

Note: This is a reference cited in AP 42, *Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

The file name refers to the reference number, the AP42 chapter and section. The file name "ref02_c01s02.pdf" would mean the reference is from AP42 chapter 1 section 2. The reference may be from a previous version of the section and no longer cited. The primary source should always be checked.

Background Report Reference

AP-42 Section Number: 12.20

Background Chapter: 4

Reference Number: 35

Title: Emissions Stack Testing at McDonnell
Douglas, Torrance, CA. Source
Tested: Etching Tank Scrubbr (Tank
193, Bldg. 2) Anodizing Tank
Scrubber (Tank 195, Bldg 2). Date
Tested: October 1 and 3, 1986

November 1986

COPY TO: RANDY P. STRAIT, MIDWEST RESEARCH INSTITUTE
(RALEIGH, NORTH CAROLINA) - sent 3/24/87

K. R&F. 4-35

3/

10.

FILE COPY

CHROMIUM EMISSIONS TESTS FOR TANK 193 AND
TANK 195 AT DOUGLAS AIRCRAFT COMPANY (DAC)

The chromium emission test conditions and results for Tank 193 and Tank 195 are shown in Appendix A. An analysis of the results is presented below.

- o "Total Cr" and " Cr^{+6} " emissions are lower for etching than for anodizing mainly because a batch of parts are etched for only 5 minutes but anodized for 40-45 minutes and because much less H_2 and other gases are generated at the solution-metal parts interface for etching than for anodizing.
- o " Cr^{+6} " emissions are a much lower per cent of "Total Cr" emissions for etching than for anodizing primarily due to differences in the reactions occurring at the solution-metal parts interface.

- Etching

Aluminum metal (Al^0) etched by nitric acid (HNO_3) and hydrofluoric acid (HF) is oxidized to trivalent aluminum (Al^{+3}). Hexavalent chromium (Cr^{+6}) picks up the 3 electrons freed by $\text{Al}^0 \rightarrow \text{Al}^{+3}$ and reduces to trivalent chromium (Cr^{+3}). This leads to a gradual buildup of Cr^{+3} in the solution as more batches of parts are processed. The chromic acid (CrO_3) concentration is maintained within a specified range by periodic addition of CrO_3 to the solution. These factors cause the Cr^{+3} to Cr^{+6} ratio to increase over time.

- Anodizing

Electric current is passed through the solution when parts are being anodized. Aluminum metal (Al^0) is oxidized to form a thin aluminum (as Al^{+3}) oxide film on the parts surface. At the same time, some complex compounds containing hexavalent chromium (Cr^{+6}) and trivalent aluminum (Al^{+3}) form in the solution and a small portion of the Cr^{+6} is reduced to trivalent chromium (Cr^{+3}). This along with periodic addition of chromic acid (CrO_3) to the solution leads to a gradual buildup of Cr^{+6} in the solution. These factors cause the Cr^{+3} to Cr^{+6} ratio to decrease over time. (The Tank 195 test with the sparge air off had a much higher Cr^{+3} to Cr^{+6} ratio than the Tank 195 tests with the sparge air on did. This indicates that the heavy Cr^{+6} compounds, i.e., complex $\text{Cr}^{+6}-\text{Al}^{+3}$ compounds, were not being entrained as readily as the other Cr^{+6} compounds and the Cr^{+3} compounds with the sparge air off.)

MAR 30 1987

K.J. SVOBODA

3/5/87

2 of 2

CHROMIUM EMISSIONS TESTS FOR TANK 193 AND
TANK 195 AT DOUGLAS AIRCRAFT COMPANY (DAC)

(cont'd.)

• " Cr^{+6} " emissions on a lb/hr/ft^2 basis are lower for either etching or anodizing than for hard chrome plating.

- Tank 193 (tri-acid etch), solution surface area = 93 ft^2

Highest measured Cr^{+6} emission on 10/1/86 = 0.0039 lb/hr

lb/hr/ft^2 basis = $(0.0039 \text{ lb/hr}) / (93 \text{ ft}^2)$

= $0.042 \times 10^{-4} \text{ lb/hr/ft}^2$ as Cr^{+6}

at control equipment inlet

- Tank 195 (anodize), solution surface area = 114 ft^2

Highest measured Cr^{+6} emission on 10/1/86 or 10/3/86 = 0.00624 lb/hr

lb/hr/ft^2 basis = $(0.00624 \text{ lb/hr}) / (114 \text{ ft}^2)$

= $0.55 \times 10^{-4} \text{ lb/hr/ft}^2$ as Cr^{+6}

at control equipment inlet

- Hard chrome plating tank data from "Report to the Scientific Review Panel on Chromium," California Air Resources Board and Department of Health Services, September 1985, page I-5

Cr^{+6} emission range = 1.5×10^{-4} to $620 \times 10^{-4} \text{ lb/hr/ft}^2$

as Cr^{+6} at control equipment inlet

APPENDIX A

*Chromium Emissions Tests for
Tank 193 and Tank 195 at C6*

K. J. SVOBODA

11/13/86

1 of 2

CHROMIUM EMISSIONS TESTS

(C6, BLDG. 2, TANKS 193 & 195)

C6 = Douglas Aircraft Company (DAC) Torrance Plant
Sampling on 10/1/86;

① 10⁴⁵_{AM} - 1⁴⁵_{PM}

Tank 193 (tri-acid etch) - sparge air on; scrubber water pump on; tank ventilation air blower on; parts in tank for 5 minutes (normal time period)

Tank 195 (anodize) - sparge air on; scrubber water pump on; tank ventilation air blower on; electric current for anodizing on at 1.5 kilamps & 21 volts and parts in tank for 40 minutes (normal time period) *

② 1⁴⁵_{PM} - 4³⁰_{PM}

Tank 193 - sparge air off; scrubber water pump on; tank ventilation air blower on; parts in tank for 5 minutes

Tank 195 - sparge air off; scrubber water pump on; tank ventilation air blower on; electric current for anodizing on at 1.5 kilamps & 21 volts and parts in tank for 10-15 minutes at start of sampling *

Tank 193 data - solution temp. = 70°F; CrO₃ conc. = 6.1-7.8 oz./gal.;
solution volume = 4,754 gal.; solution surface area = 93 ft²

Tank 195 data - solution temp. = 100°F; CrO₃ conc. = 7-8 oz./gal.;
solution volume = 4,638 gal.; solution surface area = 114 ft²

Per telecon with Rellie Rosario, Engineering Science on 11/11/86, scrubber inlet sampling results are:

Time Period	Tank 193		Tank 195	
	Total Cr	Cr +6	Total Cr	Cr +6
10 ⁴⁵ _{AM} - 1 ⁴⁵ _{PM}	0.00229 lb/hr	0.00035 lb/hr	0.00492 lb/hr	0.0036 lb/hr
1 ⁴⁵ _{PM} - 4 ³⁰ _{PM}	0.00237 lb/hr	0.00039 lb/hr	0.00441 lb/hr	0.00076 lb/hr

Tank 193 ventilation air flow = 13,187 cfm (10⁴⁵_{AM} - 1⁴⁵_{PM}) & 13,218 cfm (1⁴⁵_{PM} - 4³⁰_{PM})

Tank 195 ventilation air flow = 23,279 cfm (10⁴⁵_{AM} - 1⁴⁵_{PM}) & 23,302 cfm (1⁴⁵_{PM} - 4³⁰_{PM})

* Electric current on only when parts in tank

K.J. SVOBODA
11/13/86
2 of 2

CHROMIUM EMISSIONS TESTS (cont'd.)
(CG, BLDG. 2, TANKS 193 & 195)

Sampling on 10/3/86:

① 10⁵⁷ AM - 11³⁰ AM, 12³⁰ PM - 1⁴⁵ PM, & 1³⁰ PM - 2¹⁰ PM

Tank 195 (anodize) - sparge air on; scrubber water pump on; tank ventilation air blower on; electric current for anodizing on at 1.5-2.0 kiloamps & 21 volts and parts in tank for entire sampling period - 3 batches of parts @ 40-45 minutes each were processed and sampling was only done when parts were in tank *

Tank 195 data - solution temp. = 100°F; CrO₃ conc. = 7-8 oz./gal.;
solution volume = 4,638 gal.; solution surface area = 114 ft²

Per telecon with Rollie Rosario of Engineering Science on 11/11/86, scrubber inlet & outlet sampling results are:

Tank 195 scrubber inlet = 0.00925 lb/hr total Cr & 0.00624 lb/hr Cr⁺⁶

Tank 195 scrubber outlet = 0.00052 lb/hr total Cr & 0.00030 lb/hr Cr⁺⁶

Tank 195 ventilation air flow = 23,319 cfm to scrubber

Per pump data: Tank 195 scrubber water flow = 54 gpm

Per scrubber data **: Packing depth = 1 ft.

