

Note: This is a reference cited in *AP 42, Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

The file name refers to the reference number, the AP42 chapter and section. The file name "ref02_c01s02.pdf" would mean the reference is from AP42 chapter 1 section 2. The reference may be from a previous version of the section and no longer cited. The primary source should always be checked.

AP-42 Section 11.6
 Reference 32
 Report Sect. 4
 Reference 25

Source category: Portland Cement
 Plant name : Leeds Portland Cement Co.
 Test date : 9/17 - 9/21/84
 Process : dry

Date: 01/12/93
 Location: Leeds, AL
 Ref. No.: 25
 Basis for process rate : kiln feed

Source	Type of control	Pollutant	Run No.	Emission rate, lb/hr	Process rate, ton/hr	Emission factor		Volumetric flow rate, DSCFH	Concen ppm
						kg/Mg	lb/ton		
rotary kiln w/ suspension preheater (coal-fired)	ESP	BASED ON FEED RATE							
		filt. PM	2	24.61	140	0.0879	0.1758		
		filt. PM	3	25.40	138	0.0920	0.1841		
		filt. PM	4	16.74	135	0.0620	0.1240		
		filt. PM	5	15.66	134	0.0584	0.1169		
		AVERAGE				0.0751	0.1502	RATING: B	
		CO2	2	141,724	140	506	1,012	6,543,110	19.0
		CO2	3	140,901	138	511	1,021	6,505,131	19.0
		CO2	4	134,203	135	497	994	6,261,813	18.8
		CO2	5	131,152	134	489	979	6,119,463	18.8
		AVERAGE				501	1,002	RATING: B	
clinker cooler	cyclone + gravel bed filter	BASED ON KILN FEED RATE							
		filt. PM	8	73.85	154	0.2398	0.4795		
		filt. PM	9	38.92	163	0.1194	0.2388		
		filt. PM	10	49.90	155	0.1610	0.3219		
		AVERAGE				ERR	ERR	RATING: B	
		CO2	8	3,676	140	13.1	26.3	4,606,761	0.7
		CO2	9	4,103	138	14.9	29.7	5,141,971	0.7
		CO2	10	4,696	135	17.4	34.8	5,884,683	0.7
		AVERAGE				134	15.1	30.3	RATING: B

- Notes:
1. Raw mill data not useable because mill is ducted to same stack as kiln.
 2. Kiln emission factors for emissions with mill off only.
 3. Note that cooler emission factors are based on kilns feed rate; cooler feed rate not provided in report.

Section III

DESCRIPTION OF INSTALLATION - GENERAL

The progression of materials through the facility at Universal Atlas Cement in Leeds, Alabama, begins at the raw grinding department. The raw grinding department receives proportional raw materials and grinds them to the proper size. Acceptable size finished products are transferred from the raw mill area to blending/storage silos via a pumping system.

A feed bin receives the proportioned raw materials, and a constant level is maintained in the feed bin by a weight controller on the central control panel. From the mill feed bin the raw mix is delivered to the Loesche Mill feed chute via a weigh feeder. The chute contains a triple gate feeder which serves as an air lock for the Loesche Mill. Material out of the chute falls on a circular grinding bed and is thrown by centrifugal force under the grinding rollers. A retention ring on the periphery of the grinding bed forms the material into a predetermined, uniform thickness. The ground material spills over the edge of the retention ring and is elevated to a rotary classifier at the top of the mill by an air stream passing up through the angular louvers at the periphery of the grinding bed. The classifier separates the fines from the coarse particles - the coarse falling back into the grinding bed. The finer particles are carried by the air stream to an electrostatic precipitator. The product discharge from the precipitator is then sampled and transferred to the blending/storage silo via the pumping system.

The Loesch Mill is also air swept for simultaneous drying of the raw material by hot gases from the preheater. In case

no waste gases are available, the hot air for drying is produced by an air furnace which is equipped with automatic temperature controls. When the Loesche Mill is not operating, kiln gases are diverted through a conditioning tower to maintain suitable characteristics for particular collection by the precipitator.

Blending of raw materials occurs in a silo which utilizes a four-way cycling air valve to provide a uniform mix. Withdrawal from the blending silo is of the continuous overflow type which transports material via an air slide system to a selected storage silo. A rotary valve located at the bottom of the blending silo is used for direct removal of material into storage silo #1. Each storage silo is capable of two material withdrawal options. The normal material flow is through a level box and weighing system with the subsequent transport of material to the preheater by a pump. A bypass option routes the material around the weighing system to a pump for recirculation into the blending silo.

Dry Process

The preheater, rotary kiln, and clinker cooler transform the blended raw material into the clinker product. Raw material is clinkered by the intense heat flowing through the kiln and preheater. Clinker from the kiln discharges at approximately 2600°F and is quickly reduced to approximately 150°F at the discharge of the clinker cooler.

Hot kiln waste gases enter the refractory lined preheater through the duct beneath the kiln and stage #14 of the preheater; the gases are drawn up through the four stages by a fan. The gases from the kiln enter stage #4 at about 1900°F and leave stage #1 at about 650°F. Kiln feed enters the preheater and passes counter to the gas flow; material particles absorb heat as they mix with the hot gases.

Blended kiln feed material pumped from the storage silos enters the top of the preheater through the gas duct between stages #1 and #2. Heat is transferred from the gases to the pulverized material that is descending through the cyclones. The preheater quickly raises the temperature of the dry blended raw material from approximately 140°F to about 1450°F in four progressively hotter stages. The feed forms a suspended cloud as it mixes with the exhaust gases. Kiln waste gases provide all the heat required by the preheater. The preheater feed rate is controlled by kiln speed. The feed ratio to the kiln revolution ratio can be changed at any time.

Clinker discharges by gravity from the kiln into a six compartment grate cooler that is equipped with a clinker breaker at the discharge end. The clinker cooler performs four functions: (1) it quickly cools clinker, (2) it recovers heat from use in the kiln and other departments, (3) it reduces clinker to desired size, and (4) it drags cooled clinker to a belt conveyor.

The cooler has six fans that force air at ambient temperatures through the bed of hot clinker for quick temperature reduction. Heated air that has passed through the clinker bed has three possible exits; through the cyclone gravel bed filter, the kiln, or the coal mill. Coal is the principal fuel for firing the kiln. If needed, #2 oil is present for start-up purposes and emergency firing.

A process flow diagram is shown by Figure 3-1.

GENERAL DESCRIPTION OF THE FILTER - CLINKER COOLER

The system consists of twelve (12) cyclone/filter bed modules operating in parallel from a main inlet chamber.

The operation of each cyclone/filter module is identical. Figure 3.2 illustrates the details of a cyclone/filter module.

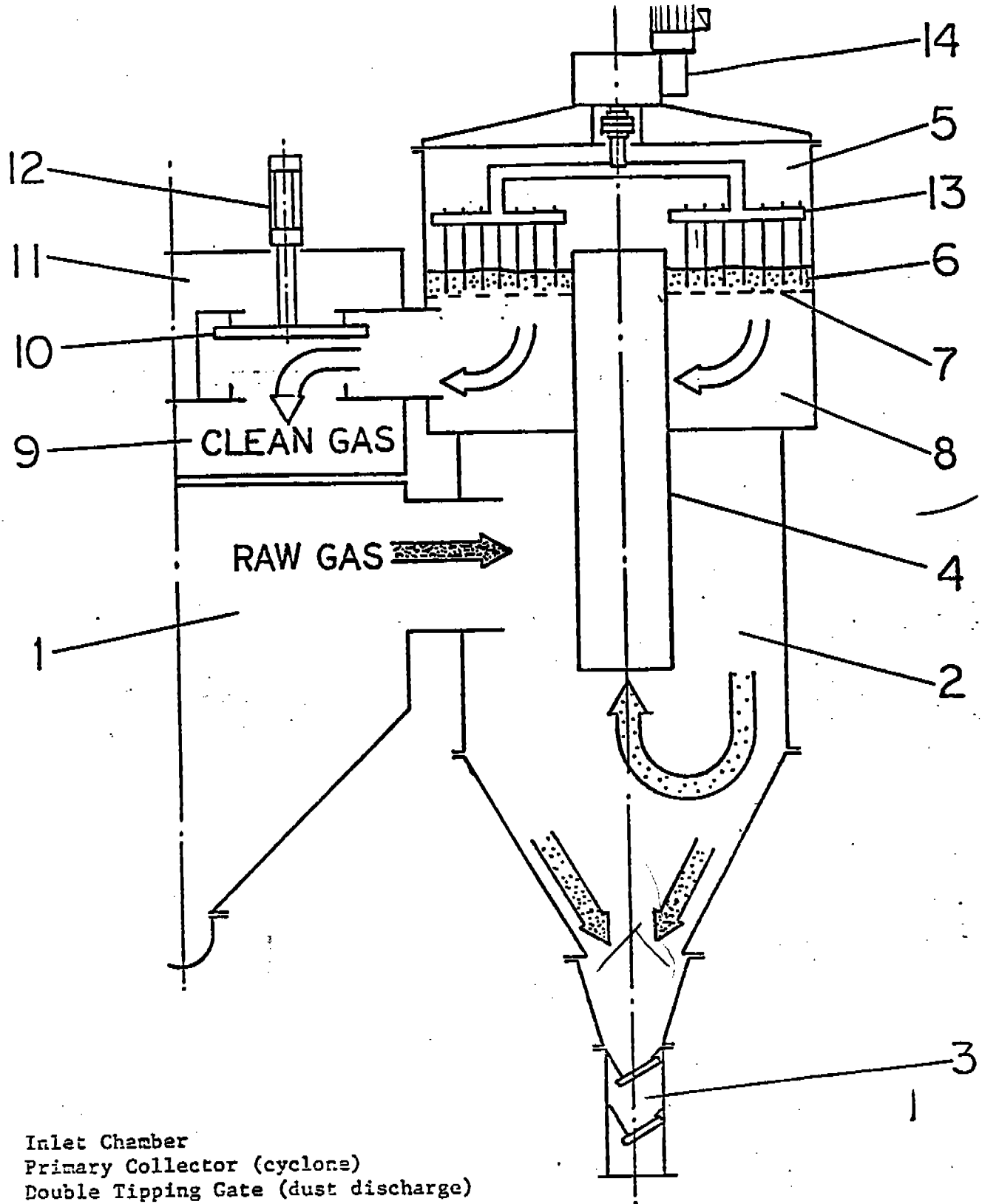
NOTE: Double filter modules differ only in the fact that there are two (2) valves, beds and rakes for each cyclone. The sequence described below would occur simultaneously in both of the filter beds located over a particular cyclone.

Raw dust-laden gas enters the module from the inlet chamber (1) into the cyclone. The cyclone (2) removes the coarse dust and the gas flow moves up the vortex tube (4) into the filter chamber (5). The removed dust is discharged via the tipping gates (3). The gas flow then is directed through the filter media (6), where the remaining dust is removed. The gas then passes through the media support screen (7), into the clean gas chamber (8), and through the clean gas outlet duct (9).

The above described process continues until the filter bed becomes sufficiently coated with dust to require cleaning. At this point, the cyclone/filter module is isolated from the clean gas stream by actuation of the backwash control valve (10) via the valve operator (12). At this point, the module has entered the bed cleaning or backflush mode. This mode is illustrated in Figure 3.3. During bed cleaning, clean air enters the unit through the backflush air inlet (11) and is directed upward through the filter media (6). During this period, the stirring rake (13) is activated via the stirring rake motor (14). During the combined rake and air flushing, the dust is removed from the filter media and directed down the vortex tube (4). This dust is an agglomerated particles significantly larger than when they first reached the gravel bed. Some of the resuspended dust

is removed in the cyclone (2); the remainder passes into the inlet chamber (1) where more of the dust is removed because of the greatly reduced velocities. The backflush gases and the balance of agglomerated dust is distributed to the other cyclone/filter modules for removal by cyclonic action. Only one (1) filter module (having one (1) filter bed on SB units and two (2) filter beds on DB units) is removed at a time for cleaning. The cleaning cycle is controlled by timers located in the control panel.

jh/MISC2/013



1. Inlet Chamber
2. Primary Collector (cyclone)
3. Double Tipping Gate (dust discharge)
4. Vortex Tube
5. Filter Chamber
6. Gravel Bed
7. Screen Support for Bed
8. Clean Gas Collection Chamber
9. Exhaust Port
10. Backwash Control Valve
11. Backwash Air Inlet
12. Valve Cylinder
13. Stirring Rake
14. Stirring Rake Motor/Reducers

FIGURE 3-2

OPERATING PHASE

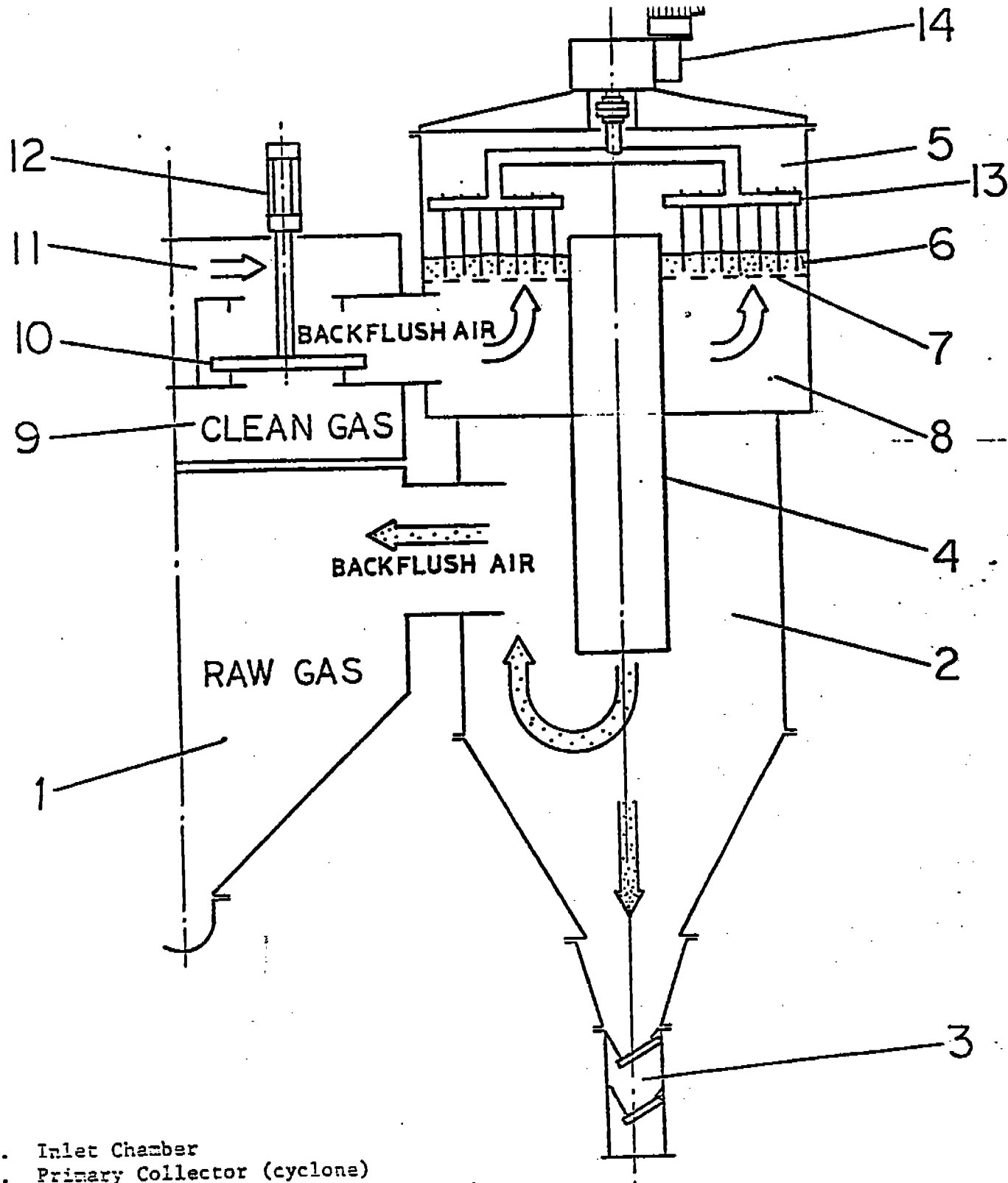


FIGURE 3-3

CLEANING PHASE

Section IV

SAMPLING AND ANALYTICAL PROCEDURES

Sampling Location and Procedures

An EPA sampling train as described in the Air Pollution Control Office publication APTD-0581 was used for this study. Figure 4-1 shows a schematic of this sampling equipment. The only modification to this configuration was the substitution of a right angle for the cyclone and flask.

Layout of the sampling ports in relation to flow disturbances for the Loesche Mill main stack is shown by Figure 4-2, and for the Clinker Cooler Stack, Figure 4-3. Based on the stack diameters and the upstream and downstream distances of the sampling ports from the closest flow disturbances the required number of sampling points according to methods prescribed in the Federal Register, was determined. Table 4-1 shows the dimensions from the stack wall to each sampling point on the main (ESP) stack, while Table 4-2 shows this for the Clinker Cooler stack. Sampling was conducted through four (4) ports on each source.

Prior to the start of the first sampling run, velocity traverses, and temperature readings were made in the stack. Moisture was estimated from previous sampling data. Using the above data, the nomograph was set up to utilize an appropriate size nozzle. Settings for the nomograph and the exact nozzle diameter are shown on the field data sheets (Appendix).

