

Note: This is a reference cited in *AP 42, Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

The file name refers to the reference number, the AP42 chapter and section. The file name "ref02_c01s02.pdf" would mean the reference is from AP42 chapter 1 section 2. The reference may be from a previous version of the section and no longer cited. The primary source should always be checked.

AP-42 Section Number: 11.17

Reference Number: 13

Title: Source Test of a Lime Plant, Standard
Lime Company, Woodville, OH

Feairheller, W. R. et al., EPA Contract
No. 68-02-1404, Task 12

Monsanto Research Corporation

December 1975

SOURCE TEST OF A LIME PLANT

Standard Lime Company
Div. Martin - Marietta
Woodville , Ohio

$$\frac{\text{Stone}}{\text{prod}} = 2.37$$

$$\frac{\text{Feed}}{\text{prod}} = 2.51 \text{ (w/coal)}$$

B-8/A-11

MONSANTO RESEARCH CORPORATION

A SUBSIDIARY OF MONSANTO COMPANY



**D A Y T O N
L A B O R A T O R Y**

DAYTON, OHIO 45407

Source Test of a Lime Plant

Plant Tested

Standard Lime Company
Div. Martin-Marietta
Woodville, Ohio

December 8, 9, and 10, 1975

Prepared for

Environmental Protection Agency
National Air Data Branch
Research Triangle Park
North Carolina 27711

by

W. R. Feairheller, H. D. Toy and
J. R. McKendree

Monsanto Research Corporation
Dayton Laboratory
1515 Nicholas Road
Dayton, Ohio 45418

Contract No. 68-02-1404, Task No. 12

TABLE OF CONTENTS

	<u>Page</u>
I. Introduction	1
II. Summary of Results	3
III. Process Description	14
IV. Location of Sampling Points	20
V. Sampling and Analytical Procedures	25
 <u>Appendix</u>	
A. Complete Particulate Results	
B. Particle Size Data from Brink® BMS-11	
C. Complete NO _x Results and Sample Calculation	
D. Field Data Sheets for Methods 5, 7, and Particle Sizing	
E. Analytical Data Sheets	
F. Sampling and Analytical Procedures	
G. Brink® BMS-11 Sampler Manual	
H. Job Participants	

LIST OF TABLES

<u>Table</u>		<u>Page</u>
1	Summary of Particulate Emissions from Kilns 4 and 6 (Metric)	4
2	Summary of Particulate Emissions from Kilns 4 and 6 (English)	5
3	Summary of NO _x Emissions	8
4	Brink® Particulate Sizing	10
5	Production Data	17

Appendix

A-1	Method 5 Data
A-2	Particulate Calculations
B-1	Particle Size Distribution for Run 1-4 (English)
B-2	Particle Size Distribution for Run 1-4 (Metric)
B-3	Particle Size Distribution for Run 1-6 (English)
B-4	Particle Size Distribution for Run 1-6 (Metric)
C-1	NO _x Results
C-2	NO _x Sample Calculations

LIST OF FIGURES

<u>Figure</u>		<u>Page</u>
1	Particle size distribution - kiln 4	12
2	Particle size distribution - kiln 6	13
3	Kiln no. 4 flow diagram	18
4	Kiln no. 6 flow diagram	19
5	Kiln stack #4	21
6	End view of kiln stacks 5 and 6	22
7	Top view, kiln stacks 5 and 6	23

SECTION I

INTRODUCTION

The purpose of the test program was to measure the uncontrolled particulate and nitrogen dioxide emissions, and to obtain particle size data on two of the lime kilns at the Standard Lime Company Plant in Woodville, Ohio. This study was initiated as part of the data acquisition program of the National Air Data Branch of the Environmental Protection Agency.

The sampling program was conducted on the inlet ducts to the baghouse controlling emissions from the No. 4 and No. 6 kilns at the plant. By sampling prior to the control device, data on the emission rate could be obtained and emission factors determined that would represent an uncontrolled facility. During the test period from Dec. 8 to Dec. 10, 1975, kiln No. 6 was producing dead burn dolomitic lime (a more complete conversion to the oxide). Dolomitic limestone, the raw material, is a natural limestone that contains both calcium and magnesium carbonates. As a result, the dolomitic lime contains both calcium and magnesium oxides. Dead burn dolomitic lime or refractory lime is a sintered form of dolomitic lime that is calcined at high temperature with the addition of iron oxide. This material is employed primarily as a refractory for lining steel furnaces.

This report presents particulate mass emission rate as determined by EPA Method 5, the NO_x emission rate as determined by EPA Method 7, and the particle size distribution as determined by Brinks® BMS-11 Cascade Impactor. The emission factors in terms of the feed of dolomitic limestone and the production of dolomitic lime or dead burn lime were calculated for both the particulate and NO_x emissions.

The following sections of this report include: (1) summary of results, (2) process description and operation, (3) details of sampling locations, and (4) sampling and analytical procedures. The field and analytical data is presented in the appendices.

SECTION II

SUMMARY OF RESULTS

Two kilns, numbers 4 and 6, were tested at the Standard Lime Company Plant during the December 8-10, 1975 sampling program. Particulate emissions were determined by EPA Method 5, NO_x emissions by EPA Method 7, and the particle size distribution by the Brinks® cascade impactor.

Particulate Mass Emissions

Kiln No. 4 - Kiln No. 4 was producing dolomitic lime at an average feed rate of 52.3 metric ton/hr (57.6 ton/hr) and an average production rate of 22.6 metric ton/hr (24.7 ton/hr). Sampling was accomplished prior to the baghouse, after multiclone and preheater, so that the uncontrolled emission rate of filterable particulate averaged 923 kg/hr (2035 lb/hr) and total particulate averaged 924 kg/hr (2037 lb/hr). The total particulate data yields an emission factor of 18.0 kg/metric ton (36.0 lb/ton) based on the feed rate and 41.7 kg/metric ton (83.5 lb/ton) based on the product.

The data from the two particulate emission measurements as shown in Table 1 (metric) and Table 2 (English) show considerable variation. Although the volume of gas sampled, the average stack temperature and the volumetric flow rate are very close for the two runs, the mass of material collected

Table 1. SUMMARY OF PARTICULATE EMISSIONS FROM KILNS 4 AND 6 (METRIC)

Run Number - Kiln Number	Kiln 4 ^c			Kiln 6 ^d		
	1-4	2-4	Average	1-6	2-6	Average
Date	12-9-75	12-9-75		12-10-75	12-10-75	
Volume of Gas Sampled - Nm ³ a	1.262	1.269	1.265	0.976	0.746	.851
Percent Moisture by Volume	6.00	2.36	4.18	3.60	3.35	3.48
Average Stack Temperature - °C	198	196	197	274	279	277
Stack Volumetric Flow Rate - Nm ³ /min	1666	1649	1658	1511	1113	1312
Percent Isokinetic	91.7	93.1	92.4	98.1	101.8	100.0
Run Time - minutes	60	60	60	72	72	72
Particulates - Front Half						
mg	17751	5624		11964	11278	
g/Nm ³	14.084	4.438	9.261	12.274	15.144	13.709
Kg/hr	1407	439	923	1112	1011	1062
	3095.4	965.8		2446.4	2224.2	
Particulates - Total						
mg	17760	5635		11995	11344	
g/Nm ³	14.091	4.447	9.269	12.306	15.233	13.770
Kg/hr	1408	440	924	1115	1017	1066
	3097.6	968.0		2453	2237.4	
Percent Impinger Catch	0.05	0.20	0.125	0.26	0.58	0.420
	2.2	2.2		66	13.2	
Limestone Feed Rate - metric ton/hr	50.4	54.1	52.3	27.2	27.2	27.2
Production Rate - metric ton/hr	21.8	23.4	22.6	10.4	10.4	10.4
	24.0	25.74		11.44	11.44	
Emission Factor (b) - kg/mTon (feed)	27.9	8.1	18.0	41.0	37.4	39.2
kg/mTon (Product)	64.6	18.8	41.7	107.2	97.8	102.5

a) Normal cubic meters at 20°C
b) Front half data only
c) Emission data collected after multicyclone and preheater
d) Emission data after low efficiency cyclone

1407 kg/hr * 2.2 lb/kg = 3095.4
1408 kg/hr * 2.2 lb/kg = 3097.6

$$21.8 \frac{\text{mt}}{\text{hr}} \times 2200 \frac{\text{lb}}{\text{mt}} \times \frac{1.1 \frac{\text{ton}}{\text{mt}}}{2000 \frac{\text{lb}}{\text{ton}}} = 24.0 \frac{\text{ton}}{\text{hr}}$$

and moisture content varied greatly, thus producing a considerable variation in the emission rate.

The plant operation during the first run varied from a feed rate of 54.1 metric ton/hr (59.6 tons/hr) for the first half of the run to 46.8 metric ton/hr (51.6 tons/hr) for the second half of the run. During the second run, the feed rate was 54.1 metric ton/hr (59.6 ton/hr) for the entire run. Filters were changed on run 1 during the port change interval due to the high loading experienced, however, this was not necessary on run 2. There is some indication, based on the filter loadings from run 1, that the lower feed rate during the second half of this run produced a higher emission rate. However, each filter from run 1 collected more material than the filter used for the entire run 2 sampling period. Similarly, the probe washing from run 1 contained over three times the amount of material collected by run 2.

A discussion with plant personnel provided no explanation for the wide variation in results, except the possibility that the feed during run 1 contained considerably more fine material. The difference in moisture content (6.00% for run 1, 2.36% for run 2) is also an indication that the feed material was different for the two runs.

Kiln No. 6 - Kiln No. 6 was producing dead burn dolomite lime at a feed rate of 27.2 metric ton/hr (30.0 tons/hr) and a production rate of 10.4 metric ton/hr (11.5 tons/hr) during both sampling runs. Sampling was accomplished prior to the baghouse, after a low efficiency cyclone so that the measured uncontrolled emission rate of filterable particulates averaged 1062 kg/hr (2340 lb/hr) and total particulate averaged 1066 kg/hr (2350 lb/hr). The total particulate data yields average emission factors of 39.2 kg/metric ton (78.4 lb/ton) based on the feed

rate and 102.5 kg/metric ton (204.4 lb/ton) based on the product. As shown in the test summaries, Table 1 (metric units) and Table 2 (English), the sampling runs at this location are in close agreement.

The production of dead burn lime is a more complete oxide formation than the dolomitic lime and is produced by higher kiln temperatures and slower feed rates. As a result, the higher kiln temperatures would increase emission rates and the lower feed rates would tend to enlarge the emission factors observed for the dead burn material.

NO_x Emissions

The NO_x emission rates and emission factors for both kiln 4 and kiln 6 are summarized in Table 3. Three samples, using the EPA Method 7 procedure were collected at each kiln.

During the sampling period on kiln 4, the feed rate was 54.1 metric ton/hr (59.6 ton/hr) and the production rate of dolomitic lime was 23.4 metric ton/hr (25.8 ton/hr) and the NO_x emission rate, varied from 89.0 to 92.8 lb/hr for an average of 91.1 lb/hr. Based on this data, the emission factors calculate to be 0.764 kg/metric ton of feed (1.53 lb/ton) and 1.77 kg/metric ton of product (3.53 lb/ton).

Kiln 6 was sampled at five minute intervals and at this time the feed rate was 27.2 metric tons/hr (30.0 tons/hr) and the production rate was 10.4 metric ton/hr (11.5 tons/hr). The emission rate as measured by Method 7 varied from 2.58 to 13.1 kg/hr (5.69 to 28.8 lb/hr) for an average value of 7.34 kg/hr (16.2 lb/hr). These values for dead burn lime production are lower and considerably more variable than the values obtained for the NO_x emission rate from dolomitic lime production. The fact that the kiln temperature for the

Table 3. SUMMARY OF NO_x EMISSIONS

	Kiln No. 4					Kiln No. 6				
	4	5	6	AVG.		4	2	3	AVG.	
Run No.	12/9	12/9	12/9			12/9	12/9	12/9		
Date	1115	1120	1125			1000	1005	1010		
Time	270.6	282.3	277.9	276.9		13.7	69.2	33.9	38.9	
Concentration (ppm) ^a										
Emission Rate										
lb/dscf x 10 ⁵	0.513	0.535	0.527	0.525		0.026	0.131	0.064	0.074	
kg/hr ^b	3.20	3.34	3.29	3.28		0.162	0.819	0.401	0.461	
lb/hr ^c	40.4	42.1	41.5	41.3		2.58	13.1	6.38	7.34	
	89.0	92.8	91.4	91.1		5.69	28.8	14.1	16.2	
Limestone Feed Rate ^d										
metric ton/hr	54.1	54.1	54.1	54.1		27.2	27.2	27.2	27.2	
ton/hr	59.6	59.6	59.6	59.6		30.0	30.0	30.0	30.0	
Production Rate ^d										
metric ton/hr	23.4	23.4	23.4	23.4		10.4	10.4	10.4	10.4	
	25.8	25.8	25.8	25.8		11.5	11.5	11.5	11.5	
Emission Factor (metric)										
kg/metric ton (feed)	0.747	0.778	0.767	0.764		0.095	0.480	0.234	0.270	
kg/metric ton (product)	1.73	1.80	1.77	1.77		0.248	1.25	0.613	0.704	
Emission Factor (English)										
lb/ton (feed)	1.49	1.56	1.53	1.53		0.190	0.959	0.469	0.539	
lb/ton (product)	3.45	3.60	3.54	3.53		0.495	2.50	1.22	1.41	

a. ppm - parts per million by volume

b. Flow rate of 1658 Nm³/min for Kiln No. 4 and 1312 Nm³/min for Kiln No. 6 based on Method 5 runs.

c. Flow rate of 58,522 DSCF/min for Kiln No. 4 and 46333 DSCF/min for Kiln No. 6 based on Method 5 runs.

d. Feed and Production based on run feed at test time for Kiln No. 4 and typical hopper delivery rates for Kiln No. 6.

dead burned material is higher than for dolomitic lime, it might be expected that the NO_x emission for kiln 6 would be higher. This was not observed. Plant personnel could offer no explanation for the lower NO_x emission rate for dead burn production or for the wide variation in the results from kiln 6.

There are two fresh air dampers on kiln 6, one at the exit of the kiln, and the other in the straight duct between the multicyclone and the baghouse, just upstream of the sampling port. The latter damper is temperature controlled and the plant personnel indicated that it should not operate until the gas reached a temperature higher than the measured temperature. It is suspected that the fresh air introduced by these dampers reduced the NO_x emission rate and caused the variations in the measured results.

Based on the data obtained, the NO_x emission factors for kiln 6 were 0.270 kg/metric ton of feed (0.539 lb/ton) and 0.704 kg/metric ton of product (1.41 lb/ton).

Particle Size Measurements

Two particle sizing runs, one on each kiln, were performed using a Brink® BMS-11 Cascade Impactor. A description of the device is given in Appendix G and computer printouts of the results are given in Appendix B. Table 4 gives a summary of the particle sizing results. It includes the stage number, the weight of material collected on the stage, the diameter of the particles in microns on the stage, the concentration of the particles, the weight percent, and the cumulative weight percent.

Table 4. BRINK® PARTICLE SIZING

Kiln 4 - 12-9-75

<u>Stage</u>	<u>Weight of Material (mg)</u>	<u>DPC^(a)</u>	<u>MG/ACF^(b)</u>	<u>Wt. %</u>	<u>Cum. Wt. %</u>
Cyclone	60.600		175.35	60.73	100.00
1	21.100	2.30	61.005	21.14	39.28
2	9.100	1.26	26.33	9.12	18.14
3	2.900	0.79	8.39	2.91	9.02
4	2.800	0.32	8.10	2.81	6.11
5	2.600	0.16	7.52	2.61	3.31
Filter	0.700		2.03	0.70	0.70

Kiln 6 - 12-9-75

<u>Stage</u>	<u>Weight of Material (mg)</u>	<u>DPC^(a)</u>	<u>MG/ACF^(b)</u>	<u>Wt. %</u>	<u>Cum. Wt. %</u>
Cyclone	88.400		409.67	78.84	100.00
1	7.360	3.09	34.11	6.56	21.16
2	9.220	1.70	42.73	8.22	14.59
3	1.460	1.07	6.77	1.30	6.37
4	1.230	0.45	5.70	1.10	5.07
5	3.650	0.23	16.92	3.26	3.97
Filter	0.800		3.71	0.71	0.71

(a) Characteristic cut point diameter of particles collected
(micrometers)

(b) MG/ACF - milligram per actual cubic foot

The results on kiln No. 4 indicate that nearly 61% of the particulate is caught in the probe and cyclone and thus the average particle diameter is larger than 2 micrometer. The rest of the particulate material is spread proportionally across the stages.

The results on kiln No. 6 indicate that almost 79% of the particulate is caught in the probe and cyclone and thus the average particulate diameter is larger than 3 micrometers. The remainder of the particulate is spread across the stages of the impactor with disproportionately large amounts caught on the second and fifth stages.

The results of the particle sizing are plotted on log-probability paper in Figures 1 and 2.