Mist eliminator depth = 6 in.
(chevron blade type)

* Electric current on only when parts in tank

** Duall F-101 - horizontal wet packed scrubber

WA01602

EMISSIONS STACK TESTING
AT McDONNELL DOUGLAS,
TORRANCE, CALIFORNIA

Source Tested:

Etching Tank Scrubber (Tank 193, Bldg. 2)
Anodizing Tank Scrubber (Tank 195, Bldg. 2)

Date Tested:

October 1 and 3, 1986

Prepared for

LATHAM AND WATKINS
ATTORNEYS AT LAW
555 South Flower Street
Los Angeles, California 90071

Revised

November 12, 1986

Prepared by

ENGINEERING-SCIENCE
75 N. Fair Oaks Avenue
P.O. Box 7107
Pasadena, California 91109

REPORT OF STACK TESTING
AT McDONNELL DOUGLAS

INTRODUCTION

On October 1 and 3, 1986, personnel from Engineering-Science (ES), Pasadena, California, conducted air pollution testing at the McDonnell Douglas facility in Torrance, California. The purpose of the tests was to determine the chromium emissions from etching tank 193 and anodizing tank 195 scrubbers (Bldg. 2). The testing program was coordinated by Mr. Ken Svoboda of McDonnell Douglas.

On October 1, the inlet to the Etching and Anodizing Tank Scrubbers was tested with the air sparger both on and off. On October 3, the inlet and outlet to the Anodizing Tank Scrubber were tested simultaneously. The ES testing team was comprised of Messrs. Rolando Rosario (team leader), Ron Diaz, Paul Farmanian, and Ms. Cammie Mok.

TEST METHODS

The chromium runs were conducted proportionally. The sampling train consisted of a Teflon probe connected to four impingers in series. The first two impingers were each charged with 100 milliliters of deionized water, the third impinger was dry, and the fourth contained about 400 grams of indicating silica gel. The first, third, and fourth impingers were straight tube Greenburg-Smith impingers and the second was an orifice tip Greenburg-Smith impinger.

RESULTS

The results of the testing program are summarized in Tables 1 through 4. Additional supporting data are presented in the Appendix. Additionally, a discussion of the determination of hexavalent chromium emissions is included in the Appendix.

ABLE 1

SUMMARY OF VOLUMETRIC FLOW DATA
 ETCHING TANK SCRUBBER (Tank 193, Bldg. 2)
 ANODIZING TANK SCRUBBER (Tank 195, Bldg. 2)

McDonnell Douglas

October 1, 1986

Parameter	Etching Tank Scrubber Inlet Sparger On	Etching Tank Scrubber Inlet Sparger Off	Anodizing Tank Scrubber Inlet Sparger On	Anodizing Tank Scrubber Inlet Sparger Off
Velocity (FPS)	30.19	30.26	41.10	41.14
Volumetric Flow				
ACFM	13,187	13,218	23,279	23,302
DSCFM	12,546	12,531	22,024	22,026
Stack Temperature (°F)	70	73	72	73
Moisture Content (%)	1.65	1.44	1.83	1.73

TABLE 2

SUMMARY OF VOLUMETRIC FLOW DATA
ANODIZING TANK SCRUBBER (Tank 195, Bldg. 2)

McDonnell Douglas

October 3, 1986

Parameter	Anodizing Tank Scrubber Inlet	Anodizing Tank Scrubber Outlet
Velocity (FPS)	41.17	50.32
Volumetric Flow		
ACFM	23,319	24,576
DSCFM	22,052	23,231
Stack Temperature (°F)	71	71
Moisture Content (%)	1.39	1.43

TABLE 3

SUMMARY OF EMISSIONS DATA
 ETCHING TANK SCRUBBER (Tank 193, Bldg. 2)
 ANODIZING TANK SCRUBBER (Tank 195, Bldg. 2)

McDonnell Douglas

October 1, 1986

Parameter	Etching Tank Scrubber Inlet Sparger On	Etching Tank Scrubber Inlet Sparger Off	Anodizing Tank Scrubber Inlet Sparger On	Anodizing Tank Scrubber Inlet Sparger Off
Total Chromium				
ppm	22.2×10^{-3}	23.0×10^{-3}	27.19×10^{-3}	24.7×10^{-3}
lbs/hr	2.29×10^{-3}	2.37×10^{-3}	4.92×10^{-3}	4.47×10^{-3}
Chromium +6				
ppm	3.38×10^{-3}	3.78×10^{-3}	19.91×10^{-3}	4.19×10^{-3}
lbs/hr	0.35×10^{-3}	0.39×10^{-3}	3.6×10^{-3}	0.76×10^{-3}

TABLE 4

SUMMARY OF VOLUMETRIC FLOW DATA
ANODIZING TANK SCRUBBER (Tank 195, Bldg. 2)

McDonnell Douglas

October 3, 1986

Parameter	Anodizing Tank Scrubber Inlet	Anodizing Tank Scrubber Outlet
Total Chromium		
ppm	51.06×10^{-3}	2.71×10^{-3}
lbs/hr	9.25×10^{-3}	0.52×10^{-3}
Chromium +6		
ppm	34.43×10^{-3}	1.55×10^{-3}
lbs/hr	6.24×10^{-3}	0.30×10^{-3}

APPENDIX

ENGINEERING-SCIENCE, ARCADIA, CALIFORNIA

PLANT: MCDONNELL DOUGLASS
 SOURCE: SCRUBBER #1 INLET
 DATE: 10-1-84
 RUN #: SPARGER ON

FLOW DATA

STACK LENGTH (INCHES)	32.875
STACK WIDTH (INCHES)	31.875
STACK CROSS-SECTIONAL AREA (SQUARE FT)	7.28
BAROMETRIC PRESSURE (INCHES MERCURY)	29.5
OXYGEN CONTENT (%)	20.8
CARBON DIOXIDE (%)	0
MOISTURE CONTENT (%)	1.65
STACK TEMPERATURE (DEG F)	70
STACK STATIC PRESSURE (INCHES WATER)	-1.3
VELOCITY PRESSURE (SQ RT INCHES WATER DELTA P)	.53
PITOT TUBE CORRECTION FACTOR	.84
VELOCITY (FEET PER SEC)	30.19
EXHAUST GAS FLOW RATE (ACFM)	13187
EXHAUST GAS FLOW RATE (DSCFM)	12546

POLLUTANT DATA

PARAMETER: TOTAL CHROMIUM
 MOLECULAR WEIGHT=52
 CONCENTRATION (PPM)
 EMISSION RATE (LBS/HR)

22.2 $\times 10^{-3}$
 2.29 $\times 10^{-3}$

PARAMETER: CHROMIUM +6
 MOLECULAR WEIGHT=52
 CONCENTRATION (PPM)
 EMISSION RATE (LBS/HR)