The sampling procedures followed were as outlined by Method 5 from the previously mentioned Federal Register. Sampling data for each run is a part of the Appendix. As noted by the data sheets, each point was sampling for three minutes

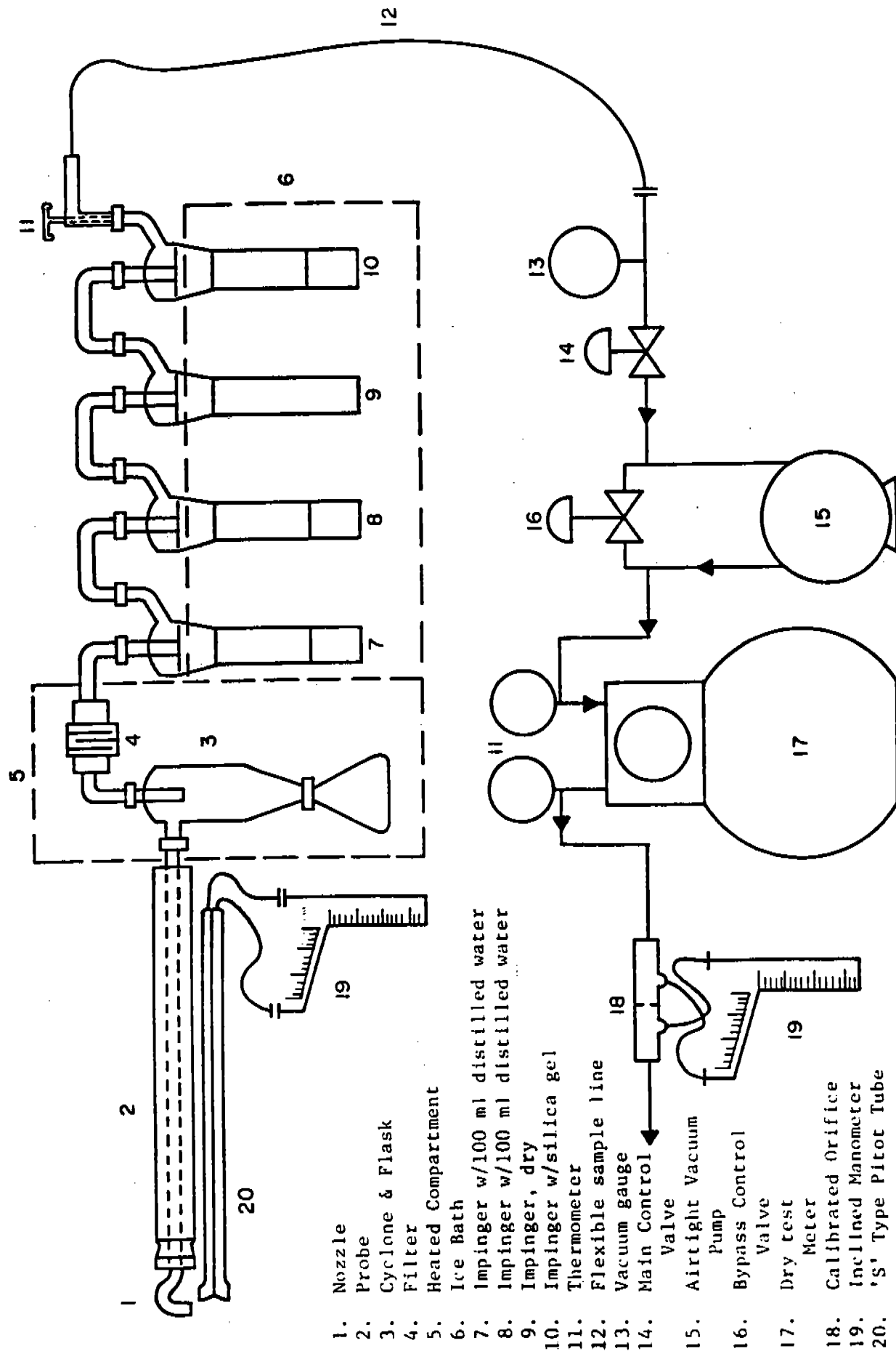


FIGURE 4-1. PARTICULATE SAMPLING TRAIN

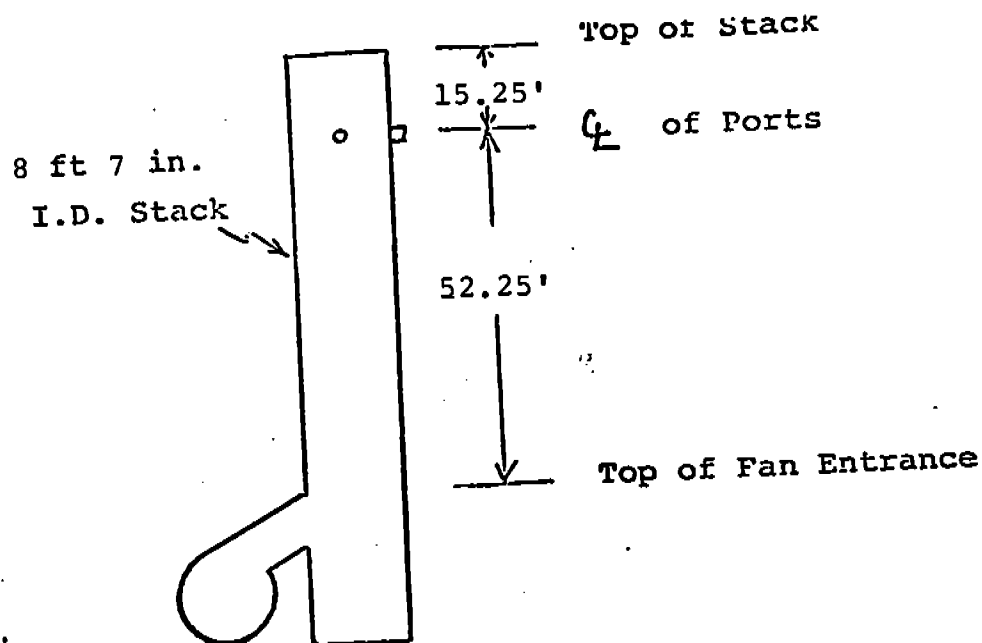


FIGURE 4-2.

Location of Sampling Ports - Main Stack
Sampling Conducted Through Four Ports

Table 4-1.
Location of Sampling Points

Point	% of Stack I.D.	X Stack I.D. (inches)	=	Distance from Port (inches)
1	3.3	103		3.3
2	10.5	103		10.8
3	19.4	103		20.0
4	32.3	103		33.3

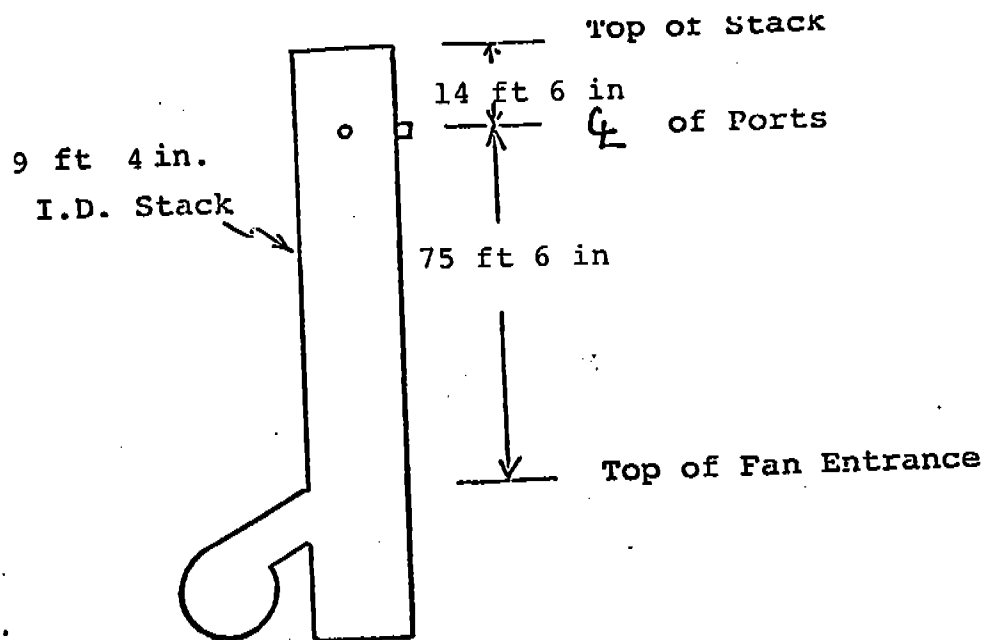


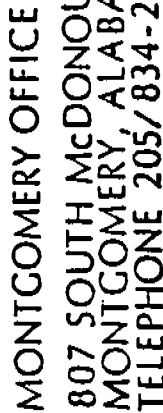
FIGURE 4-3.

Location of Sampling Ports - Clinker Cooler
Sampling Conducted Through Four Ports

Table 4-2.
Location of Sampling Points

Point	% of Stack I.D.	X Stack I.D. (inches)	=	Distance from Port (inches)
1	2.6	112		2.9
2	8.2	112		9.2
3	14.6	112		16.4
4	22.6	112		25.3
5	34.2	112		38.3

APPENDIX A
LABORATORY DATA



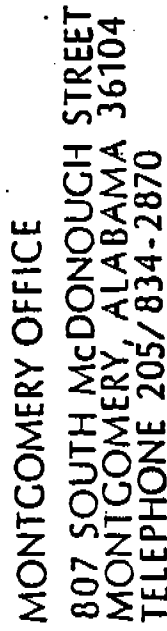
Client Lehigh
Date 9/24 Weighings By Am
Sampling Location Main
Remarks: _____

LABORATORY DATA

PARTICULATE SOURCE	RUN # 1 MILLION			RUN # 2 MILLION			RUN # 3 MILLION			RUN # 4 MILLION		
	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)
PROBE & CYCLONE WASH	15.7474g 100 ml	m-10 15.7255	.0219	104.3688g 120 ml	#5 104.3359	.0330	101.8042g 100 ml	#6 101.7732	.0310	102.5258g 80 ml	#4 102.5053	.0205
FILTER	4.985	#37 4.441	.0544	4.928	#38 4.500	.0428	4.947	#39 4.468	.0479	4.849	#40 4.529	.0320
IMPINGERS ACETONE WASH												
IMPINGERS H ₂ O												
	Gross Weight Less Blank Wt.	.0763 -.0002		Gross Weight Less Blank Wt.	.0758 -.0002		Gross Weight Less Blank Wt.	.0789 -.0002		Total Blank Wt.	.0525 -.0002	
	Net Weight	.0761		Net Weight	.0756		Net Weight	.0787			.0523	

NOTE: The weight of the probe and cyclone wash blank for each sample is in direct proportion to the weight of the actual blank based on volume.

[illegible]



Client Lehigh
Date 9/24 Weighings By Am
Sampling Location Main
Remarks: _____

LABORATORY DATA

PARTICULATE SOURCE	RUN # 45			RUN # 26			RUN # 27			BLANKS		
	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)
PROBE & CYCLONE WASH	11.6439 g 75 ml	#25 11.6781	.0158	10.7041 g 130 ml	#24 110.6863	.0178	117.7348 g 120 ml	#A 117.7193	.0153	---	---	---
FILTER	4975	4643	.0332	4754	4509	.0245	4753	4512	.0241			
IMPINGERS ACETONE WASH												
IMPINGERS H ₂ O												
	Gross Weight	0440		Gross Weight	0423		Gross Weight	0394		Total Blank Wt.		
	Less Blank Wt.	1.0002		Less Blank Wt.	0.0003		Less Blank Wt.	.0002				
	Net Weight	0488		Net Weight	0420		Net Weight	0392				

NOTE: The weight of the probe and cyclone wash blank for each sample is in direct proportion to the weight of the actual blank based on volume.

[illegible]



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807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
TELEPHONE 205/834-2870

Client Lehigh

Date 9/24 Weighings By DM

Sampling Location CLINKER COOLER

Remarks:

LABORATORY DATA

PARTICULATE SOURCE	RUN # 1 (9)			RUN # 2 (9)			RUN # 3 (10)			BLANKS		
	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)
PROBE & CYCLONE WASH	<u>101.34298</u> 110 ml	<u>101.2234</u> #36	<u>.1195</u>	<u>112.55638</u> 100 ml	<u>112.5063</u> #7	<u>.0500</u>	<u>106.11258</u> 50 ml	<u>106.0707</u> #6	<u>.0418</u>	<u>109.22068</u> 100 ml	<u>109.3204</u> #28	<u>+0002</u>
FILTER	<u>5284</u>	<u>4585</u> #45	<u>.0699</u>	<u>5171</u>	<u>4841</u> #44	<u>.0530</u>	<u>4030</u>	<u>3136</u> #21	<u>.0894</u>			
IMPINGERS ACETONE WASH												
IMPINGERS H ₂ O												
	Gross Weight			Gross Weight			Gross Weight			Total Blank Wt.		
	Less Blank Wt. ~			Less Blank Wt. -			Less Blank Wt. -			-0002		
	Net Weight			Net Weight			Net Weight			.1310		
	<u>.1894</u>			<u>.1030</u>			<u>.1312</u>			<u>.0002</u>		
	<u>.0002</u>			<u>.0002</u>			<u>.0002</u>			<u>.0002</u>		
	<u>.1892</u>			<u>.1028</u>			<u>.1310</u>			<u>.0002</u>		

NOTE: The weight of the probe and cyclone wash blank for each sample is in direct proportion to the weight of the actual blank based on volume.

SILICA GEL

Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)	Final Wt (g)	Initial Wt (g)	Diff. (g)

APPENDIX B
FIELD DATA SHEETS

LOESCHE MILL "ON"

RUNS 1, 6, 7

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

Vnn
JCOA
2/19/84

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA. <u>.258</u> in.		FILTER NUMBER	
PITOT CORR. FACTOR <u>.85</u>		SILICA GEL NUMBER	
CRIFICE CORR. FACTOR <u>1.855</u>		READING	GAS METER
FINAL			IMPINGER COND.
CO ₂	13.0	518.322	340
O ₂	10.0	462.266	200
CO	10.0		
N ₂		56.056	140

ORSAT READING	
CO ₂	13.0
O ₂	10.0
CO	10.0
N ₂	

REMARKS: T_{amb} = 93 °F at 20 = 15 (from prior data)
c = 0.85 / c_s = 2.05

PLANT: Lehigh

DATE: 9/17/84 RUN NUMBER

TIME START 1446 TIME END 1559

MATERIAL PROCESSING RATE LOESCHE MILL ON

SAMPLING LOCATION MAIN

WEATHER P/C

AMBIENT TEMP. 82 °F STACK PRESS. -0.60 (29.936) in. Hg

BAROMETRIC PRESS. 29.98 "Hg WET BULB

DRY BULB °F MOISTURE

DEW POINT °F

SKETCH: Nozzle = .257
Orifice = .258
Leak = .007/min
Pitot = .014 min/50 sec
Impact at 3.4 in.
Static at 3.5 in.

FINAL LEAK CHECK
 SAMPLE TRAIN AT 8 in. Hg
 FINAL 8.367
 INITIAL 8.360
 LEAK .007/min
 PITOT .014 min/50 sec
 IMPACT at 3.4 in.
 STATIC at 3.5 in.

STACK INFORMATION
 DISTANCE FROM DOWNSTREAM ft.
 DISTURBANCE: UPSTREAM ft.

DIA. ft.
 AREA ft²
31 silica gel = 16.5g

PORT #	POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-ΔP (in. H ₂ O)	√ΔP	ORIFICE VELOCITY HEAD-ΔP (in. H ₂ O)	GAS METER TEMP. (°F)		STACK TEMP. - T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°F)	VACUUM (in. Hg)
								INLET	OUTLET				
1	1		0		0.62	787	1.90	82	82	200	65	130	—
2	2		4	465.3	0.65	806	2.00	82	82	200	65	130	—
3	3		8	469.0	0.77	878	2.35	83	83	200	65	130	—
4	4		12	472.2	0.80	894	2.45	85	85	200	65	130	1.5
2/1	1	21.0	0	475.6	0.70	837	2.05	85	85	200	65	130	1.5
2	2	21.0	4	478.7	0.85	922	2.55	85	85	200	65	130	2.0
3	3	21.0	8	482.4	0.90	949	2.70	87	87	200	65	130	2.0
4	4	21.0	12	486.0	0.88	938	2.65	87	87	200	65	130	2.0

Lehigh - Run: 1

[illegible]

PARTICULATE EMISSION DATA CALCULATION SHEET



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807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant Lehigh
Date 9/17 Run # 1
Sampling Location MAIN -
LOESCHE MILL ON

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (.85) (.885) \sqrt{\frac{660}{(29.936)(29.398)}}$$

$$V_s = \underline{55.692} \text{ fps}$$

Sampling Data	
$\bar{P}_s = 660$ in. Hg	$\bar{T}_s = 546$ °R
$C_p = .85$	(H ₂ O) avg = .885
$P_s = 29.936$ in. Hg	$P_b = 29.398$ in. Hg
$A_s = 57.863$ sq. ft.	$V_s = 56.056$ cu. ft.
H_2O (condensate) = 140	mls
H_2O (silica gel) = 16.5	grams
$\Delta H_{avg} = 2.364$	in H ₂ O; $P_b = 29.98$ in Hg
Total particulate <u>2.0761</u> grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls H}_2\text{O}}{\text{condensate} + \text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) \left(\frac{140 + 16.5}{156.5} \right) = (0.0474) (156.5) = \underline{7.418} \text{ cu. ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64) (56.056) \left(\frac{29.98 + \frac{2.364}{13.6}}{546} \right) = (17.64) (0.552) = \underline{54.6098} \text{ cu. ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{7.418}{54.6098 + 7.418} = \underline{.1196} = \underline{12.0} \%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - .12) (55.692) \left(\frac{57.863}{57.863} \right) \left(\frac{528}{660} \right) \left(\frac{29.936}{29.92} \right) = \underline{8,171,487} \text{ cu. ft./hr}$$

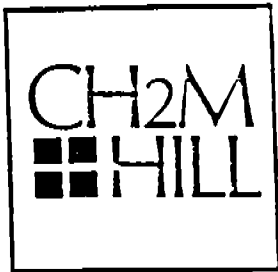
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (2.0761) / (54.6098) = \underline{0.0215}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (0.0215) (8,171,487) (1.43 \times 10^{-4}) = \underline{25.12}$$



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807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client Lehigh
Date 9/17 Run # 1
Sampling Location MainCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 12 %%CO₂ = 15% O₂ = 10%CO = —% N₂ = 75

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.120	x	1.0	x	18.0	=	2.160
CO ₂	.150	x	.88	x	44.0	=	5.810
O ₂	.100	x	.88	x	32.0	=	2.816
CO	—	x	—	x	28.0	=	—
N ₂	.750	x	.88	x	28.2	=	18.612

