

Pushing

GUARDIAN SYSTEMS

P.O. Box 300
Leeds, Alabama
205/690-6647

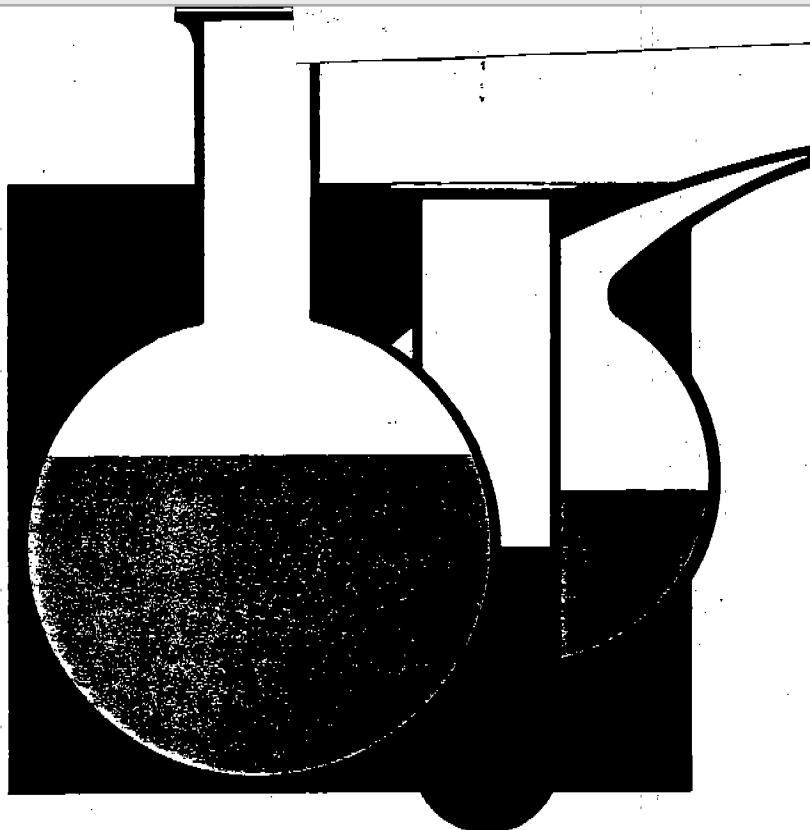
Do not use - only ✓

AP-42 Section	12.2
Reference	
Report Sect.	4
Reference	177

NC

Note: This is a reference cited in AP 42, *Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

The file name refers to the reference number, the AP42 chapter and section. The file name "ref02_c01s02.pdf" would mean the reference is from AP42 chapter 1 section 2. The reference may be from a previous version of the section and no longer cited. The primary source should always be checked.



Test Report

MASS EMISSION TESTS
CONDUCTED ON
COKE BATTERY #'s 5 & 6 POSITIVE PUSHING CONTROL SYSTEM
IN
TARRANT, ALABAMA
FOR
ALABAMA BY-PRODUCTS CORPORATION
ON
AUGUST 9, 1985

Approved by: 
Tom Lotz
Director
Field Services Division

TABLE OF CONTENTS

Chapter		Page
I	Introduction and Test Description	1
II	Summary of Test Results	2
	Tables	
	1 VE Summaries	3
	2 Summarized Particulate Results	4
	Computer Input Parameters	5
	Complete Test Results	6
III	Sampling and Analytical Procedure	8
	Figures	
	1 Stack Breeching Diagram And Sample Point Location	11
	2 Particulate Sampling Train	12
IV	Field Data	
	Nomenclature	13
	Equations	16
	Data Sheets	18
	Particulate Analysis	19
	VE Sheets for Coke Battery	22
	VE Sheets for Stack	23
V	Calibrations	24

I. INTRODUCTION AND TEST DESCRIPTION

On August 9, 1985, Guardian Systems, Inc. performed a short particulate emissions test on the Coke Battery #'s 5 & 6 Positive Pushing Control System of Alabama By-Products Corporation located in Tarrant, Alabama. The purpose of this short test was to determine if the same conditions existed as those during the compliance test. This test was conducted in accordance to the rules and regulating expressed in the Code of Federal Regulations, Title 40, Section 60, Reference Methods 1-5 as amended.

The Push Control System consists of a movable hood assembly which travels with the car and a baghouse for particulate removal. The damper on the baghouse fan is closed when not in use and is open by radio control when the coke is pushed. The sampling time for particulate emissions started when the pusher machine ram began moving forward and stopped when the emissions captured were collected by the baghouse. Only one push was sampled per traverse point and the sampling time varied for each push. Visible Emissions were monitored at the pushing emission control device outlet (stack) and at the pushing operation around the battery by Jefferson County Health Department Personnel.

The following personnel were present during the actual field sampling and represented the companies indicated:

Mr. J.D. Duren	Alabama By-Products Corporation
Mr. Buddy Wolfe	Alabama By-Products Corporation
Ms. Veda Hampton	Jefferson County Health Dept.
Mr. Frank Cargo	Jefferson County Health Dept.
Mr. Greg Karstens	Guardian Systems, Inc.
Mr. Mike Harvey	Guardian Systems, Inc.

IL SUMMARY OF TEST RESULTS

The following table gives the results of the tests conducted on the Coke Battery #'s 5 & 6 Positive Pushing Control System on August 9, 1985. The allowable emission rate 0.03 pounds per ton of coal charged. The figures are presented in two formats; one by calculation of coal and particulate in pounds per test and the second in pounds per hour.

<u>RUN #</u>	<u>1</u>
Sample time per test, Minutes	7.13
Coal charged per test, Tons	170
Particulate Mass Rate lbs/test	3.99
Emissions lbs/ton of coal charged	0.023
*Particulate Mass Rate lbs/hr	33.58
*Coal charged Tons/hour	1430.6
Emissions lbs/ton of coal charged	0.023

* The particulate mass rate and the coal charged are both presented in hourly rates even though the individual test times were less than an hour. This was accomplished by using a ratio of the time sampled to 60 minutes or one hour.

**Visual Emissions Summaries
Around Push Control**

<u>RUN #</u>	<u>1</u>
Sample time per test, Seconds	600
Time greater than 20% opacity, seconds	120
% greater than, 20 % opacity	20
Highest opacity, %	30

Data was taken from the VE Sheets found in the **Field Data Section**.

Stack Opacity

<u>RUN #</u>	<u>1</u>
Opacity, %	0

Table 1

GUARDIAN ** SYSTEMS ** INCORPORATED
P.O. BOX 300
LEEDS, AL 35094
205-699-6647

Summarized Air Test Results for ALABAMA BY-PRODUCTS CORP
Job Number 54540

RUN NUMBER	1
DATE OF TEST	08/10/85
TIME OF TEST	1121-1320
LOCATION OF TEST	586
STACK GAS TEMPERATURE DEGREES F	114
MOISTURE CONTENT, % V/V	6.69
OXYGEN CONTENT, % V/V	18.50
CARBON DIOXIDE CONTENT, % V/V	1.50
STACK GAS VELOCITY FEET PER SECOND	48.36
VOLUMETRIC FLOW ACTUAL CUBIC FEET PER MINUTE	119,778
VOLUMETRIC FLOW DRY STANDARD CUBIC FEET PER MINUTE	103,419
CONCENTRATION GRAINS PER DRY STANCARD CUBIC FOOT	0.038
CONCENTRATION GRAINS PER ACTUAL CUBIC FOOT	0.033
PARTICULATE MASS RATE (#/HR.)	33.58
% ISOKINETIC	108.19

