

AP42 Section:	11.18
Title:	<i>Correspondence and comments</i>
<p>Note: This material is related to a section in <i>AP42, Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources</i>. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/</p> <p>The file name refers to the file number, the AP42 chapter and then the section. The file name "rel01_c01s02.pdf" would mean the file relates to AP42 chapter 1 section 2. The document may be out of date and related to a previous version of the section. The document has been saved for archival and historical purposes. The primary source should always be checked. If current related information is available, it will be posted on the AP42 webpage with the current version of the section.</p>	

MEMORANDUM

DATE: August 25, 1988

TO: Ken Schuster/Jerry Clayton

FROM: Robert Wooten 

SUBJECT: Revision of Review of Emission Test Report for American Rockwool, Inc., Spring Hope, Nash Co., N.C., performed May 17-19, 1988.

The subject report was received July 1, 1988 with additional information received August 8, 1988. The test was performed by personnel of the Aluminum Company of America (ALCOA). The main purpose of the test was to determine fluoride emissions when producing rockwool with spent pot linings (SPL) salvaged from ALCOA's aluminum production operations. It was desired to show to what extent fluoride emissions increase when SPL is substituted for coke. It was also desired to show to what extent the addition of lime decreases fluoride emissions. The filter used for fluoride testing was weighed for particulate determination. Sulfur dioxide was also measured.

The method used for particulate determination involved weighing the sampling thimble used to collect particulate fluorides. Due to the filter being cellulose, it tended to gain atmospheric moisture readily which made the accuracy of the weighings slightly uncertain. Also, probe and filter holder washings were omitted. The particulate values should be somewhat less accurate than a properly performed Method 5 test.

The particulate emission test is insufficient to determine compliance. EPA Method 5 should be performed on the cupola stack and the duct scrubber stack. To meet the requirement of condition 5 of the permit, it will be necessary to determine the input to the blow chamber, the cleaning process transfer cyclone, and the bagging process transfer cyclone.

Sulfur dioxide emissions were determined using a method said to be like EPA Method 15. None of the necessary supporting data was provided. It is not possible to accept the SO₂ results since there is nothing to show that correct procedures were followed.² Also, the heat input is undetermined and SO₂ may come from other than fuel.

The method used for fluoride measurement was ALCOA Method 4075A. This method is similar to EPA Method 13B. After discussing certain details of the the test method with Gary McAllister of the EPA, I have concluded that Method 4075A should give accurate results for this source.

The attached table summarizes the fluoride emissions for the three operating conditions tested. The charge makeup and number of charges per hour given in Table 1 on page 2 of the report should be regarded as only nominal. I computed SPL, Lime, and coke charge rates from the production records covering the shift during which testing was performed. You may wish to have American Rockwool verify my numbers as their production records as reproduced in the report are a little confusing.

The first test run of condition B showed lower fluorides than the next three runs. It is not possible to say if the test run is flawed. If it is omitted, the average fluoride emission rate increases 23% (7.84 lbs/hr to 9.67 lbs/hr).

Only two fluoride runs were performed for condition D due to process problems. I accept the results of the two test runs performed.

For condition A, I calculated an emission factor of .000259 lb F/ lb coke. For condition B, I got .00557 lb F/ lb SPL (using 7.41 lb F/hr). This is 21.5 times greater than for coke pound for pound. Since more than a pound of SPL will replace a pound of coke, the actual fluoride increase should be greater.

For condition D, I found .00205 lb F/ lb SPL with .107 lb lime/ lb SPL. The emission factor for SPL with lime is 7.9 times that for coke. Note again that more SPL will be used than the weight of coke reduced.

The fluoride emission standard for this source is given in condition 11 of Permit # 3578R7. The use of 1420 lbs/hr of SPL in both cupolas combined is allowed. The fluoride emissions must not increase over 2.8 lbs/hr (total) when using SPL compared to what the fluoride emissions would be without SPL.

Test condition B shows a fluoride increase of 7.11 lb/hr over condition A with an SPL input of 1329.8 lb/hr (90 lb/hr less than the maximum allowed). If the first test run is omitted, the increase is 8.94 lb/hr. The B condition test shows that the cupolas violate the permitted fluoride increase limit set in the permit when SPL without lime is used.

Test condition D was allowed by permit for a limited time to establish emissions if lime is added along with the SPL. Condition D shows a fluoride increase of 4.37 lb/hr over condition A with a SPL input of 2382 lb/hr. If the emission factor found from this test is applied to the 1420 lb/hr SPL allowed by the permit, the fluoride emission due to SPL is 2.911 lb/hr. Of course, when SPL is used, it replaces coke at somewhat more than a pound of SPL per pound of coke. If it were a pound for a pound, the fluoride increase would be .001791 lbF/lb SPL or, for 1420 lb SPL/hr, 2.54 lbF/hr increase which is within the 2.8 lb/hr increase allowed.

As I understand it, the condition D test was not intended to determine compliance with Permit # 3578R7 operating conditions and emission limits. The results are useful for predicting fluoride emissions when lime is added to the cupolas in the proportions used for the Condition D test.

According to Ken Schuster, the real fluoride concern is that actual fluoride emissions not exceed six (6) pounds per hour and that the annual fluoride increase due to using SPL not exceed three tons. The six lb/hr figure comes from modeling. The 2.8 lb/hr increase figure in the permit comes from assumptions (apparently incorrect) about pre-SPL fluoride emissions and the six lb/hr modeling value.

Attachment

cc: Mike Aldridge
Central File via Lee Daniel

OPERATING CONDITIONS AND FLUORIDE EMISSIONS

		<u>lbs/charge</u>	<u>charges/hr</u>	<u>lbs charged/hr</u>	<u>Average Fluoride lbs/hr</u>		
Cup	loa	I	II	I	II	Total	combined
<u>17MAY88 - Condition A</u>							
SPL		0	0	0	0	0	
Lime		0	0	0	0	0	.73
Coke		388.8	388.6	3.63	3.63	1411	1411
						2822	
<u>18MAY88 - Condition B</u>							
SPL		207.3	218.7	3.25	3.00	673.7	656.1
Lime		0	0	0	0	0	0
Coke		264.8	268.5	3.25	3.00	860.6	805.5
						1666.1	
							* 9.67
<u>19MAY88 - Condition D</u>							
SPL		427.3	428.0	3.11	2.46	1329	1053
Lime		23.86	24.38	6.00	4.56	243.2	111.2
Coke		143.8	145.7	3.11	2.46	447.2	358.4
						805.6	
							5.10

* If the first of the four runs is omitted since it is much lower than the other three.

This attachment contains:

- A) Field testing raw data.
- B) Preliminary field testing reduced data.
- C) Laboratory raw and reduced data.
 - Sulfur dioxide data.
 - Fluoride.
 - 1) Nozzle data.
 - 2) Thimble data.
 - 3) Impinger data.
 - 4) SIE calibration plots.

RECEIVED
JUL 13 1988

AIR QUALITY TECH SERVICES

For ease of identification, the LSN/ID# is provided for each sample obtained in Run 4.

<u>Sample</u>	<u>LSN/ID#</u>
SO ₂ - Impinger	804222/539030
SO ₂ - Blank	804221/539026
F - Nozzle	804244/539022
F - Thimble	804233/539011
F - Impingers	804004/539000
F - Nozzle Blank	804240/539018
F - Thimble Blank	Blk Thimble
F - Impinger Blank	804229/539007

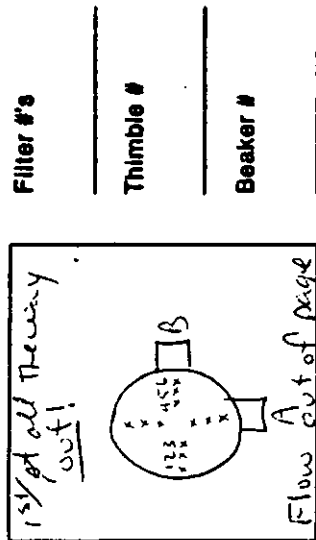
PARTICULATE TESTING RAW DATA SET



ALCOA

Plant SPRING HOPE
 Unit# Inter/Outlet OUTLET
 Operator ST ROBUCK
 Date 2/5/88
 Run No. 4
 Sample Box No. 1874
 Meter Box No. 1.90
 Meter DH @ 5.1
 K Factor 5.1
 Pilot Cp Pilot Leak Ck
 Leak Rate Before .006 CFM @ 15 Hg
 Leak Rate After .005 CFM @ 12 Hg
 Static Press., In. H₂O (+/-)

Start Time 7:45 PM AM
 Finish Time 8:59 PM AM
 Meter Yd 1.00



SCHEMATIC OF TESTING LOCATION

Ambient Temperature, °F 35
 Barometric Pressure, In. Hg
 Assumed Moisture, %
 Heater Box Setting
 Probe Length, Ft. No.
 Nozzle Diameter, In.
 Probe Heater Setting
 % O₂
 % CO₂
 H₂O, ml 10
 Silica Gel, gm 9.1
 Total V_{ic} 19.1
 Stack Area, Ft.²
 Pyrometer No.

Filter #s

Thimble #

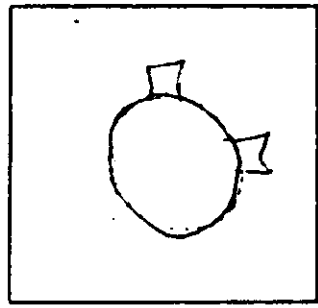
Beaker #

TRAVERSE POINT NUMBER	MIN/PT	SAMPLING TIME θ, min.	VACUUM In. Hg	STACK TEMPERATURE T _s , °F	VELOCITY HEAD DP, in.	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER DH, In. H ₂ O	Initial Volume: GAS SAMPLE VOLUME V _g , Ft. ³	GAS SAMPLE TEMPERATURE AT DRY GAS METER		TEMP. OF GAS LEAVING CONDENSER OR LAST IMPINGER T _c , °F	SAMPLE BOX TEMPERATURE T _b , °F
								INLET T _m in., °F	OUTLET T _m out., °F		
B 6	5	5	3	78	.17	1.4	497.30	52	41	33	
5	10	10	3	78	.19	1.5	500.64	59	44	36	
4	15	15	3	78	.25	2.0	504.30	64	47	40	
3	20	20	4	76	.21	1.7	507.73	67	49	40	
2	25	25	5	75	.29	2.3	511.75	72	51	41	
1	30	30	5	75	.22	1.8	515.473	72	52	41	
A 1	5	5	6	76	.21	1.7	519.05	65	52	44	
2	10	10	6	77	.28	2.3	523.22	71	53	47	
3	15	15	7	75	.28	2.3	527.21	73	53	49	
4	20	20	7	77	.30	2.6	531.77	78	54	53	
5	25	25	6	76	.32	2.7	536.17	78	55	53	
6	30	30	5	77	.18	1.5	539.725	77	55	53	
TOTAL				918	5.868			AVG. 828	AVG. 606		
AVERAGE				77	0.489	1.98	45.738	AVG. 60			

PARTICULATE TESTING
RAW DATA SET

Plant Am Refiner
Unit# 8-7N Inlet/Outlet STACK
Operator JVJ
Date 09-02-05
Run No. 4
Sample Box No. 2165
Meter Box No. 106
Meter DH@ 106
K Factor 0.13
Pilot Cp 0.34 Pilot Leak Ck ✓
Leak Rate Before 0.0013 CFM @ 15 Hg
Leak Rate After 0.0013 CFM @ 10 Hg
Static Press., In. H₂O (+/-) 0.22

Start Time 7:48 AM
Finish Time 8:59 PM
Meter Yd 0.95



Filter #'s _____
Thimble # _____
Beaker # _____

Ambient Temperature, °F _____
Barometric Pressure, In. Hg 30.28
Assumed Moisture, % _____
Heater Box Setting _____
Probe Length, Ft. _____ No. _____
Nozzle Diameter, In. _____
Probe Heater Setting _____
% O₂ _____
% CO₂ _____
H₂O, ml -5
Silica Gel, gm 12.8
Total Vic 7.8
Stack Area, Ft² _____
Pyrometer No. _____

SCHEMATIC OF TESTING LOCATION

TRAVERSE POINT NUMBER	MIN/PT	SAMPLING TIME θ, min.	VACUUM In. Hg	STACK TEMPERATURE T _s , °F	VELOCITY HEAD DP, in.	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER DH, In. H ₂ O	Initial Volume: GAS SAMPLE VOLUME V _m , Ft ³	GAS SAMPLE TEMPERATURE AT DRY GAS METER		TEMP. OF GAS LEAVING CONDENSER OR LAST IMPINGER T _c , °F	SAMPLE BOX TEMPERATURE T _b , °F
								Inlet T _m in, °F	Outlet T _m out, °F		
A-1	5	5	5	78	0.20	1.6	593.09	51	36	30	252
2	10	5	5	78	0.25	2.0	596.86	61	38	33	246
3	15	5	5	78	0.28	2.28	600.90	66	40	34	250
4	20	5	5	79	0.22	1.8	604.60	68	42	34	253
5	25	6	6	79	0.31	2.5	608.84	68	42	34	255
6	30	5	5	77	0.21	1.7	612.47	72	46	37	258
B-1	35	6.5	6.5	75	0.28	2.28	616.56	51	43	36	270
2	40	5	5	76	0.28	2.28	620.56	64	45	39	267
3	45	6	6	75	0.31	2.5	624.85	71	46	40	260
4	50	5	5	77	0.26	2.1	628.83	75	47	42	258
5	55	5	5	76	0.25	2.0	632.69	78	48	43	250
6	60	5	5	75	0.15	1.2	635.89	77	50	45	252
TOTAL				923	5.97	24.24		AVG. 802	AVG. 523		
AVERAGE				77	0.498	2.02	46.21	AVG. 54			

PARAMETER SHEET

PLANT Am Rockwood
 CITY SpHox STATE NC
 DATE 88-02-05



RUN 4 METHOD 5/6
 UNIT Both

☐ INLET
☒ OUTLET STACK
 START TIME 7:48 AM/PM
 FINISH TIME 8:59 AM/PM
 METER OPERATOR SVA

COMMENTS:

~ 99%

Preliminary Results

METER BOX NO. 2168
 DH @ 1.86

FILTER NO. 5
 THIMBLE NO. _____

DATA		
1	Pb	<u>30.28</u> "Hg
2	Static	<u>-0.22</u> "wc
3	Vlc	<u>7.8</u> ml
4	Mn	g
5	θ	<u>60</u> min
6	% O ₂	<u>19</u>
7	% CO ₂	<u>2</u>
8	DH	<u>2.02</u> "wc
9	Cp	<u>0.84</u>
10	Tm	<u>54</u> F
11	\sqrt{Dp}	<u>0.498</u> $\sqrt{"wc}$
12	Ts	<u>77</u> F
13	Vm	<u>46.21</u> ft ³
14	Dn	<u>0.294</u> in
15	As	<u>28.27</u> ft ²
16	Yd	<u>0.95</u> cf/cf

RESULTS			Assume Bw = 2%
Vmstd	<u>45.84</u>	dscf	
Vwstd	<u>0.367</u>	scf	
Bwo	<u>0.008</u>		
Md	<u>29.08</u>	lb/lb-mole	29.08
Ms	<u>28.99</u>	lb/lb-mole	28.86
Vs	<u>27.98</u>	ft/s	28.04
% I	<u>98.0</u>		99.0
acfm	<u>47460</u>		47570
dscfm	<u>46820</u>		46360
Particulate		gr/dscf	
		lb/hr	
SO ₂	<u>104.5</u>	lb/hr	
Gaseous F		lb/hr	
Particulate F		lb/hr	
Total F		lb/hr	

PARAMETER SHEET

PLANT Am Rockwood
 CITY Sp Hpe STATE NC
 DATE 88-02-05



RUN 4 METHOD 4075-A
 UNIT Both

☐ INLET
☒ OUTLET STACK
 START TIME 7:45 AM/PM
 FINISH TIME 8:59 AM/PM
 METER OPERATOR STIR

COMMENTS:

~103%

Preliminary Results

METER BOX NO. 1874
 DH @ 1.90

FILTER NO. _____
 THIMBLE NO. 4

	DATA	
1 Pb	<u>30.28</u>	"Hg
2 Static	<u>-0.22</u>	"wc
3 Vlc	<u>19.1</u>	ml
4 Mn		g
5 θ	<u>60</u>	min
6 % O ₂	<u>19</u>	
7 % CO ₂	<u>2</u>	
8 DH	<u>1.98</u>	"wc
9 Cp	<u>0.84</u>	
10 Tm	<u>60</u>	F
11 \sqrt{Dp}	<u>0.489</u>	$\sqrt{"wc}$
12 Ts	<u>77</u>	F
13 Vm	<u>45.738</u>	ft ³
14 Dn	<u>0.294</u>	in
15 As	<u>28.27</u>	ft ²
16 Yd	<u>1.00</u>	cf/cf

	RESULTS	
Vmstd	<u>47.21</u>	dscf
Vwstd	<u>0.9</u>	scf
Bwo	<u>0.019</u>	
Md	<u>29.08</u>	lb/lb-mole
Ms	<u>28.87</u>	lb/lb-mole
Vs	<u>27.53</u>	ft/s
% I	<u>103.7</u>	
acfm	<u>46700</u>	
dscfm	<u>45570</u>	
Particulate	<u>—</u>	gr/dscf
	<u>—</u>	lb/hr
SO ₂	<u>—</u>	lb/hr
Gaseous F	<u>0.08</u>	lb/hr
Particulate F	<u>0.63</u>	lb/hr
Total F	<u>0.70</u>	lb/hr

System 10: ^{miss Log - E.C.L. - L. Ferry - 80100110}
 Submitted: American Rockwell Compliance method 6
 8 Samples
 LSN 804218 thru 804225

SO₂ By EPA method-6
 Barium Perchlorate = 0.01 N

	10/100/5 (T.Vol) 760ml	10/100/5 (T.Vol) 720ml
Be ml	804218 ml	804219 ml
70	0.05	2.70
70		4.80
70	0.00	0.05
70	0.05	2.65
	0.05	2.10
	2.60 ✓	2.05
	1665.56 mg/L ✓	1,313.23 mg/L ✓
	(1265.8) Total 1266. mg ✓	(945.5) Total 946. mg ✓

10/100/5 (T.Vol) 620	20 ml (T.Vol) 1,150	10/100/5 (T.Vol) 625
804220 ml	804221 ml	804222 ml
7.35	7.40	9.40
4.80	7.35	7.40
2.55	0.05	2.80
0.05	0.05	1.95 ✓
2.50 ✓	0.00 ✓	1249.17 mg/L ✓
1,601.5 mg/L ✓		
(992.9) Total 993. mg ✓	NOT Detected ✓	(780.7) Total 781. mg ✓

$$\frac{\text{ml} \times \text{N} \times 32.03 \times 1000 \times \text{Dilution}}{\text{ml Sample}} = \text{mg/L}$$

$$\left\{ \begin{array}{l} \text{mg/L} \times \text{Total Volume (Decimal)} \\ = \text{Total mg as SO}_2 \end{array} \right.$$

22

88-021018

14H100089

8-5

10/100/5 (T.Vol)
675
804223
ml

10/100/5 (T.Vol)
675
804224
ml

10/100/5 (T.Vol)
830
804225
ml

Systa
Subm
LSI

RUN
5

11.50
9.40
2.10
2.05
2.05 J

13.30
11.50
1.80
1.05
1.75 J

15.05
13.30
1.75
1.70 J
1089.02 mg/L

Sub
Fac
0.1

1313.23 mg/L

1121.05 mg/L

886.14

886 Total
mg J

756.7

757 Total
mg J

903.9

904 Total
mg J

10/100/10 (T.Vol)
675 ml
804224 DUPLICATE
ml

10/100/5 + 5.052 mg S x 1.998
= 10.094 mg S
804218 Spk
ml

18.50
15.05
3.45 J
1.05
3.40 J

31.85
0.00
31.85
1.05
31.80 J

1089.02 mg/L

735.1

735 Total
mg J

SPIKED SAMPLE ANALYSIS = 10.186 mg
UNSPIKED SAMPLE ANALYSIS = 0.833 mg
RECOVERED = 9.353 mg
ADDED = 10.094 mg
90 Recovery = 92.66 J
= 92.790

TOTAL VOLUME OF SAMPLE RECEIVED

LSN	Total Volume	LSN	Total Volume
804218	760 ml	804222	625 ml
219	720 ml	223	675 ml
220	620 ml	224	675 ml
221	1,150 ml	225	830 ml

ml/L
10-08-18

2760:189

L. Penix
14H 1000891
amer Rods
nozzle wash

88-021020

ECL

"

- all samples directly distilled
to 500 ml \pm H₂SO₄ STE
804245 contained acetone
not analyzed.

88/01/04

07:25:50

TRAY NO. 1

TRAY I.D. 1

JOB ORDER NO.

CALIBRATION BASED ON TRAY 1

MV READOUT ONLY 16 POSITIONS

POS. 1	MV= 1441
POS. 2	MV= 1492
POS. 3	MV= 1490
POS. 4	MV= 917
POS. 5	MV= 325
POS. 6	MV= -258
POS. 7	MV= 325 20 pp 771
POS. 8	MV= 600 804 237
POS. 9	MV= 449 238
POS. 10	MV= 399 239
POS. 11	MV= 1599 240
POS. 12	MV= 1607 240
POS. 13	MV= 1715 240
POS. 14	MV= 1352 241
POS. 15	MV= 1515 242
POS. 16	MV= 1541 243

88-021020

88/01/04

09:47:33

TRAY NO. 2

TRAY I.D. 2

JOB ORDER NO.

CALIBRATION BASED ON TRAY 1

MV READOUT ONLY 8 POSITIONS

POS. 1	MV= 1254
POS. 2	MV= 1355 804 244
POS. 3	MV= 1333 246
POS. 4	MV= 1416 247
POS. 5	MV= 1451
POS. 6	MV= 1453 801 719
POS. 7	MV= -126 804 000 .5013
POS. 8	MV= -125

88-021020

88-01

1 2 MG F/L 17 MV

88/01/84 13:08:10
 TRAY NO. 1 TRAY I.D. 1 JOB ORDER NO.88-021020
 CALIBRATION BASED ON TRAY 1
 CALIBRATION: MV/KB: 16 POSITIONS-NO DISTILLED STD.;

1 2 MG F/L 917 MV
 2 20 MG F/L 325 MV MG/L FLUORIDE= 20.000
 SLOPE(MV20-MV2)= -592
 3 0.20 MG F/L 1490 MV MG/L FLUORIDE= .200
 4 200 MG F/L -258 MV MG/L FLUORIDE= 200.00
 SLOPE(MV200-MV20)= -583

POS 5 JOB ORDER#88-021020 LSN#=20 PPM
 MV= 325 ALIQUOT= 10 MLS
 ORIG. SPL.= 500 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 20
 KNOWN CONC.= 0 CALC. CONC.= 20.000(MG F/L)
 DIST. FACTOR 0 APPLIED

POS 6 JOB ORDER#88-021020 LSN#=804237
 MV= 600 ALIQUOT= 10 MLS
 ORIG. SPL.= 0 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 6.863
 KNOWN CONC.= 0 CALC. CONC.= 3.432(MG TF)
 DIST. FACTOR 0 APPLIED

POS 7 JOB ORDER#88-021020 LSN#=804238
 MV= 449 ALIQUOT= 10 MLS
 ORIG. SPL.= 0 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 12.347
 KNOWN CONC.= 0 CALC. CONC.= 6.174(MG TF)
 DIST. FACTOR 0 APPLIED

POS 8 JOB ORDER#88-021020 LSN#=804239
 MV= 399 ALIQUOT= 10 MLS
 ORIG. SPL.= 0 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 14.990
 KNOWN CONC.= 0 CALC. CONC.= 7.499(MG TF)
 DIST. FACTOR 0 APPLIED

POS 9 JOB ORDER#88-021020 LSN#=804240
 MV= 1607 ALIQUOT= 10 MLS
 ORIG. SPL.= 100 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .121
 KNOWN CONC.= 0 CALC. CONC.= .605(MG F/L)
 DIST. FACTOR 0 APPLIED

$1,040 \text{ mL} = .62 \text{ mg}$

POS 10 JOB ORDER#88-021020 LSN#=804240
 MV= 1715 ALIQUOT= 10 MLS
 ORIG. SPL.= 100 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .073
 KNOWN CONC.= 0 CALC. CONC.= .365(MG F/L)
 DIST. FACTOR 0 APPLIED

$1,040 \text{ mL} = 0.38$

$\rightarrow \text{an-c } 0.5 \text{ mg}$

POS 11 JOB ORDER#88-021020 LSN#=804241
 MV= 1352 ALIQUOT= 10 MLS
 ORIG. SPL.= 0 MLS
 S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .354
 KNOWN CONC.= 0 CALC. CONC.= .177(MG TF)
 DIST. FACTOR 0 APPLIED

QIG. SPL.= 0 MLS
WT.= 0 DIL. VOL.= 500 MLS MG F/L= 14.998
KNOWN CONC.= 0 CALC. CONC.= 7.499(MG TF)
DIST. FACTOR 0 APPLIED

POS 9 JOB ORDER#88-021020 LSN#=804240
MV= 1607 ALIQUOT= 10 MLS
ORIG. SPL.= 100 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .121
KNOWN CONC.= 0 CALC. CONC.= .605(MG F/L)
DIST. FACTOR 0 APPLIED

$\times 1,040 \text{ ml} = .62 \text{ mg}$

POS 10 JOB ORDER#88-021020 LSN#=804240
MV= 1715 ALIQUOT= 10 MLS
ORIG. SPL.= 100 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .073
KNOWN CONC.= 0 CALC. CONC.= .365(MG F/L)
DIST. FACTOR 0 APPLIED

$\times 1,040 \text{ ml} = 0.38$

} are 0.5 mg

POS 11 JOB ORDER#88-021020 LSN#=804241
MV= 1352 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .354
KNOWN CONC.= 0 CALC. CONC.= .177(MG TF)
DIST. FACTOR 0 APPLIED

POS 12 JOB ORDER#88-021020 LSN#=804242
MV= 1515 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .18
KNOWN CONC.= 0 CALC. CONC.= .090(MG TF)
DIST. FACTOR 0 APPLIED

POS 13 JOB ORDER#88-021020 LSN#=804243
MV= 1541 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .161
KNOWN CONC.= 0 CALC. CONC.= .081(MG TF)
DIST. FACTOR 0 APPLIED

POS 14 JOB ORDER#88-021020 LSN#=804244
MV= 1355 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .35
KNOWN CONC.= 0 CALC. CONC.= .175(MG TF)
DIST. FACTOR 0 APPLIED

500 ml

POS 15 JOB ORDER#88-021020 LSN#=804246
MV= 1333 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .383
KNOWN CONC.= 0 CALC. CONC.= .192(MG TF)
DIST. FACTOR 0 APPLIED

POS 16 JOB ORDER#88-021020 LSN#=804247
MV= 1416 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .272
KNOWN CONC.= 0 CALC. CONC.= .136(MG TF)
DIST. FACTOR 0 APPLIED

2760:187

Penix 88-020916

4H400891 ECL

Cedarwood 10
thimble

thimbles scraped out & total particulate weighed. Thimbles were cut in half and .5g CaO added to $\frac{1}{2}$ thimble, ashed, fused \approx NaOH. Dist H_2SO_4 SIE.

5g CaO added to x.5g part. ashed, fused, dist. SIE

803999 + 804000 had 2 thimbles each. $\frac{1}{2}$ of each thimble was taken and the 2 halves were combined.

88/01/04

08:53:16

TRAY NO. 1

TRAY I.D. 1

JOB ORDER NO.

CALIBRATION BASED ON TRAY 1

MV READOUT ONLY

16 POSITIONS

POS. 1	MV=	1481	
POS. 2	MV=	1486.2	
POS. 3	MV=	1481	
POS. 4	MV=	906.2	
POS. 5	MV=	321.20	
POS. 6	MV=	-263.200	
POS. 7	MV=	1612	
POS. 8	MV=	820	
POS. 9	47N MV=	819.804001	thimble
POS. 10	N85 MV=	915.002	"
POS. 11	4N MV=	944.003	"
POS. 12	2N MV=	512.004	
POS. 13	48N MV=	457.005	
POS. 14	N71 MV=	418.006	
POS. 15	N500 MV=	412.804007	thimble
POS. 16	MV=	1484	

} outlets
88-020916

88/01/04

09:48:58

TRAY NO. 2

TRAY I.D. 2

JOB ORDER NO.

