

Commonwealth of Pennsylvania  
Environmental Resources  
October 12, 1990

Subject: Source Test Review

To:

Data File

Commercial Stone Company, Inc.  
Sand Manufacturing Plant - Springfield Pike  
Connellsville Township, Fayette County

From:

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PARTICULATE EMISSION TESTING  
SAND PLANT BAGHOUSE  
SPRINGFIELD PIKE QUARRY  
COMMERCIAL STONE  
CONNELLSVILLE, PA

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## 1.0 INTRODUCTION

On August 27, 1990, particulate emission testing and opacity observations were performed on the Sand Plant Baghouse at Commercial Stone Company's Springfield Pike plant in Connellsville, PA.

The test program was authorized by Ms. Leah Trielle of Commercial Stone Company. Testing was performed by Mr. Richard Campbell, Mr. Darren Midberry and visible emission readings by Mr. Kevin Kennedy of Comprehensive Safety Compliance, Inc. (CSC). Testing was observed by Mr. Tim Kunz of the Pennsylvania Department of Environmental Resources Air Pollution Control Bureau.

## 2.0 SUMMARY OF RESULTS

Table No. 1 below is a summary of Particulate Emission Data and Opacity Data from tests on the Sand Plant Baghouse.

Table No. 2 is a summary of flue gas parameters.

TABLE NO. 1  
Particulate Emissions and Opacity Data

Test Date 1990	Test No.	Test Location	Particulate Emission Data			Opacity Data <sup>c</sup>		
			Conc. <sup>a</sup> Gr/dscf	Emission Rate <sup>b</sup> lb/hr.	Allowable lb/hr	Min. %	Max. %	Highest 6 min. Avg. %
8/27	SPS-1	Baghouse Outlet	0.009	1.3	18.2	0	0	0
8/27	SPS-2	Baghouse Outlet	0.010	1.5	18.3	0	0	0
8/27	SPS-3	Baghouse Outlet	0.008	1.2	18.4	0	0	0

- (a) Grains per dry standard cubic foot
- (b) Pounds per hour
- (c) Opacity data from EPA Method 9 observations

TABLE NO. 2

## Summary of Flue Gas Parameters

Test Date 1990	Test No.	Test Location	Percent Moisture	Stack Temp. °F	Percent		Flow Rates	
					CO <sub>2</sub> <sup>a</sup>	O <sub>2</sub> <sup>b</sup>	ACFM <sup>c</sup>	DSCFM <sup>d</sup>
8/27	SPS-1	Baghouse Outlet	1.3	74	0.0	21.0	18,600	17,300
8/27	SPS-2	Baghouse Outlet	1.3	78	0.0	21.0	18,900	17,500
8/27	SPS-3	Baghouse Outlet	2.0	78	0.0	21.0	18,900	17,400

a percent carbon dioxide by volume

b percent oxygen by volume

c actual cubic feet per minute

d dry standard cubic feet per minute

### 3.0 PROCESS AND TEST PROGRAM DESCRIPTION

#### Process Description

Commercial Stone operates a sand plant at the Springfield Pike location which uses a Buell size classifier (Model GI75) manufactured by Envirotech Corporation. The Buell Gravitational-Inertial Classifier utilizes classifying principles combining, gravitational, inertial centrifugal, and aerodynamic forces. Air or gas entrained feed material enters the classifier primary air inlet at the top of the unit. The primary air inlet velocity is between 3500 and 6000 feet per minute, depending on the classification required. The curtain of feed material drops in front of the air outlet provided with widely spaced vanes to almost reverse the gas flow introduced through the primary air inlet. Prior to passing through the vanes, the relatively high velocity of the entering gas stream sets up, by friction, a counter clockwise eddy current in the chamber. The eddy current is reinforced by gas entering through the secondary air inlet located just above the coarse discharge outlet. Each particle entering the classifier has a gravitational force ( $F_g$ ) proportional to its mass which, in turn, is proportional to the cube of its diameter. As the particle is introduced in the classifier at the velocity of the primary gas stream, it is also subjected to an inertial force ( $F_i$ ) also proportional to its mass. Since the gas stream flows in a downward direction, the inertial and gravitational forces ( $F_i$  and  $F_g$ ) complement each other. The gas stream changes direction as it passed through the vanes, thus exerting a drag force ( $F_d$ ) proportional to the diameter of the particle, and almost opposite in direction to the gravitational and inertial forces. As the particle is influenced by the drag force ( $F_d$ ) and changes direction, it is subjected to a small centrifugal force ( $F_c$ ) proportional to its mass, directly opposing drag force ( $F_d$ ). Under set conditions, the resultant



force ( $R$ ) acting on a particular particle diameter ( $K$ ) referred to as the cut point, will be of a magnitude and direction to give the particle a 50-50 chance of being swept by the gas stream through the vanes or to impinge on the vanes and to be thrown back into the feed curtain. The resultant force ( $R$ ) on larger particles than ( $K$ ) is in a direction at small variance with the gravitational-inertial forces and the particles will either impinge on the vanes and be knocked out, or if large enough, they will not even come in contact with the vanes but fall directly into the coarse discharge. Smaller particles will have a resultant force ( $R$ ) almost perpendicular to the gravitational-inertial forces which will permit them to be swept through the vanes by the gas stream. The eddy current flowing in a downward parallel direction to the plane formed by the vanes, provides a moving wall containing the curtain of feed material in the classifying zone without detrimental frictional drag effects of a solid wall. The particles not swept through the vanes fall on an inclined baffle plate located at the bottom of the gas outlet directly underneath the primary gas inlet. The coarse product is scrubbed by the secondary air as it slides off into the coarse discharge outlet. Any fines adhering to coarser particles are picked up by the secondary air flow to join the stray fine particles entrained by the eddy current and are returned to the classifier inlet at point, where they are reintroduced in the classifying zone. The buell gravitational-inertial classifier separates at any desired cut point between 200 to 50 mesh (74 to 297) microns). The cut point is controlled by the air velocity through the vanes, determining the magnitude of the drag force ( $F_d$ ), and the primary air inlet velocity determining inertial force, ( $F_i$ ). Regulating the inlet velocity by increasing the flow through the secondary air inlet while keeping the total air volume, i.e., vane velocity constant, is usually all that is required to meet varying cut point

requirements. There are two units which each have a capacity of 50-70 tons per hour. A process schematic is shown in Figure No. 1. Process data is provided in Table No. 3.

#### Emission Control Equipment

The particulate emissions from the two classifiers are collected by a common baghouse. The baghouse is an Eastern Control System (ECS) reverse pulse collector. The unit is a Model 270 which contains 270 ten foot length bags. This provides a surface area of 3645 square feet for collection. Test Data from the baghouse during testing is provided in Appendix D.

#### Test Program Description

Testing was conducted on the Particulate Emissions from the Sand Plant Baghouse exit stack.

The sampling location is shown in Figure No. 2. Sample duration was 84 minutes for Test No. 1 and 72 minutes for Tests 2 and 3.

# BUELL GRAVITATIONAL-INERTIAL CLASSIFIER

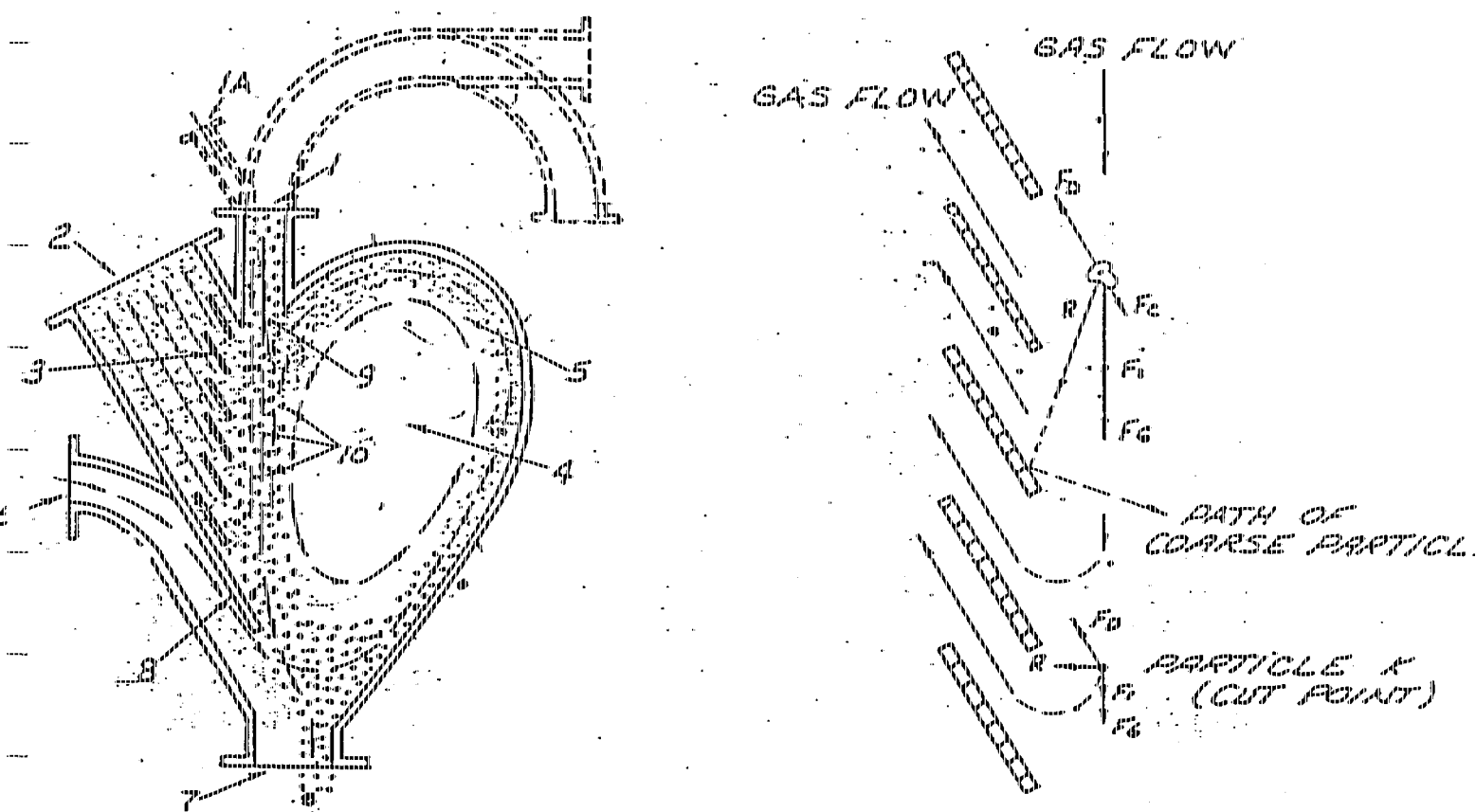


Figure No. 1

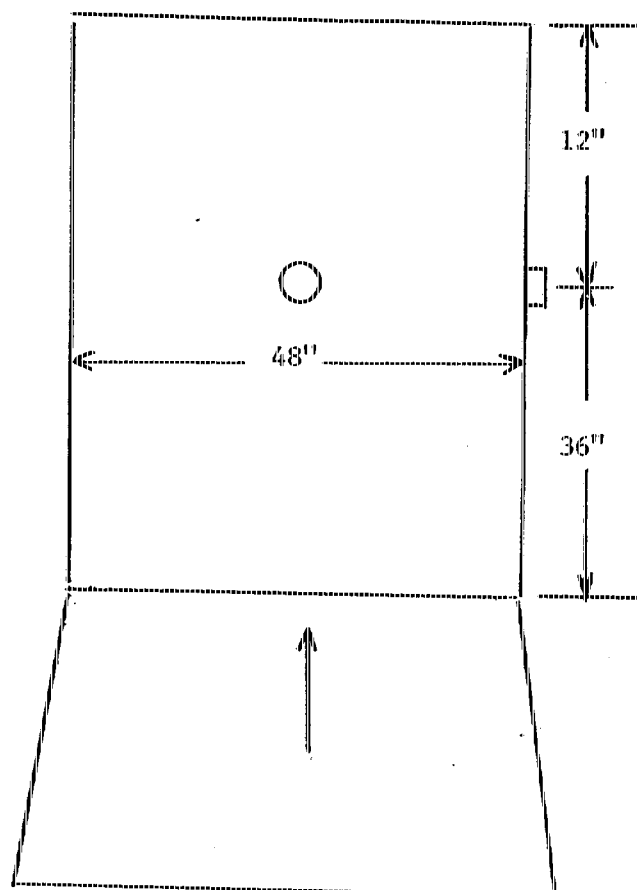
Process Schematic Springfield Pike Plant

TABEL NO. 3  
Process Production  
Springfield Pike Sand Plant

DATE 1990	TEST NO	PRODUCTION RATE TONS/HR
8/27	SPS-1	96.0
8/27	SPS-2	97.5
8/27	SPS-3	98.3

Additional process data is provided in Appendix D.

Sample Point Location



Point No.	Percent of Diameter	Point Location (inches)
1	2.1	1.0
2	6.7	3.2
3	11.8	5.7
4	17.7	8.5
5	25.0	12.0
6	35.6	17.1
7	64.4	30.9
8	75.0	36.0
9	82.3	39.5
10	88.2	42.3
11	93.3	44.8
12	97.9	47.0

2 sample ports  
 90° apart  
 12 points per port  
 24 total points

FIGURE NO. 2

Commercial Stone Sand Plant  
 Sample Location

### Method 5 Sampling Procedures

After selecting the sampling site and the minimum number of traverse points, the stack pressure, temperature, moisture and range of velocity head were measured according to the procedures described in the Federal Register\*.

Approximately 200 grams of silica gel was weighed in a sealed impinger prior to each test. Glass fiber filters\*\* (4 inch diameter) desiccated for at least 24 hours, dried at 105°C for 2 hours and weighed to the nearest 0.1mg. on an analytical balance. One hundred ml of distilled water was placed in each of the first two impingers; the third impinger was initially empty; and the impinger containing the silica gel was placed next in series. The sampling train was leak-checked at the sampling site prior to each test run by plugging the inlet to the nozzle and pulling 15-inch Hg vacuum, at the conclusion of the test by plugging the inlet to the nozzle and pulling a vacuum equal to the highest vacuum reached during the test run. A more detailed description of the sampling and analytical procedures is provided in Appendix B.

