENNISON, OHIO
 TEST DATE: 6/6/88

(25)

Isokinetic Variation

$$I = 100 T_s \left[\frac{0.002669 V_{ic} + \frac{(V_m / T_m) (P_{bar} + dH / 13.6)}{60 \theta V_s P_s A_n}}{1} \right]$$

Where:

- I = Percent isokinetic sampling.
- 100 = Conversion to percent.
- T_s = Absolute average stack gas temperature, $^{\circ}R$.
- 0.002669 = Conversion factor, $Hg - ft^3/ml - ^{\circ}R$.
- V_{ic} = Ttl vol of liquid collected in impingers and silica gel, ml.
- T_m = Absolute average dry gas meter temperature, $^{\circ}R$.
- P_{bar} = Barometric pressure at sampling site, (in. Hg).
- dH = Av pressure differential across the orifice meter, (in. H_2O).
- 13.6 = Specific gravity of mercury.
- 60 = Conversion seconds to minutes.
- θ = Total sampling time, minutes.
- V_s = Stack gas velocity, ft./sec.
- P_s = Absolute stack gas pressure, in. Hg.
- A_n = Cross sectional area of nozzle, ft^2 .

Run 1:

$$I = (100) (664) \left[\frac{(0.002669) (243.00) + \frac{54.458}{543} \left[29.95 + \frac{2.38}{13.6} \right]}{60 (70.0) (69.15) (29.95) (.000289)} \right] = 94.7\%$$

Run 2:

$$I = (100) (666) \left[\frac{(0.002669) (280.00) + \frac{54.261}{561} \left[29.95 + \frac{2.16}{13.6} \right]}{60 (70.0) (65.79) (29.95) (.000289)} \right] = 99.6\%$$

Run 3:

$$I = (100) (669) \left[\frac{(0.002669) (262.00) + \frac{52.400}{571} \left[29.95 + \frac{1.96}{13.6} \right]}{60 (70.0) (63.30) (29.95) (.000289)} \right] = 98.4\%$$

VIII. FIELD DATA

RAMCON ENVIRONMENTAL CORPORATION

Plant U.S. 200

Location San Bernardino

Operator 00000000

Date 6-6-88

Run No. 1

Sample Box No. 1

Meter Box No. 004

Meter H # 2

C Factor 472

Pitot Tube Coefficient Cp .929

Ambient Temperature 60

Barometric Pressure 29.95

Assumed Moisture, % 18

Probe Length, m(ft) 3

Nozzle Identification No. 0002885

Avg. Calibrated Nozzle Dia., (in.) 2 1/2 / 2 3/8

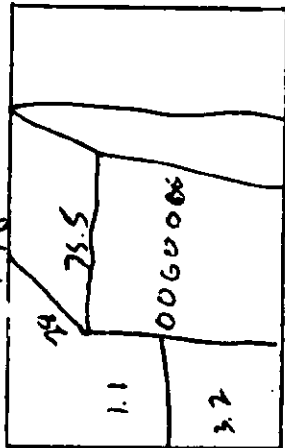
Probe Heater Setting 2.5

Leak Rate, m³/min. (cfm) 0.0308

Probe Liner Material 0.0308

Static Pressure, mm Hg (in. Hg) .02

Filter No. 56-2512



Schematic of Stack Cross Section

TRAV. PT NO.	SAMPLING TIME (h:min.)	VACUUM in. Hg	STACK TEMP (Ts) °F	VELOCITY HEAD (Pg) in H2O	PRESSURE DIFF. ORF. MTR in H2O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F		FILTER HOLDER TEMP °F	GAS TEMP LUG CONDENSER OR LAST IMPINGER °F
							Inlet	Outlet		
A-1	7:51	5	207	1.9	3.7	599.5	70	62	250	60
2	7:55	5	207	1.8	3.5	603.5	74	62	250	60
3	7:57	5	206	1.8	3.5	603.3	80	64	250	60
4	7:59	6	204	2.0	3.9	607.5	84	64	250	60
5	8:01	6	205	2.0	3.9	609.5	84	64	250	60
B-1	8:01-30	3	205	.60	1.2	610.8	84	64	250	50
2	8:05-30	3	203	.60	1.2	612.1	86	64	250	50
3	8:07-30	3	206	.60	1.2	613.1	86	64	250	50
4	8:09-30	3	208	.90	1.8	614.7	90	66	250	50
5	8:11-30	4	205	1.2	2.4	616.2	90	68	250	50
C-1	8:12-14	2	203	.40	.80	617.2	90	70	250	50
2	8:16	2	204	.40	.80	618.2	90	70	250	50
3	8:18	2	204	.40	.80	619.0	90	70	250	50