CALIBRATION BASED ON TRAY 1

MV READOUT ONLY

14 POSITIONS

POS. 1	MV=	1503	
POS. 2	1N MV=	1514	BLK sm. thimble
POS. 3	MV=	1139	
POS. 4	N58 MV=	1139	Lgc BLK thimble
POS. 5	3N MV=	122	sm. thimble + .05g NaF
POS. 6	MV=	045	
POS. 7	N76 MV=	351803998	thimble x 2
POS. 8	15 MV=	13803999	" x 2
POS. 9	N87 MV=	-42804000	" x 2
POS. 10	14 MV=	251803998	.5023
POS. 11	N11 MV=	-166803999	.5124
POS. 12	5 MV=	-179.999	.5182
POS. 13	N10 MV=	-189804000	.5013
POS. 14	16 MV=	-304.000	.5013 spike

88/01/04

14:12:56

TRAY NO. 1

TRAY I.D. 1

JOB ORDER NO.88-020916

CALIBRATION BASED ON TRAY 1

CALIBRATION; MV/KB; 16POSITIONS-NO DISTILLED STD.;

1	2 MG F/L	906 MV✓	
2	20 MG F/L	321 MV✓	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -585		
3	0.20 MG F/L	1486 MV✓	MG/L FLUORIDE= .200
4	200 MG F/L	-263 MV✓	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -584		

POS 5 JOB ORDER#88-020916 LSN#=804001
MV= 819 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 2.817
KNOWN CONC.= 0 CALC. CONC.= 1.409(MG TF)
DIST. FACTOR 0 APPLIED

POS 6 JOB ORDER#88-020916 LSN#=804002
MV= 915 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.93
KNOWN CONC.= 0 CALC. CONC.= .965(MG TF)
DIST. FACTOR 0 APPLIED

POS 7 JOB ORDER#88-020916 LSN#=804003
MV= 944 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.722
KNOWN CONC.= 0 CALC. CONC.= .861(MG TF)
DIST. FACTOR 0 APPLIED

POS 8 JOB ORDER#88-020916 LSN#=804004
MV= 512 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 9.43
KNOWN CONC.= 0 CALC. CONC.= 4.715(MG TF)
DIST. FACTOR 0 APPLIED

POS 9 JOB ORDER#88-020916 LSN#=804005
MV= 457 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 11.71
KNOWN CONC.= 0 CALC. CONC.= 5.855(MG TF)
DIST. FACTOR 0 APPLIED

POS 10 JOB ORDER#88-020916 LSN#=804006
MV= 418 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 13.653
KNOWN CONC.= 0 CALC. CONC.= 6.827(MG TF)
DIST. FACTOR 0 APPLIED

POS 11 JOB ORDER#88-020916 LSN#=804007
MV= 412 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 13.979
KNOWN CONC.= 0 CALC. CONC.= 6.99(MG TF)
DIST. FACTOR 0 APPLIED

POS 11 JOB ORDER#88-020916 LSN#=804007
MV= 412 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 13.979
KNOWN CONC.= 0 CALC. CONC.= 6.99(MG TF)
DIST. FACTOR 0 APPLIED

POS 12 JOB ORDER#88-020916 LSN#=SM BLK THIMBLE
MV= 1514 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .179
KNOWN CONC.= 0 CALC. CONC.= .09(MG TF)
DIST. FACTOR 0 APPLIED

POS 13 JOB ORDER#88-020916 LSN#=LG BLK THIMBLE
MV= 1139 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .797
KNOWN CONC.= 0 CALC. CONC.= .399(MG TF)
DIST. FACTOR 0 APPLIED

POS 14 JOB ORDER#88-020916 LSN#=SM THIMBLE .05G NAF
MV= 122 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 43.831
KNOWN CONC.= 0 CALC. CONC.= 21.916(MG TF)
DIST. FACTOR 0 APPLIED

NaF/F
.05/2.21 = 22.6 mg added

POS 15 JOB ORDER#88-020916 LSN#=803998 *thimble*
MV= 351 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 17.772
KNOWN CONC.= 0 CALC. CONC.= 8.886(MG TF)
DIST. FACTOR 0 APPLIED

x2 = 17.8 mg F in thimble

POS 16 JOB ORDER#88-020916 LSN#=803999 *thimbles*
MV= 13 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 67.364
KNOWN CONC.= 0 CALC. CONC.= 33.682(MG TF)
DIST. FACTOR 0 APPLIED

2 1/2's

x2 = 67.4 mg F in 2 thimbles

88/01/04 14:17:39
TRAY NO. 2 TRAY I.D. 2 JOB ORDER NO.88-020916
CALIBRATION BASED ON TRAY 2
CALIBRATION; MV/KB; 7POSITIONS-NO DISTILLED STD.;

88/01/04 14:20:03
TRAY NO. 2 TRAY I.D. 2 JOB ORDER NO.88-020916
CALIBRATION BASED ON TRAY 2
CALIBRATION; MV/KB; 11POSITIONS-NO DISTILLED STD.;

1	.2 MG F/L	306 MV	
2	20 MG F/L	321 MV	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -585		
3	0.20 MG F/L	1486 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-263 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -584		

TRAY NO. 2 TRAY 1.D. 2 JOB ORDER NO. 88-020916
CALIBRATION BASED ON TRAY 2
CALIBRATION: MV/KB: 11 POSITIONS-NO DISTILLED STD.;

1 2 MG F/L 906 MV
2 20 MG F/L 321 MV MG/L FLUORIDE= 20.000
SLOPE(MV20-MV2)= -585
3 0.20 MG F/L 1486 MV MG/L FLUORIDE= .200
4 200 MG F/L -263 MV MG/L FLUORIDE= 200.00
SLOPE(MV200-MV20)= -584

POS 5 JOB ORDER#88-020916 LSN#=804000 *thimble*
MV=-42 ✓ ALIQUOT= 10 MLS *2 "2's"*
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 83.677
KNOWN CONC.= 0 CALC. CONC.= 41.839(MG TF)
DIST. FACTOR 0 APPLIED $\times 2 = 83.7 \text{ mg} = 2 \text{ Thimbles}$

POS 6 JOB ORDER#88-020916 LSN#=803998
MV= 251 ✓ ALIQUOT= 10 MLS *part wt = 4.0395*
ORIG. SPL.= 0 MLS
S. WT.= .5023 DIL. VOL.= 500 MLS MG F/L= 26.357
KNOWN CONC.= 0 CALC. CONC.= 2.624(XF)
DIST. FACTOR 0 APPLIED $\times 4.0395 = 106.00 \text{ mg}$
106.0 mg + 17.8 mg = 123.8 Total mg

POS 7 JOB ORDER#88-020916 LSN#=803999
MV=-166 ✓ ALIQUOT= 10 MLS *part wt = 2.6905*
ORIG. SPL.= 0 MLS
S. WT.= .5124 DIL. VOL.= 500 MLS MG F/L= 136.43
KNOWN CONC.= 0 CALC. CONC.= 13.313(XF)
DIST. FACTOR 0 APPLIED $\times 2.6905 = 358.19 \text{ mg}$
358.2 mg + 67.4 mg in thimble = 425.6 mg

POS 8 JOB ORDER#88-020916 LSN#=803999 *duplicate*
MV=-179 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= .5182 DIL. VOL.= 500 MLS MG F/L= 143.61
KNOWN CONC.= 0 CALC. CONC.= 13.857(XF)
DIST. FACTOR 0 APPLIED $\times 2.6905 = 372 \text{ mg}$

report 13.3% and 13.8% as duplicates

POS 9 JOB ORDER#88-020916 LSN#=804000
MV=-189 ✓ ALIQUOT= 10 MLS *part wt = 2.5238*
ORIG. SPL.= 0 MLS
S. WT.= .5013 DIL. VOL.= 500 MLS MG F/L= 149.38
KNOWN CONC.= 0 CALC. CONC.= 14.899(XF)
DIST. FACTOR 0 APPLIED $\times 2.5238 = 376.02 \text{ mg F}$
376.0 mg + 83.7 mg = 459.7 mg
459.7 mg / 14.9 x .5013 = 74.7 mg F sample

POS 10 JOB ORDER#88-020916 LSN#=804000
MV=-304 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= .5013 DIL. VOL.= 500 MLS MG F/L= 235.09
KNOWN CONC.= 0 CALC. CONC.= 23.448(XF)
DIST. FACTOR 0 APPLIED $\times .5013 = 117.54 \text{ mg F in spike}$
117.54 mg F in spike
74.7 mg sample
42.81 mg / 45.2 mg = 94.7%
11 / 2.21 = 45.2 mg

POS 11 JOB ORDER#88-020916 LSN#=804000
MV=-106 ✓ ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= .5013 DIL. VOL.= 500 MLS MG F/L= 107.69
KNOWN CONC.= 0 CALC. CONC.= 10.741(XF)
DIST. FACTOR 0 APPLIED $\times 2 = 21.5 \text{ dist low}$

2760:184

PILA

+ 171000891

Rockwell compliance
impingers

88-021019

ECL

11

direct dust \pm H₂ SO₄ SIE

88/01/04

12:48:55

TRAY NO. 1

TRAY I.D. 1

JOB ORDER NO.

CALIBRATION BASED ON TRAY 2

MV READOUT ONLY

16 POSITIONS

POS. 1	MV= 1468
POS. 2	MV= 1487
POS. 3	MV= 1483
POS. 4	MV= 904
POS. 5	MV= 314
POS. 6	MV= -275
POS. 7	MV= 315 20ppm
POS. 8	MV= 1678 804236
POS. 9	MV= 1711 804236
POS. 10	MV= 968 804227
POS. 11	MV= 923 228
POS. 12	MV= 1851 229
POS. 13	MV= 1613 230
POS. 14	MV= 1700 231
POS. 15	MV= 1656 232
POS. 16	MV= 1656

88/01/04

13:40:59

TRAY NO. 2

TRAY I.D. 2

JOB ORDER NO.

CALIBRATION BASED ON TRAY 2

MV READOUT ONLY

6 POSITIONS

POS. 1	MV= 1632 804233
POS. 2	MV= 1637 233 dup
POS. 3	MV= 1678 234
POS. 4	MV= 1802 235
POS. 5	MV= 953 226
POS. 6	MV= 823 226 spiro

88/01/04

11:25:19

TRAY NO. 1

TRAY I.D. 1

JOB ORDER NO. 88-021019

CALIBRATION BASED ON TRAY 1

CALIBRATION; MV/KB; 16 POSITIONS-NO DISTILLED STD.;

1 2 MG F/L 904 MV
2 20 MG F/L 314 MV MG/L FLUORIDE= 20.000
SLOPE(MV20-MV2)= -590
3 0.20 MG F/L 1483 MV MG/L FLUORIDE= .200
4 200 MG F/L -275 MV MG/L FLUORIDE= 200.00
SLOPE(MV200-MV20)= -589

POS 5 JOB ORDER#88-021019 LSN#=
MV= 315 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 19.922
KNOWN CONC.= 0 CALC. CONC.= 19.922(MG F/L)
DIST. FACTOR 0 APPLIED

POS 6 JOB ORDER#88-021019 LSN#=804236
MV= 1711 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .076
KNOWN CONC.= 0 CALC. CONC.= .760(MG F/L)
DIST. FACTOR 0 APPLIED $x.480 = .36 \text{ Total mg}$

POS 7 JOB ORDER#88-021019 LSN#=804227
MV= 968 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.556
KNOWN CONC.= 0 CALC. CONC.= 15.560(MG F/L)
DIST. FACTOR 0 APPLIED $x.335 = 5.2 \text{ Total mg}$

POS 8 JOB ORDER#88-021019 LSN#=804228
MV= 923 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.556
KNOWN CONC.= 0 CALC. CONC.= 15.560(MG F/L)
DIST. FACTOR 0 APPLIED $x.385 = 7.1 \text{ Total mg}$

POS 9 JOB ORDER#88-021019 LSN#=804229
MV= 1851 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .05
KNOWN CONC.= 0 CALC. CONC.= .400(MG F/L)
DIST. FACTOR 0 APPLIED $x.575 = 0.2 \text{ Total mg}$

POS 10 JOB ORDER#88-021019 LSN#=804231
MV= 1513 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .05
KNOWN CONC.= 0 CALC. CONC.= .400(MG F/L)
DIST. FACTOR 0 APPLIED $x.505 = .59 \text{ Total mg}$

POS 11 LSN#=804231

DIL. VOL.= 500 MLS MG F/L= .05

KNOWN CONC.= 0 DIL. VOL.= 500 MLS
DIST. FACTOR 0 APPLIED

$$x.575 = 0.2 \text{ Total mg}$$

POS 10 JOB ORDER#88-021019
MV= 1613 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS
KNOWN CONC.= 0 CALC. CONC.= .575(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.505 = .59 \text{ Total mg}$$

LSN#=804231

DIL. VOL.= 500 MLS MG F/L= .08
KNOWN CONC.= 0 CALC. CONC.= .800(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.595 = .48 \text{ Total mg}$$

POS 12 JOB ORDER#88-021019 LSN#=804232
MV= 1656 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .097
KNOWN CONC.= 0 CALC. CONC.= .970(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.500 = .48 \text{ Total mg}$$

POS 13 JOB ORDER#88-021019 LSN#=804233
MV= 1632 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .107
KNOWN CONC.= 0 CALC. CONC.= 1.070(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.525 = .56 \checkmark$$

POS 14 JOB ORDER#88-021019 LSN#=804233 DUPLICATE
MV= 1637 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .105
KNOWN CONC.= 0 CALC. CONC.= 1.050(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.525 = .55 \checkmark$$

POS 15 JOB ORDER#88-021019 LSN#=804234
MV= 1678 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .088
KNOWN CONC.= 0 CALC. CONC.= .880(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.515 = .45$$

POS 16 JOB ORDER#88-021019 LSN#=804235
MV= 1802 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .051
KNOWN CONC.= 0 CALC. CONC.= .510(MG F/L)
DIST. FACTOR 0 APPLIED

$$x.465 = .24$$

CALIBRATION BASED ON TRAY 2
CALIBRATION: MV/KB: 6 POSITIONS-NO DISTILLED STD.;

1	2 MG F/L	904 MV	
2	20 MG F/L	314 MV	MG/L FLUORIDE= 20.000.
	SLOPE(MV20-MV2)= -590		
3	0.20 MG F/L	1483 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-275 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -589		

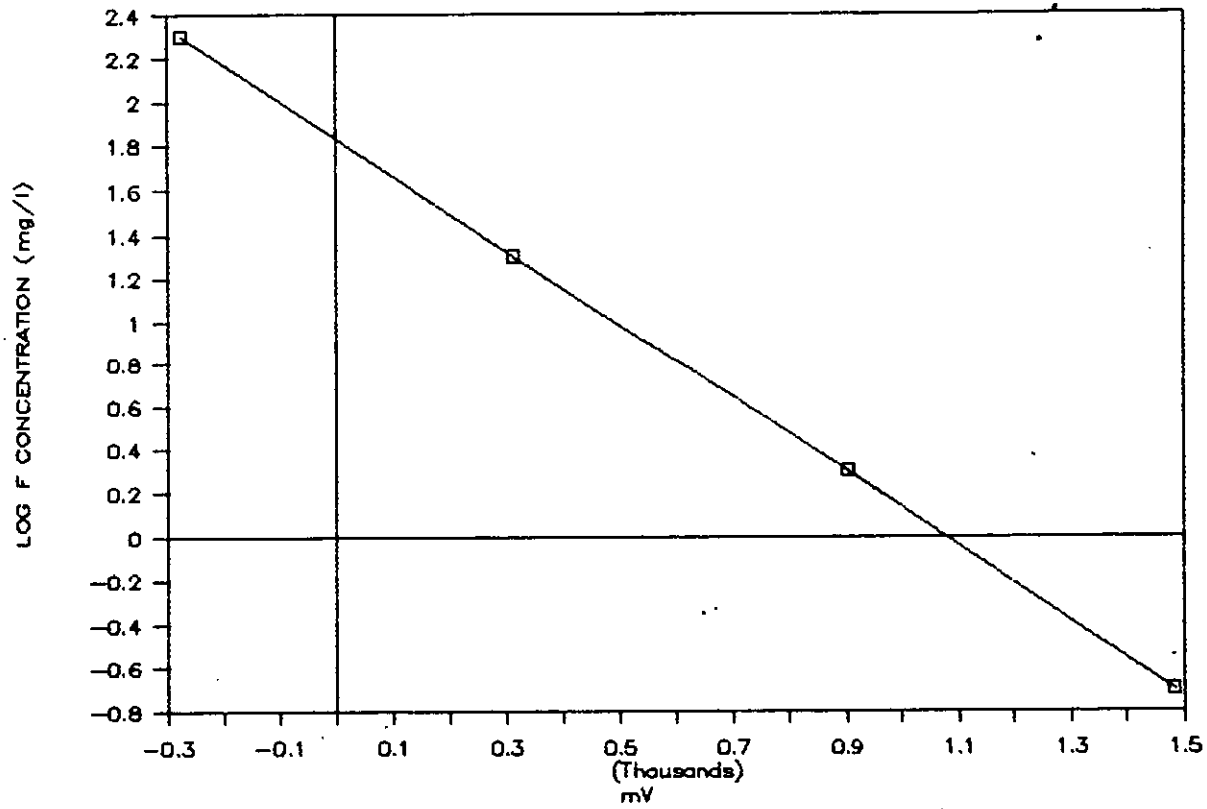
POS 5 JOB ORDER#88-021019 LSN#=804206
MV= 953 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.65
KNOWN CONC.= 0 CALC. CONC.= 16.500(MG F/L)
DIST. FACTOR 0 APPLIED $\times .255 = 4.2 \checkmark$

POS 6 JOB ORDER#88-021019 LSN#=804206 SPIKE
MV= 823 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 2.744
KNOWN CONC.= 0 CALC. CONC.= 27.440(MG F/L)
DIST. FACTOR 0 APPLIED $\frac{-16.5}{10.9 \text{ or } 109\%}$

5ml of 100 mg/L F std
50 ml samp / 500 ml
(10 ppm spike)

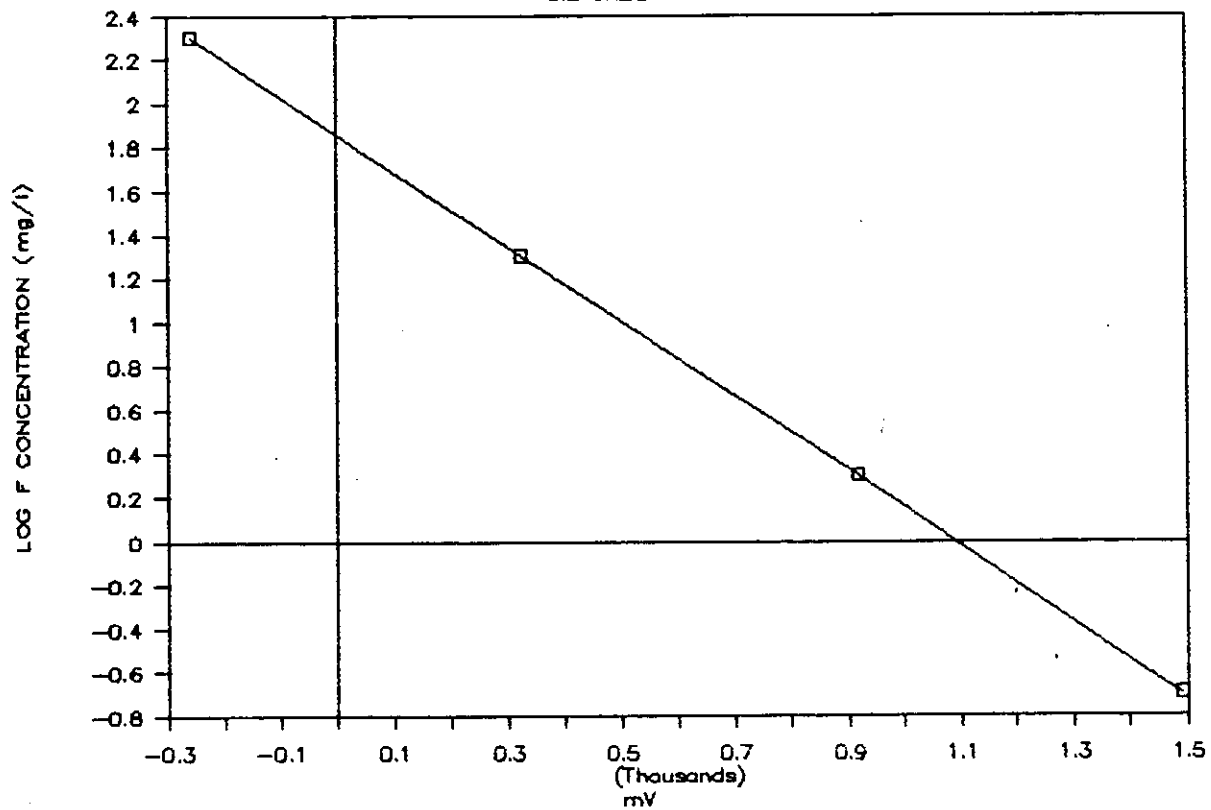
IMPINGER

SIE CALIBRATION



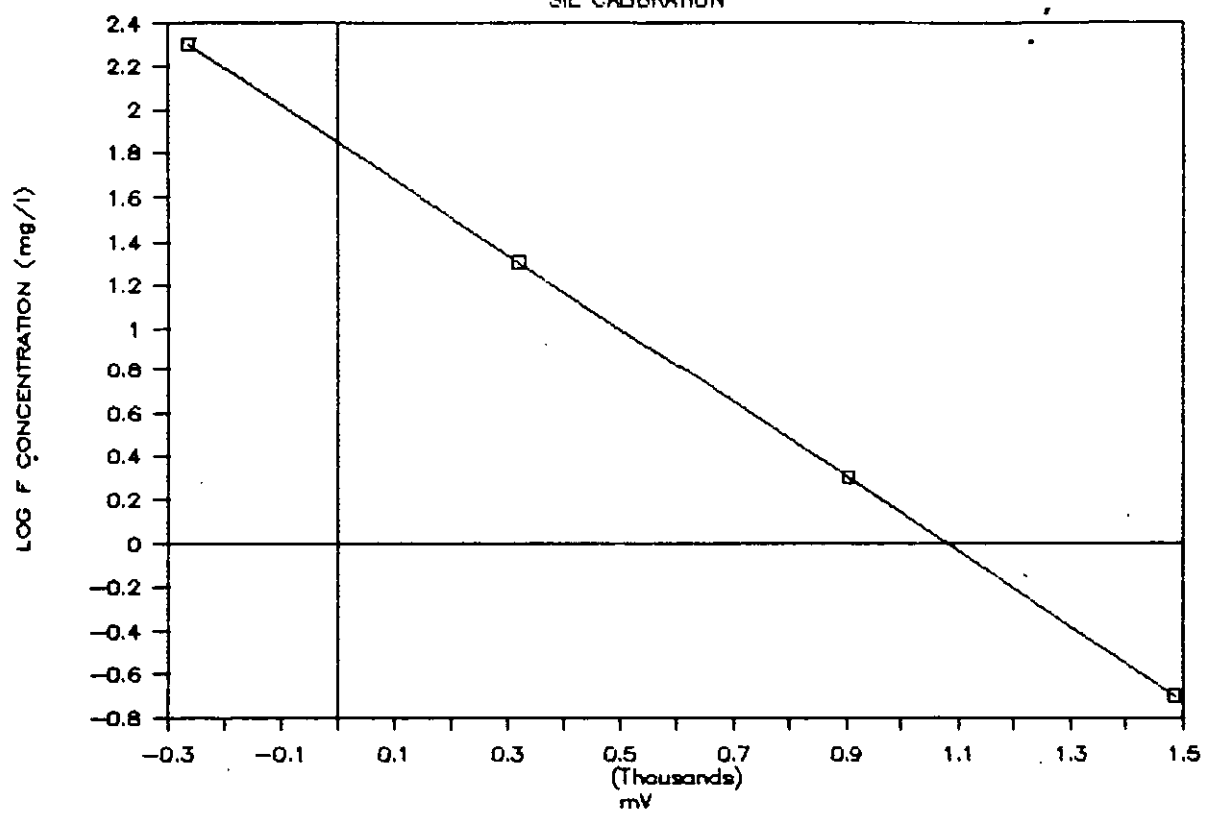
NOZZLE

SIE CALIBRATION



THIMBLE

SIE CALIBRATION



DIVISION OF ENVIRONMENTAL MANAGEMENT
AIR QUALITY SECTION
RALEIGH REGIONAL OFFICE

April 25, 1988

MEMORANDUM

TO: Mike Aldridge/Bob Wooten

FROM: Ken Schuster *KS*

SUBJECT: American Rockwool - Emission Test Reports
Spring Hope, Nash County

Several test reports for the subject company have been forwarded to you for review. The test reports forwarded were conducted on January 12-16 and February 3-5, 1988. Additional information refers to April 1986 testing and a rockwool emission test in Texas. In addition to fluorides, particulate, sulfur dioxide, and opacity were tested.

As per our earlier discussion, AR (American Rockwool Co.) initiated the use of spent pot lining (SPL) from aluminum smelting pots as a partial replacement for coke in the rockwool manufacturing process. AR began using SPL at the Spring Hope facility in August of 1986. However, AR does state that SPL was used for testing in April of 1986.

From the various tests, AR has come up with factors utilized to determine fluoride emissions both when SPL is used and when it is not. The primary concern with the fluoride emissions is that the increase due to using SPL is less than 3 tons per year. As per the permit, the hourly fluoride emissions increase is limited to 2.8 lb/hr (as per AR request) due to vegetative concerns.

At the time of the permit review (12/11/87), only information from the April 1986 testing was available. In order to determine the allowable annual increase of 1793 hr/yr and the hourly allowable of 2.8 pounds, a factor of 0.002 (.00197) lb Fl./hr/lb SPL was used (see attached Appendix I, page 9, column (B-D)/F). The January and February tests (February tests were the observed compliance tests) appear to demonstrate a lower fluoride increase of 0.0004 lb/hr/lb SPL. Due to the Jan/Feb testing showing a lower fluoride emission increase, AR plans to submit a new application for permit revision of increased SPL usage.

It should be noted that the February test appears to show compliance with the particulate and opacity standards. The source modeled SO_2 at 100 lb/hr. The actual as per testing was 144.1 lb/hr. The company will submit application for revision of modeling. Also, during the February test, 810 pounds of SPL were charged per hour per cupola. The permit application is for 710 pounds per hour per cupola. The company did not want to undercharge the amount of SPL as DEM wanted the maximum permitted charged.

Part of the permitted fluoride emissions increase is due to the crusher which is presently permitted for 13 hrs/yr of operation and as an uncontrolled source. However, AR plans to install a baghouse and application for such is forthcoming.

The Raleigh Regional Office has two primary concerns. One concern is whether the source is presently in compliance, and the other is whether the source previously operated in compliance. The February testing appears to demonstrate compliance with the present permit. The SO_2 modeling requires adjustment. The second concern arises since the source previously utilized a charge with less limestone and for which the Jan/Feb testing may not be representative. The background fluoride rate was established using the two years prior to SPL usage (October 1984 to September 1986) and determined from the actual monthly charges. The actual background fluoride rate is then dependent on a factor, most logically from the April 1986 testing. Appendix B, page 7 shows what the background could be using both the April 1986 and February 1988 factors (.02 and .002, respectively). Due to the significant difference in the factors, the background can range from 6522 lb fluoride/yr to 19,368 lb fluoride/yr. The factors discussed above are found in Appendix I, page 8, column B/A. Note that Jan/Feb SPL testing shows 0.003 vs. 0.002 as per the company. This is due to some differences in reported charges, etc. Calculations used 0.002 for the background factor. The second group of factors are listed in Appendix I, page 9, column (B-D)/(A-G). These are factors used with the SPL charge to determine fluoride out from the SPL.

On 3/17/88, AR submitted the February charge rates. Attached to this was a twelve month running average of fluoride emissions as required by the permit. What is not specified in the permit is what "factors" will be used. On this report, factors were used as follows:

No SPL Fl. x .02; SPL Fl. x .008 - until September 1987
No SPL Fl. x .002; SPL Fl. x .004 - begin October 1987

The twelve month running average of 12,242 lb is less than the 19,682 lb/yr allowable. The two different factors were used since prior to October 1987 a significantly different mix was used. The limestone going into the cupola may have a significant impact on reducing fluoride emissions. For this reason, factors from both the April 1986 and January 1988 tests were used.

In Appendix B attached, I have also run the running average for fluoride emissions using three different scenarios:

- (1) using the .02/.008 factor through September 1987
using the .002/.004 factor beginning October 1987
(results on page 8)
- (2) using the .002/.004 factors for all months
(results on page 16)
- (3) using the 0.02/0.012^{*} factor through September 1987
using the 0.002/0.004 factor beginning October 1987
(results on page 18)

^{*}The 0.012 factor comes from April 1986, Massena (M)
SPL testing (see Appendix I, page 9)

Numbers (1) and (2) above would show compliance with the running average while scenario (3) would not.

For compliance and future permitting, in addition to any future testing, the factors to be used must now be determined, in addition to any future testing. It has recently been observed that four of their five mixes do not have limestone in them (see Appendix I, page 4). The worst case of these would most likely require testing either prior to or after any permit revision.

I am also attaching a recent submittal sent by the consultant. It contains the % fluoride of the materials input during testing. The SPL % fluoride is somewhat higher than the initial permit value (12.6% vs. 10.75%). There are some other differences (both positive and negative) which total approximately the same % fluoride.

In summary, we need to determine which "factors" will be used to determine compliance for past, present, and future running averages, and what future testing, material sampling, etc. will be required. Please let me know if you would like to get together and discuss this prior to forwarding your comments. As this is still scheduled to go to hearing unless the running average can be resolved, we will need to let OLA know our decision soon.

KS/jf

att.

APPENDIX A.