\*Federal Register, CFR 40, Part 60, July 1, 1989

## A. FORMULAS AND CALCULATIONS

## NOMENCLATURE AND DIMENSIONS

$A_s$	=	cross-sectional area of stack, $\text{ft}^2$
$A_n$	=	area of sampling nozzle, $\text{ft}^2$
$B_{ws}$	=	proportional by volume of water vapor in the gas stream, dimensionless
$C_p$	=	pitot tube coefficient, dimensionless = .84
$C_s$	=	concentration of particulate matter in stack gas, $\text{gr/scf}$ , dry basis
%CO	=	percent of carbon monoxide by volume, dry basis
%CO <sup>2</sup>	=	percent of carbon dioxide by volume, dry basis
H	=	average pressure drop across the orifice meter, inches of $\text{H}_2\text{O}$
I	=	percent of isokinetic sampling
$M_d$	=	dry molecular weight, $\text{lb/lb-mole}$
$M_n$	=	total amount of particulate matter collected, $\text{mg}$ .
$M_s$	=	molecular weight of stack gas (wet basis), $\text{lb/lb-mole}$
%N <sub>2</sub>	=	percent of nitrogen by volume, dry basis
%O <sub>2</sub>	=	percent of oxygen by volume, dry basis
p	=	velocity head of stack gas, inches of $\text{H}_2\text{O}$
$P_{\text{bar}}$	=	barometric pressure, inches of Hg
$P_s$	=	absolute stack gas pressure, inches of Hg
pmr	=	particulate matter emission rate, $\text{lbs/hr}$
$Q_s$	=	volumetric flow rate, wet basis, standard conditions
$Q_{s\text{std}}$	=	volumetric flow rate, dry basis, standard
$T_m$	=	average temperature of dry gas meter, $^{\circ}\text{R}$
$T_s$	=	average temperature of stack gas, $^{\circ}\text{R}$



$V_{lc}$  = total volume of liquid collected in impingers and silica gel, ml.

$V_m$  = volume of sample through the dry gas meter at meter conditions,  $ft^3$

$V_{mstd}$  = volume of gas sample through the dry gas meter at standard conditions,  $ft^3$

$V_s$  = stack gas velocity at stack conditions, fps

$V_{wstd}$  = volume of water in the gas sample at standard conditions,  $ft^3$

$t$  = total sampling time, minutes

Note: Standard conditions =  $70^\circ F$  and 29.92 inches of Hg.

# Example Calculations for Particulate Emissions

## Test No. SPS-1

1. Volume of dry gas sampled corrected to standard conditions. Note:  $V_m$  must be corrected for leakage if any leakage rates exceed  $L_a$ .)

$$V_{mstd} = 17.65 \times V_m \times Y \left[ \frac{P_{bar} + \frac{\Delta H}{13.6}}{T_m} \right] =$$

$$V_{mstd} = 17.65 \times 87.500 \times (.995) \left[ \frac{28.6 + \frac{3.9}{13.6}}{567} \right] = 78.287$$

2. Volume of water vapor at standard conditions,  $ft^3$ .

$$V_{wstd} = 0.04707 V_{1C} =$$

$$V_{wstd} = 0.04707 \times 21.4 = 1.01$$

3. Moisture content in stack gas.

$$B_{ws} = \frac{V_{wcstd}}{V_{mstd} + V_{wcstd}} = B_{ws} = \frac{1.01}{78.287 + 1.01} = .013$$

4. Dry molecular weight of stack gas.

$$M_d = 0.440 (\% CO_2) + 0.320 (\% O_2) + 0.280 (\% N_2 + \% CO) =$$

$$M_d = 0.440 (0.0) + 0.32 (21) + 0.280 (79) = 28.84$$

5. Molecular weight of stack gas

$$M_s = M_d (1 - B_{ws}) + 18 B_{ws} =$$

$$M_s = 28.84 (1 - .013) + 18(.013) = 28.70$$

6. Stack velocity at stack conditions, fps

$$V_s = 85.49 \times C_p \times \sqrt{\Delta p_{avg.}} \times \sqrt{\frac{T_s}{P_s M_s}} =$$

SPS-1

$$V_s = 85.49 \times .84 \times (.425) \times \sqrt{\frac{534}{28.61 \times 28.70}} = 24.61$$

7. Stack gas volumetric flow rate at stack conditions, cfh

$$Q_s = 3600 \times V_s \times A_s$$

$$Q_s = 60 \times 24.61 \times 12.57 = 18,561 \text{ ACFM}$$

8. Dry stack gas volumetric flow rate at standard conditions, cfh.

$$Q_{s\text{std}} = 17.65 Q_s \frac{P_s}{T_s} (1 - B_{ws}) = \text{SCFM} \times 60 = \text{SCFH}$$

$$Q_{s\text{std}} = 17.65 \times 18,561 \frac{28.61}{534} \times (1 - .013) = 17,324 \times 60 = 1,039,440$$

9. Concentration in g/scf

$$C'_s = 0.001 \text{ g/mg} \frac{M_n}{V_{m\text{std}}} = 0.001 \times \frac{43.9}{78.287} = 0.00056$$

$$0.00056 \times 15.43 = 0.009 \text{ gr/dscf}$$

10. Particulate mass emission rate, lbs/hr.

$$\text{pmr} = \frac{C_s \times Q_{s\text{std}}}{454} = \frac{0.00056 \times 1,039,440}{454} = 1.3 \text{ lb/hr}$$

11. Isokinetic variation

$$I = \frac{100 T_s \left[ \frac{0.002669 V_{1c}}{60} + \frac{\frac{V_m}{V_s} \frac{T_m}{P_s} \gamma \frac{P_{bar}}{A_n} + \frac{\Delta H}{13.6}} \right]}{1} =$$

$$I = \frac{100 (534) \left[ \frac{0.002669 (21.4)}{60 \times 84 \times 24.61 \times 28.61 \times .000707} + \frac{87.500}{567} \frac{(.995)}{28.61} \frac{3.9}{13.6} \right]}{1} = 95.6$$

## Example Calculations for Particulate Emissions

Test No. SPS-2

1. Volume of dry gas sampled corrected to standard conditions. Note:  $V_m$  must be corrected for leakage if any leakage rates exceed  $L_a$ .)

$$V_{mstd} = 17.65 \times V_m \times Y \left[ \frac{P_{bar} + \frac{\Delta H}{13.6}}{T_m} \right] =$$

$$V_{mstd} = 17.65 \times 36.887 \times (.995) \left[ \frac{28.6 + \frac{3.9}{13.6}}{569} \right] = 32.887$$

$$V_{mstd} = 17.65 \times V_m \times Y \left[ \frac{P_{bar} + \frac{\Delta H}{13.6}}{T_m} \right] =$$

$$V_{mstd} = 17.65 \times 40.405 \times (.99) \frac{28.6 + \frac{4.0}{13.6}}{561} = 36.363$$

$$\text{Total } V_{mstd} = 32.887 + 36.363 = 69.250$$

2. Volume of water vapor at standard conditions,  $\text{ft}^3$ .

$$V_{wstd} = 0.04707 V_{1c} =$$

$$V_{wstd} = 0.04707 \times 19.6 = .922$$

3. Moisture content in stack gas.

$$B_{ws} = \frac{V_{wcstd}}{V_{mstd} + V_{wcstd}} = B_{ws} = \frac{.922}{69.250 + .922} = .013$$

4. Dry molecular weight of stack gas.

$$M_d = 0.440 (\% \text{CO}_2) + 0.320 (\% \text{O}_2) + 0.280 (\% \text{N}_2 + \% \text{CO}) =$$

$$M_d = 0.440 (0.0) + 0.32 (21.0) + 0.280 (79) = 28.84$$

# SPS-2

5. Molecular weight of stack gas

$$M_s = M_d (1-B_{ws}) + 18 B_{ws} =$$

$$M_s = 28.84 (1-.013) + 18(.013) = 28.70$$

6. Stack velocity at stack conditions, fps

$$V_s = 85.49 \times C_p \times \sqrt{\Delta p_{avg.}} \times \sqrt{\frac{T_s}{P_s M_s}} =$$

$$V_s = 85.49 \times .84 \times (.432) \times \sqrt{\frac{538}{28.61 \times 28.70}} = 25.11$$

7. Stack gas volumetric flow rate at stack conditions, cfm

$$Q_s = 3600 \times V_s \times A_s$$

$$Q_s = 60 \times 25.11 \times 12.57 = 18,938 \text{ ACFM}$$

8. Dry stack gas volumetric flow rate at standard conditions, cfm.

$$Q_{sstd} = 17.65 Q_s \frac{P_s}{T_s} (1-B_{ws}) = \text{SCFM} \times 60 = \text{SCFH}$$

$$Q_{std} = 17.65 \times 18,938 \frac{28.61}{538} \times (1-.013) = 17,544 \times 60 = 1,052,640$$

9. Concentration in g/scf

$$C'_s = 0.001 \text{ g/mg} \frac{M_n}{V_{mstd}} = 0.001 \times \frac{46.4}{69.250} = 0.00067$$

$$.00067 \times 15.43 = 0.010 \text{ gr/dscf}$$

10. Particulate mass emission rate, lbs/hr.

$$\text{pmr} = \frac{C_s \times Q_{sstd}}{454} = \frac{.00067 \times 1,052,640}{454} = 1.5 \text{ lb/hr}$$

# SPS-2

## 11. Isokinetic variation

$$I = \frac{100 T_s \left[ \frac{0.002669 V_1}{60} + \frac{\frac{V_m}{T_m} Y P_{bar}}{V_s P_s A_n} + \frac{\Delta H}{13.6} \right]}{=}$$

$$I = \frac{100 (538) \left[ \frac{0.002669(19.6)}{60 \times 72 \times 25.11 \times 28.61 \times .000707} + \frac{77.292}{565 \times (.99) (28.6 + \frac{4}{13.6})} \right]}{= 97.2}$$

## Example Calculations for Particulate Emissions

### Test No. SPS-3

1. Volume of dry gas sampled corrected to standard conditions. Note:  $V_m$  must be corrected for leakage if any leakage rates exceed  $L_a$ .)

$$V_{m_{std}} = 17.65 \times V_m \times Y \left[ \frac{P_{bar} + \frac{\Delta H}{13.6}}{T_m} \right] =$$

$$V_{m_{std}} = 17.65 \times 77.275 \times (.99) \left[ \frac{28.6 + \frac{3.8}{13.6}}{571} \right] = 68.292$$

2. Volume of water vapor at standard conditions,  $ft^3$ .

$$V_{w_{std}} = 0.04707 V_{1C} =$$

$$V_{w_{std}} = 0.04707 \times 29.2 = 1.37$$

3. Moisture content in stack gas.

$$B_{ws} = \frac{V_{wc_{std}}}{V_{m_{std}} + V_{wc_{std}}} = B_{ws} = \frac{1.37}{68.292 + 1.37} = .020$$

4. Dry molecular weight of stack gas.

$$M_d = 0.440 (\% CO_2) + 0.320 (\% O_2) + 0.280 (\% N_2 + \% CO) =$$

$$M_d = 0.440 (0.0) + 0.320 (21) + 0.280 (79) = 28.84$$

5. Molecular weight of stack gas

$$M_s = M_d (1 - B_{ws}) + 18 B_{ws} =$$

$$M_s = 28.84 (1 - .02) + 18(.02) = 28.62$$

6. Stack velocity at stack conditions, fps

$$V_s = 85.49 \times C_p \times \sqrt{\Delta p_{avg.}} \times \sqrt{\frac{T_s}{P_s M_s}} =$$

# SPS-3

$$V_s = 85.49 \times .84 \times (.430) \times \frac{538}{28.61 \times 28.62} = 25.03$$

7. Stack gas volumetric flow rate at stack conditions, cfh

$$Q_s = 3600 \times V_s \times A_s$$

$$Q_s = 60 \times 25.03 \times 12.57 = 18,878 \text{ ACFM}$$

8. Dry stack gas volumetric flow rate at standard conditions, cfh.

$$Q_{sstd} = 17.65 Q_s \frac{P_s}{T_s} (1-B_{ws}) = \text{SCFM} \times 60 = \text{SCFH}$$

$$Q_{std} = 17.65 \times 18,878 \frac{28.61}{538} \times (1-.02) = 17,364 \times 60 = 1,041,840$$

9. Concentration in g/scf

$$C'_s = 0.001 \text{ g/mg} \frac{M_n}{V_{mstd}} = 0.001 \times \frac{35.8}{68.292} = .00052$$

$$.00052 \times 15.43 = .008 \text{ gr/dscf}$$

10. Particulate mass emission rate, lbs/hr.

$$\text{pmr} = \frac{C_s \times Q_{sstd}}{454} = \frac{.00052 \times 1,041,840}{454} = 1.2 \text{ lb/hr}$$

11. Isokinetic variation

$$I = \frac{100 T_s \left[ 0.002669 \frac{V_1}{V_s} + \frac{T_m}{P_s} Y \frac{P_{bar}}{A_n} + \frac{\Delta H}{13.6} \right]}{60} =$$

$$I = \frac{100 (538) \left[ 0.002669 (29.2) + \frac{77.275}{60 \times 72 \times 25.03 \times 28.61 \times .000707} + \frac{3.8}{13.6} \right]}{60 \times 72 \times 25.03 \times 28.61 \times .000707} = 97.1$$



Formula,

$A = 0.76E^{0.42}$ , where:

A = Allowable emissions in pounds per hour

E = Emission index = F x W pounds per hour

E = Process factor in pounds per unit, and

W = Production or charging rate in units per hour

F = 20 lbs/ton

W = Tons/hr

Test No.1

$$A = 0.76 (20 \times 96.0)^{0.42} = 18.2 \text{ lbs/hr}$$

Test No. 2

$$A = 0.76 (20 \times 97.5)^{0.42} = 18.3 \text{ lbs/hr}$$

Test No. 3

$$A = 0.76 (20 \times 98.3)^{0.42} = 18.4 \text{ lbs/hr}$$

## B. SAMPLING METHODOLOGY & EQUIPMENT CALIBRATION

## SAMPLE AND VELOCITY TRAVERSES FOR STATIONARY SOURCES EPA METHOD 1

The following method was used in this test program. Sampling procedures follow those described in EPA Method 1 of the Federal Register\*.

### SELECTION OF THE MEASUREMENT SITE

The velocity measurement were taken at (\*\*) stack (or duct) diameters downstream and (\*\*) diameters upstream from any flow disturbances.

### CHOOSING THE NUMBER OF TRAVERSE POINTS (non-Cyclonic flow)

#### Particulate Traverse Sampling

At least twelve (12) traverse points were sampled for circular (or rectangular) stack diameters greater than .61 meters. At least eight (8) traverse points were sampled for circular stacks and nine (9) points for rectangular stacks for equivalent diameters for .30 to .61 meters. For other stack diameters, Figure 1-1 was referred to.

#### Non-Particulate Traverse Sampling

The above procedure was used except that Figure 1-2 is substituted for Figure 1-1.

### CROSS SECTIONAL TRAVERSE POINT LOCATIONS

The traverse points were located on two (2) perpendicular diameters. For particulate traverse samples, on of the above diameters was positioned in the plane containing the greatest concentration variation.

#### Circular Stacks

EPA Method 1, Figure 1-3 and Table 1-2 was referred to.

#### Rectangular Stacks

Using Figure 1-4 and Table 1-1 a rectangular grid, sectioned into as many equal areas as there are traverse points, was constructed. A sample was taken at the center of each elemental area in the grid.

No traverse points were located within 2.5 centimeters of the stack wall for stack diameters greater than .61 meters. For stack diameters equal to or less than .61 meters, no traverse points were located within 1.3 centimeters. For both of the above cases, where the sample nozzle inside diameter was greater than the adjusted distances, the larger distance was used.