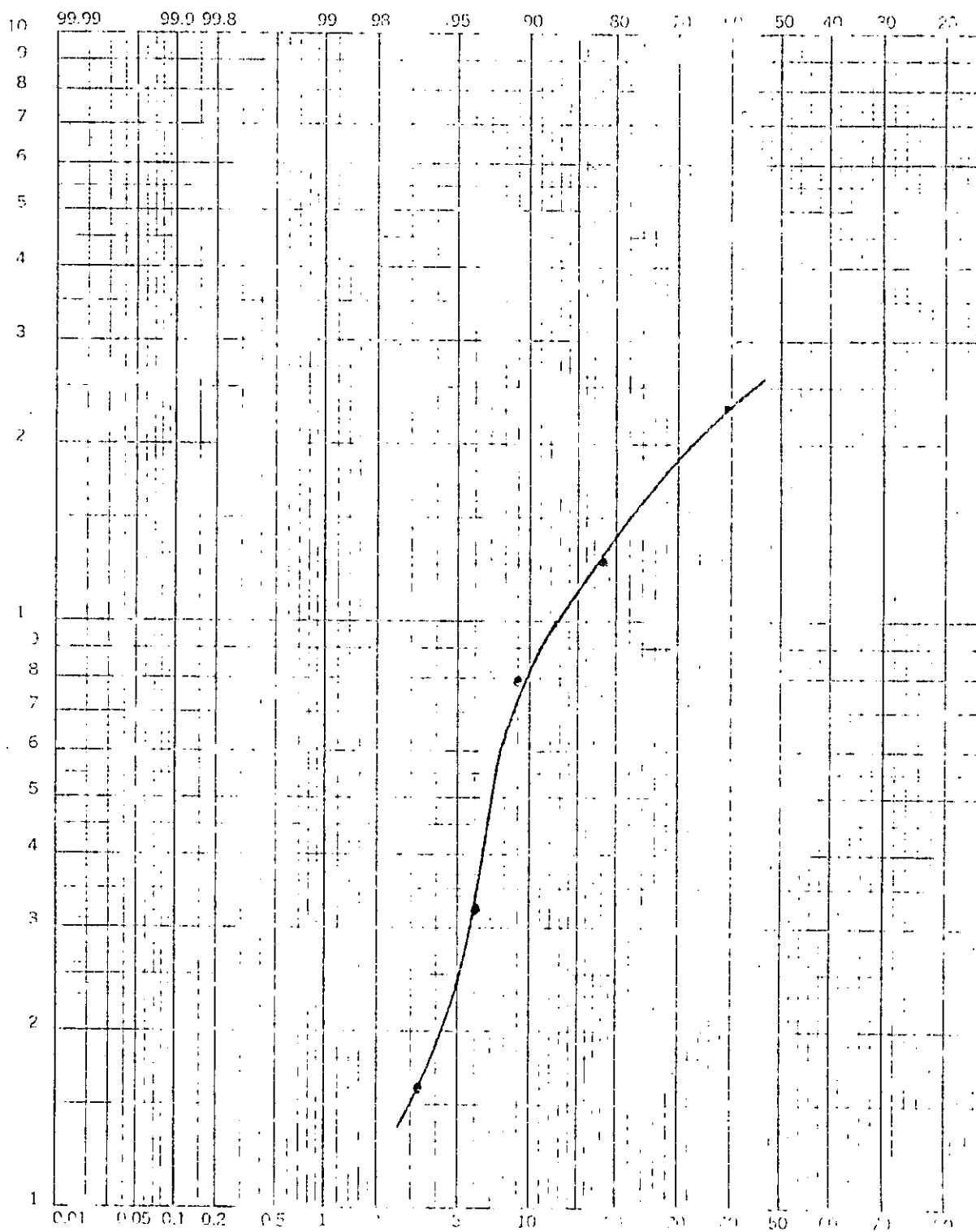


Figure 1. Particle size distribution - kiln 4

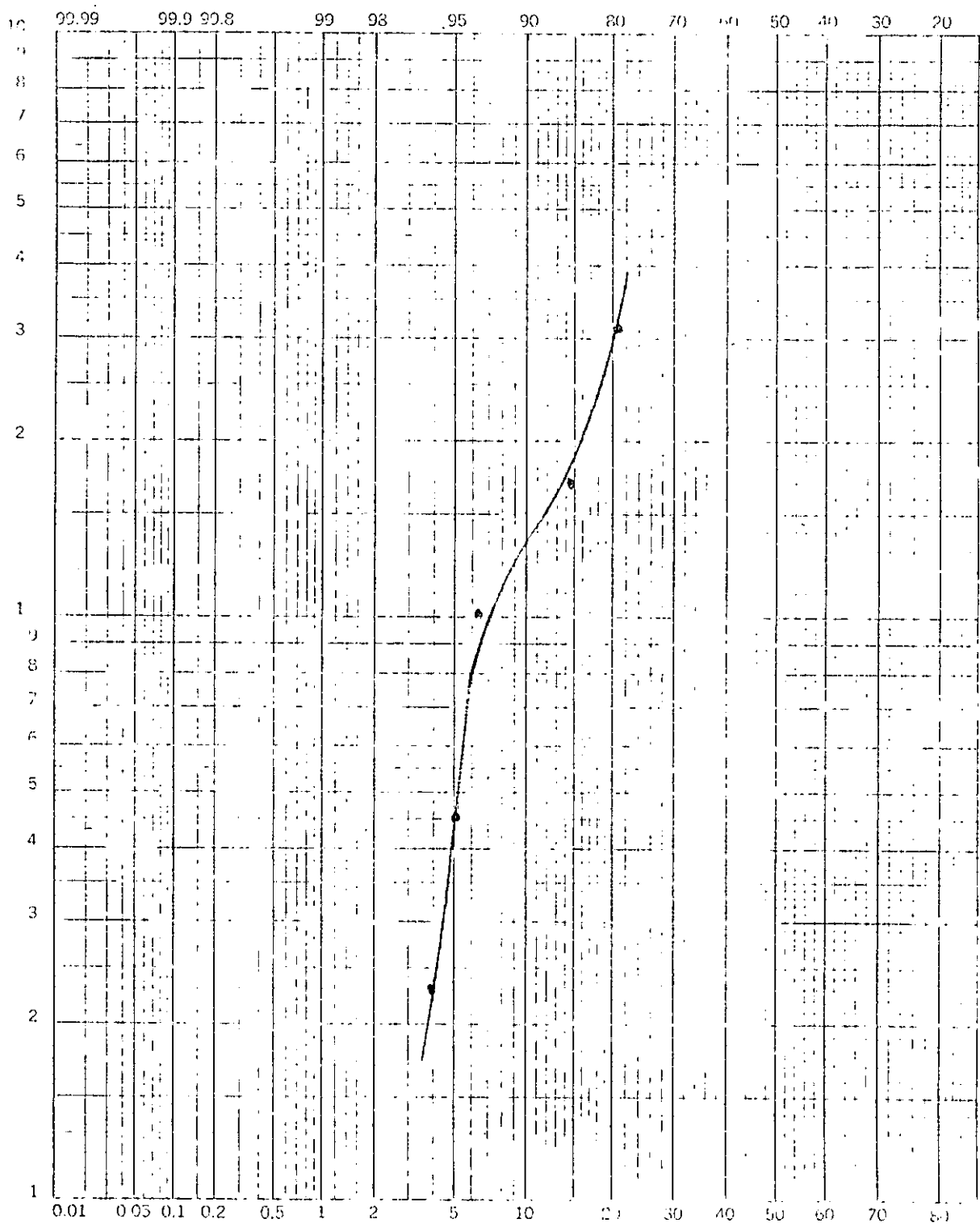


Figure 2. Particle size distribution - kiln 6

SECTION III

PROCESS DESCRIPTION AND OPERATION

The processing at Standard Lime Company involves converting limestone to either dolomitic lime or dead burn dolomite in rotary kilns. The raw material is mainly a combination of calcium and magnesium carbonates and during the process the material is converted to the oxide with loss of carbon dioxide. Dolomitic lime is not a total conversion to the oxide, whereas, the dead burn dolomitic lime is a more complete conversion. In the formation of the dead burn product, iron, in the form of Fe_2O_3 , is added to the feed resulting in fuel saving and a hotter kiln temperature. The amount of iron added is regulated in respect to the amount of fines in the feed - the finer the feed the less iron used, and in addition depends somewhat on the condition of the kiln as the fines and iron tend to coat the interior of the kiln.

The kilns are fired by either natural gas, along with pulverized coal, or pulverized coal alone when the natural gas supply is low. Natural gas generally is used during the April to November period and as a result during the December test period the only fuel was coal. The coal typically is obtained in Ohio and as a result has about a 3% sulfur content.

The plant operates on a continuous basis, 7 days a week, 24 hours a day. During the test program, kiln numbers 4 and 6

were sampled and their capabilities and approximate exit temperatures are as follows:

<u>Kiln No.</u>	<u>Capacity (Tons/day)</u>	<u>Exit Temp. (°F)</u>
4	500	450-750
6	300	700-750

The stone feed to the kilns is quarried, pulverized and classified before it is used. These operations are performed at the plant site. The limestone is fed to kiln No. 4 by a hydraulic ram that has the stated capacity of 794 lb. per stroke. The feed rate is varied by the number of strokes per hour. In kiln No. 6 the limestone is fed by a hopper and conveyor. The size of the opening in the hopper and the speed of the conveyor controls the feed rate to this kiln.

The kilns vary in diameter from 10 feet at the feed entrance to eleven feet at their discharge. The initial 320 feet of feed section are ten feet in diameter. An expansion section, a length of eighteen feet, increases the diameter to eleven feet. The final 100 feet is eleven feet in diameter.

Both kilns were equipped with mechanical collectors to separate large particulate from the gas stream after the gas leaves the kiln and prior to its entrance into the baghouse. Kiln 4 was equipped with a high efficiency (estimated at 81%*) multicyclone. However, the collector on kiln 6 was an older, low efficiency (~40%) device.

The kiln exhaust gases run through straight ducts from the outlets of the cyclones to the inlet of a baghouse that controls emissions from kilns 4, 5, and 6. The baghouse has

*Estimate by plant personnel based on 8 ton/hr out of multicyclone, 2.3 tons/hr passing through, and 1.5 ton/m residual.

22 compartments and has a total of 15,000 bags, each of which is 6" in diameter by 15 feet long. Each compartment has an exhaust fan leading to a 2' x 2' square stack, four feet high.

The gases are cooled before they enter the baghouse in order to protect the bag material. This is accomplished in several ways. Kiln 4 employs a preheater to heat the feed material and at the same time cool the exhaust gases. Kiln 6 employs dilution air to cool the gas via two dampers, one located at the exit of the kiln and the other, a temperature controlled device, in the duct between the multicyclone collector and the baghouse. In both cases, the duct between the multicyclones and the baghouse were long to provide additional cooling of the gases. On kiln 4, this duct was 186'4" long and 5'6" in diameter. On kiln 6, the duct was approximately 188 feet long and 6'9" I.D. Temperatures of 450-750°F occur in the multicyclones of kiln 4, and 800-900°F for kiln 6. The system is designed to reduce the gas temperature to ~400°F at the baghouse entrance.

All testing was done at the inlet to the baghouse and during this period kiln No. 4 was producing dolomitic lime and kiln No. 6 was producing dead burn dolomitic lime. The raw material feed rates to the kilns, the loss on ignition factor and the calculated production for kilns No. 4 and 6 are shown in Table 5. The loss on ignition factor is based on several months experience on relating measured lime production to the amount of feed. This factor is used on a daily basis to calculate production by dividing the amount of feed by the factor.

Flow diagrams of the operations on kilns 4 and 6 are shown in Figures 3 and 4.

Table 5. PRODUCTION DATA

Kiln #4

Date	Time	Coal Feed		Stone Feed		Loss on Ignition Factor	Production Rate tons/hr
		tons/hr	Ram Strokes per hour	lb/stroke	tons/hr		
12/9/75	600-930	4.3	130	794	51.6	2.315	22.3
	930-1135	4.3	150	794	59.6	2.315	25.8
	1135-1300	4.3	130	794	51.6	2.315	22.3
	1300-1700	4.6	150	794	59.6	2.315	25.8
		<u>4.375</u>					<u>24.05</u>

Kiln #6 (A)

12/10/75	930-1600	4.3	B	B	30.0	2.60	11.5
----------	----------	-----	---	---	------	------	------

A. 1.0 ton/hr of iron feed into Kiln 6

B. Stone feed is controlled by hopper opening and conveyor speed.

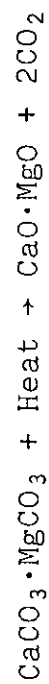
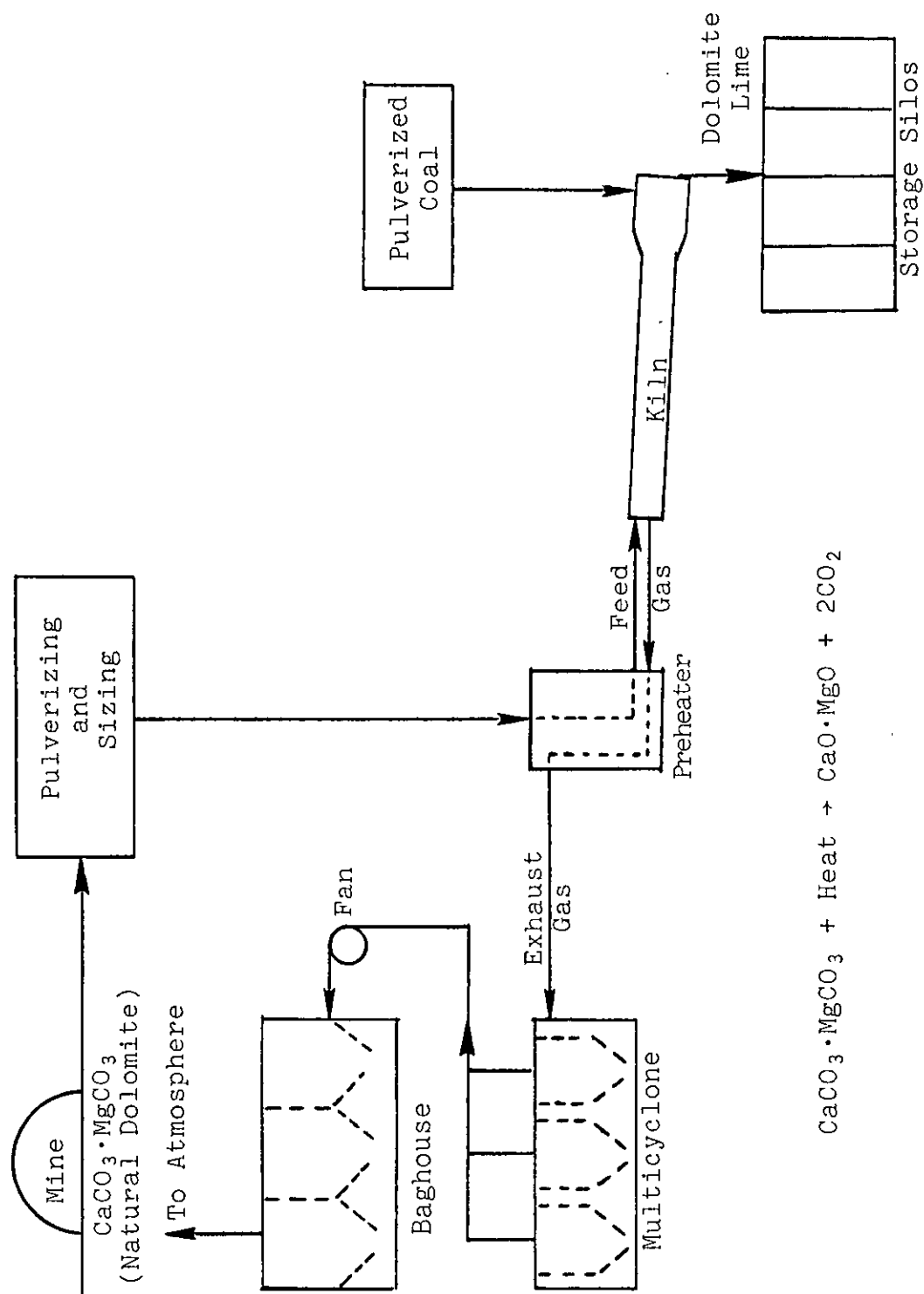