AMERICAN ROCKWOOL, NASH COUNTY 880404

II

(BOTH CUPOLAS)

TEST DATE	#SO2/HR	LB SO2/ MMBTU	PART. W/BACK- HALF	H2S	OPACITY	FLUORIDE
--------------	---------	------------------	--------------------------	-----	---------	----------

APRIL 86

(BKGRD)						1.6
(SPL,B)						2.45
(SPL,M)						3.6

JAN 88

(BKGRD)	148		1.5			0.27
(SPL350)	120		1.8			0.53
(SPL700)	110		1.7			0.87

FEB 5, 88

(BACKGRD)	144.1	1.1	3		5	0.023
(710 SPL)	111	0.9	4		5	0.86

TEXAS

(BKGRD)						0.6
(SPL1000)						1.4
(SPL1500)						1.4

PERMIT	100	2.3	10	12	20	6
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(as per
20.0515)

AMERICAN ROCKWOOL, NASH COUNTY 880404

III

#'S/CHARGE PER CUPOLA ('S)

TEST DATE	COKE	SPL	DUQ SLAG	TRAP ROCK	LIME	STEEL	TENN SLAG
-----	-----	-----	-----	-----	-----	-----	-----
APRIL 86							
(BKGRD)							
(SPL,B)							
(SPL,M)							
JAN 88							
(BKGRD)	398	0	1278	1090	49	394	396
(SPL350)	226	100	1270	1088	50	396	393
(SPL700)	259	215	1119	1006	44	371	377
FEB 5,88							
(BACKGRD)	400	0	1300	1100	50	400	400
(710 SPL)	260	225	1300	1100	50	400	400
TEXAS							
(BKGRD)	400	0	0	1050	0	0	400
(SPL1000)	278						
(SPL1500)	417						

CHARGES/
HOUR

3.3
3.3
3.3

3.37
3.37
3.37

3.6
3.6

3.3
3.3
3.3

AMERICAN ROCKWOOL, NASH COUNTY 880404
IV
CHARGE/CUPOLA (#/HR)

TEST DATE	COKE	SPL	DUQ SLAG	TRAP ROCK	LIME	STEEL	TENN SLAG

APRIL 86							
(BKGRD)	1164	0	4239	1668	83	200	2000
(SPL,B)	683	1020	0	1668	83	200	2000
(SPL,M)	683	1014	0	1668	83	200	2000
JAN 88							
(BKGRD)	1341.26	0	4306.86	3673.3	165.13	1327.78	1334.52
(SPL350)	761.62	337	4279.9	3666.56	168.5	1334.52	1324.41
(SPL700)	872.83	724.55	3771.03	3390.22	148.28	1250.27	1270.49
FEB 5, 88							
(BACKGRD)	1440	0	4680	3960	180	1440	1440
(710 SPL)	936	810	4680	3960	180	1440	1440
TEXAS							
(BKGRD)	1320	0	0	3465	0	0	1320
(SPL1000)	917.4	0	0	0	0	0	0
(SPL1500)	1376.1	0	0	0	0	0	0
	COKE	SPL	DUQ SLAG	TRAP ROCK	LIME	STEEL	TENN SLAG
	-----	-----	-----	-----	-----	-----	-----
PERMIT	1300	710	5500	2500	250	3000	3000
% FL	0	11.3	0.45	0	0.41	0	2.4
NOTE: THE MASSENA FL.=10.2%, BADIN=11.3%							
ACTUAL BLENDS;							
STANDARD ROCKWOOL							
W/O	1285	0	4329	3663	167	1332	1332
W/SPL	866	700	4329	3663	167	1332	1332
WHITE ROCKWOOL							
W/O	1285	0	7326	0	0	0	1665
W/SPL	866	700	7326	0	0	0	1665
FIBER-GRO ROCKWOOL							
W/O	1285	0	1000	4995	0	3330	

53

W/SPL	866	700	1000	4995	0	3330	
DARK ROCKWOOL							
W/O	1285	0	0	4329	0	3330	666
W/SPL	866	700	0	4329	0	3330	666
CRYOGENIC ROCKWOOL							
W/O	1285	0	0	3164	0	5162	1665
W/SPL	866	700	0	3164	0	5162	1665

FELDSPAR (NR)	BOF SLAG
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0	0
---	---

1000	
1000	

	1000
--	------

	1000
--	------

	1998
	1998

AMERICAN ROCKWOOL, NASH COUNTY 880404

V
'S OF FL INPUT/CUPOLA

TEST DATE	COKE	SPL	DUQ SLAG	TRAP ROCK	LIME	STEEL	TENN SLAG

APRIL 86							
(BKGRD)	0.0	0.0	19.1	0.0	0.3	0.0	48.0
(SPL,B)	0.0	115.3	0.0	0.0	0.3	0.0	48.0
(SPL,M)	0.0	114.6	0.0	0.0	0.3	0.0	48.0
JAN 88							
(BKGRD)	0.0	0.0	19.4	0.0	0.7	0.0	32.0
(SPL350)	0.0	38.1	19.3	0.0	0.7	0.0	31.8
(SPL700)	0.0	81.9	17.0	0.0	0.6	0.0	30.5
FEB 5, 88							
(BACKGRD)	0.0	0.0	21.1	0.0	0.7	0.0	34.6
(710 SPL)	0.0	91.5	21.1	0.0	0.7	0.0	34.6
TEXAS							
(BKGRD)							
(SPL1000)							
(SPL1500)							
	COKE	SPL	DUQ SLAG	TRAP ROCK	LIME	STEEL	TENN SLAG
	-----	-----	-----	-----	-----	-----	-----
PERMIT	1300.0	710.0	5500.0	2500.0	250.0	3000.0	3000.0
% FL	0.0	11.3	0.5	0.0	0.4	0.0	2.4
ACTUAL BLENDS;							
STANDARD ROCKWOOL							
W/O	0.0	0.0	19.5	0.0	0.7	0.0	32.0
W/SPL	0.0	79.1	19.5	0.0	0.7	0.0	32.0
WHITE ROC	0.0	0.0	0.0	0.0	0.0	0.0	0.0
W/O	0.0	0.0	33.0	0.0	0.0	0.0	40.0
W/SPL	0.0	79.1	33.0	0.0	0.0	0.0	40.0
FIBER-GRO	0.0	0.0	0.0	0.0	0.0	0.0	0.0
W/O	0.0	0.0	4.5	0.0	0.0	0.0	0.0

W/SPL	0.0	79.1	4.5	0.0	0.0	0.0	0.0
DARK ROCK	0.0	0.0	0.0	0.0	0.0	0.0	0.0
W/O	0.0	0.0	0.0	0.0	0.0	0.0	16.0
W/SPL	0.0	79.1	0.0	0.0	0.0	0.0	16.0
CRYOGENIC	0.0	0.0	0.0	0.0	0.0	0.0	0.0

7.

AMERICAN ROCKWOOL, NASH COUNTY 880404
VI
TOTALS (BOTH CUPOLAS AT SPRING HOPE)

TEST DATE	FLUORIDE IN (#,S)	FLUORIDE OUT #'S	FL OUT/ FL IN

APRIL 86			
(BKGRD)	134.8	3.2	0.02373
(SPL,B)	327.2	0.6	0.00183
(SPL,M)	325.8	1.1	0.00338
JAN 88			
(BKGRD)	104.2	0.3	0.00262
(SPL350)	179.6	0.5	0.00295
(SPL700)	259.9	0.9	0.00335
FEB 5,88			
(BACKGRD)	112.7	0.2	0.00204
(710 SPL)	295.8	0.9	0.00291
TEXAS			
(BKGRD)	45.0	0.6	0.01333
(SPL1000)	145.0	1.4	0.00966
(SPL1500)	195.0	1.4	0.00718

FLUORIDE FLUORIDE
IN (#,S) OUT #'S

PERMIT
% FL

ACTUAL BLENDS;
STANDARD ROCKWOOL
W/O 104.3
W/SPL 262.5
WHITE ROCKWOOL
W/O 145.9
W/SPL 304.1
FIBER-GRO ROCKWOOL
W/O 9.0

W/SPL 167.2
DARK ROCKWOOL
W/O 32.0
W/SPL 190.2
CRYOGENIC ROCKWOOL
W/O 79.9

AMERICAN ROCKWOOL, NASH COUNTY 880413

VIII

PER CUPOLA BASIS/HOUR

FL IN IS FOR BOTH CUPOLAS (JAN & FEB)

TEST DATE	A FLUORIDE IN (#,S)	B FL OUT (#'S)	B/A FL OUT FL IN	D FL OUT BKGRD ONLY	E SPL FL IN (#'S)	(B-D)/(A-E) FL OUT/ IN -BKGRD	(B-D)/A FL OUT/ IN TOTAL
APRIL 86							
(BKGRD)	67.4	1.60	0.0237	1.6	0.0	0.0000	0.0000
(SPL,B)	163.6	2.45	0.0150	1.6	115.3	0.0176	0.0052
(SPL,M)	162.9	3.60	0.0221	1.6	114.6	0.0414	0.0123
JAN 88							
(BKGRD)	104.2	0.27	0.0026	0.27	0.0	0.0000	0.0000
(SPL350)	179.6	0.53	0.0030	0.27	38.1	0.0018	0.0014
(SPL700)	259.9	0.87	0.0033	0.27	81.9	0.0034	0.0023
FEB 5, 88							
(BACKGRD)	72.1*	0.23	0.0032	0.23	0.0	0.0000	0.0000
(710 SPL)	274.0*	0.86	0.0031	0.23	91.5	0.0035	0.0023
TEXAS	as given from American Rockwool						
(BKGRD)	45.0	0.60	0.0133	0.6	0.0	0.0000	0.0000
(SPL1000)	145.0	1.40	0.0097	0.6	0.0	0.0055	0.0055
(SPL1500)	195.0	1.40	0.0072	0.6	0.0	0.0041	0.0041

value used in permit application

AMERICAN ROCKWOOL, NASH COUNTY 880413
IX
PER HOUR BASIS FL IN IS FOR BOTH CUPOLAS (JAN&FEB)

TEST DATE	A FLUORIDE IN (#,S)	B FL OUT (#'S)	D FL OUT BKGRD ONLY	F #OF SPL CHARGED	(B-D)/F FL INCR. FROM SPL #/HR/#SPL	G BKGRD CHARGE OF FL #/HR	(B-D)/(A-G) EMISSION FACTOR
APRIL 86							
(BKGRD)	67.4	1.60	1.6				
(SPL,B)	163.6	2.45	1.6	1020.0	0.0008		0.0052
(SPL,M)	162.9	3.60	1.6	1014.0	0.0020		0.0123
JAN 88							
(BKGRD)	104.2	0.27	0.27			104	
(SPL350)	179.6	0.53	0.27	674.0	0.0004	104	0.0034
(SPL700)	259.9	0.87	0.27	1450.0	0.0004	104	0.0038
FEB 5,88							
(BACKGRD)	72.1	0.23	0.23			72	
(710 SPL)	274.0	0.86	0.23	1620.0	0.0004	72	0.0031
TEXAS							
(BKGRD)	45.0	0.60	0.6			45	
(SPL1000)	145.0	1.40	0.6	1000.0	0.0008	45	0.0080
(SPL1500)	195.0	1.40	0.6	1500.0	0.0005	45	0.0052

APPENDIX B

AMERICAN ROCKWOOL

FLOURIDES: .02/.008 THOUGH SEPT 87

D880415

.002/.004 BEGIN OCT 87

MONTH/YR:AUGUST 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2948700	0.0045	13269.2	0.0200	265.38	
TENN SLAG	995750	0.0240	23898.0	0.0200	477.96	
LIME	53699	0.0041	220.2	0.0200	4.40	
SPL	6355	0.1075	683.2	0.0080		5.47

TOTAL FL/MONTH IN = 38070.5

TOTAL FL/MONTH OUT= 753.21

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:SEPTEMBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4275200	0.0045	19238.4	0.0200	384.77	
TENN SLAG	1553075	0.0240	37273.8	0.0200	745.48	
LIME	36336	0.0041	149.0	0.0200	2.98	
SPL	535300	0.1075	57544.8	0.0080		460.36

TOTAL FL/MONTH IN = 114205.9

TOTAL FL/MONTH OUT= 1593.58

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:OCTOBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3357900	0.0045	15110.6	0.0200	302.21	
TENN SLAG	1194000	0.0240	28656.0	0.0200	573.12	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	560850	0.1075	60291.4	0.0080		482.33

TOTAL FL/MONTH IN = 104057.9

TOTAL FL/MONTH OUT= 1357.66

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:NOVEMBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3345700	0.0045	15055.7	0.0200	301.11	
TENN SLAG	1191500	0.0240	28596.0	0.0200	571.92	
LIME	300	0.0041	1.2	0.0200	0.02	
SPL	698120	0.1075	75047.9	0.0080		600.38

TOTAL FL/MONTH IN = 118700.8

TOTAL FL/MONTH OUT= 1473.44

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:DECEMBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2460000	0.0045	11070.0	0.0200	221.40	
TENN SLAG	809150	0.0240	19419.6	0.0200	388.39	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	626825	0.1075	67383.7	0.0080		539.07

TOTAL FL/MONTH IN = 97873.3

TOTAL FL/MONTH OUT= 1148.86

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JANUARY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	1824900	0.0045	8212.1	0.0200	164.24	
TENN SLAG	618250	0.0240	14838.0	0.0200	296.76	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	579275	0.1075	62272.1	0.0080		498.18

TOTAL FL/MONTH IN = 85322.1

TOTAL FL/MONTH OUT= 959.18

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:FEBRUARY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2759050	0.0045	12415.7	0.0200	248.31	
TENN SLAG	980500	0.0240	23532.0	0.0200	470.64	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	573300	0.1075	61629.8	0.0080		493.04

TOTAL FL/MONTH IN = 97577.5

TOTAL FL/MONTH OUT= 1211.99

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:MARCH 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2979550	0.0045	13408.0	0.0200	268.16	
TENN SLAG	1048900	0.0240	25173.6	0.0200	503.47	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	612700	0.1075	65865.3	0.0080		526.92

TOTAL FL/MONTH IN = 104446.8

TOTAL FL/MONTH OUT= 1298.55

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:APRIL 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3270850	0.0045	14718.8	0.0200	294.38	
TENN SLAG	1155100	0.0240	27722.4	0.0200	554.45	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	689725	0.1075	74145.4	0.0080		593.16

TOTAL FL/MONTH IN = 116586.7

TOTAL FL/MONTH OUT= 1441.99

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:MAY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2907000	0.0045	13081.5	0.0200	261.63	
TENN SLAG	1219800	0.0240	29275.2	0.0200	585.50	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	709100	0.1075	76228.3	0.0080		609.83

TOTAL FL/MONTH IN = 118585.0

TOTAL FL/MONTH OUT= 1456.96

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JUNE 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3692850	0.0045	16617.8	0.0200	332.36	
TENN SLAG	1654300	0.0240	39703.2	0.0200	794.06	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	1182100	0.1075	127075.8	0.0080		1016.61

TOTAL FL/MONTH IN = 183396.8

TOTAL FL/MONTH OUT= 2143.03

HRS/MONTH :

#FL/HR = ERR

MONTH/YR: JULY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4028200	0.0045	18126.9	0.0200	362.54	
TENN SLAG	1761300	0.0240	42271.2	0.0200	845.42	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	1321800	0.1075	142093.5	0.0080		1136.75

TOTAL FL/MONTH IN = 202491.6

TOTAL FL/MONTH OUT= 2344.71

HRS/MONTH :

#FL/HR = ERR

MONTH/YR: AUGUST 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3816100	0.0045	17172.5	0.0200	343.45	
TENN SLAG	1748400	0.0240	41961.6	0.0200	839.23	
LIME	0	0.0041	0.0	0.0200	0.00	
SPL	647275	0.1075	69582.1	0.0080		556.66

TOTAL FL/MONTH IN = 128716.1

TOTAL FL/MONTH OUT= 1739.34

HRS/MONTH :

#FL/HR = ERR

MONTH/YR: SEPTEMBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4142600	0.0045	18641.7	0.0200	372.83	
TENN SLAG	1938600	0.0240	46526.4	0.0200	930.53	
LIME	22650	0.0041	92.9	0.0200	1.86	
SPL	0	0.1075	0.0	0.0080		0.00

TOTAL FL/MONTH IN = 65261.0

TOTAL FL/MONTH OUT= 1305.22

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:OCTOBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4445475	0.0045	20004.6	0.0020	40.01	
TENN SLAG	2147000	0.0240	51528.0	0.0020	103.06	
LIME	73637	0.0041	301.9	0.0020	0.60	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 71834.5

TOTAL FL/MONTH OUT= 143.67

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:NOVEMBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3313650	0.0045	14911.4	0.0020	29.82	
TENN SLAG	1385550	0.0240	33253.2	0.0020	66.51	
LIME	61022	0.0041	250.2	0.0020	0.50	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 48414.8

TOTAL FL/MONTH OUT= 96.83

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:DECEMBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3147350	0.0045	14163.1	0.0020	28.33	
TENN SLAG	1071550	0.0240	25717.2	0.0020	51.43	
LIME	62062	0.0041	254.5	0.0020	0.51	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 40134.7

TOTAL FL/MONTH OUT= 80.27

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JANUARY 88

6

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	1194879	0.0045	5377.0	0.0020	10.75	
TENN SLAG	331177	0.0240	7948.2	0.0020	15.90	
LIME	35359	0.0041	145.0	0.0020	0.29	
SPL	77410	0.1075	8321.6	0.0040		33.29

TOTAL FL/MONTH IN = 21791.8

TOTAL FL/MONTH OUT= 60.23

HRS/MONTH :

#FL/HR = ERR

* MONTH/YR:

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG		0.0045	0.0		0.00	
TENN SLAG		0.0240	0.0		0.00	
LIME		0.0041	0.0		0.00	
SPL		0.1075	0.0			0.00

TOTAL FL/MONTH IN = 0.0

TOTAL FL/MONTH OUT= 0.00

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG		0.0045	0.0		0.00	
TENN SLAG		0.0240	0.0		0.00	
LIME		0.0041	0.0		0.00	
SPL		0.1075	0.0			0.00

TOTAL FL/MONTH IN = 0.0

TOTAL FL/MONTH OUT= 0.00

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG		0.0045	0.0		0.00	
TENN SLAG		0.0240	0.0		0.00	
LIME		0.0041	0.0		0.00	
SPL		0.1075	0.0			0.00

'2 YEAR HISTORY (OCT 1984-SEP 1986):

140.6#/HR*423HR/YR*12MONTHS/YR=713,686#FL IN/YR

*(.02)=14273#FL IN/YR +5095 =19368 ALLOWABLE #FL OUT/YR

*(.002)=1427#FL IN/YR +5095 =6522 ALLOWABLE #FL OUT/YR

(.002/.008) / (.002/.004) SCENARIO

MONTH	FL IN/ MONTH	FL OUT/ MONTH	CUMULATIVE #FL/YEAR
AUG86	38070.5		
		753.21	
SEPT86	114205.9		
		1593.58	
OCT 86	104057.9		
		1357.66	
NOV86	118700.8		
		1473.44	
DEC86	97873.3		
		1148.86	
JAN87	85322.1		
		959.18	
FEB87	97577.5		
		1211.99	
MAR87	104446.8		
		1298.55	
APRIL87	116586.7		
		1441.99	
MAY87	118585.0		
		1456.96	
JUNE87	183396.8		
		2143.03	
JULY87	202491.6		
		2344.71	
			17183
AUG87	128716.1		
		1739.34	
			18169.3
SEPT87	65261.0		
		1305.22	
			17880.9
OCT87	71834.5		
		143.67	
			16666.9
NOV87	48414.8		

DEC87

40134.7

80.27

13290.3

14221.7

JAN88

21791.8

60.23

13322.8

AMERICAN ROCKWOOL

FLOURIDES: .002/.004 ALL MONTHS***

D880415

MONTH/YR:AUGUST 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2948700	0.0045	13269.2	0.0020	26.54	
TENN SLAG	995750	0.0240	23898.0	0.0020	47.80	
LIME	53699	0.0041	220.2	0.0020	0.44	
SPL	6355	0.1075	683.2	0.0040		2.73

TOTAL FL/MONTH IN = 38070.5

TOTAL FL/MONTH OUT= 77.51

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:SEPTEMBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4275200	0.0045	19238.4	0.0020	38.48	
TENN SLAG	1553075	0.0240	37273.8	0.0020	74.55	
LIME	36336	0.0041	149.0	0.0020	0.30	
SPL	535300	0.1075	57544.8	0.0040		230.18

TOTAL FL/MONTH IN = 114205.9

TOTAL FL/MONTH OUT= 343.50

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:OCTOBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3357900	0.0045	15110.6	0.0020	30.22	
TENN SLAG	1194000	0.0240	28656.0	0.0020	57.31	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	560850	0.1075	60291.4	0.0040		241.17

TOTAL FL/MONTH IN = 104057.9

TOTAL FL/MONTH OUT= 328.70

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:NOVEMBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3345700	0.0045	15055.7	0.0020	30.11	
TENN SLAG	1191500	0.0240	28596.0	0.0020	57.19	
LIME	300	0.0041	1.2	0.0020	0.00	
SPL	698120	0.1075	75047.9	0.0040		300.19

TOTAL FL/MONTH IN = 118700.8

TOTAL FL/MONTH OUT= 387.50

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:DECEMBER 86

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2460000	0.0045	11070.0	0.0020	22.14	
TENN SLAG	809150	0.0240	19419.6	0.0020	38.84	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	626825	0.1075	67383.7	0.0040		269.53

TOTAL FL/MONTH IN = 97873.3

TOTAL FL/MONTH OUT= 330.51

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JANUARY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	1824900	0.0045	8212.1	0.0020	16.42	
TENN SLAG	618250	0.0240	14838.0	0.0020	29.68	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	579275	0.1075	62272.1	0.0040		249.09

TOTAL FL/MONTH IN = 85322.1

TOTAL FL/MONTH OUT= 295.19

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:FEBRUARY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2759050	0.0045	12415.7	0.0020	24.83	
TENN SLAG	980500	0.0240	23532.0	0.0020	47.06	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	573300	0.1075	61629.8	0.0040		246.52

TOTAL FL/MONTH IN = 97577.5

TOTAL FL/MONTH OUT= 318.41

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:MARCH 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2979550	0.0045	13408.0	0.0020	26.82	
TENN SLAG	1048900	0.0240	25173.6	0.0020	50.35	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	612700	0.1075	65865.3	0.0040		263.46

TOTAL FL/MONTH IN = 104446.8

TOTAL FL/MONTH OUT= 340.62

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:APRIL 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3270850	0.0045	14718.8	0.0020	29.44	
TENN SLAG	1155100	0.0240	27722.4	0.0020	55.44	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	689725	0.1075	74145.4	0.0040		296.58

TOTAL FL/MONTH IN = 116586.7

TOTAL FL/MONTH OUT= 381.46

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:MAY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	2907000	0.0045	13081.5	0.0020	26.16	
TENN SLAG	1219800	0.0240	29275.2	0.0020	58.55	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	709100	0.1075	76228.3	0.0040		304.91

TOTAL FL/MONTH IN = 118585.0

TOTAL FL/MONTH OUT= 389.63

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JUNE 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3692850	0.0045	16617.8	0.0020	33.24	
TENN SLAG	1654300	0.0240	39703.2	0.0020	79.41	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	1182100	0.1075	127075.8	0.0040		508.30

TOTAL FL/MONTH IN = 183396.8

TOTAL FL/MONTH OUT= 620.95

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JULY 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4028200	0.0045	18126.9	0.0020	36.25	
TENN SLAG	1761300	0.0240	42271.2	0.0020	84.54	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	1321800	0.1075	142093.5	0.0040		568.37

TOTAL FL/MONTH IN = 202491.6

TOTAL FL/MONTH OUT= 689.17

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:AUGUST 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3816100	0.0045	17172.5	0.0020	34.34	
TENN SLAG	1748400	0.0240	41961.6	0.0020	83.92	
LIME	0	0.0041	0.0	0.0020	0.00	
SPL	647275	0.1075	69582.1	0.0040		278.33

TOTAL FL/MONTH IN = 128716.1

TOTAL FL/MONTH OUT= 396.60

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:SEPTEMBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4142600	0.0045	18641.7	0.0020	37.28	
TENN SLAG	1938600	0.0240	46526.4	0.0020	93.05	
LIME	22650	0.0041	92.9	0.0020	0.19	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 65261.0

TOTAL FL/MONTH OUT= 130.52

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:OCTOBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	4445475	0.0045	20004.6	0.0020	40.01	
TENN SLAG	2147000	0.0240	51528.0	0.0020	103.06	
LIME	73637	0.0041	301.9	0.0020	0.60	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 71834.5

TOTAL FL/MONTH OUT= 143.67

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:NOVEMBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3313650	0.0045	14911.4	0.0020	29.82	
TENN SLAG	1385550	0.0240	33253.2	0.0020	66.51	
LIME	61022	0.0041	250.2	0.0020	0.50	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 48414.8

TOTAL FL/MONTH OUT= 96.83

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:DECEMBER 87

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	3147350	0.0045	14163.1	0.0020	28.33	
TENN SLAG	1071550	0.0240	25717.2	0.0020	51.43	
LIME	62062	0.0041	254.5	0.0020	0.51	
SPL	0	0.1075	0.0	0.0040		0.00

TOTAL FL/MONTH IN = 40134.7

TOTAL FL/MONTH OUT= 80.27

HRS/MONTH :

#FL/HR = ERR

MONTH/YR:JANUARY 88

MATERIAL TYPE	AMOUNT USED(#)	FLUORIDE %*.01	FL IN(#)	FACTOR	BURDEN #FL/MONTH	SPL #FL/MONTH
DUQ. SLAG	1194879	0.0045	5377.0	0.0020	10.75	
TENN SLAG	331177	0.0240	7948.2	0.0020	15.90	
LIME	35359	0.0041	145.0	0.0020	0.29	
SPL	77410	0.1075	8321.6	0.0040		33.29

TOTAL FL/MONTH IN = 21791.8

TOTAL FL/MONTH OUT= 60.23

HRS/MONTH :

#FL/HR = ERR

II AMERICAN ROCKWOOL ANNUAL FLUORIDE TOTALS
(.002/.004) SCENARIO

MONTH	FL IN/ MONTH	FL OUT/ MONTH	CUMULATIVE #FL/YEAR
AUG86	38070.5		
		77.51	
SEPT86	114205.9		
		343.50	
OCT 86	104057.9		
		328.70	
NOV86	118700.8		
		387.50	
DEC86	97873.3		
		330.51	
JAN87	85322.1		
		295.19	
FEB87	97577.5		
		318.41	
MAR87	104446.8		
		340.62	
APRIL87	116586.7		
		381.46	
MAY87	118585.0		
		389.63	
JUNE87	183396.8		
		620.95	
JULY87	202491.6		
		689.17	4503.2
AUG87	128716.1		
		396.60	4822.2
SEPT87	65261.0		
		130.52	4609.3
OCT87	71834.5		
		143.67	4424.2
NOV87			

	48414.8	96.83	4133.6
DEC87			
	40134.7	80.27	3883.3
JAN88			
	21791.8	60.23	3648.4

* II AMERICAN ROCKWOOL ANNUAL FLUORIDE TOTALS
 .02/.012 THROUGH SEPT 87
 .002/.004 BEGIN OCT 87
 ALLOWABLE IS 19368 #FL OUT/YR

MONTH	FL IN/ MONTH	FL OUT/ MONTH	CUMULATIVE #FL/YEAR
AUG86	38070.5		
		755.94	
SEPT86	114205.9		
		1823.76	
OCT 86	104057.9		
		1598.83	
NOV86	118700.8		
		1773.63	
DEC86	97873.3		
		1418.40	
JAN87	85322.1		
		1208.27	
FEB87	97577.5		
		1458.51	
MAR87	104446.8		
		1562.01	
APRIL87	116586.7		
		1738.57	
MAY87	118585.0		
		1761.87	
JUNE87	183396.8		
		2651.33	
JULY87	202491.6		
		2913.08	
			20684.2
AUG87	128716.1		
		2017.67	
			21925.9
SEPT87	65261.0		
		1305.22	
			21407.4
OCT87	71834.5		
		143.67	

			10052.25
NOV87	48414.8	96.83	18275.4
DEC87	40134.7	80.27	16937.3
JAN88	21791.8	60.23	15789.3

RECEIVED

APR 13 1988

RALEIGH REGIONAL OFFICE

RT. 4 BOX 283
CRESTHILL DR.
LOUISVILLE, TN 37777
615/970-2313

April 12, 1988

Mr. Ken Schuster, P.E.
N.C. Dept. of Natural Resources
3800 Barrett Drive
P.O. Box 27687
Raleigh, N.C. 27611-7687



**DENNIS WEETER
ASSOCIATES**

CONSULTING ENGINEERS

Re: American Rockwool; Spring Hope, NC; Solids/Raw
Material/Product Flouride Analyses

Dear Ken:

Please find enclosed a communication from Mr. Apicella which summarizes the flouride analyses for the Feb. 1988 testing.

Based upon a communication to you dated 2/27/88, I would like to point out a few differences. The background F input in Feb. was 71-72.1 #/hr not 119.3 (derived from using April '86 % F numbers and the Feb. '88 burden mass loadings.) The effect is to change the JAN/FEB '88 "NO SPL" emission factors to 0.0030-0.0038. Note the calculated F input was 264.7#/hr when actual was 274#/hr for the maximum SPL burn. This would reduce emission factors to 0.00288 for the "SPL" burn. When we shortly submit a permit modification request, I will present detailed calculations on these emission factors.

With regard to a mass balance on F, closure was between 8.5-13.1%. The error is probably in the baghouse catch and cupola drop since methods are not available to accurately measure mass.

Note also that Duquesne and Tennessee slags measured lower F concentrations whereas SPL was slightly higher. Trap, lime, steel slag, and coke have traces of F. The analysis in Feb '88 was to the 1/100 of a percent. For future monthly reports we will use these Feb '88 F concentrations.

Sincerely,

Dennis W. Weeter, Ph.D., P.E.

Enclosure

cc: Bart Bromley; Ron Small; Lee Blayden; Jim Apicella

cc: BOB WOOTE
4.18.88

ALUMINUM COMPANY OF AMERICA

ALCOA TECHNICAL CENTER

ALCOA CENTER, PA. 15069

(412) 339-6651



1988 April 8

Dr. Dennis Weeter
Dennis Weeter Associates
Route 4, Box 283
Louisville, TN 37777

RE: FLUORIDE BALANCE AT AMERICAN ROCKWOOL'S SPRING HOPE, NC FACILITY

Dear Dr. Weeter:

Environmental Control Laboratory personnel with assistance from American Rockwool personnel conducted a fluoride material balance on the No. 1 cupola and No. 2 cupola in 1988 February. The purpose of this material balance was to show where the deposition of fluorides introduced into the system occurred and the relative accuracy (closure) of these measurements. The fluoride distribution results are shown in Table 1 with the balances of 1988 February 3 (Background Tests 1 and 2), 1988 February 4 (Background Test 3) and 1988 February 5 (710 lb/hr SPL per cupola-Tests 4 through 7) having closures of 110.7%, 113.1% and 108.5%, respectively (Table 2).

PROCEDURES AND CALCULATIONS

The mass flow rate of solids for the raw materials, products and sparkbox/baghouse catches was determined by obtaining the total process weight of each item per unit time. For the raw materials, each component of every charge was weighed and recorded with the actual charging time. Samples of all raw materials were obtained and prepared for analysis as detailed in Report 93-88-003. The mineral wool was collected in 29 lb bags and production was tracked by the number of bags produced per hour and one bag was randomly selected from each operating condition for analysis. The sparkbox/baghouse catch was collected in tared 55 gallon drums over specific time periods then weighed. Random 2 lb samples from every barrel were analyzed for each operating condition. The amount of particulate air emissions was determined by EPA Method 5 and the amount of particulate and gaseous fluorides was determined by Alcoa Method 4075-A. The cupola drop could not be directly measured because there was no practical solution to the problem of collecting the molten cupola drop material. Cupola drop was estimated by subtracting solids out from solids in and then subtracting the amount of carbon burned. A random five gallon sample was obtained for each operating condition.

Fluoride analyses on all the collected materials were completed by the Analytical Chemistry Division of Alcoa using Method 4076A.

DR. DENNIS WEETER

1988-04-08

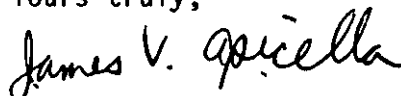
Page 2

DISCUSSION OF RESULTS

While the emission test runs were one hour in duration, the other materials were collected over longer periods of time during "steady state" operation of the system in order to obtain better representative weights and unbiased composite samples for analysis.

The data in Table 2 show the closures to range from 108.5% to 113.1%. The collection and analysis of the air emissions, product and raw materials were controlled to the extent they would not cause this deviation from 100%. The sparkbox/baghouse catch was determined by collecting what was discharged from that system. This will not give the exact sparkbox/baghouse weight values but has the best accuracy possible without shutting down the process and cleaning out each baghouse between trials. Even so, the sparkbox/baghouse should not see a total weight of solids much more/less than reported. Therefore, the majority of fluoride "overcollection" must come from the cupola drop. The cupola drop mass flow is not a measured quantity, but rather a calculated estimate that comes from a basic material balance that cannot account for losses such as unusable product and shot.