\*Federal Register, CFR 40, Part 60, July 1, 1989

\*\* Check sampling location schematic included.

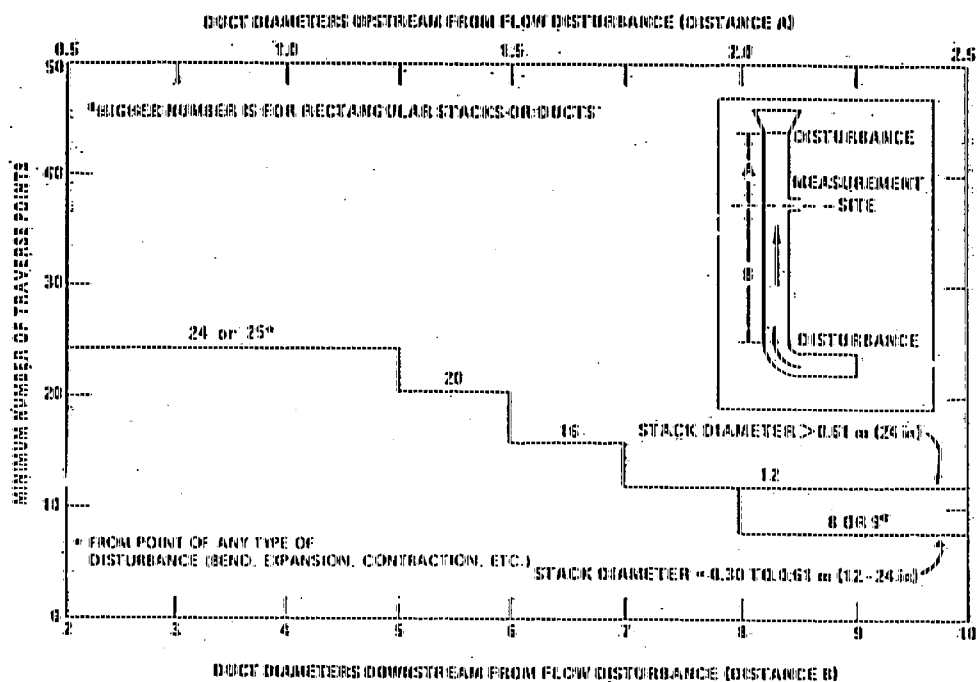


Figure 1-2. Minimum number of traverse points for particulate traverses.

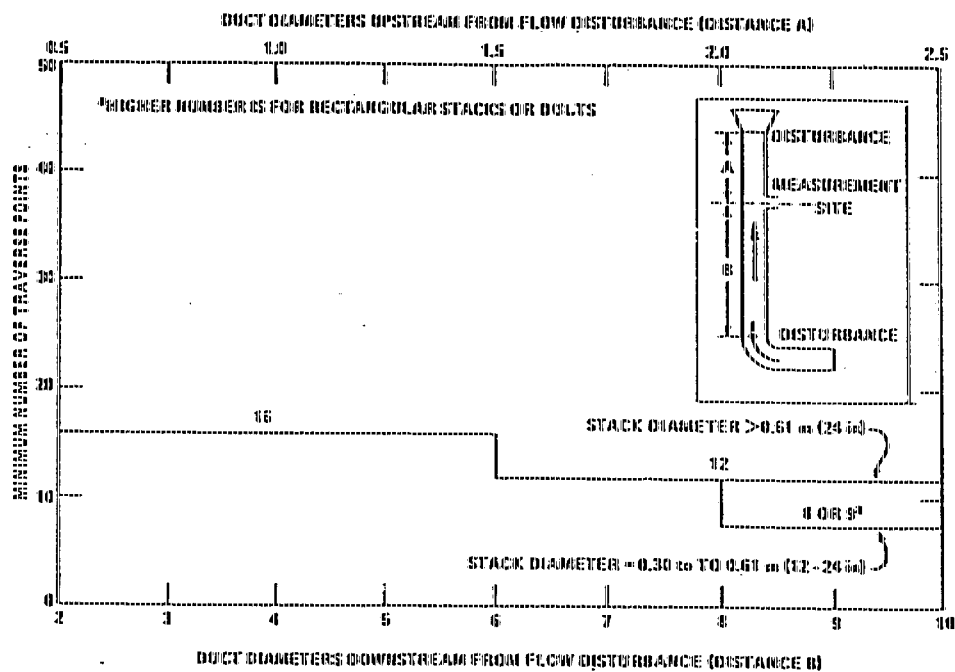


Figure 1-2. Minimum number of traverse points for velocity (nonparticulate) traverses.

TABLE 1-2. LOCATION OF TRAVERSE POINTS IN CIRCULAR STACKS  
(Percent of stack diameter from inside wall to traverse point)

Traverse point number on a diameter	Number of traverse points on a diameter--											
	2	4	6	8	10	12	14	16	18	20	22	24
1	14.6	8.7	4.4	3.2	2.6	2.1	1.8	1.6	1.4	1.3	1.1	1.0
2	35.4	25.0	14.6	10.5	8.2	6.7	5.7	4.9	4.4	3.9	3.5	3.2
3		75.0	29.6	19.4	14.6	11.8	9.9	8.5	7.5	6.7	6.0	5.5
4			93.3	70.4	32.3	22.6	17.7	14.6	12.5	10.9	9.7	8.9
5				95.4	67.7	34.2	25.0	20.1	16.9	14.6	12.9	11.6
6					95.6	65.8	35.6	26.9	22.0	18.8	16.5	14.6
7						95.5	77.4	64.4	36.6	28.3	23.6	20.4
8							95.0	85.4	75.0	63.4	37.5	29.6
9								91.8	82.3	71.1	62.5	39.2
10									97.4	88.2	79.9	71.7
11										93.3	85.4	78.0
12											90.1	83.1
13												94.3
14												
15												
16												
17												
18												
19												
20												
21												
22												
23												
24												

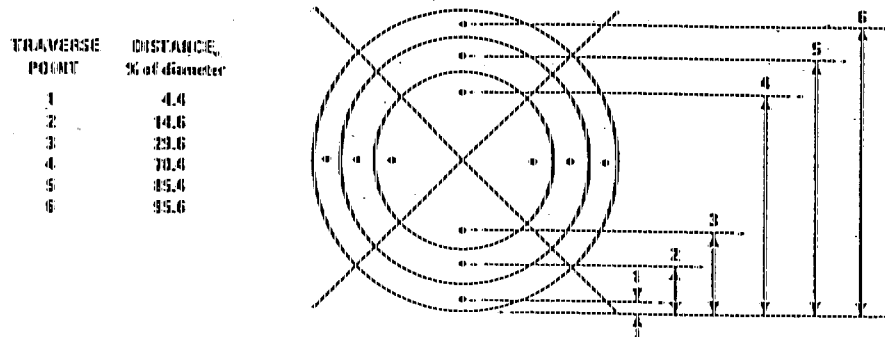


Figure 1-3. Example showing circular stack cross section divided into 12 equal areas, with location of traverse points indicated.

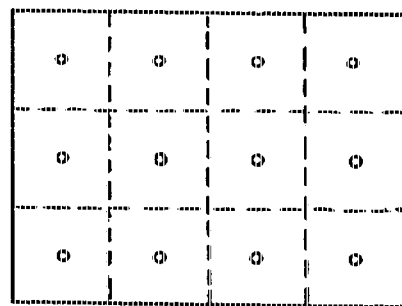


Figure 1-4. Example showing rectangular stack cross section divided into 12 equal areas, with a traverse point at centroid of each area.

### Verification of the Absence of Cyclonic Flow through Stacks

For any stack where there existed a possibility of cyclonic flow, a test for the presence of cyclonic flow was performed using a manometer and Type S pitot tube.

The manometer was leveled and zeroed before the pitot tube was connected to the probe and positioned at each traverse point. The pitot tube was rotated (if necessary) until a null reading was obtained. All rotation angles were assigned absolute values. When no rotation was necessary, a value of 0 degrees was assigned. The values were summed, and the resultant angle called alpha (  $\alpha$  ). Twenty (20) degrees was the limit of acceptability of the measurement locations for the average value of alpha with a Standard Deviation of ten (10) degrees or less.

A minimum of 24 traverse points for circular ducts and 42 points for rectangular ducts were used in the determination of gas flow angles. When the absence of cyclonic gas flow was verified, these same points were used to obtain velocity measurements.

EPA METHOD 2

DETERMINE OF STACK GAS VELOCITY AND VOLUMETRIC FLOW RATE

DETERMINATION OF STACK GAS VELOCITY AND VOLUMETRIC FLOW RATE  
EPA METHOD 2

The average velocity in a stack (or duct) was determined from the gas' density and average velocity head with a Type S pitot tube and stack gas flow was quantified.

SAMPLING APPARATUS

The apparatus consisted of the following equipment:

Pitot Tube - Type S pitot tube that meet all geometry standards was used to monitor stack gas velocity.

Draft Gauge - An inclined manometer made by Dwyer with readability of 0.01 inches H<sub>2</sub>O in the 0-1 inch range was used.

Temperature Gauge - Included a thermocouple and digital readout capable of measuring temperatures to within 1.5% of the minimum stack temperatures.

Barometer - A barometer capable of measuring atmospheric pressure to within 2.5 mm Hg.

Gas Density Determination Equipment - Fyrite analyzer.



## SAMPLING PROCEDURE

The apparatus was set up as illustrated in Figure 1.

The pretest leak-check was conducted for both the impact and static openings of the Pitot tube.

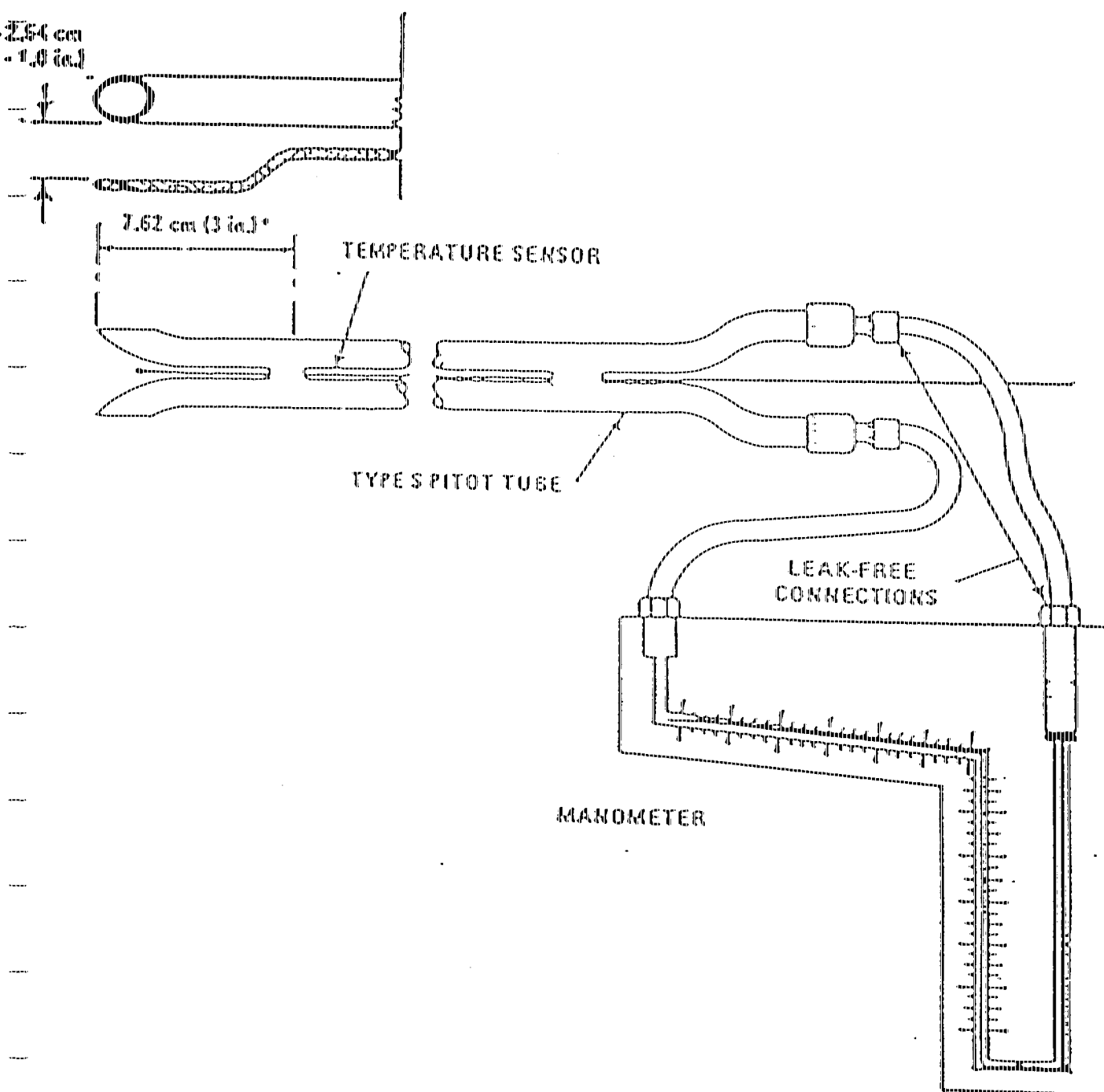
After adjusting the manometer level and zero, the velocity head and temperature were measured for each traverse point and recorded on the data sheet.

In addition, the static pressure in the stack was measured and the atmospheric pressure recorded.

The stack gas dry molecular weight was determined using EPA Method 3.

The moisture content was obtained using either EPA Method 4 or EPA Method 5.

The cross sectional area of the stack (or duct) at the sampling location was measured and recorded.



EPA Method 2  
Flow Measurement System

EPA METHOD 3

GAS ANALYSIS FOR CARBON DIOXIDE, OXYGEN, EXCESS AIR & DRY

GAS ANALYSIS FOR CARBON DIOXIDE, OXYGEN, EXCESS AIR AND DRY  
MOLECULAR WEIGHT

EPA METHOD 3

The following method was used in this test program. Sampling procedures followed those described in EPA Method 3 of the Federal Register\*.

SAMPLING APPARATUS

The gas analysis sampling train used in these tests at the exit stack meet the design specifications established by the Federal EPA and was assembled by Comprehensive Safety Compliance, Inc. (CSC) personnel.

The apparatus consisted of the following:

INTEGRATE SAMPLING: APPARATUS

PROBE - A stainless steel probe equipped with an in-stack or out-stack glass wool filter to remove particulate matter.

PUMP - A leak-free, diaphragm-type pump to transport sample gas to the flexible bag.

CONDENSER - A water-cooled condenser to remove excess moisture.

VALVE - A needle valve to adjust the sample flow rate.

SURGE TANK - A surge tank installed between the pump and rate meter to eliminate the pulsation effect of the diaphragm pump on the rate meter.

RATE METER - A rotameter capable of measuring flow rates to within  $\pm 2\%$  of a flow range of 500 to 1000 cubic centimeters per minute.

ORSAT ANALYZER - This combustible gas analyzer was used to determine dry molecular weight (DMW) for  $O_2$ ,  $CO_2$  and CO.