CO₂ = 4.0% CO₂ = 3.5% O₂ = 15.5%

RAMCON emissions test log sheet, cont. DATE 6-6-88 LOCATION Camp TEST NO. 1

TRAVERSE POINT	SAMPLING TIME (min)	VACUUM mm Hg (in. Hg)	STACK TEMP T_s (°F)	VELOCITY HEAD ΔP_s (in. H ₂ O)	ORIFICE DIFF. PRESSURE ΔH (in. H ₂ O)	GAS VOLUME V_m (ft. ³)	GAS SAMPLE TEMP. (°F)		SAMPLE BOX TEMP. (°F)	IMPIINGER TEMP (°F)
							in	out		
4	8:20	2	201	40	.80	620.0	90	70	260	55
5	8:22	2	200	50	.98	621.2	90	70	260	55
D-1	8:23-30 8:23-30	3	201	60	1.1	622.2	92	72	260	55
2	8:23-30	3	201	60	1.1	623.6	92	72	260	55
3	8:23-30	3	201	60	1.1	624.6	92	72	260	55
4	8:30-30	3	203	70	1.4	625.7	96	74	255	60
5	8:32-30	3	205	80	1.6	627.2	96	74	255	60
E-1	8:33-35 8:33-35	3	200	80	1.6	629.5	96	74	255	60
2	8:37	3	203	90	1.8	629.9	96	74	255	60
3	8:39	4	203	1.0	2.0	631.3	100	76	255	60
4	8:41	4	202	1.1	2.2	633.0	100	76	250	60
5	8:43	4	202	1.2	2.4	634.6	100	76	250	60
F-1	8:43-30 8:43-30	4	203	1.4	2.7	636.4	100	76	250	60
2	8:47-30	4	204	1.4	2.7	638.2	100	76	250	60
3	8:49-30	4	203	1.4	2.7	639.7	100	76	250	60
4	8:51-30	5	206	1.5	2.9	641.6	100	76	250	60
5	8:53-30	5	203	1.6	3.1	643.4	100	76	250	60
G-1	8:54-55 8:54-55	5	201	1.6	3.1	645.2	100	76	265	60
2	8:58	5	202	1.6	3.1	647.1	100	78	265	60
3	9:00	5	205	1.7	3.3	648.8	102	78	265	60

TEST NO. 15311

LOCATION ContDATE 6-6-84

emissions test log sheet, cont.

RAMCON

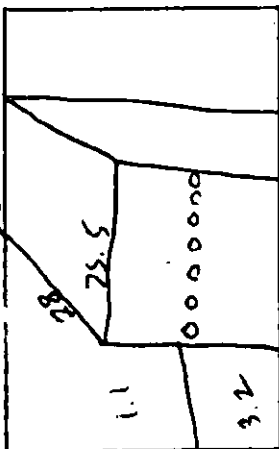
[illegible]

RAMCON ENVIRONMENTAL CORPORATION

Plant Valley Asphalt

Location Comp. Paint
 Operator Donna Jones
 Date 6-5-88
 Run No. 2
 Sample Box No. 1
 Meter Box No. 004
 Meter H @ 2.1
 C Factor .972
 Pitot Tube Coefficient Cp .879

1.96



Ambient Temperature 90
 Barometric Pressure 29.95 FINAL
 Assumed Moisture, % 1.8 INITIAL
 Probe Length, m(ft) 3 DIFFERENCE
 Nozzle Identification No. .0002985
 Avg. Calibrated Nozzle Dia., (in.) 2.50/2.50
 Probe Heater Setting 2.5
 Leak Rate, m³/min. (cfm) 0.07 6 inches
 Probe Liner Material Fluorocarbon
 Static Pressure, mm Hg (in. Hg) .05
 Filter No. 56-2508