Average Molecular Wt., wet basis (M_s) = 29.398 lb/lb-moleDensity of gas referred to air = $\frac{29.398}{28.95} = 1.015$ Gas density correction factor = $\sqrt{\frac{1.00}{1.015}} = \sqrt{.9848} = .99$



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807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LehighDate 9/17 Run# 1Sampling Location MainCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} - ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$\frac{0V_s P_s A_n}{s s n}$$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 660 °R
 V_1 = Total volume of liquid collected in impingers & silica gel = 156.5 ml
 V_m = Gas sample thru dry gas meter @ meter conditions = 56.056 cu.ft.
 T_m = Average dry gas meter temperature = 546 °R
 P_{bar} = Barometric pressure @ sampling site = 29.98 in.Hg
 ΔH = Average pressure drop across orifice = 2.364 in.H₂O
 t = Total sampling time = 64 min.
 V_s = Stack gas velocity = 55.672 fps
 P_s = Stack gas pressure = 29.936 in.Hg
 A_n = Cross-sectional area of nozzle = .000363 sq.ft.

$$I = \frac{(660) (1.667) \left[(0.00267) (156.5) + \frac{56.056}{546} \left(29.98 + \frac{2.364}{13.6} \right) \right]}{(64) (55.672) (29.936) (.000363)}$$

$$I = \frac{(1100.22) [(1.4179) + (.1027) (30.1538)]}{()}$$

$$I = \frac{(1100.22) (1.4179 + 3.0958)}{()} = \frac{(1100.22) (3.5137)}{()} = 3865.8322$$

$$I = \frac{99.809}{()} = \underline{99.8} \% \text{ of isokinetic sampling}$$

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA. 25.0 in. FILTER NUMBER _____
PITOT CORR. FACTOR 0.85 SILICA GEL NUMBER _____
ORIFICE CORR. FACTOR 1.855

ORSAT READING

% CO₂ 15.5 16.0 16.5
% O₂ 9.5 11.5 11.0
% CO _____
% N₂ _____

REMARKS: T_g = 200 T_m = 70 %H₂O = 15
c = 0.85 T_g = 0.60

READING	GAS METER	IMPINGER COND.
FINAL	774.920	360
INITIAL	718.050	200
NET	56.870 2.412 54.458	160

PLANT

DATE 9/20 RUN NUMBER 6TIME START 0804 TIME END 0924MATERIAL PROCESSING RATE LOESCHE WTC ONSAMPLING LOCATION MANWEATHER HazyAMBIENT TEMP. 58 °F STACK PRESS. -0.62 in. HgBAROMETRIC PRESS. 29.98 in. Hg WET BULB

DRY BULB °F MOISTURE

DEW POINT °F

SKETCH:

FINAL LEAK CHECK
SAMPLE TRAIN AT 10 in. Hg
FINAL 0.030
INITIAL 0.065
LEAK 0.015 /min.
PITOT
IMPACT 0 at 3.8 in.
STATIC 0 at 5.4 in.

SEE PREVIOUS
pump off from C 729.614
2,412

STACK INFORMATION

DIA. _____ in.
AREA _____ ft²
DISTANCE FROM DOWNSTREAM _____ ft.
DISTURBANCE: UPSTREAM _____ ft.

Silica gel = 17.0g

PORT # POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	√Δp	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°F)	VACUUM (in. Hg)
							INLET	OUTLET				
3/1	0.00	0	—	0.76	872	2.30	58	—	200	55	125	—
2	4	4	720.6	0.95	975	2.80	60	—	200	60	125	2.0
3	12	8	723.2	1.00	1,000	3.00	62	—	200	60	125	2.0
4	12	12	726.2	0.95	975	2.80	63	—	200	60	125	2.0
4/1	0.00	0	—	6.63	794	1.90	64	—	200	55	125	2.5
2	4	4	729.6	0.76	872	2.40	65	—	200	55	125	3.0
3	8	8	738.2	0.83	911	2.50	67	—	195	55	125	3.0
4	12	12	741.9	0.85	902	2.60	69	—	195	55	125	3.5

PARTICULATE EMISSION DATA CALCULATION SHEET

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGH

Date 9/20/84 Run # 6

Sampling Location MAN;
LOESCHE MILL ON



1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (0.85) (0.93) \sqrt{\frac{659}{29.934(29.38)}}$$

$$V_s = \underline{58.50} \text{ fps}$$

Sampling Data	
$T_s = 659^\circ R$	$\bar{T}_s = 528^\circ R$
$C_p = 0.85$	$(\Delta p)_{avg} = 0.93$
$P_s = 29.934 \text{ in. Hg}$	$M_s = 29.38 \text{ g/mole}$
$A_s = 57.863 \text{ sq. ft.}$	$V_s = 54.458 \text{ cu. ft.}$
$H_2O \text{ (condensate)} = 160 \text{ mls}$	
$H_2O \text{ (silica gel)} = 17.0 \text{ gms}$	
$\Delta H_{avg} = 2.63 \text{ in H}_2O$	$P_b = 29.98 \text{ in Hg}$
Total particulate = 0.0420 grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls H}_2O}{\text{condensate} + \text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) \left(\frac{160 + 17}{177} \right) = (0.0474) (177) = \underline{8.39} \text{ cu. ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64) (54.458) \left(\frac{29.98 + \frac{2.63}{13.6}}{528} \right) = () (0.0571) = \underline{54.90} \text{ cu. ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{8.39}{54.90 + 8.39} = \underline{0.133} = \underline{13.3\%}$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - 0.133) (58.50) (57.863) (528/659) (29.934/29.92) = \underline{8468960} \text{ cu. ft.}$$

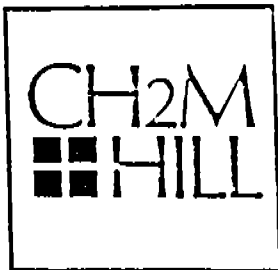
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (0.0420) / (54.90) = \underline{0.12}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (0.12) (8468960) (1.43 \times 10^{-4}) = \underline{14.30}$$



MONTGOMERY OFFICE

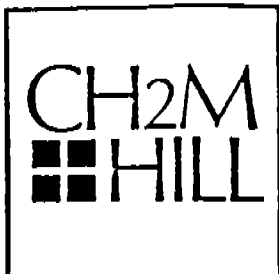
807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/634-2670Client LEHIGHDate 9/20/84 Run # 6Sampling Location MAIN;
LOESCHE MILL ONCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 13.3 %%CO₂ = 16.0% O₂ = 10.7%CO = —% N₂ = 73.3

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.133	x	1.0	x	18.0	=	2.39
CO ₂	.16	x	.867	x	44.0	=	6.10
O ₂	.107	x	.867	x	32.0	=	2.97
CO	—	x	—	x	28.0	=	—
N ₂	.733	x	.867	x	28.2	=	17.92

Average Molecular Wt., wet basis (M_s) = 29.38 lb/lb-molDensity of gas referred to air = $\frac{29.38}{28.95} = 1.01$ Gas density correction factor = $\sqrt{\frac{1.00}{1.01}} = \sqrt{0.985} = 0.99$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/20/84 Run# 6Sampling Location MAINLOESCHE MILL ONCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} \cdot ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(p_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$\frac{OV_s P_s A_n}{s s n}$$
Where I = % of isokinetic sampling T_s = Avg. stack temperature = 659°R V_1 = Total volume of liquid collected in impingers & silica gel = 177 ml V_m = Gas sample thru dry gas meter @ meter conditions = 54.458 cu.ft. T_m = Average dry gas meter temperature = 528 °R p_{bar} = Barometric pressure @ sampling site = 29.98 in.Hg ΔH = Average pressure drop across orifice = 2.63 in.H₂O t = Total sampling time = 64 min. V_s = Stack gas velocity = 58.50 fps P_s = Stack gas pressure = 29.934 in.Hg A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(659) (1.667) \left[(0.00267) (177) + \frac{54.458}{528} \left(29.98 + \frac{2.63}{13.6} \right) \right]}{(64) (58.50) (29.934) (0.000363)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) (\quad) \right]}{(\quad)}$$

$$I = \frac{(\quad) (0.4726 + 3.1121)}{(\quad)} = \frac{(\quad)}{(\quad)} =$$

$$I = \frac{(\quad)}{(\quad)} = \underline{96.30} \% \text{ of isokinetic sampling}$$

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA. .258 in.	FILTER NUMBER _____		
PITOT CORR. FACTOR .85	SILICA GEL NUMBER _____		
CRIFICE CORR. FACTOR 1.855			
	READING	GAS METER	IMPINGER COND.
	FINAL	834.867	375
	INITIAL	777.582	200
	NET	57.285	175

ORSAT READING	
% CO2	5.5
% O2	20.0
% CO	0
% N2	

ORSAT READING

% CO2 _____
% O2 58.4
% CO 6
% N2 _____

REMARKS: $T_g = 200$, $\% H_2O = 15$, $T_M = 80$
 $C = 0.86$, $\Delta P = 0.61$

PLANT

Lehigh

DATE

9/20/84

RUN NUMBER

7

TIME START

1024

TIME END

1133

MATERIAL PROCESSING RATE

LOESCHTE MILL ON

SAMPLING LOCATION

MAIN

WEATHER

SUNNY

AMBIENT TEMP.

72 °F

STACK PRESS.

-0.62 "H₂O

BAROMETRIC PRESS.

29.98 "Hg

WET BULB

DRY BULB

°F

MOISTURE

DEW POINT

°F

SKETCH:

SEE
PREVIOUS
RUNS

Silica gel = 13.5g

FINAL LEAK CHECK
SAMPLE TRAIN AT <u>10</u> in. Hg
FINAL <u>.032</u>
INITIAL <u>.015</u>
LEAK <u>.017</u> /min.
PITOT
IMPACT <u>0</u> at <u>3.2</u> in.
STATIC <u>0</u> at <u>4.4</u> in.

STACK INFORMATION

DIA. _____ in. AREA _____ ft²

DISTANCE FROM DOWNSTREAM _____ ft.

DISTURBANCE: UPSTREAM _____ ft.

PORT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	$\sqrt{\Delta p}$	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. -T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°C)	VACUUM (in. Hg)
							INLET	OUTLET				
2/1	<u>1024</u>	<u>0</u>	<u>—</u>	<u>0.83</u>	<u>.911</u>	<u>2.50</u>	<u>74</u>	<u>—</u>	<u>200</u>	<u>55</u>	<u>125</u>	<u>3.0</u>
2	<u>28</u>	<u>4</u>	<u>780.8</u>	<u>0.95</u>	<u>.975</u>	<u>2.80</u>	<u>75</u>	<u>—</u>	<u>200</u>	<u>60</u>	<u>125</u>	<u>3.0</u>
3	<u>32</u>	<u>8</u>	<u>784.6</u>	<u>1.00</u>	<u>1.00</u>	<u>2.95</u>	<u>78</u>	<u>—</u>	<u>200</u>	<u>60</u>	<u>125</u>	<u>3.5</u>
4	<u>36</u>	<u>12</u>	<u>788.5</u>	<u>0.98</u>	<u>.990</u>	<u>2.90</u>	<u>80</u>	<u>—</u>	<u>200</u>	<u>60</u>	<u>125</u>	<u>3.5</u>
	<u>1020</u>											
1/1	<u>1024</u>	<u>0</u>	<u>772.1</u>	<u>0.65</u>	<u>.806</u>	<u>1.95</u>	<u>80</u>	<u>—</u>	<u>200</u>	<u>60</u>	<u>125</u>	<u>2.5</u>
2	<u>28</u>	<u>4</u>	<u>776.0</u>	<u>0.83</u>	<u>.911</u>	<u>2.50</u>	<u>83</u>	<u>—</u>	<u>200</u>	<u>65</u>	<u>125</u>	<u>3.0</u>
3	<u>32</u>	<u>8</u>	<u>779.2</u>	<u>0.85</u>	<u>.922</u>	<u>2.55</u>	<u>84</u>	<u>—</u>	<u>200</u>	<u>65</u>	<u>125</u>	<u>3.0</u>
4	<u>36</u>	<u>12</u>	<u>802.7</u>	<u>0.90</u>	<u>.949</u>	<u>2.65</u>	<u>85</u>	<u>—</u>	<u>200</u>	<u>65</u>	<u>125</u>	<u>3.0</u>

9/20/84- 12m '7-

[illegible]

PARTICULATE EMISSION DATA CALCULATION SHEET

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGH

Date 9/20/34 Run # 7

Sampling Location MAIN,

LOESCHE MILL ON

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (0.86) (0.929) \sqrt{\frac{660}{29.934(29.34)}}$$

$$V_s = \underline{59.21} \text{ fps}$$

Sampling Data	
$F_s = 660$	$T_s = 544$
$C_p = 0.86$	$(\sqrt{\Delta p})_{avg} = 0.929$
$P_s = 29.934$	$M_s = 29.34$
$A_s = 57.863$	$V_s = 57.285$
$H_2O \text{ (condensate)} = 175$	als
$H_2O \text{ (silica gel)} = 13.5$	$grams$
$\Delta H_{avg} = 2.58$	$in H_2O; P_b = 29.98$
Total particulate = 0.0392 grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\begin{matrix} mls H_2O \\ \text{condensate} + \end{matrix} \begin{matrix} grams H_2O \\ \text{silica gel} \end{matrix} \right)$$

$$V_{wstd} = (0.0474) (175 + 13.5) = (0.0474) (188.5) = \underline{8.93} \text{ cu.ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64) (57.285) \left(\frac{29.98 + \frac{2.58}{13.6}}{544} \right) = () (0.0555) = \underline{56.04} \text{ cu.ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{8.93}{56.04 + 8.93} = \underline{0.137} = \underline{13.7} \%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - 0.137) (59.21) (57.863) (528 / 660) (29.934 / 29.92) = \underline{8519271} \text{ cu.ft.}$$

6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (0.0392) / (56.04) = \underline{0.01}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (0.01) (8519271) (1.43 \times 10^{-4}) = \underline{13.15}$$



MONTGOMERY OFFICE

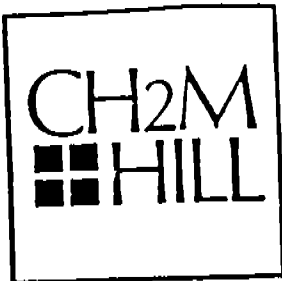
807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/20/84 Run # 7Sampling Location MAIN;
LOESCHE MILL ONCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 13.7 %%CO₂ = 16.0% O₂ = 10.7%CO = —% N₂ = 73.3

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.137	x	1.0	x	18.0	=	2.47
CO ₂	.16	x	.863	x	44.0	=	6.08
O ₂	.107	x	.863	x	32.0	=	2.95
CO	—	x	—	x	28.0	=	—
N ₂	.733	x	.863	x	28.2	=	17.84

Average Molecular Wt., wet basis (M_s) = 29.34 lb/lb-molDensity of gas referred to air = $\frac{29.34}{28.95} = 1.01$ Gas density correction factor = $\sqrt{\frac{1.00}{1.01}} = \sqrt{0.987} = 0.99$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/20/84 Run# 7Sampling Location MAIN;LOESCHE MILL ONCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} - ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$OV_s P_s A_n$$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 660 °R
 V_1 = Total volume of liquid collected in impingers & silica gel = 188.5 ml
 V_m = Gas sample thru dry gas meter @ meter conditions = 57.285 cu.ft.
 T_m = Average dry gas meter temperature = 544 °R
 P_{bar} = Barometric pressure @ sampling site = 29.98 in.Hg
 ΔH = Average pressure drop across orifice = 2.58 in.H₂O
 ϕ = Total sampling time = 64 min.
 V_s = Stack gas velocity = 59.21 fps
 P_s = Stack gas pressure = 29.934 in.Hg
 A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(660) (1.667) \left[(0.00267) (188.5) + \frac{57.285}{544} \left(29.98 + \frac{2.58}{13.6} \right) \right]}{(64) (59.21) (29.934) (0.000363)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) (\quad) \right]}{(\quad) (\quad) (\quad) (\quad)}$$

$$I = \frac{(\quad) (0.5033 + 3.1770)}{(\quad) (\quad) (\quad) (\quad)} = \frac{(\quad) (\quad) (\quad) (\quad)}{(\quad) (\quad) (\quad) (\quad)}$$

$$I = \frac{(\quad)}{(\quad)} = \underline{93.34} \% \text{ of isokinetic sampling}$$

LOESCHE MILL "OFF"

RUNS 2, 3, 4, 5

MONTGOMERY CRUISE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA.	250 in.	FILTER NUMBER	
PITOT CORR. FACTOR	.85	SILICA GEL NUMBER	
CRIFICE CORR. FACTOR	1.55		
READING	GAS METER	IMPINGER COND.	
FINAL	569.968	325	
INITIAL	525.797	200	
NET	44.171	125	

ORSAT READING

CO₂ 3.0
O₂ 14.0
CO 54.40
N₂ 28.60

REMARKS: $T_a = 70$, $T_s = 270$, $\%H_2O = 15$, $c = 0.82$
 $\Delta P = 0.66$

PLANT: LehighDATE: 9/18/84 RUN NUMBER: 2TIME START: 0742 TIME END: 0853MATERIAL PROCESSING RATE: LOESCHIG MILL OFFSAMPLING LOCATION: MAINWEATHER: V/CAMBIENT TEMP. 64 °F STACK PRESS. -0.1 "H₂O (29.88)BAROMETRIC PRESS. 29.44 "Hg WET BULBDRY BULB 64 °F MOISTURE ASSUME ±15%DEW POINT 64 °F from previous work

SKETCH:



Silica gel = 15-52

DIA. in.
AREA ft²

STACK INFORMATION

DISTANCE FROM

DISTURBANCE:

DOWNSTREAM ft.
UPSTREAM ft.

FINAL LEAK CHECK
SAMPLE TRAIN AT 8 in. Hg
FINAL -0.98
INITIAL .095
LEAK .003 / 0.06 min. sec.
PITOT .006 min. sec.
IMPACT 0 at 3 in.
STATIC 0 at 3 in.