\*Federal Register, CFR 40, Part 60, July 1, 1989

## INTEGRATED SAMPLING

The sample train was set up as shown in Figure No. 1.

For single point sampling the sample point location was at the center of the cross-section or at least 1 meter away from the stack wall. The analyzer leak test was performed, and the flexible bag as well as the sample train was also leak tested.

For multi-point sampling at least eight traverse points were sampled in circular stacks with diameters less than .61 meters (9 points for rectangular stacks of .61 meter equivalent diameter). At least twelve points were sampled for all other cases.

Traverse point location was performed; with respect to EPA Method 1. All points were traversed and sampled for an equal length of time.

After positioning the probe and purging the sample line, the sample bag was connected to the train.

The sample was taken at a constant rate. The run was simultaneous with the lasted the total duration of the pollutant emission rate determination.

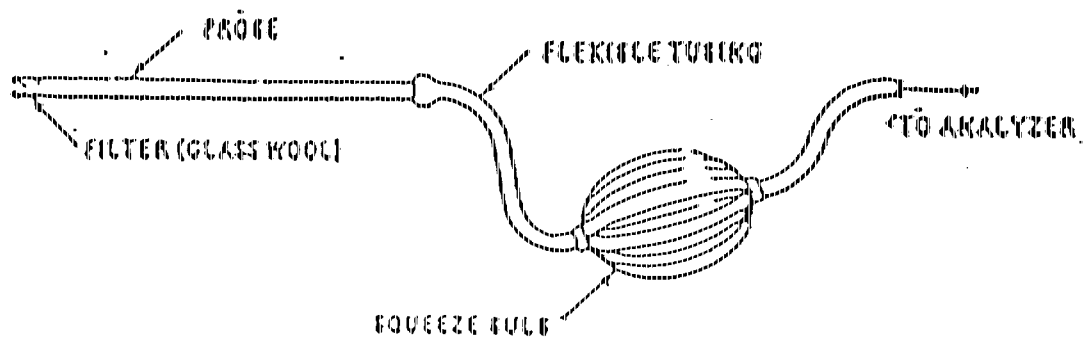
One flue gas sample was taken for each pollutant emission rate determination. Within eight hours the samples were analyzed for %  $\text{CO}_2$  and  $\text{O}_2$ . The %  $\text{N}_2$  and CO were determined and the dry molecular weight was calculated.

## SAMPLE ANALYSIS

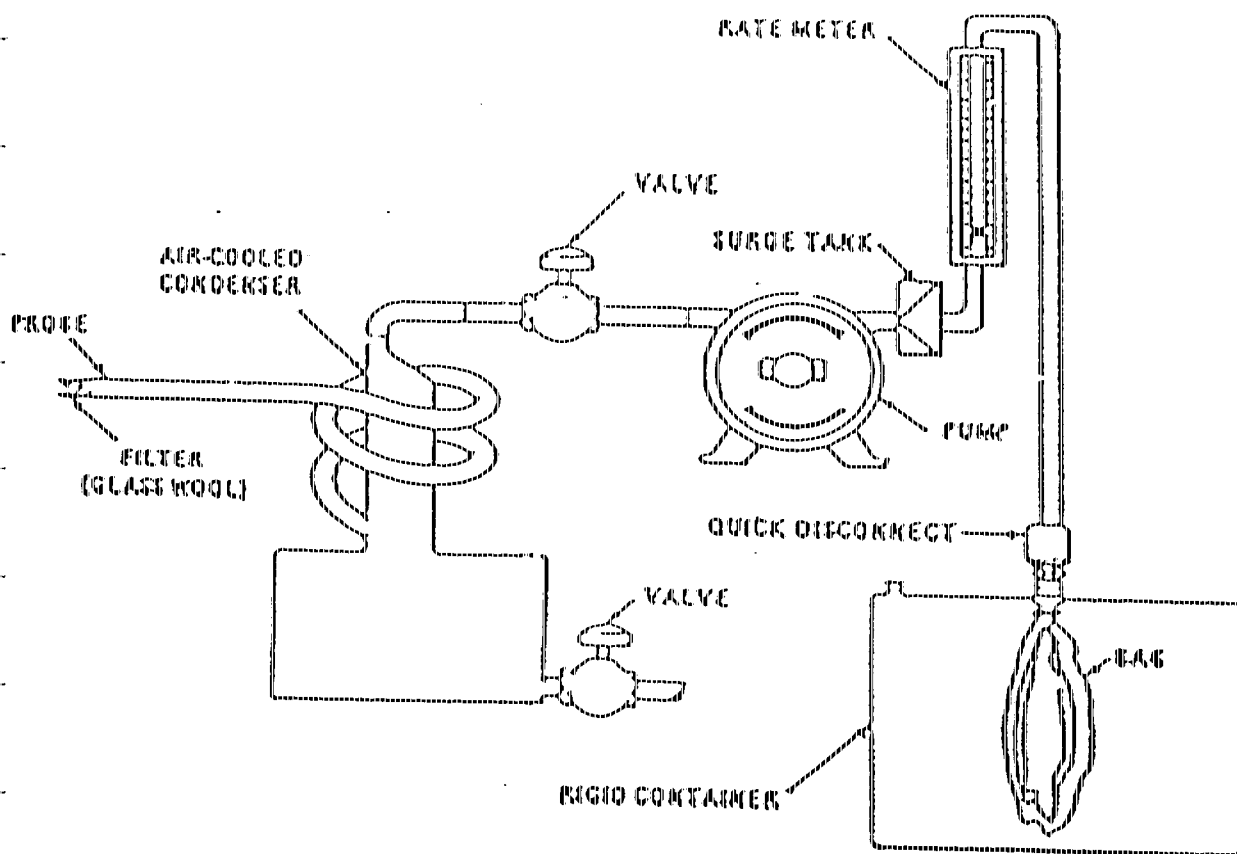
After a sample was drawn into a combustible gas analyzer, it was immediately analyzed for percent  $\text{CO}_2$  and  $\text{O}_2$ . The percentage of the gas that is  $\text{N}_2$  and CO was determined by subtracting the sum of the percents  $\text{CO}_2$  and  $\text{O}_2$  from 100 percent.

During sample analysis for a given gas, consecutive passes through the analyzer absorbing solutions were made until two consecutive readings were obtained. All values were averaged and recorded to the nearest 0.1%.

The analyzer was leak-tested before and after the analyses was performed.



Grab-sampling train.



EPA Method 3

Gas Composition Sampling Train

### CO<sub>2</sub> ANALYSIS

The results of three analyses differ by no more than:

- a) .3% volume when the CO<sub>2</sub> was greater than 4% of the total gas volume.
- b) .2% volume when CO<sub>2</sub> was less than or equal to 4% of the total gas volume.

### O<sub>2</sub> ANALYSIS

The analytical procedure was repeated until the results of three analyses differ by no more than:

- a) .3% volume when O<sub>2</sub> was less than or equal to 15% of total gas volume.
- b) .2% by volume when O<sub>2</sub> was greater than or equal to 15% of total gas volume.

### CO AND N<sub>2</sub> ANALYSIS

For percent CO and N<sub>2</sub>, the analytical procedure was repeated until three analyses differ by no more than .3%.

### DETERMINATION OF THE DRY MOLECULAR WEIGHT

The DMW of each sample was determined using the following formula:

$$\text{DMW} = 0.440 (\% \text{ CO}_2) + .322 (\% \text{ O}_2) + .280 (\% \text{ N}_2 + \% \text{ CO})$$

### EMISSION RATE CORRECTION FACTOR OR EXCESS AIR DETERMINATION

An EPA approved combustible gas analyzer was used to determine the excess air content in the stack emissions.

For single point and grab samples, the values obtained for % O<sub>2</sub>, CO and N<sub>2</sub> were substituted in the following equation to determine the % excess air:

$$\% \text{ EA} = \frac{\% \text{ O}_2 - .5\% \text{ CO}}{.264\% \text{ N}_2 - (\% \text{ O}_2 - .5\% \text{ CO})} \times 100\%$$

The fuel factor, F, calculated as follows:

$$F = \frac{20.9 - \% \text{ O}_2}{\% \text{ CO}_2}$$

\*20.9 = The percent of O<sub>2</sub> by volume in ambient air.

## SAMPLING PROCEDURE

After selecting the sampling site and the minimum number of traverse points, the stack pressure and temperature was measured according to the procedures described in the Federal Register\*.

Approximately 200 grams of silica gel were weighed in a sealed impinger prior to each test. One-hundred ml of distilled water was placed in each of the first two impingers; the third impinger was initially empty; and the impinger containing the silica gel was placed next in series. The train was set up with the probe as shown in Figure A-1. The sampling train was leak-checked at the sampling site prior to each test run by plugging the inlet to the nozzle and pulling a 15 inch Hg vacuum, and at the conclusion of the test by plugging the inlet to the nozzle and pulling a vacuum equal to the highest vacuum reached during the test run.

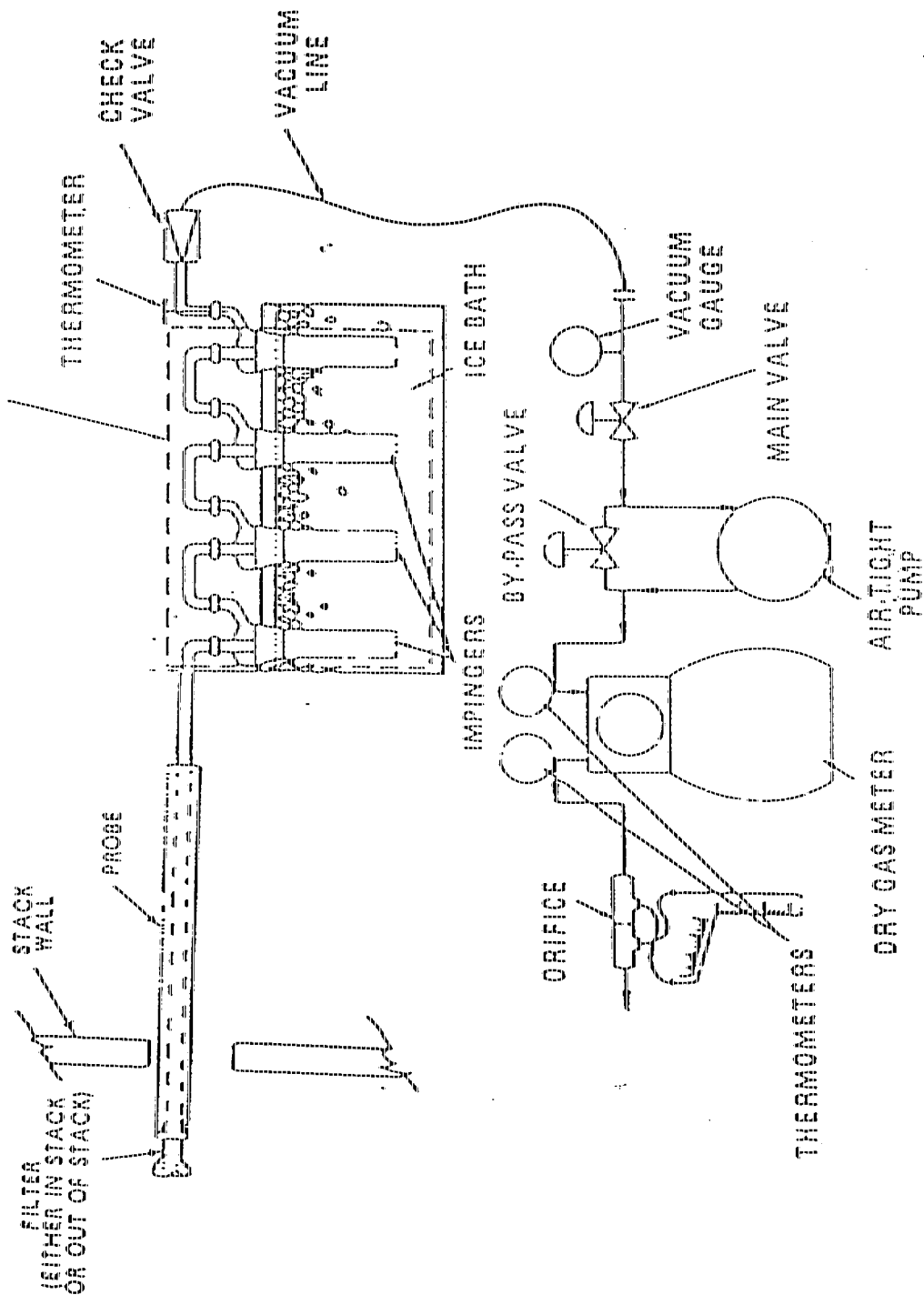
Crushed ice was placed around the impingers to keep the temperature of the gases leaving the last impinger at 68 F or less.

During sampling, stack gas and sampling train data were recorded at each sampling point and when significant changes in stack flow conditions occur. All sampling data were recorded on the Field Data Sheet.

\*Federal Register, CFR 40, Part 60, July 1, 1989



IMPINGER TRAIN OPTIONAL, MAY BE REPLACED  
BY AN EQUIVALENT CONDENSER



Moisture sampling train-reference method.

#### SAMPLE RECOVERY PROCEDURE

The sampling train was moved carefully from the test site to the cleanup area. Sample fractions were recovered as follows:

1. The volume of water from the first three impingers was measured and recorded on the Recovery Data Sheet.
2. The silica gel from the fourth impinger was weighed and the gain recorded on the Sample Recovery Data Sheet with other pertinent data.

EPA METHOD 5  
DETERMINATION OF PARTICULATE EMISSIONS

DETERMINATION OF PARTICULATE EMISSIONS  
EPA METHOD 5

The following method was used in this test program. Sampling procedures followed those described in Method 5 of the Federal Register\*.

SAMPLING APPARATUS

The particulate sampling train used in these tests at the exit stack met design specifications established by the Federal EPA and was assembled by Comprehensive Safety Compliance, Inc. (CSC) personnel. It consisted of:

Nozzle - Stainless steel (316) with sharp, tapered leading edge and accurately measured round opening.

Probe - Glass lined with a heating system capable of maintaining a minimum gas temperature of 250 F at the exit end during sampling.

Filter Holder - Pyrex glass with heating system capable of maintaining a filter temperature of approximately 250 F.

Draft Gauge - An inclined manometer made by Dwyer with readability of 0.01 inches H<sub>2</sub>O in the 0-1 inch range was used.

Impingers - Four impingers connected in series with glass ball joints. The first, third and fourth impingers were of the Greensburg-Smith design, modified by replacing the tip with a 1/2 inch I.D. glass tube extending to 1/2 inch from the bottom of the flask.

Metering System - Vacuum gauge, leak-free pump, thermometers capable of measuring temperature to within 5 , calibrated dry gas meter, and related equipment, to maintain an isokinetic sampling rate and to determine sample volume. The dry gas meter is made by Rockwell and the fiber vane pump is made by Gast.

Barometer - Aneroid type to measure atmospheric pressures to  $\pm 0.1$  inch Hg.