Figure 3. Kiln no. 4 flow diagram

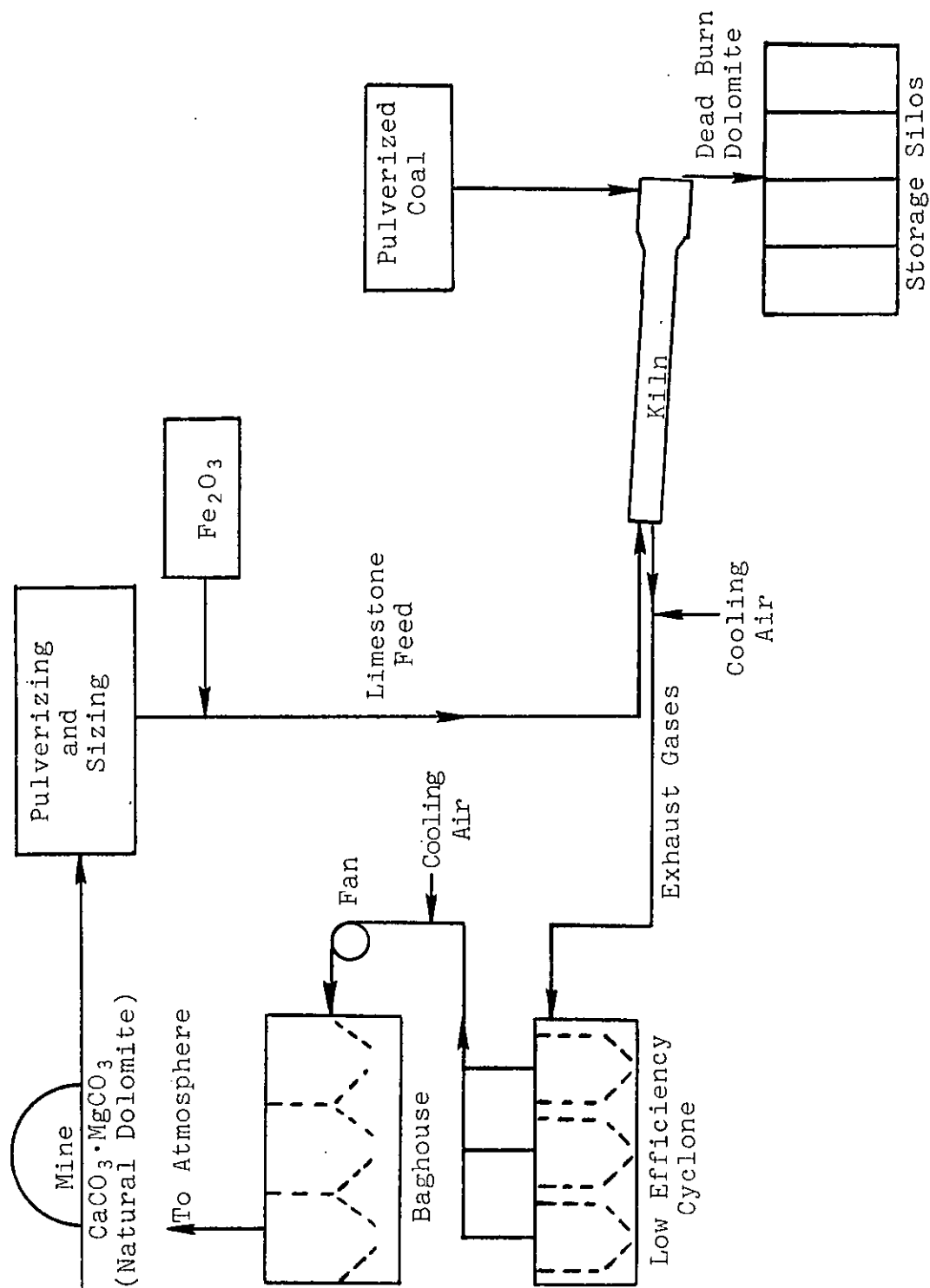


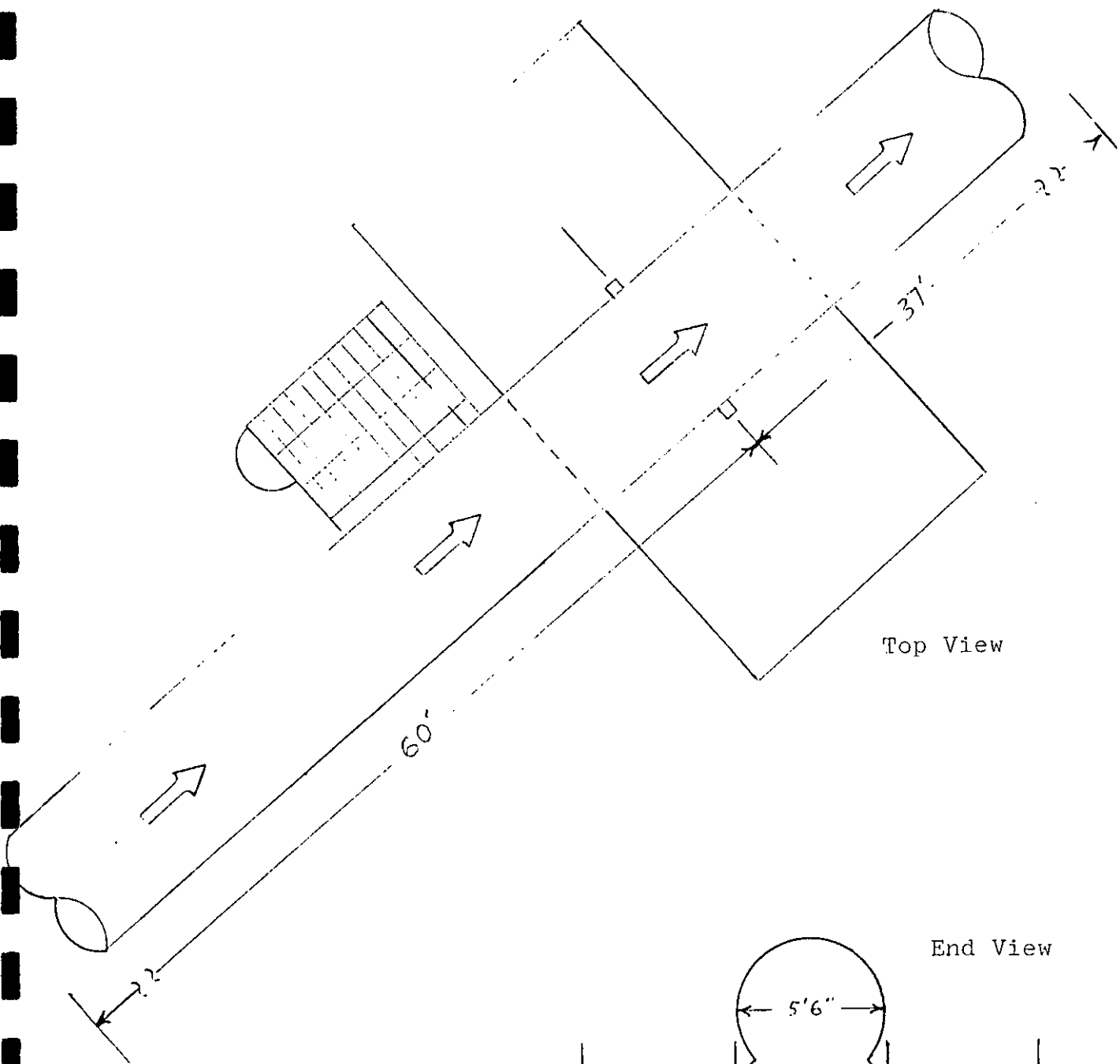
Figure 4. Kiln no. 6 flow diagram

SECTION IV

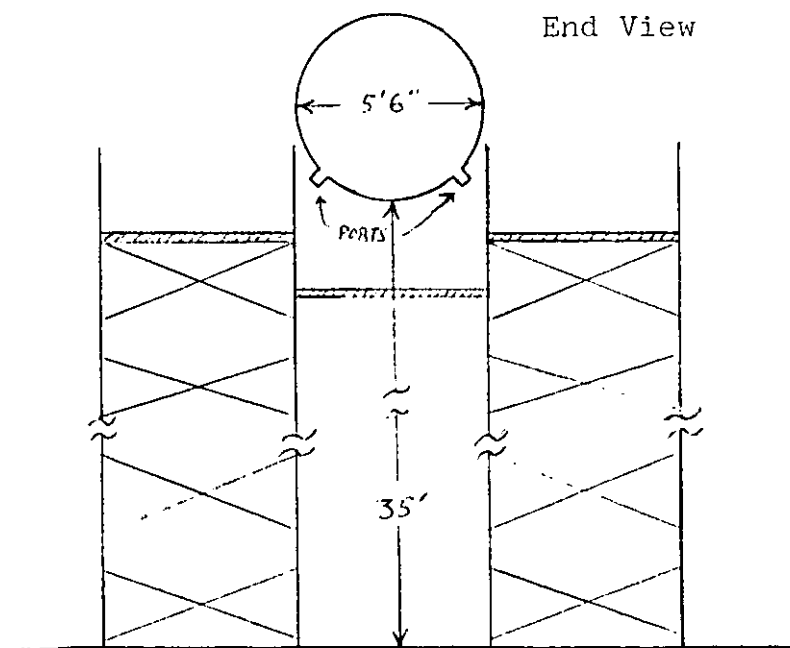
LOCATION OF SAMPLING POINTS

The exhaust duct on kiln No. 4 and the sampling location are shown on Figure 4. The sampling ports (3" pipe couplings welded to the side of the duct) were located 60 feet (18.29 meters or 10.91 diameters) from the nearest upstream disturbance and 37 feet (11.28 meters or 6.73 diameters) from the nearest downstream disturbance. The ports were installed at a 45 degree angle from horizontal so as to make sampling easier and to eliminate a vertical traverse which may collect dust from the bottom of the duct. The sampling platforms on both sides of the duct were 35 feet above ground level on scaffolding. Since the ports were located over 8 diameters downstream from the nearest disturbance and over 2 diameters upstream from the nearest disturbance, only 12 traverse points (6 along each of the two perpendicular diameters) were required (as specified in the Federal Register, Vol. 36, No. 247, December 23, 1971, Method 1, "Sample and Velocity Traverses for Stationary Sources").

The exhaust duct for kiln No. 6 and the sampling location are shown on Figures 5 and 6. The end view is shown in Figure 5 and the top view in Figure 6. The sampling ports (3" pipe couplings welded to the side of the duct) were located 21'9" (6.64 meters or 3.23 diameters) from the nearest upstream disturbance and 30 feet (9.14 meters or 4.44 diameters)



Top View



End View

Figure 5. Kiln Stack #4

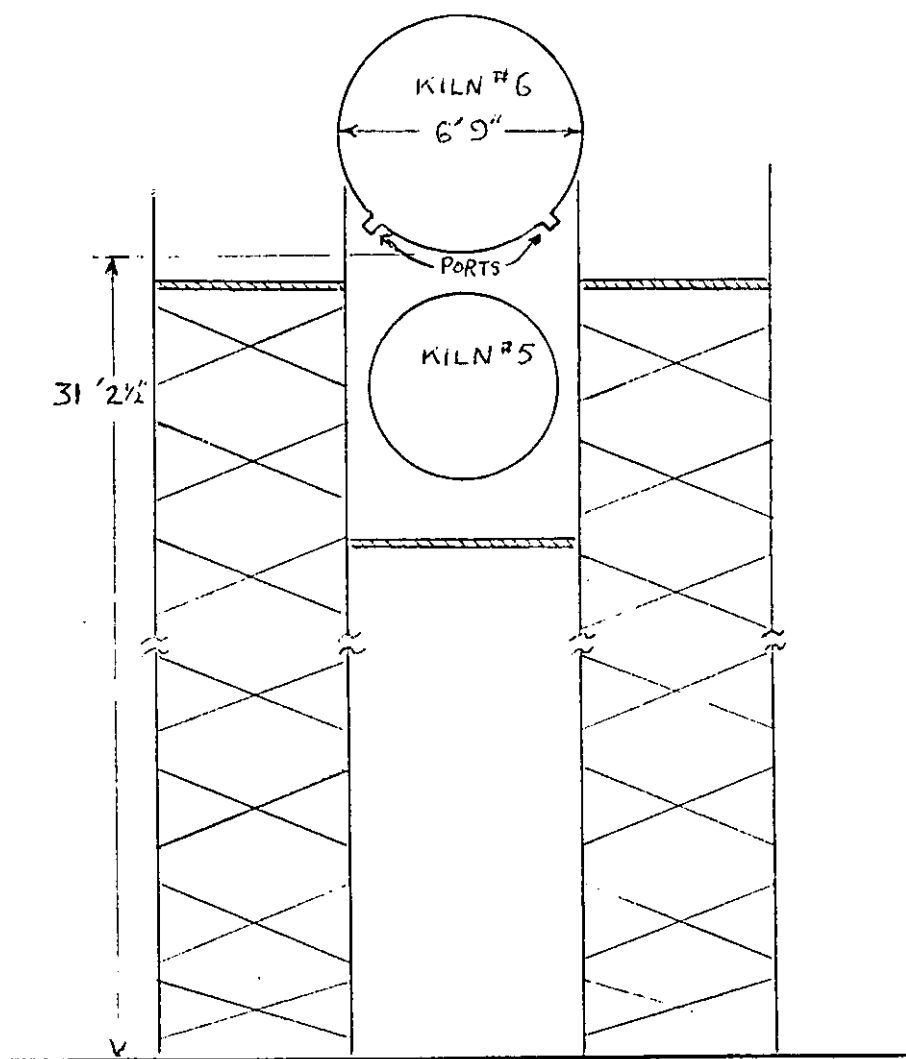


Figure 6. End View of Kiln Stacks 5 and 6

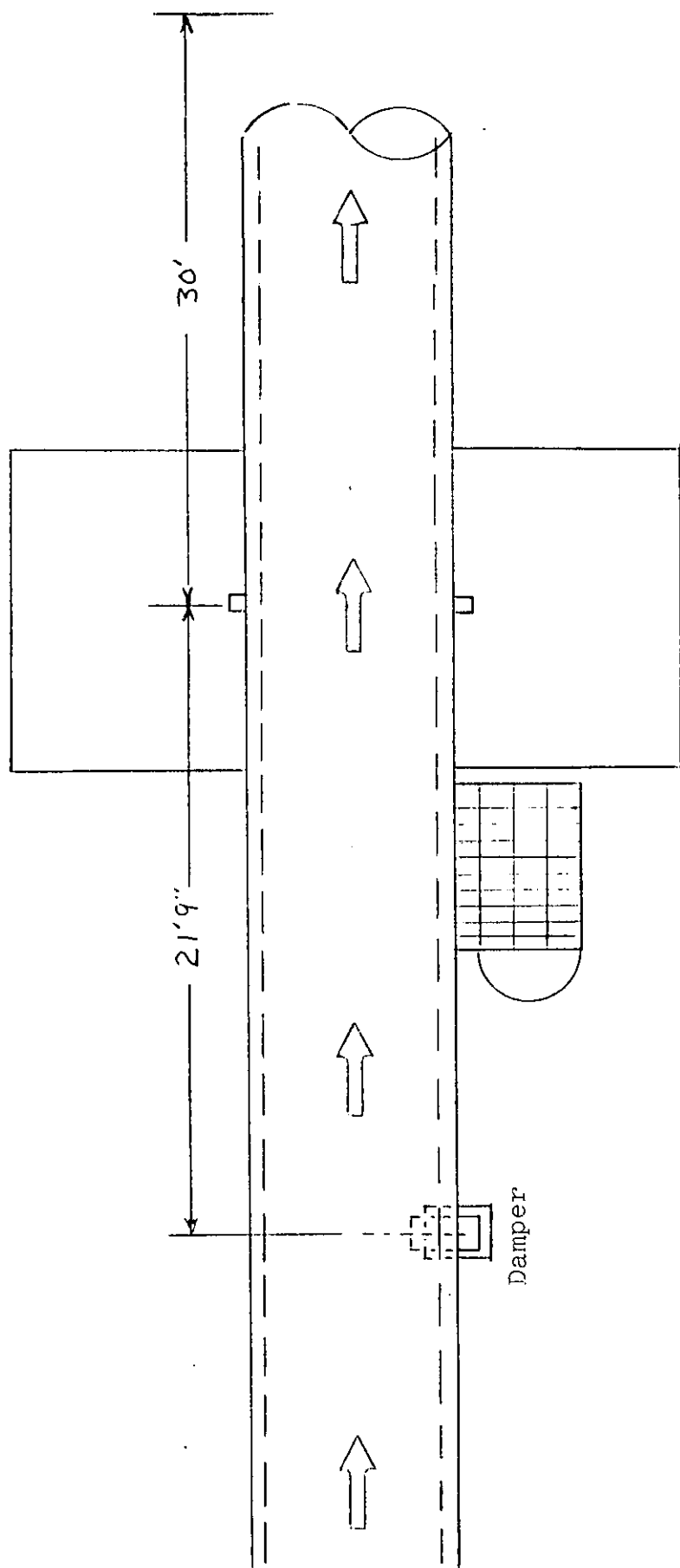


Figure 7. Top View, Kiln Stacks 5 and 6

from the nearest downstream disturbance. The nearest upstream disturbance in this case was a damper mechanism. The ports were installed at a 45 degree angle from the horizontal. The sampling platforms were approximately 30 feet above ground level on scaffolding. Since the ports were located 3.23 diameters downstream from the nearest disturbance, 44 traverse points (22 along each of the two perpendicular diameters) were required (as specified in the above mentioned Federal Register, Method 1). Information obtained on the presurvey indicated that the ports would be closer to the damper (2.5 diameters) and, therefore, a 48 point traverse would be required. The sampling runs were made using the 48 point traverse layout rather than the 44 point layout. The extra 4 sampling points will have no effect on the results. This error occurred due to the fact that the ports were not installed prior to the presurvey and the location suggested by the presurvey team was not followed when the ports were installed.

SECTION V

SAMPLING AND ANALYTICAL PROCEDURES

Sampling Procedures

Gas velocities were measured with a calibrated type S pitot tube and inclined manometer. Velocities were measured at each sampling point across the stack diameter to determine an average value according to procedures described in the Federal Register, Vol. 36, No. 247, December 23, 1971, Method 2. Temperatures were measured with a Type K (Chromel-Alumel) thermocouple connected to a digital thermometer.

An integrated sample of the stack gases was collected during each particulate test by pumping the gas into a Tedlar plastic bag at the rate of approximately 0.033 cfm. This bag sample was then analyzed with an Orsat analyzer for CO₂, O₂, and CO as described in the above mentioned Federal Register, Method 3.

The sampling of particulate emissions from kilns 5 and 6 was done in accordance with Method 5 as given in the above Federal Register, with two exceptions. The procedure given does not provide for the collection of particulate from the "back half" (behind the filter and including the impingers) of the sampling train. The procedure used for the collection of the "back half" samples is given in "Specifications for

Incinerator Testing at Federal Facilities", U. S. Department of Health, Education and Welfare, October 1967.

The second exception to the Federal Register method was to employ a heated teflon-lined flexible connection between the probe and the sample box containing the filter and impinger portions of the train. This flexible connection has ball joints to match the probe and filter holder and was heated to the same temperature as the probe, the temperature being monitored with a type K thermocouple. After the run was completed, the flexible line was cleaned in the same manner as the probe, and the washings were included in the probe wash portion of the samples.

The particle sizing tests were run using a Brink® BMS-11 Sampler. The sampler was operated according to the accompanying manual (Appendix G). The flow rate through the sampler was determined with the use of the preliminary velocity traverses on each kiln and the various calibration curves provided with the instrument. A predetermined pressure drop across the particle sizer resulted in a calculated velocity and therefore a desired particle size distribution.

Nitrogen oxides were collected in evacuated 2-liter flasks containing 25 ml of a dilute sulfuric acid/hydrogen peroxide absorbing solution. The sampling and analytical procedure was described in Method 7 of the Federal Register. Each flask was evacuated and vacuum tested for one minute, and the initial flask temperature, pressure, and barometric pressure recorded. The sampling probe was inserted into the stack and heated, and the sample flask connected. The 3-way stopcock was turned to the "purge" position and stack gas drawn through the system with a vacuum pump to check for condensation in the probe line. The 3-way stopcock was then

turned to the sample position for 15 to 30 seconds. The flask valve was then closed and disconnected from the probe. The contents were shaken for at least 5 minutes. The flasks were allowed to set for at least 16 hrs. They were then shaken for at least 2 minutes, the final pressure, temperature, and barometric pressure were taken, and the sample was transferred to a storage bottle.

Analytical Procedures

Analytical procedures used on the collected samples generally follow the methods outlined in the Federal Register.

Samples from the Method 5 sampling trains were recovered as outlined in the August 17, 1971 Federal Register. After removal of the filter, all sample exposed surfaces were washed with distilled water or reagent grade acetone as specified. All sample bottles for liquid samples were obtained from Wheaton Scientific, Catalogue No. 219630. Prior to use each of these bottles was acid soaked with 1:1 HNO₃ for one day, rinsed with distilled water and soaked with distilled water for one day.

Analytical procedures for the Method 5 samples follow the Federal Register guidelines, with one exception. Container No. 3 as indicated in the method contains water from the impingers and washing of the glassware of the train. The solution was extracted with chloroform and ether, and then the extracted portion was dried to constant weight, as specified. In addition, the water remaining after extraction was evaporated to dryness at 100°C to constant weight.¹ Both weights

¹See "Specifications for Incinerator Testing at Federal Facilities", U. S. Department of Health, Education, and Welfare publication, October 1967, page 31.

were included in the total mass of particulate. Sample weights from the Method 5 samplers were reported as "front half" (probe washings and filter collection weights) and "total" (front half plus water, chloroform-ether extract and impinger acetone washing weights).

Nitrogen Oxide samples were analyzed using the Phenoldisulfonic acid, photometric analysis described in the above mentioned December 23, 1971 Federal Register, Method 7.

Clean-up and sample collection from the Brink's® particle sizing runs is achieved by cooling and disassembling the impactor, removing the collection plates or substrates, and placing them in plastic containers for shipment back to the laboratory. The impactor is then cleaned thoroughly with methylene chloride solvent. The final filter is removed and placed in its container for shipment. The nozzle, probe, cyclone, and cyclone catch bottle is washed with methylene chloride and all of the washings were saved for subsequent analysis.