PORT # POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-ΔP (in. H ₂ O)	√ΔP	ORIFICE VELOCITY HEAD-ΔP (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°F)	VACUUM (in. Hg)
							INLET	OUTLET				
3/1	1.12	0	—	0.58	762	1.55	63	—	270	60	130	—
2	1.12	4	528.3	0.58	762	1.55	64	—	270	60	135	—
3	1.12	8	531.1	0.62	787	1.68	65	—	270	60	140	1.5
4	1.12	12	533.9	0.62	787	1.68	66	—	270	60	140	2.0
4/1	1.12	0	536.9	0.42	648	1.18	67	—	270	60	140	1.5
2	1.12	4	540.0	0.50	707	1.35	67	—	270	60	140	1.5
3	1.12	8	542.4	0.58	762	1.58	67	—	265	60	140	1.5
4	1.12	12	544.7	0.58	762	1.58	68	—	265	60	140	1.5

PARTICULATE EMISSION DATA CALCULATION SHEET

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGH

Date 9/18/84 Run # 2

Sampling Location MAIN

LOESCHE MILL OFF

**CH2M
HILL**

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\Delta p)_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (0.85) (0.7583) \sqrt{\frac{727}{29.896(29.83)}}$$

$$V_s = \underline{449.752} \text{ fps}$$

Sampling Data	
$T_s = 727^\circ R$	$T_a = 528^\circ R$
$C_p = 0.85$	$(\Delta p)_{avg} = 0.7583$
$P_s = 29.896 \text{ in. Hg}$	$P_a = 29.83 \text{ in. Hg}$
$A_s = 57.863 \text{ sq. ft.}$	$V_s = 44.171 \text{ cu. ft.}$
$H_2O \text{ (condensate)} = 12.5 \text{ mls}$	
$H_2O \text{ (silica gel)} = 15.5 \text{ gms}$	
$\Delta H_{avg} = 1.565 \text{ in. H}_2\text{O}$	$P_b = 29.94 \text{ in. Hg}$
Total particulate = 0.0756 gram	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls H}_2\text{O}}{\text{condensate}} + \frac{\text{grams H}_2\text{O}}{\text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) (12.5 + 15.5) = (0.0474) (140.5) = \underline{6.66} \text{ cu. ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{T_m}}{T_m} \right)$$

$$V_{mstd} = (17.64) (44.171) \left(\frac{29.94 + \frac{1.565}{13.6}}{528} \right) = (\underline{\quad}) (0.0569) = \underline{44.35} \text{ cu. ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{6.66}{44.35 + 6.66} = \underline{0.1307} = \underline{13} \%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - 0.13) (49.79) (57.863) (528/727) (29.896/29.92) = \underline{6543110} \text{ cu. ft.}$$

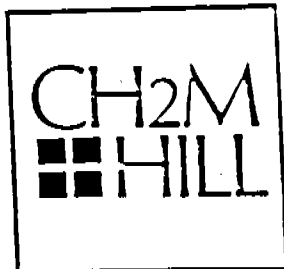
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (0.0756) / (44.35) = \underline{0.0263}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (\underline{6543110}) (0.0263) (1.43 \times 10^{-4}) = \underline{24.61}$$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/18/84 Run # 2Sampling Location MAIN,
LOESCHE MILL OFFCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

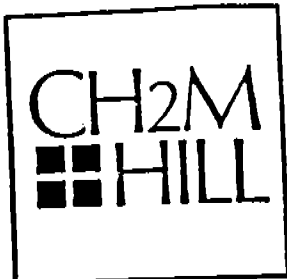
Moisture Content = 13 %

%CO₂ = 19
% O₂ = 10.5
%CO = —
% N₂ = 70.5

FROM
RUN #5

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	13/100	x	1.0	x	18.0	=	2.34
CO ₂	.19	x	.87	x	44.0	=	7.27 7.20
O ₂	.105	x	.87	x	32.0	=	2.92
CO	—	x	—	x	28.0	=	—
N ₂	70.5	x	.87	x	28.2	=	17.30 135

Average Molecular Wt., wet basis (M_s) = 29.83 lb/lb-molDensity of gas referred to air = $\frac{29.83}{28.95} = 1.03$ Gas density correction factor = $\sqrt{\frac{1.00}{1.03}} = \sqrt{0.97} = 0.985$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/18/84 Run# 2Sampling Location MAIN;LOESCHE MILL OFFCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} - ^\circ\text{R}} \right) v_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$\frac{O V_s P_s A_n}{s s n}$$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 727 °R V_1 = Total volume of liquid collected in impingers & silica gel = 140.5 ml V_m = Gas sample thru dry gas meter @ meter conditions = 44.171 cu.ft. T_m = Average dry gas meter temperature = 528 °R P_{bar} = Barometric pressure @ sampling site = 29.94 in.Hg ΔH = Average pressure drop across orifice = 1.565 in.H₂O t = Total sampling time = 64 min. V_s = Stack gas velocity = 49.752 fps P_s = Stack gas pressure = 29.896 in.Hg A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(727)(1.667) \left[(0.00267)(140.5) + \frac{44.171}{528} \left(29.94 + \frac{1.565}{13.6} \right) \right]}{(64)(49.752)(29.896)(0.000363)}$$

$$I = \frac{(1211.90) [(0.3751) + (0.0337)(32.0551)]}{(33.34)}$$

$$I = \frac{(1211.90)(0.3751 + 2.5143)}{(33.34)} = \frac{(1211.90)(2.8894)}{(33.34)} =$$

$$I = \frac{(1211.90)(2.8894)}{(33.34)} = \underline{101.3} \text{ \% of isokinetic sampling}$$

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA. .250 in. FILTER NUMBER _____
PITOT CORR. FACTOR .85 SILICA GEL NUMBER _____
ORIFICE CORR. FACTOR 1.85

ORSAT READING

% CO₂ _____
% O₂ _____
% CO _____
% N₂ _____

READING	GAS METER	IMPINGER COND.
FINAL	617.266	318
INITIAL	572.272	200
NET	44.994	118

REMARKS: T_c = 270, T_m = 70, %H₂O = 15c = 0.82ΔP = 0.66PLANT LehighDATE 9/18/84 RUN NUMBER 3TIME START 1012 TIME END 1122MATERIAL PROCESSING RATE LOLSCH MILL OFFSAMPLING LOCATION W4NWEATHER P/CAMBIENT TEMP. 70 °F STACK PRESS. -0.6 "H₂O (24.896) in. HgBAROMETRIC PRESS. 24.94 "Hg WET BULB _____ °F

DRY BULB _____ °F MOISTURE _____

DEW POINT _____ °F

SKETCH:

SEE
RUNS 1, 2

Silica gel = 15.0g

FINAL LEAK CHECK
SAMPLE TRAIN AT 10 in. Hg
FINAL .101
INITIAL .094
LEAK .007 / min. 30 sec
PITOT .014 / min.
IMPACTO at 3.2 in.
STATICO at 3.2 in.

STACK INFORMATION

DIA. _____ in. DISTANCE FROM DOWNSTREAM _____ ft.
AREA _____ ft² DISTURBANCE: _____ ft.

PORT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-ΔP (in. H ₂ O)	ORIFICE VELOCITY HEAD-ΔP (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. - T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°C)	VACUUM (in. Hg)
						INLET	OUTLET				
POINT #											
211	1012	0	—	0.54	0.735	70	70	265	60	140	—
2	1016	4	574.9	0.60	775	70	70	265	60	140	1.5
3	1020	8	577.8	0.62	787	71	71	265	60	140	2.0
4	1024	12	580.7	0.58	762	72	72	265	60	130	2.0
111	1000	6	563.6	0.47	686	73	73	265	65	130	2.0
2	1004	4	586.2	0.60	775	75	75	265	65	130	2.5
3	1008	8	588.9	0.58	762	76	76	270	65	130	2.5
4	1012	12	591.9	0.58	762	78	78	270	65	130	2.5

RUN-3- behind - 9/18

[illegible]

PARTICULATE EMISSION DATA CALCULATION SHEET

MONTGOMERY OFFICE

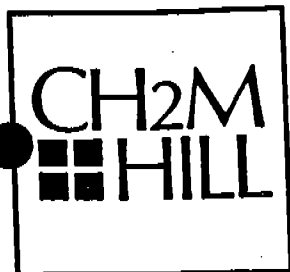
807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGH

Date 9/18/84 Run # 3

Sampling Location MAIN,

LOESCHE MILL OFF



1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (0.85) (0.75) \sqrt{\frac{728}{29.896 (29.89)}}$$

$$V_s = \underline{49.192} \text{ fps}$$

Sampling Data	
$\bar{T}_s = 728^\circ R$	$\bar{T}_a = 536^\circ R$
$C_p = 0.85$	$(\Delta p)_{avg} = 0.75$
$P_s = 29.896 \text{ in. Hg}$	$P_a = 29.89 \text{ in. Hg}$
$\lambda_s = 57.863 \text{ ft.}$	$V_s = 44.944 \text{ cu. ft.}$
$H_2O \text{ (condensate)} = 118 \text{ mls}$	
$H_2O \text{ (silica gel)} = 15.0 \text{ gms}$	
$\Delta H_{avg} = 1.64 \text{ in } H_2O$	$P_b = 29.94 \text{ in Hg}$
Total particulate = 0.0787 grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls } H_2O}{\text{condensate} + \text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) (118 + 15) = (0.0474) (133) = \underline{6.30} \text{ cu. ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{T_m}}{T_m} \right)$$

$$V_{mstd} = (17.64) (44.94) \left(\frac{29.94 + \frac{1.64}{13.6}}{536} \right) = (\underline{\quad}) (0.0561) = \underline{44.51} \text{ cu. ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{6.30}{44.51 + 6.30} = \underline{0.124} = \underline{12.4} \%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - 0.124) (44.94) (57.863) (528 / 728) (29.896 / 29.92) = \underline{6,505,131} \text{ cu. ft.}$$

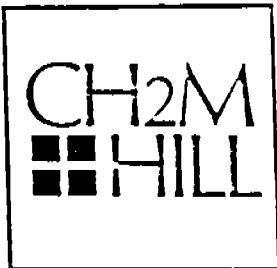
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (0.0787) / (44.51) = \underline{0.0273}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (0.0273) (6,505,131) (1.43 \times 10^{-4}) = \underline{25,40}$$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/634-2670Client LEHIGHDate 9/18/84 Run # 3Sampling Location MAINLOESCHE MILL OFFCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 12.4 %%CO₂ = 19% O₂ = 10.5% CO = —% N₂ = 70.5

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.124	x	1.0	x	18.0	=	2.23
CO ₂	.19	x	.875	x	44.0	=	7.32
O ₂	.105	x	.875	x	32.0	=	2.94
CO	—	x	—	x	28.0	=	—
N ₂	.705	x	.875	x	28.2	=	17.40

Average Molecular Wt., wet basis (M_s) = 29.89 lb/lb-molDensity of gas referred to air = $\frac{29.89}{28.95} = 1.03$ Gas density correction factor = $\sqrt{\frac{1.00}{1.03}} = \sqrt{0.97} = 0.98$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/18/84 Run# 3Sampling Location MAIN;LOESCHE MILL OFFCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} \cdot ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$0V_s P_s A_n$$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 728°R V_1 = Total volume of liquid collected in impingers & silica gel = 133 ml V_m = Gas sample thru dry gas meter @ meter conditions = 44.994 cu.ft. T_m = Average dry gas meter temperature = 536 °R P_{bar} = Barometric pressure @ sampling site = 29.94 in.Hg ΔH = Average pressure drop across orifice = 1.64 in.H₂O t = Total sampling time = 64 min. V_s = Stack gas velocity = 49.192 fps P_s = Stack gas pressure = 29.896 in.Hg A_n = Cross-sectional area of nozzle = .000363 sq.ft.

$$I = \frac{(728)(1.667) \left[(0.00267)(133) + \frac{44.994}{536} \left(29.94 + \frac{1.64}{13.6} \right) \right]}{(64)(49.192)(29.896)(.000363)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) (\quad) \right]}{(\quad)}$$

$$I = \frac{(\quad)(0.3551) + 2.5234}{(\quad)} = \frac{(\quad)(2.8785)}{(\quad)} =$$

$$I = \frac{(\quad)}{(\quad)} = \underline{102.2} \% \text{ of isokinetic sampling}$$

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

VMA
JCOB
9/18/84

PLANT Lehigh
DATE 9/18/84 RUN NUMBER 4
TIME START 1255 TIME END 1407
MATERIAL PROCESSING RATE
SAMPLING LOCATION Loeschte mill OFF
WEATHER Sunny
AMBIENT TEMP. 78 °F STACK PRESS. 0.6 "H₂O
BAROMETRIC PRESS. 29.94 "Hg WET BULB
DRY BULB — °F MOISTURE —
DEW POINT — °F

TEST CONDUCTED BY: Martin/Cooper

NOZZLE DIA. <u>2.58</u> in.		FILTER NUMBER	
PITOT CORR. FACTOR <u>.85</u>		SILICA GEL NUMBER	
CRIFICE CORR. FACTOR <u>1.855</u>			
READING	GAS METER	IMPINGER COND.	
FINAL	<u>664.307</u>	<u>330</u>	
INITIAL	<u>620.002</u>	<u>200</u>	
NET	<u>44.305</u>	<u>130</u>	

ORSAT READING

CO₂ 0.5
O₂ 0.5
CO 0.5
N₂ 0.5

REMARKS: T_g = 270, T_m = 90 % H₂O = 15%
c = 0.82
ΔP = 0.66

SKETCH:

see runs #2

Silica gel = 16.5g

FINAL LEAK CHECK
SAMPLE TRAIN AT 9 in. Hg
FINAL -042
INITIAL .033
LEAK .009 /min. 30 sec
PITOT 0.018 /min
IMPACT 8 at 4 in.
STATIC 0 at 39 in.

STACK INFORMATION

DIA. — in. DISTANCE FROM DOWNSTREAM — ft.
AREA — ft² DISTURBANCE: — ft.

PORT # POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	√Δp	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. -T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°C)	VACUUM (in. Hg)
							INLET	OUTLET				
3/1	13.5	0	—	0.44	0.663	1.28	78	78	270	60	130	—
2	12.5	4	622.3	0.55	0.742	1.55	78	78	270	60	130	—
3	13.0	8	624.9	0.58	0.762	1.65	80	80	270	60	130	—
4	13.07	12	627.9	0.55	0.742	1.55	82	82	270	60	130	—
4/1	13.3	0	620.7	0.40	0.633	1.15	83	83	270	60	130	—
2	13.7	4	622.7	0.51	0.714	1.45	83	83	270	60	130	—
3	14.1	8	625.9	0.53	0.728	1.52	83	83	270	60	130	—
4	14.5	12	628.4	0.53	0.728	1.52	83	83	270	60	130	—

PARTICULATE EMISSION DATA CALCULATION SHEET

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGHDate 9/18/84 Run # 4Sampling Location MAINLOESCHE MILL OFF

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (0.85) (0.73) \sqrt{\frac{730}{29.896(29.37)}}$$

$$V_s = \underline{48.368} \text{ fps}$$

Sampling Data	
$\bar{T}_s = 730$	$\bar{T}_a = 545$
$C_p = 0.85$	$(\sqrt{\Delta p})_{avg} = 0.73$
$P_s = 29.896$	$M_s = 29.37$
$A_s = 57.863$	$V_s = 44.305$
$H_2O \text{ (condensate)} = 130$	mls
$H_2O \text{ (silica gel)} = 16.5$	grams
$\Delta H_{avg} = 1.52$	$\text{in } H_2O; P_b = 29.94$
Total particulate = 0.0523 grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\begin{array}{l} \text{mls } H_2O \\ \text{condensate} \end{array} + \begin{array}{l} \text{grams } H_2O \\ \text{silica gel} \end{array} \right)$$

$$V_{wstd} = (0.0474) \left(\underline{130} + \underline{16.5} \right) = (0.0474) (\underline{146.5}) = \underline{6.94} \text{ cu.ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{T_m}}{T_m} \right)$$

$$V_{mstd} = (17.64) (\underline{44.305}) \left(\frac{29.94 + \frac{1.52}{545}}{545} \right) = (\underline{\quad}) (\underline{0.0551}) = \underline{43.09} \text{ cu.ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{6.94}{43.09 + 6.94} = \underline{0.139} = \underline{14} \%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - \underline{.14}) (\underline{48.368}) (57.86) (528 / \underline{730}) (\underline{29.896} / 29.92) = \underline{6,261.813} \text{ cu.ft.}$$

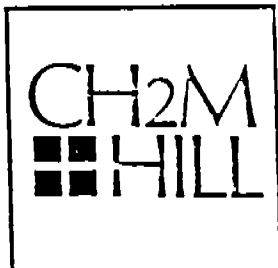
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (\underline{0.0523}) / (\underline{43.09}) = \underline{0.0187}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (\underline{6261.813}) (\underline{0.0187}) (1.43 \times 10^{-4}) = \underline{16.74}$$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/18/84 Run # 4Sampling Location MAIN;LOESCHE MILL OFFCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 14 %%CO₂ = 12.8% O₂ = 10.5%CO = —% N₂ = 70.7

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.14	x	1.0	x	18.0	=	2.52
CO ₂	.128	x	.86	x	44.0	=	6.81
O ₂	.105	x	.86	x	32.0	=	2.89
CO	—	x	—	x	28.0	=	—
N ₂	.707	x	.86	x	28.2	=	17.15

Average Molecular Wt., wet basis (M_s) = 29.37 lb/lb-molDensity of gas referred to air = $\frac{29.37}{28.95} = 1.01$ Gas density correction factor = $\sqrt{\frac{1.00}{1.01}} = \sqrt{0.986} = 0.99$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/13/84 Run# 4Sampling Location MAIN;LOESCHE MILL OFFCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} - ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$\frac{OV_s P_s A_n}{s s n}$$
Where I = % of isokinetic sampling T_s = Avg. stack temperature = 730 °R V_1 = Total volume of liquid collected in impingers & silica gel = 146.5 ml V_m = Gas sample thru dry gas meter @ meter conditions = 44.305 cu.ft. T_m = Average dry gas meter temperature = 545 °R P_{bar} = Barometric pressure @ sampling site = 29.94 in.Hg ΔH = Average pressure drop across orifice = 1.52 in.H₂O t = Total sampling time = 64 min. V_s = Stack gas velocity = 48.368 fps P_s = Stack gas pressure = 29.896 in.Hg A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(730) (1.667) \left[(0.00267) (146.5) + \frac{44.305}{545} \left(29.94 + \frac{1.52}{13.6} \right) \right]}{(64) (48.368) (29.896) (0.000363)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) (\quad) \right]}{(\quad) (\quad) (\quad) (\quad)}$$

$$I = \frac{(\quad) (0.3112 + 2.4439)}{(\quad) (\quad) (\quad) (\quad)} = \frac{(\quad) (\quad) (\quad) (\quad)}{(\quad) (\quad) (\quad) (\quad)} =$$

$$I = \frac{(\quad) (\quad) (\quad) (\quad)}{(\quad) (\quad) (\quad) (\quad)} = \underline{102.7} \% \text{ of isokinetic sampling}$$

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA. 0.258 in. FILTER NUMBER 4
 PITOT CORR. FACTOR 0.85 SILICA GEL NUMBER
 ORIFICE CORR. FACTOR 1.855

READING	GAS METER	IMPINGER COND.
FINAL	708.360	330
INITIAL	664.546	200
NET	43.814	130

ORSAT READING

% CO ₂	19.0	19.5
% O ₂	10.5	10.5
% CO		
% N ₂		

REMARKS:

SEE RUN#4

PLANT Lehigh

DATE 9/18/84 RUN NUMBER 5

TIME START 1510 TIME END 1621

MATERIAL PROCESSING RATE DESICHE MILL OFF

SAMPLING LOCATION MATIN

WEATHER

AMBIENT TEMP. 82 °F STACK PRESS. - 0.5 "H₂O

BAROMETRIC PRESS. 29.94 "Hg WET BULB

DRY BULB °F MOISTURE

DEW POINT °F

SKETCH:

SEE RUNS 1, 2

Silica gel = 14.0g

FINAL LEAK CHECK
 SAMPLE TRAIN AT 9 in. Hg
 FINAL .013
 INITIAL .000
 LEAK .013 /min.
 PITOT
 IMPACT 0 at 3.2 in.
 STATIC 0 at 3.8 in.