Schematic of Stack Cross Section

TRAV. PT NO.	SAMPLING TIME (θ) min.	VACUUM in. Hg	STACK TEMP (Tg) °F	VELOCITY HEAD (Pg) in H2O	PRESSURE DIFF. ORF. MTR in H2O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F		FILTER HOLDER TEMP °F	GAS TEMP LVG CONDENSER OR LAST IMPINGER °F
							Inlet	Outlet		
1	10:08 10:10	5	200	2.1	4.1	653.2 655.3	90	86	255	60
2	10:12	5	200	1.9	3.7	657.0	94	94	255	60
3	10:14	5	195	1.9	3.7	659.5	94	84	255	60
4	10:16	5	202	1.8	3.5	661.3	102	84	255	60
5	10:18	5	202	1.8	3.5	663.5	102	84	255	60
1	10:18-20 10:20-30	4	200	1.2	2.4	665.2	102	84	250	60
2	10:22-30	4	204	1.1	2.2	666.9	106	84	250	60
3	10:25-30	4	207	1.1	2.2	668.4	106	84	250	60
4	10:26-30	4	206	1.4	2.7	670.1	106	84	250	60
5	10:28-30	4	205	1.5	2.9	672.0	110	84	250	60
1	10:29 10:31	4	207	1.20	1.4	673.4	108	88	250	60
2	10:33	3	205	.80	1.6	674.8	108	88	250	60
3	10:35	4	205	1.1	2.2	676.4	110	88	260	60

Corr = 4.0% Cor = 15.5% Or = 15.5%

RAMCON emissions test log sheet, cont. DATE 6-6-88 LOCATION Comp. Plant TEST NO. 2

TRAVERSE POINT	SAMPLING TIME (min)	VACUUM (in. Hg)	STACK TEMP (°F)	VELOCITY HEAD (in. H ₂ O)	ORIFICE DIFF. PRESSURE (in. H ₂ O)	GAS VOLUME V _m (ft. ³)	GAS SAMPLE TEMP. (°F)		SAMPLE BOX TEMP. (°F)	IMPINGER TEMP (°F)
							in	out		
4	10:32	4	208	1.2	2.4	678.0	110	88	260	60
5	10:39	4	208	1.3	2.5	679.8	110	88	260	60
1	10:40 10:42	3	208	.60	1.2	681.0	110	88	260	60
2	10:44	3	208	.60	1.2	682.3	110	88	260	60
3	10:46	3	208	.60	1.2	683.5	110	88	260	60
4	10:48	3	209	.80	1.6	684.8	110	90	260	60
5	10:50	3	209	.80	1.6	686.3	110	90	255	60
1	10:52 10:52:30	2	208	.50	.98	687.4	110	90	250	60
2	10:54:30	2	212	.50	.98	688.4	110	90	250	60
3	10:55:30	2	211	.50	.98	689.5	110	90	250	60
4	10:59:30	2	210	.60	1.2	690.8	110	90	250	60
5	11:00:30	2	208	.60	1.2	691.9	110	90	250	55
1	11:01 11:03	3	207	.70	1.4	693.1	110	90	250	55
2	11:05	3	206	.70	1.4	694.3	110	90	260	55
3	11:07	3	204	.80	1.6	695.6	110	90	260	55
4	11:09	3	204	.80	1.6	697.0	110	90	260	55
5	11:11	3	207	.80	1.6	698.5	110	90	260	55
1	11:13 11:13:30	4	203	1.6	3.1	700.2	112	92	265	60
2	11:15:30	4	207	1.5	2.9	701.8	112	92	265	60
3	11:17:30	4	200	1.5	2.9	702.0	112	92	265	60

RAMCON

emissions test log sheet, cont. DATE: 6-6-88 LOCATION Camp Pineda TEST NO. 2

Scrap Paper TEST NO. 2

[illegible]

RAMCON ENVIRONMENTAL CORPORATION

Plant Goodyear

1.96

Location Comp. Denison

Operator John J. Baker

Date 6-5-88

Run No. 3

Sample Box No. 1

Meter Box No. 004

Meter H # 21

C Factor 272

Pitot Tube Coefficient Cp 0.99

Ambient Temperature 90

Barometric Pressure 29.95

Assumed Moisture, % 18

Probe Length, m(ft) 2

Nozzle Identification No. 0001885

Avg. Calibrated Nozzle Dia., (in.) 230/230/230

Probe Heater Setting 2.5

Leak Rate, m³/min. (cfm) 1001 at 6 in.