Yours truly,



James V. Apicella
Environmental Scientist
Environmental Control Laboratory

TABLE 1

Fluoride Distribution

	Background 1&2 88-02-03			Background 3 88-02-04			710 SPL 88-02-05		
	Solids	%F	lb F/hr	Solids	%F	lb F/hr	Solids	%F	lb F/hr
Raw Materials	26267	-- ^{1a}	72.1	25527	-- ^{1b}	71.0	27308	-- ^{1c}	274.0
Products	14616	0.3	43.8	15157	0.3	45.5	13322	1.1	146.5
SP Box and Baghouse ²	294.7	--	8.9	383.3	--	11.8	307.5	--	30.8
Emissions ³	2.2	--	0.23	2.2	--	0.23	3.7	--	0.86
Cupola Drop	8963 ⁴	0.3	26.9	7602 ⁴	0.3	22.8	11125 ⁴	1.07	119.0

1a See breakdown of raw materials Table R1

1b See breakdown of raw materials Table R2

1c See breakdown of raw materials Table R3

2 See breakdown of spark box and baghouse Table R4

3 See ECL Report 93-88-003, Compliance Report on Particulate and Sulfur Dioxide, Fluoride and Visual Emissions from Mineral Wool Production, American Rockwool, Inc., Spring Hope, NC, Test Period: 1988 February 3-5

4 Estimate from carbon balance

TABLE 2
Fluoride Balance

	<u>Background 1&2</u> <u>88-02-03</u>	<u>Background 3</u> <u>88-02-04</u>	<u>710 SPL</u> <u>88-02-05</u>
Total Fluoride in lb/hr	72.1	71.0	274.0
Total Fluoride out lb/hr	79.8	80.3	297.2
% Closure	110.7	113.1	108.5

TABLE R1

SPRING HOPE PRODUCTION DATA 88-02-03

TIME		10:30AM - 5:30PM BACKGROUND TESTS				
		UNIT 1	UNIT 2	TOTAL		
PRODUCT	BAGS	1764	1764		PRODUCT	LB/HR
	POUNDS	51156	51156		0.3 %F	43.8
	TIME (hr)	7	7			
	LB/HR	7308	7308	14616		
COKE	POUNDS	9464	10301		COKE	
	TIME (hr)	7.02	7.00		0.04 %F	1.1
	LB/HR	1348	1472	2820		
SPL	POUNDS	0	0		SPL	
	TIME (hr)	7.02	7.00		12.6 %F	0.0
	LB/HR	0	0	0		
DUQUESNE	POUNDS	31201	34575		DUQUESNE	
	TIME (hr)	7.02	7.00		0.14 %F	13.1
	LB/HR	4445	4939	9384		
TRAP	POUNDS	26338	29188		TRAP	
	TIME (hr)	7.02	7.00		0.05 %F	4.0
	LB/HR	3752	4170	7922		
LIME	POUNDS	1162	1317		LIME	
	TIME (hr)	7.02	7.00		0.02 %F	0.1
	LB/HR	166	188	354		
STEEL	POUNDS	9958	10970		STEEL	
	TIME (hr)	7.17	7.22		0.07 %F	2.0
	LB/HR	1389	1519	2908		
TENNESSEE	POUNDS	9669	10515		TENNESSEE	
	TIME (hr)	7.02	7.00		1.8 %F	51.8
	LB/HR	1377	1502	2879		
TOTAL CHARGE LB/HR		12476	13790	26267	FLUORIDES	72.1

TABLE R2

SPRING HOPE PRODUCTION DATA 88-02-04

TIME		8:00AM - 11:00AM		BKGRD	TESTS
		UNIT 1	UNIT 2	TOTAL	
PRODUCT	BAGS	784	784		
	POUNDS	22736	22736		
	TIME (hr)	3	3		
	LB/HR	7579	7579	15157	
COKE	POUNDS	4491	4440		
	TIME (hr)	3.18	3.12		
	LB/HR	1412	1423	2835	
SPL	POUNDS	0	0		
	TIME (hr)	3	3		
	LB/HR	0	0	0	
DUQUESNE	POUNDS	14109	14190		
	TIME (hr)	3.18	3.12		
	LB/HR	4437	4548	8985	
TRAP	POUNDS	12142	12160		
	TIME (hr)	3.18	3.12		
	LB/HR	3818	3897	7716	
LIME	POUNDS	504	499		
	TIME (hr)	3.18	3.12		
	LB/HR	158	160	318	
STEEL	POUNDS	4477	4441		
	TIME (hr)	3.17	3.15		
	LB/HR	1412	1410	2822	
TENNESSEE	POUNDS	4479	4501		
	TIME (hr)	3.18	3.12		
	LB/HR	1408	1443	2851	
=====					
TOTAL CHARGE LB/HR		12646	12881	25527	
PRODUCT				0.3 %F	LB/HR
COKE				0.04 %F	1.1
SPL				12.6 %F	0.0
DUQUESNE				0.14 %F	12.6
TRAP				0.05 %F	3.9
LIME				0.02 %F	0.1
STEEL				0.07 %F	2.0
TENNESSE				1.8 %F	51.3
=====					
FLUORIDES					71.0

TABLE R3

SPRING HOPE PRODUCTION DATA 88-02-05

TIME		8:00AM - 4:00PM 710# SPL				
		UNIT 1	UNIT 2	TOTAL		
PRODUCT	BAGS	1862	1813		PRODUCT	LB/HR
	POUNDS	53998	52577		1.1 %F	146.5
	TIME (hr)	8	8			
	LB/HR	6750	6572	13322		
<hr/>						
COKE	POUNDS	7155	8222		COKE	
	TIME (hr)	8	8		0.04 %F	0.8
	LB/HR	894	1028	1922		
SPL	POUNDS	6056	6715		SPL	
	TIME (hr)	8	8		12.6 %F	201.1
	LB/HR	757	839	1596		
DUQUESNE	POUNDS	36435	39675		DUQUESNE	
	TIME (hr)	8	8		0.14 %F	13.3
	LB/HR	4554	4959	9514		
TRAP	POUNDS	30468	33600		TRAP	
	TIME (hr)	8	8		0.05 %F	4.0
	LB/HR	3809	4200	8009		
LIME	POUNDS	1436	1606		LIME	
	TIME (hr)	8	8		0.02 %F	0.1
	LB/HR	180	201	380		
STEEL	POUNDS	11343	12384		STEEL	
	TIME (hr)	8	8		0.07 %F	2.1
	LB/HR	1418	1548	2966		
TENNESSEE	POUNDS	11142	12224		TENNESSEE	
	TIME (hr)	8	8		1.8 %F	52.6
	LB/HR	1393	1528	2921		
<hr/>						
TOTAL CHARGE LB/HR		13004	14303	27308	FLUORIDES	274.0

TABLE R4

AMERICAN ROCKWOL SPRING HOPE PLANT
SPARKBOX AND BAGHOUSE CATCH

COMPLIANCE TESTS 88-02-03 THRU 88-02-05
SOLIDS AND FLUORIDES

		UNIT 1		UNIT 2		BOTH UNITS		SUMMARY	
BACKGROUND # 1&2	BAGHOUSE	101.0 LB/HR TOTAL	3.5 %F	101.0 LB/HR TOTAL	4.2 %F	202.0 LB/HR TOTAL	7.8 LB/HR F	294.7 LB/HR	8.9 LB/HR F
		3.5 LB/HR F		4.2 LB/HR F					
	SPARK BOX	37.8 LB/HR TOTAL	1.2 %F	54.8 LB/HR TOTAL	1.3 %F	92.7 LB/HR TOTAL	1.2 LB/HR F		
		0.5 LB/HR F		0.7 LB/HR F					
		UNIT 1		UNIT 2		BOTH UNITS		SUMMARY	
BACKGROUND #3	BAGHOUSE	133.5 LB/HR TOTAL	4.2 %F	147.8 LB/HR TOTAL	3.3 %F	281.3 LB/HR TOTAL	10.5 LB/HR F	383.3 LB/HR	11.8 LB/HR F
		5.6 LB/HR F		4.9 LB/HR F					
	SPARK BOX	37.3 LB/HR TOTAL	1.3 %F	64.8 LB/HR TOTAL	1.3 %F	102.0 LB/HR TOTAL	1.3 LB/HR F		
		0.5 LB/HR F		0.8 LB/HR F					
		UNIT 1		UNIT 2		BOTH UNITS		SUMMARY	
710 SPL	BAGHOUSE	117.5 LB/HR TOTAL	11.5 %F	124.3 LB/HR TOTAL	12.8 %F	241.8 LB/HR TOTAL	29.4 LB/HR F	307.5 LB/HR	30.8 LB/HR F
		13.5 LB/HR F		15.9 LB/HR F					
	SPARK BOX	19.8 LB/HR TOTAL	2.1 %F	46.0 LB/HR TOTAL	2.0 %F	65.8 LB/HR TOTAL	1.3 LB/HR F		
		0.4 LB/HR F		0.9 LB/HR F					

EXPRESS

QUESTIONS? CALL 800-238-5355 TOLL FREE

RECEIVED

7161840805

7161840805

AUG 08 1988

CF

500

From (Your Name) Please Print

JAMES Y. APICELLA

Date
8/5/88

Your Phone Number (Very Important)

(412) 337-2975

Company

Department/Floor No.

AVC-C

Street Address

City

State

ZIP Required For Correct Invoicing

City

State

ZIP Required For Correct Invoicing

YOUR BILLING REFERENCE INFORMATION (FIRST 24 CHARACTERS WILL APPEAR ON INVOICE)

83A00070

PAYMENT

☐ Bill Sender

☐ Cash

☐ Bill Recipient's Fedex Acct. No.

☐ Bill 3rd Party Fedex Acct. No.

☐ Bill Credit Card

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

☐ Other

RALEIGH REGIONAL OFFICE

To (Recipient's Name) Please Print

MR. ROBERT MOOTEN

Company

NC DNR

Department/Floor No.

NC DEPT. NATURAL RESOURCE

Exact Street Address (Use of P.O. Boxes or R.F.D. Zip Codes Will Delay Delivery and Result in Extra Charge)

3800 BARRETT DRIVE

City

RALEIGH

State

NC

ZIP Street Address Zip Required

27611

HOLD FOR PICK-UP AT THIS FEDERAL EXPRESS LOCATION

Street Address (See Service Guide or Call 800-238-5355)

3800 BARRETT DRIVE

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RALEIGH

State

NC

ZIP Code of Street Address Required

27611

Emp. No.

27611

Date

8/5/88

City

RALEIGH

State

NC

Recipient's Phone Number (Very Important)

(919) 733-3940

Department/Floor No.

NC DEPT. NATURAL RESOURCE

Exact Street Address (Use of P.O. Boxes or R.F.D. Zip Codes Will Delay Delivery and Result in Extra Charge)

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RALEIGH

State

NC

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ZIP Code of Street Address Required

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Emp. No.

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Date

8/5/88

City

RALEIGH

State

NC

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PRINTED U.S.A. WGSSEL

007



1988 August 4

NC NRDC
Air Quality Section
P.O. Box 27687
Raleigh, NC 27611
Attn: Mr. Robert Wooten

RE: 1988 MAY AMERICAN ROCKWOOL, SPRING HOPE, NC, EMISSION CALCULATIONS

Dear Mr. Wooten:

As per your request, this letter demonstrates how fluoride emissions were calculated from the 1988 May testing at American Rockwools Spring Hope, NC, Facility. Run 1, 1988 May 17, has been chosen for specific examples. All referenced data can be found in the formal report, #93-88-005, dated 1988 May 14, or in the attachment.

Fluoride Emissions - Fluoride emissions were determined using Alcoa Sampling Method 4075A coupled with Alcoa Analytical Method 4076A. The equation used for mass emission of F is as follows:

$$\text{lb/hr F} = \frac{M_F}{V_{mstd}} (Q_{std}) \left[\frac{1 \text{ lb}}{453600 \text{ mg}} \right] \left[\frac{60 \text{ min}}{1 \text{ hr}} \right] \quad (A)$$

In equation (A) V_{mstd} is the dry standard volume of stack gas sampled during Run 1 and Q_{std} is the dry standard volumetric flow rate for the stack gas during Run 1. The term M_F stands for the total mass of fluoride recovered from the Method 4075-A sampling train. This particular train has three areas that collect fluoride - the nozzle, the thimble and the impingers. Particulate fluoride is collected in the nozzle and thimble while gaseous fluoride is captured in the impingers. For analysis, the nozzle wash and a 100 ml sample from the impinger solution were directly distilled (sulfuric acid distillation) to 500 ml volume. They were then directly analyzed by Specific Ion Electrode (SIE) and the impinger value was corrected for a dilution. The contents of the thimble had CaO added, were ashed and then fused with NaOH before being distilled (sulfuric acid distillation) for analysis by SIE. (Refer to Method 4076A.)

why not filter and analyze?

no visible particulate - in nozzle wash. Gary McAllister says it should be OK

For an SIE analysis, a calibration is completed with known concentrations (analytical standards) of fluoride solutions. For each fluoride solution SIE yields a specific millivolt reading. From the millivolt readings and their respective fluoride concentrations a line can be determined. The millivolt reading is plotted vs the log of the F concentration. The F concentration of a sample solution (usually a 10 ml aliquot from the H_2SO_4 distillate) is determined from this graph (or by the equation of the line) by taking the antilog of the Y-axis value for the corresponding X-axis voltage. The fluoride concentration obtained is then factored for dilution (when applicable) and converted to total fluoride mass. Sample calculations for the nozzle wash, thimble and impinger (F_n , F_t and F_i , respectively) F content follow.

Please note that the total mass of fluoride (M_F) collected by this train is equal to the sum of F_n , F_t and F_i .

The nozzle:

	Y	X	LOG Y
Calibrations:	2 mg F/l	828 mV	0.301
	20 mg F/l	244 mV	1.301
	0.20 mg F/l	1341 mV	-0.699
	200 mg F/l	-346 mV	2.301

These values can either be plotted (X vs log Y) or an equation can be developed:

$$Y = 10^{mx+b} \text{ where } m = \text{slope} \\ b = \text{intercept}$$

The Run 1 nozzle wash sample produced an SIE millivolt reading of 1205 mV which corresponded to a concentration of 0.395 mg F/l. Since the distillation volume was 500 ml (0.5 l):

$$F_n = (0.395 \text{ mg F/l}) (0.5 \text{ l}) = 0.198 \text{ mg F or } 0.20 \text{ mg F}$$

Calibration plots can be found in the attachments.

The Thimble:

	Y	X	LOG Y
Calibrations:	2 mg F/l	844 mV	0.301
	20 mg F/l	266 mV	1.301
	0.20 mg F/l	1357 mV	-0.699
	200 mg F/l	-327 mV	2.301

SIE for the Run 1 thimble sample gave a reading of 728 mV which corresponded to a concentration of 3.175 mg F/l. Therefore, total fluoride for the thimble is equal to:

$$F_t = (3.175 \text{ mg F/l}) (0.5 \text{ l}) = 1.588 \text{ mg F or } 1.59 \text{ mg F}$$

The Impingers:

	Y	X	LOG Y
Calibrations:	2 mg F/l	898 mV	0.301
	20 mg F/l	308 mV	1.301
	0.20 mg F/l	1420 mV	-0.699
	200 mg F/l	-279 mV	2.301

The analysis of the impingers differs from that of the nozzle and thimble. The nozzle and thimble analyses make use of the entire submitted sample. Due to the large volume of the impinger field sample ($V_f = 0.510 \text{ l}$) only a 100 ml aliquot (V_a) is used for analysis. This fluoride containing aliquot is distilled by sulfuric acid distillation to a fluoride containing distillate volume (V_d) of 500 ml. This is equivalent to a five-fold dilution in the fluoride concentration because the same mass of fluoride that was originally in the 100 ml aliquot is now in the 500 ml distillate.

From the distillate a 10 ml aliquot is taken for SIE analysis where it produced a reading of 1238 mV. This value corresponded to a distillate fluoride concentration (Cd) of 0.479 mg F/l. The distillate concentration is not equal to field sample concentration (Ca).

Therefore:

$$Ca = (Vd/Va) Cd = (500 \text{ ml}/100 \text{ ml}) (0.479 \text{ mg F/l}) = 2.395 \text{ mg F/l}$$

Since the field sample volume was 0.510 l and its concentration was 2.395 mg F/l, the same as the 100 ml aliquot, the total mass of fluoride contained in the impingers (Fi) equals:

$$Fi = V_f Ca = (0.510 \text{ l}) (2.395 \text{ mg F/l}) = 1.22 \text{ mg F.}$$

Total Fluoride:

$$M_F = F_n + F_t + F_i = 0.20 \text{ mg} + 1.59 \text{ mg} + 1.22 \text{ mg} = 3.01 \text{ mg F.}$$

Therefore, equation (A) yields:

$$1b/hr F = \frac{3.01 \text{ mg F}}{38.28 \text{ dsft}} \left[\frac{38019 \text{ ds ft}^3}{\text{min}} \right] \left[\frac{1 \text{ lb}}{453600 \text{ mg}} \right] \left[\frac{60 \text{ min}}{\text{hr}} \right] = 0.39 \text{ lb F/hr}$$

Review of the laboratory data in the attachment shows QA/QC measures (spiked samples, duplicate samples and blanks) that have been taken.

I hope this letter addresses your needs regarding laboratory calculations. If I can be of any further assistance, please do not hesitate to call.

Sincerely,

James V. Apicella

JAMES V. APICELLA

ext. 2975

Attachments

CC: L. C. Blayden, ATC-C
R. G. Small, Pittsburgh 23
D. Weeter Associates, Louisville, TN
B. W. Bromley, American Rockwool, Spring Hope, NC.

This attachment contains:

- A) Field testing raw data.
- B) Preliminary field testing reduced data.
- C) Fluoride laboratory raw and reduced data.

- 1) Nozzle data.
- 2) Thimble data.
- 3) Impinger data.
- 4) SIE calibration plots.

For ease of identification, the LSN/ID# is provided for each fluoride sample obtained in Run 1.

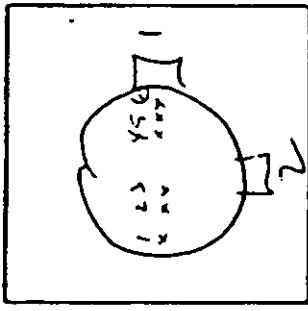
<u>SAMPLE</u>	<u>LSN/ID#</u>
Nozzle	816053/539375
Thimble	816064/539386
Impingers	816041/539363

RAW DATA SET



Plant Spring Hill
 Unit# 6th Inlet/Outlet Stack
 Operator JUH
 Date 88-05-17
 Run No. 1
 Sample Box No. 27
 Meter Box No. 133
 Meter DH# 7312 90/135
 K Factor 0.34 Pilot Leak CK ✓
 Leak Rate Before 0 CFM @ 19 Hg
 Leak Rate After 0 CFM @ 5 Hg
 Static Press., In. H₂O (+/-) 0.12

Start Time 9:25 AM
 Finish Time 10:28 PM
 Meter Yd 1,000



SCHEMATIC OF TESTING LOCATION

K: 7.50 75/105

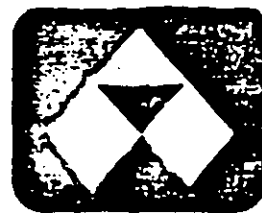
Ambient Temperature, °F 70
 Barometric Pressure, In. Hg 29.84
 Assumed Moisture, % N/A
 Heater Box Setting N/A
 Probe Length, Ft. 1.0 No. 1
 Nozzle Diameter, In. 0.243
 Probe Heater Setting N/A
 % O₂ 13
 % CO₂ 2
 H₂O, ml 11.8
 Silica Gel, gm 16.3
 Total Vic 28.27
 Stack Area, Ft.² 28.27
 Pyrometer No. 1

Filter # 1
 Thimble # 1
 Beaker # 1

RAVERSE POINT NUMBER	MIN/PT	SAMPLING TIME θ, min.	VACUUM In. Hg	STACK TEMPERATURE T _s , °F	VELOCITY HEAD DP, in. H ₂ O	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER DH, In. H ₂ O	Initial Volume: GAS SAMPLE VOLUME V _m , Ft. ³	GAS SAMPLE TEMPERATURE AT DRY GAS METER		TEMP. OF GAS LEAVING CONDENSER OR LAST IMPINGER T _c , °F	SAMPLE BOX TEMPERATURE T _b , °F
								INLET T _a , In. °F	OUTLET T _a , Out. °F		
1-1	5	5	3	102	0.13	1.3	178.40	74	69	39	NA
2	10	10	3	102	0.19	1.4	181.71	78	71	41	
3	15	15	4	100	0.21	1.6	185.26	83	72	42	
4	20	20	4	101	0.21	1.6	188.83	85	73	44	
5	25	25	3	101	0.19	1.4	192.17	86	75	46	
6	30	30	2	101	0.10	0.75	194.65	86	76	48	
2-1	35	35	3	104	0.16	1.2	197.76	83	76	55	
2	40	40	2	106	0.13	0.98	200.41	86	77	53	
3	45	45	4	104	0.21	1.6	204.13	98	78	59	
4	50	50	4	104	0.21	1.6	207.65	87	78	60	
5	55	55	4	105	0.24	1.8	211.38	86	78	60	
6	60	60	2	106	0.13	0.98	214.16	86	78	62	
TOTAL				1236	5.056	16.21		AVG. 1008	AVG. 901		✓
VERAGE				(103)	(0.421)	(1.35)	(81.14)	AVG. (80)			

PARAMETER SHEET

PLANT Am. Rockwell
 CITY Spring Hope STATE NC
 DATE 88-05-17



ALCOA

RUN 7 METHOD 4075-A
 UNIT 6041

O INLET

X OUTLET STACKSTART TIME 9:25 AM/PMFINISH TIME 10:28 AM/PMMETER OPERATOR SVA

COMMENTS:

Preliminary Reduced
Data

METER BOX NO. 2 P
 DH @ 1.93

FILTER NO. _____
 THIMBLE NO. 1

DATA

1	Pb	<u>29.84</u>	"Hg
2	Static	<u>-0.12</u>	"wc
3	Vlc	<u>16.8</u>	ml
4	Mn		g
5	θ	<u>60</u>	min
6	% O ₂	<u>18</u>	
7	% CO ₂	<u>2</u>	
8	DH	<u>1.35</u>	"wc
9	Cp	<u>0.84</u>	
10	Tm	<u>80</u>	F
11	√Dp	<u>0.421</u>	√"wc
12	Ts	<u>103</u>	F
13	Vm	<u>39.14</u>	ft ³
14	Dn	<u>0.243</u>	in
15	As	<u>28.27</u>	ft ²
16	Yd	<u>1.00</u>	cf/cf

RESULTS

Vmstd	<u>38.28</u>	dscf
Vwstd	<u>0.791</u>	scf
Bwo	<u>0.02</u>	
Md	<u>29.04</u>	lb/lb-mole
Ms	<u>28.92</u>	lb/lb-mole
Vs	<u>24.47</u>	ft/s
% I	<u>101.4</u>	
acfm	<u>41500</u>	
dscfm	<u>38020</u>	
Particulate		gr/dscf
		lb/hr
SO ₂		lb/hr
Gaseous F		lb/hr
Particulate F		lb/hr
Total F		lb/hr
K-F	<u>7.6</u>	

ALCOA TECHNICAL CENTER
ANALYTICAL CHEMISTRY DIVISION

88-6-15 8:55 PAGE 1

* FINAL REPORT *

J.O. : 88-052325

NO. SAMP. : 28

APPROVED : 88-06-03

SUBMITTED BY : L.D.PENIX
LOCATION : ENVIRONMENTAL CONTROL LAB
SHOP ORDER : 14H1000891
SYSTEM ID. : SPRING HOPE 4075A'S AM. ROCKWOOL
PROJ. LEADER : JOE GIBB

PHONE : 2572

TID :

PHONE : 2597

REPORT CC : L.C.BLAYDEN, J.E.GIBB, J.V.APICELLA

LSN	SAMPLE IDENTIFICATION
=====	=====
816041	539363 IMPINGERS 1
816042	539364 IMPINGERS 2
816043	539365 IMPINGERS 3
816044	539366 IMPINGERS 4
816045	539367 IMPINGERS 5
816046	539368 IMPINGERS 6
816047	539369 IMPINGERS 7
816048	539370 IMPINGERS 8
816049	539371 IMPINGERS 9
816052	539374 BLANK
816053	539375 NOZZLE 1
816054	539376 NOZZLE 2
816055	539377 NOZZLE 3
816056	539378 NOZZLE 4
816057	539379 NOZZLE 5
816058	539380 NOZZLE 6
816059	539381 NOZZLE 7
816060	539382 NOZZLE 8
816061	539383 NOZZLE 9
816064	539386 THIMBLE 1
816065	539387 THIMBLE 2
816066	539388 THIMBLE 3
816067	539389 THIMBLE 4
816068	539390 THIMBLE 5
816069	539391 THIMBLE 6
816070	539392 THIMBLE 7
816071	539393 THIMBLE 8
816072	539394 THIMBLE 9

N.D. (OR ND) = NOT DETECTED
- (OR BLANK) = NOT DETERMINED

ALCOA TECHNICAL CENTER
ANALYTICAL CHEMISTRY DIVISION

88-06-15 8:55 PAGE 2
J.O. NO. 88-052325

AREA: 201 - FLUORIDE LAB

APPROVED: 88-06-03

ANALYSIS: FLUORIDE

UNITS: TOTAL MG

LSN/ID # FLUORIDE

=====

816041 1.2

539363 IMP

INGERS 1

816042 4.8

539364 IMP

INGERS 2

816043 4.9

539365 IMP

INGERS 3

816044 12.8

539366 IMP

INGERS 4

816045 46.9 ✓

539367 IMP

INGERS 5

816046 87.2

539368 IMP

INGERS 6

816047 66.0

539369 IMP

INGERS 7

816048 26.0

539370 IMP

INGERS 8

816049 37.5

539371 IMP

INGERS 9

816052 #

539374 BLA

NK

ANALYSIS: FLUORIDE

UNITS: TOTAL MG

LSN/ID FLUORIDE
=====

816053 0.2
539375 NOZ
ZLE 1

816054 0.2
539376 NOZ
ZLE 2

816055 0.1
539377 NOZ
ZLE 3

816056 0.7
539378 NOZ
ZLE 4

816057 0.6 ✓
539379 NOZ
ZLE 5

816058 0.3
539380 NOZ
ZLE 6

816059 0.2
539381 NOZ
ZLE 7

816060 0.2
539382 NOZ
ZLE 8

816061 0.3
539383 NOZ
ZLE 9

ANALYSIS: TOTAL PARTICULATE FLUORIDE

UNITS: TOTAL MG

LSN/ID TOTAL PARTICULATE FLUORIDE

=====

816064 1.6

539386 THI

MBLE 1

816065 1.8

539387 THI

MBLE 2

816066 1.9

539388 THI

MBLE 3

816067 4.3

539389 THI

MBLE 4

816068 5.8 ✓

539390 THI

MBLE 5

816069 5.8

539391 THI

MBLE 6

816070 6.6

539392 THI

MBLE 7

816071 6.1

539393 THI

MBLE 8

816072 6.7

539394 THI

MBLE 9

COMMENTS:

SAMPLES WERE ANALYZED BY METHODS 913C (DISTILLATION) AND 914F (SIE)

DUPLICATE: RUN 1 MG/L RUN 2 MG/L

LSN#816047 122 123

SPIKE: LSN# 816045 - 5ML OF A 1000 MG/L FLUORIDE STD WAS ADDED
TO 50 ML OF SAMPLE IN THE DISTILLATION FLASK. FINAL DISTILLATION
VOLUME WAS 500 ML.

SPIKED SAMPLE ANALYSIS 200.8 MG/L

UNSPIKED SAMPLE ANALYSIS 99.9 MG/L

RECOVERED FLUORIDE 100.9 MG/L

ADDED FLUORIDE 100.0 MG/L

% RECOVERY 101 %

- NO VOLUME WAS RECEIVED ON LSN# 816052 (CAUSTIC BLANK)

THE FLUORIDE RESULT WAS 0.3 MG/L

NOZZLE WASH DUPLICATE: RUN1 MG/L RUN2 MG/L

LSN# 816055 1.6 1.6

SPIKE: LSN# 816058 - 5 ML OF A 1000 MG/L FLUORIDE STD WAS ADDED TO
50 ML OF SAMPLE IN DISTILLATION FLASK. FINAL DILUTION VOLUME
WAS 500 ML.

SPIKED SAMPLE ANALYSIS 103.5 MG/L

UNSPIKED SAMPLE ANALYSIS 2.6 MG/L

RECOVERED FLUORIDE 100.9 MG/L

ADDED FLUORIDE 100.0 MG/L

% RECOVERY 101 %

SMALL BLK THIMBLE - 0.1 TOTAL MG FLUORIDE

SPIKE: .1050 G NAF WAS ADDED TO A BLK THIMBLE. TO ALL THIMBLES
.5G CAO WAS ADDED. ALL WERE ASHED, FUSED WITH NaOH, DISTILLED
AND ANALYZED SIE.