3.38 $\times 10^{-3}$
 .35 $\times 10^{-3}$

ENGINEERING-SCIENCE, ARCADIA, CALIFORNIA

PLANT: MCDONNELL DOUGLASS
SOURCE: SCRUBBER #1 INLET
DATE: 10-1-86
RUN #: SPARGER OFF

FLOW DATA

STACK LENGTH (INCHES)	32.875
STACK WIDTH (INCHES)	31.875
STACK CROSS-SECTIONAL AREA (SQUARE FT)	7.28
BAROMETRIC PRESSURE (INCHES MERCURY)	29.5
OXYGEN CONTENT (%)	20.8
CARBON DIOXIDE (%)	0
MOISTURE CONTENT (%)	1.44
STACK TEMPERATURE (DEG F)	73
STACK STATIC PRESSURE (INCHES WATER)	-1.3
VELOCITY PRESSURE (SQ RT INCHES WATER DELTA P)	.53
PITOT TUBE CORRECTION FACTOR	.84
VELOCITY (FEET PER SEC)	30.26
EXHAUST GAS FLOW RATE (ACFM)	13218
EXHAUST GAS FLOW RATE (DSCFM)	12531

POLLUTANT DATA

PARAMETER: TOTAL CHROMIUM
MOLECULAR WEIGHT=52
CONCENTRATION (PPM)
EMISSION RATE (LBS/HR)

23 $\times 10^{-3}$
2.37 $\times 10^{-3}$

PARAMETER: CHROMIUM +6
MOLECULAR WEIGHT=52
CONCENTRATION (PPM)
EMISSION RATE (LBS/HR)

3.78 $\times 10^{-3}$
.39 $\times 10^{-3}$

ENGINEERING-SCIENCE, ARCADIA, CALIFORNIA

PLANT: MCDONNELL DOUGLASS
SOURCE: SCRUBBER #2 INLET
DATE: 10-1-86
RUN #: SPARGER ON

FLOW DATA

STACK LENGTH (INCHES)	37.875
STACK WIDTH (INCHES)	35.875
STACK CROSS-SECTIONAL AREA (SQUARE FT)	9.44
BAROMETRIC PRESSURE (INCHES MERCURY)	29.5
OXYGEN CONTENT (%)	20.8
CARBON DIOXIDE (%)	0
MOISTURE CONTENT (%)	1.83
STACK TEMPERATURE (DEG F)	72
STACK STATIC PRESSURE (INCHES WATER)	-1.4
VELOCITY PRESSURE (SQ RT INCHES WATER DELTA P)	.72
PITOT TUBE CORRECTION FACTOR	.84
VELOCITY (FEET PER SEC)	41.1
EXHAUST GAS FLOW RATE (ACFM)	23279
EXHAUST GAS FLOW RATE (DSCFM)	22024

POLLUTANT DATA

PARAMETER: TOTAL CHROMIUM	
MOLECULAR WEIGHT=52	
CONCENTRATION (PPM)	27.19 $\times 10^{-3}$
EMISSION RATE (LBS/HR)	4.92 $\times 10^{-3}$
PARAMETER: CHROMIUM +6	
MOLECULAR WEIGHT=52	
CONCENTRATION (PPM)	19.91 $\times 10^{-3}$
EMISSION RATE (LBS/HR)	3.6 $\times 10^{-3}$

ENGINEERING-SCIENCE, ARCADIA, CALIFORNIA

PLANT: MCDONNELL DOUGLASS
 SOURCE: SCRUBBER #2 INUGT
 DATE: 10-1-86
 RUN #: SPARGER OFF

FLOW DATA

STACK LENGTH (INCHES)	37.875
STACK WIDTH (INCHES)	35.875
STACK CROSS-SECTIONAL AREA (SQUARE FT)	9.44
BAROMETRIC PRESSURE (INCHES MERCURY)	29.5
OXYGEN CONTENT (%)	20.8
CARBON DIOXIDE (%)	0
MOISTURE CONTENT (%)	1.73
STACK TEMPERATURE (DEG F)	73
STACK STATIC PRESSURE (INCHES WATER)	-1.4
VELOCITY PRESSURE (SQ RT INCHES WATER DELTA P)	.72
PITOT TUBE CORRECTION FACTOR	.84
VELOCITY (FEET PER SEC)	41.14
EXHAUST GAS FLOW RATE (ACFM)	23302
EXHAUST GAS FLOW RATE (DSCFM)	22026

POLLUTANT DATA

PARAMETER:	TOTAL CHROMIUM	
MOLECULAR WEIGHT=52		
CONCENTRATION (PPM)		24.7 $\times 10^{-3}$
EMISSION RATE (LBS/HR)		4.47 $\times 10^{-3}$
PARAMETER:	CHROMIUM +6	
MOLECULAR WEIGHT=52		
CONCENTRATION (PPM)		4.19 $\times 10^{-3}$
EMISSION RATE (LBS/HR)		.76 $\times 10^{-3}$

ENGINEERING-SCIENCE, ARCADIA, CALIFORNIA

PLANT: MCDONNELL DOUGLASS
SOURCE: SCRUBBER #2 INLET
DATE: 10-3-66
RUN #: INLET

FLOW DATA

STACK LENGTH (INCHES)	37.875
STACK WIDTH (INCHES)	35.875
STACK CROSS-SECTIONAL AREA (SQUARE FT)	9.44
BAROMETRIC PRESSURE (INCHES MERCURY)	29.3
OXYGEN CONTENT (%)	20.8
CARBON DIOXIDE (%)	0
MOISTURE CONTENT (%)	1.39
STACK TEMPERATURE (DEG F)	71
STACK STATIC PRESSURE (INCHES WATER)	-1.3
VELOCITY PRESSURE (SQ RT INCHES WATER DELTA P)	.72
PITOT TUBE CORRECTION FACTOR	.84
VELOCITY (FEET PER SEC)	41.17
EXHAUST GAS FLOW RATE (ACFM)	23319
EXHAUST GAS FLOW RATE (DSCFM)	22052

POLLUTANT DATA

PARAMETER:	TOTAL CHROMIUM	
MOLECULAR WEIGHT=52		
CONCENTRATION (PPM)		51.06×10^{-3}
EMISSION RATE (LBS/HR)		9.25×10^{-3}

PARAMETER:	CHROMIUM +6	
MOLECULAR WEIGHT=52		
CONCENTRATION (PPM)		34.43×10^{-3}
EMISSION RATE (LBS/HR)		6.24×10^{-3}

ENGINEERING-SCIENCE, ARCADIA, CALIFORNIA

PLANT: MCDONNELL DOUGLASS
 SOURCE: SCRUBBER #2 OUTLET
 DATE: 10-3-86
 RUN #: OUTLET

FLOW DATA

STACK LENGTH (INCHES)	38.75
STACK WIDTH (INCHES)	30.25
STACK CROSS-SECTIONAL AREA (SQUARE FT)	8.14
BAROMETRIC PRESSURE (INCHES MERCURY)	29.3
OXYGEN CONTENT (%)	20.8
CARBON DIOXIDE (%)	0
MOISTURE CONTENT (%)	1.43
STACK TEMPERATURE (DEG F)	71
STACK STATIC PRESSURE (INCHES WATER)	-1.3
VELOCITY PRESSURE (SQ RT INCHES WATER DELTA P)	.88
PITOT TUBE CORRECTION FACTOR	.84
VELOCITY (FEET PER SEC)	50.32
EXHAUST GAS FLOW RATE (ACFM)	24576
EXHAUST GAS FLOW RATE (DSCFM)	23231

POLLUTANT DATA

PARAMETER: TOTAL CHROMIUM
 MOLECULAR WEIGHT=52
 CONCENTRATION (PPM)
 EMISSION RATE (LBS/HR)

2.71 $\times 10^{-3}$
 .52 $\times 10^{-3}$

PARAMETER: CHROMIUM +6
 MOLECULAR WEIGHT=52
 CONCENTRATION (PPM)
 EMISSION RATE (LBS/HR)

1.55 $\times 10^{-3}$
 .3 $\times 10^{-3}$

FIELD DATA

PLANT MCD TORONTO
DATE 10-1-86
SAMPLING LOCATION SC# 1
SAMPLE TYPE _____
RUN NUMBER 2
OPERATOR FAL
AMBIENT TEMPERATURE _____
BAROMETRIC PRESSURE _____
STATIC PRESSURE, (P_s) -1.0
FILTER NUMBER (S) _____