Upon arrival at the laboratory the aluminum collection pans were reweighed to determine the amount of material collected. The final filter was also weighed and its weight gain determined. The methylene chloride washes from the probe and cyclone and subsequent stages were evaporated in a tared container and the weight of residue was determined.

APPENDIX A

COMPLETE PARTICULATE RESULTS

Table A-1. METHOD 5 DATA

14 JAN 1976

STANDARD LIME - DIVISION OF MARTIN MARIETTA
PARTICULATE SAMPLING - 6912-12

ABBR	DESCRIPTION (68 DEG F)	UNITS	1-4	2-4	1-6	2-6
TT	DURATION OF RUN	MINUTES	60.0	60.0	72.0	72.0
PB	BAROMETRIC PRESSURE	IN HG	29.40	29.40	29.48	29.48
DELH	AVG ORIFICE PRESS DROP	IN H2O	2.176	2.002	1.073	0.463
VM	VOL DRY GAS(METER CON)	DCF	43.928	45.011	33.244	25.238
TM	AVG GAS METER TEMP	DEG F	55.9	65.5	44.8	40.8
VMSTD	VOL DRY GAS (STD COND)	DSCF	44.43	44.67	34.36	26.25
VW	TOTAL H2O COLLECTED	ML	59.8	22.8	27.1	19.2
VWG	VOL H2O VAPOR(STD CON)	SCF	2.835	1.081	1.285	0.910
PCNTM	PERCENT MOISTURE BY VOL		6.00	2.36	3.60	3.35
MD	MOLE FRACTION DRY GAS		0.940	0.976	0.964	0.966
CO2	PERCENT CO2		18.2	18.2	16.2	16.2
O2	PERCENT O2		10.8	10.8	10.1	10.1
N2	PERCENT N2		71.0	71.0	73.7	73.7
MWD	MOL WT OF DRY GAS		31.3	31.3	31.0	31.0
MW	MOL WT OF STACK GAS		30.5	31.0	30.5	30.6
DELF	AVG STACK VELOCITY HEAD	IN H2O	1.039	0.958	0.538	0.225
CP	PITOT COEFFICIENT		0.850	0.850	0.855	0.855
TS	STACK TEMPERATURE	DEG F	388.	384.	526.	533.
PM	STACK PRESSURE(STATIC)	IN H2O	-2.90	-2.90	-6.00	-6.00
PS	STACK PRESSURE (ABS)	IN HG	29.19	29.19	29.04	29.04
VS	AVG STACK GAS VELOCITY	FPM	4322.	4099.	2967.	2194.
DS	STACK DIAMETER	INCHES	66.00	66.00	81.00	81.00
AS	STACK AREA	SQ IN	3421.2	3421.2	5153.0	5153.0
GS	STACK FLOW RT(DRY STD)	DSCFM	58825.	58218.	53366.	39299.
QA	STACK FLOW RT(ACTUAL)	ACFM	102665.	97372.	106144.	78516.
DN	PROBE TIP DIAMETER	INCHES	0.245	0.245	0.245	0.245
PCT1	PERCENT ISOKINETIC		91.7	93.1	98.1	101.8
MF	PARTICULATE (FRONT)	MG	17751.10	5623.90	11963.50	11277.80
MT	PARTICULATE (TOTAL)	MG	17760.10	5634.90	11995.30	11344.30
CAN	PARTICULATE (FRONT)	GR/DSCF	6.1532	1.9390	5.3622	6.6161
CAO	PARTICULATE (TOTAL)	GR/DSCF	6.1563	1.9428	5.3765	6.6551
CAT	PARTICULATE (FRONT)	GR/ACF	3.5127	1.1551	2.6861	3.2994
CAU	PARTICULATE (TOTAL)	GR/ACF	3.5145	1.1574	2.6932	3.3188
CAW	PARTICULATE (FRONT)	LB/HR	3101.999	967.439	2452.388	2228.229
CAX	PARTICULATE (TOTAL)	LB/HR	3103.572	969.331	2458.906	2241.367

Table A-2
PARTICULATE CALCULATIONS

Example: Run No. 1-4

1. Volume of dry gas sampled at standard conditions (dscf^a)

$$\begin{aligned} \text{VMSTD} &= \frac{17.65 \times \text{VM} \left(\text{PB} + \frac{\text{DELH}}{13.6} \right)}{(\text{TM} + 460)} \\ &= \frac{17.65 \times 43.928 \left(29.40 + \frac{2.176}{13.6} \right)}{(55.9 + 460)} \\ &= 44.43 \text{ dscf} \end{aligned}$$

2. Volume of water vapor at standard conditions (scf^b)

$$\begin{aligned} \text{VWG} &= 0.0645 \times \text{VW} \\ &= 0.0645 \times 43.928 \\ &= 2.835 \end{aligned}$$

3. Percent moisture in stack gas

$$\begin{aligned} \text{PCNTM} &= \frac{100 \times \text{VWG}}{\text{VMSTD} + \text{VWG}} \\ &= \frac{100 \times 2.835}{44.43 + 2.835} \\ &= 6.00 \end{aligned}$$

Table A-2 (Continued)
PARTICULATE CALCULATIONS

4. Mole fraction of dry gas

$$\begin{aligned} MD &= \frac{100 - PCNTM}{100} \\ &= \frac{100 - 6.00}{100} \\ &= 0.940 \end{aligned}$$

5. Molecular weight of dry stack gas

$$\begin{aligned} MWD &= (\% \text{ CO}_2 \times 0.44) + (\% \text{ O}_2 \times 0.32) + [(\% \text{ C} + \% \text{ N}_2) \times 0.28] \\ &= (18.2 \times 0.44) + (10.8 \times 0.32) + [(0 + 71.0) \times 0.28] \\ &= 31.3 \end{aligned}$$

6. Molecular weight of wet stack gas

$$\begin{aligned} MW &= MWD \times MD + 18 (1 - MD) \\ &= 31.3 \times 0.94 + 18 (1 - 0.94) \\ &= 30.5 \end{aligned}$$

Table A-2 (Continued)
PARTICULATE CALCULATIONS

7. Stack gas velocity at stack conditions (fpm^c)

$$VS = 4360 \times \left[\frac{\sum_{i=1}^{i=n} \sqrt{DELP \times (TS + 460)}}{n} \right] \times \left[\frac{1}{PS \times MW} \right]^{\frac{1}{2}}$$

where n = the number of data points

$$= 4360 \times 29.58 \times \left(\frac{1}{29.19 \times 30.5} \right)^{\frac{1}{2}}$$

$$= 4322 \text{ fpm}$$

8. Stack gas volumetric flow rate at standard conditions (dscfm^d)

$$QS = \frac{0.123 \times VS \times AS \times MD \times PS}{TS + 460}$$

$$= \frac{0.123 \times 4322 \times 3421 \times 0.940 \times 29.19}{388 + 460}$$

$$= 58825 \text{ dscfm}$$

9. Stack gas volumetric flow rate at stack conditions (acfm^e)

$$QA = \frac{0.05645 \times QS (TS + 460)}{PS \times MD}$$

$$= \frac{0.05645 \times 58825 \times (388 + 460)}{29.19 \times 0.940}$$

$$= 102665 \text{ acfm}$$

Table A-2 (Continued)
PARTICULATE CALCULATIONS

10. Area of nozzle (sq. ft.)

$$\begin{aligned} AN &= 54.54 \times 10^{-4} (DN)^2 \\ &= 54.54 \times 10^{-4} (0.245)^2 \\ &= 3.27 \times 10^{-4} \text{ sq. ft.} \end{aligned}$$

11. Percent Isokinetic

$$\begin{aligned} PCTI &= \frac{100.0(TS + 460) \left[0.00267(VW) + \frac{VM}{(TM + 460)} (PB + \frac{DELH}{13.6}) \right]}{(TT)(VS)(PS)(AN)} \\ &= \frac{100.0(388 + 460) \left[0.00267(59.8) + \frac{43.928}{(55.9 + 460)} (29.40 + \frac{2.176}{13.6}) \right]}{(60)(4322)(29.19)(3.27 \times 10^{-4})} \\ &= 91.7 \end{aligned}$$

12. Particulate - probe, cyclone, and filter (gr/dscf)

$$\begin{aligned} CAN &= 0.0154 \times \frac{MF}{VMSTD} \\ &= 0.0154 \times \frac{17751}{44.43} \\ &= 6.15 \text{ gr/dscf} \end{aligned}$$

Table A-2 (Continued)
PARTICULATE CALCULATIONS

13. Particulate - total (gr/dscf)

$$\begin{aligned} \text{CAO} &= 0.0154 \times \frac{\text{MT}}{\text{VMSTD}} \\ &= 0.0154 \times \frac{17760}{44.43} \\ &= 6.156 \text{ gr/dscf} \end{aligned}$$

14. Particulate - probe, cyclone, and filter at stack conditions (gr/acf)

$$\begin{aligned} \text{CAT} &= \frac{17.65 \times \text{CAN} \times \text{PS} \times \text{MD}}{(\text{TS} + 460)} \\ &= \frac{17.65 \times 6.15 \times 29.19 \times 0.94}{(388 + 460)} \\ &= 3.5127 \text{ gr/acf} \end{aligned}$$

15. Particulate - total at stack conditions (gr/acf)

$$\begin{aligned} \text{CAU} &= \frac{17.65 \times \text{CAO} \times \text{PS} \times \text{MD}}{(\text{TS} + 460)} \\ &= \frac{17.65 \times 6.156 \times 29.19 \times 0.94}{(388 + 460)} \\ &= 3.5145 \text{ gr/acf} \end{aligned}$$

Table A-2 (Continued)
PARTICULATE CALCULATIONS

16. Particulate - probe, cyclone, and filter (lb/hr)

$$\begin{aligned} \text{CAW} &= 0.00857 \times \text{CAN} \times \text{QS} \\ &= 0.00857 \times 6.1532 \times 58825 \\ &= 3101.999 \text{ lb/hr} \end{aligned}$$

17. Particulate - total (lb/hr)

$$\begin{aligned} \text{CAX} &= 0.00857 \times \text{CAO} \times \text{QS} \\ &= 0.00857 \times 6.1563 \times 58825 \\ &= 3103.571 \text{ lb/hr} \end{aligned}$$

^aDry standard cubic feet @ 68°F, 29.92 in. Hg

^bStandard cubic feet @ 68°F, 29.92 in. Hg

^cFeet per minute @ stack conditions; n = number of data points

^dDry standard cubic feet per minute @ 68°F, 29.92 in. Hg

^eActual cubic feet per minute @ stack conditions

APPENDIX B

PARTICLE SIZE DATA FROM BRINK® BMS-11

CASCADE IMPACTOR PARTICLE SIZE DISTRIBUTION FOR RUN 1-4

STAGE	WT OF MATERIAL	DPC	MG/ACM	WT PCNT	CUM WT PCNT
CYCLONE	60.600		4.97	60.72	100.00
1	21.100	2.30	1.73	21.14	39.28
2	9.100	1.26	0.75	9.12	18.14
3	2.900	0.79	0.24	2.91	9.02
4	2.800	0.32	0.23	2.81	6.11
5	2.600	0.16	0.21	2.61	3.31
FILTER	0.700		0.06	0.70	0.70

Table B-2. PARTICLE SIZE DISTRIBUTION FOR RUN 1-4 (ENGLISH)

CASCADE IMPACTOR PARTICLE SIZE DISTRIBUTION FOR RUN 1-4						
INPUT VARIABLE	UNITS	INPUT DATA				
SAMPLING TIME	MIN	5.0				
PRESSURE DROP	IN HG	1.80				
STATIC PRESSURE	IN H2O	-4.36				
PARTICLE DENSITY	G/CC	3.50				
BAROMETRIC PRESSURE	IN HG	29.40				
GAS MOL WT		30.7				
GAS TEMPERATURE	DEG F	128.0				
GAS VISCOSITY	POISE	0.00025				
GAS DENSITY	G/CC	0.00017				
STATE	WT OF MATERIAL	DPC	MG/ACF	WT PCNT	CUM WT PCNT	
CYCLONE	60.600		175.35	60.72	100.00	
1	21.100	2.30	61.05	21.14	39.28	
2	9.100	1.26	26.33	9.12	18.14	
3	2.900	0.79	8.39	2.91	9.02	
4	2.800	0.32	8.10	2.81	6.11	
5	2.600	0.16	7.52	2.61	3.31	
FILTER	0.700		2.03	0.70	0.70	

CASCADE IMPACTOR PARTICLE SIZE DISTRIBUTION FOR RUN 1-6

STAGE	WT OF MATERIAL	DPC	MG/ACM	WT PCNT	CUM WT PCNT
CYCLONE	88.400		11.60	78.84	100.00
1	7.360	3.09	0.97	6.56	21.16
2	9.220	1.70	1.21	8.22	14.59
3	1.460	1.07	0.19	1.30	6.37
4	1.230	0.45	0.16	1.10	5.07
5	3.650	0.23	0.48	3.26	3.97
FILTER	0.800		0.10	0.71	0.71

Table B-4. PARTICLE SIZE DISTRIBUTION FOR RUN 1-6 (ENGLISH)

CASCADE IMPACTOR PARTICLE SIZE DISTRIBUTION FOR RUN 1-6						
STATE	WT OF MATERIAL	DPC	MG/ACF	WT PCNT	CUM WT PCNT	
CYCLONE	88.400		409.67	78.84	100.00	
1	7.360	3.09	34.11	6.56	21.16	
2	9.220	1.70	42.73	8.22	14.59	
3	1.460	1.07	6.77	1.30	6.37	
4	1.230	0.45	5.70	1.10	5.07	
5	3.650	0.23	16.92	3.26	3.97	
FILTER	0.800		3.71	0.71	0.71	

APPENDIX C

COMPLETE NO_x RESULTS AND SAMPLE CALCULATION

Table C-1. NO_x RESULTS

NOX CALCULATION (68 DEG F)

STANDARD LINE COMPANY
16 JANUARY 1976

VSC - SAMPLE VOLUME AT STANDARD CONDITIONS (DRY BASIS), ML
MASS - MASS OF NO₂ IN GAS SAMPLE, UG
PPM - PPM NO_x AS NO₂
LB/DSCF - CONCENTRATION OF NO_x AS NO₂ (DRY BASIS), LB/DSCF

VSC	MASS	PPM	LB/DSCF	KG/HR	G/NCM	LB/HR
0.174581E+04	0.456397E+02	0.136986E+02	0.162083E-05	0.258155E+01	0.259603E-01	0.569125E+01
0.174188E+04	0.230191E+03	0.692473E+02	0.819339E-05	0.130499E+02	0.131231E+00	0.287696E+02
0.178052E+04	0.115031E+03	0.338532E+02	0.400553E-05	0.637975E+01	0.641554E-01	0.140647E+02
0.180956E+04	0.934312E+03	0.270552E+03	0.320119E-04	0.403670E+02	0.512724E+00	0.889924E+02
0.160093E+04	0.970096E+03	0.282260E+03	0.333971E-04	0.421138E+02	0.534911E+00	0.928434E+02
0.175603E+04	0.931359E+03	0.277918E+03	0.328835E-04	0.414660E+02	0.526684E+00	0.914154E+02

Table C-2

METHOD 7NO_x CALCULATIONSPlant Standard LimeDate 1-12-76Test No. Sample Calculation Test #4 Flask #15

Volume of flask and value (ml)	= VF =	<u>2112</u>
Initial absolute pressure of flask (in Hg)	= PI =	<u>2.60</u>
Final absolute pressure of flask (in Hg)	= PF =	<u>28.46</u>
Initial temperature of flask (°F)	= TI =	<u>63</u>
Final temperature of flask (°F)	= TF =	<u>72</u>
Mass of NO _x as NO ₂ in gas sample (μg)	= M =	<u>934</u>

Volume of sample at standard conditions, dry basis(ml) = VS

$$VS = \frac{17.65}{\cancel{17.71}} \times (VF - 25) \left(\frac{PF}{TF + 460} - \frac{PI}{TI + 460} \right)$$

$$VS = \frac{17.65}{\cancel{17.71}} \times (2112 - 25) \left(\frac{28.46}{72 + 460} - \frac{2.60}{63 + 460} \right) = \underline{1784} \text{ ml}$$

Concentration of NO_x as NO₂ (dry basis) (lbs/scf) = C

$$C = 6.2 \times 10^{-5} \times \left(\frac{M}{VS} \right)$$

$$C = 6.2 \times 10^{-5} \times \left(\frac{934}{1784} \right) = \underline{3.246 \times 10^{-5}} \text{ lbs/scf}$$

$$PPM = \frac{24.5}{MW} \times 1.6018 \times 10^7 \times \frac{\#}{scf}$$

$$PPM = \frac{24.5}{46} \times 1.6018 \times 10^7 \times 3.246 \times 10^{-5} = \underline{277} \text{ ppm}$$

APPENDIX D

FIELD TEST DATA SHEETS

TRAVERSE POINT LOCATION FOR CIRCULAR DUCTS

Standard Lime Div. of

PLANT Martin-Marietta Woodville

DATE 12/8/75

SAMPLING LOCATION #4 k.l.m.