47.5 MG OF FLUORIDE WAS ADDED TO THE THIMBLE - 48.0 MG
OF FLUORIDE WAS RECOVERED OR 101 %

LAB REFERENCES: 2760:213, 2760:214, 2760:215

ANALYST(S): SANDRA DEISEROTH

APPROVED BY: NORMA J. HORNUNG

FINAL APPROVED BY: NANCY M. FITZGERALD

*** END OF REPORT ***

2760; 213

L. Penix
14 H/1000891
Rockwood
- Spring type

88-052325
ECL
34

impingers dent
2 1/2 550 SIE

check direct SIE

2760; 213-2

88/05/23 13:18:34
TRAY NO. 1 TRAY I.D. 1 JOB ORDER NO.
CALIBRATION BASED ON TRAY 1
MV READOUT ONLY 16 POSITIONS

POS. 1	MV= 1417
POS. 2	MV= 1419
POS. 3	MV= 3020 1420.2
POS. 6	MV= -279 898.2
POS. 7	MV= 326 308.26
POS. 8	MV= 1232 279.200
POS. 9	MV= 1238 20.1.10
POS. 10	MV= 967 816041
POS. 11	MV= 943 042
POS. 12	MV= 662 18.052325
POS. 13	MV= 486
POS. 14	MV= 387
POS. 15	MV= 372 816047
POS. 16	MV= 434

88/05/23 14:14:27
TRAY NO. 2 TRAY I.D. 1 JOB ORDER NO.
CALIBRATION BASED ON TRAY 1
MV READOUT ONLY 9 POSITIONS

POS. 1	MV= 429
POS. 2	MV= 433 816047 0.47
POS. 3	MV= 522 048
POS. 4	MV= 371 049
POS. 5	MV= 604 050
POS. 6	MV= 918 051
POS. 7	MV= 1553 052
POS. 8	MV= 1080 814274
POS. 9	MV= 1237 814275 18.052325

88/05/23 14:43:32
TRAY NO. 3 TRAY I.D. 1 JOB ORDER NO.
CALIBRATION BASED ON TRAY 1
MV READOUT ONLY 4 POSITIONS

POS. 1	MV= 389 20 pp 10.1.15
POS. 2	MV= 1572
POS. 3	MV= 1601 816042

88/05/23

15:10:25

TRAY NO. 1

TRAY I.D. 1

JOB ORDER NO.88-052325

CALIBRATION BASED ON TRAY 1

CALIBRATION; MV/KB; 16 POSITIONS-NO DISTILLED STD.;

1 2 MG F/L 898 MV
 2 20 MG F/L 308 MV MG/L FLUORIDE= 20.000
 SLOPE(MV20-MV2)= -590
 3 0.20 MG F/L 1420 MV MG/L FLUORIDE= .200
 4 200 MG F/L -279 MV MG/L FLUORIDE= 200.00
 SLOPE(MV200-MV20)= -587

dist

POS 5 JOB ORDER#88-052325 LSN#816041

MV= 1238 ALIQUOT= 10 MLS

ORIG. SPL.= 100 MLS

S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .479

KNOWN CONC.= 0 CALC. CONC.= 2.395(MG F/L)

DIST. FACTOR 0 APPLIED

x .510 ml = 1.22 Total/mg ✓

POS 6 JOB ORDER#88-052325 LSN#816042

MV= 967 ALIQUOT= 10 MLS

ORIG. SPL.= 100 MLS

S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.511

KNOWN CONC.= 0 CALC. CONC.= 7.555(MG F/L)

DIST. FACTOR 0 APPLIED

x .643 ml = 4.8 mg ✓

POS 7 JOB ORDER#88-052325 LSN#816043

MV= 943 ALIQUOT= 10 MLS

ORIG. SPL.= 100 MLS

S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.601

KNOWN CONC.= 0 CALC. CONC.= 8.335(MG F/L)

DIST. FACTOR 0 APPLIED

x .590 ml = 4.9 ✓

POS 8 JOB ORDER#88-052325 LSN#816044

MV= 662 ALIQUOT= 10 MLS

ORIG. SPL.= 100 MLS

S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 5.024

KNOWN CONC.= 0 CALC. CONC.= 25.12 (MG F/L)

DIST. FACTOR 0 APPLIED

x .510 = 12.8 ✓

POS 9 LSN#816045

MV= 486 ALIQUOT= 10 MLS

ORIG. SPL.= 50 MLS

S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 8.985

KNOWN CONC.= 0 CALC. CONC.= 99.856(MG F/L)

DIST. FACTOR 0 APPLIED

x .470 = 41.6 ✓

CALIBRATION: 7 POSITIONS-NO DISTILLED STD.

1	2 MG F/L	898 MV	
2	20 MG F/L	308 MV	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -590		
3	0.20 MG F/L	1420 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-279 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -587		

POS 6 JOB ORDER#88-052325 LSN#=316052
MV= 1601 ALIQUOT= 10 MLS
ORIG. SPL.= 100 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .063
KNOWN CONC.= 0 CALC. CONC.= .315(MG F/L)
DIST. FACTOR 0 APPLIED

POS 9 JOB ORDER# LSN#=20 PPM
MV= 309 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 19.922
KNOWN CONC.= 0 CALC. CONC.= 19.922(MG F/L)
DIST. FACTOR 0 APPLIED

S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 9.985
KNOWN CONC. = 0 CALC. CONC. = 99.850 (MG F/L)
DIST. FACTOR 0 APPLIED $\times 470 = 46.9 \text{ mg} \checkmark$

POS 10 JOB ORDER#88-052325 LSN# = 816045 SPIKE
MV = 307 ALIQUOT = 10 MLS

ORIG. SPL. = 50 MLS
S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 20.079
KNOWN CONC. = 0 CALC. CONC. = 200.79 (MG F/L)
DIST. FACTOR 0 APPLIED $- \frac{99.85}{100.94} \div 100 = 101\% \checkmark$

5 ml of 4000 ppm added to 50 ml

POS 11 JOB ORDER#88-052325 LSN# = 816046
MV = 372 ALIQUOT = 10 MLS

ORIG. SPL. = 50 MLS
S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 15.58
KNOWN CONC. = 0 CALC. CONC. = 155.80 (MG F/L)
DIST. FACTOR 0 APPLIED $\times 560 = 87.2 \checkmark$

POS 12 JOB ORDER#88-052325 LSN# = 816047
MV = 434 ALIQUOT = 10 MLS

ORIG. SPL. = 50 MLS
S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 12.231
KNOWN CONC. = 0 CALC. CONC. = 122.31 (MG F/L)
DIST. FACTOR 0 APPLIED $\times 540 = 66.0 \checkmark$

POS 13 JOB ORDER#88-052325 LSN# = 816047 DUPLICATE
MV = 433 ALIQUOT = 10 MLS

ORIG. SPL. = 50 MLS
S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 12.279
KNOWN CONC. = 0 CALC. CONC. = 122.79 (MG F/L)
DIST. FACTOR 0 APPLIED $\times 540 = 66.3 \checkmark$

POS 14 JOB ORDER#88-052325 LSN# = 816048
MV = 522 ALIQUOT = 10 MLS

ORIG. SPL. = 100 MLS
S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 8.670
KNOWN CONC. = 0 CALC. CONC. = 43.380 (MG F/L)
DIST. FACTOR 0 APPLIED $\times 600 = 26.0 \text{ mg} \checkmark$

POS 15 JOB ORDER#88-052325 LSN# = 816049
MV = 371 ALIQUOT = 10 MLS

ORIG. SPL. = 100 MLS
S. WT. = 0 DIL. VOL. = 500 MLS MG F/L = 15.64
KNOWN CONC. = 0 CALC. CONC. = 78.200 (MG F/L)
DIST. FACTOR 0 APPLIED $\times 480 = 37.5 \checkmark$

1988/06 04:30:25
TRAY NO. 1 TRAY I.D. 1 JOB ORDER NO.88-052325
CALIBRATION BASED ON TRAY 1
CALIBRATION; MV/KB; 16POSITIONS-NO DISTILLED STD.;

1	2 MG F/L	828 MV	
2	20 MG F/L	244 MV	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -584		
3	0.20 MG F/L	1341 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-346 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -590		

POS 5 JOB ORDER#88-052325 LSN#=20 PPM
MV= 242 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 20.157
KNOWN CONC.= 0 CALC. CONC.= 20.157(MG F/L)
DIST. FACTOR 0 APPLIED

POS 6 JOB ORDER#88-052325 LSN#=816053
MV= 1205 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .395
KNOWN CONC.= 0 CALC. CONC.= .198(MG TF)
DIST. FACTOR 0 APPLIED

POS 7 JOB ORDER#88-052325 LSN#=816054
MV= 1213 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .38
KNOWN CONC.= 0 CALC. CONC.= .190(MG TF)
DIST. FACTOR 0 APPLIED

POS 8 JOB ORDER#88-052325 LSN#=816055
MV= 1424 ALIQUOT= 10 MLS
ORIG. SPL.= 39 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .123
KNOWN CONC.= 0 CALC. CONC.= 1.577(MG F/L)
DIST. FACTOR 0 APPLIED

79ml. via vol
x .079 = .12 Total mg ✓

760:214

1988/05 02:49:37
 TRAY NO. 2 TRAY I.D. JOB ORDER NO.
 CALIBRATION BASED ON TRAY 1
 MV READOUT ONLY 16 POSITIONS

POS. 1	MV= 1196
POS. 2	MV= 1345
POS. 3	MV= 1341.2
POS. 4	MV= 828.2
POS. 5	MV= 244.20
POS. 6	MV= -346.200
POS. 7	MV= 258
POS. 8	MV= 242.2011
POS. 9	MV= 1205.216053
POS. 10	MV= 1213.050
POS. 11	MV= 1424.055
POS. 12	MV= 1414.055
POS. 13	MV= 906.050
POS. 14	MV= 949.057
POS. 15	MV= 1294.058
POS. 16	MV= 411.058

1988/06 04:04:38
 TRAY NO. 2 TRAY I.D. JOB ORDER NO.
 CALIBRATION BASED ON TRAY 1
 MV READOUT ONLY 6 POSITIONS

POS. 1	MV= 1129
POS. 2	MV= 1175.816059
POS. 3	MV= 1181.050
POS. 4	MV= 1085.050
POS. 5	MV= 1168.050
POS. 6	MV= 956.050

L. Penix 88 05.0725

14141000891 EQL

rockwood 34

nozzle work - directly dist & H₂ SOV
 SIE & Orion.

spine 50ml dist 818058 - total mol 0.300C

MV= 1414 ALIQUOT= 10 MLS
ORIG. SPL.= 40 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .131
KNOWN CONC.= 0 CALC. CONC.= 1.638(MG F/L)
DIST. FACTOR 0 APPLIED $x.079 = .13$ Total mg dup

POS 10 JOB ORDER#88-052325 LSN#=816056
MV= 906 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.451
KNOWN CONC.= 0 CALC. CONC.= .726(MG TF)
DIST. FACTOR 0 APPLIED

POS 11 JOB ORDER#88-052325 LSN#=816057 ✓
MV= 949 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 1.213
KNOWN CONC.= 0 CALC. CONC.= .607(MG TF)
DIST. FACTOR 0 APPLIED

POS 12 JOB ORDER#88-052325 LSN#=816058
MV= 1294 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .256
KNOWN CONC.= 0 CALC. CONC.= 2.560(MG F/L)
DIST. FACTOR 0 APPLIED $x.103 \text{ ml} = 0.26$ Total mg ✓
103 ml orig vol

POS 13 JOB ORDER#88-052325 LSN#=816058 SPIKE
MV= 411 ALIQUOT= 10 MLS
ORIG. SPL.= 50 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 10.353
KNOWN CONC.= 0 CALC. CONC.= 103.53(MG F/L)
DIST. FACTOR 0 APPLIED
5 ml of 1000 ppm std $\frac{0.6}{100.9} \text{ or } 101\% \checkmark$

POS 14 JOB ORDER#88-052325 LSN#=816059
MV= 1175 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .454
KNOWN CONC.= 0 CALC. CONC.= .227(MG TF)
DIST. FACTOR 0 APPLIED

POS 15 JOB ORDER#88-052325 LSN#=816060
MV= 1181 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .441
KNOWN CONC.= 0 CALC. CONC.= .221(MG TF)
DIST. FACTOR 0 APPLIED

POS 16 JOB ORDER#88-052325 LSN#=816061
MV= 1085 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= .679
KNOWN CONC.= 0 CALC. CONC.= .34(MG TF)
DIST. FACTOR 0 APPLIED

389/06 04:34:41
TRAY NO. 2 TRAY I.D. 2 JOB ORDER NO. 88-052325
CALIBRATION BASED ON TRAY 2
CALIBRATION; MV/KB; 6 POSITIONS-NO DISTILLED STD.;

1	2 MG F/L	828 MV	
2	20 MG F/L	244 MV	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -584		
3	0.20 MG F/L	1341 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-346 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -590		

816055 dup

run 1 mg/L

1.6

run 2 mg/L

1.6

816058 spike

5ml of 1000 ppm added to 50ml of
sample in that flask. Final dil vol 500ml

spiked 103.5 mg/L
un " 2.6 mg/L
recoo 100.9 mg/L
added 100 mg/L
to recoo 101.9

2760: 215

L. Penix
14H 100089
Rockwood
Spring Hope

88-052325
ECL
34

filters - wet up .5g CaO added, ashed fused
E NaOH. dist E H₂SO₄ 32E

No BIK filter received. ashed sm. filter.
added .1050 g NaF to BIK thimble for 3 piles.

1988/06 01:03:42

TRAY NO. 1 TRAY I.D. 1

JOB ORDER NO.

CALIBRATION BASED ON TRAY 1

MV READOUT ONLY 16 POSITIONS

POS. 1	MV=	1358
POS. 2	MV=	1362
POS. 3	MV=	1357.2
POS. 4	MV=	844 2
POS. 5	MV=	266 20
POS. 6	MV=	-327 200
POS. 7	MV=	258
POS. 8	N58 MV=	728 816064
POS. 9	N87 MV=	702 005
POS. 10	N2 MV=	687 000
POS. 11	N4 MV=	477 000
POS. 12	N15 MV=	403 000
POS. 13	N6 MV=	404 069
POS. 14	N70 MV=	372 070
POS. 15	N81 MV=	389 071
POS. 16	N5 MV=	367 072

1988/06 02:07:46

TRAY NO. 2 TRAY I.D.

JOB ORDER NO.

CALIBRATION BASED ON TRAY 1

MV READOUT ONLY 8 POSITIONS

POS. 1	MV=	26220 ppm H ₂ O std repeat
POS. 2	MV=	26120 ppm dist std repeat
POS. 3	N14 MV=	551 073
POS. 4	N1 MV=	453 074
POS. 5	MV=	1354
POS. 6	MV=	1354 BIK thimble
POS. 7	MV=	-138
POS. 8	MV=	-138 BIK thimble + .1059 NaF

2760 : 215

1988/06 07:09:15
TRAY NO. 1 TRAY I.D. 1 JOB ORDER NO.88-052325
CALIBRATION BASED ON TRAY 1
CALIBRATION; MV/KB; 16POSITIONS-NO DISTILLED STD.;

1	2 MG F/L	844 MV	
2	20 MG F/L	266 MV	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -578		
3	0.20 MG F/L	1357 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-327 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -593		

POS 5 JOB ORDER#88-052325 LSN#=20 PPM
MV= 261 ALIQUOT= 10 MLS
ORIG. SPL.= 500 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 20.392
KNOWN CONC.= 0 CALC. CONC.= 20.392(MG F/L)
DIST. FACTOR 0 APPLIED

POS 6 JOB ORDER#88-052325 LSN#=816064
MV= 728 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 3.175
KNOWN CONC.= 0 CALC. CONC.= 1.588(MG TF)
DIST. FACTOR 0 APPLIED

POS 7 JOB ORDER#88-052325 LSN#=816065
MV= 702 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 3.521
KNOWN CONC.= 0 CALC. CONC.= 1.761(MG TF)
DIST. FACTOR 0 APPLIED

POS 8 JOB ORDER#88-052325 LSN#=816066
MV= 687 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 3.738
KNOWN CONC.= 0 CALC. CONC.= 1.869(MG TF)
DIST. FACTOR 0 APPLIED

POS 9 JOB ORDER#88-052325 LSN#=816067
MV= 477 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 8.629
KNOWN CONC.= 0 CALC. CONC.= 4.315(MG TF)
DIST. FACTOR 0 APPLIED

POS 10 JOB ORDER#88-052325 LSN#=816068
MV= 403 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 11.588 *OK*
KNOWN CONC.= 0 CALC. CONC.= 5.794(MG TF) *OK*
DIST. FACTOR 0 APPLIED

POS 11 JOB ORDER#88-052325 LSN#=816069
MV= 404 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 11.542
KNOWN CONC.= 0 CALC. CONC.= 5.771(MG TF)
DIST. FACTOR 0 APPLIED

POS 12 JOB ORDER#88-052325 LSN#=816070
MV= 372 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 13.111
KNOWN CONC.= 0 CALC. CONC.= 6.556(MG TF)
DIST. FACTOR 0 APPLIED

POS 13 JOB ORDER#88-052325 LSN#=816071
MV= 389 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 12.253
KNOWN CONC.= 0 CALC. CONC.= 6.127(MG TF)
DIST. FACTOR 0 APPLIED

POS 14 JOB ORDER#88-052325 LSN#=816072
MV= 367 ALIQUOT= 10 MLS
ORIG. SPL.= 0 MLS
S. WT.= 0 DIL. VOL.= 500 MLS MG F/L= 13.375
KNOWN CONC.= 0 CALC. CONC.= 6.688(MG TF)
DIST. FACTOR 0 APPLIED

1988/06

07:13:33

TRAY NO. 2

TRAY I.D. 2

JOB ORDER NO. 88-052325

CALIBRATION BASED ON TRAY 2

CALIBRATION; MV/KB; 6 POSITIONS-NO DISTILLED STD.;

1	2 MG F/L	844 MV	
2	20 MG F/L	266 MV	MG/L FLUORIDE= 20.000
	SLOPE(MV20-MV2)= -578		
3	0.20 MG F/L	1357 MV	MG/L FLUORIDE= .200
4	200 MG F/L	-327 MV	MG/L FLUORIDE= 200.00
	SLOPE(MV200-MV20)= -593		

POS 5 JOB ORDER#88-052325

LSN#=BLK THIMBLE

MV= 1354

ALIUOT= 10 MLS

ORIG. SPL.= 0 MLS

S. WT.= 0

DIL. VOL.= 500 MLS

MG F/L= .203

KNOWN CONC.= 0

CALC. CONC.= .102(MG TF)

DIST. FACTOR 0 APPLIED

POS 6 JOB ORDER#88-052325 LSN#=THIMBLE, .10500 NAF

MV=-138

ALIUOT= 10 MLS

ORIG. SPL.= 0 MLS

S. WT.= 0

DIL. VOL.= 500 MLS

MG F/L= 96.009

KNOWN CONC.= 0

CALC. CONC.= 48.005(MG TF)

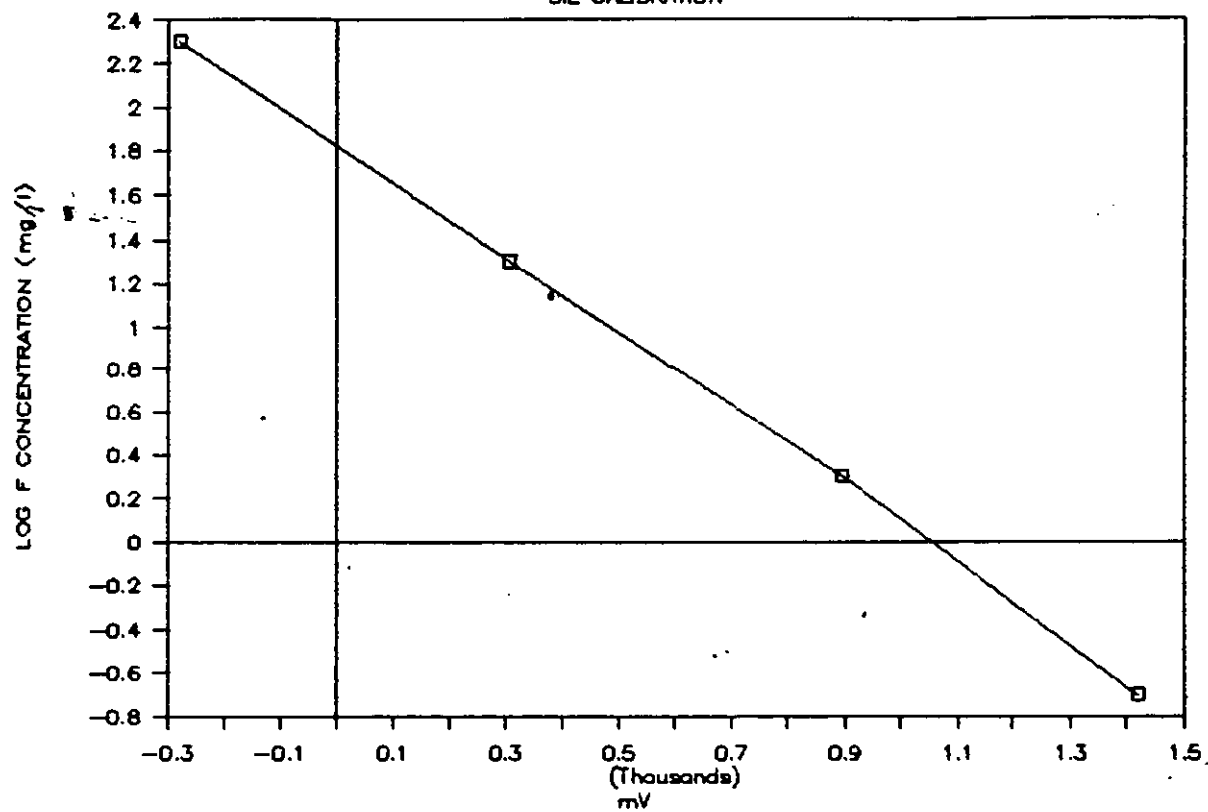
DIST. FACTOR 0 APPLIED

.1050g NaF / 2.21 = 47.5 mg F added

48 ÷ 47.5 = 101 % ✓

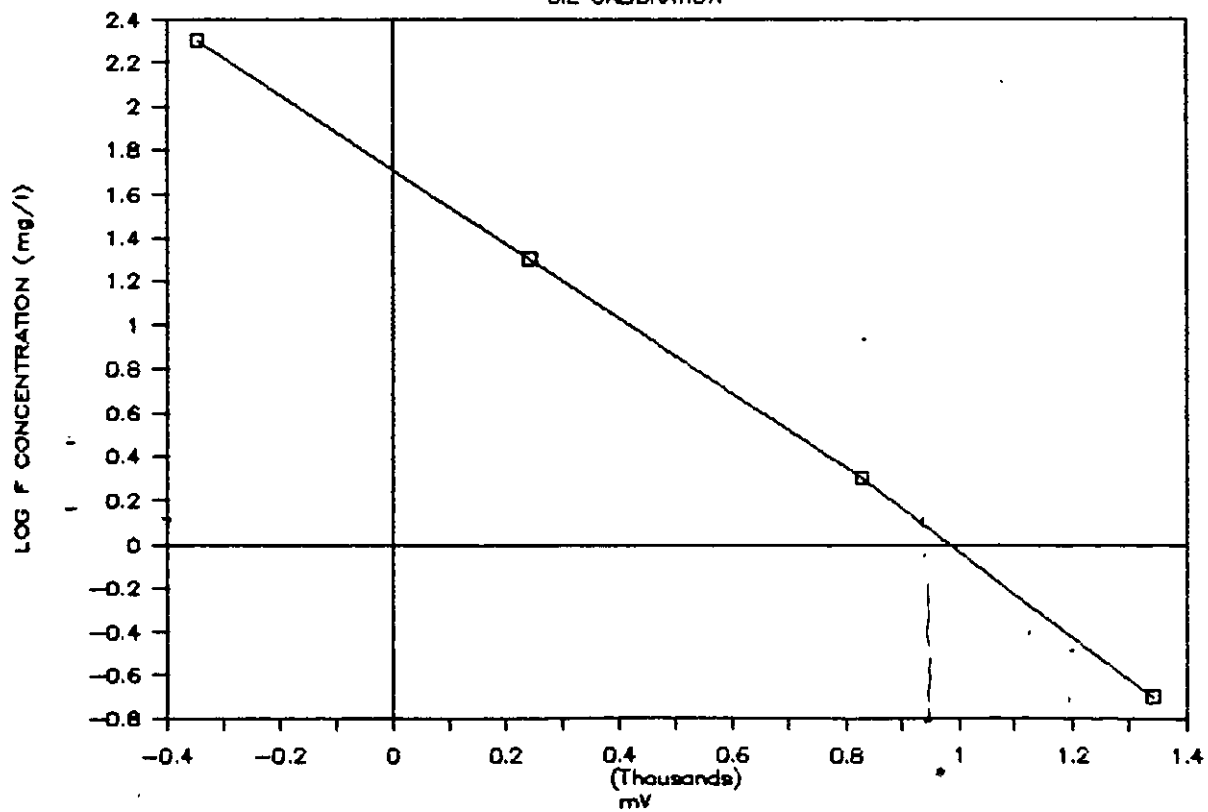
IMPINGER

SIE CALIBRATION



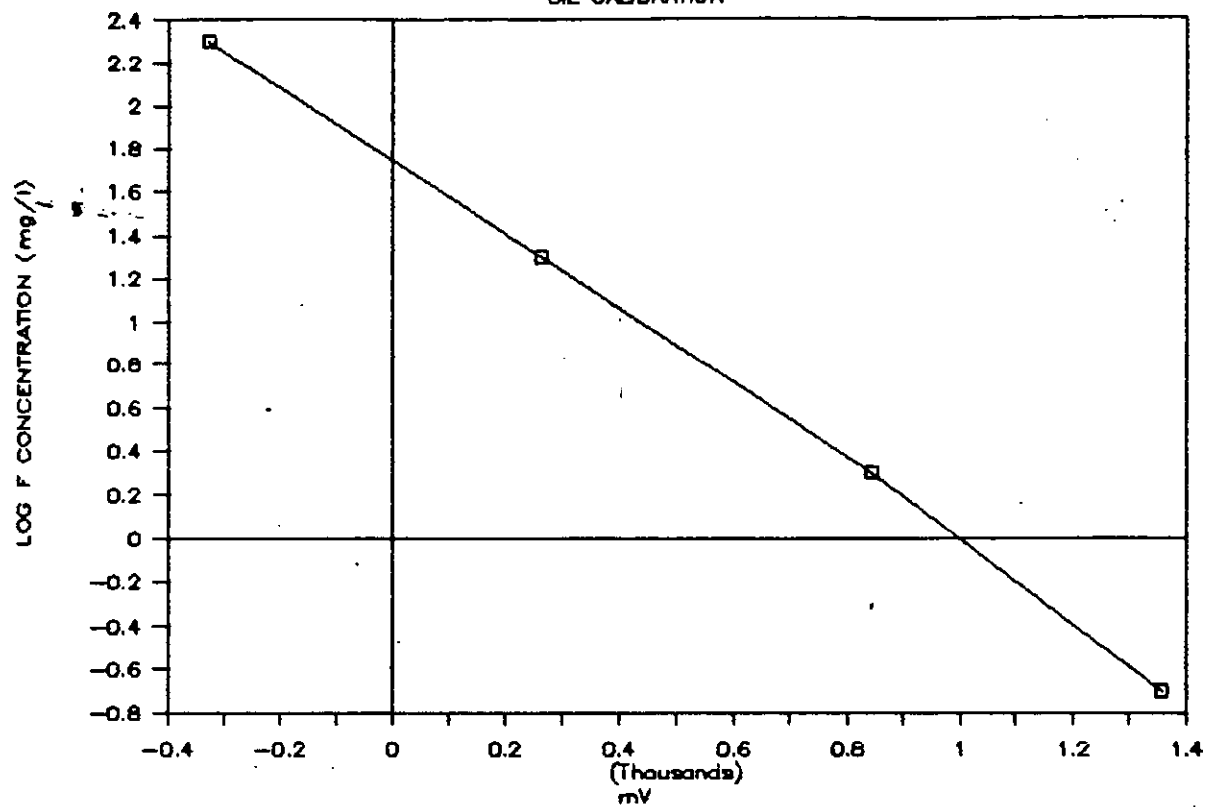
NOZZLE

SIE CALIBRATION



THIMBLE

SIE CALIBRATION



MINERAL WOOL EMISSION FACTORS DEVELOPED FROM AP-40

Process	Pollutant	Production rate, tons/hr	Mass flux, lbs/hr	Emission factor		%	flow rate, DSCFM
				kg/Mg	lbs/ton		
Cupola	PM (filt.)	3525/2000	50	14	28	Concen., mg/DSCF	
	PM (filt.)	4429/2000	46	10	21		
	PM (filt.)	3625/2000	29	8.0	16		
	AVERAGE			11	22		
	SO2	3525/2000	20	5.6	11	32.6	4550
	SO3	3525/2000	11	3.2	6.3	18.5	4550
						%	
	CO	3525/2000	160	45	91	0.9	4550
Reverberatory furnace	PM (filt.)	3050/2000	7.3	2.4	4.8	Concen., mg/DSCF	
Blow chamber	PM (filt.)	3525/2000	9.2	2.6	5.2		
	PM (filt.)	3625/2000	7.1	2.0	3.9		
	PM (filt.)	3525/2000	98	28	56		
	PM (filt.)	4120/2000	8.3	2.0	4.0		
	AVERAGE			8.6	17		
	SO2	3525/2000	1.5	0.43	0.87	1.04	11,100
	aldehydes	3525/2000	1.5	0.43	0.86	1.03	11,100
Curing oven	PM (filt.)	3525/2000	9.0	2.5	5.1	8.2	1,530
	PM (filt.)	3625/2000	5.2	1.4	2.9	Concen., mg/DSCF	
	PM (filt.)	3050/2000	2.3	0.7	1.5		
	PM (filt.)	5180/2000	15	2.9	5.9		
	PM (filt.)	3500/2000	5.0	1.4	2.9		
	AVERAGE			1.8	3.6		
	SO2	3525/2000	2.0	0.58	1.2	3.23	4,740
	aldehydes	5180/2000	1.9	0.37	0.73		
	aldehydes	3500/2000	2.2	0.63	1.3		
	AVERAGE			0.50	1.00		
	NO2	5180/2000	0.60	0.12	0.23		
	NO2	3500/2000	0.15	0.043	0.086		
	AVERAGE			0.079	0.16		
cooler	PM (filt.)	3525/2000	0.75	0.21	0.43	Concen., mg/DSCF	
	PM (filt.)	3700/2000	2.6	0.69	1.4		
	PM (filt.)	3050/2000	3.6	1.2	2.3		
	PM (filt.)	3050/2000	8.4	2.8	5.5		
	AVERAGE			1.2	2.4		
	SO2	3525/2000	0.12	0.034	0.068	0.49	1,850
	aldehydes	3525/2000	0.074	0.021	0.042	0.30	1,850

**AQMD PROGRAM
REVIEW TRACKING FORM**

Document name: <i>WOOL</i>	MRI Lead: <i>Rick Marinshaw</i>
Project name: <i>Mineral Products AP-42</i>	EPA Lead: <i>Ron Myers</i>
Charge no. (project-task-subtask): <i>9711-30-53</i>	Due to client: <i>/ /</i>
Last reviewer return document to: <i>Brian Shrager</i>	Return by: <i>/ /</i>
Route to: <i>Joanne Darryl</i>	On: <i>8/20/92</i>
Review instructions: <i>Joanne can't get to this until next week. (AP-42 Mineral Wool)</i>	
Review comments: <i>Brian - I'm sorry I did not get but half way thru this today. I am returning it because I will be out next week. I got to page 22. I be happy to review</i>	
Ready for transmit? <input type="checkbox"/> yes <input type="checkbox"/> no	Initials: <i>DLH</i> <i>8/21/92</i>
Route to: <i>Darryl Joanne</i>	No later than: <i>/ /</i>
Review instructions:	
Review comments: <i>Looks good! need to make Times 11 pt (9-pt supers & sub's). Questions/comments noted - see me if you have any questions/want to go over this. Thanks!</i>	
Ready for transmit? <input type="checkbox"/> yes <input type="checkbox"/> no	Initials: <i>Jmk</i> <i>8/26/92</i>
Route to: <i>Dennis</i>	No later than: <i>/ /</i>
Review instructions: <i>Dennis, several comments that change the meaning of what you wrote were made. I'm not sure what to do w/ these comments!</i>	
Review comments:	
Ready for transmit? <input type="checkbox"/> yes <input type="checkbox"/> no	Initials: <i>/ /</i>

on Aug 31. back

**AQMD PROGRAM
REVIEW TRACKING FORM**

Document name:	MRI Lead:
Project name:	EPA Lead:
Charge no. (project-task-subtask):	Due to client: / /
Last reviewer return document to:	Return by: / /
Route to: Rick M.	On: 8 / 20 / 92
Review instructions: Rick, Heres mineral wool. I thought you might want to look @ it before WP does final corrections. I question - see Reference 4 on last page (also Ref 4 pg 10).	
Review comments:	
Ready for transmit? yes no	Initials: / /
Route to:	No later than: / /
Review instructions:	
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5. DRAFT AP-42 SECTION 8.16

8.16 MINERAL WOOL MANUFACTURING

8.16.1 General^{1,2}

Mineral wool often is defined as any fibrous glassy substance made from minerals (typically natural rock materials such as basalt or diabase) or mineral products such as slag and glass. Because glass wool production is covered separately in AP-42 (Section 8.11), this section deals only with the production of mineral wool from natural rock and slags such as iron blast furnace slag, the primary material, and copper, lead, and phosphate slags. These materials are processed into insulation and other fibrous building materials that are used for structural strength and fire resistance. Generally, these products take one of four forms: "blowing" wool or "pouring" wool, which is put into the structural spaces of buildings; batts, which may be covered with a vapor barrier of paper or foil and are shaped to fit between the structural members of buildings; industrial and commercial products such as high-density fiber felts and blankets, which are used for insulating boilers, ovens, pipes, refrigerators, and other process equipment; and bulk fiber, which is used as a raw material in manufacturing other products, such as ceiling tile, wall board, spray-on insulation, cement, and mortar.