PROBE LENGTH AND TYPE _____
NOZZLE I.D. _____
ASSUMED MOISTURE, % _____
SAMPLE BOX NUMBER _____
METER BOX NUMBER 3
METER AN. 1.95 1.07
C FACTOR _____
PROBE HEATER SETTING _____
HEATER BOX SETTING _____
REFERENCE AD _____

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 10 MINUTES

CONDITION SPALGEX OFF

TRAVERSE POINT NUMBER	CLOCK TIME (24-hr CLOCK)		GAS METER READING (V _m), ft ³	VELOCITY HEAD (Δp _s), in. H ₂ O	ORIFICE PRESSURE DIFFERENTIAL (Δh), in. H ₂ O		STACK TEMPERATURE (T _s), °F	DRY GAS METER TEMPERATURE		PUMP VACUUM, in. Hg	SAMPLE BOX TEMPERATURE, °F	IMPINGER TEMPERATURE, °F
	SAMPLING TIME, min				DESIRED	ACTUAL		INLET (T _{m in}), °F	OUTLET (T _{m out}), °F			
	0	1355	221.095	2.7	.75	.75	73			✓		
	10	1405	226.1			.75				✓		
	20	1415	231.8			.75	73			✓		
	30	1425	238.3			.75						
	40	1435	242.2			.75	73			✓		
	50	1445	247.3			.75						
	60	1455	253.2			.75	73					
	70	1505	258.5			.75				2.1		
	80	1515	264.0			.75	73					
	90	1525	269.8			.75				✓		
	100	1535	274.5			.75						
	110	1545										
	120	1555	285.6			.75	73			✓		
	130	1605	291.1			.75				✓		
	140	1615	296.7			.75				✓		
	150	1625	301.7			.75				✓		
	160	1635	307.0			.75				✓		
161	170	1645	307.158	END RUN								
	180	1655		END RUN			73					
			26.0638	15.5V	.75		73					

COMMENTS.

FIELD DATA

1 Qv = 0.87
2 Qv = 0.87

PLANT McD
DATE 10-1-80
SAMPLING LOCATION SCHLAGER #2 INERT
SAMPLE TYPE _____
RUN NUMBER _____
OPERATOR _____
AMBIENT TEMPERATURE _____
BAROMETRIC PRESSURE _____
STATIC PRESSURE, (P_s) _____
FILTER NUMBER (s) _____

PROBE LENGTH AND TYPE _____
NOZZLE I.D. _____
ASSUMED MOISTURE, % _____
SAMPLE BOX NUMBER _____
METER BOX NUMBER _____
METER ΔH _____
C FACTOR _____
PROBE HEATER SETTING _____
HEATER BOX SETTING _____
REFERENCE ΔP _____

SCHEMATIC OF TRAVERSE POINT LAYOUT
READ AND RECORD ALL DATA EVERY 10 MINUTES

TRAVERSE POINT NUMBER	SAMPLING TIME, min	CLOCK TIME (24 hr CLOCK)	GAS METER READING (V _m), in ³	VELOCITY HEAD (ΔP _s), in. H ₂ O	ORIFICE DIFFERENTIAL (ΔH), in. H ₂ O		STACK TEMPERATURE (T _s), °F	DRY GAS METER TEMPERATURE		PUMP VACUUM, in. Hg	SAMPLE BOX TEMPERATURE, °F	IMPIRGER TEMPERATURE, °F
					DESIRED	ACTUAL		INLET (T _{m in}), °F	OUTLET (T _{m out}), °F			
	10:45	01	746.789	.55	.75	.75	872	81		3		
	10:55	10	751.5		.75	.75	73	83				
	10:05	20	756.1		.75	.75	70	86		3		
	10:15	30	760.8		.75	.75		81				
	10:25	40	766.0			.75		88				
	11:35	50	770.1			.75	72	88				
	11:45	60	775.2			.75	73	88				
	70	1155	780.5			.75	73	88		3		
	80	1205	784.6			.75						
	90	1215	789.3			.75						
	100	1225	794.1			.75						
	110	1235	798.9			.75				3		
	120	1245	803.6			.75						
	130	1255	808.4			.75	73			3		
	140	1305	812.1			.75						
	150	1315	817.9			.75						
	160	1325	822.7			.75						
	170	1335	827.9			.75						
	180	1345	832.249			.75						
			854	.55		.75	71	84			77.5	
				15 = .74								

COMMENTS:

FIELD DATA

PLANT McD
 DATE 10-1-86
 SAMPLING LOCATION scrubber #2
 SAMPLE TYPE V
 RUN NUMBER 1
 OPERATOR 10001
 AMBIENT TEMPERATURE _____
 BAROMETRIC PRESSURE _____
 STATIC PRESSURE, (P_s) _____
 FILTER NUMBER (s) _____

PROBE LENGTH AND TYPE _____
 NOZZLE I.D. _____
 ASSUMED MOISTURE, % 24
 SAMPLE BOX NUMBER 2
 METER BOX NUMBER _____
 METER ΔH_g 1.99 1.03
 C FACTOR _____
 PROBE HEATER SETTING _____
 HEATER BOX SETTING _____
 REFERENCE ΔP _____

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 10 MINUTES

TRAVERSE POINT NUMBER	CLOCK TIME (24 hr CLOCK)	GAS METER READING (V _m), ft ³	VELOCITY HEAD (ΔP _v), in. H ₂ O	ORIFICE PRESSURE DIFFERENTIAL (ΔH), in. H ₂ O		STACK TEMPERATURE (T _s), °F	DRY GAS METER TEMPERATURE		PUMP VACUUM, in. Hg	SAMPLE BOX TEMPERATURE, °F	IMPINGER TEMPERATURE, °F
				DESIRED	ACTUAL		INLET (T _{m in}), °F	OUTLET (T _{m out}), °F			
	1355	0	832.351	.75	.75	74	83		2		
	1405	10	837.9	.75	.75	72			2		
	1415	20	842.1	.75	.75	72			2		
	1425	30	846.9	.75	.75	72			2		
	1435	40	851.6	.75	.75	73			2		
	1445	50	856.4	.75	.75	73	83		2		
	1455	60	861.2	.75	.75	76	84		2		
	70	1505	866.0	.75	.75	73	84		2		
	80	1515	870.9	.75	.75	73	83		2		
	90	1525	875.6	.75	.75	73	82		2		
	100	1535	880.4	.75	.75	73	82		2		
	110	1545	885.3	.75	.75	73	82		2		
	120	1555	890.1	.75	.75	73	82		2		
	130	1605	894.8	.75	.75	73	82		2		
	140	1615	899.6	.75	.75	73	82		2		
	150	1625	904.5	.75	.75	72	82		2		
	160	1635	909.355								
	170	1645									
	180	1655									
			77504	.75	.75	73	83			88.7	

COMMENTS:

FIELD DATA

PLANT McCOMBET DOUGLAS
DATE 10-3-86
SAMPLING LOCATION SUBSEA PZ INLET
SAMPLE TYPE METHOD 5
RUN NUMBER 1
OPERATOR C. MOX
AMBIENT TEMPERATURE 75
BAROMETRIC PRESSURE 29.5
STATIC PRESSURE (P_s) -
FILTER NUMBER (S) NONE

PROBE LENGTH AND TYPE 100K / THERMISTOR
NOZZLE I.D. _____
ASSUMED MOISTURE, % _____
SAMPLE BOX NUMBER _____
METER BOX NUMBER 2A
METER ΔH 1.47 / 0 = 1.03
C FACTOR _____
PROBE HEATER SETTING _____
HEATER BOX SETTING _____
REFERENCE ΔP _____