Table 1

GUARDIAN ** SYSTEMS ** INCORPORATED

P.O. BOX 300
LEEDS, AL 35094
205-699-6647

Computer Input Parameters for ALABAMA BY-PRODUCTS CORP
Job Number 54540

RUN NUMBER	1
DATE	08/10/85
LOCATION	586
TIME	1121-1320
BAROMETRIC PRESSURE (IN. HG)	30.12
STATIC PRESSURE (IN. H2O)	- 0.310
RUN TIME (MINUTES)	7.13
METER VOLUME (CORRECTED)	3.509
STACK TEMPERATURE (°F)	114.00
METER TEMPERATURE (°F)	87.00
METER PRESSURE (IN. H2O)	0.62
SGR VELOCITY PRESSURE	0.810
MASS OF PARTICULATE (MG.)	8.4
ML. OF WATER COLLECTED	5.2
% OXYGEN	18.5
% CARBON DIOXIDE	1.5
% CARBON MONOXIDE	0.0
STACK AREA (SQ. FT.)	41.28
PITOT CORRECTION FACTOR	0.85
NOZZLE DIAMETER (IN.)	0.180

P.O. BOX 300
LEEDS, AL 35094
205-699-6647

Computed Air Test Results for ALABAMA BY-PRODUCTS CORP
Job Number 54540

RUN NUMBER	1
DATE OF TEST	08/10/85
LOCATION OF TEST	586
1. STACK PRESSURE	
INCHES HG	30.10
MILLIMETERS HG	764.47
2. METER PRESSURE	
INCHES HG	30.17
MILLIMETERS HG	766.21
3. METER VOLUME	
DRY STANDARD CUBIC FEET	3.415
DRY STANDARD CUBIC METERS	0.097
4. WATER VOLUME	
STANDARD CUBIC FEET	0.245
STANDARD CUBIC METERS	0.007
5. MOISTURE CONTENT (%)	6.69
6. MOLECULAR WEIGHT DRY	28.98
7. MOLECULAR WEIGHT WET	28.25
8. STACK VELOCITY	
FEET PER SECOND	48.36
METERS PER SECOND	14.74
9. VOLUMETRIC FLOW	
ACTUAL CUBIC FEET PER MINUTE	119,778
ACTUAL CUBIC METERS PER SECOND	56.53
10. VOLUMETRIC FLOW	
DRY STANDARD CUBIC FEET PER MINUTE	103,419
DRY STANDARD CUBIC METER PER SECOND	48.81

GUARDIAN ** SYSTEMS ** INCORPORATED

P.O. BOX 300
LEEDS, AL 35094
205-699-6647

Computed Air Test Results for ALABAMA BY-PRODUCTS CORP
Job Number 54540

RUN NUMBER	1
DATE OF TEST	08/10/85
LOCATION OF TEST	586
11. CONCENTRATION GRAINS PER DRY STANDARD CUBIC FOOT GRAMS PER DRY STANDARD CUBIC METER	0.0379 0.0867
15. PARTICULATE MASS RATE (#/HR.)	33.58
16. VOLUME AT NOZZLE ACTUAL CUBIC FEET ACTUAL CUBIC METERS	3.955 0.112
17. CONCENTRATION GRAINS PER ACTUAL CUBIC FOOT GRAMS PER ACTUAL CUBIC METER	0.0327 0.0748
18. % ISOKINETIC	108.19

III. METHOD 5 SAMPLING AND ANALYTICAL PROCEDURES

General

All sampling and analytical procedures for determination of the particulate emissions from this source were conducted in strict adherence with the Code of Federal Regulations, Title 40, Part 60, Appendix A, Methods 1-5 as amended. The equipment used in this test was manufactured by Research Appliance Corporation and was properly calibrated before these tests (See **Calibrations**). The particulate mass was determined gravimetrically after removal of uncombined water.

METHOD 1

This method was used to determine the number of sampling points and the required matrix. The dimensions of the stack (see **Figure 1**) indicated that 24 points would be required to sample this source. For this short test we sampled 10 points (5 per diameter) for the time of a cycle at each point. Verification of absence of cyclonic flow was not determined because there was no tangential entry and cyclonic flow did not seem to exit from the stack exhaust.

METHOD 2

Velocity measurements were taken using a properly calibrated S type pitot tube and a 0-10 inch inclined manometer.

METHOD 3

Gas analysis for CO_2 , O_2 , and N_2 by difference was performed by using the grab sample technique twice during each test and analyzed with a Fyrite Gas Analyzer (0-20% scale with 0.5% divisions).

METHOD 4

Moisture content of the stack gases was determined in conjunction with Method 5 for each test by measuring the amount of water collected in the condenser and the increase in weight of the silica gel. These weights were combined to give the total amount of water collected.

METHOD 5

Sampling Techniques

Previous data was used to determine stack temperature, pressure, range of velocity heads, moisture content and stack gas dry molecular weight prior to sampling. A Monroe 1830 programable calculator or a Radio Shack PC-4 computer and a TI55 were used to calculate the sampling rate for isokinetic sampling. The particulate determinations were made by utilizing the sample train in **Figure 2**. The meter box was leak checked from the pump to the orifice and initial and final leak checks of the sampling system and pitot lines were performed as outlined in Method 5 and were recorded on the data sheets (**See Field Data**). The nozzle (stainless steel 316 with an angle of taper of 30° and of button hook design) was calibrated before and after each test using a micrometer and was also recorded on the data sheets.

The sampling probe was placed at each sampling point and the gases were drawn through a stainless steel nozzle attached to a heated glass lined probe. The probe was maintained at the proper setting to obtain an exit temperature of $248 \pm 25^{\circ}\text{F}$ (**See Calibrations**). The gases then pass through a glass fiber filter (Gelman, Class A) of 0.3 micron retention to remove particulate matter. The filter was maintained at a temperature of $248 \pm 25^{\circ}\text{F}$. These temperatures were recorded at each point on the data sheets.

The gases then pass through a condenser placed in an ice bath to maintain a maximum exit temperature of 68°F . This temperature was also recorded on the data sheets. The gases then pass through a pre-weighed drying column filled with indicating silica gel to remove any remaining moisture. The clean and cool gases then entered the meter box where the gas flow and temperature were measured and recorded.

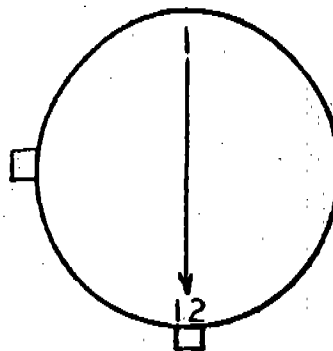
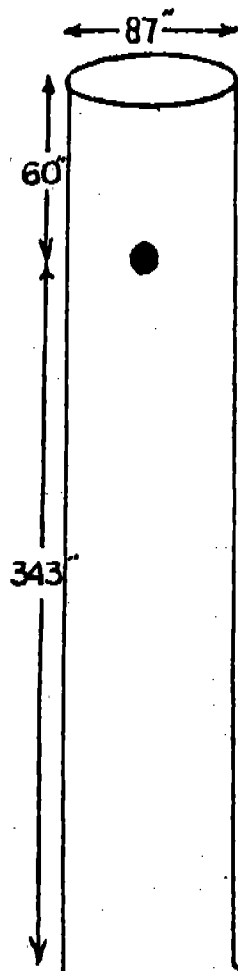
Upon completion of the sampling and the final leak checks the gel column and the probe nozzle-filter holder assembly was transferred to the cleanup area making sure no particulate was lost. The cleanup data was recorded on **Particulate Sample Recovery and Intergrity Sheet** located in the Field Data Section. Particulate catches were placed in sealed petri dishes. Each acetone rinse from the nozzle, probe, and filter holder was combined and placed in a sealed container. For the group of tests an acetone blank of approximately 200 milliliters was placed in a sealed container. These containers were clearly marked and transported to our laboratory for analysis.