Probe Liner Material Aluminum

Static Pressure, mm Hg (in. Hg) 0.5

Filter No. SG-2509

Schematic of Stack Cross Section

TRAV. PT NO.	SAMPLING TIME (θ) min.	VACUUM in. Hg	STACK TEMP (Tg) °F	VELOCITY HEAD (Pg) in H2O	PRESSURE DIFF. ORF. MR in H2O	GAS SAMPLE VOLUME ft ³	GAS SAMPLE TEMP. AT DRY GAS METER °F		FILTER HOLDER TEMP °F	GAS TEMP LVG CONDENSER OR LAST IMPINGER °F
							Inlet	Outlet		
1	11:50-30	4	200	1.5	2.9	707.6	104	100	230	60
2	11:54-30	4	200	1.5	2.9	710.9	104	100	230	60
3	11:58-30	4	203	1.5	2.9	713.0	104	100	230	60
4	12:02-30	4	212	1.6	3.1	714.8	110	98	230	60
5	12:06-30	5	213	1.7	3.3	716.7	116	98	230	60
1	12:10-30	3	215	2.0	1.4	718.3	116	98	240	60
2	12:05	3	210	2.0	1.4	719.5	116	98	240	60
3	12:07	3	210	2.0	1.4	720.8	118	98	240	60
4	12:09	3	210	1.0	2.0	722.6	118	98	240	60
5	12:11	3	210	1.0	2.0	723.9	118	98	240	60
1	12:14-30	3	204	1.0	1.0	725.2	118	98	240	60
2	12:15-30	3	204	1.0	1.0	726.3	118	98	245	60
3	12:17-30	3	205	1.0	1.0	727.4	118	98	245	60

602=40 602=4.5%

Or=14.8

Or=14.8

RAMCON emissions test log sheet, cont. DATE 6-5-87 LOCATION Camp TEST NO. 3

TRAVERSE POINT	SAMPLING TIME (min)	VACUUM (in. Hg)	STACK TEMP T_s (°F)	VELOCITY HEAD ΔP_s (in. H ₂ O)	ORIFICE DIFF. PRESSURE ΔH (in. H ₂ O)	GAS VOLUME V_m (ft. ³)	GAS SAMPLE TEMP. (°F)		SAMPLE BOX TEMP. (°F)	IMPIGNER TEMP (°F)
							in	out		
4	12:33:30	3	205	.60	1.2	728.6	118	98	250	60
5	12:34:30	3	204	.60	1.2	729.7	120	98	250	65
1	12:35:30	3	204	.60	1.2	730.8	118	98	250	65
2	12:36:30	3	207	.70	1.4	731.9	120	99	230	65
3	12:37:30	3	207	.70	1.4	733.4	120	98	230	65
4	12:38:30	3	208	.80	1.6	734.7	120	100	250	65
5	12:39:30	3	209	.80	1.6	736.2	120	100	250	65
1	12:40:30	3	206	.60	1.2	737.4	120	100	250	65
2	12:41:30	3	207	.70	1.4	738.8	120	100	270	65
3	12:42:30	3	210	.80	1.6	740.1	120	100	270	65
4	12:43:30	4	212	1.2	2.4	741.8	120	100	270	65
5	12:44:30	4	214	1.2	2.4	743.5	120	100	270	65
1	12:45:30	4	212	1.0	2.0	745.0	120	100	260	65
2	12:46:30	4	212	1.0	2.0	746.6	120	100	260	65
3	12:47:30	4	214	1.1	2.2	748.1	120	100	260	65
4	12:48:30	4	216	1.2	2.4	749.8	120	100	260	65
5	12:49:30	4	218	1.2	2.4	751.5	120	100	260	65
1	12:50:30	4	214	1.2	2.4	753.0	120	100	260	65
2	12:51:30	4	216	1.3	2.5	754.7	120	100	260	65
3	12:52:30	5	213	1.4	2.7	756.3	120	100	260	65

[illegible]