STACK INFORMATION

DIA. in. DISTANCE FROM DOWNSTREAM ft.
 AREA ft² DISTURBANCE: UPSTREAM ft.

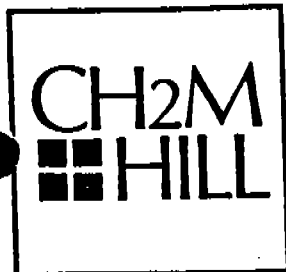
PORT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	√Δp	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)
POINT #						
2/1	1510	0	—	0.50	707	1.40
2	14	4	667.0	0.55	742	1.55
3	18	8	669.9	0.55	742	1.55
4	22	12	672.6	0.53	728	1.50
1/1	1529	0	675.4	0.43	656	1.25
2	33	4	678.0	0.52	721	1.47
3	37	8	680.7	0.53	728	1.50
4	41	12	683.4	0.52	721	1.47

GAS METER TEMP. T _m (°F)	STACK TEMP. -T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°C)	VACUUM (in. Hg)
INLET				
OUTLET				
83	270	60	125	—
84	270	60	125	—
85	270	60	125	—
87	270	60	125	2.0
			°C ↓	
88	270	60	125	1.5
89	270	60	125	2.0
90	270	60	125	2.0
92	270			2.0

35

1000

PARTICULATE EMISSION DATA CALCULATION SHEET



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2670

Plant LEHIGH

Date 9/18/84 Run # 5

Sampling Location MAIN;

LOESCHE MILL OFF

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (0.85) (0.717) \sqrt{\frac{730}{29.896(29.67)}}$$

$$V_s = \underline{47.266} \text{ fps}$$

Sampling Data	
$T_s = 730$	$P_s = 551$
$C_p = 0.85$	$(\Delta p)_{avg} = 0.717$
$P_s = 29.896$	$M_s = 29.67$
$A_s = 57.863$	$V_s = 43.814$
$H_2O \text{ (condensate)} = 130$	mls
$H_2O \text{ (silica gel)} = 14.0$	grams
$\Delta H_{avg} = 1.47$	$P_b = 29.94$
Total particulate = <u>0.0488</u> grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls } H_2O}{\text{condensate} + \text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) (\underline{130} + \underline{14}) = (0.0474) (\underline{144}) = \underline{6.83} \text{ cu.ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64) (\underline{43.814}) \left(\frac{29.94 + \frac{1.47}{13.6}}{551} \right) = (\underline{\quad}) (\underline{0.0545}) = \underline{42.15} \text{ cu.ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{6.83}{42.15 + 6.83} = \underline{0.1395} = \underline{14}\%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - \underline{.14}) (\underline{47.266}) (\underline{57.863}) (528 / \underline{730}) (\underline{29.896} / 29.92) = \underline{6,119,463} \text{ cu.ft.}$$

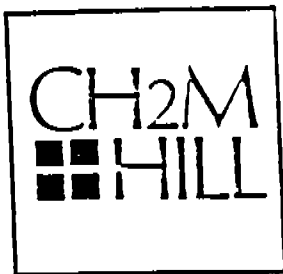
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (\underline{0.0488}) / (\underline{42.15}) = \underline{0.079}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (\underline{0.079}) (\underline{6,119,463}) (1.43 \times 10^{-4}) = \underline{15.66}$$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/18/84 Run # 5Sampling Location MANLOESCHE MILL OFFCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 14 %%CO₂ = 18.8% O₂ = 10.5%CO = —% N₂ = 70.7

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	14.0	x	1.0	x	18.0	=	2.52
CO ₂	18.8	x	.86	x	44.0	=	7.11
O ₂	10.5	x	.86	x	32.0	=	2.89
CO	—	x	—	x	28.0	=	—
N ₂	70.7	x	.86	x	28.2	=	17.15

Average Molecular Wt., wet basis (M_s) = 29.67 lb/lb-molDensity of gas referred to air = $\frac{29.67}{28.95} = 1.02$ Gas density correction factor = $\sqrt{\frac{1.00}{1.02}} = \sqrt{0.976} = 0.99$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/18/84 Run# 5Sampling Location MAINLOESCHE MILL OFFCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} - ^\circ\text{R}} \right) V_l + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$\frac{OV_s P_s A_n}{s s n}$$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 730 R
 V_l = Total volume of liquid collected in impingers & silica gel = 144 ml
 V_m = Gas sample thru dry gas meter @ meter conditions = 43.814 cu.ft.
 T_m = Average dry gas meter temperature = 551 °R
 P_{bar} = Barometric pressure @ sampling site = 29.94 in.Hg
 ΔH = Average pressure drop across orifice = 1.47 in.H₂O
 t = Total sampling time = 64 min.
 V_s = Stack gas velocity = 47.266 fps
 P_s = Stack gas pressure = 29.896 in.Hg
 A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(730)(1.667) \left[(0.00267)(144) + \frac{43.814}{551} \left(29.94 + \frac{1.47}{13.6} \right) \right]}{(64)(47.266)(29.896)(0.000363)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) (\quad) \right]}{(\quad) (\quad) (\quad) (\quad)}$$

$$I = \frac{(\quad) (0.3845 + 2.3893)}{(\quad) (\quad) (\quad) (\quad)} = \frac{(\quad) (\quad) (\quad) (\quad)}{(\quad) (\quad) (\quad) (\quad)}$$

$$I = \frac{(\quad)}{(\quad)} = \underline{102.8} \% \text{ of isokinetic sampling}$$

CLINKER COOLER

RUNS 8, 9, 10

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

VMA
9/24/84
VMA
JCOA

PLANT Lehigh RUN NUMBER 188
DATE 9/24/84 TIME END 0724
TIME START 0813
MATERIAL PROCESSING RATE
SAMPLING LOCATION Clincher Coaker
WEATHER HAZY
AMBIENT TEMP. (WET) 70 °F STACK PRESS. -0.22 "H₂O "H₂O
BAROMETRIC PRESS. 29.96 "Hg WET BULB 11.0 (26.6) °F
DRY BULB 38.0 °F MOISTURE ASSUMED 4%
DEW POINT 38.0 °F PREVIOUS DATA

TEST CONDUCTED BY: Martin/Cooper

NOZZLE DIA. 2.58 in. FILTER NUMBER 45
PITOT CORR. FACTOR .85 SILICA GEL NUMBER
ORIFICE CORR. FACTOR 1.85

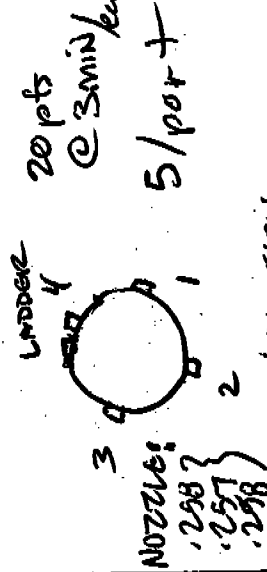
ORSAT READING

CO₂ 14.2
O₂ 84.9
CO 0.9
N₂ 0.0

READING	GAS METER	IMPINGER COND.
FINAL	882.039	215
INITIAL	854.095	200
NET	27.944	15

REMARKS: $T_s = 350$, $\%H_2O = 4$, $T_m = 110$
 $C = 1.10$, $\Delta P = 0.54$

SKETCH:



FINAL LEAK CHECK
SAMPLE TRAIN AT 9 in. Hg
FINAL .052
INITIAL .042
LEAK .010/min.
PITOT
IMPACT 0 at 3.6 in.
STATIC 0 at 3.2 in.

LOW FLOW STACK INFORMATION

DIA. in. 42 DISTANCE FROM DOWNSTREAM ft.
AREA ft² 0.1-0.2 DISTURBANCE: Silica gel = 12.0 g

PORT # POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	√Δp	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)	GAS METER TEMP. (°F)		STACK TEMP. (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°F)	VACUUM (in. Hg)
							INLET	OUTLET				
1/1	41.9	0	—	0.13	361	0.46	97	97	350	60	125	—
2	17	3	855.4	0.15	387	0.53	97	97	360	60	125	—
3	20	6	856.1	0.17	412	0.59	98	98	365	60	125	—
4	23	9	858.3	0.20	447	0.70	98	98	370	60	125	—
5	26	12	859.8	0.22	447	0.70	98	98	375	65	125	—
2/1	46.9	0	861.1	0.13	361	0.46	99	99	355	65	130	—
2	48.36	3	862.4	0.12	346	0.41	103	103	370	65	130	—
3	48.9	6	863.7	0.07	265	0.25	104	104	380	65	125	—

PARTICULATE EMISSION DATA CALCULATION SHEET

MONTGOMERY OFFICE



807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGH

Date 9/21/84 Run # 1

Sampling Location CLINKER COOLER

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (\frac{410}{0.45}) (0.415) \sqrt{\frac{787}{29.944(27.97)}}$$

$$V_s = \frac{37.83}{29.23} \text{ fps } \text{Dm}$$

Sampling Data	
$\bar{T}_s = 737^\circ R$	$\bar{T}_a = 568^\circ R$
$C_p = \frac{410}{0.45}$	$(\bar{K} \bar{p})_{avg} = 0.415$
$P_s = 29.944$	$P_b = 29.96$ in Hg; $P_a = 27.97$ in Hg
$\lambda_s = 68.417$ sq.ft.	$\lambda_a = 27.944$ sq.ft.
H_2O (condensate) = 15	mls
H_2O (silica gel) = 12	grams
$\Delta H_{avg} = 0.619$ in H_2O	$P_b = 29.96$ in Hg
Total particulate = 0.1892 grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls } H_2O}{\text{condensate} + \text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) (\frac{15 + 12}{27}) = (0.0474) (27) = 1.28 \text{ cu.ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64) (27.944) \left(\frac{29.96 + \frac{0.619}{13.6}}{568} \right) = (\text{---}) (0.0528) = 26.04 \text{ cu.ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{1.28}{26.04 + 1.28} = 0.047 = 4.7\%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - 0.047) (\frac{37.83}{29.23}) (68.417) (528/787) (29.944/29.92) = 4606761 \text{ cu.ft./1}$$

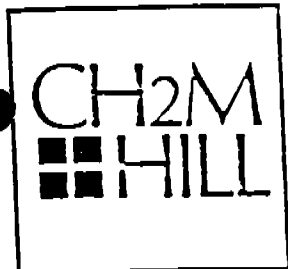
6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (0.1892) / 26.04 = 0.112$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (\frac{5962154}{4606761}) (0.112) (1.43 \times 10^{-4}) = 73.85 \text{ Dm}$$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/634-2670Client LEHIGHDate 9/21/84 Run # 1Sampling Location CLINKER COOLERCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

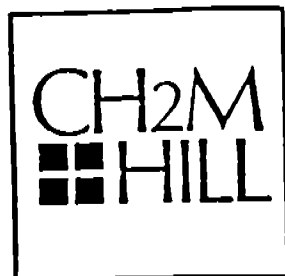
Average Orsat Analysis (dry)

Moisture Content = 4.7 %

%CO₂ = 0.7
% O₂ = 4.0
%CO = —
% N₂ = 95.3

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.047	x	1.0	x	18.0	=	0.85
CO ₂	0.007	x	.953	x	44.0	=	0.29
O ₂	0.04	x	.953	x	32.0	=	1.22
CO	—	x	—	x	28.0	=	—
N ₂	95.3	x	.953	x	28.2	=	25.61

Average Molecular Wt., wet basis (M_s) = 27.97 lb/lb-mole✓ Density of gas referred to air = $\frac{27.97}{28.95} = 0.966$ ✓ Gas density correction factor = $\sqrt{\frac{1.00}{0.966}} = \sqrt{1.035} = 1.02$



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MONTGOMERY, ALABAMA 36104
205/834-2870

Client LEHIGH

Date 9/21/84 Run# 1

Sampling Location CLINKER COOLER

CALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu. ft.}}{\text{ml} - ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$OV_s P_s A_n$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 787°R
 V_1 = Total volume of liquid collected in impingers & silica gel = 27 ml
 V_m = Gas sample thru dry gas meter @ meter conditions = 27.944 cu.ft.
 T_m = Average dry gas meter temperature = 568 °R
 P_{bar} = Barometric pressure @ sampling site = 29.96 in.Hg
 ΔH = Average pressure drop across orifice = 0.619 in.H₂O
 t = Total sampling time = 60 min.
 V_s = Stack gas velocity = 29.23 37.83 fps
 P_s = Stack gas pressure = 29.944 in.Hg
 A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(787) (1.667) \left[(0.00267) (27) + \frac{27.944}{568} \left(29.96 + \frac{0.619}{13.6} \right) \right]}{(60) (37.83) (29.944) (0.000363)}$$

$$I = \frac{() \left[() + () \right]}{() () () ()}$$

$$I = \frac{() (0.0721 + 1.4762)}{() () () ()} = \frac{() () () ()}{() () () ()}$$

$$I = \frac{()}{()} = \frac{82.33}{106.6} \% \text{ of isokinetic sampling}$$

106.6 gpm

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martin Cooper

NOZZLE DIA. <u>1.258</u> in.	FILTER NUMBER _____		
PITOT CORR. FACTOR <u>.85</u>	SILICA GEL NUMBER _____		
CRIFICE CORR. FACTOR <u>1.053</u>			
	READING	GAS METER	IMPINGER COND.
	FINAL	<u>913.619</u>	<u>210</u>
	INITIAL	<u>882.152</u>	<u>200</u>
	NET	<u>31.467</u>	<u>10</u>

ORSAT READING	
% CO ₂ <u>2</u>	_____
% O ₂ <u>44.6</u>	_____
% CO <u>44.6</u>	_____
% N ₂ _____	_____

ORSAT READING

CO ₂	_____
O ₂	_____
CO	_____
N ₂	_____

REMARKS: SEE RUN 1

VMA
SCDA
9/24/84

PLANT Lehigh RUN NUMBER 2-19
DATE 9/21/84 TIME END _____
TIME START 1010
MATERIAL PROCESSING RATE _____
SAMPLING LOCATION CLINKER COOLER
WEATHER Hazy
AMBIENT TEMP. 86 °F STACK PRESS. -0.22 "H₂O"
BAROMETRIC PRESS. 29.96 "Hg WET BULB _____
DRY BULB _____ °F
DEW POINT _____ °F

SKETCH:

SEE RUN 1

Silica gel = 13.5g

(Higher Flow / T_g in / ft² than RUN #1)

STACK INFORMATION
DISTANCE FROM DOWNSTREAM _____ ft.
DISTURBANCE: _____ ft.

FINAL LEAK CHECK
SAMPLE TRAIN AT 12 in. Hg
FINAL .036
INITIAL .030
LEAK .006 / min.
PITOT
IMPACT 0 at 3.8 in.
STATIC 0 at 5 in.