Analysis

The Filter (Gelman, Class A, without organic binder, minimum 99.9% retention for particles of 0.3 microns as determined by DOP tests) were prepared for the field test by first heating for 2 hours 220°F then desiccating at $68 \pm 10^{\circ}\text{F}$ at ambient pressure for 24 hours and weighing at intervals of at least six (6) hours to a constant weight to the nearest 0.1 milligrams (less than 0.5 milligrams change from previous reading). Upon return to the laboratory, the filters were subjected to the same procedures as outlined above. The weights were recorded in a bound laboratory book and transferred to the sheets in this report (**Method 5 Train Analytical Particulate Data**). During each weighing the filter was not exposed to laboratory atmosphere for more than two (2) minutes and a relative humidity of less than fifty percent (50%).

The acetone blank (reagent grade, 0.001 % residue, in glass bottles) (**Blank Analytical Data**) for the group of tests and the acetone rinses for each test were evaporated to dryness in tared glass beakers. They were then desiccated for twenty-four (24) hour and weighed to a constant weight as describe above.

SAMPLING POINTS



POINT *DISTANCE FROM WALL

1	5.9
2	9.8
3	14.3
4	19.4
5	25.8
6	34.9
7	60.1
8	69.3
9	75.6
10	80.7
11	85.2
12	89.1

*INCLUDES 4" NIPPLE

FROM FAN

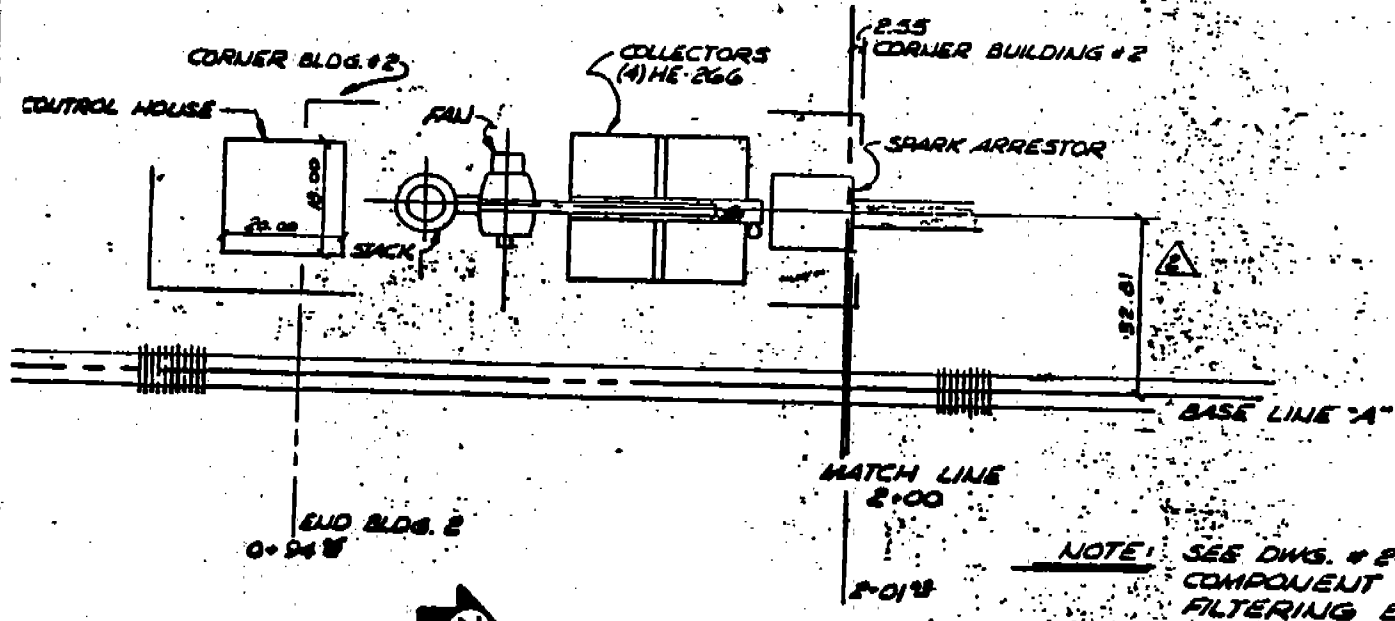
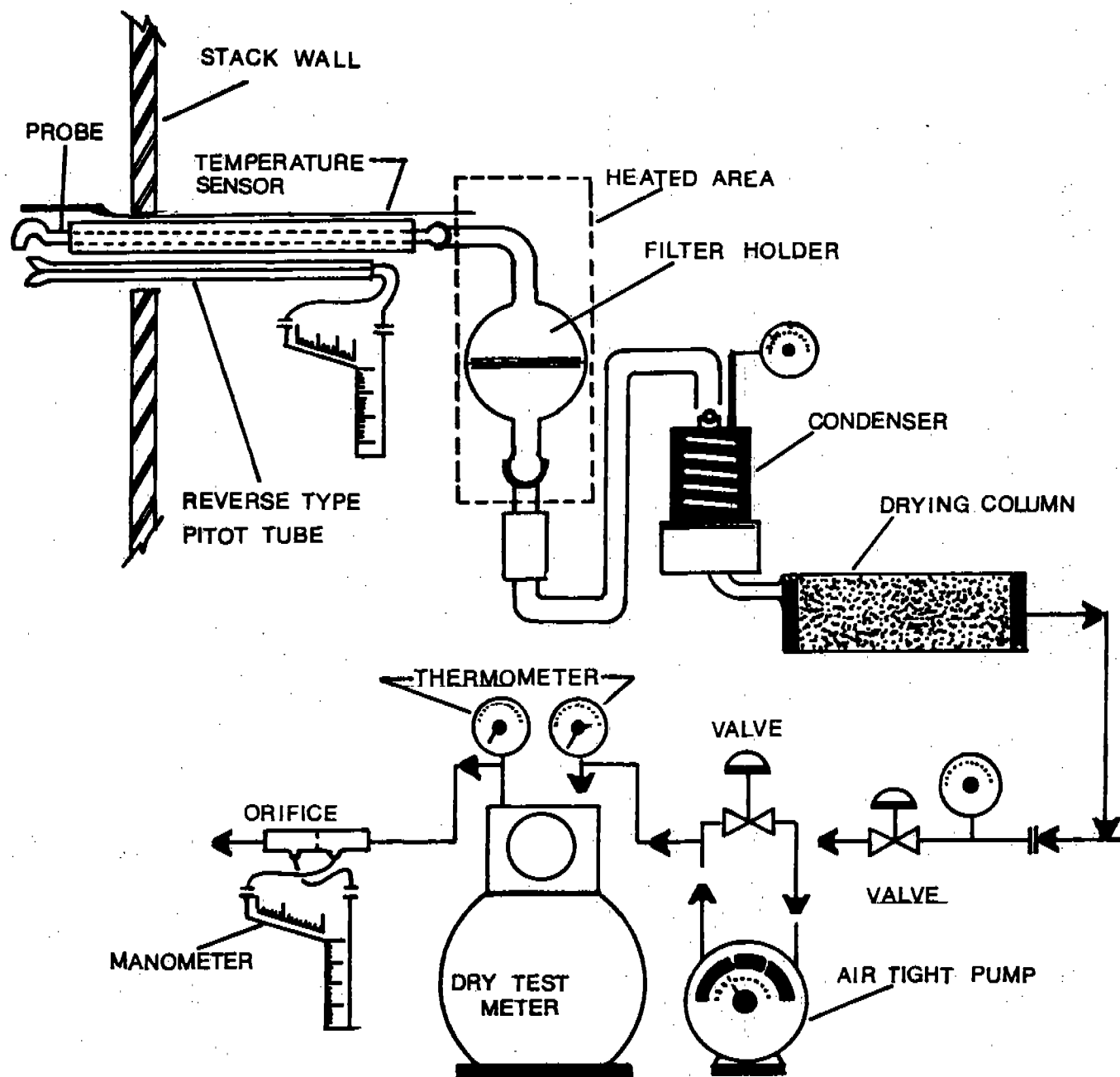