\*Federal Register, CFR 40, Part 60, July 1, 1989

## SAMPLING PROCEDURE

After selecting the sampling site and the minimum number of traverse points, the sack pressure, temperature, moisture and range of velocity head were measured according to the procedures described in the Federal Register\*.

Approximately 200 grams of silica gel was weighed in a sealed impinger prior to each test. Glass fiber filters\*\* (4 inch diameter) were desiccated for at least 24 hours, dried at 105 C for 2 hours and weighed to the nearest 0.1 mg on an analytical balance. One hundred ml of distilled water was placed in each of the first two impingers; the third impinger was initially empty; and the impinger containing the silica gel was placed next in series. The train was set up with the probe as shown in Figure A-1. The sampling train was leak-checked at the sampling site prior to each test run by plugging the inlet to the nozzle and pulling 15 inch Hg vacuum, and at the conclusion of the test by plugging the inlet to the nozzle and pulling a vacuum equal to the highest vacuum reached during the test run.

The pitot tube and line were leak-checked at the test site prior to an following the initial velocity traverse. The check was made by blowing into the impact opening of the pitot tube\* until 3 or more inches of water were recorded on the manometer and then capping the impact opening and holding it for 15 seconds to assure it was leak free. The static pressure side of the pitot tube was leak checked using the same procedure, except suction was used to obtain the 3 inch H<sub>2</sub>O manometer reading. Crushed ice was placed around the impingers to keep the temperature of the gases leaving the last impinger at 68 F or less.

During sampling, stack gas and sampling train data were recorded at each sampling point and when significant changes in stack flow conditions occurred. Isokinetic sampling rates were set throughout the sampling period with the said of a nomograph or calculator. All sampling were recorded on the Particulate Field Data Sheet.

\* Federal Register, CFR 40, Part 60, July 1, 1989

\*\*934 AH type

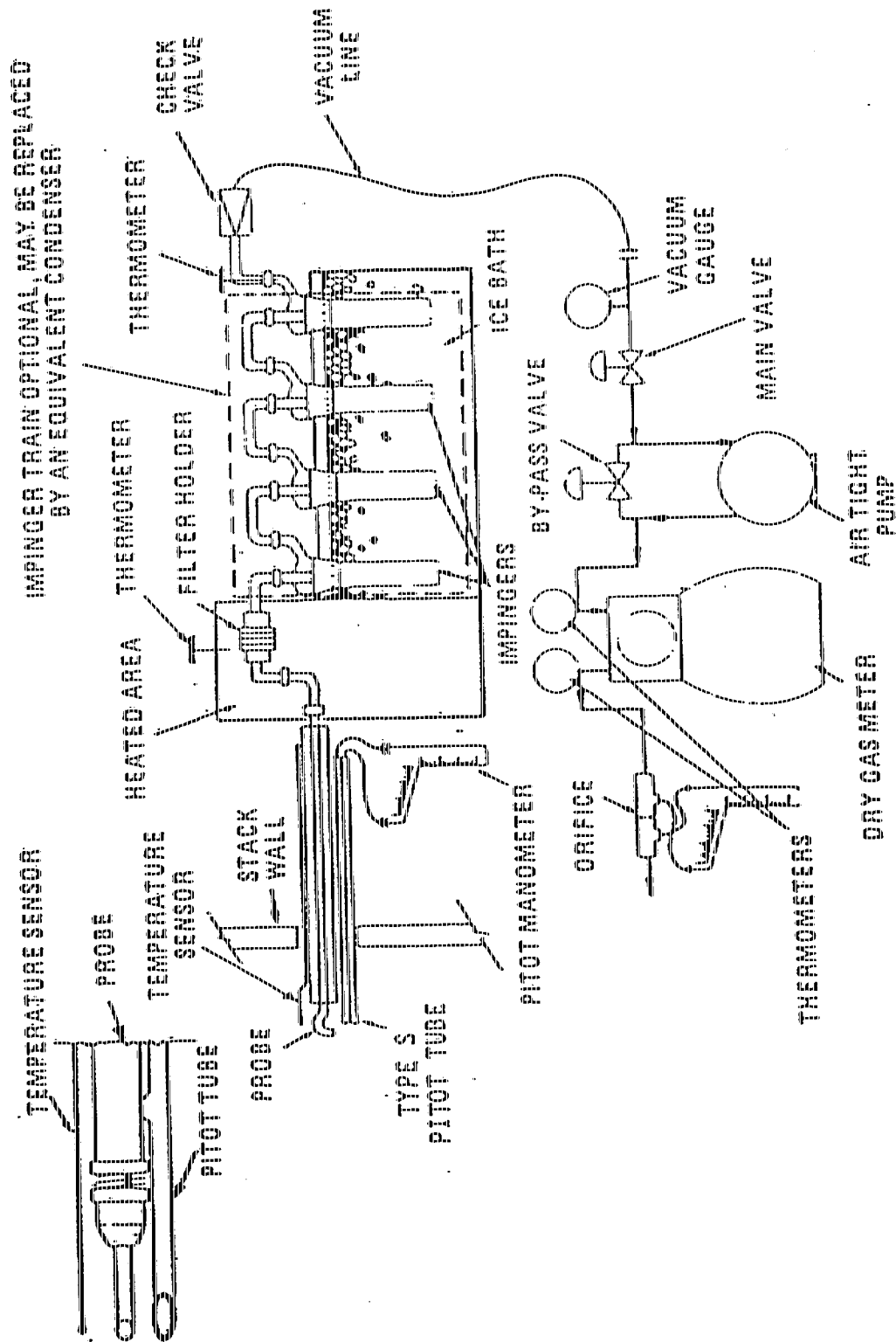


Figure 1. Schematic of Method 5 sampling train.

## SAMPLE RECOVERY PROCEDURE

The sampling train was moved carefully from the test site to the cleanup area. Samples of the acetone and distilled water used in the sample recovery were taken for use as blanks. The volume of water from the first three impingers was measured. Sample fractions were recovered as follows:

Container No. 1 - The filter was removed from its holder and placed in a petri dish and sealed.

Container No. 2 - Loose particulate and H<sub>2</sub>O washings from all sample-exposure surface prior to the filter were placed in a sample container, sealed and labelled. Particulate was removed from the probe with the aid of a brush and H<sub>2</sub>O rinsing. The liquid level was marked after the container was sealed.

Container No. 3 - Loose particulate and acetone washings from all sample-exposure surface prior to the filter were placed in a sample container, sealed and labeled. Particulate was removed from the probe with the aid of a brush and acetone rinsing. The liquid level was marked after the container was sealed.

Container No. 4 - A minimum of 200 ml of acetone was taken for the blank analysis. The blank was obtained and treated in a similar manner as the acetone washing.

Container No. 5 - Distilled water in the impinger section of the sampling train was measured and placed in a sample container. The impingers and connecting glassware were rinsed with distilled H<sub>2</sub>O and this rinse was added to the container for shipment to the laboratory.

Container No. 6 - The impinger section of the sampling train of the connecting glassware was rinsed with acetone and this rinse was put in a container for shipment to the laboratory.

Container No. 7 - A minimum of 200 ml of distilled water was taken for the blank analysis. The blank was obtained and treated in a similar manner as the water rinse.

The silica gel from the fourth impinger was weighed and the gain recorded on the Sample Recovery Data Sheet with other pertinent data.

## ANALYTICAL PROCEDURES

The following procedures were used and follow the methods described in the DER Source Testing Manual\*.

Container No. 1 - The filter and any loosed particulate matter from this sample container were placed into a tared glass weighing dish, baked at 105 C for 2 hours, desiccated for 24 hours to a constant weight and weighed to the nearest 0.1 mg.

Container No. 2 - The H<sub>2</sub>O washings were transferred to a tared beaker and evaporated to dryness at 105 C temperature and pressure, desiccated for 24 hours to a constant weight, and weighed to the nearest 0.1 mg.

Container No. 3 - the acetone washings were transferred to a tared beaker and evaporated to dryness at ambient temperature and pressure, desiccated for 24 hours to a constant weight, and weighed to the nearest 0.1 mg.

Container No. 4 - The acetone blank was transferred to a tared breaker and evaporated to dryness at ambient temperature and pressure. The blank was then desiccated for 24 hours to a constant weight and weighed to the nearest 0.1 mg.

Container No. 5 - The contents of this container were filtered through 0.8-.22 micron tared filters to remove insoluble particulate. The filters and filtrates in tared breakers were evaporated to dryness at 105 C, then desiccated to a constant weight and weighed on an analytical balance to the nearest 0.1 mg.

Container No. 6 - The acetone was transferred to a tared beaker and evaporated to dryness at ambient temperature and pressure. The sample was then desiccated for 24 hours to a constant weight and weighed to the nearest 0.1 mg.

Container No. 7 - The distilled water blank was transferred to a tared beaker and evaporated to dryness at 105 C. The blank was desiccated to a constant weight and weighed on analytical balance to the nearest 0.1 mg.

The term "constant weight" means a difference of no more than 0.5 mg or 1% of total weight less tare weight, whichever is greater between two consecutive readings, with no less than 6 hours of desiccation between weighings.



EPA Method 9  
Visual Determination of the  
Opacity of Emissions from Stationary Sources

Determination of Opacity of Emissions from  
Stationary Sources

EPA Method 9

The following method was used in this test program. Observation procedures follow those described in EPA Method 9 of the Federal Register \*.

Observation Procedures

The opacity of emissions from stationary sources was determined visually by a qualified observer. The qualified observer stood at a distance sufficient to provide a clear view of the emissions with the sun oriented in the 140 sector to his back. Consistent with maintaining the above requirement, the observer made his observations from a position such that his line of vision was approximately perpendicular to the plume direction, and approximately perpendicular to the longer axis of the outlet. The observer recorded the name of the plant, emissions location, type facility, observer's name and affiliation, a sketch of the observer's position relative to the source, and the date on a field data sheet (Figure 1). The time, estimated distance to the emission location, approximate wind direction, estimated wind speed, description of the sky condition (presence and color of clouds), and plume background were recorded on a field data sheet at the time opacity readings were taken.

Opacity observations were made at the point of greatest opacity in the portion of the plume where condensed water vapor was not present. The observer did not look continuously at the plume, but momentarily at 15-second intervals.

SOURCE NAME			OBSERVATION DATE				START TIME		STOP TIME			
ADDRESS			SEC MIN	0	15	30	45	SEC MIN	0	15	30	45
			1					31				
			2					32				
			3					33				
			4					34				
			5					35				
			6					36				
			7					37				
			8					38				
			9					39				
			10					40				
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			22					52				
			23					53				
			24					54				
			25					55				
			26					56				
			27					57				
			28					58				
			29					59				
			30					60				
Source Layout Sketch      Draw North Arrow 			AVERAGE OPACITY FOR HIGHEST PERIOD				NUMBER OF READINGS ABOVE % WERE					
			RANGE OF OPACITY READINGS MINIMUM      MAXIMUM									
			OBSERVER'S NAME (PRINT)									
COMMENTS			OBSERVER'S SIGNATURE					DATE				
			ORGANIZATION									
I HAVE RECEIVED A COPY OF THESE OPACITY OBSERVATIONS			CERTIFIED BY					DATE				
SIGNATURE												
TITLE			DATE					VERIFIED BY				
								DATE				

If condensed water vapor was present within the plume as it emerged from the emission outlet, opacity observations were made beyond the point in the plume at which condensed water vapor was no longer visible. The observer recorded the approximate distance from the emission outlet to the point in the plume at which the observations were made.

If water vapor in the plume condensed and became visible at a distinct distance from the emission outlet, the opacity of emissions were evaluated at the emission outlet prior to the condensation of water vapor and the formation of the steam plume.

Opacity observations were recorded to the nearest 5 percent at 15-second intervals on an observational record sheet (Figure 1). A minimum of 24 observations were recorded. Each momentary observation recorded was deemed to represent the average opacity of emissions for a 15-second period.

Opacity was determined as an average of 24 consecutive observations recorded at 15-second intervals. The observations recorded on the record sheet were divided into sets of 24 consecutive observations. A set was composed of any 24 consecutive observations. For each set of 24 observations, the average was calculated by summing the opacity of the 24 observations and dividing this sum by 24.

To receive certification as a qualified observer, a candidate was tested and demonstrated the ability to assign opacity readings in 5 percent increments to 25 different black plumes and 25 different white plumes, with an error not to exceed 15 percent opacity on any one reading and an average error not exceed 7.5 percent opacity in each category.

The certification is valid for a period of 6 months, at which time the qualification procedure must be repeated by any observer in order to retain certification. The certification test consists of showing the candidate a complete run of 50 plumes-25 black plumes and 25 white plumes-generated by a smoke generator. Plumes within each set of 25 black and 25 white runs were presented in random order. The candidate assigned an opacity value to each plume and recorded his observation on a suitable form. At the completion of each run of 50 readings, the score of the candidate was determined. If a candidate failed to qualify, the complete run of 50 readings was repeated in any retest. The smoke test was administered as part of a smoke school and was preceded by training of familiarization runs of the smoke generator during which candidates were shown black and white plumes of known opacity.

# METER BOX CALIBRATION DATA AND CALCULATION FORM

Date 8/9/90

Meter box number 444

Barometric pressure,  $P_b =$  29.05 in. Hg Calibrated by Kevin Kennedy

Orifice manometer setting ( $\Delta H$ ), in. $H_2O$	Gas volume		Temperatures				Time ( $\theta$ ), min	$Y_i$	$\Delta H @$ in. $H_2O$
	Wet test meter ( $V_w$ ), ft <sup>3</sup>	Dry gas meter ( $V_d$ ), ft <sup>3</sup>	Wet test meter ( $t_w$ ), °F	Dry gas meter					
				Inlet ( $t_{d_i}$ ), °F	Outlet ( $t_{d_o}$ ), °F	Avg <sup>a</sup> ( $t_d$ ), °F			
0.5	5	348.589 353.836	73°F	101 105	82 85	93	12:00	.987	1.61
1.0	5	357.515 362.751	73°F	105 120	88 89	100	9:30	.993	
1.5	10	363.123 373.203	73°F	108 110	87 87	98	13:00	.986	1.41
2.0	10	374.035 384.653	73°F	111 121	88 91	103	11:40	.990	
3.0	10	385.132 395.751	73°F	119 125	88 90	106	10:00	.992	1.64
4.0	10	401.271 411.889	73°F	119 125	88 92	106	8:50	.991	
Avg								.990	1.55

$\Delta H$ , in. $H_2O$	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)}$	$\Delta H @ i = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[ \frac{(t_w + 460) \theta}{V_w} \right]^2$
0.5	0.0368	5.2905 (29.05) (533) 5.271 (29.01) (533) 5.2995 (29.05) (533)	
1.0	0.0737	5.276 (29.12) (533) 10.6105 (29.05) (533)	
1.5	0.110	10.58 (29.16) (533) 10.6105 (29.05) (533)	
2.0	0.147	10.618 (29.22) (533) 10.6105 (29.05) (533)	
3.0	0.221	10.623 (29.27) (533) 10.6105 (29.05) (533)	
4.0	0.294	10.611 (29.34) (533)	

<sup>a</sup> If there is only one thermometer on the dry gas meter, record the temperature under  $t_d$ .