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY 10 MINUTES

TRAVERSE POINT NUMBER	CLOCK TIME		GAS METER READING (V_m) , ft ³	VELOCITY HEAD (ΔP_s) , In. H ₂ O	ORIFICE PRESSURE DIFFERENTIAL (Δh) , In. H ₂ O		STACK TEMPERATURE (T_g) , °F	DRY GAS METER TEMPERATURE		PUMP VACUUM, In. Hg	SAMPLE BOX TEMPERATURE, °F	IMPINGER TEMPERATURE, °F.
	SAMPLING TIME, min	CLOCK TIME 124-HR CLOCK			DESIRED	ACTUAL		INLET $(T_{m \text{ in}})$, °F	OUTLET $(T_{m \text{ out}})$, °F			
	0	11:00	41.425		0.75	0.75	71		Inlet 63			
	10	11:10	46.12		0.75	0.75	74		81			
	20	11:20	57.13		0.75	0.75	73		83			
*	30	11:30	56.07		0.75	0.75	73		83	9		
	40	12:40	61.1		0.75	0.75	78		90	9		
	50	12:50	65.69		0.75	0.75	76		92	9		
	60	13:00	70.4		0.75	0.75				9		
	70	13:10	75.55		0.75	0.75	73		94	9		
	80	13:20	80.64		0.75	0.75				9		
	90	13:30	84.72		0.75	0.75	78		93	9		
	100	13:40	89.34		0.75	0.75				9		
	110	13:50	94.12		0.75	0.75	71		94	9		
	120	14:00	98.933		0.75	0.75				9		
	TT = 120 min		V _m = 57.508	v _p = .72	Δh =	6.75	T _s = 74	T _m =	86	9"	16.8 mH	

COMMENTS: ~~★~~ pump shut off at 11:31 ($V_m = 56.705$) per request of client. Pump turned back on at 12:31.

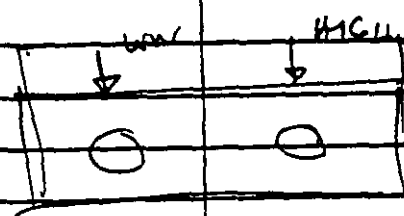
PRELIMINARY VELOCITY TRAVERSE

PLANT TURRAN
 DATE 10-1-86
 LOCATION _____
 STACK I.D. _____
 BAROMETRIC PRESSURE, in. Hg _____
 STACK GAUGE PRESSURE, in. H₂O 1.3
 OPERATORS _____

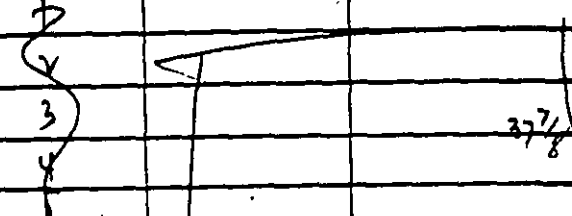
TRI-ACID
 #1

37 7/8" x 31 7/8"

STD

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp _s), in. H ₂ O	STACK TEMPERATURE (T _s), °F
1	.15	63
2	.12	1
3	.15	7
4	.15	
5	.12	
6	.18	
1	.4	63
2	.3	1
3	.25	7
4	.22	
5	.3	
6	.35	
ΣV = (416)		63
		
AVERAGE		

1. ANODIZE TANK
 SCHEMATIC OF TRAVERSE POINT LAYOUT 00
 Scrubber #2 35 7/8" x 37 7/8"

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp _s), in. H ₂ O	STACK TEMPERATURE (T _s), °F
1	.55	63
2	.52	63
3	.52	70
4	.47	70
5		
6		
1	.45	70
2	.45	70
3	.40	70
4	.30	70
5	.35	70
6		
ΣV = (874) CANT READ w/ STD.		
1.3 ps		
		
AVERAGE		

FIELD DATA

PLANT McD + D
DATE 10-3-86
SAMPLING LOCATION scrubber #2 A
SAMPLE TYPE _____
RUN NUMBER _____
OPERATOR RON DIAZ
AMBIENT TEMPERATURE _____
BAROMETRIC PRESSURE _____
STATIC PRESSURE, (P_s) -
FILTER NUMBER (S) NONE

PROBE LENGTH AND TYPE TEFLON RUBBER
NOZZLE I.D. _____
ASSUMED MOISTURE, % _____
SAMPLE BOX NUMBER _____
METER BOX NUMBER 3
METER AM. 193/8-1.02
C FACTOR _____
PROBE HEATER SETTING _____
HEATER BOX SETTING _____
REFERENCE AD _____

SCHEMATIC OF TRAVERSE POINT LAYOUT

READ AND RECORD ALL DATA EVERY _____ MINUTES

[illegible]

COMMENTS

SCHEMATIC OF TRAVERSE POINT LAYOUT

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_s) , in. H ₂ O	STACK TEMPERATURE (T_s) , °F
1	.27	70
2	.30	
3	.27	70
4	.21	
5	.28	70
6	.34	
AVERAGE		

PLANT McDonnell Douglas
DATE 10-1-84
LOCATION SKUNKWORKS #2 (MUST)
STACK I.D. 37" / 2
BAROMETRIC PRESSURE, in. Hg 24.5
STACK GAUGE PRESSURE, in. H₂O _____
OPERATORS _____

SCHEMATIC OF TRAVERSE POINT LAYOUT

[illegible][illegible]

EPA (Dur) 233

472

$$P_{scm} = 233K \left(\frac{50}{533} \right) \left(\frac{14.}{29.9V} \right)$$

$$= 22057 \text{ dscfm}$$

PRELIMINARY VELOCITY TRAVERSE

PLANT MCCORMICK DOWNS
DATE 10-3-86
LOCATION SCRUBBER 2 OUTLET
STACK I.D. _____
BAROMETRIC PRESSURE, in. Hg _____
STACK GAUGE PRESSURE, in. H₂O _____
OPERATORS CM/RD

width 38 ⁶/₈"
30 ¹/₄"

38.75
70.25

SCHEMATIC OF TRAVERSE POINT LAYOUT

0221, 0201.

1

X X
A B

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_g) , in. H ₂ O	STACK TEMPERATURE (T_g) , °F
1	0.60	70
2	0.90	69
3	1.0	70
4	1.0	63
5	1.4	63
6	1.2	63
1	0.65	63
2	0.92	70
3	1.0	70
4	1.1	70
5	1.6	70
6	1.1	70
AVERAGE		

TRAVERSE POINT NUMBER	VELOCITY HEAD (Δp_s), in. H ₂ O	STACK TEMPERATURE (T_s), °F
1	0.25	70
2	0.36	63
3	0.60	70
4	0.66	70
5	0.68	70
6	0.84	70
	$\sqrt{\Delta p_s} =$	87
		68°F
1	0.27	70
2	0.37	65
3	0.62	70
4	0.65	70
5	0.65	70
6	0.80	70
	$\sqrt{\Delta p_s} =$	88
		69°F
AVERAGE		

The Determination of Hexavalent Chromium Emissions using a Diphenylcarbazide Colorimetric Method

SAMPLE ANALYSIS

The EPA Method 5 sampling trains were recovered in the laboratory. The probe wash and impinger solutions were combined, diluted to 500 ml and stored at room temperature. These samples were analyzed for hexavalent chromium within 24 hours after sampling following the procedure below.

The analytical procedure consisted of three steps. The initial step consisted of digesting the samples with an alkaline solution of sodium hydroxide and anhydrous sodium carbonate. The digested samples were then filtered. The second step involved reacting the digested and filtered samples with the diphenylcarbazide solution to produce the characteristic red-violet color which allowed the colorimetric determination of hexavalent chromium. The third step consisted of analyzing the reacted samples spectrophotometrically.

The samples were prepared for analysis by an initial digestion and filtration step. A 50 ml aliquot of each sample was taken and combined with approximately 40 ml of the digesting solution. This mixture was heated to near boiling with constant stirring for thirty minutes and then allowed to cool. After the solution had cooled, it was filtered through a 47-mm diameter, 3.0- μ m pore size Zeflur filter. The filtered solution was then transferred quantitatively into a 100 ml volumetric flask and diluted to the mark. A deionized water blank was prepared for analysis in the same manner.