INSIDE OF FAR WALL TO

OUTSIDE OF NIPPLE, (DISTANCE A) 5 11/4

INSIDE OF NEAR WALL TO

OUTSIDE OF NIPPLE, (DISTANCE B) 54

STACK I.D., (DISTANCE A - DISTANCE B)

NEAREST UPSTREAM DISTURBANCE >12' (20)

NEAREST DOWNSTREAM DISTURBANCE >45' (810)**CALCULATOR**

SCHEMATIC OF SAMPLING LOCATION

Pilot #5A

[illegible]

PRELIMINARY VELOCITY TRAVERSE

PLANT Standard Line - Warville

DATE 12-8-75

LOCATION #4 Kila

STACK I.D. 5' 6"

BAROMETRIC PRESSURE, in. Hg 29.60

STACK GAUGE PRESSURE, in. H₂O ~~4~~ 2.90

OPERATORS Taylor - Etter - Hughes

SCHEMATIC OF TRAVERSE POINT LAYOUT

South East Port

Patient # 5A
Cp = 0.85

Northwest Port

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_s) , in. H ₂ O	STACK TEMPERATURE (T_s) , °F
1	1.20	401
2	1.05	411
3	1.10	416
4	1.20	418
5	1.25	395
6	1.00	398
AVERAGE		

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_s), in. H ₂ O	STACK TEMPERATURE (T_s), °F
7	0.80	390
8	1.30	404
9	1.30	411
10	1.10	413
11	1.15	415
12	1.10	413
AVERAGE	0.06 1.13	407

NOMOGRAPH DATA

PLANT Standard Line - Woodville

DATE 12/8/75

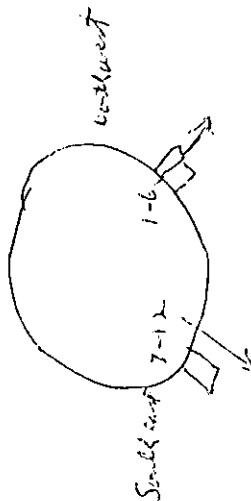
SAMPLING LOCATION #4 Kelm

Pitot # 5A : 0.85 = C_p

CALIBRATED PRESSURE DIFFERENTIAL ACROSS ORIFICE, in. H ₂ O	ΔH_0	1.95
AVERAGE METER TEMPERATURE (AMBIENT + 20 °F), °F	$T_{m\text{ avg.}}$	50
PERCENT MOISTURE IN GAS STREAM BY VOLUME	B_{wo}	10
BAROMETRIC PRESSURE AT METER, in. Hg	P_m	29.60
STATIC PRESSURE IN STACK, in. Hg ($P_m \pm 0.073 \times$ STACK GAUGE PRESSURE in in. H ₂ O)	P_s	29.81 0.910
RATIO OF STATIC PRESSURE TO METER PRESSURE	P_s/P_m	1.007
AVERAGE STACK TEMPERATURE, °F	$T_{s\text{ avg.}}$	407
AVERAGE VELOCITY HEAD, in. H ₂ O	$\Delta p_{\text{ avg.}}$	1.13
MAXIMUM VELOCITY HEAD, in. H ₂ O	$\Delta p_{\text{ max.}}$	1.30
C FACTOR		0.94
CALCULATED NOZZLE DIAMETER, in.		0.228
ACTUAL NOZZLE DIAMETER, in.		.245?
REFERENCE Δp , in. H ₂ O		0.84

PLANT Waverlyville - Standard Oil
DATE 12-9-75
SAMPLING LOCATION K.L.W #4
SAMPLE TYPE EPA-5
RUN NUMBER 1-4
OPERATOR J. Kautzke & E. H. H.
AMBIENT TEMPERATURE 30°F
BAROMETRIC PRESSURE 27.40
STATIC PRESSURE, (P_s) -2.42" H₂O
FILTER NUMBER (S) 53-6 + 53

PROBE LENGTH AND TYPE 6' gless
NOZZLE I.D. 0.245"
ASSUMED MOISTURE, % 10
SAMPLE BOX NUMBER 3
METER BOX NUMBER 6
METER ΔH 1.91
G-FACTOR 2.41 ΔE 2.3K
PROBE HEATER SETTING 14
HEATER BOX SETTING H1
REFERENCE ΔH 1.13



SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 5 MINUTES LEAK RATE 100 CFM @ 15" HG

[illegible]

CLEAN-UP DATA

Plant MARTIN MARIETTA Comments:

Date 12/9/75

Sampling Location #4 Kiln

Sample Type EPA-5

Run Number 1-4

Sample Box Number 3

Clean-up Man ETTER & HUGHES

IMPINGERS

	#1	#2	#3	#4	#5
Final Vol.	150 ml	100 ml	0 ml	ml	ml
Initial Vol.	100 ml	100 ml	0 ml	ml	ml
Net Vol.	50 ml	0 ml	0 ml	ml	ml

Total Net Volume in Impingers 50 ml

SILICA GEL

Final Weight	209.8 g	g	g
Initial Weight	200.0 g	g	g
Net Weight	9.8 g	g	g

Total Net Weight in Silica Gel 9.8 g

Total Moisture 59.8 g

Filter Number(s) 53-6 _____
53-7 _____

DRY MOLECULAR WEIGHT DETERMINATION

COMMENTS:

PLANT Standard Lime Woodville
 DATE 12/9/75
 SAMPLING TIME (24-hr CLOCK) 1300
 SAMPLING LOCATION #4 Kiln
 SAMPLE TYPE (BAG, INTEGRATED, CONTINUOUS) Integrated
 ANALYTICAL METHOD O₂ set
 AMBIENT TEMPERATURE 72°F
 OPERATOR Thalman

RUN GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) M _D , lb/lb-mole
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	18.2	18.2	18.2	18.2%				44/100	8.008
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)	29	10.8%	29	10.8%				32/100	3.456
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)	29	0%	29	0%				28/100	
N ₂ (NET IS 100 MINUS ACTUAL CO READING)				71%				28/100	19.88
TOTAL									31.344

ISOKINETIC CALCULATION

Run# 1-4, Date 12-9-75

Location Kiln #4 at P.L., Initials J.M.

MOISTURE

Net volume of liquid collected
in impingers and silica gel

$$= V_L = \underline{59.8} \text{ ml}$$

Net volume of gas through dry
gas meter at meter conditions

$$= V_m = \underline{43.928} \text{ cu.ft.}$$

Barometric Pressure - absolute

$$= BP = \underline{29.40} \text{ in.Hg}$$

Average absolute meter tem-
perature ($^{\circ}\text{F} + 460$)

$$= T_m = \underline{516} \text{ }^{\circ}\text{R}$$

Percent moisture = M

$$M = \frac{100}{1 + \frac{373.63 (V_m) (BP)}{(T_m) (V_L)}} + 2.5$$

$$= \frac{100}{1 + \frac{373.63 (43.928) (29.40)}{(516) (59.8)}} + 2.5 = \underline{8.51} \%$$

MOLECULAR WEIGHT

Percent O_2 by volume dry basis

$$= \text{O}_2 = \underline{\hspace{2cm}} \%$$

" CO " " " "

$$= \text{CO} = \underline{\hspace{2cm}} \%$$

" CO_2 " " " "

$$= \text{CO}_2 = \underline{\hspace{2cm}} \%$$

$$\text{N}_2 = 100 - (\text{O}_2 + \text{CO}_2 + \text{CO}) = 100 - ($$

$$+ \quad + \quad)$$

Percent N_2 by volume dry basis

$$= \text{N}_2 = \underline{\hspace{2cm}} \%$$

Dry molecular weight = MD

$$\text{MD} = 0.44 (\text{CO}_2) + 0.32 (\text{O}_2) + 0.28 (\text{N}_2 + \text{CO})$$

$$\text{MD} = 0.44 (\quad) + 0.32 (\quad) + 0.28 (\quad + \quad)$$

$$\text{MD} = \underline{31.344}$$

ISOKINETIC CALCULATION (Cont.d)

Run # 1-4

MOLECULAR WEIGHT (Cont.d)

Stack gas molecular weight = MS

$$MS = (MD) [1 - (M \times 10^{-2})] + 18 (M \times 10^{-2})$$

$$MS = (57.31344) [1 - (0.0851)] + 18 (0.0851) = \underline{30.21}$$

VELOCITY

Static pressure of stack = p = -2.90 H₂O

Pitot tube coefficient, type S = CP = 0.85

Average Absolute Stack Temp. (°F + 460) = T_s = 848 °R

Average of the square root of Δp = $\sqrt{\Delta p}$ = 1.02 (in. H₂O)^{1/2}

Absolute pressure of stack (BP ± $\frac{p}{13.6}$) = P_s = 29.19 in. Hg

Stack gas velocity = V_s

$$V_s = (85.48) (CP) \left[\sqrt{\frac{(T_s)}{(P_s) (MS)}} \right] \left[\sqrt{\Delta p} \right]$$

$$V_s = (85.48) (0.85) \left[\sqrt{\frac{(848)}{(29.19) (30.21)}} \right] \left[1.02 \right] = \underline{72.676} \text{ FPS}$$

ISOKINETIC PERCENT

Average pressure drop across orifice = ΔH = 2.37 in. H₂O

Total sampling time = T = 60 min.

Diameter of nozzle = d = 0.245 in.

Area of nozzle = A

$$A = 54.54 \times 10^{-4} (d^2) = 54.54 \times 10^{-4} (0.245)^2 = \underline{0.00033} \text{ sq. ft.}$$

I = percent Isokinetic

$$I = (1.667) (T_s) \left[0.00267 (V_L) + \frac{(V_m)}{(T_m)} \left(PB + \frac{\Delta H}{13.6} \right) \right]$$

$$(T) (V_s) (P_s) (A)$$

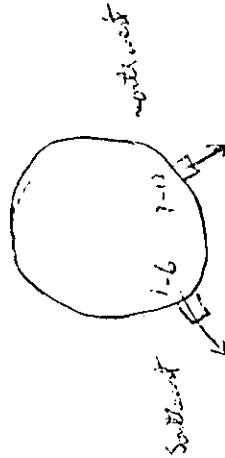
$$I = (1.667) (848) \left[0.00267 (59.8) + \frac{(43.908)}{(516)} \left(29.4 + \frac{2.37}{13.6} \right) \right]$$

$$(60) (72.676) (29.19) (0.00033)$$

$$I = \underline{90.1} \%$$

PLANT Standard Lime-woodville

DATE 12/9/75
SAMPLING LOCATION Kiln #4
SAMPLE TYPE EPA-5
RUN NUMBER 2-4
OPERATOR Thelma-Hughes
AMBIENT TEMPERATURE 35°F
BAROMETRIC PRESSURE 39.40
STATIC PRESSURE, (P_s) -2.9"H₂O
FILTER NUMBER (S) 52-3



PROBE LENGTH AND TYPE 5.6 ft
NOZZLE I.D. .245
ASSUMED MOISTURE, % 10
SAMPLE BOX NUMBER IVRC
METER BOX NUMBER IVRC
METER ΔH 1.91
C FACTOR
PROBE HEATER SETTING 14.5
HEATER BOX SETTING HI
REFERENCE Δp 1.1302 2.38 ΔH

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 5 MINUTES LEAK RATE .01 CFM @ 23"Hg

[illegible]

CLEAN-UP DATA

Plant MARTIN MARIETTA Comments:
Date 12/9/75
Sampling Location #4 KILN
Sample Type EPA 5
Run Number 2-4
Sample Box Number 6
Clean-up Man ETTER & HUGHES

IMPINGERS

	#1		#2		#3		#4		#5
Final Vol.	114	ml	100	ml	0	ml		ml	
Initial Vol.	100	ml	100	ml	0	ml		ml	
Net Vol.	14	ml	0	ml	0	ml		ml	

Total Net Volume in Impingers 14 ml

SILICA GEL

Final Weight	208.8	g		g		g
Initial Weight	200.0	g		g		g
Net Weight	8.8	g		g		g

Total Net Weight in Silica Gel 8.8 g

Total Moisture 22.8 g

Filter Number(s) 53-8

ISOKINETIC CALCULATION

Run# 2-4/, Date 12-9-75

Location KLn #4 Std Line, Initials JA

MOISTURE

Net volume of liquid collected

in impingers and silica gel

$$= V_L = \underline{22.8} \text{ ml}$$

Net volume of gas through dry

gas meter at meter conditions

$$= V_m = \underline{45.011} \text{ cu.ft.}$$

Barometric Pressure - absolute

$$= BP = \underline{29.40} \text{ in.Hg}$$

Average absolute meter tem-

perature ($^{\circ}\text{F} + 460$)

$$= T_m = \underline{525.5} \text{ }^{\circ}\text{R}$$

Percent moisture = M

$$M = \frac{100}{1 + \frac{373.63 (V_m) (BP)}{(T_m) (V_L)}} + 2.5$$

$$= \frac{100}{1 + \frac{373.63 (45.011) (29.40)}{(525.5) (22.8)}} + 2.5 = \underline{4.87} \%$$

MOLECULAR WEIGHT

Percent O_2 by volume dry basis

$$= \text{O}_2 = \underline{\hspace{2cm}} \%$$

" CO " " " "

$$= \text{CO} = \underline{\hspace{2cm}} \%$$

" CO_2 " " " "

$$= \text{CO}_2 = \underline{\hspace{2cm}} \%$$

$$\text{N}_2 = 100 - (\text{O}_2 + \text{CO}_2 + \text{CO}) = 100 - ($$

$$+ \quad + \quad)$$

Percent N_2 by volume dry basis

$$= \text{N}_2 = \underline{\hspace{2cm}} \%$$

Dry molecular weight = MD

$$\text{MD} = 0.44 (\text{CO}_2) + 0.32 (\text{O}_2) + 0.28 (\text{N}_2 + \text{CO})$$

$$\text{MD} = 0.44 (\quad) + 0.32 (\quad) + 0.28 (\quad + \quad)$$

$$\text{MD} = \underline{29.31344}$$

ISOKINETIC CALCULATION (Cont.d)

Run # 2-4

MOLECULAR WEIGHT (Cont.d)

Stack gas molecular weight = MS

$$MS = (MD) [1 - (M \times 10^{-2})] + 18 (M \times 10^{-2})$$

$$MS = (31.344) [1 - (0.0487)] + 18 (0.0487) = \underline{30.69}$$

VELOCITY

Static pressure of stack = p = -2.90 H₂O

Pitot tube coefficient, type S = CP = 0.85

Average Absolute Stack Temp. (°F + 460) = T_s = 844 °R

Average of the square root of Δp = $\sqrt{\Delta p}$ = 0.97 (in. H₂O)^{1/2}

Absolute pressure of stack (BP ± $\frac{p}{13.6}$) = P_s = 29.19 in. Hg

Stack gas velocity = V_s

$$V_s = (85.48) (CP) \left[\sqrt{\frac{(T_s)}{(P_s) (MS)}} \right] \left[\sqrt{\Delta p} \right]$$

$$V_s = (85.48) (0.85) \left[\sqrt{\frac{(844)}{(29.19) (30.69)}} \right] \left[0.97 \right]$$

$$= \underline{68.41} \text{ FPS}$$

ISOKINETIC PERCENT

Average pressure drop across orifice = ΔH = 2.00 in. H₂O

Total sampling time = T = 60 min.

Diameter of nozzle = d = 0.245 in.

Area of nozzle = A

$$A = 54.54 \times 10^{-4} (d^2) = 54.54 \times 10^{-4} ()^2 = \underline{0.00033} \text{ sq. ft.}$$

I = percent Isokinetic

$$I = (1.667) (T_s) \left[0.00267 (V_L) + \left(\frac{V_m}{T_m} \right) \left(PB + \left(\frac{\Delta H}{13.6} \right) \right) \right]$$

$$(T) (V_s) (P_s) (A)$$

$$I = (1.667) (844) \left[0.00267 (22.8) + \left(\frac{45.011}{525.5} \right) \left(29.4 + \left(\frac{2}{13.6} \right) \right) \right]$$

$$(60) (68.41) (29.19) (0.00033)$$

$$I = \underline{92.2} \%$$

TRAVERSE POINT LOCATION FOR CIRCULAR DUCTS

new location

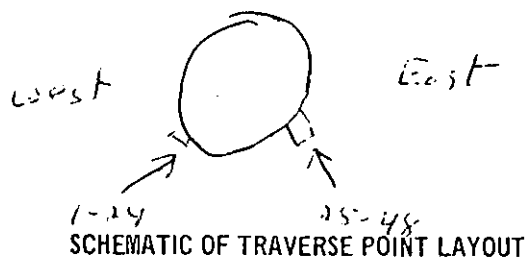
PLANT Standard Lume
 DATE 12/10/75
 SAMPLING LOCATION #6 Kiln
 INSIDE OF FAR WALL TO
 OUTSIDE OF NIPPLE, (DISTANCE A) _____
 INSIDE OF NEAR WALL TO
 OUTSIDE OF NIPPLE, (DISTANCE B) _____
 STACK I.D., (DISTANCE A - DISTANCE B) 81"
 NEAREST UPSTREAM DISTURBANCE 2400 SD
 NEAREST DOWNSTREAM DISTURBANCE 2 1/2 D
 CALCULATOR J. Joy 57D

SCHEMATIC OF SAMPLING LOCATION

TRAVERSE POINT NUMBER	FRACTION OF STACK I.D.	STACK I.D.	PRODUCT OF COLUMNS 2 AND 3 (TO NEAREST 1/8 INCH)	DISTANCE B	TRAVERSE POINT LOCATION FROM OUTSIDE OF NIPPLE (SUM OF COLUMNS 4 & 5)
1	.011	81"	78 1/8 1"	5 1/4 3	6 1/4 14
2	.032		2 1/2		7 3/4 5 1/4
3	.055		4 1/2		9 3/4 7 1/2
4	.079		6 3/8		11 5/8 9 3/8
5	.105		8 1/2		13 3/4 11 1/2
6	.132		10 3/4		16 13 3/4
7	.161		13 "		18 1/4 16
8	.194		15 3/4		21 18 3/4
9	.23		18 5/8		23 7/8 21 5/8
10	.272		22		27 1/4 25
11	.323		26 1/8		31 3/8 29 1/8
12	.398		32 1/4		37 1/2 35 1/4
13	.602		48 3/4		54 51 3/4
14	.677		54 7/8		60 1/8 57 3/8
15	.728		59		64 1/4 62
16	.77		62 3/8		67 5/8 65 3/8
17	.806		65 1/4		70 1/2 68 1/4
18	.839		68		73 1/4 71
19	.868		70 1/4		75 1/2 73 1/4
20	.895		72 1/2		77 3/4 75 1/2
21	.921		74 5/8		79 7/8 77 3/8
22	.945		76 1/2		81 3/4 79 1/2
23	.968		78 3/8		83 7/8 81 3/8
24	.989		80		85 1/4 83

PRELIMINARY VELOCITY TRAVERSE

PLANT Standard Limestone - Woodville
 DATE 12-10-75
 LOCATION Kip # 6
 STACK I.D. 6' 9"
 BAROMETRIC PRESSURE, in. Hg 29.48
 STACK GAUGE PRESSURE, in. H₂O -6.0
 OPERATORS M. Koubicek & Etter



West Pit # 6A

East

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_s), in. H ₂ O	STACK TEMPERATURE (T_s), °F
1	0.09	468
2	0.10	514
3	0.11	522
4	0.13	526
5	0.12	524
6	0.125	531
7	0.13	536
8	0.14	539
9	0.165	545
10	0.185	551
11	0.185	553
12	0.185	558
13	0.175	555
14	0.185	554
15	0.19	552
16	0.19	550
17	0.185	547
18	0.18	544
19	0.18	542
20	0.17	534
21	0.17	527
22	0.15	516
23	0.13	504
24	0.14	507
AVERAGE		

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_s), in. H ₂ O	STACK TEMPERATURE (T_s), °F
25	0.39	290
26	0.30	375
27	0.21	456
28	0.19	478
29	0.21	502
30	0.22	512
31	0.23	521
32	0.23	530
33	0.25	537
34	0.25	537
35	0.26	543
36	0.25	548
37	0.23	551
38	0.25	551
39	0.23	550
40	0.24	548
41	0.25	550
42	0.23	551
43	0.23	548
44	0.22	545
45	0.20	538
46	0.19	529
47	0.18	523
48	0.17	520
AVERAGE		