FIGURE 1



PARTICULATE SAMPLING TRAIN.. METHOD 5

NOMENCLATURE

- ACF - Actual Cubic Feet
- ACFM - Actual Cubic Feet per minute
- ACM - Actual Cubic Meters
- ACMS - Actual Cubic Meters per second
- An - Cross sectional area of nozzle, (ft²)
- As - Area of Stack, (ft²)
- Bws - Water vapor in the gas stream, proportion by volume (dimensionless)
- Ca - Acetone blank residue concentration, mg/g
- c_a - Particulate Concentration, ACF
- CFM - Cubic feet per minute
- Cp - Pitot tube coefficient, (dimensionless)
- c_s - Particulate Concentration, grains/DSCF
- Cso₂ - Concentration of sulfur dioxide (dry basis) corrected to standard conditions, lb/DSCF
- C₁₂ - Particulate concentration (c_s adjusted to 12% excess air), grains/DSCF
- C₅₀ - Particulate concentration (c_s adjusted to 50% excess air), grains/DSCF
- DSCF - Dry Standard Cubic Feet
- DSCFM - Dry Standard Cubic Feet per minute
- DSCM - Dry Standard Cubic Meters
- DSCMS - Dry Standard Cubic Meters per second
- EA - Excess Air, %
- I - Isokinetic Sampling, %
- Km - Orifice Correction Factor, (dimensionless)
- Kp - Pitot tube constant, 85.49 $\left[\frac{(\text{lb/lb-mole})(\text{in. Hg})}{(^{\circ}\text{R})(\text{in. H}_2\text{O})} \right]^{1/2}$

NOMENCLATURE - continued

- La - Maximum acceptable leakage rate for either a pretest leak check or for a leak check following a component change; equal to 0.02 CFM or 4 percent of the average sampling rate, whichever is less.
- Li - Individual leakage rate observed during the leak check conducted prior to the "ith" component change ($i = 1, 2, 3, \dots, n$), CFM.
- Lp - Leakage rate observed during the post test leak check, $\text{ft}^3/\text{min. (cfm)}$.
- Ma - Mass of residue of acetone after evaporation, mg.
- Md - Molecular weight of stack gas; dry basis, lb/lb-mole.
- Mn - Total amount of particulate matter collected, mg.
- Ms - Molecular weight of stack gas; wet basis, lb/lb-mole.
- Mw - Molecular weight of water, 18.0 g/g-mole (18.01 lb/lb-mole)
- AP - Velocity head of stack gas, in. H_2O
- Pa - Density of acetone, mg/ml
- Pbar - Barometric pressure at the sampling site, in. Hg
- Pg - Stack static pressure, in. H_2O
- Pm - Meter pressure, in. Hg
- PMR - Particulate Mass Rate, lbs per hour
- Ps - Absolute stack pressure, in. Hg
- Pstd - Standard absolute pressure, 29.92 in. Hg
- Pw - Density of water, 0.9982 g/ml (0.002201 lb/ml)
- Qa - Volumetric flow rate, ACFM
- Qs - Volumetric flow rate, DSCFM
- R - Ideal gas constant, 0.06236 mm Hg - $\text{m}^3/^\circ\text{K-g-mole}$ (21.85 in. Hg- $\text{ft}^3/^\circ\text{R-lb-mole}$)
- SCF - Standard Cubic Foot
- ta - Ambient Temperature, $^\circ\text{F}$
- tm - Average Temperature of meter, $^\circ\text{F}$
- t_s - Average Temperature of stack, $^\circ\text{F}$

NOMENCLATURE - continued

tstd - Standard Temperature, 68°F

NOTE: Capital "T" denotes degrees Rankin

Va - Volume of acetone blank, ml

Vaw - Volume of acetone used in wash, ml

Vlc - Total volume of liquid collected in condenser and silica gel, ml

Vm - Volume of gas sample, as measured by the dry gas meter, ACF

Vmc - Volume of gas sample, corrected for leak, ACF

Vm(std) - Volume of gas sample measured by the dry gas meter, corrected to standard conditions, DSCF

Vn - Volume collected at stack conditions through nozzle, ACF

Vs - Average stack gas velocity, ft/sec.

Vw(std) - Volume of water in the gas sample, corrected to standard conditions, SCF

Wa - Weight of residue in acetone wash, mg

Y - Dry gas meter calibration factor, (dimensionless)

ΔH - Average pressure differential across the calibrated orifice, in. H₂O

ΔH_a - Value of ΔH measured for a specific orifice when operated under the following conditions: 0.75 cfm of dry air (M.W. = 29) at 68°F, 29.92 in. Hg.

$\sqrt{\Delta P}$ - Average of the square roots of the velocity pressure, in. H₂O

θ - Total sampling time, min.

θ_1 - Sampling time interval from the beginning of a run until the first component change, min.

θ_i - Sampling time interval between two successive component changes, beginning with the interval between the first and second changes, min.

θ_p - Sampling time interval from the final (n^{th}) component change until the end of the sampling run, min.

%CO₂, %O₂, %N₂, %CO - Number percent (%) by volume (dry basis) of each compound in the stack gas.

EQUATIONS

$$1. \quad P_s = P_{\text{bar}} + \frac{P_g}{13.6}$$

$$2. \quad P_m = P_{\text{bar}} + \frac{\Delta H}{13.6}$$

$$3. \quad V_{m(\text{std})} = V_m Y \left(\frac{T_{\text{std}}}{T_m} \right) \left[\frac{P_{\text{bar}} + \frac{\Delta H}{13.6}}{P_{\text{std}}} \right]$$

$$4. \quad V_w(\text{std}) = 0.04707 V_{lc}$$

$$5. \quad B_{ws} = \frac{V_{w(\text{std})}}{V_{m(\text{std})} + V_{w(\text{std})}}$$

$$6. \quad M_d = 0.44 (\%CO_2) + 0.32 (\%O_2) + 0.28 (\%N_2 + \%CO)$$

$$7. \quad M_s = M_d (1 - B_{ws}) + 18(B_{ws})$$

$$8. \quad v_s = K_p C_p (\sqrt{\Delta p}) \text{ avg.} \sqrt{\frac{T_s}{M_s P_s}}$$

$$9. \quad Q_a = (v_s) (A_s) (60)$$

$$10. \quad Q_s = Q_a (1 - B_{ws}) \left(\frac{528}{T_s} \right) \left(\frac{P_s}{29.92} \right)$$

$$11. \quad c_s = [0.0154 (Mn/Vmstd)]$$

$$12. \quad EA = \left[\frac{\%O_2 - 0.5 \%CO}{0.264 \%N_2 (\%O_2 - 0.5 \%CO)} \right] 100$$

$$13. \quad c_{50} = \frac{c_s}{1 - \left[\frac{(1.5)(\%O_2) - 0.133(\%N_2) - 0.75 (\%CO)}{21} \right]}$$

$$14. \quad c_{12} = c_s \frac{12}{\%CO_2}$$

EQUATIONS - continued

$$15. \quad \text{PMR} = (c_s)(Q_s) \left(\frac{60}{7000} \right)$$

$$16. \quad V_n = \frac{T_s}{P_s} \left[(0.002669)(V_{1c}) + \frac{V_m}{T_m} \left(P_{\text{bar}} + \frac{\overline{\Delta H}}{13.6} \right) \right]$$

$$17. \quad C_a = (0.0154)(M_n) / V_n$$

$$18. \quad I = \frac{100 V_n}{60 \theta v_s A_n}$$

$$19. \quad V_{mc} = V_m - (L_p - L_a) \theta$$

$$20. \quad W_a = C_a V_a w p_a$$

JCAO 8/9/85

PLANT

ABC

RUN #

1

LOCATION

Tarrant

DATE

8/10/85

BARO. PRES.

30.12

TIME

1121-1320

GUARDIAN SYSTEMS INC.