# METER BOX CALIBRATION DATA AND CALCULATION FORM

Date 8/6/90

Meter box number 443

Barometric pressure,  $P_b =$  28.9 in. Hg Calibrated by Kevin Kennedy

Orifice manometer setting ( $\Delta H$ ), in. $H_2O$	Gas volume		Temperatures				Time ( $\theta$ ), min	$y_i$	$\Delta H @$ in. $H_2O$
	Wet test meter ( $V_w$ ), ft <sup>3</sup>	Dry gas meter ( $V_d$ ), ft <sup>3</sup>	Wet test meter ( $t_w$ ), °F	Dry gas meter					
				Inlet ( $t_{d_i}$ ), °F	Outlet ( $t_{d_o}$ ), °F	Avg <sup>a</sup> ( $t_d$ ), °F			
0.5	5	420.900 426.150	72°F	108 120	79 83	97	12:10	.996	
1.0	5	426.260 431.560	73°F	122 103	88 91	101	8:14	.990	1.47
1.5	10	435.000 445.327	73°F	133 105	92 79	102	13:00	1.017	
2.0	10	445.942 456.395	73°F	111 115	80 84	97	8:30 <sup>50</sup>	.994	
3.0	10	456.939 467.998	73°F	111 140	85 90	106	10:00	.998	1.65
4.0	10	468.026 478.629	73°F	112 140	86 90	107	8:39 <sup>50</sup>	.993	1.64
Avg								.998	

$\Delta H$ , in. $H_2O$	$\frac{\Delta H}{13.6}$	$Y_i = \frac{V_w P_b (t_d + 460)}{V_d (P_b + \frac{\Delta H}{13.6}) (t_w + 460)}$	$\Delta H @_i = \frac{0.0317 \Delta H}{P_b (t_d + 460)} \left[ \frac{(t_w + 460) \theta^2}{V_w} \right]$
0.5	0.0368	$\frac{5(28.9)(552)}{5.25(28.97)(532)} = .996$	
1.0	0.0737	$\frac{5(28.9)(561)}{5.30(28.97)(533)} = .990$	
1.5	0.110	$\frac{10(28.9)(562)}{10.327(29.01)(533)} = 1.017$	
2.0	0.147	$\frac{10(28.9)(557)}{10.453(29.00)(533)} = .994$	
3.0	0.221	$\frac{10(28.9)(560)}{10.584(29.12)(533)} = .998$	
4.0	0.294	$\frac{10(28.9)(562)}{10.603(29.19)(533)} = .993$	

<sup>a</sup> If there is only one thermometer on the dry gas meter, record the temperature under  $t_d$ .

# POSTTEST DRY GAS METER CALIBRATION DATA FORM

*Edward B. Maffery* 9-10-96

Test numbers \_\_\_\_\_ Date 9-10-90 Meter box number 444 Plant \_\_\_\_\_  
 Barometric pressure,  $P_b = 29.40$  in. Hg Dry gas meter number 806 Pretest  $Y = .990$

Orifice manometer setting, ( $\Delta H$ ), in. H <sub>2</sub> O	Gas volume		Temperature			Time ( $\Theta$ ), min	Vacuum setting, in. Hg	$Y_i$	$Y_i$	
	Wet test meter ( $V_w$ ), ft <sup>3</sup>	Dry gas meter ( $V_d$ ), ft <sup>3</sup>	Wet test meter ( $t_w$ ), °F	Inlet ( $t_{d_i}$ ), °F	Outlet ( $t_{d_o}$ ), °F	Average ( $t_d$ ), °F			$V_w P_b (t_d + 460)$	$V_d P_b + \frac{\Delta H}{13.6} t_w + 460$
2"	10	10.456	72	104/110	86/90	97	2.29	.996	10.456(29.4)(557)	10.456(29.4)(532)
2"	10	10.498	72	104/117	88/91	100	2.30	.998	10.498(29.4)(560)	10.498(29.4)(532)
2"	10	10.532	72	109/120	88/94	103	2.32	.999	10.532(29.4)(574)	10.532(29.4)(532)
								$Y = .998$		

a If there is only one thermometer on the dry gas meter, record the temperature under  $t_d$ .

$V_w$  = Gas volume passing through the wet test meter, ft<sup>3</sup>

$V_d$  = Gas volume passing through the dry gas meter, ft<sup>3</sup>

$t_w$  = Temperature of the gas in the wet test meter, °F.

$t_{d_i}$  = Temperature of the inlet gas of the dry gas meter, °F.

$t_{d_o}$  = Temperature of the outlet gas of the dry gas meter, °F.

$t_d$  = Average temperature of the gas in the dry gas meter, obtained by the average of  $t_{d_i}$  and  $t_{d_o}$ , °F.

$\Delta H$  = Pressure differential across orifice, in H<sub>2</sub>O.

$Y_i$  = Ratio of accuracy of wet test meter to dry gas meter for each run.

$Y$  = Average ratio of accuracy of wet test meter to dry gas meter for all three runs;  
 tolerance = pretest  $Y \pm 0.05Y$

$P_b$  = Barometric pressure, in. Hg.

$\Theta$  = Time of calibration run, min.



# POSTTEST DRY GAS METER CALIBRATION DATA FORM

James D. M. Kelly 9-16-90

Test numbers \_\_\_\_\_ Date 9-10-90 Meter box number 443 Plant \_\_\_\_\_  
 Barometric pressure,  $P_b = 29.40$  in. Hg Dry gas meter number 806 Pretest  $Y = .995$

Orifice manometer setting, ( $\Delta H$ ), in. H <sub>2</sub> O	Gas volume		Temperature			Time ( $\theta$ ), min	Vacuum setting, in. Hg	$Y_i$	$Y_i$
	Wet test meter ( $V_w$ ), ft <sup>3</sup>	Dry gas meter ( $V_d$ ), ft <sup>3</sup>	Wet test meter ( $t_w$ ), °F	Inlet ( $t_{d_i}$ ), °F	Outlet ( $t_{d_o}$ ), °F	Average ( $t_d$ ), °F			
2"	10	10.667	72	<del>113</del> 136	<del>86</del> 96	108	2"	.996	$V_w P_b (t_d + 460)$ $V_d P_b + \Delta H \frac{t_w}{13.6}$ $\frac{10(29.4)}{13.6} = 56.8$ $\frac{10.667(29.4 + 460)}{10(29.4) + 56.8} = .996$
2"	10	10.704	72	<del>116</del> 138	<del>89</del> 93	109	2"	.994	$\frac{10.704(29.4 + 460)}{10(29.4) + 57.8} = .994$
2"	10	10.757	72	<del>124</del> 139	<del>90</del> 94	112	2"	.994	$\frac{10.757(29.4 + 460)}{10(29.4) + 58} = .994$
								$Y = .995$	

a If there is only one thermometer on the dry gas meter, record the temperature under  $t_d$ .

$V_w$  = Gas volume passing through the wet test meter, ft<sup>3</sup>.

$V_d$  = Gas volume passing through the dry gas meter, ft<sup>3</sup>.

$t_w$  = Temperature of the gas in the wet test meter, °F.

$t_{d_i}$  = Temperature of the inlet gas of the dry gas meter, °F.

$t_{d_o}$  = Temperature of the outlet gas of the dry gas meter, °F.

$t_d$  = Average temperature of the gas in the dry gas meter, obtained by the average of  $t_{d_i}$  and  $t_{d_o}$ , °F.

$\Delta H$  = Pressure differential across orifice, in H<sub>2</sub>O.

$Y_i$  = Ratio of accuracy of wet test meter to dry gas meter for each run.

$Y$  = Average ratio of accuracy of wet test meter to dry gas meter for all three runs;

tolerance = pretest  $Y \pm 0.05Y$

$P_b$  = Barometric pressure, in. Hg.

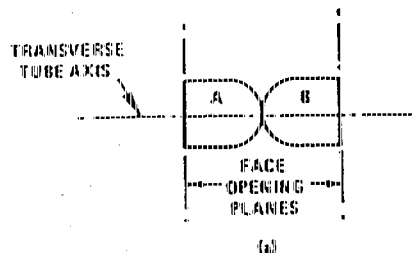
$\theta$  = Time of calibration run, min.

# PITOT TUBE INSPECTION DATA SHEET

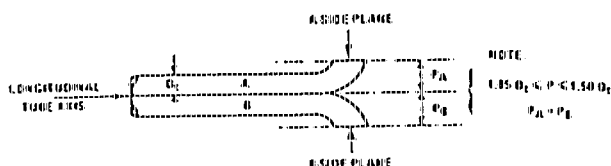
Pitot Tube Identification Number: #7

Date: 2/23/90

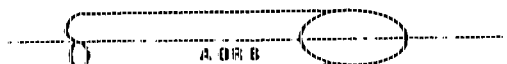
1. What is the external diameter of the tubing in inches?  
3/8"
2. Are the face opening planes perpendicular to the transverse axis as in Figure A (end view)? yes
3. Are the face opening planes parallel to the longitudinal axis as in Figure B (top view)? yes
4. Are both legs of equal length and center lines coincident, when viewed from both sides as in Figure (C)? yes
5. Does the tube meet all the geometric specifications to be assigned a .84 coefficient? yes



(a)



(b)



(c)

CALIBRATED BY:

Kenn Kennedy

# NOZZLE CALIBRATION DATA FORM

Date 8/23/90 Calibrated by R. Campbell

Nozzle identification number	$D_1$ (in.)	$D_2$ (in.)	$D_3$ (in.)	$\Delta D$ (in.)	$D_{avg}$
#4	.360	.360	.360	0.0	.360

where:

$D_{1,2,3}$  = Three different nozzles diameters, in.; each diameter measure within 0.001 in.

$\Delta D$  = Maximum difference between any two diameters, in.,  
 $\Delta D \leq 0.004$  in.

$D_{avg}$  = Average of  $D_1$ ,  $D_2$ , and  $D_3$ .

## Visible Emission Observation Form

SOURCE NAME			OBSERVATION DATE				START TIME				STOP TIME			
Comm Stone			8/27/90				8:50				10:40			
ADDRESS			Springfield Pike											
CITY			STATE				ZIP							
PHONE			SOURCE ID NUMBER											
PROCESS EQUIPMENT			OPERATING MODE											
CONTROL EQUIPMENT			OPERATING MODE											
DESCRIBE EMISSION POINT														
START			STOP											
HEIGHT ABOVE GROUND LEVEL			HEIGHT RELATIVE TO OBSERVER											
START			STOP											
DISTANCE FROM OBSERVER			DIRECTION FROM OBSERVER											
START			STOP											
DESCRIBE EMISSIONS														
START			STOP											
EMISSION COLOR			PLUME TYPE: CONTINUOUS <input type="checkbox"/>											
START			STOP											
WATER DROPLETS PRESENT:			IF WATER DROPLET PLUME											
NO <input type="checkbox"/> YES <input type="checkbox"/>			ATTACHED <input type="checkbox"/> DETACHED <input type="checkbox"/>											
POINT IN THE PLUME AT WHICH OPACITY WAS DETERMINED														
START			STOP											
DESCRIBE BACKGROUND														
START			STOP											
BACKGROUND COLOR			SKY CONDITIONS											
START			STOP											
WIND SPEED			WIND DIRECTION											
START			STOP											
AMBIENT TEMP.			WET BULB TEMP				RH, percent							
START			STOP											
<p>Source Layout Sketch</p> <p>Draw North Arrow</p> <p>X Emission Point</p> <p>Observer's Position</p> <p>Sun</p> <p>Wind</p> <p>Plume and Stack</p> <p>Sun Location Line</p> <p>40°</p>			24				54							
			25				55							
			26				56							
			27				57							
			28				58							
			29				59							
			30				60							
			AVERAGE OPACITY FOR HIGHEST PERIOD				NUMBER OF READINGS ABOVE % WERE							
			RANGE OF OPACITY READINGS				MINIMUM							
			OBSERVER'S NAME (PRINT)				MAXIMUM							
OBSERVER'S SIGNATURE				DATE										
ORGANIZATION														
I HAVE RECEIVED A COPY OF THESE OPACITY OBSERVATIONS				CERTIFIED BY										
SIGNATURE				DATE										
TITLE				DATE										

# VISIBLE EMISSION OBSERVATION FORM

No.

COMPANY NAME <i>Comm Stone</i>		
STREET ADDRESS <i>Springfield Pike</i>		
CITY <i>Cannonsville</i>	STATE <i>PA</i>	ZIP
PHONE (KEY CONTACT)		SOURCE ID NUMBER

PROCESS EQUIPMENT	OPERATING MODE
CONTROL EQUIPMENT	OPERATING MODE

DESCRIBE EMISSION POINT

HEIGHT ABOVE GROUND LEVEL	HEIGHT RELATIVE TO OBSERVER
	Start End

DISTANCE FROM OBSERVER	DIRECTION FROM OBSERVER
Start <i>40</i> End	Start <i>SW</i> End

DESCRIBE EMISSIONS	
Start	End
EMISSION COLOR	IF WATER DROPLET PLUME
Start End	Attached <input type="checkbox"/> Detached <input type="checkbox"/>

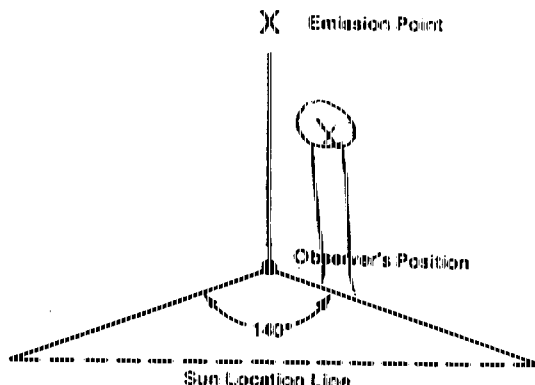
POINT IN THE PLUME AT WHICH OPACITY WAS DETERMINED
Start End

DESCRIBE PLUME BACKGROUND	
Start <i>HAZY SKY</i> End	SKY CONDITIONS
BACKGROUND COLOR	Start <i>HAZY</i> End
Start <i>Blue</i> End	WIND DIRECTION
WIND SPEED	Start <i>EAST</i> End
Start <i>0</i> End	WET BULB TEMP
AMBIENT TEMP	RH, percent
Start <i>60</i> End	

Stack with Plume	Sun	Wind

SOURCE LAYOUT SKETCH

Draw North Arrow



ADDITIONAL INFORMATION
<i>TEST-1</i>

OBSERVATION DATE <i>8/27/90</i>					START TIME <i>8:50</i>	END TIME <i>10:46</i>
SEC MIN	0	15	30	45	COMMENTS	
1	0	0	0	0		
2	0	0	0	0		
3	0	0	0	0		
4	0	0	0	0		
5	0	0	0	0		
6	0	0	0	0		
7	0	0	0	0		
8	0	0	0	0		
9	0	0	0	0		
10	0	0	0	0		
11	0	0	0	0		
12	0	0	0	0		
13	0	0	0	0		
14	0	0	0	0		
15	0	0	0	0		
16	0	0	0	0		
17	0	0	0	0		
18	0	0	0	0		
19	0	0	0	0		
20	0	0	0	0		
21	0	0	0	0		
22	0	0	0	0		
23	0	0	0	0		
24	0	0	0	0		
25						
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28						
29						
30						