Then the samples were reacted with a diphenylcarbazide solution. A 50 ml aliquot of the digested and filtered sample was transferred to another 100 ml volumetric flask and diluted with approximately 70 ml of deionized water. The pH of each sample was adjusted to 2 ± 0.5 with the dropwise addition of $\approx 10\% (V/V)$ H_2SO_4 . Two ml of the diphenylcarbazide solution were added to each volumetric flask which was then diluted to the mark. This step was also carried out for seven calibration standards prepared from a standard potassium dichromate stock solution. The seven calibration standard solutions, ranging in concentration from $0.5 \mu g Cr^{+6}$ to $100 \mu g Cr^{+6}$, were used to obtain the spectrophotometric calibration factor, K_c . The samples and calibration standard solutions were allowed to stand for 15 minutes to allow full color development.

After full color development, the samples and calibration standard solutions were analyzed spectrophotometrically at the redetermined optimum wavelength.

EMISSION CALCULATIONS

~~The seven calibration~~

The spectrophotometric calibration factor, K_c , was calculated from the absorbance values obtained by the spectrophotometric measurement of the seven calibration standard solutions. The spectrophotometer calibration factor, K_c , was calculated using the following equation:

$$K_c = 15 \frac{A_1 + 2A_2 + 5A_3 + 10A_4 + 15A_5 + 20A_6}{A_1^2 + A_2^2 + A_3^2 + A_4^2 + A_5^2 + A_6^2}$$

Where:

K_c = calibration factor

A_1 = absorbance of the 5 $\mu\text{g Cr}^{+6}$ standard

A_2 = " " " 10 $\mu\text{g Cr}^{+6}$ "

A_3 = " " " 25 $\mu\text{g Cr}^{+6}$ "

A_4 = " " " 50 $\mu\text{g Cr}^{+6}$ "

A_5 = " " " 75 $\mu\text{g Cr}^{+6}$ "

A_6 = " " " 100 $\mu\text{g Cr}^{+6}$ "

The total $\mu\text{g Cr}^{+6}$ in each sample, m , was calculated using the following equation:

$$m = \frac{500 K_c A F}{V_a}$$

Where

500 = volume in ml of total sample

A = absorbance of sample

F = dilution factor (if required)

V_a = volume in ml of aliquot analyzed (50)

Douglas Aircraft Test Data

12/1/87 RPS

A.) Test Run No. 1

- tested for 3 hours but anodized for only 40 min. of 3h
- averaged 1.5 kiloamperes or 1,500 amps during 40 min of cr+6 anodizing = 1,000 amp-hours
- uncontrolled emission rate (space air on) = 3.6×10^{-3} lbs/h
- Tank surface area = $114 \text{ ft}^2 = 10.6 \text{ m}^2$
- calculation of uncontrolled emissions (mg/Ah)

$$\frac{3.6 \times 10^{-3} \text{ lb}}{\text{h}} \left| \frac{0.4536 \text{ kg}}{\text{lb}} \right| = 1.633 \times 10^{-3} \frac{\text{kg}}{\text{h}} \text{ or } 1.633 \times 10^3 \frac{\text{mg}}{\text{h}}$$

$$\frac{1.633 \times 10^3 \text{ mg}}{\text{h}} \left| \frac{0.67 \text{ h}}{1,000 \text{ Ah}} \right| = 1.1 \text{ mg/Ah} \checkmark$$

$$\cdot (1.633 \times 10^{-3} \text{ kg/h}) \div 10.6 \text{ m}^2 = 1.54 \times 10^{-4} \frac{\text{kg/h}}{\text{m}^2}$$

B.) Test Run No. 2

- tested for 3h and anodized during all 3h (testing was stopped to change pants).
- averaged 1.5 kiloamperes or 1,500 amps during anodizing = 4,500 Ah
- cr+6 uncontrolled emission rate (space air on) = 6.24×10^{-3} lbs/h
- calculation of uncontrolled emissions (mg/Ah)

$$\frac{6.24 \times 10^{-3} \text{ lb}}{\text{h}} \left| \frac{0.4536 \text{ kg}}{\text{lb}} \right| = 2.83 \times 10^{-3} \frac{\text{kg}}{\text{h}} \text{ or } 2.83 \times 10^3 \frac{\text{mg}}{\text{h}}$$

$$\frac{2.83 \times 10^3 \text{ mg}}{\text{h}} \left| \frac{3 \text{ h}}{4,500 \text{ Ah}} \right| = 1.9 \frac{\text{mg}}{\text{Ah}} \checkmark$$

$$\cdot (2.83 \times 10^{-3} \text{ kg/h}) \div (10.6 \text{ m}^2) = 2.67 \times 10^{-4} \frac{\text{kg/h}}{\text{m}^2}$$

Reliable Plating

$$\begin{aligned} \text{Tank dimension} &= \overbrace{11.5, 2.0, 2.0}^{23 \text{ ft}^2} \text{ ft} \\ &= \overbrace{3.5, 0.61, 0.61}^{2.13 \text{ m}^2} \text{ m} \end{aligned}$$

A) Run No. 1

$$\bullet 1.9 \times 10^{-3} \frac{\text{kg}}{\text{h}} = \frac{1.9 \times 10^3 \frac{\text{mg}}{\text{h}}}{1 \text{ h}} \left| \frac{1 \text{ h}}{86.5 \text{ Ah}} \right| = 22 \frac{\text{mg}}{\text{Ah}}$$

$$\bullet (1.9 \times 10^{-3} \text{ kg/h}) \div (2.13 \text{ m}^2) = 8.92 \times 10^{-4} \frac{\text{kg}}{\text{h m}^2}$$

B) Run No. 2

$$\bullet 0.39 \times 10^{-3} \frac{\text{kg}}{\text{h}} = \frac{0.39 \times 10^3 \frac{\text{mg}}{\text{h}}}{1 \text{ h}} \left| \frac{1 \text{ h}}{25.8 \text{ Ah}} \right| = 15 \frac{\text{mg}}{\text{Ah}}$$

$$\bullet (0.39 \times 10^{-3} \text{ kg/h}) \div (2.13 \text{ m}^2) = 1.83 \times 10^{-4} \frac{\text{kg}}{\text{h m}^2}$$

C) Run No. 3

$$\bullet 0.15 \times 10^{-3} \frac{\text{kg}}{\text{h}} = \frac{0.15 \times 10^3 \frac{\text{mg}}{\text{h}}}{1 \text{ h}} \left| \frac{1 \text{ h}}{20 \text{ Ah}} \right| = 7.5 \frac{\text{mg}}{\text{Ah}}$$

$$\bullet (0.15 \times 10^{-3} \text{ kg/h}) \div (2.13 \text{ m}^2) = 7.04 \times 10^{-5} \frac{\text{kg}}{\text{h m}^2}$$