NOMOGRAPH DATA

PLANT Standard Lime - Woodville

DATE 12/10/75

SAMPLING LOCATION #6 Kiln

Pitot # 6A = 855 = C_p

CALIBRATED PRESSURE DIFFERENTIAL ACROSS ORIFICE, in. H ₂ O	ΔH_{θ}	1.98	1.98
AVERAGE METER TEMPERATURE (AMBIENT + 20 °F), °F	$T_{m\text{ avg.}}$	55 40°F	50°F
PERCENT MOISTURE IN GAS STREAM BY VOLUME	B_{wo}	85	
BAROMETRIC PRESSURE AT METER, in. Hg	P_m	29.60	29.48
STATIC PRESSURE IN STACK, in. Hg ($P_m \pm 0.073 \times$ STACK GAUGE PRESSURE in in. H ₂ O)	P_s	29.04 29.57	
RATIO OF STATIC PRESSURE TO METER PRESSURE	P_s/P_m	.985	
AVERAGE STACK TEMPERATURE, °F	$T_{s\text{ avg.}}$	540	540
AVERAGE VELOCITY HEAD, in. H ₂ O	$\Delta p_{\text{ avg.}}$	.16	.16
MAXIMUM VELOCITY HEAD, in. H ₂ O	$\Delta p_{\text{ max.}}$		0.16
C FACTOR			
CALCULATED NOZZLE DIAMETER, in.			
ACTUAL NOZZLE DIAMETER, in.		.245	
REFERENCE Δp , in. H ₂ O	$\Delta P = 1$	$\Delta H = 1.96$	

FIELD DATA

PLANT Standard Oil - Live Oak Co
 DATE 12-10-75
 SAMPLING LOCATION Kiln #1
 SAMPLE TYPE EAH-S
 RUN NUMBER 1-6
 OPERATOR McKee & Carter
 AMBIENT TEMPERATURE 72
 BAROMETRIC PRESSURE 29.48
 STATIC PRESSURE, (P_s) -6.0"
 FILTER NUMBER (s) 6.3-4

PROBE LENGTH AND TYPE 7' glass
 NOZZLE I.D. 2.245
 ASSUMED MOISTURE, % 8
 SAMPLE BOX NUMBER 4
 METER BOX NUMBER 3
 METER ΔH₆ 1.983
 C FACTOR _____
 PROBE HEATER SETTING h
 HEATER BOX SETTING h
 REFERENCE ΔP for sp of 1 2 Hg 1.96

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 1.5 MINUTES LEAK RATE @ 0.25 CFM @ 15" Hg

TRAVERSE POINT NUMBER	SAMPLING TIME, min	CLOCK TIME (24 hr CLOCK)	GAS METER READING (V _m), ft ³	VELOCITY HEAD (ΔP _s), in. H ₂ O	ORIFICE PRESSURE DIFFERENTIAL (ΔH), in. H ₂ O		STACK TEMPERATURE (T _s), °F	DRY GAS METER TEMPERATURE		PUMP VACUUM, in. Hg	SAMPLE BOX TEMPERATURE, °F	IMPIRGER TEMPERATURE, °F
					DESIRED	ACTUAL		INLET (T _{m in}), °F	OUTLET (T _{m out}), °F			
1	0	1010	076.804	.11	.23	.23	429	48	46	1.0	176	37
2	1.5	1015	77.341	.10	.21	.21	430	47	46	1.0	177	37
3	3	1013	77.810	.14	.29	.29	429	47	46	1.0	177	36
4	4.5	1014.5	78.245	.12	.26	.25	459	47	46	1.0	184	35
5	6	1016	78.561	.15	.315	.315	483	47	46	1.0	184	39
6	7.5	1017.5	79.341	.17	.36	.36	492	47	46	1.0	184	39
7	9	1019	79.862	.14	.295	.295	491	47	45	1.0	184	39
8	10.5	1020.9	80.541	.20	.42	.42	512	46	46	1.0	184	36
9	12	1022	80.543	.20	.42	.42	524	47	46	1.0	182	36
10	13.5	1023.5	81.146	.22	.45	.45	527	46	46	1.0	192	36
11	15	1025	81.620	.23	.47	.47	543	47	46	1.0	194	36
12	16.5	1026.5	82.147	.21	.43	.43	546	46	45	1.0	195	36
13	18	1028	82.677	.39	.78	.78	557	45	45	1.0	195	36
14	19.5	1029.5	82.344	.49	.98	.98	560	45	45	1.0	195	36
15	21	1031	84.148	.51	1.02	1.02	566	47	46	1.0	198	36
16	22.5	1032.5	84.710	.46	.93	.93	564	46	45	1.0	206	38

Probe length 7' glass
 Nozzle I.D. 2.245
 Assumed moisture 8%
 Sample box number 4
 Meter box number 3
 Meter ΔH₆ 1.983
 C factor _____
 Probe heater setting h
 Heater box setting h
 Reference ΔP for sp of 1 2 Hg 1.96

BUN NO 1-6

BUN NO 1-6

BUN NO 1-6

[illegible]

RUN NO. 1-6

IMPINGER
TEMPERATURE,
OF

[illegible]

CLEAN-UP DATA

Plant STANDARD LIME Comments:
Date 12-10-75
Sampling Location KILN # 6
Sample Type EPA 5
Run Number 1-6
Sample Box Number 4
Clean-up Man ETTER, HUGHES, & MCKENDREE

IMPINGERS

	#1	#2	#3	#4	#5
Final Vol.	116 ml	103 ml	1 ml	ml	ml
Initial Vol.	100 ml	100 ml	0 ml	ml	ml
Net Vol.	16 ml	3 ml	1 ml	ml	ml

Total Net Volume in Impingers 20 ml

SILICA GEL

Final Weight	207.1 g	g	g
Initial Weight	200.0 g	g	g
Net Weight	7.1 g	g	g

Total Net Weight in Silica Gel 7.1 g

Total Moisture 27.1 g

Filter Number(s) 63-4 _____

DRY MOLECULAR WEIGHT DETERMINATION

PLANT Standard Lime
 DATE 12/10/75
 SAMPLING TIME (24-hr CLOCK) 1300 hr. - 1215 hr
 SAMPLING LOCATION #6 Kiln
 SAMPLE TYPE (BAG, INTEGRATED, CONTINUOUS) Bag
 ANALYTICAL METHOD Oxstat
 AMBIENT TEMPERATURE 70°
 OPERATOR H. Joy

COMMENTS:

RUN GAS	1		2		3		AVERAGE NET VOLUME	MULTIPLIER	MOLECULAR WEIGHT OF STACK GAS (DRY BASIS) M_d , lb/lb-mole
	ACTUAL READING	NET	ACTUAL READING	NET	ACTUAL READING	NET			
CO ₂	16.1	16.1	16.2	16.2			16.2	44/100	7.13
O ₂ (NET IS ACTUAL O ₂ READING MINUS ACTUAL CO ₂ READING)	26.2	10.1	26.3	10.1			10.1	32/100	3.23
CO (NET IS ACTUAL CO READING MINUS ACTUAL O ₂ READING)	26.2	0.0	26.3	0.0			0.0	28/100	0.0
N ₂ (NET IS 100 MINUS ACTUAL CO READING)		73.8		73.7			73.7	28/100	20.64
TOTAL									31.00

STANDARD LIME

ISOKINETIC CALCULATION

Run# 1-6, Date 12-10-75
 Location KILN #6, Initials R.H.E.

MOISTURE

Net volume of liquid collected

in impingers and silica gel

$$= V_L = \underline{27.1} \text{ ml}$$

Net volume of gas through dry

gas meter at meter conditions

$$= V_m = \underline{33.244} \text{ cu.ft.}$$

Barometric Pressure - absolute

$$= BP = \underline{29.48} \text{ in.Hg}$$

Average absolute meter tem-

perature ($^{\circ}\text{F} + 460$)

$$= T_m = \underline{505} \text{ }^{\circ}\text{R}$$

Percent moisture = M

$$M = \frac{100}{1 + \frac{373.63 (V_m) (BP)}{(T_m) (V_L)}} + 2.5$$

$$= \frac{100}{1 + \frac{373.63 (33.244) (29.48)}{(505) (27)}} + 2.5 = \underline{6.09} \%$$

MOLECULAR WEIGHT

Percent O_2 by volume dry basis

$$= \text{O}_2 = \underline{10.1} \%$$

" CO " " " "

$$= \text{CO} = \underline{0.0} \%$$

" CO_2 " " " "

$$= \text{CO}_2 = \underline{16.2} \%$$

$$\text{N}_2 = 100 - (\text{O}_2 + \text{CO}_2 + \text{CO}) = 100 - ($$

$$+ +)$$

Percent N_2 by volume dry basis

$$= \text{N}_2 = \underline{73.7} \%$$

Dry molecular weight = MD

$$\text{MD} = 0.44 (\text{CO}_2) + 0.32 (\text{O}_2) + 0.28 (\text{N}_2 + \text{CO})$$

$$\text{MD} = 0.44 (16.2) + 0.32 (10.1) + 0.28 (73.7 + 0.0)$$

$$\text{MD} = \underline{31.00}$$

ISOKINETIC CALCULATION (Cont.d)

Run # 1-6

MOLECULAR WEIGHT (Cont.d)

Stack gas molecular weight = MS

$$MS = (MD) [1 - (M \times 10^{-2})] + 18 (M \times 10^{-2})$$

$$MS = (31.00) [1 - (6.09 \times 10^{-2})] + 18 (6.09 \times 10^{-2}) = \underline{30.21}$$

VELOCITY

Static pressure of stack

$$= p = \underline{-6.0''} \text{ H}_2\text{O}$$

Pitot tube coefficient, type S

$$= CP = \underline{.855}$$

Average Absolute Stack Temp. ($^{\circ}\text{F} + 460$)

$$= T_s = \underline{986} \text{ } ^{\circ}\text{R}$$

Average of the square root of Δp

$$= \sqrt{\Delta p} = \underline{.652} (\text{in. H}_2\text{O})^{1/2}$$

Absolute pressure of stack ($BP \pm \frac{p}{13.6}$)

$$= P_s = \underline{29.48} \text{ in. Hg}$$

Stack gas velocity = V_s

$$\underline{29.04}$$

$$V_s = (85.48) (CP) \left[\sqrt{\frac{(T_s)}{(P_s) (MS)}} \right] \left[\sqrt{\Delta p} \right]$$

$$V_s = (85.48) (.855) \left[\sqrt{\frac{(986)}{(29.48) (30.21)}} \right] \left[.652 \right]$$

$$\underline{29.04}$$

$$= \underline{50.52} \text{ FPS}$$

ISOKINETIC PERCENT

Average pressure drop across orifice

$$= \Delta H = \underline{1.07} \text{ in. H}_2\text{O}$$

Total sampling time

$$= T = \underline{72.0} \text{ min.}$$

Diameter of nozzle

$$= d = \underline{.245} \text{ in.}$$

Area of nozzle = A

$$A = 54.54 \times 10^{-4} (d^2) = 54.54 \times 10^{-4} (.245)^2 = \underline{3.274 \times 10^{-4}} \text{ sq. ft.}$$

I = percent Isokinetic

$$I = (1.667) (T_s) \left[0.00267 (V_L) + \frac{(V_m)}{(T_m)} \left(PB + \frac{\Delta H}{13.6} \right) \right]$$

$$(T) (V_s) (P_s) (A)$$

$$I = (1.667) (986) \left[0.00267 (27) + \frac{(33.244)}{(505)} \left(29.48 + \frac{1.07}{13.6} \right) \right]$$

$$(72.0) (\underline{50.52}) (\underline{29.04}) (3.274 \times 10^{-4})$$

$$I = \underline{95.91} \%$$

FIELD DATA

PLANT Standard Line - Woodville
 DATE 12/10/75
 SAMPLING LOCATION #6 R/W
 SAMPLE TYPE EPA-5
 RUN NUMBER 2-6
 OPERATOR Tom Hughes, Effer
 AMBIENT TEMPERATURE 35.0°F
 BAROMETRIC PRESSURE 29.48
 STATIC PRESSURE, (P_s) -6.0" H₂O
 FILTER NUMBER (s) 53-9

PROBE LENGTH AND TYPE Pitot #6 A (65 ft. MR)
 NOZZLE I.D. .245
 ASSUMED MOISTURE, % 8
 SAMPLE BOX NUMBER _____
 METER BOX NUMBER _____
 METER ΔH_g 1.98
 C FACTOR _____
 PROBE HEATER SETTING HF
 HEATER BOX SETTING HI
 REFERENCE ΔP ΔP=1, ΔH=1.96

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 1.5 MINUTES LEAK RATE 0 CFM @ 25 Hg

TRAVERSE POINT NUMBER	SAMPLING TIME, min	CLOCK TIME (24 hr CLOCK)	GAS METER READING (V _m), ft ³	VELOCITY HEAD (ΔP _s), in. H ₂ O	ORIFICE PRESSURE DIFFERENTIAL (ΔH), in. H ₂ O		STACK TEMPERATURE (T _s), °F	DRY GAS METER TEMPERATURE		PUMP VACUUM, in. Hg	SAMPLE BOX TEMPERATURE, °F	IRPINGER TEMPERATURE, °F
					DESIRED	ACTUAL		INLET (T _{m in}), °F	OUTLET (T _{m out}), °F			
1	0	1335	110.627	.23024	.49	.49	416	43	39	1	153	
2	1.5		111.195	.16	.33	.33	416	43	39	1	16	38
3	3		111.678	.20	.41	.41	495	39	42	1	173	38
4	4.5		112.215	.20	.41	.41	527	39	40	1	171	40
5	6		112.634	.19	.39	.39	532	39	39	1	176	41
6	7.5		113.115	.24	.49	.49	552	40	41	1		39
7	9		113.664	.23	.47	.47	553	41	40	1		40
8	10.5		114.161	.28	.57	.57	556	40	39	1		39
9	12		114.805	.27	.49	.49	557	40	40			38
10	13.5		115.275	.27	.55	.55	568	40	40			39
11	15		115.836	.28	.57	.57	569	40	39	1.1	2-11	39
12	16.5		116.437	.27	.55	.55	563	42	39			40
13	18		116.984	.29	.59	.59	567	41	39			39
14	19.5		117.563	.28	.57	.57	555	41	39	1.1		29
15	21		118.161	.23	.47	.47	547	41	40			42
16	22.5		118.691	.27	.55	.55	553	41	39			42

77.12

12

219

2-6
RUN NO.

RUN NO.

[illegible]

CLEAN-UP DATA

Plant Standard Lins Comments:
Date 12/10/75
Sampling Location #6 Rele
Sample Type EPA-5
Run Number 2-6
Sample Box Number 56B-5
Clean-up Man F. Toy Elder Hughes

IMPINGERS

	#1	#2	#3	#4	#5
Final Vol.	111 ml	101 ml	ml	ml	ml
Initial Vol.	100 ml	100 ml	ml	ml	ml
Net Vol.	11 ml	1 ml	ml	ml	ml

Total Net Volume in Impingers 12 ml

SILICA GEL

Final Weight	207.2 g	g	g
Initial Weight	200 g	g	g
Net Weight	7.2 g	g	g

Total Net Weight in Silica Gel 7.2 g

Total Moisture 19.2 g

Filter Number(s) 53.9

292.7
85.5 g
7.2

Date 12/9/15

Project EVA

Analyst Thalman

[illegible]

Brink BMS-11 Data

Run 1-6

12-9-75

5 Minute Sampling Time
2 mm Probe Tip

Started Run 1035 hrs

Finished Run 1040 hrs

<u>Time</u>	<u>ΔP in. Hg</u>	<u>Inlet Temp. °F</u>	<u>Outlet Temp. °F</u>	<u>Static Pressure in. H₂O</u>
1035	0.55	182	230	4.2
1036	0.55	181	210	3.4
1037	0.55	183	206	4.4
1038	0.55	182	201	4.4
1039	0.55	182	223	3.6
1040	Stopped			

Brink BMS-11 Data

Run 1-4

12-9-75

5 Minute Sampling Time

1.5 mm Probe Tip

Started Run 1500 hrs

Finished Run 1505 hrs

<u>Time</u>	<u>ΔP in. Hg</u>	<u>Inlet Temp °F</u>	<u>Outlet Temp °F</u>	<u>Static Pressure in. H₂O</u>
1500	1.8	83	173	4.6
1501	1.8	82	174	4.0
1502	1.8	81	174	4.6
1503	1.8	81	174	4.2
1504	1.8	83	176	4.4
1505	stopped			

APPENDIX E

ANALYTICAL DATA SHEETS

ANALYTICAL DATA

PLANT STANDARD Lime Woodville

COMMENTS:

DATE 12-9-75SAMPLING LOCATION KILN #4SAMPLE TYPE EPA Method 5RUN NUMBER 1-4SAMPLE BOX NUMBER MRPCLEAN-UP MEN McDONALD McKendree & ETTERANALYST W. Mc DONALDFRONT HALF

LABORATORY RESULTS

ACETONE WASH OF NOZZLE, PROBE, CYCLONE (BYPASS),
FLASK, FRONT HALF OF FILTER HOLDERCONTAINER 904-3
53-615255.3 mg
1113.5FILTER NUMBER 53-6 F1/2
53-7 2nd 1/2CONTAINER 53-71382.3 mgFRONT HALF SUBTOTAL 17751.1 mgBACK HALFIMPINGER CONTENTS AND WATER WASH OF
IMPINGERS, CONNECTORS, AND BACK
HALF OF FILTER HOLDERCONTAINER 904-5
ETHER-CHLOROFORM 904-6
EXTRACTION4.9 mg
0 mgACETONE WASH OF IMPINGERS, CONNECTORS,
AND BACK HALF OF FILTER HOLDERCONTAINER 904-44.1 mgBACK HALF SUBTOTAL 9.0 mgTOTAL WEIGHT 17760.1 mgMOISTURE

IMPINGERS

FINAL VOLUME _____ ml

INITIAL VOLUME _____ ml

NET VOLUME _____ ml

SILICA GEL

FINAL WEIGHT _____ g _____ g _____ g

INITIAL WEIGHT _____ g _____ g _____ g

NET WEIGHT _____ g _____ g _____ g

TOTAL MOISTURE _____ g

EPA (Dur) 231

4/72

ANALYTICAL DATA

PLANT STANDARD Lime
DATE 12-9-75
SAMPLING LOCATION #4 KILN
SAMPLE TYPE EPA 5
RUN NUMBER 2-4
SAMPLE BOX NUMBER 6
CLEAN-UP MEN ETTER & HUGHES
ANALYST W. McDonald