PORT # POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)	GAS METER TEMP. (°F)		STACK TEMP. -T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°F)	VACUUM (in. Hg)
						INLET	OUTLET				
1	100	0		0.23	0.75	93	93	420	60	125	—
2	150	3	883.5	0.30	0.95	93	93	420	60	125	—
3	200	6	885.2	0.32	1.02	93	93	420	60	125	—
4	250	9	887.2	0.32	1.02	93	93	425	65	125	—
5	300	12	888.9	0.25	0.78	93	93	430	65	130	—
6	350	0	890.6	0.37	1.15	95	95	420	65	125	—
7	400	3	892.7	0.41	1.30	95	95	415	65	125	—
8	450	6	894.8	0.45	1.45	96	96	410	65	125	2.0
9	500	9	897.1	0.43	1.35	97	97	410	65	125	2.0
10	550					90	90	410	65	125	2.0

RUN 2 - Cooler - 9121

$T_M = 100$, $T_s = 420$, $\%H_2O = 4$, $C = 1.13$, $DP = 0.660$

POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD- P (in. H ₂ O)	$\sqrt{\Delta P}$	ORIFICE VELOCITY HEAD- II (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. - T _s (°F)	IMPINGER TEMP. (°F) ICE	BOX TEMP. (°F)	VACUUM (in. Hg)
							INLET	OUTLET				
201	1049	0	900.9	0.16	1400	0.51	97	97	410	65	130	2.0
2	52	3	902.3	0.16	1400	0.51	98	98	410	65	125	2.0
3	55	6	903.6	0.18	1424	0.57	98	98	410	60	125	2.0
4	58	9	905.1	0.16	1400	0.51	98	98	410	60	125	2.0
5	101	12	906.3	0.13	961	0.42	98	98	410	60	125	2.0
101	104	0	907.6	0.15	987	0.48	97	97	440	60	125	2.0
2	105	3	909.0	0.13	361	0.42	97	97	445	65	125	2.0
3	12	6	910.1	0.13	361	0.42	97	97	450	65	125	2.0
4	15	9	911.2	0.13	387	0.42	97	97	450	65	125	2.0
5	18	12	912.4	0.13	387	0.42	97	97	450	65	125	2.0
	1121				9.7280	15.7000	19.20.00		8465.00			
	OFF				0.4864	0.7850	96.00		423.25			
							554.0000		883.25			

PARTICULATE EMISSION DATA CALCULATION SHEET



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant Vehigh
Date 9/21/84 Run # 2(9)
Sampling Location Clinker
Cooler

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 (.85) (.1486) \sqrt{\frac{883}{(29.944)(28.085)}}$$

$$V_s = \underline{36.188} \text{ fps}$$

Sampling Data	
$F_s = 883$	$F_a = 556$
$C_p = .85$	$(\Delta p)_{avg} = .1486$
$P_s = 29.944$	$M_s = 28.085$
$A_s = 68.417$	$V_s = 31.467$
H_2O (condensate) =	10 mls
H_2O (silica gel) =	13.5 gms
$\Delta H_{avg} = 1.785$	$P_b = 29.96$
Total particulate = 0.1028 grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\frac{\text{mls } H_2O}{\text{condensate} + \text{silica gel}} \right)$$

$$V_{wstd} = (0.0474) \left(\frac{10 + 13.5}{23.5} \right) = \underline{1.11} \text{ cu.ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64)(V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64)(31.467) \left(\frac{29.96 + \frac{1.785}{13.6}}{556} \right) = (1)(.0540) = \underline{29.970} \text{ cu.ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{1.11}{29.97 + 1.11} = \underline{0.0357} = \underline{3.6}\%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600(1 - B_{ws})V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600(1 - .036)(36.188)(68.417)(528/883)(29.944/29.92) = \underline{5,141,971} \text{ cu.ft./}$$

6. Calculate grain loading--

$$\text{grains/sdcf} = (15.43)(\text{grams})/V_{mstd} = (15.43)(.1028)/(29.97) = \underline{0.0529}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf})(Q_{stpd})(1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (.0529)(5,141,971)(1.43 \times 10^{-4}) = \underline{38.92}$$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client Lehigh
Date 9-21 Run # (9) 2
Sampling Location ClinkerCALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

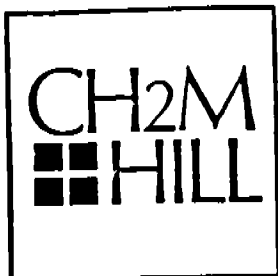
Average Orsat Analysis (dry)

Moisture Content = 3.6 %

%CO₂ = 0.7
% O₂ = 4.0
%CO = -
% N₂ = 95.3

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	1.036	x	1.0	x	18.0	=	1.648
CO ₂	1.007	x	.964	x	44.0	=	.297
O ₂	1.040	x	.964	x	32.0	=	.233
CO	-	x	-	x	28.0	=	-
N ₂	1.953	x	.964	x	28.2	=	25.907

Average Molecular Wt., wet basis (M_s) = 28.085 lb/lb-moDensity of gas referred to air = 28.95 =Gas density correction factor = $\sqrt{\frac{1.00}{28.95}}$ = $\sqrt{\quad}$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LehighDate 9/21/81 Run# 2 (9)Sampling Location ClinchesterCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} - ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

$$\frac{OV_s P_s A_n}{s s n}$$

Where I = % of isokinetic sampling T_s = Avg. stack temperature = 883 °R
 V_1 = Total volume of liquid collected in impingers & silica gel = 23.5 ml
 V_m = Gas sample thru dry gas meter @ meter conditions = 31.467 cu.ft.
 T_m = Average dry gas meter temperature = 556 °R
 P_{bar} = Barometric pressure @ sampling site = 29.96 in.Hg
 ΔH = Average pressure drop across orifice = 0.785 in.H₂O
 t = Total sampling time = 60 min.
 V_s = Stack gas velocity = 36.188 fps
 P_s = Stack gas pressure = 29.944 in.Hg
 A_n = Cross-sectional area of nozzle = 0.000363 sq.ft.

$$I = \frac{(883) (1.667) \left[(0.00267) (23.5) + \frac{31.467}{556} \left(29.96 + \frac{.785}{13.6} \right) \right]}{(60) (36.188) (29.944) (0.000363)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) (\quad) \right]}{(\quad) (\quad) (\quad) (\quad)}$$

$$I = \frac{(\quad) \left[(\quad) + (\quad) \right]}{(\quad) (\quad) (\quad) (\quad)} = \frac{(\quad) (\quad)}{(\quad) (\quad)} = \underline{\hspace{2cm}}$$

$$I = \frac{109.86}{\underline{\hspace{1cm}}} = \underline{109.9} \text{ \% of isokinetic sampling}$$

MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870 A DIVISION OF CH2M HILL

CH2M HILL

TEST CONDUCTED BY: Martinez/Cooper

NOZZLE DIA. 0.256 in. FILTER NUMBER _____
 PITOT CORR. FACTOR 0.85 SILICA GEL NUMBER _____
 CRIFICE CORR. FACTOR 0.855

READING	GAS METER	IMPINGER COND.
FINAL	949.428	215
INITIAL	913.707	200
NET	35.721	15

ORSAT READING

O ₂	10	0.0	1.0
CO	10	0.0	4.0
N ₂	2		

REMARKS: T_s = 360, T_A = 100, 0/0420 = 4, 0.1.10
ΔP = 0.54

Vmt
9/24/84
JCOA

PLANT Lehigh

DATE 9/21/84 RUN NUMBER 3 (10)
 TIME START 1200 TIME END 1307
 MATERIAL PROCESSING RATE _____
 SAMPLING LOCATION CLinker Cooler
 WEATHER Sunny
 AMBIENT TEMP. 86 °F STACK PRESS. -0.22 "H₂O"
 BAROMETRIC PRESS. 29.96 "Hg WET BULB _____
 DRY BULB _____ °F MOISTURE _____
 DEW POINT _____ °F

SKETCH:

FINAL LEAK CHECK
 SAMPLE TRAIN AT 12 in. Hg
 FINAL .056
 INITIAL .042
 LEAK .014 /min.
 PITOT _____
 IMPACT @ 49 in.
 STATICO @ 44 in.

STACK INFORMATION
 DISTANCE FROM DOWNSTREAM _____ ft.
 DISTURBANCE: _____ ft.

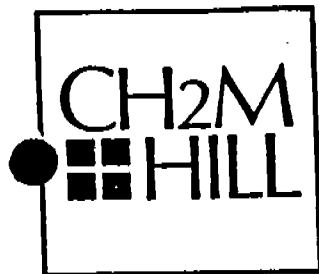
DIA. _____ in.
 AREA _____ ft²

SEE RUN 1
 Silica gel = 12.0g

PORT # POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD-Δp (in. H ₂ O)	ORIFICE VELOCITY HEAD-Δp (in. H ₂ O)	GAS METER TEMP. T _m (°F)		STACK TEMP. -T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. 65	VACUUM (in. Hg)
						INLET	OUTLET				
1/1	100	0	✓	0.18	0.61	93		370	55	125	—
2	03	3	914.9	0.18	0.61	93		390	60	125	—
3	06	6	916.3	0.18	0.61	93		390	60	125	—
4	09	9	917.7	0.20	0.69	93		400	60	125	—
5	12	12	919.2	0.20	0.69	93		400	60	125	—
2/1	17	0	920.7	0.25	0.84	93		370	65	125	—
2	20	3	922.4	0.25	0.84	94		370	65	125	—
3	23	6	923.9	0.24	0.82	94		360	65	125	—
4	26	9	925.8	0.25	0.84	94		340	65	125	—
						94		210	105	125	—

POINT #	DISTANCE FROM PORT (in.)	TIME	GAS METER VOLUME (ft ³)	STACK VELOCITY HEAD- P (in. H ₂ O)	$\sqrt{\Delta P}$	ORIFICE VELOCITY HEAD- H (in. H ₂ O)	GAS METER TEMP.		STACK TEMP. - T _s (°F)	IMPINGER TEMP. (°F)	BOX TEMP. (°C)	VACUUM (in. Hg)
							T _m (°F)					
							INLET	OUTLET				
1/1	120.30	0	929.0	0.32	.566	1.10	95		340	65	125	—
2	37	3	930.7	0.37	.608	1.30	95		330	65	125	2.0
3	40	6	932.7	0.39	.625	1.38	95		335	65	125	2.0
4	43	9	935.1	0.35	.592	1.22	96		350	60	125	2.0
5	46	12	936.7	0.32	.566	1.10	96		370	60	125	2.0
4/1	48.31	0	938.5	0.37	.608	1.30	95		395	60	125	2.0
2	53	3	940.6	0.42	.648	1.45	96		395	65	125	2.5
3	58	6	943.1	0.50	.707	1.75	96		395	65	125	3.0
4	1301	9	945.1	0.42	.648	1.45	97		395	65	125	3.0
5	1304	12	947.3	0.39	.625	1.38	97		400	65	125	3.0
				0.7900	10.7900	20.6800	1392		735.00			
	0.5395			0.5395	0.5395	1.0340	946000		371.7500			
	1307						5546000		931.7500			

PARTICULATE EMISSION DATA CALCULATION SHEET



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Plant LEHIGH
Date 9/21/84 Run # 3
Sampling Location CLINKER COOLER

1. Calculate Stack gas velocity--

$$V_s = 85.49 C_p (\sqrt{A_p})_{avg} \sqrt{\frac{T_s}{P_s M_s}}$$

$$V_s = 85.49 \left(\frac{110}{100} \right) (0.54) \sqrt{\frac{832}{29.944(28.09)}}$$

$$V_s = \underline{50.50} \text{ fps}$$

39.023

Sampling Data	
$T_s = 832^\circ R$	$T_m = 555^\circ R$
$C_p = \frac{110}{100}$	$(\sqrt{A_p})_{avg} = 0.54$
$P_s = 29.944 \text{ in. Hg}$	$M_s = 28.09 \text{ g/g-mole}$
$A_s = 68.417 \text{ sq. ft.}$	$V_s = 35.721 \text{ cu. ft.}$
$H_2O \text{ (condensate)} = 15 \text{ mls}$	
$H_2O \text{ (silica gel)} = 12 \text{ gms}$	
$\Delta H_{avg} = 1.034 \text{ in H}_2O$	$P_b = 29.96 \text{ in H}_2O$
Total particulate = <u>0.1310</u> grams	

2. Calculate volume of water vapor in gas sample--

$$V_{wstd} = (0.0474) \left(\begin{matrix} \text{mls H}_2O \\ \text{condensate} \end{matrix} + \begin{matrix} \text{grams H}_2O \\ \text{silica gel} \end{matrix} \right)$$

$$V_{wstd} = (0.0474) (\underline{15} + \underline{12}) = (0.0474) (\underline{27}) = \underline{1.28} \text{ cu. ft.}$$

3. Calculate gas sample volume at standard conditions--

$$V_{mstd} = (17.64) (V_m) \left(\frac{P_b + \frac{\Delta H}{13.6}}{T_m} \right)$$

$$V_{mstd} = (17.64) (\underline{35.721}) \left(\frac{29.96 + \frac{1.034}{13.6}}{\underline{555}} \right) = (\underline{\quad}) (0.0541) = \underline{34.10} \text{ cu. ft.}$$

4. Calculate percent moisture in gas stream--

$$B_{ws} = \frac{V_{wstd}}{V_{mstd} + V_{wstd}} = \frac{1.28}{34.10 + 1.28} = \underline{0.0362} = \underline{3.6}\%$$

5. Calculate stack gas volumetric flow rate, dry basis, std. conditions--

$$Q_{stpd} = 3600 (1 - B_{ws}) V_s A_s \frac{528}{T_s} \frac{P_s}{29.92}$$

$$Q_{stpd} = 3600 (1 - \underline{0.036}) (\underline{50.5}) (\underline{68.417}) \left(\frac{528}{\underline{832}} \right) \left(\frac{29.944}{29.92} \right) = \underline{765418} \text{ cu. ft.}$$

5,884,683

6. Calculate grain loading--

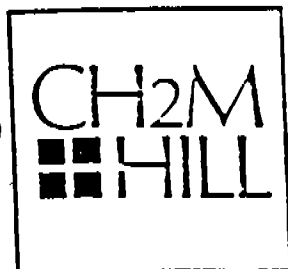
$$\text{grains/sdcf} = (15.43) (\text{grams}) / V_{mstd} = (15.43) (\underline{0.1310}) / (\underline{34.10}) = \underline{0.059}$$

7. Calculate mass emission rate--

$$\text{lbs/hr} = (\text{grains/sdcf}) (Q_{stpd}) (1.43 \times 10^{-4})$$

$$\text{lbs/hr} = (\underline{765418}) (\underline{0.059}) (1.43 \times 10^{-4}) = \underline{64.55}$$

5,884,683 49.90



MONTGOMERY OFFICE
807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870

Client LEHIGH
Date 9/21/84 Run # 3
Sampling Location CLINKER COOL

CALCULATION OF STACK GAS MOLECULAR WEIGHT (M_s)

Average Orsat Analysis (dry)

Moisture Content = 3.6 %

% CO₂ = 0.7

% O₂ = 4.0

% CO = —

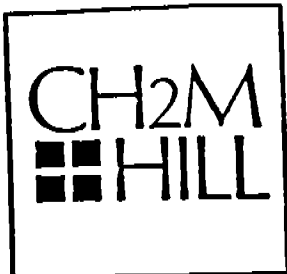
% N₂ = 95.3

Component	% Vol/100	X	Moisture Correction	X	Molecular Weight	=	Wt./Mole (wet base)
H ₂ O	.036	x	1.0	x	18.0	=	0.65
CO ₂	.007	x	.964	x	44.0	=	0.30
O ₂	.04	x	.964	x	32.0	=	1.23
CO	—	x	—	x	28.0	=	—
N ₂	.953	x	.964	x	28.2	=	25.91

Average Molecular Wt., wet basis (M_s) = 28.09 lb/lb-mo

Density of gas referred to air = $\frac{28.09}{28.95} = 0.97$

Gas density correction factor = $\sqrt{\frac{1.00}{0.97}} = \sqrt{1.03} = 1.015$



MONTGOMERY OFFICE

807 SOUTH McDONOUGH STREET
MONTGOMERY, ALABAMA 36104
205/834-2870Client LEHIGHDate 9/21/84 Run# 3Sampling Location CLINKER COOLERCALCULATION OF ISOKINETIC RATE

$$I = T_s \left(1.667 \frac{\text{min}}{\text{sec}} \right) \left[\left(0.00267 \frac{\text{in. Hg-cu.ft.}}{\text{ml} \cdot ^\circ\text{R}} \right) V_1 + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\Delta H}{13.6} \right) \right]$$

 $OV_s P_s A_n$ Where I = % of isokinetic sampling T_s = Avg. stack temperature = 832 °R V_1 = Total volume of liquid collected in impingers & silica gel = 27 ml V_m = Gas sample thru dry gas meter @ meter conditions = 35.721 cu.ft. T_m = Average dry gas meter temperature = 555 °R P_{bar} = Barometric pressure @ sampling site = 29.96 in.Hg ΔH = Average pressure drop across orifice = 1.034 in.H₂O t = Total sampling time = 60 min. V_s = Stack gas velocity39.023 = 50.50 ~~fps~~ ²⁰⁰ P_s = Stack gas pressure= 29.944 in.Hg A_n = Cross-sectional area of nozzle= 0.000363 sq.ft.

$$I = \frac{(832) (1.667) \left[(0.00267) (27) + \frac{35.721}{555} \left(29.96 + \frac{1.034}{13.6} \right) \right]}{(60) (50.50) (29.944) (0.000363)}$$

$$I = \frac{() \left[() + () \right]}{() () () ()}$$

$$I = \frac{() (0.0721 + 1.4332)}{() () () ()} = \frac{() () () ()}{() () () ()}$$

$$I = \frac{1}{1} = \frac{84.44}{109.3} \text{ \% of isokinetic sampling}$$

109.3 ^{Don}

APPENDIX C
QUALITY ASSURANCE

METER BOX CALIBRATION DATA AND CALCULATION FORM

(English units)

Date 3/1/24

Meter box number Nutech

Barometric pressure, $P_b = 30.03$ in. Hg Calibrated by D. Martin

Orifice manometer setting (ΔH), in. H_2O	Gas volume		Temperature				Time (θ), min	Y_i	$\Delta H \theta_i$ in. H_2O
	Wet test meter (V_w), ft ³	Dry gas meter (V_d), ft ³	Wet test meter (t_w), °F	Dry gas meter					
				Inlet (t_{d_i}), °F	Outlet (t_{d_o}), °F	Avg ^a (t_d), °F			
0.5	5	222.207 227.285	62	63/69	66	12:52	.991	1.811	
1.0	86	221.971 234.100	62	71/75	73	11:03	.997	1.830	
1.5	10	234.551 244.882	62	75/82	78.5	13:13	.995	1.856	
2.0	10	245.262 255.700	62	82/86	84.0	13:17	.994	1.866	
3.0	10	256.040 266.574	62	87/90	88.5	10:58	.990	1.876	
4.0	10	266.958 277.438	62	90/95	92.5	9:32	1.000	1.891	
							Avg	.995	1.855

ΔH , in. H_2O	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)}$	$\Delta H \theta_i = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[\frac{(t_w + 460) \theta}{V_w} \right]^2$
0.5	0.0368	$\frac{5(30.03)(526)}{5.376(30.0662)(522)} = .991$	$\frac{0.0317(0.5)}{30.03(526)} \left[\frac{522(12.867)}{5} \right]^2 = 1.811$
1.0	0.0737	$\frac{6(30.03)(533)}{6.129(30.1031)(522)} = .997$	$\frac{0.0317(1.0)}{30.03(533)} \left[\frac{522(11.05)}{6} \right]^2 = 1.830$
1.5	0.110	$\frac{10(30.03)(538.5)}{10.331(30.14)(522)} = .995$	$\frac{0.0317(1.5)}{30.03(538.5)} \left[\frac{522(15.22)}{10} \right]^2 = 1.856$
2.0	0.147	$\frac{10(30.03)(544)}{10.326(30.177)(522)} = .994$	$\frac{0.0317(2.0)}{30.03(544)} \left[\frac{522(13.282)}{10} \right]^2 = 1.866$
3.0	0.221	$\frac{10(30.03)(548.5)}{10.534(30.251)(522)} = .9902$	$\frac{0.0317(3.0)}{30.03(548.5)} \left[\frac{522(10.972)}{10} \right]^2 = 1.876$
4.0	0.294	$\frac{10(30.03)(552.5)}{10.412(30.324)(522)} = 1.000$	$\frac{0.0317(4.0)}{30.03(552.5)} \left[\frac{522(9.33)}{10} \right]^2 = 1.891$

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d .