IX. CALIBRATIONS

METER BOX CALIBRATION DATA AND CALCULATION FORM

(English units)

Date

6/18/88

Meter box number

004/638809Barometric pressure, $P_b =$ 29.99

in. Hg

Calibrated by

Shawn Greenwood

Orifice manometer setting (ΔH), in. H_2O	Gas volume		Temperature				Time (θ), min	Y_i	ΔH_i in. H_2O
	Wet test meter (V_w), ft^3	Dry gas meter (V_d), ft^3	Wet test meter (t_w), $^{\circ}F$	Dry gas meter					
				Inlet (t_{d_i}), $^{\circ}F$	Outlet (t_{d_o}), $^{\circ}F$	Avg ^a (t_d), $^{\circ}F$			
0.5	5	041.00 046.21	75°	74° 158°	80° 82°	90°	13.38	985	2.04
1.0	5	047.00 55.72	75°	96° 164°	80° 83°	91°	9.27	984	1.96
1.5	10	053.00 063.14	75°	96° 164°	82° 85°	91°	15.32	982	1.99
2.0	10								
3.0	10								
4.0	10								
Avg								984	2.00

ΔH , in. H_2O	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t + 460)}$	$\Delta H_i = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460) \theta}{V_w} \right]^2$
0.5	0.0368		
1.0	0.0737		
1.5	0.110		
2.0	0.147		
3.0	0.221		
4.0	0.294		

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d .

METER BOX CALIBRATION DATA AND CALCULATION FORM

(English units)

Date 5-31-88Meter box number 004/63121Barometric pressure, $P_b = 30.11$ in. Hg Calibrated by 0000-10000

Orifice manometer setting (ΔH), in. H_2O	Gas volume		Temperature				Time (θ), min	Y_i	$\Delta H \theta$ in. H_2O
	Wet test meter (V_w), ft ³	Dry gas meter (V_d), ft ³	Wet test meter (t_w), °F	Dry gas meter					
				Inlet (t_{d_i}), °F	Outlet (t_{d_o}), °F	Avg ^a (t_d), °F			
0.5	5								
1.0	5	351.0 333.226	75	99 101	81 81	90.5	9.63	.482	2.0
1.5	10								
2.0	10	349.2 327.469	75	101 104	82 82	92.25	13.87	.460	2.1
3.0	10	336.609 347.203	75	105 105	83 82	93.75	11.36	.474	2.1
4.0	10								
							Avg	.472	2.1

ΔH , in. H_2O	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t + 460)}$	$\Delta H \theta_i = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460) \theta}{V_w} \right]^2$
0.5	0.0368		
1.0	0.0737		
1.5	0.110		
2.0	0.147		
3.0	0.221		
4.0	0.294		

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d .

RAMCON ENVIRONMENTAL CORPORATION

EPA QA MANUAL VOL. III
 Section No. 3.4.2
 Revision No. 0
 Date January 15, 1980
 Page 17 of 22

Date 2-10-88 Thermocouple number Outlet
 Ambient temperature 55°F °C Barometric pressure 29.96 in. Hg
 Calibrator J. Greenwood Reference: mercury-in-glass ☒
 other ☐

Reference point number ^a	Source ^b (specify)	Reference Thermometer Temperature, °C	Thermocouple Potentiometer Temperature, °C	Temperature Difference, % ^c
A	ICE BATH	33°F	32°F	.03%
B	OVEN	152°F	150°F	.01%
C	OVEN	175°F	175°F	0%
D	Ambient	55°F	55°F	0%
	6-6-88	60°F	60°F	0%

^aEvery 30°C (50°F) for each reference point.

^bType of calibration system used.

^c
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%$$

Figure 2.5 stack temperature sensor calibration data form.

RAMCON ENVIRONMENTAL CORPORATION

EPA QA MANUAL VOL. III
 Section No. 3.4.2
 Revision No. 0
 Date January 15, 1980
 Page 17 of 22

Date 2-10-88 Thermocouple number Andit
 Ambient temperature 55°F °C Barometric pressure 29.96 in. Hg
 Calibrator S. Greenwood Reference: mercury-in-glass ✓
 other _____

Reference point number ^a	Source ^b (specify)	Reference Thermometer Temperature, °C	Thermocouple Potentiometer Temperature, °C	Temperature Difference, % ^c
A	ICE BATH	33°F	33°F	0%
B	OVEN	150°F	151°F	.0079%
C	OVEN	175°F	173°F	.01%
D	AMBIENT 6-6-88	55°F 6-6-88 80°F	54°F 80°F	.02% 0%

^aEvery 30°C (50°F) for each reference point.