P. O. Box 300
Leeds, Alabama 35094
205/699-6647

	TIME		METER VOL FT 3	STACK TEMP OF	METER TEMP. OF			IMP TEMP OF	BOX TEMP OF	METER PRESSURE "H ₂ O	VELOCITY PRESSURE "H ₂ O	VACUUM PRESSURE "Hg	STATIC PRESSURE "H ₂ O	
	MIN	DUR			IN	OUT	AVG						BEFORE	AFTER
A			876.028	(1)	(2)	(3)		(4)	(5)					
1	39	53	876.4	101	87	87	87	40	729	.68	.71			
2	37	92	876.1	109	84	84	86	40	232	.75	.79			
3	39	17	877.0	118	86	86	86	41	236	.61	.65			
4	38	39	877.3	114	88	88	88	40	241	.54	.57			
5	1:14	46	877.9	120	84	84	85	40	242	.58	.63			
B														
1	38	98	878.2	119	90	88	89	43	237	.62	.66			
2	41	78	878.5	112	88	89	89	45	228	.54	.57			
3	39	15	878.9	115	88	88	88	50	231	.53	.56			
4	39	24	879.2	119	90	90	90	52	245	.60	.64			
5	38	53	879.579	115	86	87	87	56	261	.75	.80			
10	7.13	25	3.551	114				87		.62	.810			
			.9852											
			3.509											

WATER

 ml. 3ml
 gram 2.2
 Total 5.2

OPERATOR

Karstens / Harvey

 Sample Box No. 4
 Meter Box No. 400
 Meter ΔH@ 1.71
 Pitot, CP .84
 Temp. Device 52 II
 Probe No. 8-1
 Probe Liner Glass
 Probe Htr. Set 78%
 Ambient Temp. 85
 % O₂ 18.5
 % CO₂ 1.5

NOTES

 Orifice ✓
 Pitot ✓
 Meter ✓

 R68
 S74
 FINAL LEAK AT
 5" Hg
 .008

BLANK ANALYTICAL DATA

Plant ABC

Sample location SYSTEM 5+6

Relative humidity 20%

Type of blank ACETONE

Liquid level at mark and container sealed ✓

Density of blank (ρ_a) .7857 g/ml

Blank volume (V_a) 200 ml

Date and time of wt. 0800 8-12-85 Gross wt. 110.5859 mg

Date and time of wt. 1400 8-12-85 Gross wt. 110.5861 mg

Average gross wt. 110.5860 mg

Tare wt. 110.5858 mg

Weight of blank (m_a) .2 mg

$$Ca = \frac{m_a}{V_a \rho_a} = \frac{(0.2)}{(200)(.7857)} = .001273 \text{ mg/g}$$

Note: In no case shall a blank residue greater than (0.01 mg/g) or 0.001% of the weight of blank used be subtracted from the sample weight.

Remarks: _____

Signature of analyst

Signature of reviewer

Greg O. Karsten

[Signature]

METHOD 5

PARTICULATE SAMPLE RECOVERY AND INTEGRITY SHEET

Plant: ABC Sample date: 8-10-85
 Sample location: SYSTEM 5+6 Run no.: #1
 Sample recovery person: KARSTEN'S Recovery date: SAME
 Filter(s) no.: C-15

MOISTURE

Impingers	Silica gel
Final volume (wt) <u>3</u> ml (gm)	Final wt. <u>945.7</u> g
Initial volume (wt) <u>0</u> ml (gm)	Initial wt. <u>943.5</u> g
Net volume (wt) <u>3</u> ml (gm)	Net wt. <u>2.2</u> g
Total moisture <u>5.2</u> g	
Color of silica gel <u>BLUE</u>	
Description of impinger water <u>CLEAR</u>	

RECOVERED SAMPLE

Filter container no. C-15 sealed ✓
 Description of particulate on filter CLEAN

Acetone rinse container no. <u>RUN #1</u>	Liquid level marked <u>✓</u>
Acetone blank container no. <u>BLK</u>	Liquid level marked <u>✓</u>

Samples stored and locked _____
 Remarks: _____

Date of laboratory custody _____
 Laboratory personnel taking custody _____
 Remarks: _____

METHOD 5 TRAIN ANALYTICAL PARTICULATE DATA

Plant ABC Run No. #1
 Sample location SYSTEM 5+6
 Relative humidity 20%
 Density of acetone (pa) .7857 g/ml

Sample type	Sample identifiable	Liquid level marked and/or container sealed
Acetone rinse	✓	✓
filter(s)	✓	✓

Acetone rinse container no. #4
 Acetone rinse volume (Vaw) 175 ml
 Acetone blank residue concentration (Ca) .001273 mg/g
 $Wa = Ca \text{ Vaw pa} = (.001273) (175) (.7857) = .2$ mg
 Date and time of wt 0800 8-12-85 Gross wt 106.6721 mg
 Date and time of wt 1400 8-12-85 Gross wt 106.6717 mg
 Average gross wt 106.6719 mg
 Tare wt 106.6600 mg
 Less acetone blank wt (Wa) .2 mg
 Weight of particulate in acetone rinse 11.7 mg

Filter(s) container no. C-15
 Date and time of wt 0800 8-12-85 Gross wt 170.0 mg
 Date and time of wt 1400 8-12-85 Gross wt 170.0 mg
 Average gross wt 170.0 mg
 Tare wt 173.3 mg
 Weight of particulate on filter(s) -3.3 mg
 Weight of particulate in acetone rinse 11.7 mg
 Total weight of particulate 8.4 mg

Note: In no case shall a blank residue greater than (.01 mg/g) or .001% of the weight of acetone used be subtracted from the sample weight.

Remarks: _____

Signature of analyst Greg A. Kauters

Signature of reviewer _____

Alabama By Products August 9, 1985
Pushing Emission Study

<u>Oven No.</u>	<u>Isms</u> <u>Start / Stop</u>	<u>Opacity</u>	<u>Capture</u>	<u>Duration</u>
* 2	11:33 / 11:34	30%	70%	60sec
12	11:54 / 11:55	25%	75%	60sec
22	12:14 / 12:15	20%	80%	50sec
32	12:44 / 12:45	15%	85%	25sec
52	1:18 / 1:19	5%	95%	-
4	1:39 / 1:40	10%	90%	-
14	1:56 / 1:57	5%	95%	-
24	2:11 / 2:12	20%	80%	10sec
34	2:39 / 2:40	20%	80%	4sec
44	2:53 / 2:54	10%	90%	-

17 tons of coal per oven

Observer: Vado Hampton, JCHD

* Emissions from coke oven

Pushes Observed: 10

Total Pushes Recr 85% Capture. 5

Total Pushes Less Than 85% 5



JEFFERSON COUNTY DEPARTMENT OF HEALTH
BUREAU OF ENVIRONMENTAL HEALTH
AIR POLLUTION CONTROL PROGRAM

VISIBLE EMISSIONS OBSERVATION REPORT

Company Name Alabama By Products V.E.E. No.
Source #546 Pink control baghouse Permit No. 0000018503
Start/Stop Time _____ Sky Condition 50% clouds Date 080975
Observation Point E of source Distance from Source 125'
Stack Height 40' Background sky Plume Color black
Method of Evaluation ☐ Wind Direction SE Minutes Exceeding:
Total Minutes Observed 10% Opacity =
Highest 6 Minute Average 20% Opacity =
Violation Observed ☐ 60% Opacity =
Notice Number % Opacity =

Seconds					Seconds					Seconds					Seconds				
Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45	Min.	0	15	30	45
0	11:52	0	0	0	15					30					45				
1	0	0	0	0	16	1:35	0	0	0	31					46				
2	11:53	0	0	0	17	0	0	0	0	32					47				
3	0	0	0	0	18	1:53	0	0	0	33					48				
4	12:13	0	0	0	19	0	0	0	0	34					49				
5	0	0	0	0	20					35					50				
6	12:14	0	0	0	21					36					51				
7	0	0	0	0	22					37					52				
8	1:18	0	0	0	23					38					53				
9	0	0	0	0	24					39					54				
10	1:39	0	0	0	25					40					55				
11	0	0	0	0	26					41					56				
12	1:55	0	0	0	27					42					57				
13	0	0	0	0	28					43					58				
14					29					44					59				

Results and Recommendations:

Observer Cargo

43

Time Analysis

VL METHOD 5 CALIBRATIONS

Meter Box

The meter box shall be initially calibrated every 6 months at the fixed settings of 0.5, 1.0, 1.5, 2.0, and 2.5 inches of water. Upon return to the laboratory from this test the meter box was recalibrated using the average meter pressure for the test series. The recalibration point shall be the closest point to the initial calibration (for an average meter pressure of 1.40 a 1.50 "H₂O was used as the recalibration point). These recalibrations produced a single point MCF and a H@ which shall be compared to the original calibration to see if they are within the 5% allowed at the highest vacuum seen. These recalibration sheets are located behind the original calibration sheets and were used to modify the meter volumes.