POSTTEST DRY GAS METER CALIBRATION DATA FORM (English units)

Test number 1-10 Date 10/26/84 Meter box number Nutech Plant L-6.9h
 Barometric pressure, $P_b = 30.01$ in. Hg Dry gas meter number Nutech Pretest $Y = 0.99$

Orifice manometer setting, (ΔH), in. H_2O	Gas volume		Temperature				Time (θ), min	Vacuum setting, in. Hg	Y_i	Y_i	
	Wet test meter (V_w), ft^3	Dry gas meter (V_d), ft^3	Wet test meter (t_w), $^{\circ}F$	Inlet (t_{d_i}), $^{\circ}F$	Outlet (t_{d_o}), $^{\circ}F$	Average (t_d), $^{\circ}F$				$V_w P_b (t_d + 460)$	$V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)$
10.280 2.0	10	72.138 82.468	76	78	90	84	13:23	8	9825	$\frac{(10) 30.01 (544)}{10.280 (30.157)}$	536
10.379 2.0	10	82.468 92.847	76	88	98	93	12:56	8	9892	$\frac{10 (30.01) 532}{16.379 (30.157)}$	536
10.507 2.0	10	92.847 103.357	76	98	108	103	12:40	8	9948	$\frac{10 (30.01) 563}{10.507 (30.157)}$	536
									$Y = 9888$		

^a If there is only one thermometer on the dry gas meter, record the temperature under t_d where

V_w = Gas volume passing through the wet test meter, ft^3 .

V_d = Gas volume passing through the dry gas meter, ft^3 .

t_w = Temperature of the gas in the wet test meter, $^{\circ}F$.

t_{d_i} = Temperature of the inlet gas of the dry gas meter, $^{\circ}F$.

t_{d_o} = Temperature of the outlet gas of the dry gas meter, $^{\circ}F$.

t_d = Average temperature of the gas in the dry gas meter, obtained by the average of t_{d_i} and t_{d_o} , $^{\circ}F$.

ΔH = Pressure differential across orifice, in. H_2O .

Y_i = Ratio of accuracy of wet test meter to dry gas meter for each run.

Y = Average ratio of accuracy of wet test meter to dry gas meter for all three runs; tolerance = pretest $Y \pm 0.05Y$.

P_b = Barometric pressure, in. Hg.

θ = Time of calibration run, min.



BY W. H. H. H. H. H. DATE 1/2/54

SHEET NO. 1 OF 2

PROJECT NO

240°, After 15 min.

APPENDIX I

CALIBRATION FORM A
(Must Be Included With Field Data Sheet)

Gauge	Average Test Condition Temperature	Gauge Reading	Reference Reading Room Temp	Difference
Impinger (Pre)	65	72	72	0
Meter In (Pre)	~ 90	73	72	+1
Meter Out (Pre)			Room Temp	
Filter Heated Area (Pre) (IN OIL)	250	274	248 ± 25°F 272	+2
Stack (Post)	200-400	270	Stack Temp ° F 272	-2

Post Test

Train Leak Rate

Run 1

CFM

Run 2

CFM

Run 3

CFM

ALL RUNS
 < 0.02 CFM
 O.K.

Pitot Leak Check

Pre Test (Answer Yes or No)

Orsat Leak Check

N/A

Metering System Leak Check

YES

Barometer Calibrated

YES

Sample Bag Leak Check (if used)

N/A

Signature

Dan A. Monte

Title

AIR QUALITY SPEC. Date 10/84

APPENDIX D
PROCESS RECORDS

9/17/84

LEEDS RAW MILL PROCESS DATA SHEET

DATE _____

[illegible]

DATE 7/17/84

LEEDS KILN PROCESS DATA SHEET

Main Stack Test

TIME	PREHEATER										KILN										COOLER										SPRAY TOWER				COAL MILL								REMARKS
	Feed Rate	ID-30 Speed	ID-30 Draft	Stage 1 Exit Temp.	Stage 1 Gas Temp.	Stage 4 Gas Temp.	Reh. Exit 02	ID-24 Speed	Kiln Speed	Kiln Amps	Kiln Exit 02	Sec. Air Temp.	#1 Grate Speed	#1 Undergrate Press.	#2 Undergrate Press.	#4 Undergrate Press.	#1 Fan Flow	#2 Fan Flow	#3 Fan Flow	#4 Fan Flow	Kiln Hood Draft	Vent fan Damper	Filter Inlet Temp.	Spray Tower Inlet Temp.	Spray Tower Outlet Temp.	Water Spray	Coal Feed	Mil Inlet Temp.	Mil Outlet Temp.	Mil Inlet Draft	Mil Drive Amps	System Fan Damper	Prim. Air Fan Damper	Dust Trap Temp.									
7:15	134	100	1	34	705	100	X	100	112	100	1	100	5	10	11	72	X	16	17	19	19	100	100	100	1	1	1	10.6	410	110	9	29	60	37									
7:20	134	100	1	34	705	100	X	100	112	100	1	100	5	10	12	76	X	16	17	17	17	100	100	100	1	1	1	10.7	405	155	85	29	60	37									
7:25	134	100	1	34	705	100	X	100	112	100	9	100	5	12	12	7	X	15	16	155	155	100	60	100	1	1	1	10.8	408	162	6	28	60	37									
7:30	134	100	1	34	705	100	X	100	112	100	8	100	5	12	14	18	X	15	16	20	20	100	70	100	1	1	1	10.8	405	165	9	28	60	37									
7:35	134	100	9	35	700	100	X	100	112	100	9	100	7	14	15	102	X	17	15	18	18	100	78	110	1	1	1	10.8	405	165	25	28	60	37									
7:40	134	100	11	33	705	100	X	100	112	100	11	100	55	14	15	48	X	16	18	16	16	100	87	100	1	1	1	10.8	408	160	25	28	60	37									
7:45	134	100																																									
7:50	134	100																																									
7:55	134	100																																									
8:00	134	100																																									
8:05	134	100																																									
8:10	134	100																																									
8:15	134	100																																									
8:20	134	100																																									
8:25	134	100																																									
8:30	134	100																																									
8:35	134	100																																									
8:40	134	100																																									
8:45	134	100																																									
8:50	134	100																																									
8:55	134	100																																									
9:00	134	100																																									
9:05	134	100																																									
9:10	134	100																																									
9:15	134	100																																									
9:20	134	100																																									
9:25	134	100																																									
9:30	134	100																																									
9:35	134	100																																									
9:40	134	100																																									
9:45	134	100																																									
9:50	134	100																																									

DATE 9-18-84

LEEDS KILN PROCESS DATA SHEET

MAIN STACK Test

MAIN STACK Test										COOLER										SPRAY TOWER				COAL MILL							REMARKS				
PREHEATER					KILN					COOLER										SPRAY TOWER				COAL MILL							REMARKS				
TIME	Feed Rate	ID-30 Speed	ID-30 Draft	Stage 1 Exit Draft	Stage 1 Gas Temp.	Stage 4 Gas Temp.	Preh. Exit O ₂	ID-24 Speed	Kiln Speed	Kiln Amps	Kiln Exit O ₂	Sec. Air Temp.	#1 Grate Speed	#1 Undergrate Press.	#2 Undergrate Press.	#4 Undergrate Press.	#1 Fan Flow	#2 Fan Flow	#3 Fan Flow	#4 Fan Flow	Kiln Hood Draft	Vent Fan Damper	Filter Inlet Temp.	Spray Tower Inlet Temp.	Spray Tower Outlet Temp.	Water Spray	Coal Feed	Mill Inlet Temp.	Mill Outlet Temp.	Mill Inlet Draft	Mill Drive Amps	System Fan Damper	Preh. Air Fan Damper	Dust Trap Temp.	REMARKS
10:00 AM	140	1040	-5	33 1/2	700	1505	7	70	118	7 1/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	650	Test #3					
10:30 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660	M:11					
11:00 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660	OFF #2					
11:30 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
12:00 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
12:30 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
1:00 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
1:30 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
2:00 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
2:30 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
3:00 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
3:30 PM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
4:00 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
4:30 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
5:00 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
5:30 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						
6:00 AM	138	1040	-5	33 1/2	700	1505	5	70	118	9/11	17 1/2	5	12 1/2	13	5 1/2	17	16	20	17	10.2	10.2	10.2	415	155	1	28	60	37	660						

138 TP
Average
Feed 15

138 TPH
Average
Feed to Kiln

LEEDS RAW MILL PROCESS DATA SHEET

DATE 9/18/84

REMARKS
Test #4
Mill OFF (#3)

TIME	LOESCHE MILL								AUXILIARIES								REMARKS	
	Diff. Pressure 11 WC	Air Flow 1000 CFM	Inlet Temp. OF	Outlet Temp. OF	Feed Rate T/H	Vibration 11/sec	Classifier Speed RPM	M:11 Amps A	D24 Amps A	D24 Speed RPM	O11 Elevator A	O418 Elevator A	D52 Pump Speed RPM	D52 Pump Press. PSI	FEED Bin Level Tons	Opacity %		M:11 ΔP Set Point %
3 PM	T	T	T	T	T	T	T	T	162	680	T	11	405	2	T	6	T	
15	T	T	T	T	T	T	T	T	162	680	T	11	405	2	T	6	T	
30	T	T	T	T	T	T	T	T	162	680	T	11	405	2	T	10	T	
45	T	T	T	T	T	T	T	T	162	680	T	11	405	3	T	8	T	
1 PM	T	T	T	T	T	T	T	T			T				T		T	
8 PM	T	T	T	T	T	T	T	T			T				T		T	
15	T	T	T	T	T	T	T	T			T				T		T	
2 PM	T	T	T	T	T	T	T	T			T				T		T	
30	T	T	T	T	T	T	T	T			T				T		T	
45	T	T	T	T	T	T	T	T			T				T		T	
12 AM	T	T	T	T	T	T	T	T			T				T		T	
2 AM	T	T	T	T	T	T	T	T			T				T		T	
4 AM	T	T	T	T	T	T	T	T			T				T		T	
6 AM	T	T	T	T	T	T	T	T			T				T		T	

LEEDS KILN PROCESS DATA SHEET

DATE

Main Stack Test

TIME	PREHEATER										KILN				COOLER										SPRAY TOWER				COAL MILL							REMARKS
	Feed Rate	ID-30 Speed	ID-30 Draft	Stage 1 Exit Draft	Stage 1 Gas Temp.	Stage 4 Gas Temp.	ID-24 Speed	Kiln Speed	Kiln Amps	Kiln Exit O ₂	Sec. Air Temp.	#1 Grate Speed	#1 Undergrate Press.	#2 Undergrate Press.	#4 Undergrate Press.	#1 Fan Flow	#2 Fan Flow	#3 Fan Flow	#4 Fan Flow	Kiln Hood Draft	Vent Fan Damper	Filter Inlet Temp.	Spray Tower Inlet Temp.	Spray Tower Outlet Temp.	Water Spray	Coal Feed	Mill Inlet Temp.	Mill Outlet Temp.	Mill Inlet Draft	Mill Drive Amps	System Fan Damper	Prim. Air Fan Damper	Dust-Trip Temp.			
1 PM	135	1000	-5	33	710	1510	680	110	8/15	1	1850	5 1/2	14	15	5.4	16	19	15	15	.06	95	550	540	305	72	10.2	360	140	1	28	60	37	640	135 TPH Feed to Kiln		
2 PM	135	1000	-5	33	710	1510	680	110	11/16	1	1900	7	15	9.5	16	18	17	17	.06	100	600	550	305	68	10.2	480	150	.8	28	60	37	730				
3 PM	135	1000	-5	33	710	1510	680	110	11/16	1	1850	7	15.5	17	5	15 1/2	16 1/2	15	.07	90	510	530	305	68	10.2	355	140	1	28	60	37	590				
4 PM	135	1000	-5	32 1/2	710	1510	680	110	9/15	1	1800	5	12	13	6.5	16	19	17	.04	100	600	550	305	67	10.2	480	150	1	28	60	37	720				
5 PM	135	1000	-5	32 1/2	710	1510	680	110	8/15	1	1720	7 1/2	14.5	16.2	10	17	18	17	.01	86	550	540	305	68	10.2	470	150	.8	27	60	37	730				
6 PM	135	1000	-5	33	710	1510	680	110	6/12	1	1750	5	13	14	4.9	15	17	15	.09	62	440	590	305	67	10.2	340	135	.8	27	60	37	560				
7 PM																																				
8 PM																																				
9 PM																																				
10 PM																																				
11 PM																																				
12 AM																																				
1 AM																																				
2 AM																																				
3 AM																																				
4 AM																																				
5 AM																																				
6 AM																																				

Mill off (#3)

135 TPH Feed to Kiln

DATE 2/19/87

REMARKS

Test 5

$$M:=(\#4)(\#4)$$

SSS

TIME

10

3 AM

50

20

12 PM

2

11

4 PM

52

01/28

4:14 PM

1

10 PM

12 AM

1

2 AM

4 AM

1.

6 AM

DATE 2/19/84

LEEDS KILN PROCESS DATA SHEET

Main Stack Test

TIME			PREHEATER										KILN				COOLER										SPRAY TOWER						COAL MILL						REMARKS
			Feed Rate	ID-30 Speed	ID-30 Draft	Stage 1 Exit Draft	Stage 1 Gas Temp.	Stage 4 Gas Temp.	Preh. Exit 02	ID-24 Speed	Kiln Speed	Kiln Amps	Kiln Exit 02	Sec. Air Temp.	#1 Grate Speed	#1 Undergrate Press.	#2 Undergrate Press.	#4 Undergrate Press.	#1 Fan Flow	#2 Fan Flow	#3 Fan Flow	#4 Fan Flow	Kiln Hood Draft	Vent Fan Damper	Filter Inlet Temp.	Spray Tower Inlet Temp.	Spray Tower Outlet Temp.	Water Spray	Coal Feed	Mill Inlet Temp.	Mill Outlet Temp.	Mill Inlet Draft	Mill Drive Amps	System Fan Damper	Prim. Air Fan Damper	Dust Trap Temp.			
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630	134 TPH feed to kiln	Test #5									
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630											
8 PM	15	1040	-5	33	700	1500	1	135	5	13.8	11.6	7.5	16	17	10.5	100	480	340	305	67	10.2	305	140	1.1	27	60	37	630								</			

LEEDS RAW MILL PROCESS DATA SHEET DATE 2/20/84
Main Stack Test

TIME	LOESCHE MILL										AUXILIARIES							REMARKS	
	Diff. Pressure	Air Flow	Inlet Temp.	Outlet Temp.	Feed Rate	Vibration	Classifier Speed	Mill Amps	D24 Amps	D24 Speed	D11 Elevator	D4/B Elevator	D52 Pump Speed	D52 Pump Press.	Feed Bin Level	Opacity	Mill ΔP Set Point		
05:30 AM	36	140	600	190	8.4	10/18	48	160	205	1180	26	19	1180	22	68	4	73	Test #6 Mill ON (#2)	
06:20 AM	37	190	620	195	8.5	10/18	46	150/165	205	1180	30	20	1180	25	67	5	73		
07:35 AM	37	190	620	195	8.6	11/18	46	150/165	208	1180	25	24	1180	26	64	4	73		
08:50 AM	37	190	615	195	8.5	10/18	45	150/165	207	1180	23	21	1180	25	56	4	73		
09:05 AM	37	190	600	195	8.6	10/18	45	150/165	205	1180	24	22	1180	25	56	4	73		
09:20 AM	37	190	600	190	8.4	11/18	46	150/165	205	1180	24	21	1180	25	56	3	73		
10:35 AM																			
10 PM																			
12 AM																			
2 AM																			
4 AM																			
6 AM																			

LEEDS KILN PROCESS DATA SHEET
Main Stack Test

TIME	PREHEATER							KILN			COOLER													
	Feed Rate	ID-30 Speed	ID-30 Draft	Stage 1 Exit Draft	Stage 1 Gas Temp.	Stage 4 Gas Temp.	Reh. Exit O2	ID-24 Speed	Kiln Speed	Kiln Amps	Kiln Exit O2	Sec. Air Temp.	#1 Grate Speed	#1 Undergrate Press.	#2 Undergrate Press.	#4 Undergrate Press.	#1 Fan Flow	#2 Fan Flow	#3 Fan Flow	#4 Fan Flow	Kiln Hood Draft	Vent Fan Damper	Filter Inlet Temp.	
8 AM	150	100	-1	28	150	150	100	100	100	100	100	100	9	5	16	5	1	16	15	15	16	06	60	30
9 AM	145	100	-2	28	150	150	100	100	100	100	100	100	7	14	15	7	1	16	16	16	07	60	30	
9:35 AM	145	100	-1 1/2	28	150	150	100	100	100	100	2	100	6	14	15	6	1	15 1/2	18	16	07	58	40	
10 AM	152	100	-1 1/2	28	150	150	100	100	100	100	3 1/2	100	8	14 1/2	16	7 1/2	1	16	17 1/2	15	07	61	40	
10:05 AM	150	100	-2	28	150	150	100	100	100	100	3 1/2	100	7	14 1/2	15	6 1/2	1	16	14	15	07	71	50	
10:20 AM	150	100	-2	28	150	150	100	100	100	100	3 1/2	100	8	14 1/2	15 1/2	9	1	17	17	16	06	71	50	
10:35 AM																								
10 PM																								
12 AM																								
2 AM																								
4 AM																								
6 AM																								

LEEDS RAW MILL PROCESS DATA SHEET

DATE 9/20/84

TIME	LOESCHE MILL								AUXILIARIES								REMARKS	
	Diff. Pressure	Air Flow	Inlet Temp.	Outlet Temp.	Feed Rate	Vibration	Classifier Speed	Mill Amps	D24 Amps	D24 Speed	O11 Elevator	O4/B Elevator	D52 Pump Speed	D52 Pump Press.	Feed Bin Level	Opacity	Mill ΔP Set Point	Test #7 Mill On #3
30 AM	37	185	610	200	8.3	10/18	47	150/115	203	1180	30	21	1180	26	—	4	73	
45 AM	37	185	610	200	8.5	11/18	47	150/110	202	1180	22	20	1180	26	—	4	73	
10 AM	37	185	610	200	8.5	11/18	46	150/110	200	1180	30	24	1180	25	—	6	73	
15 AM	37	185	610	200	8.5	11/18	46	150/110	201	1180	22	19	1180	26	—	4	73	
30 PM	37	190	610	200	8.4	11/19	46	150/110	201	1180	30	19	1180	27	—	4	73	
45 PM																		
8 PM																		
10 PM																		
12 AM																		
2 AM																		
4 AM																		
6 AM																		

TIME	Feed Rate
30 AM	52
45 AM	45
10 AM	45
15 PM	45
30 PM	45
45 PM	45
8 PM	45
10 PM	45
12 AM	45
2 AM	45
4 AM	45
6 AM	45

APPENDIX E
VISIBLE EMISSIONS

Bureau of Environmental Health
Paul Pate, Director



September 25, 1984

Mr. David Martin
Scientist
CH₂M Hill
807 South McDonough Street
P.O. Box 2499
Montgomery, Alabama 36104

Dear Mr. Martin:

As per our conversation Friday, September 21, 1984, enclosed are copies of visible emission readings performed by Sanitarians from this office on the Loesche Mill September 17 - 18, 1984 at Lehigh Cement, Leeds, Alabama. As we had discussed, please include these visible emissions readings in the stack test report.