^bType of calibration system used.

^c
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%$$

Figure 2.5 stack temperature sensor calibration data form.

(39)

RAMCON ENVIRONMENTAL CORPORATION

Lear Siegler Stack SamplerNozzle Diameter Calibration

Date _____ Signature _____

Nozzle No.	Average Diameter	Nozzle No.	Average Diameter
1	_____	7	_____
2	_____	8	_____
3	_____	9	_____
4	_____	10	_____
5	_____	11	_____
6	_____	12	_____

Pitot Tube Calibration (S Type)Pitot Tube Identification No. 34 Date 2-5-88Calibrated by: G. A. Turner"A" SIDE CALIBRATION

Run No.	Δp std cm H ₂ O (in. H ₂ O)	Δp (s) cm H ₂ O (in. H ₂ O)	C_p (s)	DEVIATION $C_p(s) - \bar{C}_p(A)$
1	1.05	1.5	.836	.007
2	.87	1.3	.818	.011
3	.58	0.83	.835	.006
$\bar{C}_p$ (SIDE A)			.829	

"B" SIDE CALIBRATION

Run No.	Δp std cm H ₂ O (in. H ₂ O)	Δp (s) cm H ₂ O (in. H ₂ O)	C_p (s)	DEVIATION $C_p(s) - \bar{C}_p(B)$
1	1.05	1.51	.833	.009
2	.87	1.32	.812	.012
3	.58	0.85	.826	.002
$\bar{C}_p$ (SIDE B)			.824	

$$\text{AVERAGE DEVIATION} = \sigma(A \text{ OR } B) = \frac{\sum |C_p(s) - \bar{C}_p(A \text{ OR } B)|}{3} \quad + \text{MUST BE } \leq 0.01$$

$$|\bar{C}_p(\text{SIDE A}) - \bar{C}_p(\text{SIDE B})| + \text{MUST BE } \leq 0.01$$

Form No. EED-17-1

$$C_p(s) = C_p(\text{std}) \sqrt{\frac{\Delta p \text{ std}}{\Delta p_s}}$$

(40)