COMMENTS:

FRONT HALF

LABORATORY RESULTS

ACETONE WASH OF NOZZLE, PROBE, CYCLONE (BYPASS),
FLASK, FRONT HALF OF FILTER HOLDER

CONTAINER 904-7 4683.0 mgFILTER NUMBER 53-8 _____CONTAINER 53-8 940.9 mgFRONT HALF SUBTOTAL 5623.9 mgBACK HALF

IMPINGER CONTENTS AND WATER WASH OF
IMPINGERS, CONNECTORS, AND BACK
HALF OF FILTER HOLDER

CONTAINER 904-9 0.8 mgETHER-CHLOROFORM
EXTRACTION 904-10 1.3 mg

ACETONE WASH OF IMPINGERS, CONNECTORS,
AND BACK HALF OF FILTER HOLDER

CONTAINER 904-8 8.5 mgBACK HALF SUBTOTAL 10.6 mg

TOTAL WEIGHT	<u>5634.5</u> mg
--------------	------------------

MOISTURE

IMPINGERS

FINAL VOLUME _____ ml

INITIAL VOLUME _____ ml

NET VOLUME _____ ml

SILICA GEL

FINAL WEIGHT _____ g _____ g _____ g

INITIAL WEIGHT _____ g _____ g _____ g

NET WEIGHT _____ g _____ g _____ g

TOTAL MOISTURE _____ g

EPA (Dur) 231

4/72

ANALYTICAL DATA

PLANT STANDARD Lime
DATE 12-10-75
SAMPLING LOCATION KILN #6
SAMPLE TYPE EPA 5
RUN NUMBER 1-6
SAMPLE BOX NUMBER MRC 4
CLEAN-UP MEN ETTER, HUGHES, McKendree
ANALYST W. McDONALD

COMMENTS:

FRONT HALF

LABORATORY RESULTS

ACETONE WASH OF NOZZLE, PROBE, CYCLONE (BYPASS),
FLASK, FRONT HALF OF FILTER HOLDER

CONTAINER 904-11 11305.1 mg

FILTER NUMBER 63-4

CONTAINER 63-4 658.4 mg

FRONT HALF SUBTOTAL 11963.5 mg

BACK HALF

IMPINGER CONTENTS AND WATER WASH OF
IMPINGERS, CONNECTORS, AND BACK
HALF OF FILTER HOLDER

CONTAINER 904-13 23.1 mg

ETHER-CHLOROFORM
EXTRACTION 904-14 5.7 mg

ACETONE WASH OF IMPINGERS, CONNECTORS,
AND BACK HALF OF FILTER HOLDER

CONTAINER 904-12 3.0 mg

BACK HALF SUBTOTAL 31.8 mg

TOTAL WEIGHT 11995.3 mg

MOISTURE

IMPINGERS

FINAL VOLUME _____ ml

INITIAL VOLUME _____ ml

NET VOLUME _____ ml

SILICA GEL

FINAL WEIGHT _____ g _____ g _____ g

INITIAL WEIGHT _____ g _____ g _____ g

NET WEIGHT _____ g _____ g _____ g

TOTAL MOISTURE _____ g

EPA (Dur) 231

4/72

ANALYTICAL DATA

PLANT STANDARD Lime
DATE 12-10-75
SAMPLING LOCATION #16 KILN
SAMPLE TYPE EPA 5
RUN NUMBER 2-6
SAMPLE BOX NUMBER 5GB-5
CLEAN-UP MAN H. TOY & ENTER & HUGHES
W. MC DONALD

COMMENTS:

FRONT HALF

LABORATORY RESULTS

ACETONE WASH OF NOZZLE, PROBE, CYCLONE (BYPASS),
FLASK, FRONT HALF OF FILTER HOLDER

CONTAINER 904-1510420.0 mgFILTER NUMBER 53-9CONTAINER 53-9857.8 mgFRONT HALF SUBTOTAL 11277.8 mgBACK HALF

IMPINGER CONTENTS AND WATER WASH OF
IMPINGERS, CONNECTORS, AND BACK
HALF OF FILTER HOLDER

CONTAINER 904-1757.4 mgETHER-CHLOROFORM
EXTRACTION 904-186.7 mg

ACETONE WASH OF IMPINGERS, CONNECTORS,
AND BACK HALF OF FILTER HOLDER

CONTAINER 904-162.4 mg

BACK HALF SUBTOTAL

66.5 mg

TOTAL WEIGHT

11344.3 mgMOISTURE

IMPINGERS

FINAL VOLUME _____ ml

INITIAL VOLUME _____ ml

NET VOLUME _____ ml

SILICA GEL

FINAL WEIGHT _____ g _____ g _____ g

INITIAL WEIGHT _____ g _____ g _____ g

NET WEIGHT _____ g _____ g _____ g

TOTAL MOISTURE _____ g

Standard Lime Brinks Liquid

ANALYSIS REPORT	PROJECT NUMBER 6912-12
TO	FROM
DATE	CC

4 Kiln

Probe	mg Samples	48.9
Cyclone		11.7
Stage 1		4.8
" 2		2.0
" 3		1.7
" 4		2.1
" 5		2.4
		TOTAL mg. 73.6

6 Kiln

Probe	80.0
Cyclone	8.4
Stage 1	2.5
" 2	1.4
" 3	1.3
" 4	1.1
" 5	3.6
TOTAL mg 98.3	

Alum Filters Brinks

ANALYSIS REPORT	PROJECT NUMBER
TO	FROM
DATE	CC

#6 Kiln

65-1	4.86 mg	
65-2	7.82 mg	
65-3	0.156 mg	
65-4	0.129 mg	
65-5	0.045 mg	
		TOTAL 13.01 mg

#4 Kiln

65-6	= 16.3 mg	
65-7	7.1 mg	
65-8	1.2 mg	
65-9	0.7 mg	
65-10	0.2 mg	
		TOTAL 25.5 mg

#4	thimble filter	# 102	wgt = 0.7 mg
#6	"	# 108	wgt = 0.8 mg

APPENDIX F

SAMPLING AND ANALYTICAL PROCEDURES

Method 5 MRC Clean Up Procedure

1. Transportation to Clean Up Area

After the run is finished the train can be removed to the clean up area. This can be done with the probe in place or it can be removed from the train and sealed at both ends. If the probe is removed, the cyclone entrance must be sealed to avoid any loss of contamination of its contents. Caution must be exercised when sealing the system for transportation so that cooling of the front half of the train will not cause water to be pulled into it from the back half.

2. Probe Clean Up

Remove any material from the outside of the probe and nozzle that may contaminate the sample by falling or being washed into the sample bottle. Remove the nozzle and rinse the inside with acetone into the front 1/2 acetone sample bottle. Brush the inside with a clean test tube brush while flushing with acetone into the sample bottle. Wash the brush with acetone into the bottle. Rinse the inside of the nozzle a final time with acetone.

Elevate one end (the nozzle end) of the probe and rinse through with acetone into the front 1/2 sample bottle while slowly rotating through 720°. Run a clean probe brush through the probe while flushing with acetone and slowly rotating the brush. When the brush is exposed on the down stream end of the probe, rinse it into the sample bottle with acetone. Do not pull the brush back through the probe. Repeat the brushing step. Rinse the probe a final time by flushing with acetone while rotating it 360°. Wash the funnel into the sample bottle. Observe the probe inside and repeat the clean up procedure if sample particles are obvious.

3. Cyclone and Filter Clean Up

Remove the cyclone and connecting glassware. Rinse the cyclone with acetone into the collection flask and pour into the front 1/2 sample bottle. Repeat until all loose material is collected. If sample material adheres to the inside surfaces, a small test tube brush can be used to free it, and it should be rinsed into the sample bottle. If a brush or spatula is used to clean the cyclone and catch bottle, it must be cleaned into the sample bottle. All connecting glassware between the cyclone system and filter must be treated in the same manner i.e., flushing and or brushing the sample exposed surfaces into the sample container with acetone until no sample material is visible.

The filter assemble must be disassembled in a dust free area. The filter membrane should be loosened from the silicon rubber gasket by running a knife blade or spatula between them being careful not to scrape or loosen particles of glass from the glass frit. Lift the filter, using a spatula, and place into its storage container. Carefully scrape all particles of filter material from both mating surfaces (i.e.: the silicon rubber gasket and the front-half glass holder) and place them in the storage container. Rinse and brush the front-half glass holder's sample exposed surfaces into the sample bottle with acetone. Do not wash or rinse the frit. Wash brushes, spatulas and funnels into the sample and store.

4. Back 1/2 Clean Up

Rinse the back half filter holder, connecting glass to the first impinger, and impinger "U" bends with distilled water into the back half water wash sample bottle by filling or flushing twice. Remove each of the first three impingers and pour the content into a graduated cylinder for measurement. Transfer these contents into the sample bottle. Wash each of the three impingers two times by squirting about 100 ml of distilled water into each, covering the inlet and outlet openings with clean caps or clean fingers, and shaking to cause the wash water to contact all of the inside surfaces. Pour this wash water into the sample bottle and rinse the graduated cylinder and funnel into sample bottle with distilled water. Seal the bottle and store.

Rinse the back half filter holder, the filter to impinger connectors, and the impinger "U" bends with acetone into the back half acetone wash sample bottle by filling or flushing two times. Wash each of the three impingers two times by adding about 100 ml of acetone and shaking as with the water wash. Transfer these washings to the sample bottle. Rinse the graduated cylinder used to measure the water and the funnel with acetone into the sample bottle. Cap the bottle and store.

The sampling and analysis procedures used on Methods 3 and 7 were from the Federal Register, Vol. 36, No. 247, December 23, 1971 were followed during sampling at Standard Lime Company.

APPENDIX G

BRINK® BMS-11 SAMPLER MANUAL

Cascade Impactor for Adiabatic Measurements

J. A. BRINK, Jr.¹ Monsanto Chemical Co., Everett, Mass.

Data on development, design, and use of a practical instrument in the field make it possible to build a convenient tested device for special purposes. Particle-size measurements make possible determination of acceptable stack discharges of aerosols, evaluation of installed collection equipment, rational selection and design of equipment, and recognition of potential problems early in the development of new processes.

IMPINGEMENT DEVICES have been used for years for sampling gas-borne particles (4-6). Owens (9) and Ferry, Farr, and Hartmann (7) used a single jet and a dry glass slide to collect samples of gas particulates suspended in a gaseous medium, and measured the sizes of impacted particles with a microscope by laborious particle counts. In May's cascade system of impactor stages (8) particles were separated into five fractions ranging in particle size from 1 to 50 microns. Laskin (7) used May's impactor on heavy aerosol particles of micron and submicron sizes. A modified cascade impactor extended measurements into the submicron range (13).

Extensive theoretical and experimental work with cascade impactors has been done by Ranz and Wong (11, 12), Gillespie and Johnstone (2, 3), and others at the University of Illinois. Pilcher, Mitchell, and Thomas (10) and Wilcox (14, 15) investigated the characteristics of impactors and made significant improvements in design.

In many industrial plants aerosols are found in high concentrations in gases saturated with vapor at 30° to 200° C. For measuring particle-size distributions of aerosols in such gases, special equipment is required. An "in-line" cascade impactor and accessory equipment for adiabatic measurements on industrial processes were developed at Monsanto in 1954. They are small, light, and compact, and can be carried in a suitcase for tests at plants throughout the country.

Impactor and Auxiliary Apparatus

The cascade impactor has five in-line stages, each of which has a jet that utilizes a collection cup as an impaction plate. A spring holds each collection cup in place. The particles suspended in the gases pass through a jet, particles with sufficient inertia impact against a cup, and the remainder pass through annular slots located around the cup. Each collection cup has annular slots with a total cross-sectional area 30 times the area of the largest jet; turbulent effects at the slots were negligible with this design. The dimensions of the impactor jets (Table I) were selected as described by Ranz and Wong (11). The impactor is 15½ inches long and was machined from 316 stainless steel. Particles in the 0.3- to 3.0-micron range are collected.

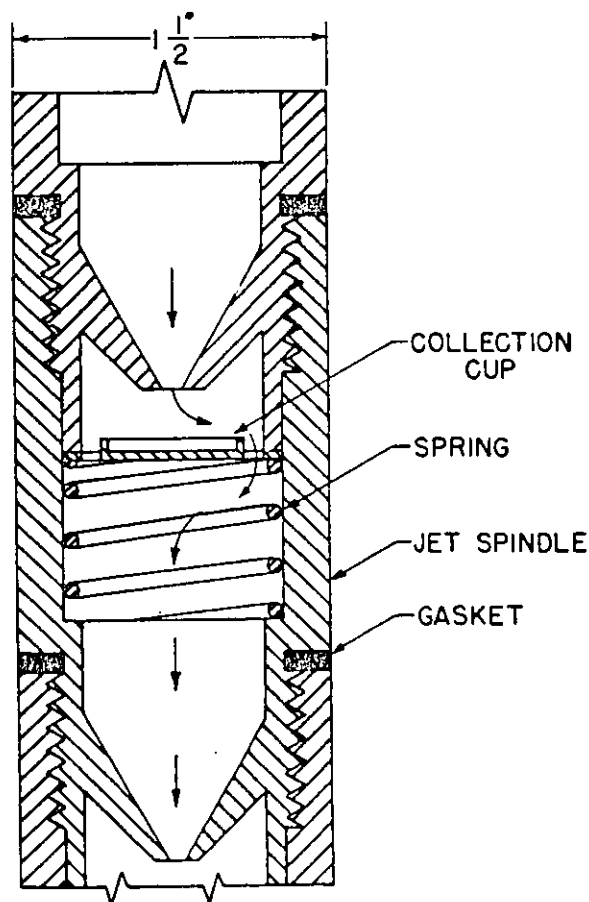
A glass cyclone, with the same dimensions as that described by Gillespie (2, 3),

is used upstream from the impactor. Two filters, consisting of Corning No. 9480 filter tubes packed tightly with No. 800 Pyrex glass wool, collect particles less than 0.3 micron in diameter. The cyclone, impactor, and filters are mounted in a box with a removable side. A No. 2½ L-R Manufacturing Co. blower, inside the box, is driven by a 8000 r.p.m. Fairchild Industries (Model

Table I. Dimensions of Cascade Impactor Jets

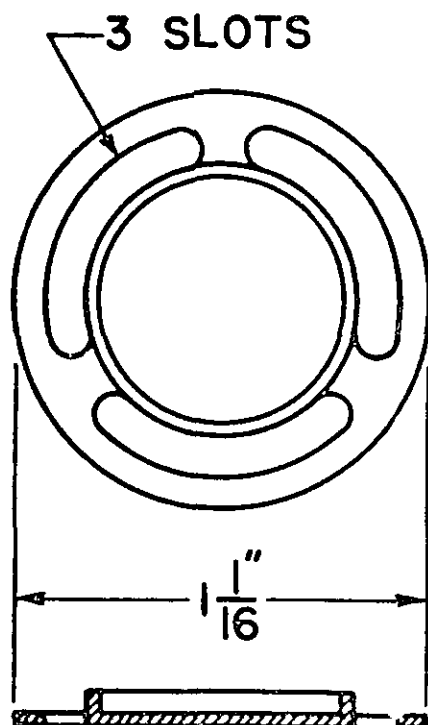
Jet No.	Dimensions, Cm.	
	Jet diam.	Spacing of jet opening*
1	0.249	0.747
2	0.1775	0.533
3	0.1396	0.419
4	0.0946	0.282
5	0.0731	0.220

* From collection cup surface.



The in-line impactor has five stages. Particles in the range of 0.3 to 3.0 microns are collected by successive impingement

¹ Present address, Research Department, Monsanto Chemical Co., St. Louis, Mo.



Collection cups are positioned so that the distance from the jet decreases as the jet diameter becomes smaller. Annular slots around cup minimize turbulence

1401) motor on the back side of the box. A heater, consisting of three 12-inch sections of Nichrome resistance wire

(Catalog No. 416, Size 1, coiled type, Eagle Electric Mfg. Co.) is mounted in the box. The three sections of the heater are connected in series and operated on 110 volts. The air temperature is controlled with a 100° to 400° F. thermostat (Fenwall, Inc., Catalog No. 1731-0). The box was designed so that air could be circulated at a high velocity past the cyclone, impactor, and filters. The blower discharges air through a duct below the heater and then over the heater, impactor, and cyclone back to the blower inlet. Thermometers, mounted on the removable side of the box, indicate the temperature of the air entering the blower and leaving the heater. Two manometers, mounted on the removable side of the box and connected to the impactor, measure the static pressure at the inlet and the pressure drop across the impactor.

A sample line, heated with flexible tape heaters (electromagnetic heating tape, Howe and French), is controlled with a Variac. Sample probes and lines are sized for isokinetic sampling.