If you have any questions, please contact me at 933-9110, extension 1280.

Sincerely,

Veda M. Hampton
Air Pollution Control Engineer
Air Pollution Control Program

VMH/bb

Enc.



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Company Name Lehigh Cement (Leeds) Location 18
Source Kiln Permit No. 0002901501
Start/Stop Time 3:13/3:42 Sky Condition 50% clouds Date 091884
Observation Point SSW of source Distance from Source 100 yds
Stack Height 101' Background blue sky Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☐ If Notice Given, Notice Number ☐
Total Minutes Observed 30 Highest 6 Minute Average 10
Minutes Exceeding: ☐ % Opacity = ☐ ☐ % Opacity = ☐
Observer Cargo 43 Time Analysis 02 ID V E

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	10	10	10	10	15	10	10	10	10	30					45				
1	10	10	10	10	16	10	10	10	10	31					46				
2	10	10	10	10	17	10	10	10	10	32					47				
3	10	10	10	10	18	10	10	10	10	33					48				
4	10	10	10	10	19	10	10	5	10	34					49				
5	10	10	10	10	20	10	10	5	10	35					50				
6	10	10	10	5	21	10	10	10	10	36					51				
7	10	10	5	10	22	5	5	5	5	37					52				
8	10	10	10	10	23	5	5	5	5	38					53				
9	10	15	10	10	24	10	5	5	5	39					54				
10	10	10	10	10	25	5	5	10	10	40					55				
11	10	10	10	10	26	10	10	10	10	41					56				
12	10	10	10	10	27	10	10	10	15	42					57				
13	10	10	10	10	28	10	10	10	10	43					58				
14	10	10	10	10	29	10	10	10	10	44					59				

Results and Recommendations:

stack test min 9-14 - 10% opacity

run #5

reading point of maximum opacity (detached plume)

MILL OFF



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Company Name Lehigh Cement (Leeds) Location 18
Source Kiln Permit No. 0002901501
Start/Stop Time 1:04/1:33 Sky Condition 50% clouds Date 09/18/84
Observation Point SSW of source Distance from Source 100 yards
Stack Height 101 ft Background blue sky Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☐ If Notice Given, Notice Number ☐
Total Minutes Observed 30 Highest 6 Minute Average 10
Minutes Exceeding: ☐ % Opacity = ☐ ☐ % Opacity = ☐
Observer Cargo 43 Time Analysis 02 ID VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	5	5	5	10	15	15	15	10	10	30					45				
1	5	5	10	5	16	10	10	10	10	31					46				
2	5	10	10	5	17	10	10	10	5	32					47				
3	5	5	5	5	18	5	5	5	5	33					48				
4	0	0	0	0	19	5	5	5	5	34					49				
5	0	5	5	5	20	5	5	5	5	35					50				
6	10	10	5	5	21	5	5	5	10	36					51				
7	5	5	10	10	22	5	5	5	5	37					52				
8	5	5	5	5	23	5	5	5	5	38					53				
9	10	10	10	10	24	5	10	10	10	39					54				
10	5	10	5	10	25	10	10	10	10	40					55				
11	10	10	10	15	26	5	5	10	5	41					56				
12	10	10	10	10	27	5	10	10	10	42					57				
13	10	10	10	5	28	10	5	5	5	43					58				
14	10	10	15	15	29	5	5	5	5	44					59				

Results and Recommendations:

Stack test min 11-16 = 10% opacity
run # 4

Reading point of highest opacity (detached plume)



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS, OBSERVATION REPORT

Company Name Lohigh Cement (Leeds) Location 18
Source Kiln Permit No. 0002901501
Start/Stop Time 10:13/10:42 Sky Condition clear Date 09/18/84
Observation Point East of source Distance from Source 200 yards
Stack Height 101 ft Background sky Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other ☐
Violation Observed ☐ If Notice Given, Notice Number ☐
Total Minutes Observed 30 Highest 6 Minute Average 12
Minutes Exceeding: ☐ % Opacity = ☐ % Opacity = ☐
Observer Cargo 43 Time Analysis 02 ID VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	15	10	5	5	15	5	10	5	5	30					45				
1	10	5	5	40	16	10	5	5	5	31					46				
2	10	10	10	15	17	10	5	5	5	32					47				
3	5	5	5	10	18	30	10	5	5	33					48				
4	10	15	10	20	19	5	10	10	5	34					49				
5	10	10	10	5	20	10	10	10	10	35					50				
6	10	20	30	15	21	10	10	15	15	36					51				
7	10	10	10	5	22	5	10	10	15	37					52				
8	5	10	10	15	23	15	15	10	10	38					53				
9	10	10	5	10	24	10	10	5	10	39					54				
10	15	10	10	10	25	5	5	10	5	40					55				
11	10	10	5	10	26	5	5	5	10	41					56				
12	5	5	10	5	27	10	5	5	5	42					57				
13	10	10	10	5	28	5	5	10	5	43					58				
14	10	5	5	10	29	5	5	5	5	44					59				

Results and Recommendations:

stack test min 1-6 = 12% opacity
run #3

pull off



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Company Name Lehigh Cement (Leeds) Location 18
Source Kilm Permit No. 0002901501
Start/Stop Time 7:45/8:14 Sky Condition 20% clouds Date 09/18/84
Observation Point East of source Distance from Source 200 yards
Stack Height 101 ft Background blue sky Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☐ If Notice Given, Notice Number ☐
Total Minutes Observed 30 Highest 6 Minute Average 11
Minutes Exceeding: ☐ % Opacity = ☐ ☐ % Opacity = ☐
Observer Cargo Time Analysis 02 ID VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	15	20	20	15	15	5	5	5	5	30					45				
1	15	15	10	10	16	5	5	5	5	31					46				
2	10	15	10	15	17	5	5	10	10	32					47				
3	15	20	15	5	18	5	10	10	10	33					48				
4	5	5	10	5	19	5	5	5	5	34					49				
5	10	5	10	10	20	5	10	5	5	35					50				
6	5	5	10	5	21	5	10	5	5	36					51				
7	10	10	5	5	22	10	5	5	5	37					52				
8	5	5	10	5	23	5	5	5	5	38					53				
9	5	10	5	5	24	5	10	5	5	39					54				
10	5	5	10	5	25	5	5	10	5	40					55				
11	10	5	5	10	26	5	5	10	5	41					56				
12	10	10	5	5	27	5	5	10	5	42					57				
13	5	10	10	5	28	5	5	5	5	43					58				
14	10	5	10	5	29	5	5	5	5	44					59				

Results and Recommendations:

Stack test min 0-5 = 11% opacity
run # 2

MILL OFF



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Company Name Lehigh Cement (L-2 ds) Location 18
Source Kiln Permit No. 0002901501
Start/Stop Time 2:48/3:17 Sky Condition 75% clouds Date 09/19/84
Observation Point SW of source Distance from Source 100 yds
Stack Height 101 ft Background cloudy sky Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☐ If Notice Given, Notice Number ☐
Total Minutes Observed 30 Highest 6 Minute Average 06
Minutes Exceeding: ☐ % Opacity = ☐ , ☐ % Opacity = ☐
Observer Cargo 43 Time Analysis 02 ID V F

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	0	0	0	0	15	5	5	5	5	30					45				
1	0	0	0	0	16	5	10	5	5	31					46				
2	0	0	0	0	17	10	5	5	10	32					47				
3	0	0	0	0	18	5	5	5	5	33					48				
4	0	0	0	0	19	5	5	10	5	34					49				
5	0	0	0	0	20	5	5	5	5	35					50				
6	0	0	0	0	21	10	5	5	5	36					51				
7	0	0	0	0	22	10	5	10	5	37					52				
8	0	0	0	0	23	5	5	5	5	38					53				
9	0	0	0	0	24	5	10	5	5	39					54				
10	0	0	0	0	25	5	5	5	10	40					55				
11	0	0	0	0	26	5	5	5	5	41					56				
12	0	0	0	0	27	5	5	10	5	42					57				
13	0	0	0	0	28	5	5	5	10	43					58				
14	0	0	0	5	29	5	5	5	10	44					59				

Results and Recommendations:

min 17-22 = 6% opacity

Stack test run #1

mill on



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Company Name

Lehigh Cement (Roads)

Location

18

Source

Kilm

Permit No.

0002901501

Start/Stop Time

8:04/8:33

Sky Condition

clear

Date

09 25 84

Observation Point

East of source

Distance from Source

200 yards

Stack Height

101'

Background

blue sky

Plume Color

white

Method of Evaluation:

☐ 1 Hour Cumulative

☒ 6 Minute Average

☐ Other

Violation Observed

☐

If Notice Given, Notice Number

☐

Total Minutes Observed

30

Highest 6 Minute Average

3

Minutes Exceeding:

32

% Opacity =

34

% Opacity =

38

Observer

Cargo

43

Time Analysis

02

ID

VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	5	5	5	5	15	5	5	5	5	30					45				
1	5	10	10	5	16	0	0	0	5	31					46				
2	0	0	0	0	17	0	0	5	5	32					47				
3	0	5	5	5	18	0	0	5	5	33					48				
4	5	0	5	5	19	5	0	5	5	34					49				
5	5	5	5	5	20	5	0	0	5	35					50				
6	5	0	0	0	21	5	5	5	5	36					51				
7	5	5	5	5	22	5	5	5	5	37					52				
8	5	5	0	5	23	5	5	0	5	38					53				
9	5	0	0	0	24	5	5	0	0	39					54				
10	5	5	0	0	25	0	5	5	0	40					55				
11	0	0	5	0	26	5	5	0	0	41					56				
12	0	0	0	0	27	5	5	0	0	42					57				
13	0	0	0	0	28	5	0	5	5	43					58				
14	0	0	0	5	29	0	5	0	5	44					59				

Results and Recommendations:

Stack test

min 0-5 = 39% opacity

run #6

mill on



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Vmt
9/21/84
JCBH

Company Name Lohigh Cement Co Location 18
Source Kiln Permit No. 0002901501
Start/Stop Time 10:23/10:52 Sky Condition clear Date 092084
Observation Point SE of source Distance from Source 200 yds
Stack Height 101' Background blue sky Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☐ If Notice Given, Notice Number ☐
Total Minutes Observed 30 Highest 6 Minute Average 03
Minutes Exceeding: ☐ % Opacity = ☐ , ☐ % Opacity = ☐
Observer C. A. G. Time Analysis 02 ID V F

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	5	5	5	5	15	0	0	0	5	30					45				
1	5	5	5	5	16	5	5	0	5	31					46				
2	5	5	5	5	17	5	5	5	5	32					47				
3	5	5	5	5	18	5	5	5	5	33					48				
4	5	5	5	5	19	5	5	5	5	34					49				
5	5	5	5	5	20	5	5	5	5	35					50				
6	5	5	5	5	21	5	5	5	5	36					51				
7	5	5	5	5	22	5	5	5	5	37					52				
8	5	5	5	5	23	5	5	5	5	38					53				
9	5	5	5	5	24	5	5	5	5	39					54				
10	5	5	5	5	25	5	5	5	5	40					55				
11	5	5	5	5	26	5	5	5	5	41					56				
12	5	5	5	5	27	5	5	5	5	42					57				
13	5	5	5	5	28	5	5	5	5	43					58				
14	5	5	5	5	29	5	5	5	5	44					59				

Results and Recommendations:

stack test
run # 7

min 17-22 = 3% opacity

min on



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

2/4
9/21/84
JCH

Company Name Lehigh Cement Location 18

Source Clinker cooler Permit No. 0002901502

Start/Stop Time 8:17/8:46 Sky Condition 10% clouds Date 092184

Observation Point SE of source Distance from Source 1000 ft

Stack Height 80 feet Background Pro-hat tower Plume Color white

Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other

Violation Observed ☒ If Notice Given, Notice Number 23

Total Minutes Observed 30 Highest 6 Minute Average 11

Minutes Exceeding: 32 % Opacity = 34 36 % Opacity = 38

Observer Cargo 43 Time Analysis 02 ID VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	15	10	10	10	15	5	5	10	15	30					45				
1	15	10	10	10	16	10	10	5	5	31					46				
2	10	10	10	10	17	5	5	5	5	32					47				
3	5	5	5	15	18	5	5	5	5	33					48				
4	5	15	10	10	19	5	15	10	10	34					49				
5	10	15	15	10	20	5	5	5	5	35					50				
6	10	15	10	10	21	5	5	5	5	36					51				
7	10	10	10	15	22	5	5	5	5	37					52				
8	15	10	10	10	23	10	10	10	5	38					53				
9	10	10	5	10	24	5	5	5	10	39					54				
10	5	5	10	10	25	5	5	5	5	40					55				
11	10	5	5	10	26	10	5	5	5	41					56				
12	10	10	10	10	27	10	5	5	5	42					57				
13	5	5	5	10	28	5	5	5	5	43					58				
14	10	5	5	5	29	5	5	5	5	44					59				

Results and Recommendations:

Stack test
run #8 min 4-9 = 11% opacity



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

VMA
9/21/84
JCOH

Company Name High Cement (Road) Location 18
Source clinker cooler Permit No. 0002901902
Start/Stop Time 10:17/10:46 Sky Condition 10% clouds Date 092184
Observation Point SE of source Distance from Source 1000 ft
Stack Height 80 ft Background pre heat tower Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☒ If Notice Given, Notice Number 23
Total Minutes Observed 30 Highest 6 Minute Average 12
Minutes Exceeding: 32 % Opacity = 34 % Opacity = 36
Observer Cargot Time Analysis 02 ID VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	5	5	10	10	15	5	5	5	5	30					45				
1	10	5	5	5	16	5	5	5	5	31					46				
2	5	5	5	5	17	5	5	10	10	32					47				
3	5	5	5	5	18	5	5	10	5	33					48				
4	5	5	15	5	19	5	5	5	5	34					49				
5	5	5	5	5	20	5	30	5	5	35					50				
6	5	5	5	5	21	5	5	5	5	36					51				
7	10	15	5	5	22	5	5	5	5	37					52				
8	5	15	15	20	23	5	5	5	5	38					53				
9	5	5	0	15	24	10	15	15	10	39					54				
10	10	20	20	5	25	10	10	5	5	40					55				
11	5	5	10	5	26	5	5	5	5	41					56				
12	5	10	10	5	27	10	10	5	5	42					57				
13	0	0	5	5	28	35	10	5	5	43					58				
14	5	5	15	15	29	5	5	5	10	44					59				

Results and Recommendations:

stack test
run #9

min 7-13 = 12% opacity
11.46



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

VCH
9/21/84
JCDH

Company Name L. High Cement (Leeds) Location 18
Source Clinker Cooler Permit No. 0002901502
Start/Stop Time 12:29/12:58 Sky Condition 10% clouds Date 09 21 84
Observation Point 500 ft Distance from Source 1000 ft
Stack Height 80 ft Background no heat towers Plume Color white
Method of Evaluation: ☐ 1 Hour Cumulative ☒ 6 Minute Average ☐ Other
Violation Observed ☒ If Notice Given, Notice Number 23
Total Minutes Observed 30 Highest 6 Minute Average 30
Minutes Exceeding: 32 % Opacity = 34 % Opacity = 38
Observer Conrad Time Analysis 00 ID VE

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	15	15	15	15	15	5	10	10	10	30					45				
1	25	25	25	25	16	10	10	15	50	31					46				
2	20	10	15	15	17	40	40	30	30	32					47				
3	15	15	15	20	18	30	30	35	30	33					48				
4	15	15	15	10	19	40	40	40	30	34					49				
5	20	15	15	15	20	50	25	25	30	35					50				
6	15	15	10	10	21	30	30	25	30	36					51				
7	10	5	10	5	22	25	25	25	20	37					52				
8	5	10	10	50	23	10	25	30	20	38					53				
9	50	15	10	10	24	20	25	25	20	39					54				
10	5	10	10	5	25	25	30	20	20	40					55				
11	5	5	10	5	26	30	20	30	20	41					56				
12	5	5	5	15	27	25	15	30	15	42					57				
13	15	15	10	10	28	30	20	25	15	43					58				
14	10	10	10	10	29	5	15	15	10	44					59				

Results and Recommendations:

stack test
run #10
min 17-22 = 30% opacity