Pitot Tubes

The S type pitot tubes shall be calibrated against a standard pitot tube ($C_p = 0.99$) every six months in a wind tunnel with a capacity to generate a test section velocity of approximately 3000 feet/ minute. Additionally the pitot tube shall be measured as to its specifications and alignment. Upon return, the intercomponent spacings and the face opening alignment of the pitot tube assembly shall be rechecked and if no changes are noticed; it shall be assumed that the coefficient of the assembly had not changed.

Temperature measurements

All temperature devices (impinger, meter box, hot box and stack) shall be calibrated every 6 months against an ASTM mercury-in-glass reference thermometer or a reference thermocouple and potentiometer calibrated by fixed points, e.g., ice bath and boiling water (corrected to barometric pressure). Upon return the stack temperature device shall be recalibrated within 10% of the average absolute stack temperature. If the device being tested agrees within 1.5% of the reference device, the temperature data taken in the field shall be considered valid.

VL METHOD 5 CALIBRATIONS CONTINUED

Barometric Pressure

An aneroid barometer capable of measuring atmospheric pressure to within 0.1 inches Hg shall be used. If this device is defective the following alternate method shall be used. The barometric reading may be obtained from a nearby national weather service station, in which case the station value (which is the absolute barometric pressure) shall be requested and an adjustment for elevation differences between the weather station and sampling point shall be applied at a rate of minus 0.1 inches Hg per 100 feet elevation increase or vice versa for elevation decrease.

Specific Test equipment and measurements

The equipment used during these tests was as follows:

Probe: 8-1

Meter Box: #400

Stack Temperature: OMEGA II

Sample Box: #4

Condenser: #1

The average temperature of the stack was 114 °F. The OMEGA II was recalibrated at 115 °F and agreed exactly with an ASTM mercury-in-glass reference thermometer. The intercomponent spacings and the face opening alignment of the Pitot Tube assembly was rechecked and no changes were noticed; therefore, it was assumed that the coefficient of the assembly had not changed.

Type A Glass Fiber Filters

FEATURES

- High tensile strength. ■ Excellent handling characteristics. ■ Good wetting properties. ■ Minimum of 99.9% retention for particles of .3 μ m as determined by DOP tests. ■ Binder free.

This is the original glass fiber filter pioneered by Gelman Instrument Company over 15 years ago. It continues to be widely used for high volume sampling. Since zinc is one of the raw materials incorporated in the glass fibers, Type A Filters have a variable zinc content. Another component of the filter, sulfuric acid, is used as a dispersion medium, making the sheets unsuitable for measurement of sulfates.

Type A Glass Fiber Filters are less likely to develop static charge or tear than other glass fiber media types. They are used extensively in applications where zinc and iron content is not important, or where sulfate content is not being determined.

Size	37 mm	47 mm	102 mm	8"x10"
Product No.	61715	61694	61696	61701
Filters/Pkg.	500	100	100	100

TYPE A GLASS FIBER FILTER SPECIFICATION REPORT

The following physical/chemical properties represent typical, average values obtained in accordance with accepted test methods. They are subject to normal manufacturing variations and are supplied as a technical service. The analysis has been made in accordance with EPA procedures (micrograms/8" x 10" sheet).

ELEMENTS:

Antimony	.30	Manganese	.200
Arsenic	.30	Mercury	.100
Beryllium	.1	Molybdenum	.10
Bismuth	.10	Nickel	.10
Cadmium	.5	Selenium	.5000
Chromium	.10	Tin	.10
Cobalt	.10	Titanium	.170
Copper	.2	Vanadium	.10
Iron	.2300	Zinc	.5000
Lead	.20		to 25,000

OTHER PHYSICALS:

BSO	.522	Flow Resistance (Max.)	
*pH	8.5	@ 320 cm/min.	.80 mm
DOP @320/cm/min		Flow Rate (air)	
(ASTM Method 2986)	.99.9%	lpm/cm ² @ 70 cm Hg	.50
Tensile Strength		Max. Use Temp.	.400°C
(Fed. Spec. UUP31B)	.750 gr	Static Properties	Low
Weight,		Ability to	
8"x10" sheet	4.0±.3 gr.	Fold	Excellent

WATER EXTRACTABLE IONS:

Sulfate	.100	Chloride	.1500
Nitrate	.50	Fluoride	.15
Ammonia	.20		

*pH—Gelman Procedure:

- 500 ml distilled water.
- Add 15 drops saturated KCl solution
- Shred one 8"x10" sheet and soak in prepared water for one hour.
- Run pH at ambient temperature.

Type A/E Glass Fiber Filters

FEATURES

- Low trace metals. ■ Medium Handling characteristics. ■ Available in all sizes. ■ Minimum of 99.9% retention for particles of .3 μ m as determined by DOP tests. ■ Binder free.

Type A/E Glass Fiber Filters are composed of low acid soluble glass fiber. They contain low levels of both zinc and iron. The filters do react with atmospheric sulfur dioxide; and therefore, when high levels of sulfur are expected, corrections for this reaction should be accounted for.

Type A/E Glass Fiber Filters are binder free and ideal for gravimetric analysis of air pollutants. This pure, organic free filter is the basis for procedures widely used in determining municipal and industrial air polluting substances.

Size	25 mm	37 mm	47 mm	102 mm	8"x10"
Product No.	61630	61652	61631	61633	61638
Filters/Pkg.	500	500	100	100	100

TYPE A/E GLASS FIBER FILTER SPECIFICATION REPORT

The following physical/chemical properties represent typical, average values obtained in accordance with accepted test methods. They are subject to normal manufacturing variations and are supplied as a technical service. The analysis has been made in accordance with EPA procedures (micrograms/8" x 10" sheet).

ELEMENTS:

Antimony	.20	Manganese	.2
Arsenic	.20	Mercury	.80
Beryllium	.1	Molybdenum	.10
Bismuth	.10	Nickel	.10
Cadmium	.2	Selenium	.200
Chromium	.10	Tin	.10
Cobalt	.10	Titanium	.10
Copper	.2	Vanadium	.10
Iron	.100-1800	Zinc	.90
Lead	.10		

OTHER PHYSICALS:

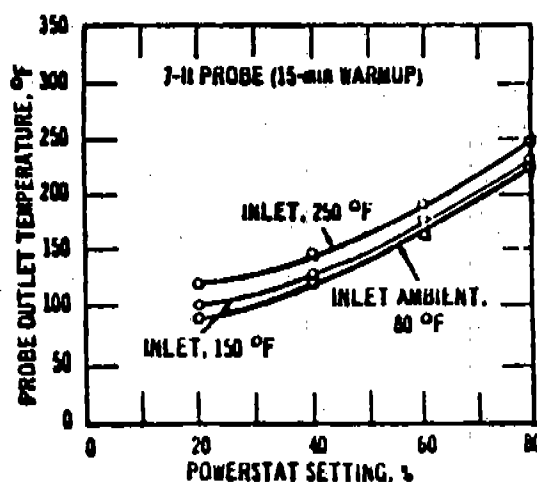
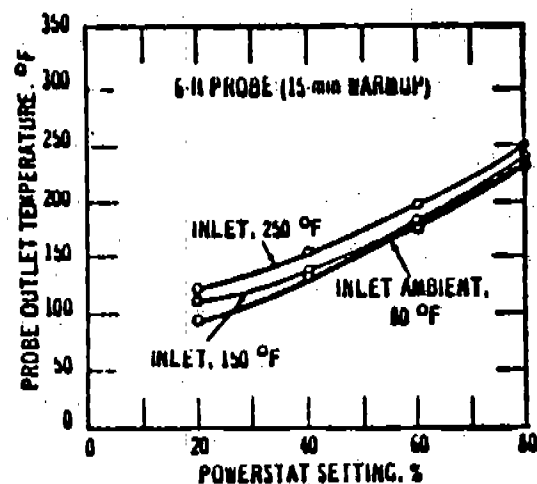
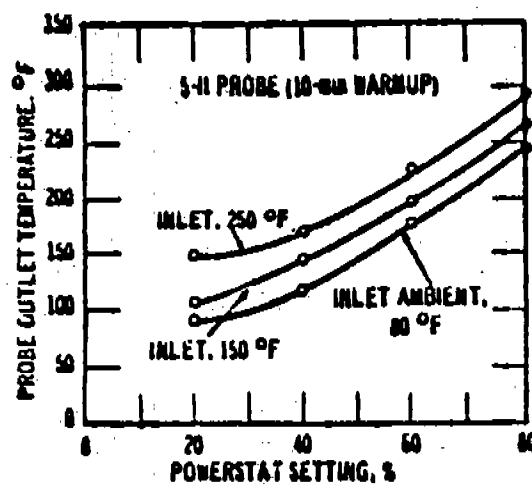
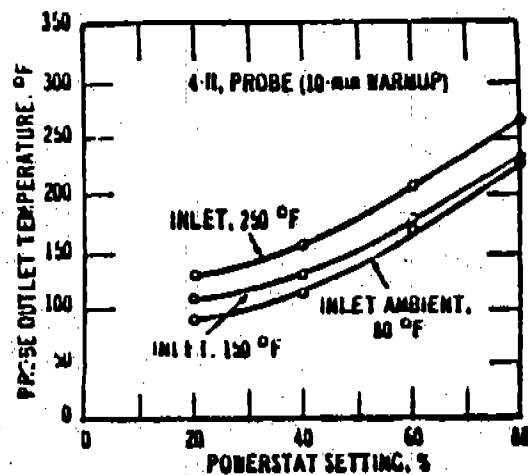
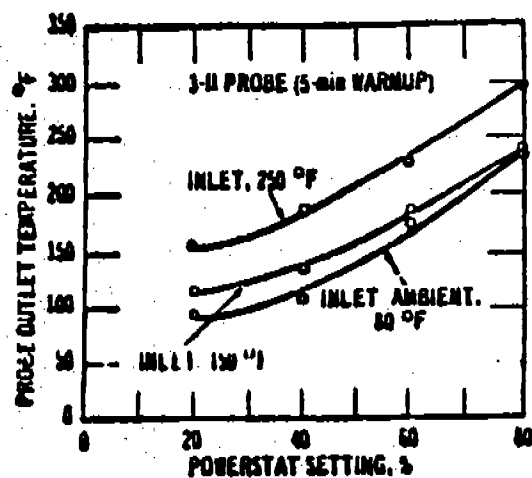
BSO	.522	Flow Resistance (Max.)	
*pH	9.0	@ 320 cm/min.	.80 mm
DOP @320/cm/min		Flow Rate (air)	
ASTM Methods 2986)	.99.9%	lpm/cm ² @ 70 cm Hg	.60
Tensile Strength		Max. Use Temp.	.400°C
(Fed. Spec. UUP31B)	.600 gr.	Static Properties	Medium
Weight		Ability to	
8"x10" sheet	4.0±.3 gr.	Fold	Good

WATER EXTRACTABLE IONS:

Sulfate	.600	Chloride	.1500
Nitrate	.115	Fluoride	.87
Ammonia	.13		

*pH—Gelman Procedure:

- 500 ml distilled water.
- Add 15 drops saturated KCl solution.
- Shred one 8"x10" sheet and soak in prepared water for one hour.
- Run pH at ambient temperature.



NOTE: Flow rate held constant at 0.75; 50% change in flow rate has little effect on probe temperature.

METER CALIBRATION FORM

Date 3/18/85

Box No. 400

P_{bar} = 30.00 in. Hg

Calibrated by LOTZ

Orifice Manometer Set- ting ΔH (in. H ₂ O)	Δ wtm (in. H ₂ O)	Gas Volume Wet Test Meter		Gas Volume Dry Gas Meter		Temperature Wet Test Meter		Avg. Temp. Dry Gas Meter	
		Initial	Final	Initial	Final	Initial	Final	Inlet	Outlet
2.0	-1.03	0.006	13.55	736.675	750.248	12.2	12.5	75 94	61 72
2.5	-1.25	13.575	26.765	750.248	763.914	12.5	13.0	90 101	72 78
1.5	-0.78	26.765	37.010	763.984	774.522	13.0	13.2	97 98	78 51
1.0	-0.58	37.010	45.505	774.522	783.227	13.2	13.5	98 97	51 52
0.5	-0.36	45.505	51.557	783.227	785.497	13.5	13.5	95 94	52 52

Pump must be operated for at least 15 minutes at each ΔH setting (.5, 1, 1.5, 2 and 3)

T_{dgm} = average temperature of dry gas meter (inlet and outlet) + 460°F

T_{wtm} = average temperature of wet test meter + 460°F

P_{wtm} = P_{bar} - $\frac{\Delta wtm}{13.6}$

P_{dgm} = P_{bar} + $\frac{\Delta H}{13.6}$

Δwtm = pressure on wet test meter in inches of H₂O

Y = meter calibration factor

28.316 = conversion factor when using a wet test meter calibrated in liters

Calculations

$$Y = \frac{(\text{wet final} - \text{wet initial})(T_{dgm})(P_{wtm})}{(\text{dry final} - \text{dry initial})(28.316)(T_{wtm})(P_{dgm})}$$

or (1)

$$\bar{Y} = \frac{\sum Y}{5}$$

Meter Tolerance = 1.00 ± 0.01

If the meter calibration factor is not within the allowable tolerance, the calibration factory Y may be used to mathematically correct the gas meter dial readings to the proper values instead of physically adjusting the dry gas meter dials to correspond to the wet test meter readings.

POST TEST METER CALIBRATION

PROJECT

ABC

STACK

546

Date

8-12-85

Box No.

400

P_{bar} =

30.08

in. Hg

Calibrated by

Signature

Orifice Manometer Set- ting ΔH (in. H ₂ O)	Δ wtm (in. H ₂ O)	Gas Volume Wet Test Meter		Gas Volume Dry Gas Meter		Temperature Wet Test Meter		Avg. Temp. Dry Gas Meter	
		Initial	Final	Initial	Final	Initial	Final	Inlet	Outlet
.5 @ 6" Hg	-0.38	0	6.375	556.313	92.78	26.0	25.8	101/101	53/91
.5 @ 6" Hg	-0.38	0	6.377	902.774	908.223	25.8	25.6	101/105	91/92
.5 @ 6" Hg	-0.38	0	6.37	909.223	915.78	25.6	25.5	103/105	92/92

Pump must be operated for at least 15 minutes at each ΔH setting

T_{dgm} = average temperature of dry gas meter (inlet and outlet) + 460°F

T_{wtm} = average temperature of wet test meter + 460°F

P_{wtm} = P_{bar} - $\frac{\Delta wtm}{13.6}$ (includes vapor pressure of water at saturation at the temperature of the wet test meter)