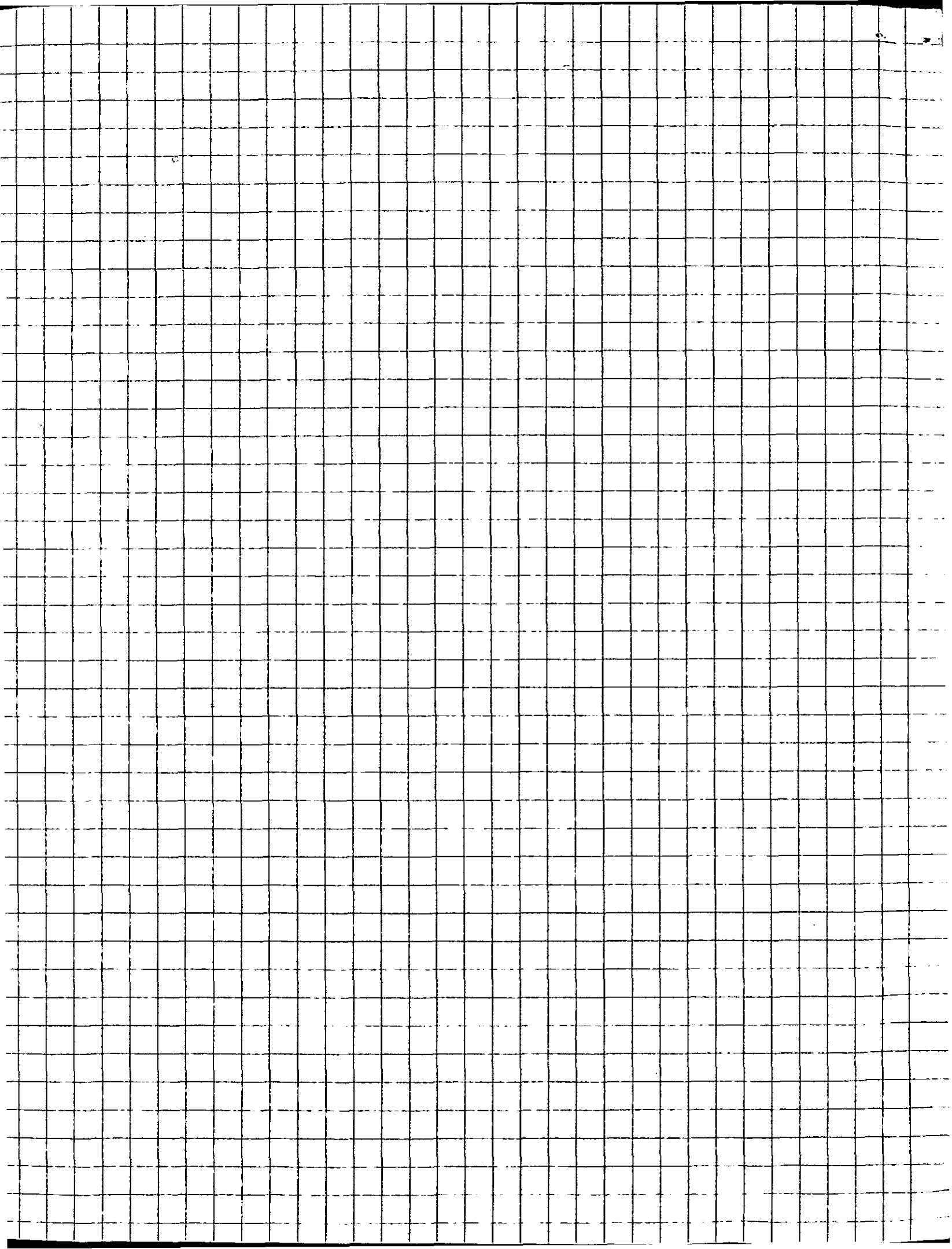
D) Run No. 4

$$\bullet 2.5 \times 10^{-3} \frac{\text{kg}}{\text{h}} = \frac{2.5 \times 10^3 \frac{\text{mg}}{\text{h}}}{1 \text{ h}} \left| \frac{1 \text{ h}}{100 \text{ Ah}} \right| = 25 \frac{\text{mg}}{\text{Ah}}$$

$$\bullet (2.5 \times 10^{-3} \text{ kg/h}) \div (2.13 \text{ m}^2) = 1.17 \times 10^{-3} \frac{\text{kg}}{\text{h m}^2}$$

Avg $\bullet 1.2 \times 10^{-3} \frac{\text{kg}}{\text{h}}$

$$\bullet 5.79 \times 10^{-4} \frac{\text{kg}}{\text{h m}^2} \Rightarrow 6 \times 10^{-4} \frac{\text{kg}}{\text{h m}^2}$$



11/17/86

~~McDonnell~~ Douglas Aircraft Co.

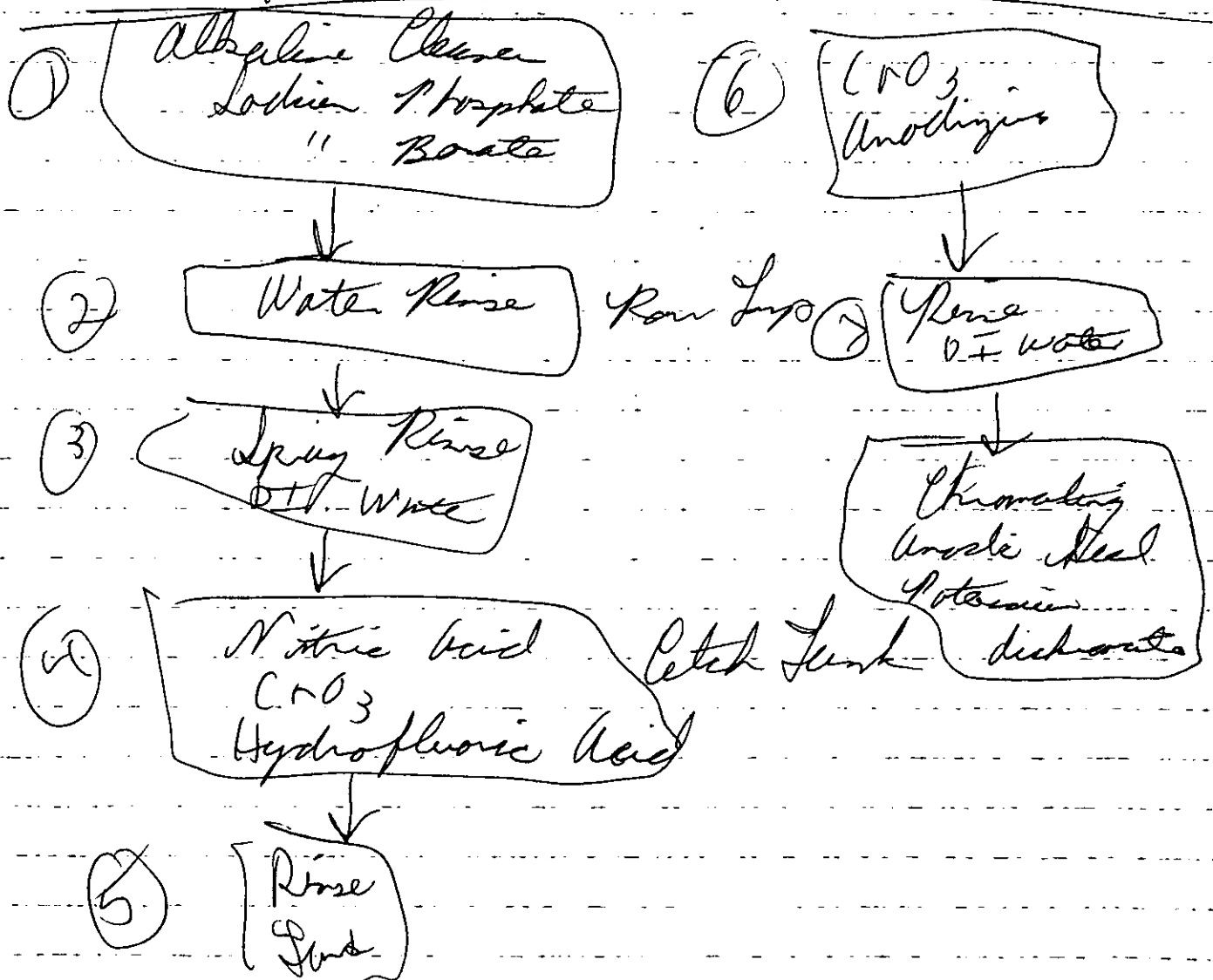
~~Ken Sroboda~~

Ken Sroboda
3855 Lakewood Blvd.
Long Beach, Ca 90846
Mail Code 73-40
(Dept. C1-711)

Kent Adams
Sen. Engr.
Env. Compliance

Env. Compliance Dept.