OBSERVER'S NAME (PRINT)	<i>Kevin Kennedy</i>
OBSERVER'S SIGNATURE	<i>Kevin Kennedy</i>
DATE	<i>8/27/90</i>
ORGANIZATION	<i>PSC INC</i>
CERTIFIED BY	<i>ETA</i>
DATE	<i>MARCH 1990</i>

CONTINUED ON VEO FORM NUMBER
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## Visible Emission Observation Form

SOURCE NAME			OBSERVATION DATE				START TIME		STOP TIME			
Comm. Stanz			8/27/90				11:10					
ADDRESS			Springfield Pike									
CITY			STATE		ZIP							
PHONE			SOURCE ID NUMBER									
PROCESS EQUIPMENT			OPERATING MODE									
CONTROL EQUIPMENT			OPERATING MODE									
DESCRIBE EMISSION POINT												
START			STOP									
HEIGHT ABOVE GROUND LEVEL			HEIGHT RELATIVE TO OBSERVER									
START			STOP		START		STOP					
DISTANCE FROM OBSERVER			DIRECTION FROM OBSERVER									
START			STOP		START		STOP					
DESCRIBE EMISSIONS												
START			STOP									
EMISSION COLOR			PLUME TYPE CONTINUOUS <input type="checkbox"/>									
START			STOP		FUGITIVE <input type="checkbox"/>		INTERMITTENT <input type="checkbox"/>					
WATER DROPLETS PRESENT			IF WATER DROPLET PLUME									
NO <input checked="" type="checkbox"/> YES <input type="checkbox"/>			ATTACHED <input type="checkbox"/> DETACHED <input type="checkbox"/>									
POINT IN THE PLUME AT WHICH OPACITY WAS DETERMINED												
START			STOP									
DESCRIBE BACKGROUND												
START			STOP									
BACKGROUND COLOR			SKY CONDITIONS									
START			STOP		START		STOP					
WIND SPEED			WIND DIRECTION									
START			STOP		START		STOP					
AMBIENT TEMP.			WET BULB TEMP.		RH. percent							
START			STOP									
Source Layout Sketch			Draw North Arrow									
<p>Source Layout Sketch</p> <p>Draw North Arrow</p> <p>X Emission Point</p> <p>Sun - Wind -&gt;</p> <p>Plume and Stack</p> <p>Observers Position</p> <p>140°</p> <p>Sun Location Line</p>												
AVERAGE OPACITY FOR HIGHEST PERIOD			NUMBER OF READINGS ABOVE % WERE									
RANGE OF OPACITY READINGS			MINIMUM				MAXIMUM					
OBSERVER'S NAME (PRINT)			Kevin Kennedy									
OBSERVER'S SIGNATURE			Kevin Kennedy				DATE		8/27/90			
ORGANIZATION			CSC INC.									
I HAVE RECEIVED A COPY OF THESE OPACITY OBSERVATIONS			CERTIFIED BY				DATE		March 1990			
SIGNATURE			ETA									
TITLE			DATE				VERIFIED BY		DATE			

# VISIBLE EMISSION OBSERVATION FORM

No. \_\_\_\_\_

COMPANY NAME <i>Comm. Stone</i>		
STREET ADDRESS <i>Springfield Pike</i>		
CITY	STATE <i>PA</i>	ZIP
PHONE (KEY CONTACT)	SOURCE ID NUMBER	

PROCESS EQUIPMENT	OPERATING MODE
CONTROL EQUIPMENT	OPERATING MODE

DESCRIBE EMISSION POINT	
HEIGHT ABOVE GROUND LEVEL	HEIGHT RELATIVE TO OBSERVER Start      End
DISTANCE FROM OBSERVER Start <i>40'</i> End	DIRECTION FROM OBSERVER Start <i>SW</i> End

DESCRIBE EMISSIONS	
Start	End
EMISSION COLOR	IF WATER DROPLET PLUME Attached <input type="checkbox"/> Detached <input type="checkbox"/>
POINT IN THE PLUME AT WHICH OPACITY WAS DETERMINED Start      End	

DESCRIBE PLUME BACKGROUND	
Start <i>HAZY SKY</i>	End
BACKGROUND COLOR Start <i>Blue</i> End	SKY CONDITIONS Start <i>HAZY</i> End
WIND SPEED Start <i>0-2</i> End	WIND DIRECTION Start <i>EAST</i> End
AMBIENT TEMP Start <i>80° F</i> End	WET BULB TEMP      RH, percent

Stack with Plume

Sun

Wind

SOURCE LAYOUT SKETCH

Draw North Arrow

X Emission Point

Observer's Position

Sun Location Line

140°

ADDITIONAL INFORMATION <i>TEST-2</i>
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OBSERVATION DATE <i>8/27/90</i>				START TIME <i>9:10</i>	END TIME <i>12:45</i>
SEC	0	15	30	45	COMMENTS
MIN					
1	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
2	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
3	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
4	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
5	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
6	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
7	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
8	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
9	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
10	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
11	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
12	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	
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OBSERVER'S NAME (PRINT) <i>Kevin Kennedy</i>	
OBSERVER'S SIGNATURE <i>Kevin Kennedy</i>	DATE <i>8/27/90</i>
ORGANIZATION <i>CSC INC.</i>	
CERTIFIED BY <i>ETA</i>	DATE <i>MARCH 1998</i>
CONTINUED ON VEO FORM NUMBER	

## Visible Emission Observation Form

SOURCE NAME <i>Comm Stone</i>			OBSERVATION DATE <i>8/27/90</i>				START TIME <i>1:33</i>		STOP TIME <i>2:56</i>	
ADDRESS <i>Springfield Pike</i>			SEC MIN 0 15 30 45				SEC MIN 0 15 30 45			
CITY			STATE <i>PA</i>				ZIP			
PHONE			SOURCE ID NUMBER							
PROCESS EQUIPMENT			OPERATING MODE							
CONTROL EQUIPMENT			OPERATING MODE							
DESCRIBE EMISSION POINT										
START			STOP							
HEIGHT ABOVE GROUND LEVEL START <i>45'</i> STOP			HEIGHT RELATIVE TO OBSERVER START <i>10'</i> STOP							
DISTANCE FROM OBSERVER START <i>40'</i> STOP			DIRECTION FROM OBSERVER START <i>SW</i> STOP							
DESCRIBE EMISSIONS										
START			STOP							
EMISSION COLOR START STOP			PLUME TYPE: CONTINUOUS <input type="checkbox"/> FUGITIVE <input type="checkbox"/> INTERMITTENT <input type="checkbox"/>							
WATER DROPLETS PRESENT NO <input checked="" type="checkbox"/> YES <input type="checkbox"/>			IF WATER DROPLET PLUME ATTACHED <input type="checkbox"/> DETACHED <input type="checkbox"/>							
POINT IN THE PLUME AT WHICH OPACITY WAS DETERMINED										
START			STOP							
DESCRIBE BACKGROUND										
START <i>HAZY SKY</i> STOP										
BACKGROUND COLOR START <i>BLUE</i> STOP			SKY CONDITIONS START <i>HAZY</i> STOP							
WIND SPEED START <i>0-2 mph</i> STOP			WIND DIRECTION START <i>EAST</i> STOP							
AMBIENT TEMP START <i>80°F</i> STOP			WET BULB TEMP				RH, percent			
Source Layout Sketch			Draw North Arrow							
COMMENTS <i>TEST-3</i>										
I HAVE RECEIVED A COPY OF THESE OPACITY OBSERVATIONS			CERTIFIED BY <i>ETA</i>				DATE <i>MARCH 1990</i>			
SIGNATURE			VERIFIED BY				DATE			
TITLE										



# VISIBLE EMISSION OBSERVATION FORM

No.

COMPANY NAME <i>Comm Stone</i>		
STREET ADDRESS <i>Springfield Pike</i>		
CITY	STATE <i>PA</i>	ZIP
PHONE (KEY CONTACT)		SOURCE ID NUMBER

PROCESS EQUIPMENT	OPERATING MODE
CONTROL EQUIPMENT	OPERATING MODE

DESCRIBE EMISSION POINT	
HEIGHT ABOVE GROUND LEVEL <i>45'</i>	HEIGHT RELATIVE TO OBSERVER Start <i>10'</i> End
DISTANCE FROM OBSERVER Start <i>40'</i> End	DIRECTION FROM OBSERVER Start <i>SW</i> End

DESCRIBE EMISSIONS	
Start	End
EMISSION COLOR	IF WATER DROPLET PLUME Attached <input type="checkbox"/> Detached <input type="checkbox"/>
POINT IN THE PLUME AT WHICH OPACITY WAS DETERMINED	
Start	End

DESCRIBE PLUME BACKGROUND	
Start <i>HAZY SKY</i>	End
BACKGROUND COLOR	SKY CONDITIONS
Start <i>Blue</i> End	Start <i>HAZY</i> End
WIND SPEED	WIND DIRECTION
Start <i>0-2 mph</i> End	Start <i>EAST</i> End
AMBIENT TEMP	WET BULB TEMP RH, percent
Start	End

Stack with Plume Sun Wind	SOURCE LAYOUT SKETCH Draw North Arrow
X Emission Point Observer's Position Sun Location Line	

ADDITIONAL INFORMATION <i>TEST-3</i>
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OBSERVATION DATE <i>8/27/90</i>				START TIME <i>1:33</i>	END TIME <i>2:56</i>
SEC MIN	0	15	30	45	COMMENTS
1	0	0	0	0	
2	0	0	0	0	
3	0	0	0	0	
4	0	0	0	0	
5	0	0	0	0	
6	0	0	0	0	
7	0	0	0	0	
8	0	0	0	0	
9	0	0	0	0	
10	0	0	0	0	
11	0	0	0	0	
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OBSERVER'S NAME (PRINT) <i>Kevin Kennedy</i>	DATE <i>8/27/90</i>
OBSERVER'S SIGNATURE <i>Kevin Kennedy</i>	DATE <i>8/27/90</i>
ORGANIZATION <i>OSCEANO</i>	
CERTIFIED BY <i>ETA</i>	DATE <i>MARCH 1990</i>

CONTINUED ON VEO FORM NUMBER	
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AMBIENT TEMPERATURE 74  
 BAROMETRIC PRESSURE 30.2  
 ASSUMED MOISTURE, % 1A  
 PROBE LENGTH, (in) 44  
 NOZZLE IDENTIFICATION NO. 1A  
 AVERAGE CALIBRATED NOZZLE DIAMETER, (in) 0.006 at 15H  
 PROBE HEATER SETTING 250  
 LEAK RATE, (dlm) 0.006 at 15H  
 PROBE LINER MATERIAL Glass  
 STATIC PRESSURE, (in Hg) +1.20  
 FILTER NO. 1003Z

SCHEMATIC OF STACK CROSS SECTION  
 Location 48H  
 Stack diam. 48H  
 SAMPLE BOX NO. SPS-1  
 METER BOX NO. # 443  
 METER AND KE-10  
 GFACTOR 10  
 PITOT TUBE COEFFICIENT, 0.84

TRAVERSE POINT NUMBER	TIME	SAMPLING TIME (in min.)	VACUUM (in Hg)	STACK TEMPERATURE (°F)	VELOCITY HEAD (in H <sub>2</sub> O)	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER (in H <sub>2</sub> O)	GAS SAMPLE VOLUME (lit)	GAS SAMPLE TEMPERATURE - AT ORIFICE METER		FILTER HOLDER TEMPERATURE, LAST IMPINGING, (°F)	TEMPERATURE OF GAS LEAVING CONDENSER OR LAST IMPINGING, (°F)
								INLET (°F)	OUTLET (°F)		
A-1	8:50	0	10.0	73	.45	8.0	6.800	74	76	248	47
2		3.5	9.9	73	.37	6.6	12.370	123	78	250	47
3		7.0	5.5	73	.28	5.0	17.545	124	82	250	47
4		10.5	5.0	73	.24	4.2	20.210	124	84	255	47
5		14.0	5.0	73	.20	3.6	26.170	125	84	256	48
6		17.5	3.5	74	.16	2.9	29.690	123	86	257	48
7		21	4.5	75	.19	3.4	33.570	124	86	260	48
8		24.5	3.0	74	.14	2.5	37.110	125	86	261	48
9		28	4.5	75	.20	3.6	40.285	126	89	260	49
10		31.5	5.0	74	.24	4.2	44.110	127	89	259	49
11		35	5.0	74	.22	4.1	48.370	132	91	260	49
12		42	5.0	74	.20	3.6	52.510	133	91	258	49
3-1	8:57	45.5	14.0	73	.57	10.0	56.321	127	90	256	51
2		49	13.0	73	.50	9.0	60.360	128	91	255	51
3		52.5	11.0	73	.44	7.8	73.975	133	94	254	51
4		56	9.0	73	.38	6.8	79.360	139	94	256	51
5		59.5	5.5	74	.25	4.5	83.600	141	95	258	52
6		63	2.0	74	.05	0.9	85.400	129	97	257	52
TOTAL								Avg.	Avg.		
AVERAGE				74	.425	3.9	87.500		107		

IMPINGING WATER Final 202 ml  
 DESTICANT Final 244.9 gm  
 COND. Final 225.5 gm  
 COND. Initial 2 ml  
 COND. Final 19.4 gm

[illegible]

AMBIENT TEMPERATURE	72°F
BACKSTRIKING PRESSURE	
ASSUMED MOISTURE %	
PROB LENGTH, IN	
NOZZLE IDENTIFICATION NO.	
AVERAGE CALCULATED NOZZLE DIAMETER, (in.)	
PROBE HEATER SETTING	
CURRENT RATE, (in.)	0.005 in 154
PROBE LINER MATERIAL	Glass
STATIC PRESSURE, (in. Hg)	
WILLARD NO.	100032

**SCHMATIC OF STACK CROSS SECTION**

**Location** \_\_\_\_\_

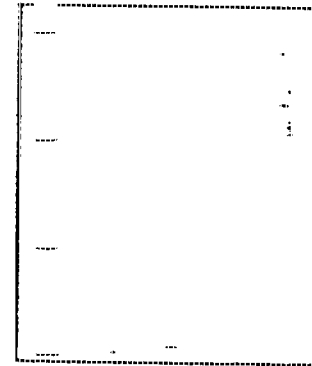
**Date** \_\_\_\_\_

TRAVERSE POINT NUMBER	TIME	SAMPLING TIME min.	VACUUM (in. Hg)	STACK TEMPERATURE (°F)	VELOCITY HEAD (AP <sub>s</sub> ) (ft./sec.)	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER (in. H <sub>2</sub> O)	GAS SAMPLE VOLUME (lit.)	GAS SAMPLE TEMPERATURE - AT ORY GAS METER		FILTER HOLDER TEMPERATURE (°F)	TEMPERATURE OF GAS LEAVING OR CONDENSER OR LAST IMPINGER, (°F)
								INLET (°F)	OUTLET (°F)		
B-7	66.5		1.0	74	.04	0.10	87.130	120	95	259	51
8	70		1.0	74	.02	0.36	88.600	120	96	260	51
9	73.5		1.0	74	.02	0.36	90.010	120	93	261	50
10	77		1.0	74	.02	0.36	91.400	119	95	260	50
11	80.5		1.0	74	.02	0.36	92.905	119	93	257	50
12	84	10:27	1.0	74	.02	0.36	94.300	119	92	256	51
TOTAL								Avg.			
AVERAGE								Avg.			