P_{dgm} = P_{bar} + $\frac{\Delta H}{13.6}$

Δwtm = pressure on wet test meter in inches of H₂O

Y = meter calibration factor

28.316 = conversion factor when using a wet test meter calibrated in liters

Calculations

$$Y = \frac{(\text{wet final} - \text{wet initial})(T_{dgm})(P_{wtm})}{(\text{dry final} - \text{dry initial})(28.316)(T_{wtm})(P_{dgm})}$$

or (1)

$$\bar{Y} = \frac{\sum Y}{3} = .9882$$

Pre test Y 1.0210

% Difference -3.21

ORIFICE CALIBRATION FORM

Date 3/18/85Meter Box No. 400P_{bar} 70.0Calibrated By LOTZ

ΔH	V ₁	V ₂	θ	t ₁	t ₂	V ₂ - V ₁	Q _m	K _m	
in. H ₂ O	CF	CF	Sec.	°F	°F	CF			
2.0	736.75	750.38	10.20	84.5	68	13.673	.7554	.7265	1.75
2.5	750.38	763.984	9.00	95.5	75	13.636	.8861	.7249	1.75
1.5	763.984	774.522	9.00	97.5	79.5	10.538	.7001	.7273	1.74
1.0	774.522	783.237	9.00	96.5	81.5	8.715	.5833	.7403	1.68
0.5	783.237	788.453	9.00	94.5	82	5.222	.4187	.7506	1.63
Average K _m								.7339	

1.71

V₁ = Dry gas meter reading at the start of each testV₂ = Dry gas meter reading at the end of each testt₁ = Dry gas meter inlet temperaturet₂ = Dry gas meter outlet temperatureCalculations

$$1. \quad Q_m = \frac{V_2 - V_1}{\theta} \left[\frac{t_2 + 460}{t_1 + t_2 + 460} \right] \quad (60)(Y)$$

$$2. \quad K_m = \frac{Q_m}{\sqrt{\frac{P_m M_m}{T_m \Delta H}}} \quad \text{(for each } \Delta H \text{)}$$

$$\begin{aligned} M_m &= 29 \\ T_m &= t_2 + 460 \\ P_m &= P_{\text{bar}} + \frac{\Delta H}{13.6} \end{aligned}$$

Y = meter calibration factor

3. Calculate the average K_m as follows:

$$K_m = \frac{\sum K_m}{5}$$

4. Calculate ΔH_a as follows:

$$\Delta H_a = \frac{Q @^2 P @ M @}{T @} \left[\frac{1}{K_m^2} \right] = \frac{0.921}{K_m^2}$$

$$\begin{aligned} Q @ &= 0.75 \text{ cfm} \\ T @ &= 528^\circ \text{R} \\ P @ &= 29.92 \text{ in.Hg} \\ M @ &= 29 \end{aligned}$$

5. Orifice Tolerance = $1.84 \pm .25$

POST TEST ORIFICE CALIBRATION

Date 8-12-85

Meter Box No. 400

P_{bar} 30.08

Calibrated By _____
Signature

Δ H	V ₁	V ₂	θ	t ₁	t ₂	V ₂ - V ₁	Q _m	K _m
in. H ₂ O	CF	CF	Sec.	°F	°F	CF		
0.5	88.33	902.778	900	101	90	6.415	.4158	.7464
0.5	902.778	909.222	900	103	92	6.445	.4210	.7489
0.5	909.222	915.708	910	105.5	94	6.485	.4219	.7492
Average K _m								<u>.7482</u>

V₁ = Dry gas meter reading at the start of each test

Average H@ 1.66

V₂ = Dry gas meter reading at the end of each test

Pre test H@ 1.63

t₁ = Dry gas meter inlet temperature

% Difference 0.94

t₂ = Dry gas meter outlet temperature

Calculations

$$1. Q_m = \frac{V_2 - V_1}{\theta} \left[\frac{t_2 + 460}{\frac{t_1 + t_2 + 460}{2}} \right] (60)(Y)$$

$$2. K_m = \frac{Q_m}{\sqrt{\frac{P_m M_m}{T_m \Delta H}}}$$

$$\begin{aligned} M_m &= 29 \\ T_m &= t_2 + 460 \\ P_m &= P_{bar} + \frac{\Delta H}{13.6} \end{aligned}$$

Y = meter calibration factor

3. Calculate the average K_m as follows:

$$\bar{K}_m = \frac{\sum K_m}{3}$$

4. Calculate Δ H_a as follows:

$$\Delta H_a = \frac{Q \theta^2 P \theta M \theta}{T \theta} \left[\frac{1}{\bar{K}_m^2} \right] = \frac{0.921}{\bar{K}_m^2}$$

$$\begin{aligned} Q \theta &= 0.75 \text{ cfm} \\ T \theta &= 5280R \\ P \theta &= 29.92 \text{ in.Hg} \\ M \theta &= 29 \end{aligned}$$

TEMPERATURE CALIBRATION

FOR OMEGA II

DATE INITIAL

3-6-85

DATE FINAL

DEVICE READING (DEGREES F)

ACTUAL TEMPERATURE (DEGREES F)

TEMPERATURE CALIBRATION

FOR OMEGA II

DATE INITIAL 3-6-85 DATE FINAL

3-6-85

DATE INITIAL

DEVICE READING (DEGREES F)

(2500)

300

600

900

1200

1500

1800

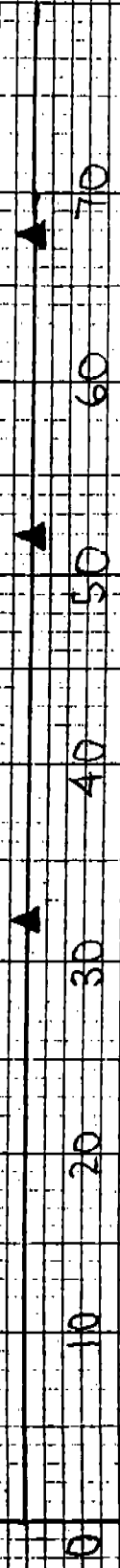
2100

2400

ACTUAL TEMPERATURE (DEGREES F)

TEMPERATURE CALIBRATION
 FOR TEMP INGR # 1
 DATE INITIAL 3-6-85 DATE FINAL _____

DEVICE READING (DEGREES F)



ACTUAL TEMPERATURE (DEGREES F)

TEMPERATURE CALIBRATION

FOR HOT BOX # 4

DATE INITIAL

3-6-85

DATE FINAL

DEVICE READING (DEGREES F)

ACTUAL TEMPERATURE (DEGREES F)

PITOT CALIBRATION FORM

Date 3-6-85 Probe # 8-1
 Calibrated By G KARSTENS
 Nozzle Size N/A

SIDE A

Run #	ΔP_{std} (in. H ₂ O) Standard	$\Delta P(s)$ (in. H ₂ O) Type "s"	$C_p(s)$	Deviation $C_p(s) - \bar{C}_p(A)$
1	.483	.663	.845	-.002
2	.480	.653	.849	+.002
3	.480	.653	.849	+.002
			$\bar{C}_p(A)$	.847

SIDE B

Run #	ΔP_{std} (in. H ₂ O) Standard	$\Delta P(s)$ (in. H ₂ O) Type "s"	$C_p(s)$	Deviation $C_p(s) - \bar{C}_p(B)$
1	.493	.673	.847	—
2	.480	.656	.847	—
3	.486	.666	.846	-.001
			$\bar{C}_p(B)$	.847

CALCULATIONS

$$C_p(s) = C_{p(std)} \sqrt{\frac{\Delta P_{std}}{\Delta P(s)}}$$

(or 0.99)

$$\text{Average Deviation} = \frac{\sum |C_p(s) - \bar{C}_p(A \text{ or } B)|}{3}$$

3

← Must be ≤ 0.01

$$|\bar{C}_p(A) - \bar{C}_p(B)| \leftarrow \text{Must be } \leq 0.01$$