Aluminum Aircraft Parts.
80% of Commercial & Defense
70% of Parts into Plane



CrO_3 Anodizing Tank

95-100°F

Vol. $\approx$ 5,000 gal

$[\text{CrO}_3]$

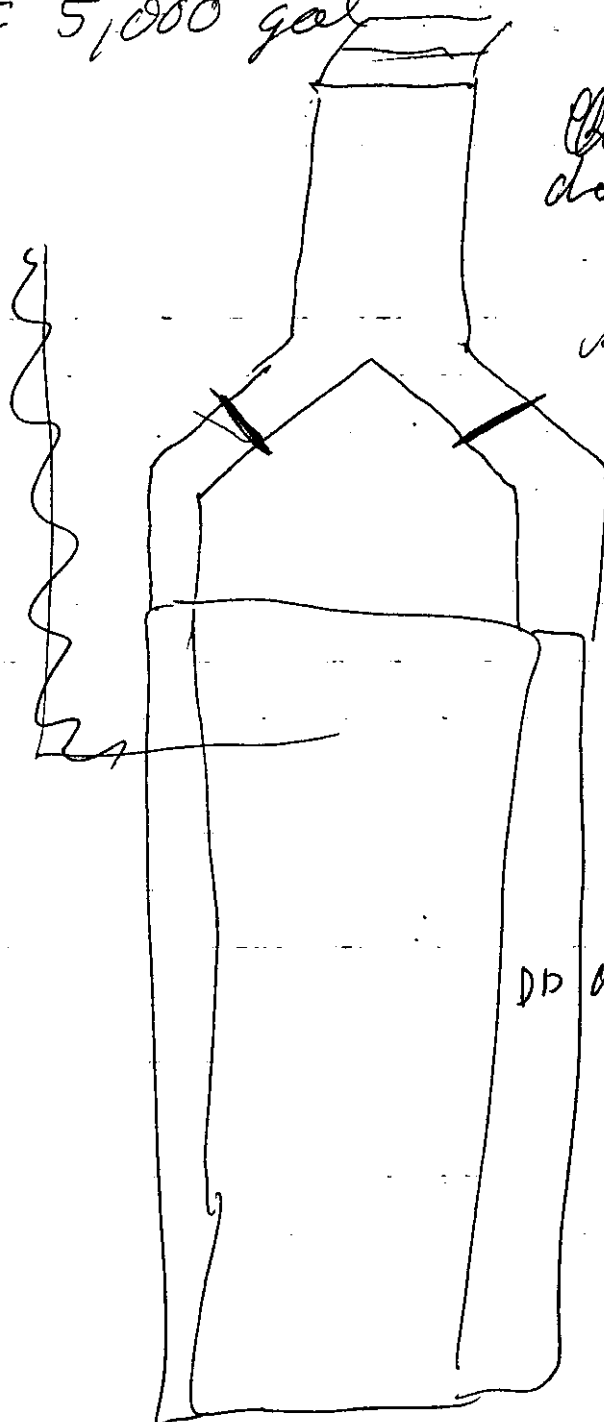
$\approx$ 5% by wt.

7-8 oz/gal.

Obtain more when
drops below 7 oz/gal

Leads &
circuit

1 1/2 batteries
30-40 volts



square
dent

2'

DD draft

etch tank

Deall F100 LFM = 16,000
Serial No. ~~3942~~ 9331
2" SP

20 HP blower

3 HP pump

Anodizing tank

F101 LFM = 21,000

Deall

LFM = 54 SP = 2"

ser no. 3942

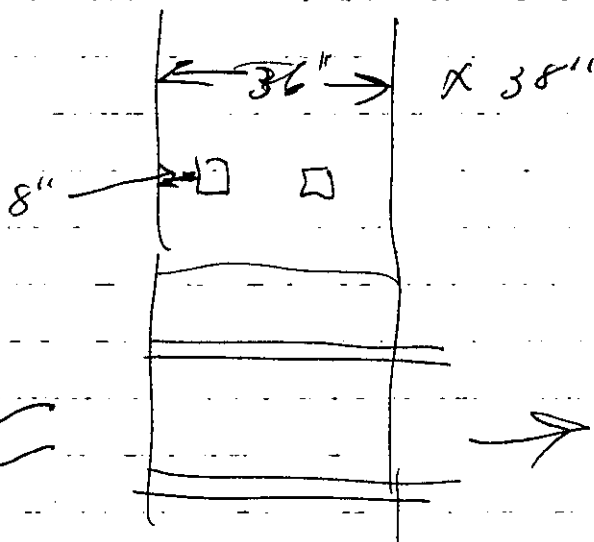
Installed

~~1979~~

1979

(314) 644-1000
Enviroment System
Monsanto
- Bright mist eliminators
- St. Louis MO.
- Bioavia not in
10-40 ft/min
velocity

Anodizing tank



Harrington Ind. Plastics, Inc.
Annaleim

Vortals
packed-bed
(2') scrubbers

(314) 829-9030

Kenofom ME

11/30/87

Ken Svoboda

Anodizing

Run No. 1.: 40 min. anodized
1.5 kamps for 40 min.
3 h of test

Run No. 2.: 3 h anodized
1.5 - 2 kamps on avg use 1.5 kamps
3 h test
3 batches at 40 min each

Cetch Tank No current parts etched for 5 min.
smooth surface

~~Hydrofluoric acid~~
Hydrofluoric acid

Seattle
Wichita, KS

Boeing subcontract anodizing to Northrup

Lockheed in Marietta, Georgia
C-5A military transport planes

Headquarters in
Burbank

Houston Dallas are

General Dynamics

~~Rockwell~~ Rohr, San Diego - subcontractor
for Douglas

Grumman - East Coast

McDonald Douglas

Israel Dyn

McDonald - Douglas

→ Rohr

→

Space equipment manufacturers

1.) Hughes

2.) Rockwell

3.) Martin Marietta

Test report + rich results
→ Will be totally public by middle of Jan.

April & May will be testing new baths

- scrubber + mist eliminators
- Herington - Single packed-bed w/ 2 ft. packing. Vertical flow unit
 - mist eliminator sign wave upstream + mesh pad downstream.
 - inlet + outlet
- got spikes over 1st 5 min. in mean voltage.
- last 35-40 min. avg was constant.
- avg. 2-3 $\mu\text{g}/\text{ft}^2$ over 40 min period
time weighted average.

Long Beach, California - will be testing a new analytical tank in March or April of 1988.

$$\left(3.6 \times 10^{-3} \frac{\text{lb}}{\text{h}} \right) \left(0.4535 \frac{\text{kg}}{\text{lb}} \right) \left($$

$$\left(1.633 \times 10^{-3} \frac{\text{kg}}{\text{h}} \right) \left(10^6 \frac{\text{ng}}{\text{kg}} \right) = 1.633 \times 10^3 \text{ ng/h}$$

$$\left(1.633 \times 10^3 \frac{\text{ng}}{\text{h}} \right) (3 \text{ h}) = 4.9 \times 10^3 \text{ ng}$$

$$\frac{4.9 \times 10^3}{4,500 \text{ Ah}} = 1.0 \text{ ng/Ah}$$

Douglas Aircraft

Long Beach, Calif.

80,000 gal chromic anodizing tanks
South Coast Air Resource

29,000 gal

Eng. Science

(213) 496-6335

Ken Ivorra

Shows a lot about Mist Eliminators
Have test data - 3 runs.
- Need to know what was tested
Where tested, how tested,

700,000 cfm

Coming up w/ proposal

Need 99% control

Carmen Finance

State of Calif. Air Resource Board

(SCAM)

Address: Next to Long Beach Airport off
Freeway.

Best way to get there?

12:15

Test inlet + outlet?
ie Uncontrolled Condition?

SCAND

Test - 4,000 - 5,000 gal tanks

South on ~~Long Beach~~ Harbor Freeway → (S)
405 Freeway (San Diego Freeway) → to
San Diego Freeway route on 405 → Past
Long Beach Freeway → (W) Lakewood Blvd.
→ to (R) on Spring Street →
R at first stoplight onto Airport Plaza
Drive → Second Bldg on Rt next to
Marriott → turn Rt by Marriott

4800 Airport Plaza Drive
Ken Swoda: X 66335