Impinger Water	Desiccant
100 ml	100 ml
100 ml	100 ml
100 ml	100 ml

[illegible]

AMBIENT TEMPERATURE \_\_\_\_\_  
 BAROMETRIC PRESSURE 29.6  
 ASSUMED MOISTURE, % \_\_\_\_\_  
 PROBE LENGTH, (in) \_\_\_\_\_  
 NOZZLE IDENTIFICATION NO. #4  
 AVERAGE CALIBRATED NOZZLE DIAMETER, (in) .360  
 PROBE HEATER SETTING 250  
 LEAK RATE, (cc/min) 0.0003 15"  
 PROBE LINER MATERIAL Glass  
 STATIC PRESSURE, (in. Hg) 1.20  
 FILTER NO. 100021



SCHEMATIC OF STACK CROSS-SECTION  
 Location \_\_\_\_\_  
 Stack diam. 48"

PLANT LAUREL STATION  
 LOCATION Spent Acid Pond  
 OPERATOR P. Campbell  
 DATE 8/27/96  
 RUN NO. SPS-2  
 SAMPLE BOX NO. \_\_\_\_\_  
 METER BOX NO. 4417 44444  
 METER ANGLE RF  
 PIVOT TUBE COEFFICIENT, % 84

TRAVERSE POINT NUMBER	SAMPLING TIME (in. min.)		VACUUM (in. Hg)	STACK TEMPERATURE (°F)	VELOCITY HEAD (in. H <sub>2</sub> O)	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER (in. H <sub>2</sub> O)	GAS SAMPLE VOLUME (in <sup>3</sup> )	GAS SAMPLE TEMPERATURE "AT DRY GAS METER"		FILTER HOLDER TEMPERATURE (°F)	TEMPERATURE OF GAS LEAVING CONDENSER OR LAST IMPIINGER (°F)
	TIME	IN. MIN.						INLET (°F)	OUTLET (°F)		
1	11:10	0					90-150	91	91	262	41
2		3	13.0	80	.55	9.8	101.410	128	97	263	41
3		6	13.0	80	.55	9.8	106.770	139	99	260	41
4		7	12.0	80	.46	8.2	111.760	147	99	261	42
5		12	9.0	79	.37	6.6	116.910	154	104	264	42
6		15	8.0	79	.30	5.4	120.480	152	106	269	42
7		18	5.0	79	.17	3.8	123.570	139	106	272	42
8		21	2.0	79	.10	1.8	125.820	136	105	268	42
9		24	1.0	78	.06	1.1	124.530	130	103	269	42
10		27	1.0	78	.04	0.70	129.025	127	105	264	42
11		30	1.0	78	.02	0.36	130.340	123	103	260	43
12		33	1.0	78	.02	0.36	131.670	120	102	258	42
A-1	Stop /	36	1.0	75	.45	8.0	133.637	95	87	260	44
2		39	10.0	75	.45	8.0	135.720	98	87	259	44
3		42	11.0	75	.45	8.0	136.740	111	87	256	44
4		45	8.0	78	.35	6.2	135.000	115	86	253	44
5		48	7.0	76	.32	5.6	138.530	117	87	254	45
6		51	5.0	76	.24	4.3	142.180	117	88	255	45
7		54	4.0	76	.15	2.7	145.770	117	88	255	45
TOTAL						72 (38)	136.887	Avg. (109)	Avg.		
AVERAGE				78	.432	(40)	(40.405)	Avg. (101)			

IMPIINGER WATER Final 284 ml  
 Int. 200 ml  
 Cond. 9 ml  
 DESTIGANT Final 214.7 gms  
 Int. 199.1 gms  
 gms 15.6 gms

ORISAT ANALYSES

CO	CO <sub>2</sub>	CH <sub>4</sub>	HCN
0.0	0.0	0.0	0.0

\* Changed Meters  
 Pump stopped  
 681.400 new meter

SCHEMATIC OF STAR CROSS SECTION  
 Location \_\_\_\_\_  
 Scale: 1 cm. = 100 ft.

SERIAL	DATE
BAROMETRIC PRESSURE	28.6
ASSUMED ALTITUDE	
PROBE LENGTH, (in)	
NOZZLE IDENTIFICATION NO.	
AVERAGE CALCULATED NOZZLE DIAMETER, (in)	2.50
PROBE WATER PASTING	
TEMPERATURE, (°m)	0.00 ± 1.5"
PROBE INNER MATERIAL	Class
STATION PRESSURE	(in. Hg)
INTERNO.	100021

[illegible][illegible]

01			
02			
03			
04			
05			
06			
07			
08			
09			
10			

PLANT Cam-Sand  
 LOCATION St. Paul Island (SAND)  
 OPERATOR 10 km/m  
 DATE 6/27/66  
 RUN NO. SPS-3  
 SAMPLE BOX NO. #444  
 METER BOX NO. #444  
 METER ANG. KF-18  
 G FACTOR 1.84  
 PIVOT TUBE COEFFICIENT, 0.84

SCHEMATIC OF STACK CROSS SECTION  
 Location 48W  
 Stack diam. 48W

BUREAU PERAL  
 BAROMETRIC PRESSURE 28.6  
 ASSUMED MOISTURE, %  
 PROBE LENGTH, (in)  
 NOZZLE IDENTIFICATION NO. #4  
 AVERAGE CALIBRATED NOZZLE DIAMETER, (in) 2.60  
 PROBE HEATER SETTING 2.50  
 LEAK RATE, (in)  
 PROBE LINER MATERIAL Glass  
 STATIC PRESSURE, (in. Hg) 4.20  
 FILTER NO. 108030

TRAVEL POINT NUMBER	TIME (hr:min)	SAMPLING TIME (hr:min)	VACUUM (in. Hg)	STACK TEMPERATURE (°F)	VELOCITY HEAD (ft/sec)	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER (in. H <sub>2</sub> O)	GAS SAMPLE VOLUME (ft <sup>3</sup> )	GAS SAMPLE TEMPERATURE "AT DRY GAS METER"		TEMPERATURE OF GAS LEAVING CONDENSER OR FILTER HOUSING (°F)	TEMPERATURE OF GAS LEAVING CONDENSER OR FILTER HOUSING (°F)
								INLET (°F)	OUTLET (°F)		
1-1	12:30	0	10.0	81	.45	8.0	721.840	93	93	262	50
2		3	10.0	81	.45	8.0	726.670	127	93	266	50
3		6	7.0	80	.32	5.8	731.510	125	94	268	50
4		9	6.0	80	.25	4.4	735.680	134	92	270	50
5		12	5.0	80	.20	3.6	737.330	134	92	264	50
6		15	5.0	79	.14	2.5	740.740	133	90	260	51
7		18	3.5	79	.10	1.8	745.540	130	93	260	51
8		21	4.0	79	.15	2.7	748.665	131	94	259	51
9		24	5.0	78	.20	3.6	751.130	130	98	261	53
10		27	5.5	78	.25	4.4	754.590	136	97	264	53
11		30	5.5	78	.25	4.4	758.390	138	96	264	52
12	5:07	33	5.0	78	.20	3.6	762.220	137	96	261	52
13	5:07	36	13.0	76	.55	9.8	765.523	119	89	259	54
14		39	13.0	77	.48	8.5	770.870	136	88	259	54
15		42	13.0	77	.46	8.2	775.370	138	89	261	57
16		45	5.0	77	.15	2.7	780.295	133	88	259	54
17		48	5.0	77	.15	2.7	785.670	134	88	260	54
18		51	5.0	78	.10	1.8	788.590	136	88	260	54
19		54	5.0	78	.10	1.8	789.005	136	88	260	54
TOTAL								Av.	Av.		
AVERAGE				78	.430	3.8	77.275	Av.	Av.	///	

C-049

ORSAT ANALYSES

Time	CO <sub>2</sub>	O <sub>2</sub>	CO
Final	0.0	0.0	0.0
Init.	21.0	21.0	21.0

IMPINGER WATER  
 Final 2476 ml  
 Init. 226.4 ml  
 Cond. 8 ml  
 Desiccant  
 Final 2476 gm  
 Init. 226.4 gm  
 Gain 212 gm

RUN: SP-37-90  
 LOCATION: SP-37-90 (dark) (SP-37)  
 OPERATOR: R. Campbell  
 DATE: 9-27-90  
 RUN NO.: SPS-3  
 SAMPLE BOX NO.: 444  
 METR BOX NO.: 444  
 METR ANG: 444  
 CFACOR: 444  
 PILOT TUBE COEFFICIENT, C: 444

SCHEMATIC OF STACK CROSS SECTION  
 Located for  
 Stack diam.

UNITED STATES  
DEPARTMENT OF JUSTICE  
FEDERAL BUREAU OF INVESTIGATION  
WASHINGTON, D. C. 20535

TRAVERSE POINT NUMBER	TIME	SAMPLING TIME (ft, min.)	VACUUM (in. Hg)	STACK TEMPERATURE (°F)	VELOCITY HEAD (in. H <sub>2</sub> O)	PRESSURE DIFFERENTIAL ACROSS ORIFICE METER (in. H <sub>2</sub> O)	GAS SAMPLE VOLUME (ft <sup>3</sup> )	GAS SAMPLE TEMPERATURE - AT DRY GAS METER		FILTER HOLDER TEMPERATURE, LAST IMPINGER, (°F)	TEMPERATURE OF GAS LEAVING CONDENSER OR LAST IMPINGER, (°F)
								INLET (°F)	OUTLET (°F)		
8-7		57	2.0	77	.05	0.90	790.995	128	91	260	54
9		60	2.0	77	.05	0.90	792.530	129	92	264	54
10		63	2.0	77	.04	0.70	794.001	121	96	261	55
11		66	2.0	77	.04	0.70	795.490	121	96	258	56
12		69	2.0	78	.05	0.90	797.250	121	97	261	56
	2.53	72	2.0	78	.05	0.90	799.215	121	98	264	56
TOTAL											
AVERAGE						3.87346					

[illegible]

7808

INVESTIGANT  
\_\_\_\_\_  
SEE

Year	1997	1998	1999	2000	2001	2002	2003	2004	2005	2006	2007	2008	2009	2010	2011	2012	2013	2014	2015	2016	2017	2018	2019	2020	2021	2022	2023	2024	2025	2026	2027	2028	2029	2030	2031	2032	2033	2034	2035	2036	2037	2038	2039	2040	2041	2042	2043	2044	2045	2046	2047	2048	2049	2050	2051	2052	2053	2054	2055	2056	2057	2058	2059	2060	2061	2062	2063	2064	2065	2066	2067	2068	2069	2070	2071	2072	2073	2074	2075	2076	2077	2078	2079	2080	2081	2082	2083	2084	2085	2086	2087	2088	2089	2090	2091	2092	2093	2094	2095	2096	2097	2098	2099	2100
1997	1998	1999	2000	2001	2002	2003	2004	2005	2006	2007	2008	2009	2010	2011	2012	2013	2014	2015	2016	2017	2018	2019	2020	2021	2022	2023	2024	2025	2026	2027	2028	2029	2030	2031	2032	2033	2034	2035	2036	2037	2038	2039	2040	2041	2042	2043	2044	2045	2046	2047	2048	2049	2050	2051	2052	2053	2054	2055	2056	2057	2058	2059	2060	2061	2062	2063	2064	2065	2066	2067	2068	2069	2070	2071	2072	2073	2074	2075	2076	2077	2078	2079	2080	2081	2082	2083	2084	2085	2086	2087	2088	2089	2090	2091	2092	2093	2094	2095	2096	2097	2098	2099	2100	

#### D. PROCESS DATA



## Springfield Pike

Sand Plant B-27-90

(3)

Time	1A (Total Production)	Mfg. Sand	Dust (Baghouse)
9:50 - 10:00 am	96 tons/hr	72 tons/hr	24 tons/hr
11 am - 1:11 pm	97.5 tons/hr	73.13 tons/hr	24.37 tons/hr
3:00 - 2:53 pm	98.3 tons/hr	73.73 tons/hr	24.57 tons/hr

(a)

SP

Sieve	Mfg. Sand	Wt. Passing	% Passing
3/8		626.5	100
4		624.4	99.7
8		447	71.3
16		311.4	49.7
30		235.4	37.6
50		166.4	26.7
100		99.3	15.8
200		49.5	7.9

Dust (Baghouse)			
3/8		565.4	100
4		565.4	100
8		564.8	99.9
16		563.3	99.6
30		561	99.2
50		550.2	97.3
100		474.9	84
200		300.4	53.1

1A (#10)			
3/8		620	100
4		615	99.2
8		454.7	73.4
16		331.1	53.4
30		259	41.8
50		199.6	32.2
100		135.4	21.8
200		73.6	11.9

Facility Springfield Pike Sanitary Boiler No. \_\_\_\_\_  
Monitor Name Bryant Test No. 1-3  
Design Efficiency \_\_\_\_\_ Test Date 6/27/90

[illegible]

E. LABORATORY DATA

# ANALYTICAL DATA

Test No.	SPS-1	SPS-2	SPS-3
Test Date	8/27/90	8/27/90	8/27/90
Filter No.	100031	100021	100030
Filter Tare, mg	620.2	627.3	617.3
Filter Final, mg	628.0	633.7	623.3
Filter Net, mg	7.8	6.4✓	6.0
Probe Acetone, mg	20.5	5.1✓	13.7
Probe Water	2.4	24.7✓	7.1
Impinger Sol. mg.	19.4	1.4	11.1
Impinger Insol., mg	9.0	3.6✓	6.0
Impinger Acetone, mg	4.2	6.6	3.0
Acetone Blank (mg)*	0.2	-	-
Water Blank (mg)*	0.4	-	-
Total particulate, mg	43.9	46.4	35.8

\*Data not corrected for blank values.

Analyst Kevin Kennedy  
 Reviewer Richard L. Campbell

F. V. E. CERTIFICATION

# VISIBLE EMISSIONS EVALUATOR

This is to certify that

*Technical Associates of Raleigh, North Carolina*

met the specifications of Federal Reference Method 9 and qualified as a visible emissions evaluator. Maximum deviation on white and black smoke did not exceed 7.5% opacity and no single error exceeding 15% opacity was incurred during the certification test conducted by Eastern Technical Associates of Raleigh, North Carolina. This certificate is valid for six months from date of issue.

*Thomas H. Rose*

President

*W. Lee*

Vice President

*David Savage*

Program Manager

*225154*

Certificate Number

*Atlanta, Ga.*

Location

*March 21, 1990*

Date of Issue