RAMCON

Lear Siegler Stack Sampler

Heating Probe Calibration

Probe No. 34

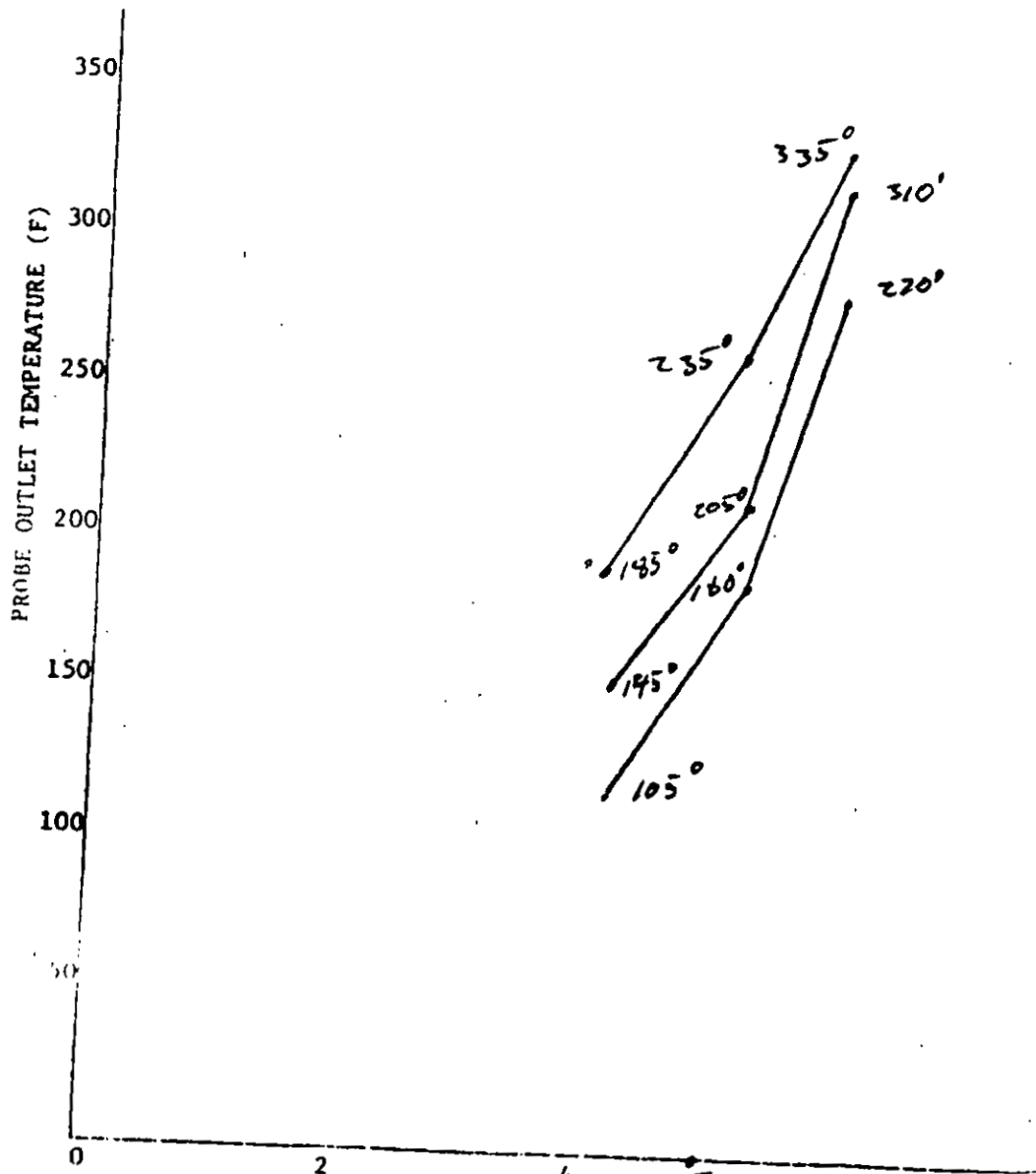
Probe Length 3 ft

Date of Calibration 2/21/88

Signature STuck

Name of Company to be tested _____

Note: 3 ft. probe - 5 min. warmup
6 ft. probe - 15 min. warmup
10 ft. probe - 30 min. warmup
Calibration flow rate = .75 CFM



STACK TEMPERATURE SENSOR CALIBRATION DATA FORM

Date 2/9/88 Thermocouple number 34
 Ambient temperature 55 °F Barometric pressure 29.95 in. Hg
 Calibrator G. Greenwood Reference: mercury-in-glass ☒
 other ☐

Reference point number	Source ^a (specify)	Reference thermometer temperature, °C	Thermocouple potentiometer temperature, °C	Temperature difference, ^b %
A	Ice water	32°F	31°F	.03%
B	Boiling water	212°F	212°F	0%
C	Boiling oil	381°F	377°F	.01%
D	Ambient 6-6-88	60°F	60°F	0%

^a Type of calibration system used.

^b
$$\left[\frac{(\text{ref temp, } ^\circ\text{C} + 273) - (\text{test thermom temp, } ^\circ\text{C} + 273)}{\text{ref temp, } ^\circ\text{C} + 273} \right] 100 \leq 1.5\%$$

X. RAMCON PERSONNEL

RAMCON Environmental Stack Test Team

Sumner Buck - President

Sumner Buck is the President of RAMCON Environmental. He is a graduate of the EPA 450 "Source Sampling for Particulate Pollutants" course and the 474 "Continuous Emissions Monitoring" course all given at RTP. Mr. Buck is a qualified V.E. reader with current certification. Mr. Buck has personally sampled over 300 stacks including over 200 asphalt plants. He is 43 years old and a graduate of the University of Mississippi with graduate studies at Memphis State University and State Technical Institute of Memphis.

Allen Turner - Team Leader

Allen Turner has been employed by RAMCON for three years. He has undergone extensive training in Methods 1 through 9. He is qualified as a team leader and is currently certified as a V.E. reader.