Mineral wool manufacturing facilities are included in Standard Industrial Classification (SIC) Code 3296, mineral wool. This SIC code also includes the production of glass wool insulation products, but those facilities engaged in manufacturing textile glass fibers are included in SIC Code 3229. The six digit source category code (SCC) for mineral wool manufacturing is 3-05-017.

8.16.2 Process Description^{1,4,5}

Most mineral wool produced in the United States today is produced from slag or a mixture of slag and rock. Most of the slag used by the industry is generated by integrated iron and steel plants as a blast furnace byproduct from pig iron production. Other sources of slag include the copper, lead, and phosphate industries. The production process has three primary components—molten

mineral generation in the cupola, fiber formation and collection, and final product formation. Figure 8.16-1 illustrates the mineral wool manufacturing process.

The first step in the process involves melting the mineral feed. The raw material (slag and rock) is loaded into a cupola in alternating layers with coke at weight ratios of about 5 to 6 parts mineral to 1 part coke. As the coke is ignited and burned, the mineral charge is heated to the molten state at a temperature of 1300° to 1650°C (2400° to 3000°F). Combustion air is supplied through tuyeres located near the bottom of the furnace. Process modifications at some plants include air enrichment and the use of natural gas auxiliary burners to reduce coke consumption. One facility also reported using an aluminum flux byproduct to reduce coke consumption.

The molten mineral charge exits the bottom of the cupola in a water-cooled trough and falls onto a fiberization device. Most of the mineral wool produced in the United States is made by variations of two fiberization methods. The Powell process uses groups of rotors revolving at a high rate of speed to form the fibers. Molten material is distributed in a thin film on the surfaces of the rotors and then is thrown off by centrifugal force. As the material is discharged from the rotor, small globules develop on the rotors and form long, fibrous tails as they travel horizontally. Air or steam may be blown around the rotors to assist in fiberizing the material. A second fiberization method, the Downey process, uses a spinning concave rotor with air or steam attenuation. Molten material is distributed over the surface of the rotor, from which it flows up and over the edge and is captured and directed by a high-velocity stream of air or steam.

During the spinning process, not all globules that develop are converted into fiber. The nonfiberized globules that remain are referred to as "shot." In raw mineral wool, as much as half of the mass of the product may consist of shot. As shown in Figure 8.16-1, shot is usually separated from the wool by gravity immediately following fiberization.

Depending on the desired product, various chemical agents may be applied to the newly formed fiber immediately following the rotor. In almost all cases, an oil is applied to suppress dust and, to some degree, anneal the fiber. This oil can be either a proprietary product or a medium-weight fuel or lubricating oil. If the fiber is intended for use as loose wool or bulk products, no further chemical treatment is necessary. If the mineral wool product is required to have structural rigidity, as in batts and industrial felt, a binding agent is applied with or in place of the oil treatment. This binder is

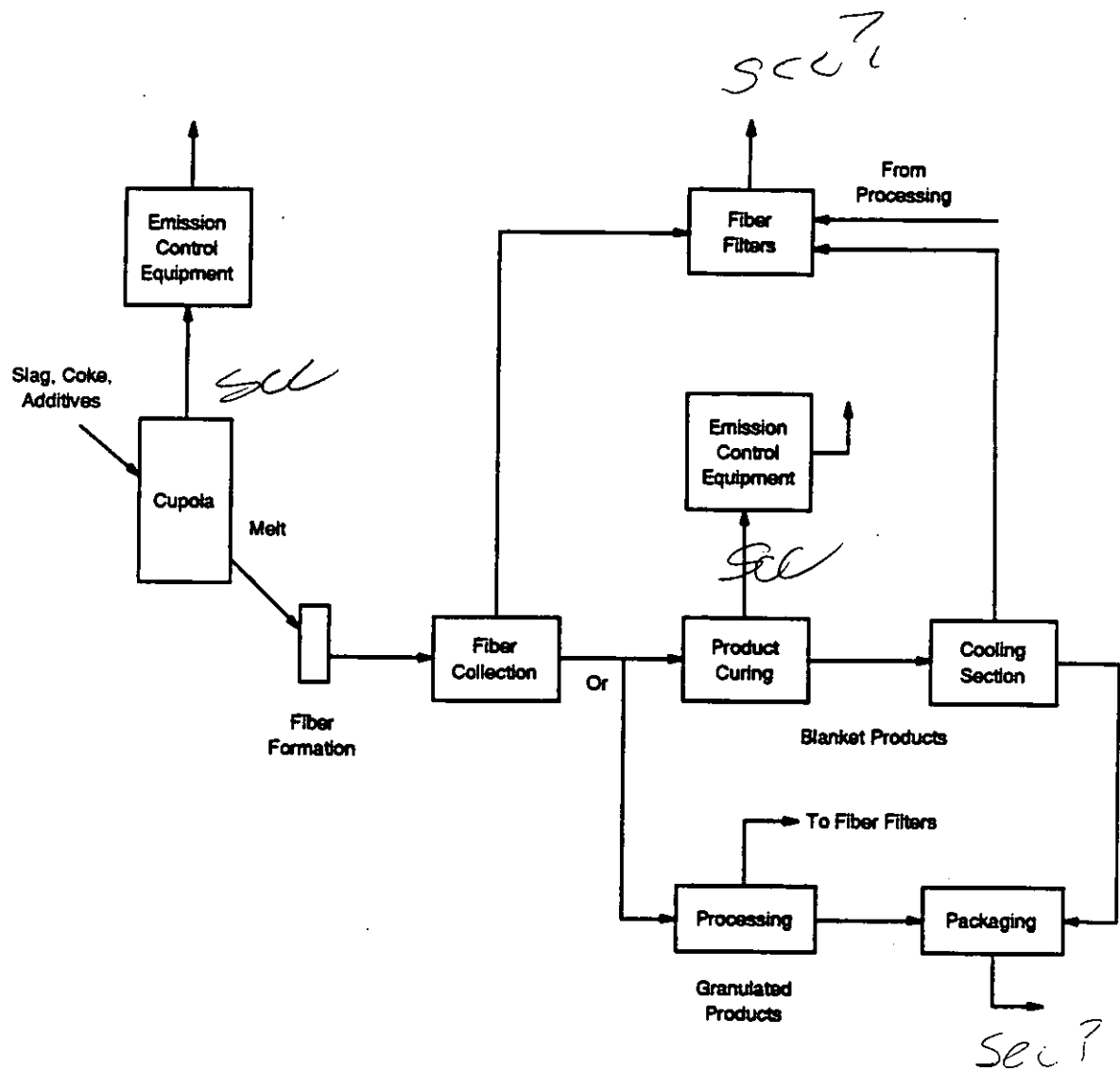


Figure 8.16-1. Mineral wool manufacturing process flow diagram.

typically a phenol-formaldehyde resin that requires curing at elevated temperatures. Both the oil and the binder are applied by atomizing the liquids and spraying the agents to coat the airborne fiber.

After formation and chemical treatment, the fiber is collected in a blowchamber. Resin-and/or oil-coated fibers are drawn down on a wire mesh conveyor by fans located beneath the collector. The speed of the conveyor is set so that a wool blanket of desired thickness can be obtained.

Mineral wool containing the binding agent is carried by conveyor to a curing oven, where the wool blanket is compressed to the appropriate density and the binder is baked. Hot air, at a temperature of 150° to 320°C (300° to 600°F), is forced through the blanket until the binder has set. Curing time and temperature depend on the type of binder used and the mass rate through the oven. A cooling section follows the oven, where blowers force air at ambient temperatures through the wool blanket.

To make batts and industrial felt products, the cooled wool blanket is cut longitudinally and transversely to the desired size. Some insulation products are then covered with a vapor barrier of aluminum foil or asphalt-coated kraft paper on one side and untreated paper on the other side. The cutters, vapor barrier applicators, and conveyors are sometimes referred to collectively as a batt machine. Those products that do not require a vapor barrier, such as industrial felt and some residential insulation batts, can be packed for shipment immediately after cutting.

Loose wool products consist primarily of blowing wool and bulk fiber. For these products, no binding agent is applied, and the curing oven is eliminated. For granulated wool products, the fiber blanket leaving the blowchamber is fed to a shredder and pelletizer. The pelletizer forms small, 1-inch diameter pellets and separates shot from the wool. A bagging operation completes the processes. For other loose wool products, fiber can be transported directly from the blowchamber to a baler or bagger for packaging.

8.16.3 Emissions and Controls¹

The sources of emissions in the mineral wool manufacturing industry are the cupola, the blow chamber, the curing oven, the mineral wool cooler, and possibly materials handling and bagging operations. With the exception of lead, the industry emits the full range of criteria pollutants. Also,

depending on the particular types of slag and binding agents used, the facilities may emit both metallic and organic hazardous air pollutants (HAP's).

The primary source of emissions in the mineral wool manufacturing process is the cupola. It is a significant source of particulate matter (PM) emissions and is likely to be a source of PM less than 10 micrometers (μm) in diameter (PM-10) emissions, although no particle size data are available. Coke combustion in the furnace produces carbon monoxide (CO), carbon dioxide (CO_2), and nitrogen oxide (NO_x) emissions. Finally, because blast furnace slags contain sulfur, the cupola is also a source of sulfur dioxide (SO_2) and hydrogen sulfide (H_2S) emissions.

The blowchamber is a source of PM (and probably PM-10) emissions. Also, the annealing oils used in the process can lead to VOC emissions from the process. Other sources of VOC emissions include batt application and the curing oven. Finally, fugitive PM emissions can be generated during cooling, handling, and bagging operations. Table 8.16-1 presents emission factors for filterable PM emissions from various mineral wool manufacturing processes; Table 8-16.2 shows emission factors for CO, CO_2 , SO_2 , and sulfates; and Table 8.16-3 presents emission factors for fluorides.

Mineral wool manufacturers use a variety of air pollution control techniques, but most are directed toward PM control with minimal control of other pollutants. The industry has given greatest attention to cupola PM control, with two-thirds of the cupolas in operation having fabric filter control systems. Some cupola exhausts are controlled by wet scrubbers and electrostatic precipitators (ESP's); cyclones are also used for cupola PM control either alone or in combination with other control devices. About half of the blow chambers in the industry also have some level of PM control, with the predominant control device being low-energy wet scrubbers. Cyclones and fabric filters have been used to a limited degree on blow chambers. Finally, afterburners have been used to control VOC emissions from blow chambers and curing ovens and CO emissions from cupolas.

TABLE 8.16-1. (METRIC UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

*No PM-10
data of
assumptions?*

All emission factors in kg/Mg of product unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	Filterable PM ^b	
Cupola ^c (30501701)	8.2	E
Cupola with fabric filter ^d (30501701)	0.051	D
Reverberatory furnace ^e (30501702)	2.4	E
Batt curing oven ^e (30501704)	1.8	E
Batt curing oven with ESP ^f (30501704)	0.36	D
Blow chamber ^e (30501703)	6.0	E
Blow chamber with wire mesh filter ^g (30501703)	0.45	D
Cooler ^e (30501705)	1.2	E

^aFactors represent uncontrolled emissions unless otherwise noted.

^bFilterable PM is that PM collected on or prior to the filter of an EPA Method 5 (or equivalent) sampling train.

^cReferences 1, 12. Activity level is assumed to be total feed charged.

^dReferences 6, 7, 8, 10, and 11. Activity level is total feed charged.

^eReference 12.

^fReference 9.

^gReference 7. Activity level is mass of molten mineral feed charged.

TABLE 8.16-1. (ENGLISH UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in lb/ton of product unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	Filterable PM ^b	
Cupola ^c (30501701)	16	E
Cupola with fabric filter ^d (30501701)	0.10	D
Reverberatory furnace ^e (30501702)	4.8	E
Batt curing oven ^e (30501704)	3.6	E
Batt curing oven with ESP ^f (30501704)	0.72	D
Blow chamber ^e (30501703)	12	E
Blow chamber with wire mesh filter ^g (30501703)	0.91	D
Cooler ^e (30501705)	2.4	E

^aFactors represent uncontrolled emissions unless otherwise noted.

^bFilterable PM is that PM collected on or prior to the filter of an EPA Method 5 (or equivalent) sampling train.

^cReference 1, 12. Activity level is assumed to be total feed charged.

^dReferences 6, 7, 8, 10, and 11. Activity level is total feed charged.

^eReference 12.

^fReference 9.

^gReference 7. Activity level is mass of molten mineral feed charged.

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TABLE 8.16-2 (METRIC UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in kg/Mg of total feed charged unless noted
Ratings (A-E) follow each emission factor

Source (SCC)	CO ^b		CO ₂ ^b		SO ₂		SO ₃	
Cupola (30501701)	125	D	260	D	4.0 ^c	D	3.2 ^d	E
Cupola with fabric filter (30501701)	NA		NA		NA		0.077 ^b	E
Cupola with fabric filter (30501701)	NA		NA		NA		e	
Batt curing oven (30501704)	e		e		0.58 ^d	E	e	
Blow chamber (30501703)	e		80 ^f	E	0.43 ^d	E	e	
Cooler (30501705)	e		e		0.034 ^d	E	e	

NA = Not applicable.

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 6.

^cReferences 6, 10, and 11.

^dReference 12.

^eNo data available.

^fReference 9.

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TABLE 8.16-2 (ENGLISH UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in lb/ton of total feed charged unless noted
Ratings (A-E) follow each emission factor

Source (SCC)	CO ^b		CO ₂ ^b		SO ₂		SO ₃	
Cupola (30501701)	250	D	520	D	8.0 ^a	D	6.3 ^d	E
Cupola with fabric filter (30501701)	NA		NA		NA		0.15 ^b	E
Cupola with fabric filter (30501701)	NA		NA		NA		e	
Batt curing oven (30501704)					1.2 ^d	E		
Blow chamber (30501703)			160 ^f	E	0.087 ^d	E		
Cooler (30501705)					0.068 ^d	E		

NA = Not applicable.

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 6.

^cReferences 6, 10, and 11.

^dReference 12.

^eNo data available.

^fReference 9.

TABLE 8.16-3 (METRIC UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in kg/Mg of total feed charged unless noted
 Ratings (A-E) follow each emission factor

Process (SCC)	NO _x		N ₂ O		H ₂ S		Fluorides	
Cupola (30501701)	0.8 ^b	E	c ND		1.5 ^b	E	c ND	
Cupola with fabric filter (30501701)	c ND		c ND		c ND		0.019 ^d	D
Cupola with fabric filter (30501701)	c ND		c ND		c ND		0.19 ^e	D
Batt curing oven (30501714)	c ND		0.079	E	c ND		c ND	

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 1.

^cNo data available.

^dReferences 10 and 11. Coke only used as fuel.

^eReferences 10 and 11. Fuel combination of coke and aluminum smelting byproducts.

TABLE 8.16-3 (ENGLISH UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in lb/ton of total feed charged unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	NO _x		N ₂ O		H ₂ S		Fluorides	
Cupola (30501701)	1.6 ^b	E	c		3.0 ^b	E	c	
Cupola with fabric filter (30501701)	c		c		c		0.038 ^d	D
Cupola with fabric filter (30501701)	c		c		c		0.38 ^e	D
Batt curing oven (30501714)	c		0.16	E	c		c	

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 1.

^cNo data available.

^dReferences 10 and 11. Coke only used as fuel.

^eReferences 10 and 11. Fuel combination of coke and aluminum smelting byproducts.

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w/ 1 set foot notes*

REFERENCES FOR SECTION 8.16

1. Source Category Survey: Mineral Wool Manufacturing Industry, EPA-450/3-80-016, U. S. Environmental Protection Agency, Research Triangle Park, NC, March 1980.
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4. Personal communication between F. May, U.S.G. Corporation, Chicago, Illinois, and R. Marinshaw, Midwest Research Institute, Cary, NC, June 5, 1992.
5. Memorandum from K. Schuster, N.C. Department of Environmental Management, to M. Aldridge, American Rockwool, April 25, 1988.
6. Sulfur Oxide Emission Tests Conducted on the #1 and #2 Cupola Stacks in Leeds, Alabama for Rock Wool Company, November 8 & 9, 1988, Guardian Systems, Inc., Leeds, AL, Undated.
7. Particulate Emissions Tests for U.S. Gypsum Company on the Number 4 Dry Filter and Cupola Stack Located in Birmingham, Alabama on January 14, 1981, Guardian Systems, Inc., Birmingham, AL, Undated.
8. Untitled Test Report, Cupolas Nos. 1, 2, and 3, U.S. Gypsum, Birmingham, AL, June 1979.
9. Particulate Emission Tests on Batt Curing Oven for U.S. Gypsum, Birmingham, Alabama on October 31-November 1, 1977, Guardian Systems, Inc., Birmingham, AL, Undated.
10. J.V. Apicella, Particulate, Sulfur Dioxide, and Fluoride Emissions from Mineral Wood Emission, with Varying Charge Compositions, American Rockwool, Inc. Spring Hope, N.C. 27882, Alumina Company of America, Alcoa Center, PA, June 1988.
11. J.V. Apicella, Compliance Report on Particulate, Sulfur Dioxide, Fluoride, and Visual Emissions from Mineral Wood Production, American Rockwool, Inc., Spring Hope, NC 27882, Aluminum Company of America, Alcoa Center, PA, February 1988.
12. J.L. Spinks, "Mineral Wool Furnaces," In: Air Pollution Engineering Manual, J.A. Danielson, ed., U. S. DHEW, PHS, National Center for Air Pollution Control, Cincinnati, OH. PHS Publication Number 999-A-40, 1967, pp. 343-347.

EMISSION FACTOR DOCUMENTATION FOR AP-42 SECTION 8.16

Mineral Wool Manufacturing

1. INTRODUCTION

The document Compilation of Air Pollutant Emission Factors (AP-42) has been published by the U. S. Environmental Protection Agency (EPA) since 1972. Supplements to AP-42 have been routinely published to add new emission source categories and to update existing emission factors. AP-42 is routinely updated by EPA to respond to new emission factor needs of EPA, State and local air pollution control programs, and industry.

An emission factor relates the quantity (weight) of pollutants emitted to a unit of activity of the source. The uses for the emission factors reported in AP-42 include:

1. Estimates of areawide emissions;
2. Estimates of emissions for a specific facility; and
3. Evaluation of emissions relative to ambient air quality.

The purpose of this report is to provide background information from test reports and other information to support preparation of AP-42 Section 8.16, Mineral Wool Manufacturing.

This background report consists of five sections. Section 1 includes the introduction to the report. Section 2 gives a description of the mineral wool industry. It includes a characterization of the industry, an overview of the different process types, a description of emissions, and a description of the technology used to control emissions resulting from mineral wool production operations. Section 3 is a review of emission data collection and laboratory analysis procedures. It describes the literature search, the screening of emission data reports, and the quality rating system for both emission data and emission factors. Section 4 details revisions to the existing AP-42 section narrative and pollutant emission factor development. It includes a review of specific data sets and the results of data analyses. Section 5 presents AP-42 Section 8.16, Mineral Wool Manufacturing.

2. INDUSTRY DESCRIPTION^{1,2}

Mineral wool often is defined as any fibrous glassy substance made from minerals (typically natural rock materials such as basalt or diabase) or mineral products such as slag and glass. Because glass wool production is covered separately in AP-42 (Section 8.11), this section deals only with the production of mineral wool from natural rock and slags such as iron blast furnace slag, the primary material, and copper, lead, and phosphate slags. These materials are processed into insulation and other fibrous building materials that are used for structural strength and fire resistance. Generally, these products take one of four forms: "blowing" wool or "pouring" wool, which is put into the structural spaces of buildings; batts, which may be covered with a vapor barrier of paper or foil and are shaped to fit between the structural members of buildings; industrial and commercial products such as high-density fiber felts and blankets, which are used for insulating boilers, ovens, pipes, refrigerators, and other process equipment; and bulk fiber, which is used as a raw material in manufacturing other products, such as ceiling tile, wall board, spray-on insulation, cement, and mortar.

Mineral wool manufacturing facilities are included in Standard Industrial Classification (SIC) Code 3296, mineral wool. This SIC code also includes the production of glass wool insulation products, but those facilities engaged in manufacturing textile glass fibers are included in SIC Code 3229. The six digit source category code (SCC) for mineral wool manufacturing is 3-05-017.

2.1 CHARACTERIZATION OF THE INDUSTRY^{1,3}

Because the U.S. Department of Commerce aggregates the mineral wool manufacturing industry, as defined in this document, into a single SIC category with glass wool manufacturing, industry statistics are difficult to obtain. The available U. S. Government publications do not present information on rock and slag wool production, nor was such information found in the open literature. The most recent data related strictly to rock and slag wool production appear to be those generated by EPA in 1980. These data form the basis for the discussion below.

TABLE 2-1. DISTRIBUTION OF MINERAL WOOL
MANUFACTURING FACILITIES¹

State	No. of facilities	
Alabama	3	4
California	1	0
Colorado	1	0
Illinois	1	✓
Indiana	5	4
Minnesota	2	1
Missouri	2	✓ 0
New Jersey	1	0
North Carolina	1	✓
Ohio	1	✓
Pennsylvania	3	2
Tennessee	1	✓
Texas	3	1
Virginia	1	0
Washington	1	✓

Wisconsin

1

These are the plants in each state
that are currently operating based
on information we have gathered,

①. Dispute?
a. Dispute? ...

As of 1980, approximately 26 mineral wool manufacturing facilities were operating in the United States. Table 2-1 lists the number of facilities by State. These facilities were estimated to have shipped about 2.7×10^5 megagrams (Mg) (3.0×10^5 tons) of structural mineral wool insulation products with a value of about \$100 million during 1980. A growth rate of less than 2 percent per year was projected at that time.

PROCESS DESCRIPTION^{1,4,5}

Most mineral wool produced in the United States today is produced from slag or a mixture of slag and rock. Most of the slag used by the industry is generated by integrated iron and steel plants as a blast furnace byproduct from pig iron production. Other sources of slag include the copper, lead, and phosphate industries. The production process has three primary components—molten mineral generation in the cupola, fiber formation and collection, and final product formation. Figure 2-1 illustrates the mineral wool manufacturing process.

The first step in the process involves melting the mineral feed. The raw material (slag and rock) is loaded into a cupola in alternating layers with coke at weight ratios of about 5 to 6 parts mineral to 1 part coke. As the coke is ignited and burned, the mineral charge is heated to the molten state at a temperature of 1300° to 1650°C (2400° to 3000°F). Combustion air is supplied through tuyeres located near the bottom of the furnace. Process modifications at some plants include oxygen enrichment and the use of natural gas auxiliary burners to reduce coke consumption. One facility also reported using an aluminum flux byproduct to reduce coke consumption.

The molten mineral charge exits the bottom of the cupola in a water-cooled trough and falls onto a fiberization device. Most of the mineral wool produced in the United States is made by variations of two fiberization methods. The Powell process uses groups of rotors revolving at a high rate of speed to form the fibers. Molten material is distributed in a thin film on the surfaces of the rotors and then is thrown off by centrifugal force. As the material leaves the surface, small globules develop and form long, fibrous tails as they travel horizontally. Air or steam may be blown around the rotors to assist in fiberizing the material. A second fiberization method, the Downey process, uses a spinning concave rotor with air or steam attenuation. Molten material is distributed over the

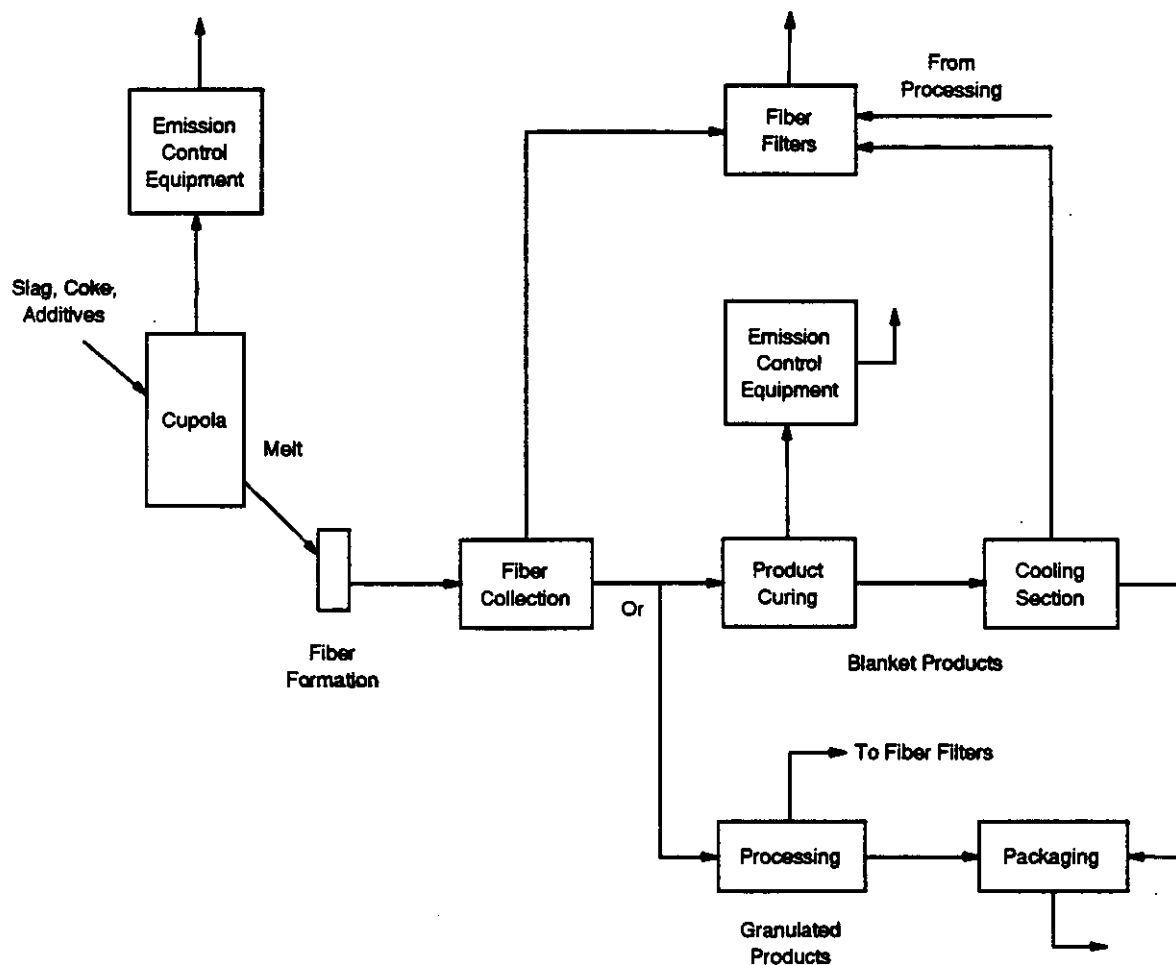


Figure 2-1. Mineral wool manufacturing process flow diagram.

surface of the rotor, from which it flows up and over the edge and is captured by a high-velocity stream of air or steam.

During the spinning process, not all globules that develop are converted into fiber. The nonfiberized globules that remain are referred to as "shot." In raw mineral wool, as much as half of the mass of the product may consist of shot. Shot is usually separated from the wool by gravity immediately following fiberization.

Depending on the desired product, various chemical agents may be applied to the newly formed fiber immediately following the rotor. In almost all cases, an oil is applied to suppress dust and, to some degree, anneal the fiber. This oil can be either a proprietary product or a medium-weight fuel or lubricating oil. If the fiber is intended for use as loose wool or bulk products, no further chemical treatment is necessary. If the mineral wool product is required to have structural rigidity, as in batts and industrial felt, a binding agent is applied with or in place of the oil treatment. This binder is typically a phenol-formaldehyde resin that requires curing at elevated temperatures. Both the oil and the binder are applied by atomizing the liquids and spraying the agents to coat the airborne fiber.

After formation and chemical treatment, the fiber is collected in a blowchamber. Resin- and/or oil-coated fibers are drawn down on a wire mesh conveyor by fans located beneath the collector. The speed of the conveyor is set so that a wool blanket of desired thickness can be obtained.

Mineral wool containing the binding agent is carried by conveyor to a curing oven, where the wool blanket is compressed to the appropriate density and the binder is baked. Hot air, at a temperature of 150° to 320°C (300° to 600°F), is forced through the blanket until the binder has set. Curing time and temperature depend on the type of binder used and the mass rate through the oven. A cooling section follows the oven, where blowers force air at ambient temperatures through the wool blanket.

To make batts and industrial felt products, the cooled wool blanket is cut longitudinally and transversely to the desired size. Some insulation products are then covered with a vapor barrier of aluminum foil or asphalt-coated kraft paper on one side and untreated paper on the other side. The

Loose wool products consist primarily of blowing wool and bulk fiber. For these products, no binding agent is applied, and the curing oven is eliminated. For granulated wool products, the fiber blanket leaving the blowchamber is fed to a shredder and pelletizer. The pelletizer forms small, 1-inch-diameter pellets and separates shot from the wool. A bagging operation completes the processes. For other loose wool products, fiber can be transported directly from the blowchamber to a baler or bagger for packaging.

3) 22 sources of wastewater storage & treatment emissions?
 { binder storage & mixing.