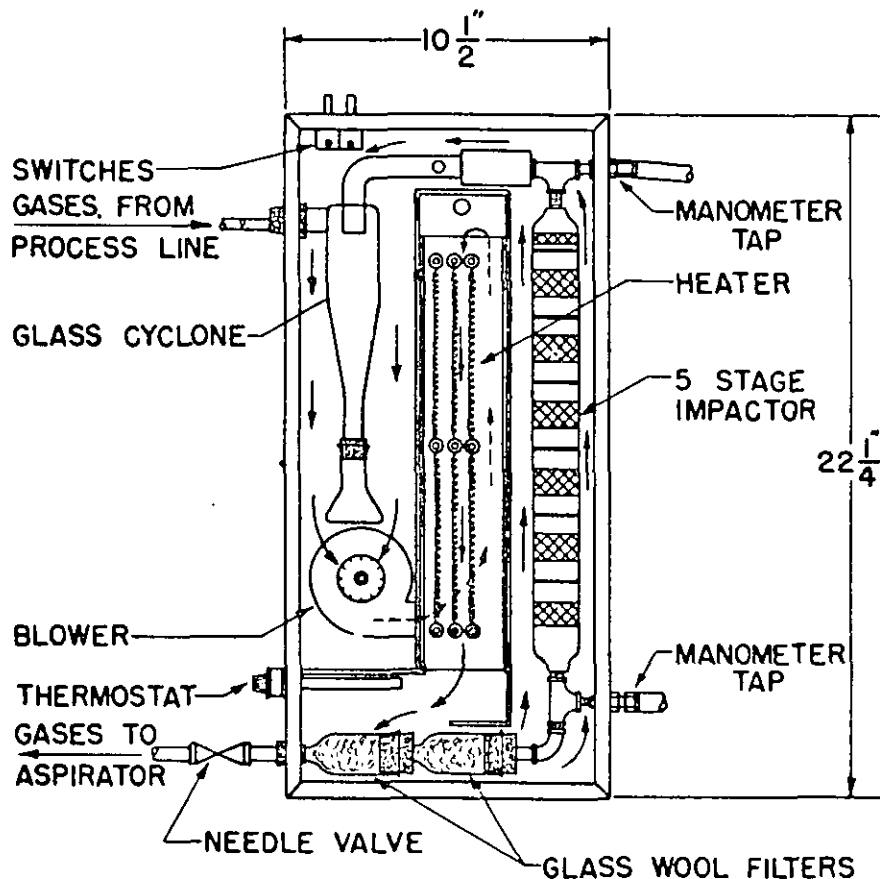
Experimental Procedure

Prior to each test, the impactor and auxiliary apparatus were tested for leaks under 8-inch mercury vacuum. Then the sample line and the impactor box were heated to the temperature of the process stream. The sample probe was inserted in the process stream and gas flow through the impactor was started

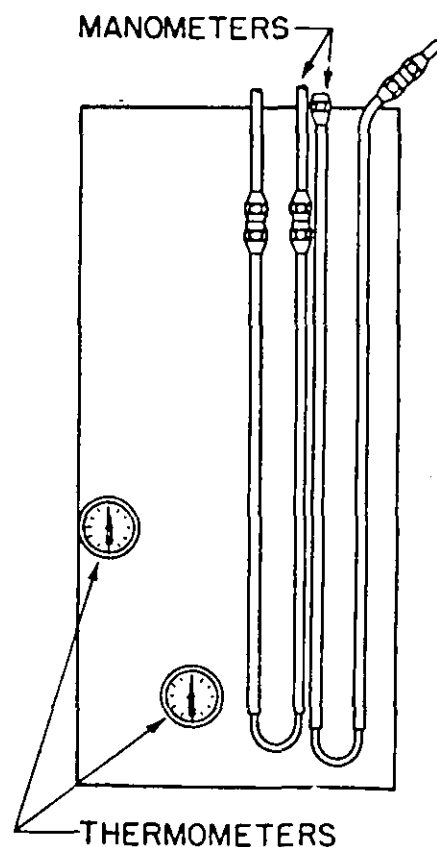
and held constant until the end of the test. A vacuum of 4 to 8 inches of mercury was usually required, giving flow rates of 2700 to 3700 cc. per minute.

Usually, 40 to 80% of the aerosol was collected on one cup or stage of the impactor. The sample time was limited by re-entrainment from this cup; a 25- to 50-mg. sample could be collected on one cup without re-entrainment. The maximum sample time was used, so that reasonable quantities were collected on other cups. With one trial run, the flow rate for subsequent runs was set at a level that would give maximum collection on the third or middle stage of the impactor, and the sample time was set at the maximum time feasible without re-entrainment.

After each test the sample cups were removed from the impactor with tweezers. The quantity collected on each cup was determined by weighing and/or chemical analyses. The glass cyclone was washed with water and the solution analyzed. The glass wool filters were leached with water and the resulting solution was analyzed; or the glass wool was removed, the filter holder was rinsed, and the glass wool and solution were analyzed either volumetrically or gravimetrically, depending on the type of aerosol. Both weighing and chemical analyses were used when condensation or evaporation was a potential source of



Compact make-up of impactor and its auxiliary equipment make it suitable for tests throughout the plant



Temperature and pressure during sampling period are controlled by observing and adjusting instruments. Interpretation of data aids in identifying aerosols collected

error; double analyses showed that errors due to condensation or evaporation were negligible and sampling was adiabatic.

Calibration of Impactor

The in-line impactor was designed with the internal dimensions used by Ranz and Wong (17), except that the gases leaving any stage passed through annular slots around the collection cups rather than out the side. As the velocities through the slots were $1/30$ to $1/350$ the jet velocities, the in-line impactor was expected to perform in the same way as impactors used by Ranz and Wong.

Sulfuric acid mists were generated in the laboratory and passed through different retention chambers in the manner described by Gillespie (2, 3). Particle-size determinations were made on these mists with the in-line impactor (Figure 1). These data are in good agreement with data reported by Gillespie, who used impactors calibrated by Ranz and Wong (17). This work showed that the in-line impactor should have the same calibration as determined by them.

With a gas meter, the in-line impactor was calibrated as a flowmeter. The relationship between gas flow through the impactor and pressure drop across the impactor, for air at 25° C. and 14.7 p.s.i.a., was determined to be:

$$V_0 = 24.5 (\Delta P)^{0.48} \quad (1)$$

The gas flow, V_0 , for process gases at different temperatures and pressures, was calculated from this equation, using suitable corrections for differences in gas density.

Particle-Size Distribution Calculations

Calculation of particle-size distribution was based on the generalized calibration curve determined by Ranz and Wong (17). The characteristic diameter, D_{pc} , for each stage of the impactor may be calculated from

$$D_{pc} = \left(\frac{18 \mu D_c}{C \rho_p v_1} \right)^{1/3} (\psi_{30})^{1/2} (10)^4 \quad (2)$$

where $(\psi_{30})^{1/2} = 0.38$ for round jets (17). However, as the C factor is a function of D_{pc} , calculation of D_{pc} is not as straightforward as indicated by Equation 2. C is defined in general as:

$$C = 1 + \left(\frac{2L}{D_p} \right) [1.23 + 0.41 e^{-0.44 D_p (10)^{-4}/L}] (10)^4 \quad (3)$$

for $0.1 < [2L(10)^4/D_p] < 134$

When

$$\left(\frac{D_p(10)^{-4}}{L} \right) > 2.7,$$

the exponential term in Equation 3 may be neglected and

$$C = 1 + (2.46)^2 (10)^4 (L/D_p) \quad (4)$$

This condition existed in most applications of the impactor. For air at normal

room temperature and pressures, Equation 4 becomes

$$C = 1 + 0.165/D_p \quad (5)$$

When gases are expanded through a jet, the temperature change can be estimated on the basis of frictionless adiabatic flow for a perfect gas (3). However, in most cases this temperature change is negligible and the velocity of the gases at the jet throat, v_2 , can be estimated as follows:

$$v_2 = \frac{4 V_0 (P_0)}{\pi (D_c)^2 (P_2)} \quad (6)$$

When D_p is taken equal to D_{pc} , Equations 2, 4, and 6 can be combined to give

$$D_{pc} = \frac{-15.3 \mu}{\sqrt{\rho_p P_2}} + \sqrt{\frac{234 \mu^2}{\rho_p P_2} + \frac{(2.05)(10)^4 D_c^2 P_2}{\rho_p V_0 P_0}} \quad (7)$$

The characteristic diameters for each stage of the impactor can be calculated directly from Equation 7. When $D_p/L \times (10)^{-4} < 2.7$, Equations 2, 3, and 6 must be solved by approximation. For any test, after D_{pc} was calculated for

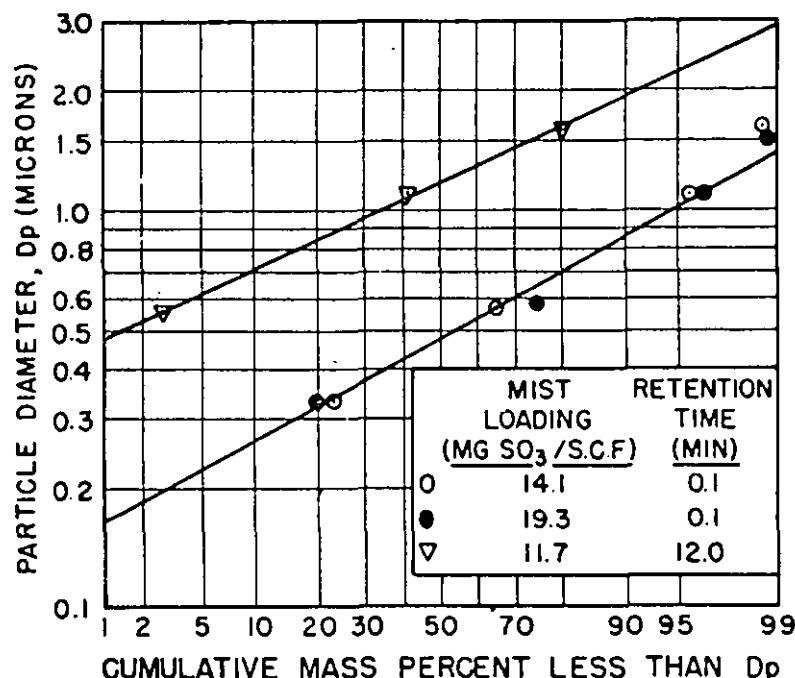


Figure 1. Cumulative particle size distribution of sulfuric acid mist generated in the laboratory is a function of mist loading and retention time

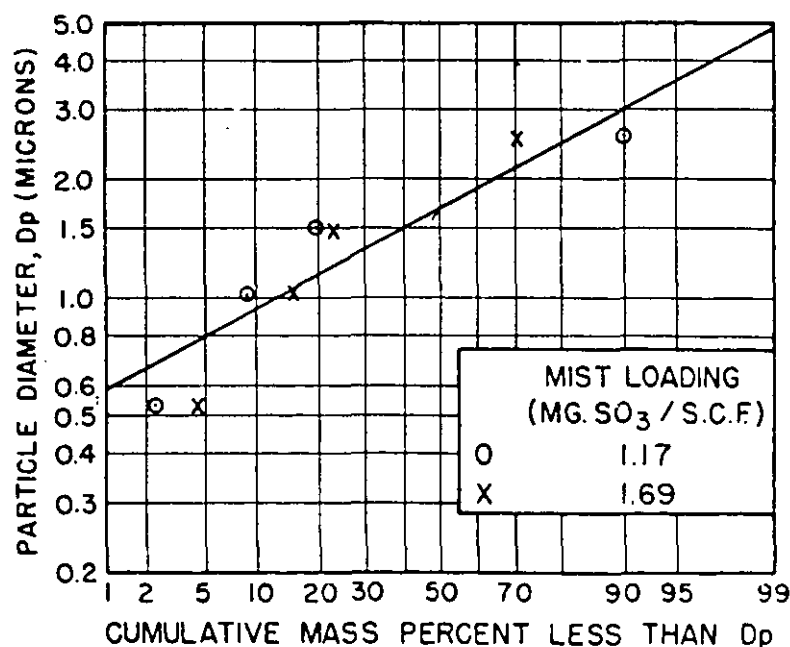


Figure 2. Cumulative particle size distribution of sulfuric acid plant aerosols

Table II. Particle-Size Distribution Curve Calculation

Jet No.	D_{pe}	% Collected	Cumulative % < D_{pe}
1	3.14	0.39	99.61
2	1.63	0.68	98.93
3	1.10	2.74	96.19
4	0.57	21.60	74.59
5	0.33	54.73	19.86
Glass filter			
1	...	19.86	
2	...	0	

each stage, cumulative distributions as shown in Figures 1 to 3 were calculated. For example, consider the curve given for a mist loading of 19.3 mg. of sulfur trioxide per standard cubic foot and a retention time of 0.1 minute in Figure 1. The data for this test are given in Table II. As 19.86% of the mist was not collected by the last stage, this percentage of the mist has diameters smaller than 0.33 micron, the D_{pe} for the last stage. The cumulative per cent not collected by stage 4 was 19.86% + 54.73% = 74.59%. This calculation was continued as shown in Table II and then the curve was plotted as shown in Figure 1.

Field Measurements

The impactor and auxiliary apparatus have been used extensively at plants throughout the country. Measurements of particle-size distributions of aerosols within and leaving eight different acid plants have been measured, and collection efficiencies of different types of full-scale dust and mist collectors for different particle sizes have shown that the impactor is a valuable and practical instrument for field use.

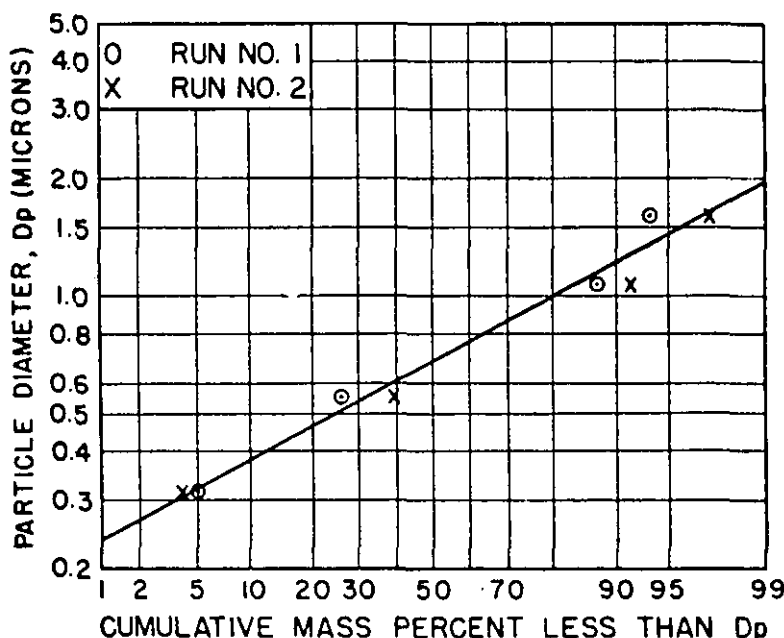


Figure 3. Distribution of phosphoric acid mist leaving typical plant can be used to control process. Analysis of aerosols under varying plant conditions aids in the selection of new equipment

The particle-size distribution and loading of the mist leaving a new contact sulfuric acid plant are shown in Figure 2. This plant is of the Leonard-Monsanto design, a single unit with a rated capacity of 400 tons per day on a 100% sulfuric acid basis. Process gases in phosphoric acid plants are usually saturated with 5 to 40% water vapor, depending on the process design and operation, and adiabatic measurements are required. A typical particle-size distribution for the acid mist leaving a phosphoric acid plant is given in Figure 3.

Particle-size determinations by chemical company personnel make possible determination of acceptable stack discharges of aerosols, evaluation of installed collection equipment, rational selection and design of equipment, and recognition of potential problems early in the development of new processes.

Acknowledgment

The many helpful suggestions made by H. F. Johnstone, W. E. Ranz, Herbert Kraemer, J. B. Wong, and G. R. Gillespie on the design and operation of cascade impactors are gratefully acknowledged.

H. P. Willett and other engineers at the Chemical Construction Corp. assisted in the design of the in-line impactor.

D. A. Moore assisted in the experimental aspects of the work and in the preparation of a company report on which this discussion is based. W. M. Davis made several helpful suggestions.

Nomenclature

C = empirical correction factor for resistance of gas to movement

of small particles as defined in Equation 3, 4, or 5

D_c = diameter of impactor jet, cm.

D_p = diameter of aerosol particle, microns

D_{pe} = characteristic diameter of aerosol particle for impactor stage, microns

g_c = 1 (g.)/(atm.)(cm.)(sec.)²

L = mean free path for gas molecules, cm. $L = 2\mu/\rho\bar{v}$

P_0 = absolute pressure at inlet to impactor, atm.

P_2 = pressure after jet, atm.

ΔP = pressure drop across impactor, inches Hg

T_2 = temperature after jet, °K.

V_0 = gas flow at inlet to impactor, cc./second. $V_0 = 24.5 (\Delta P)^{0.44}$ for air at 25° C. and 14.7 p.s.i.a.

v_2 = average linear velocity of gas through jet, at P_2 and T_2 , cm. per second

$\bar{v}$ = average molecular velocity of gas, cm. per second $\bar{v} = (10)^{1/2}$

ψ = dimensionless inertial parameter, $\psi = C_p v_2 D_p^2 / 18 \mu D_c (10)^{1/2}$

ψ_{50} = inertial parameter for impactor efficiency of 50%

ρ = gas density after jet at T_2 and P_2 , gram per cc.

ρ_p = density of aerosol particle, gram per cc.

μ = viscosity of gas after jet at T_2 and P_2 , g./(cm.)(sec.)

Literature Cited

- (1) Ferry, R. M., Farr, L. E., Hartmann, M. G., *Chem. Revs.* **44**, 389 (1949).
- (2) Gillespie, G. R., Eng. Expt. Sta. Univ. Illinois, Tech. Rept. 9, SO-1010 (1953).
- (3) Gillespie, G. R., Johnstone, H. F., *Chem. Eng. Progr.* **51**, 74F (1955).
- (4) Greenburg, L., Smith, G. W., U. S. Bur. Mines Rept. Invest. 2392 (1922).
- (5) Hatch, T., Warren, H., Drinker, P., *J. Ind. Hyg. Toxicol.* **14**, 301 (1932).
- (6) Katz, S. H., others, U. S. Pub. Health Bull., **144**, 69 (1925).
- (7) Laskin, S., in "Pharmacology and Toxicology of Uranium Compounds," ed. by C. Voegtlin and H. C. Hodge, McGraw-Hill, New York, 1949.
- (8) May, J. R., *J. Sci. Instr.* **22**, 187 (1945).
- (9) Owens, J. S., *Proc. Roy. Soc. (London)* **A 101**, 18 (1922).
- (10) Pilcher, J. M., Mitchell, R. I., Thomas, R. E., *Proc. Chem. Spec. Man. Assoc.* (December 1955).
- (11) Ranz, W. E., Wong, J. B., *AMA Arch. Ind. Hyg. and Occupational Med.* **5**, 464 (1952).
- (12) Ranz, W. E., Wong, J. B., *IND. ENG. CHEM.* **44**, 1371 (1952).
- (13) Sonkin, L. S., *J. Ind. Hyg. Toxicol.* **28**, 269 (1946).
- (14) Wilcox, J. D., *AMA Arch. Ind. Hyg. and Occupational Med.* **7**, 376 (1953).
- (15) Wilcox, J. D., Van Antwerp, W. R., Jr., *AMA Arch. Ind. Health* **11**, 422 (1955).

RECEIVED for review March 21, 1957

ACCEPTED August 14, 1957

Annual Meeting; AIChE, Boston, Mass., December 1956.

APPENDIX H

PROJECT PARTICIPANTS

Project Participants

Environmental Protection Agency
Office of Air Quality Planning and Standards
Emission Standards and Engineering Division

Thomas F. Lahre

Project Officer
National Air Data Branch

MONSANTO RESEARCH CORPORATION

William R. Fearheller, Jr.

Research Group Leader

Harlan D. Toy

Research Engineer, Team
Leader

Mark T. Thalman

Research Chemist

Jesse R. McKendree

Technician (Field Sampling)

Patrick Hughes

Technician (Field Sampling)

Roger Etter

Technician (Field Sampling)

Francine A. Kulik

Research Chemist (Analysis)

Windle H. McDonald

Technician (Analysis)

Joan T. Hecathorn

Secretary