The primary source of emissions in the mineral wool manufacturing process is the cupola. It is a significant source of particulate matter (PM) emissions and is likely to be a source of PM less than 10 micrometers in diameter (PM-10) emissions, although no particle size data are available. Coke combustion in the furnace produces carbon monoxide (CO), carbon dioxide (CO₂), and nitrogen oxide (NO_x) emissions. Finally, because coke and blast furnace slags contain sulfur, the cupola is also a source of sulfur dioxide (SO₂) and H₂S emissions.

✓ Binder application → organic emissions (e.g. phenol, formaldehyde)

2.4 CONTROL TECHNOLOGY¹

Mineral wool manufacturers use a variety of air pollution control techniques, but most are directed toward PM control with minimal control of other pollutants. The industry has given greatest attention to cupola PM control, with two-thirds of the cupolas in operation having fabric filter control systems. Some cupola exhausts are controlled by wet scrubbers and electrostatic precipitators (ESP's); cyclones are also used for cupola PM control either alone or in combination with other control devices. About half of the blow chambers in the industry also have some level of PM control, with the predominant control device being low-energy wet scrubbers. Wire mesh filters also are often used to control PM emissions from blow chambers. Cyclones and fabric filters have been used to a limited degree on blow chambers. Finally, afterburners have been used to control VOC emissions from blow chambers and curing ovens and CO emissions from cupolas. Table 2-2 provides a summary of the extent of control in the industry as of 1980.

TABLE 2-2. SUMMARY OF AIR POLLUTION CONTROLS OPERATING IN
THE U.S. MINERAL WOOL INDUSTRY

Process source	Total	Number of process sources controlled by indicated devices						
		Fabric Filters	ESP	Wet scrubbers	Cyclones	After- burners	Other	None
Cupolas ^a	53	35	2	3	20	2	2 ^b	3
Blowchambers ^c	46	2	0	21	3	2	0	21 ^d
Curing ovens	15	1	0	0	0	6	0	8
Coolers	6	0	0	0	0	1	0	5

^aTwo cupolas are controlled with fabric filters followed by direct-flame afterburners; two cupolas are controlled by wet scrubbers followed by ESP's; seven cupolas are controlled by cyclones followed by fabric filters; and one cupola is controlled by a cyclone followed by a wet scrubber.

^bCarbon monoxide control system is operating on two cupolas with a fabric filter in one plant.

^cThree blowchambers use two control devices in series; two plants use afterburners plus wet scrubbers, and one plant has cyclones plus a fabric filter.

^dIncludes nine units reported to use wire mesh filters.

We'll have updated information from responses to
an industry questionnaire — probably mid-summer
until we have a summary prepared

REFERENCES FOR SECTION 2

1. Source Category Survey: Mineral Wool Manufacturing Industry, EPA-450/3-80-016. U. S. Environmental Protection Agency, Research Triangle Park, NC, March 1980.
2. The Facts on Rocks and Slag Wool, Pub. No. N 020, North American Insulation Manufacturers Association, Alexandria, VA, Undated.
3. ICF Corporation, Supply Response to Residential Insulation Retrofit Demand, Report to the Federal Energy Administration, Contract No. P-14-77-5438-0, Washington, D.C., June 1977.
4. Personal communication between F. May, U.S.G. Corporation, Chicago, Illinois, and R. Marinshaw, Midwest Research Institute, Cary, NC, June 5, 1992.
5. Memorandum from K. Schuster, N.C. Department of Environmental Management, to M. Aldridge, American Rockwool, April 25, 1988.

3. GENERAL DATA REVIEW AND ANALYSIS

3.1 LITERATURE SEARCH AND SCREENING¹⁻³

Data for this investigation were obtained from a number of sources within the Office of Air Quality Planning and Standards (OAQPS) and from outside organizations. The AP-42 Background Files located in the Emission Inventory Branch (EIB) were reviewed for information on the industry, processes, and emissions. The Crosswalk/Air Toxic Emission Factor Data Base Management System (XATEF) and VOC/PM Speciation Data Base Management System (SPECIATE) data bases were searched by SCC code for identification of the potential pollutants emitted and emission factors for those pollutants. A general search of the Air CHIEF CD-ROM also was conducted to supplement the information from these two data bases.

The Minerals Yearbook and Census of Manufactures were reviewed for information on the industry, including number of plants, plant location, and annual production capacities. However, because the data from these sources could not be disaggregated for mineral wool manufacturing, this information was obtained from the Source Category Survey Report. The Aerometric Information Retrieval System (AIRS) data base also was searched for data on the number of plants, plant location, and estimated annual emissions of criteria pollutants.

A number of sources of information were investigated specifically for emission test reports and data. A search of the Test Method Storage and Retrieval (TSAR) data base was conducted to identify test reports for sources within the mineral wool industry. Copies of these test reports were obtained from the files of the Emission Measurement Branch (EMB). The EPA library was searched for additional test reports. A list of plants that have been tested within the past 5 years was compiled from the AIRS data base. State and Regional offices were contacted about the availability of test reports. However, the information obtained from these offices was limited. Publications lists from the Office of Research and Development (ORD) and Control Technology Center (CTC) were also searched for reports on emissions from the mineral wool industry. In addition, representative trade associations, including the North American Insulation Manufacturers Association (NAIMA), were contacted for assistance in obtaining information about the industry and emissions.

To screen out unusable test reports, documents, and information from which emission factors could not be developed, the following general criteria were used:

1. Emission data must be from a primary reference:
 - a. Source testing must be from a referenced study that does not reiterate information from previous studies.
 - b. The document must constitute the original source of test data. For example, a technical paper was not included if the original study was contained in the previous document. If the exact source of the data could not be determined, the document was eliminated.
2. The referenced study must contain test results based on more than one test run.
3. The report must contain sufficient data to evaluate the testing procedures and source operating conditions (e.g., one-page reports were generally rejected).

A final set of reference materials was compiled after a thorough review of the pertinent reports, documents, and information according to these criteria.

3.2 EMISSION DATA QUALITY RATING SYSTEM

As part of the analysis of the emission data, the quantity and quality of the information contained in the final set of reference documents were evaluated. The following data were excluded from consideration:

1. Test series averages reported in units that cannot be converted to the selected reporting units;
2. Test series representing incompatible test methods (i.e., comparison of EPA Method 5 front half with EPA Method 5 front and back half);
3. Test series of controlled emissions for which the control device is not specified;

4. Test series in which the source process is not clearly identified and described; and

5. Test series in which it is not clear whether the emissions were measured before or after the control device.

Test data sets that were not excluded were assigned a quality rating. The rating system used was that specified by EIB for preparing AP-42 sections. The data were rated as follows:

A--Multiple tests that were performed on the same source using sound methodology and reported in enough detail for adequate validation. These tests do not necessarily conform to the methodology specified in EPA reference test methods, although these methods were used as a guide for the methodology actually used.

B--Tests that were performed by a generally sound methodology but lack enough detail for adequate validation.

C--Tests that were based on an untested or new methodology or that lacked a significant amount of background data.

D--Tests that were based on a generally unacceptable method but may provide an order-of-magnitude value for the source.

The following criteria were used to evaluate source test reports for sound methodology and adequate detail:

1. Source operation. The manner in which the source was operated is well documented in the report. The source was operating within typical parameters during the test.

2. Sampling procedures. The sampling procedures conformed to a generally acceptable methodology. If actual procedures deviated from accepted methods, the deviations are well documented. When this occurred, an evaluation was made of the extent to which such alternative procedures could influence the test results.

3. Sampling and process data. Adequate sampling and process data are documented in the report, and any variations in the sampling and process operation are noted. If a large spread between test results cannot be explained by information contained in the test report, the data are suspect and are given a lower rating.

4. Analysis and calculations. The test reports contain original raw data sheets. The nomenclature and equations used were compared to those (if any) specified by EPA to establish equivalency. The depth of review of the calculations was dictated by the reviewer's confidence in the ability and conscientiousness of the tester, which in turn was based on factors such as consistency of results and completeness of other areas of the test report.

3.3 EMISSION FACTOR QUALITY RATING SYSTEM ¹

The quality of the emission factors developed from analysis of the test data was rated using the following general criteria:

A--Excellent: Developed only from A-rated test data taken from many randomly chosen facilities in the industry population. The source category is specific enough so that variability within the source category population may be minimized.

B--Above average: Developed only from A-rated test data from a reasonable number of facilities. Although no specific bias is evident, it is not clear if the facilities tested represent a random sample of the industries. The source category is specific enough so that variability within the source category population may be minimized.

C--Average: Developed only from A- and B-rated test data from a reasonable number of facilities. Although no specific bias is evident, it is not clear if the facilities tested represent a random sample of the industry. In addition, the source category is specific enough so that variability within the source category population may be minimized.

D--Below average: The emission factor was developed only from A- and B-rated test data from a small number of facilities, and there is reason to suspect that these facilities do not represent a

random sample of the industry. There also may be evidence of variability within the source category population. Limitations on the use of the emission factor are noted in the emission factor table.

E--Poor: The emission factor was developed from C- and D-rated test data, and there is reason to suspect that the facilities tested do not represent a random sample of the industry. There also may be evidence of variability within the source category population. Limitations on the use of these factors are always noted.

The use of these criteria is somewhat subjective and depends to an extent upon the individual reviewer. Details of the rating of each candidate emission factor are provided in Chapter 4 of this report.

REFERENCES FOR SECTION 3

1. 1987 Census of Manufacturers, Industry Series-Abrasives, Asbestos and Miscellaneous Nonmetallic Mineral Products, U.S. Department of Commerce, Washington, D.C., May 1990.
2. Minerals Yearbook, Vol. I--Metals and Minerals, 1989, U.S. Department of the Interior, Bureau of Mines, Washington, D.C., 1991.
3. Source Category Survey: Mineral Wool Manufacturing Industry, EPA-450/3-80-016, U. S. Environmental Protection Agency, Research Triangle Park, NC, March 1980.
4. Technical Procedures for Developing AP-42 Emission Factors and Preparing AP-42 Sections (Draft), Office of Air Quality Planning and Standards, U. S. Environmental Protection Agency, Research Triangle Park, NC, March 6, 1992.

4. AP-42 SECTION DEVELOPMENT

4.1 REVISION OF SECTION NARRATIVE¹

Section 8.16, Mineral Wool Manufacturing, was last revised in 1972. The narrative provided in that version was quite limited, and the discussion of emissions and emission controls provided almost no information. Consequently, the narrative was completely rewritten for this version. The draft section, which is based primarily on information presented in the Source Category Survey Report and in test reports reviewed as a part of this study, contains an expanded discussion of the process, emissions, and emission controls and provides a process flow diagram.

4.2 POLLUTANT EMISSION FACTOR DEVELOPMENT

In addition to a review of the data available in the background file for Section 8.16, this evaluation included an examination of the emission data contained in the Source Category Survey Report and reviews of nine emission test reports. All tests described in these nine reports were conducted by facilities to demonstrate compliance with State or local regulations. The tests documented in References 8, 9, and 10 were conducted at the same facility. However, no process data were provided for these tests. In addition, the two stacks that were sampled served several emission sources, including cupolas, fugitive dust collection systems, a curing oven, and pipe manufacturing machines. (Based on exhaust gas flow rates provided, the cupolas accounted for 5 to 8 percent of the total flow exiting the stacks.) For these reasons, these three references were not used to develop emission factors.

The remainder of this section is divided into five parts. First the data presented in the Source Category Survey Report are discussed. Then the six test reports that contain sufficient data for emission factor development are discussed individually. Emission factors for mineral wool manufacturing included in the XATEF and SPECIATE data bases were also reviewed, and a discussion of these emission factors is presented. Then a discussion of the review of the existing test data in the AP-42 background file is presented. Finally, the results of the data review and analysis are presented.

4.2.1 Review of Source Category Survey Data (Reference 1)

As part of a review of the mineral wool manufacturing industry to assess the need for a new source performance standard, EPA compiled a substantial amount of emission data from State and local agencies. Because the data were only presented in summary form, their quality cannot be evaluated. Consequently, they cannot be averaged with other available test data to obtain emission factors. In view of these limitations, the emission factors developed from these data were deemed to be useful for order of magnitude estimates only and are rated E. Table 4-1 summarizes the information on uncontrolled emission factors for mineral wool cupolas. For each pollutant, the table shows the number of tests reviewed during the study and a range and average emission factor. Table 4-2 summarizes uncontrolled PM emission factor information for blow chambers. Finally, one test on a cupola in San Bernadino County, California, generated particle size data that were obtained with an Andersen cascade impactor. These data are presented in Table 4-3.

4.2.2 Review of Specific Data Sets

4.2.2.1 Reference 2. This test was sponsored by the facility in 1988 to demonstrate that SO₂ emissions from the Nos. 1 and 2 cupolas were in compliance with State requirements. While the primary purpose of the test was to measure SO₂ levels, sufficient data were obtained from the associated Method 2 and 3 tests to calculate CO and CO₂ emission factors. The SO₂ measurements were made with a Standard EPA Method 8 train; sulfur trioxide (SO₃) measurements also were obtained from this train. Volumetric flow rates were obtained via EPA Method 2, and CO and CO₂ concentrations were obtained from Orsat measurements per EPA Method 3.

The process information contained in the test report was quite sparse. In fact, the only data that were available in the test report were process rate data sheets, which were contained in an appendix. Subsequently, the State agency supplied a process flow diagram for the facility. The information contained in the process flow diagram and in the process data appendix indicated that emissions from each cupola were controlled by a fabric filter, but no design or operating data on the system are available. During the tests, cupola No. 1 fired a mixture of coke (~ 15 percent) and slag

**TABLE 4-1. SUMMARY OF SOURCE CATEGORY SURVEY EMISSION
FACTOR DATA FOR UNCONTROLLED MINERAL WOOL CUPOLAS**

		Emission factor			
		kg/Mg feed		lb/ton feed	
Pollutant	No. of tests	Range	Average	Range	Average
PM	3	2.3-6.8	5.3	4.6-13.7	10.6
SO ₂	10	NA	5.3	NA	10.6
H ₂ S	3	NA	1.5	NA	3.0
CO	9	3-156	78	6-312	156
NO _x	6	0.1-1.9	0.8	0.2-3.7	1.6

NA = not available.

**TABLE 4-2. SUMMARY OF SOURCE CATEGORY SURVEY EMISSION
FACTOR DATA FOR UNCONTROLLED MINERAL WOOL BLOW CHAMBERS**

		Emission factor			
		kg/Mg feed		lb/ton feed	
Source	No. of tests	Range	Average	Range	Average
PM	2	0.7 -0.9	0.8	1.4-1.8	1.6
VOC's	2	NA	0.2	NA	0.4

**TABLE 4-3. SOURCE CATEGORY PARTICLE SIZE DATA FOR
UNCONTROLLED MINERAL WOOL CUPOLAS**

Particle size range, μ m	Percent by weight
+30	5.6
9.2-30	0.1
5.5-9.2	0.5
3.3-5.5	1.0
2.0-3.3	5.0
1.0-2.0	67.8
0.2-1	20.0

(~ 85 percent), while cupola No. 2 fired a mixture of coke (~ 15 percent), slag (~ 80 percent), and ore (~ 5 percent).

The data are rated A for CO, CO₂, and SO₂ because standard methodology was used, no problems were reported, and all results were fully documented. Unlike these gaseous pollutants (CO, CO₂, and SO₂), which generally are not controlled by fabric filters, SO₃ is emitted as PM, and, thus, would be controlled by a fabric filter. Because the report did not include adequate information on the design and operation of the fabric filter, the SO₃ data are rated B.

4.2.2.2 Reference 3. This test program was sponsored by the facility in January 1981 to demonstrate that PM emissions from the cupola complied with State emission limits. The PM measurements were made on each operation at the outlet to an air pollution control device using EPA Method 5. Fyrite was used to quantify CO₂ emissions. Three runs were completed on the blow chamber; four runs were conducted on the cupola, but one was declared invalid because of sampling equipment problems. The results from that run were not reported.

The process information contained in the test report was limited to process data sheets contained in the appendix. However, the State agency provided flow diagrams indicating that the cupola was controlled by a fabric filter and the blowchamber was controlled by a wire mesh filter. No other information is available on the process.

The PM test data from this report are rated B. Tests were conducted with standard EPA methods, and no problems were reported. However, the process information contained in the report was insufficient to characterize the processes or control systems adequately. The CO₂ data are rated C due to the relative inaccuracy of the Fyrite analysis.

4.2.2.3 Reference 4. This test program was sponsored in June 1979 by the facility to demonstrate that the PM emissions from cupolas Nos. 1, 2, and 3 complied with State emission limits. Some data also were collected on organic emissions from the blow chamber. The sampling train used to collect the hydrocarbons included a heated glass probe with glass wool plug to collect the PM, followed by two tubes filled with activated charcoal. The samples were analyzed by placing carbon disulfide in the activated carbon tubes for 24 hours, then filtering and evaporating the liquid to dryness at room temperature. However, it is likely that a significant amount of sample was lost in the

evaporation step. For that reason, the test method used was not considered to be acceptable for AP-42 emission factor development, and the results are not included in this review. The PM tests were conducted with EPA Methods 1 through 5, and no problems were noted. Fyrite was used to quantify CO₂ emissions.

The process information for the cupolas is limited to a process flow diagram supplied by the State agency and process data sheets contained in the report appendix. The process diagrams indicate that each cupola is equipped with a fabric filter. During the tests, the process data indicated that the cupola was fired with a blend of coke (~10.5 percent), shale (~6.4 percent), slag (probably blast furnace slag) (~62.3 percent), and phosphate slag (~20.8 percent). Some process data were supplied on the blow chamber operation, but they were insufficient to determine the basis for the process weights associated with these operations.

The PM test data are rated B. Tests were conducted with standard EPA methods and no problems were reported. However, the process information was inadequate to warrant a higher rating. The CO₂ data are rated C due to the relative inaccuracy of Fyrite analysis.

4.2.2.4 Reference 5. This facility-sponsored test was conducted to demonstrate that the PM emissions from the batt curing oven complied with State emission limits. Five test runs were conducted using EPA Methods 1 through 5. Run 1 was discarded because of a failed posttest leak check, and Run 2 was discontinued because of a process malfunction, leaving three valid runs. The report does not provide process or emission data for the two discarded runs. Fyrite was used to quantify CO₂ emissions.

The process information in the report is quite limited. The introduction does note that emissions are directed through an ESP, but no other process description is provided. Operational data are presented in Chapter IV of the report. However, these data are difficult to read, and the raw data could not be clearly related to the process weights presented in summary tables. The process weights appear to be in units of batt produced, but the exact basis for the process weights could not be confirmed from the raw data.

The test data from this report are rated C. Tests were conducted with standard EPA methods, and no problems were reported. However, the process information contained in the report was

insufficient to characterize the processes or control systems adequately. Also, the basis for the process rates given in summary tables is unclear.

4.2.2.5 Reference 6. This test program was conducted by the facility to measure emissions of PM, SO₂, and fluorides. The tests were designed to evaluate the effect of substituting an aluminum smelting cell byproduct material (SPL) for coke on a pound-per-pound carbon basis. The typical charge compositions for the different test conditions are shown below.

AVERAGE CHARGE MAKEUP

Charge (1 lb)	Condition A	Condition B	Condition D
SPL ^a	0	210	450
Lime	0	0	50
Coke	385	260	140
Duquesne slag	1,300	1,300	1,300
Trap rock	1,100	1,100	1,100
Steel slag	400	400	400
Tennessee slag	400	400	400
Avg. No. charges/hr	3.5	3.4	3.0

The test design for this program was somewhat unusual. The facility operates two cupolas, each with its own spark arrestor and fabric filter. The exhaust from the fabric filters is combined and ducted to the atmosphere through a common stack. The sampling was conducted in this common stack. Sampling for fluorides was conducted using Alcoa Method 4075A (which was approved by the State and EPA) in conjunction with EPA Methods 1 through 4. Particulate matter emissions were obtained from a cellulose thimble in the front half of the Method 4075A. This procedure provides results that are comparable to EPA Method 5 front half results but are less accurate for emissions that include significant levels of condensible PM. However, for the reported stack gas temperatures, which ranged from 34° to 44°C (93° to 111°F), the condensible PM fraction should be negligible. Therefore, the filterable PM results should be relatively accurate for AP-42 emission factor development with a one-step quality down-rating. The SO₂ samples were obtained with a glass-bulb technique that is purported to be similar to EPA Method 15 procedures. The concentration of SO₂ was measured with a gas chromatograph/flame photometric detector. Although the test method appears to be acceptable, there is inadequate information to evaluate the validity of the analytical

method used or to demonstrate that this method is equivalent to EPA Method 6 or 8. Consequently, the SO₂ data are rated D. Fyrite was used to quantify CO₂ emissions.

The test data for fluorides are rated A. Reference or equivalent methods were used, no problems were reported, and results were fully documented. The PM data were rated B because the method used is somewhat less reliable than EPA Method 5. The SO₂ data were rated D because a nonstandard method was used and no information was presented on its reliability, accuracy, precision, or equivalence to other methods. The CO₂ data are rated C due to the relative inaccuracy of Fyrite analysis.

4.2.2.6 Reference 7. This test program was sponsored by the facility to demonstrate that PM, SO₂, and fluoride emissions from the cupola were in compliance with State requirements. The tests were conducted in the common stack for the two cupolas as described in Reference 6 above. Tests were conducted with two different charge conditions as shown below.

CHARGE MAKEUP

Baseline		710 lb/hr SPL	
Charge	lb	Charge	lb
Coke	385-400	Coke	260
SPL	0	SPL	225
Duquesne slag	1,300	Duquesne slag	1,300
Trap rock	1,100	Trap rock	1,100
Lime	50	Lime	50
Steel	400	Steel	400
Tennessee slag	400	Tennessee slag	400

Three test runs were completed for the baseline conditions, and four were completed for the SPL runs. Standard EPA methods were used for PM (Methods 1 through 5) and SO₂ (Method 6). An Alcoa method (Method 4075A) that was approved by the State and EPA was used for fluorides. Fyrite was used to quantify CO₂ emissions.

The PM, SO₂, and fluoride test data from this report are rated A. Standard methods or acceptable equivalents were used, no problems were reported, and the test report fully documented results. The CO₂ data are rated C due to the relative inaccuracy of Fyrite analysis.

4.2.3 Review of XATEF and SPECIATE Data Base Emission Factors

The XATEF data base does not contain emission factors for mineral wool manufacturing.

The SPECIATE data base contains emission factors for emissions from mineral wool furnaces, curing ovens, and coolers. However, all of the emission factors are based on surrogate profiles. Consequently, they will not be used in the revised AP-42 section.

4.2.4 Review of Test Data in AP-42 Background Files

The current version of AP-42 contains uncontrolled PM emission factors for the cupola, reverberatory furnace, blow chamber, curing oven, and cooler and an uncontrolled SO₂ emission factor for the cupola. A review of the background file indicated that these emission factors are based on averaging a limited quantity of emission data that were reported in an early (1967) version of AP-40 (Reference 11). In addition, Reference 11 includes emission data on uncontrolled emissions of SO₃ and CO from cupolas; SO₂ and aldehydes from blow chambers; SO₂, nitrogen dioxide (NO₂), and aldehydes from curing ovens; and SO₂ and aldehydes from coolers. This reference reported average rather than run-specific test results, and the test methods were not documented. Given these limitations, the emission factors developed from these data were deemed to be useful for order of magnitude estimates only. The emission factors developed from these data are rated E, with the exception of the emission factors for aldehydes. Because the lack of documentation on the aldehyde emission tests and the fact that a reliable method for testing aldehydes was not available at the time of these tests, the aldehyde emission results are highly suspect and are unrated.

4.2.5 Results of Data Analysis

For mineral wool manufacturing cupolas, the test reports and documents described above provided sufficient data to develop emission factors for uncontrolled and controlled filterable PM emissions; uncontrolled CO, CO₂, SO₂, H₂S, and NO_x emissions; uncontrolled and controlled SO₃

OK
Not sure if this furnace is still in operation

emissions; and controlled fluoride emissions. For reverberatory furnaces, an uncontrolled filterable PM emission factor was developed. For mineral wool batt curing ovens, emission factors were developed for uncontrolled and controlled filterable PM emissions and for uncontrolled SO₂ and N₂O emissions. For mineral wool blow chambers, emission factors were developed for uncontrolled and controlled filterable PM emissions and for uncontrolled SO₂ and CO₂ emissions. Finally, for mineral wool coolers, emission factors were developed for filterable PM and SO₂ emissions. The data used in the analysis are summarized in Tables 4-4 and 4-5. Table 4-6 summarizes the emission factors developed from data found in AP-40 (Reference 11). The final emission factors that were incorporated into the revised AP-42 section and their ratings are tabulated in Table 4-7. The paragraphs below describe how the emission factors were calculated and summarize the rationale for the ratings.

The filterable PM emission factor for cupola emissions was developed by averaging the data in the source category survey report (Reference 1) and AP-40 Reference 11. When compared to the fabric-filter-controlled data from References 3, 4, 6, and 7, the uncontrolled PM data indicate a control efficiency of 99 percent. Thus, although the emission factor is based on secondary references, the uncontrolled data are consistent with the controlled data from primary references. Because the emission factor is based on secondary data, it is rated E.

The emission factors included in the revised AP-42 section on mineral wool manufacturing for uncontrolled CO and CO₂ emissions from cupolas were developed from Reference 2. Emission factors for CO₂ emissions from cupolas also were developed from References 3, 4, 6, and 7. The Reference 2 data are rated A and indicated an emission factor of 125 kg/Mg (250 lb/ton), and the data from the other four references are rated C and average 205 kg/Mg (410 lb/ton). However, emission factors developed from C-rated data can only be rated E. For that reason, the emission factor for CO₂ emissions from cupolas developed from Reference 2 was used.

Although the CO and CO₂ emissions were measured downstream from a fabric filter, these emission factors are considered to be uncontrolled because fabric filters are not expected to affect CO and CO₂ emissions. These emission factors are rated D, because they are based on A- and B-rated data from only one plant.

**TABLE 4-4. SUMMARY OF EMISSION TEST DATA FOR MINERAL
WOOL MANUFACTURING CUPOLAS**

Facility	Source ID	APCD	Pollutant	No. of runs	Data rating	Emission factor, kg/Mg (lb/ton) feed		Ref.
						Range	Average	
A	Cupola No. 1 (slag)	FF	CO	3	A	130-140 (260-280)	130 (260)	2
			SO ₂	3	A	2.5-3.8 (5.1-7.6)	3.1 (6.2)	2
			CO ₂	3	A	230-270 (470-550)	250 (510)	2
			SO ₃	3	B	0.0085-0.43 (0.017-0.86)	0.15 (0.30)	2
	Cupola No. 2 (slag/ore)	FF	CO	3	A	120-130 (230-260)	120 (250)	2
			SO ₂	3	A	3.3-3.8 (6.5-7.6)	3.4 (6.9)	2
			CO ₂	3	A	250-310 (490-620)	270 (540)	2
			SO ₃	3	B	0.0010-0.0050 (0.0040-0.010)	0.0034 (0.0067)	2
B	Cupola No. 4 (slag)	FF	PM	3	B	0.0020-0.049 (0.0041-0.098)	0.025 (0.050)	3
		None	CO ₂	3	C	310-350 (610-690)	330 (650)	3
	Cupola No. 1	FF	PM	3	B	0.015-0.072 (0.030-0.14)	0.041 (0.082)	4
		None	CO ₂	3	C	150-290 (290-580)	220 (430)	4
	Cupola No. 2	FF	PM	3	B	0.016-0.053 (0.033-0.11)	0.032 (0.065)	4
		None	CO ₂	3	C	110-200 (210-390)	150 (290)	4
	Cupola No. 3 (slag, shale, phosphate slag)	FF	PM	3	B	0.037-0.073 (0.073-0.15)	0.050 (0.099)	4
		None	CO ₂	3	C	200-250 (390-500)	230 (450)	4
C	Cond. 1 ^a	FF	PM	3	B	0.035-0.057 (0.069-0.11)	0.049 (0.098)	6
			SO ₂	3	D	3.7-4.5 (7.4-8.9)	4.2 (8.3)	6
			Fluorides	3	A	0.016-0.036 (0.031-0.072)	0.029 (0.058)	6
		None	CO ₂	3	C	160-170 (320-330)	170 (330)	6
	Cond. 2	None	PM	4	B	0.019-0.095 (0.038-0.19)	0.074 (0.15)	6
			SO ₂	4	D	1.9-4.1 (3.8-8.1)	3.8 (7.6)	6
			Fluorides	4	A	0.059-0.49 (0.19-0.98)	0.32 (0.63)	6
		None	CO ₂	4	C	100-180 (200-360)	120 (240)	6
	Cond. 3	None	PM	2	B	0.076-0.079 (0.15-0.16)	0.18 (0.37)	6
			SO ₂	2	D	2.0-2.9 (4.1-5.8)	2.5 (5.0)	6
			Fluorides	2	A	0.19-0.26 (0.37-0.51)	0.22 (0.44)	6
		None	CO ₂	2	C	110-170 (220-340)	140 (280)	6
C	Cond. 1 ^a	FF	PM	3	A	0.061-1.1 (0.12-2.2)	0.084 (0.17)	7
			SO ₂	3	A	5.0-6.5 (10-13)	5.5 (11.0)	7
			Fluorides	3	A	0.0073-0.011 (0.015-0.021)	0.0085 (0.017)	7
		None	CO ₂	3	C	240-260 (480-510)	250 (500)	7
	Cond. 2	None	PM	4	A	0.12-0.17 (0.24-0.33)	0.14 (0.28)	7
			SO ₂	4	A	3.7-4.5 (7.5-9.0)	4.1 (8.2)	7
			Fluorides	4	A	0.026-0.036 (0.052-0.073)	0.032 (0.064)	7
		None	CO ₂	3	C	230-240 (460-480)	240 (470)	7

^aRefer to Section 4.2.2.5 for composition of charge material.

**TABLE 4-5. SUMMARY OF EMISSION TEST DATA FOR MINERAL WOOL
MANUFACTURING CURING AND BLOWING**

Facility	Source ID	APCD	Pollutant	No. of runs	Data rating	Emission factor, kg/Mg (lb/ton) feed		Ref.
						Range	Average	
B	Batt curing oven	ESP	PM	3	C	0.23-0.60 (0.46-1.2)	0.36 (0.72)	5
		None	CO ₂	3	C	60-110 (110-220)	80 (160)	5
B	Blow chamber	Wire mesh filter	PM	3	B	0.30-5.9 (0.59-1.2)	0.45 (0.91)	3

TABLE 4-6. SUMMARY OF UNCONTROLLED EMISSION FACTORS
DEVELOPED FROM AP-40^{11,a}

Process	Pollutant	No. of tests	Range		Average	
			kg/Mg	lb/ton	kg/Mg	lb/ton
Cupola	PM (filterable)	3	8.0-14	16-28	11	22
	SO ₂	1			5.6	11
	SO ₃	1			3.2	6.3
	CO	1			45	91
Reverberatory furnace	PM (filterable)	1			2.4	4.8
Blow chamber	PM (filterable)	4	2.0-28	4.0-56	8.6	17
	SO ₂	1			0.43	0.87
	aldehydes ^b	1			0.43	0.86
Curing oven	PM (filterable)	5	0.74-2.9	1.5-5.9	1.8	3.6
	SO ₂	1			0.58	1.2
	aldehydes ^b	2	0.37-0.63	0.73-1.3	0.50	1.00
	NO ₂	2	0.043-0.12	0.086-0.23	0.079	0.16
Cooler	PM (filterable)	4	0.21-2.8	0.43-5.5	1.2	2.4
	SO ₂	1			0.034	0.068
	aldehydes ^b	1			0.021	0.042

^aAll emission factors rated E except where indicated.

^bEmission factors are unrated.

TABLE 4-7. MINERAL WOOL MANUFACTURING EMISSION FACTORS

Source	Control	Pollutant	Emission factor		Rating	Ref. No.
			kg/Mg	lb/ton		
Cupola ^a	Uncontrolled	Filterable PM	8.2	16	E	1, 11
		CO	125	250	D	2
		CO ₂	260	520	D	2
		SO ₂	4.0	8.0	D	2,7
		SO ₃	3.2	6.3	E	11
		H ₂ S	1.5	3.0	E	1
		NO ₂	0.8	1.6	E	1
	Fabric filter	Filterable PM	0.051	0.10	D	2,3,4,6,7
		SO ₃	0.077	0.15	E	2
		Fluorides ^b	0.019	0.038	D	7
		Fluorides ^c	0.19	0.38	D	6
Reverberatory Furnace	Uncontrolled	Filterable PM	2.4	4.8	E	11
Batt curing oven ^d	Uncontrolled	Filterable PM	1.8	3.6	E	11
		SO ₂	0.58	1.2	E	11
		NO ₂	0.079	0.16	E	11
		CO ₂	80	160	E	5
	ESP	Filterable PM	0.36	0.72	D	5
Blow chamber ^e	Uncontrolled	Filterable PM	6.0	12	E	1,11
		SO ₂	0.43	0.87	E	11
	Wire mesh filter	Filterable PM	0.45	0.91	D	3
Cooler	Uncontrolled	Filterable PM	1.2	2.4	E	11
		SO ₂	0.034	0.068	E	11

^aActivity level is total feed-charged.^bOnly coke was used as fuel.^cFuel was a combination of coke and aluminum smelting byproducts.^dActivity level is mass of product.^eActivity level is mass of molten mineral feed.

For SO₂ emissions from cupolas, A- or B-rated data are available for three operating units--cupola Nos. 1 and 2 at Facility A and the combined stream at Facility C under different operating conditions. Examination of the data shows the data to fall in a reasonably narrow range. Consequently, the emission factor was obtained by simply averaging the four A- and B-rated test data. This emission factor is rated D because of the limited number of tests and facilities used.

Uncontrolled emission factors for SO₃, H₂S, and NO_x emissions from cupolas were developed from secondary references (References 1 and 11). The uncontrolled SO₃ emission factor for cupolas, which was developed from Reference 11, indicates a fabric filter control efficiency of 98 percent when compared to the data in Reference 2. For that reason, the uncontrolled SO₃ emission factor also appears to be reasonable. The uncontrolled emission factors for H₂S and NO_x also were developed from Reference 1. However, there are no controlled data to which these emission factors can be compared.

The controlled filterable PM emission factor for cupolas was obtained by averaging the average of the data from five tests at Facility C with the data from tests on the four cupolas at Facility B. Again, the emission factor is rated D because it is based on a very limited quantity of data. The remaining emission factors developed for mineral wool cupolas are for controlled SO₃ and fluoride emissions. The SO₃ emission factor is based on two B-rated tests at the same facility (Reference 2) and is rated D. Fluoride emission factors for two types of fuel were developed from one test each and are also rated D.

The emission factors for controlled filterable PM emissions from batt curing ovens and blow chambers were developed from References 5 and 3, respectively. These emission factors are each rated D, because they are based on a single emission test. All other emission factors for mineral wool manufacturing are based on secondary references and are rated E.

The uncontrolled filterable PM emission factor for blow chambers is based on an average of the data in References 1 and 11. This emission factor, when compared to the wire mesh filter-controlled emission factor developed from Reference 3, indicates a control efficiency of 93 percent, which seems reasonable. Also, a comparison of the uncontrolled (based on secondary data) and ESP-controlled PM emission factors for curing ovens indicates a control efficiency of 80 percent, which also appears to be reasonable. There are no other data with which the other emission factors

developed from secondary data can be compared. However, they are considered to be useful for order-of-magnitude estimates and have been included in the revised AP-42 section.

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3. Particulate Emissions Tests for U.S. Gypsum Company on the Number 4 Dry Filter and Cupola Stack Located in Birmingham, Alabama on January 14, 1981, Guardian Systems, Inc., Birmingham, AL, Undated.
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5. DRAFT AP-42 SECTION 8.16

8.16 MINERAL WOOL MANUFACTURING

8.16.1 General^{1,2}

Mineral wool often is defined as any fibrous glassy substance made from minerals (typically natural rock materials such as basalt or diabase) or mineral products such as slag and glass. Because glass wool production is covered separately in AP-42 (Section 8.11), this section deals only with the production of mineral wool from natural rock and slags such as iron blast furnace slag, the primary material, and copper, lead, and phosphate slags. These materials are processed into insulation and other fibrous building materials that are used for structural strength and fire resistance. Generally, these products take one of four forms: "blowing" wool or "pouring" wool, which is put into the structural spaces of buildings; batts, which may be covered with a vapor barrier of paper or foil and are shaped to fit between the structural members of buildings; industrial and commercial products such as high-density fiber felts and blankets, which are used for insulating boilers, ovens, pipes, refrigerators, and other process equipment; and bulk fiber, which is used as a raw material in manufacturing other products, such as ceiling tile, wall board, spray-on insulation, cement, and mortar.

Mineral wool manufacturing facilities are included in Standard Industrial Classification (SIC) Code 3296, mineral wool. This SIC code also includes the production of glass wool insulation products, but those facilities engaged in manufacturing textile glass fibers are included in SIC Code 3229. The six digit source category code (SCC) for mineral wool manufacturing is 3-05-017.

8.16.2 Process Description^{1,4,5}

Most mineral wool produced in the United States today is produced from slag or a mixture of slag and rock. Most of the slag used by the industry is generated by integrated iron and steel plants as a blast furnace byproduct from pig iron production. Other sources of slag include the copper, lead, and phosphate industries. The production process has three primary components--molten

mineral generation in the cupola, fiber formation and collection, and final product formation. Figure 8.16-1 illustrates the mineral wool manufacturing process.

The first step in the process involves melting the mineral feed. The raw material (slag and rock) is loaded into a cupola in alternating layers with coke at weight ratios of about 5 to 6 parts mineral to 1 part coke. As the coke is ignited and burned, the mineral charge is heated to the molten state at a temperature of 1300° to 1650°C (2400° to 3000°F). Combustion air is supplied through tuyeres located near the bottom of the furnace. Process modifications at some plants include air enrichment and the use of natural gas auxiliary burners to reduce coke consumption. One facility also reported using an aluminum flux byproduct to reduce coke consumption.

The molten mineral charge exits the bottom of the cupola in a water-cooled trough and falls onto a fiberization device. Most of the mineral wool produced in the United States is made by variations of two fiberization methods. The Powell process uses groups of rotors revolving at a high rate of speed to form the fibers. Molten material is distributed in a thin film on the surfaces of the rotors and then is thrown off by centrifugal force. As the material is discharged from the rotor, small globules develop on the rotors and form long, fibrous tails as they travel horizontally. Air or steam may be blown around the rotors to assist in fiberizing the material. A second fiberization method, the Downey process, uses a spinning concave rotor with air or steam attenuation. Molten material is distributed over the surface of the rotor, from which it flows up and over the edge and is captured and directed by a high-velocity stream of air or steam.

During the spinning process, not all globules that develop are converted into fiber. The nonfiberized globules that remain are referred to as "shot." In raw mineral wool, as much as half of the mass of the product may consist of shot. As shown in Figure 8.16-1, shot is usually separated from the wool by gravity immediately following fiberization.

Depending on the desired product, various chemical agents may be applied to the newly formed fiber immediately following the rotor. In almost all cases, an oil is applied to suppress dust and, to some degree, anneal the fiber. This oil can be either a proprietary product or a medium-weight fuel or lubricating oil. If the fiber is intended for use as loose wool or bulk products, no further chemical treatment is necessary. If the mineral wool product is required to have structural rigidity, as in batts and industrial felt, a binding agent is applied with or in place of the oil treatment. This binder is

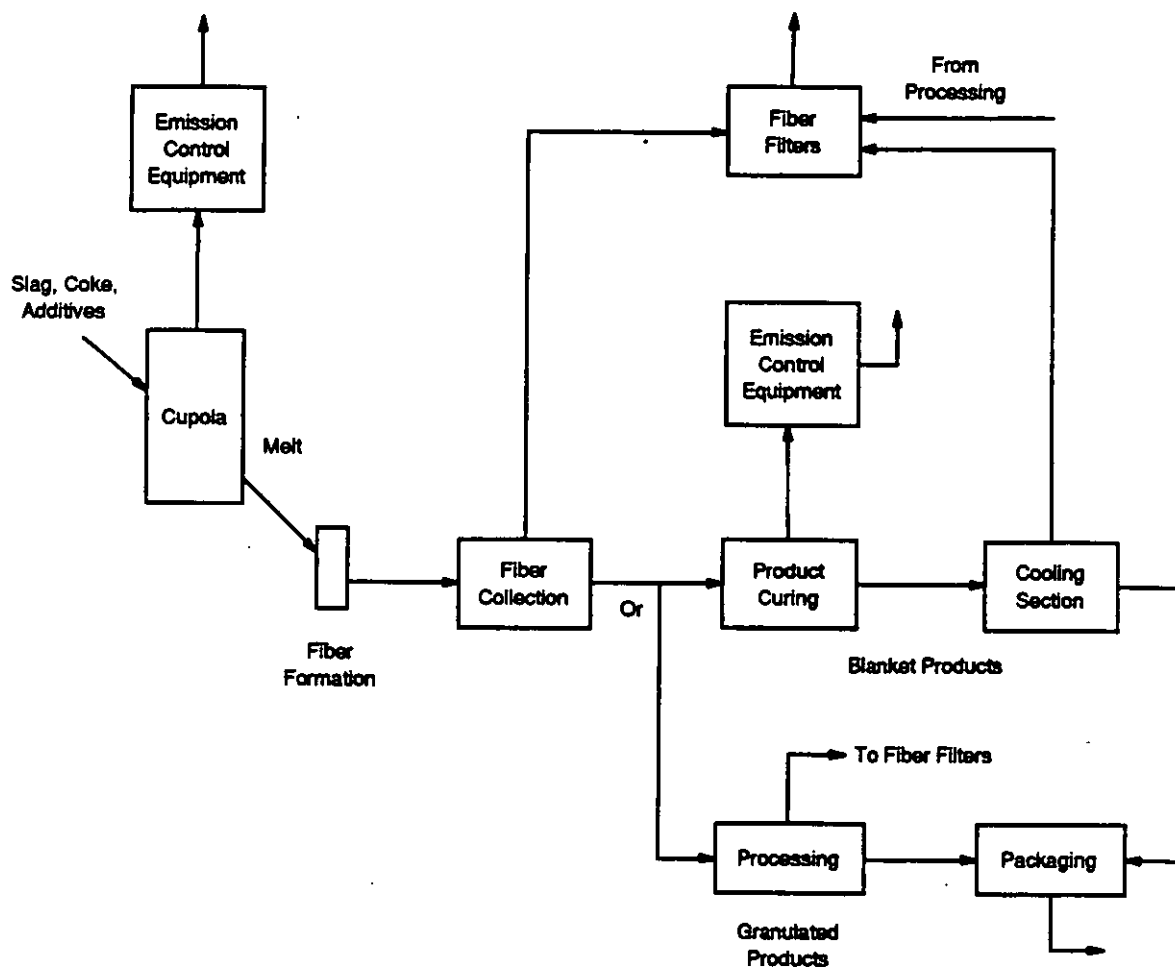


Figure 8.16-1. Mineral wool manufacturing process flow diagram.

typically a phenol-formaldehyde resin that requires curing at elevated temperatures. Both the oil and the binder are applied by atomizing the liquids and spraying the agents to coat the airborne fiber.

After formation and chemical treatment, the fiber is collected in a blowchamber. Resin-and/or oil-coated fibers are drawn down on a wire mesh conveyor by fans located beneath the collector. The speed of the conveyor is set so that a wool blanket of desired thickness can be obtained.

Mineral wool containing the binding agent is carried by conveyor to a curing oven, where the wool blanket is compressed to the appropriate density and the binder is baked. Hot air, at a temperature of 150° to 320°C (300° to 600°F), is forced through the blanket until the binder has set. Curing time and temperature depend on the type of binder used and the mass rate through the oven. A cooling section follows the oven, where blowers force air at ambient temperatures through the wool blanket.

To make batts and industrial felt products, the cooled wool blanket is cut longitudinally and transversely to the desired size. Some insulation products are then covered with a vapor barrier of aluminum foil or asphalt-coated kraft paper on one side and untreated paper on the other side. The cutters, vapor barrier applicators, and conveyors are sometimes referred to collectively as a batt machine. Those products that do not require a vapor barrier, such as industrial felt and some residential insulation batts, can be packed for shipment immediately after cutting.

Loose wool products consist primarily of blowing wool and bulk fiber. For these products, no binding agent is applied, and the curing oven is eliminated. For granulated wool products, the fiber blanket leaving the blowchamber is fed to a shredder and pelletizer. The pelletizer forms small, 1-inch diameter pellets and separates shot from the wool. A bagging operation completes the processes. For other loose wool products, fiber can be transported directly from the blowchamber to a baler or bagger for packaging.

8.16.3 Emissions and Controls¹

binder storage & mixing?
wastewater storage & treatment?

The sources of emissions in the mineral wool manufacturing industry are the cupola, the blow chamber, the curing oven, the mineral wool cooler, and possibly materials handling and bagging operations. With the exception of lead, the industry emits the full range of criteria pollutants. Also,

depending on the particular types of slag and binding agents used, the facilities may emit both metallic and organic hazardous air pollutants (HAP's).

The primary source of emissions in the mineral wool manufacturing process is the cupola. It is a significant source of particulate matter (PM) emissions and is likely to be a source of PM less than 10 micrometers (μm) in diameter (PM-10) emissions, although no particle size data are available. Coke combustion in the furnace produces carbon monoxide (CO), carbon dioxide (CO_2), and nitrogen oxide (NO_x) emissions. Finally, because blast furnace slags contain sulfur, the cupola is also a source of sulfur dioxide (SO_2) and hydrogen sulfide (H_2S) emissions.

buried
The blowchamber is a source of PM (and probably PM-10) emissions. Also, the annealing oils used in the process can lead to VOC emissions from the process. Other sources of VOC emissions include batt application and the curing oven. Finally, fugitive PM emissions can be generated during cooling, handling, and bagging operations. Table 8.16-1 presents emission factors for filterable PM emissions from various mineral wool manufacturing processes; Table 8-16.2 shows emission factors for CO, CO_2 , SO_2 , and sulfates; and Table 8.16-3 presents emission factors for fluorides.

Mineral wool manufacturers use a variety of air pollution control techniques, but most are directed toward PM control with minimal control of other pollutants. The industry has given greatest attention to cupola PM control, with two-thirds of the cupolas in operation having fabric filter control systems. Some cupola exhausts are controlled by wet scrubbers and electrostatic precipitators (ESP's); cyclones are also used for cupola PM control either alone or in combination with other control devices. About half of the blow chambers in the industry also have some level of PM control, with the predominant control device being low-energy wet scrubbers. Cyclones and fabric filters have been used to a limited degree on blow chambers. Finally, afterburners have been used to control VOC emissions from blow chambers and curing ovens and CO emissions from cupolas.

TABLE 8.16-1. (METRIC UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in kg/Mg of product unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	Filterable PM ^b	
Cupola ^c (30501701)	8.2	E
Cupola with fabric filter ^d (30501701)	0.051	D
Reverberatory furnace ^e (30501702)	2.4	E
Batt curing oven ^e (30501704)	1.8	E
Batt curing oven with ESP ^f (30501704)	0.36	D
Blow chamber ^e (30501703)	6.0	E
Blow chamber with wire mesh filter ^g (30501703)	0.45	D
Cooler ^e (30501705)	1.2	E

^aFactors represent uncontrolled emissions unless otherwise noted.

^bFilterable PM is that PM collected on or prior to the filter of an EPA Method 5 (or equivalent) sampling train.

^cReferences 1, 12. Activity level is assumed to be total feed charged.

^dReferences 6, 7, 8, 10, and 11. Activity level is total feed charged.

^eReference 12.

^fReference 9.

^gReference 7. Activity level is mass of molten mineral feed charged.

**TABLE 8.16-1. (ENGLISH UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a**

All emission factors in lb/ton of product unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	Filterable PM ^b	
Cupola ^c (30501701)	16	E
Cupola with fabric filter ^d (30501701)	0.10	D
Reverberatory furnace ^e (30501702)	4.8	E
Batt curing oven ^e (30501704)	3.6	E
Batt curing oven with ESP ^f (30501704)	0.72	D
Blow chamber ^e (30501703)	12	E
Blow chamber with wire mesh filter ^g (30501703)	0.91	D
Cooler ^e (30501705)	2.4	E

^aFactors represent uncontrolled emissions unless otherwise noted.

^bFilterable PM is that PM collected on or prior to the filter of an EPA Method 5 (or equivalent) sampling train.

^cReference 1, 12. Activity level is assumed to be total feed charged.

^dReferences 6, 7, 8, 10, and 11. Activity level is total feed charged.

^eReference 12.

^fReference 9.

^gReference 7. Activity level is mass of molten mineral feed charged.

TABLE 8.16-2 (METRIC UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in kg/Mg of total feed charged unless noted
Ratings (A-E) follow each emission factor

Source (SCC)	CO ^b		CO ₂ ^b		SO ₂		SO ₃	
Cupola (30501701)	125	D	260	D	4.0 ^c	D	3.2 ^d	E
Cupola with fabric filter (30501701)	NA		NA		NA		0.077 ^b	E
Cupola with fabric filter (30501701)	NA		NA		NA		e	
Batt curing oven (30501704)	e		e		0.58 ^d	E	e	
Blow chamber (30501703)	e		80 ^f	E	0.43 ^d	E	e	
Cooler (30501705)	e		e		0.034 ^d	E	e	

NA = Not applicable.

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 6.

^cReferences 6, 10, and 11.

^dReference 12.

^eNo data available.

^fReference 9.

TABLE 8.16-2 (ENGLISH UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in lb/ton of total feed charged unless noted
Ratings (A-E) follow each emission factor

Source (SCC)	CO ^b		CO ₂ ^b		SO ₂		SO ₃	
Cupola (30501701)	250	D	520	D	8.0 ^a	D	6.3 ^d	E
Cupola with fabric filter (30501701)	NA		NA		NA		0.15 ^b	E
Cupola with fabric filter (30501701)	NA		NA		NA		e	
Batt curing oven (30501704)					1.2 ^d	E		
Blow chamber (30501703)			160 ^f	E	0.087 ^d	E		
Cooler (30501705)					0.068 ^d	E		

NA = Not applicable.

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 6.

^cReferences 6, 10, and 11.

^dReference 12.

^eNo data available.

^fReference 9.

TABLE 8.16-3 (METRIC UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in kg/Mg of total feed charged unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	NO _x		N ₂ O		H ₂ S		Fluorides	
Cupola (30501701)	0.8 ^b	E	c		1.5 ^b	E	c	
Cupola with fabric filter (30501701)	c		c		c		0.019 ^d	D
Cupola with fabric filter (30501701)	c		c		c		0.19 ^e	D
Batt curing oven (30501714)	c		0.079	E	c		c	

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 1.

^cNo data available.

^dReferences 10 and 11. Coke only used as fuel.

^eReferences 10 and 11. Fuel combination of coke and aluminum smelting byproducts.

TABLE 8.16-3 (ENGLISH UNITS)
EMISSION FACTORS FOR MINERAL WOOL MANUFACTURING^a

All emission factors in lb/ton of total feed charged unless noted
Ratings (A-E) follow each emission factor

Process (SCC)	NO _x		N ₂ O		H ₂ S		Fluorides	
Cupola (30501701)	1.6 ^b	E	c		3.0 ^b	E	c	
Cupola with fabric filter (30501701)	c		c		c		0.038 ^d	D
Cupola with fabric filter (30501701)	c		c		c		0.38 ^e	D
Batt curing oven (30501714)	c		0.16	E	c		c	

^aFactors represent uncontrolled emissions unless otherwise noted.

^bReference 1.

^cNo data available.

^dReferences 10 and 11. Coke only used as fuel.

^eReferences 10 and 11. Fuel combination of coke and aluminum smelting byproducts.

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COMPARISON OF MINERAL WOOL MANUFACTURING EMISSION
FACTORS IN AP-40 (1967) AND SOURCE CATEGORY SURVEY (1980)

Process	Pollutant	Uncontrolled emission factors in kg/Mg			
		AP-40		Source cat. survey	
		EF	No. tests	EF	No. tests
cupola	PM	11	3	5.3	3
	SO ₂	5.6	1	5.3	10
	SO ₃	3.2	1		
	H ₂ S			1.5	3
	CO	45	1	78	9
	NO _x			0.8	6
Blowchamber	PM	8.6	4	0.8	2
	SO ₂	0.58	1		
	VOC's			0.2	2
	aldehydes	0.43	1		
Curing oven	PM	1.8	5		
	SO ₂	0.58	1		
	NO ₂	0.079	2		
	aldehydes	0.5	2		
Cooler	PM	1.2	4		
	SO ₂	0.034	1		
	aldehydes	0.021	1		

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is it part of SO₂

average all

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✓ use

check with EMB SO₂ vs H₂S

can we convert H₂S to SO₂

SO₃ to SO₂

compare with controlled (new data) ;
see if realistic

MINERAL WOOL--CO2 BY FYRITE--SUMMARY OF EMISSION FACTORS

Process	Ref. No.	Run No.	CO2, %	Volum. flow rate, DSCFM	Mass flux lb/hr	Process rate ton/hr	Emission factor,	
							kg/Mg	lb/ton
Cupola	3	1	12	4,095	3,388	4.9	350	690
		2	11	4,138	3,138	4.9	320	640
		3	10	4,342	2,993	4.9	310	610
		Average					330	650
Cupola	4	1	6	5,460	2,258	5.4	210	420
		2	6	5,502	2,276	3.9	290	580
		3	4	5,664	1,562	5.4	150	290
		Average					220	430
		1	6.5	4,494	2,014	5.2	200	390
		2	4	4,535	1,251	4.9	130	260
		3	4	4,281	1,181	5.6	110	210
		Average					150	290
		1	7.5	4,093	2,116	5.4	200	390
		2	7.5	5,015	2,593	5.2	250	500
		3	8.1	4,377	2,444	5.2	240	470
		Average					230	450
Batt curing oven	5	1	0.5	8,942	308	2	80	150
		2	0.5	8,943	308	1.42	110	220
		3	0.5	8,779	303	2.79	60	110
		Average					80	160
Cupola	6	1	2	30,019	4,139	12.54	170	330
		2	1.5	38,947	4,028	12.54	160	320
		3	1.5	39,723	4,108	12.54	170	330
		Average					170	330
		1	1	35,997	2,482	12.48	100	200
		2	1	37,813	2,607	12.48	110	210
		3	1	35,985	2,481	12.48	100	200
		4	1.5	43,313	4,479	12.48	180	360
		Average					120	240
		1	1.5	38,129	3,943	11.52	170	340
		2	1	36,781	2,536	11.52	110	220
		Average					140	280
Cupola	7	1	2	48,160	6,640	13.09	260	510
		2	2	48,730	6,719	13.09	260	510
		3	2	45,790	6,314	13.09	240	480
		Average					250	500
		1	2	46,360	6,392	13.45	240	480
		2	2	45,870	6,325	13.45	240	470
		3	2	45,360	6,254	13.45	230	460
		Average					240	470

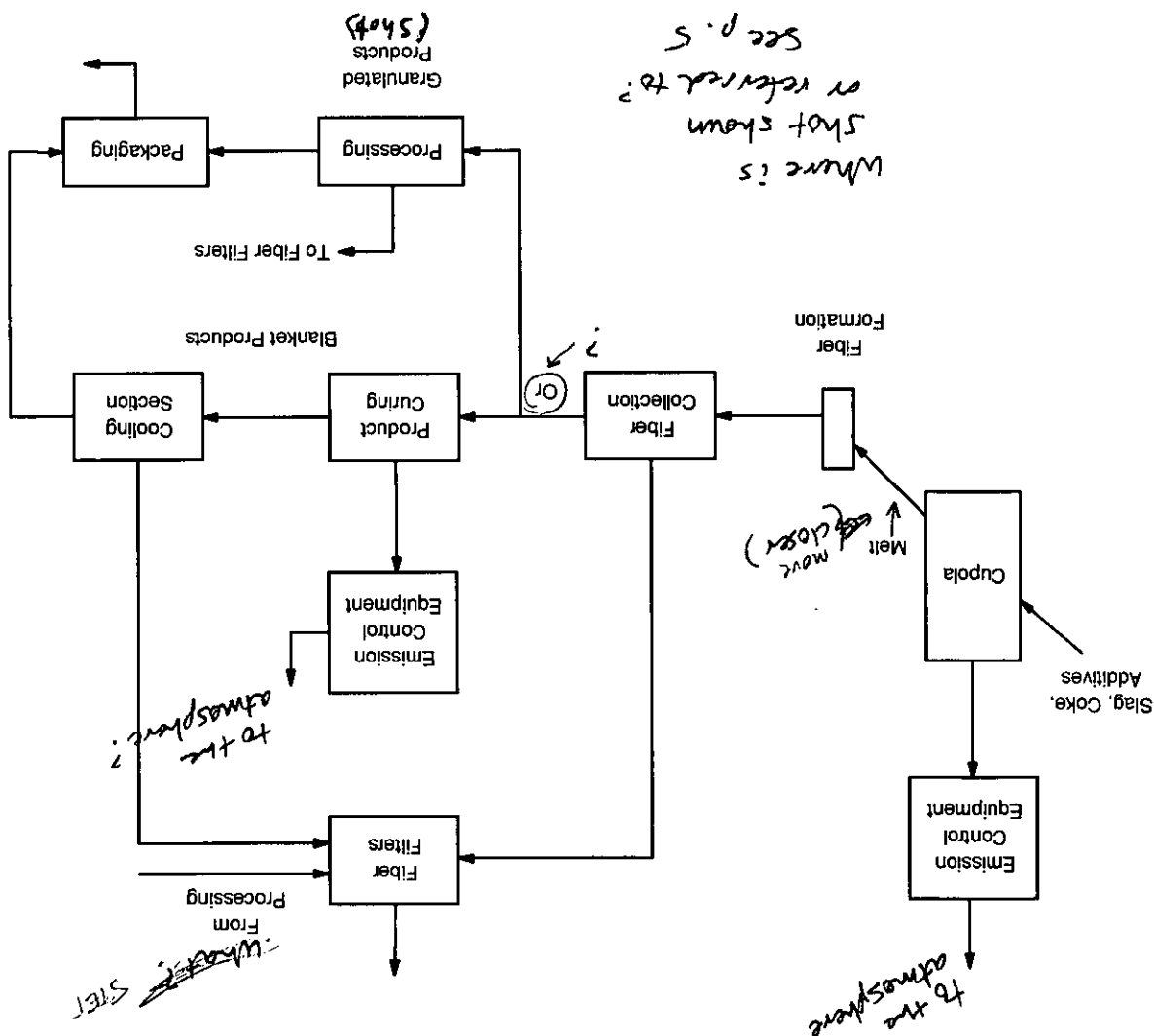
COMPARISON OF UNCONTROLLED MINERAL WOOL MANUFACTURING EMISSION
FACTORS IN AP-40 (1967) AND SOURCE CATEGORY SURVEY (1980) WITH
CONTROLLED EMISSION FACTORS DEVELOPED FROM EMISSION TESTS

CONTROLLED EMISSION FACTORS DEVELOPED FROM EMISSION TESTS										
Process	Pollutant	Uncontrolled emission factors in kg/Mg				Average emission factor,		Controlled emission factors,		Control efficiency
		AP-40		Source cat. survey						
		EF	No. tests	EF	No. tests	kg/Mg	lb/ton	APCD	kg/Mg	
cupola	→ PM	11	3	5.3	3	8.2	16	fabric filter	0.051	0.994
	× SO2	5.6	1	5.3	10	5.3	11			
	→ SO3	3.2	1			3.2	6.4	fabric filter	0.077	0.976
	→ H2S			1.5	3	1.5	3.0			
	× CO	45	1	78	9	75	149			
	→ NOx			0.8	6	0.8	1.6			
Blowchamber	→ PM	8.6	4	0.8	2	6.0	12	dry filter	0.45	0.925
	→ SO2	0.58	1			0.58	1.2			
	× VOC's			0.2	2	0.20	0.40			
	× aldehydes	0.43	1			0.43	0.86			
Curing oven	→ PM	1.8	5			1.8	3.6	ESP	0.36	0.800
	→ SO2	0.58	1			0.58	1.2			
	→ NO2	0.079	2			0.079	0.16			
	× aldehydes	0.5	2			0.5	1			
Cooler	→ PM	1.2	4			1.2	2.4			
	→ SO2	0.034	1			0.034	0.068			
	× aldehydes	0.021	1			0.021	0.042			

Figure 2-1. Mineral Wool Manufacturing Process Flow Diagram.

must be
at least
11 pts.
Times

Where is last machine? cutting?



COMPARISON OF EMISSION CONCENTRATIONS
MINERAL WOOL AP-42 REVISIONS

Section 4 refer. No.	Process	PM gr/dscf	SO2 PPM	CO	NOx	TVOC's PPM
2	cupola		0.86	7.8		
3	cupola	0.007				
3	blow cha	0.008				
4	cupola	0.011				
4	cupola	0.0094				
4	cupola	0.0093				
5	batt curin					
8	cupola+	0.018				6.26
8	cupola+	0.013				6.72
9	cupola+	0.025	0.278	5.78		
10	cupola+	0.014	16.17	10.7	0	0