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Background Report Reference

AP-42 Section Number: 12.20

Background Chapter: 4

Reference Number: 61

Title: Source Emission Evaluation

Boeing Company

January 1990

Memo to: Lalit Banker
Standards Development Branch
MD - 13

From: Frank Clay

Subject: Annodizing Test Report
Boeing Company
Plant 2
Chromic Acid Anodizing Tank 5A
Seattle, Washington
August-December 1989

Robin Barker asked me to review the Boeing annodizing test report and I find it unacceptable as far as correct test procedures are concerned. While the calculations were done properly and the report contains all the information needed to verify emission results, there are two deficiencies in their test procedure. These are the port location and the number of points sampled. An extension should have been placed on the stack so that the ports could have been located properly, and 24 instead of 12 sample points should have been used. The incorrect location of the ports may not be as significant as the lack of sufficient sampling points. Past experience has shown that the distribution of chromic acid mist in the stack gas is not uniform and thus the number of sampling points becomes significant if the true emissions are to be known.

There is no preliminary velocity traverse data in the report and consequently, no evidence of any check for cyclonic flow. If significant cyclonic flow did exist and the "blind man's approach" was used in sampling, there is no way to salvage the data.

It is regrettable that this is the only annodizing test data that we have, especially since a few elementary changes in the testing procedure would have produced an acceptable report.

AMTEST

ELECTROPLATING
REF. 4-61

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EMISSION
EVALUATION

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JANUARY 30, 1990

Prepared For:

BOEING COMPANY
PLANT 2
CHROMIC ACID ANODIZING TANK 5A
SEATTLE, WASHINGTON
AUGUST-DECEMBER 1989

Submitted by:

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AM TEST, INC.
REDMOND, WASHINGTON

We certify that the information contained herein is accurate and complete to the best of our knowledge.

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INTRODUCTION

The purpose of this testing program performed at the Boeing Company's Plant No. 2 in Seattle, Washington was to quantify the hexavalent chromium (Cr^{+6}) and total chromium (Cr) emissions exhausted from the #5A chromic acid anodizing tank which is one of the tanks in the 80 anodic tank line located in the 210 Building. The fumes which result from the plating process are controlled by blowing air horizontally across the top of the tank and pulling off the fumes through ductwork with an axial fan to the exhaust stack. The plating cycle takes 55 minutes during which the vat is heated to a temperature of approximately 95° F. Samples were collected over a four (4) month period as coordination would allow.

Three (3) samples were collected while no parts were in the bath (passive or "blank" runs) on August 22, August 29, and September 19, 1989. Eight (8) samples were collected during anodizing cycles on August 22, August 31, September 19, November 30 and December 1, 1990. Three (3) particle size distribution (droplet size) samples were collected during anodizing cycles on August 31 and November 30, 1989. An ambient workroom sample was collected over a 6.5 hour period on August 31, 1989. Samples of the acid in the bath were collected on September 19 and December 1, 1989.

As part of this study, the ventilation system which pulls off air at the surface of the acid bath into vents along one side of the tank was evaluated by measuring the gas velocity at the individual air vents. The velocity was measured on August 22, August 31 and November 30, 1989.

The chromium sample method used is similar to EPA Method 5, which is detailed in the Title 40 Code of Federal Regulations, Part 60 (40 CFR 60), Appendix A. The sample train was modified to exclude the front-half filter and utilized 0.1 N sodium hydroxide (NaOH) solution in the first two (2) impingers, followed by an impinger containing 0.1 N nitric acid (HNO₃), followed by an empty impinger, a final back-half Teflon^R filter to collect any chromium which might break through the absorbing liquid, and a bubbler containing indicating silica gel desiccant. EPA Methods 1-4 were also performed during each chromium test. Methods 1 and 2 were performed to measure the stack gas velocity, temperature and volumetric flow rate. Method 3 was used to calculate the molecular weight of the stack gas. Method 4 was performed to measure the moisture content of the stack gas.

Mr. Kris A. Hansen, Mr. James A. Guenthoer, Mr. K. Steven Mackey, Ms. Jan M. Widmeyer and Ms. Angela F. Blaisdell of Am Test, Inc.'s Air Quality Division based in Redmond, Washington performed the field sampling, data reduction, and final report preparation. Am Test, Inc.'s Trace Metals Division performed the hexavalent chromium and chromium analysis. Mr. Paul Siebenaler coordinated this project for Boeing Aerospace & Electronics.

PRESENTATION OF RESULTS

This report is divided into separate sections for each type of test conducted. Please refer to the Table of Contents to locate information for a specific type of sample. Within each section a summary sheet is included for each run. On the summary sheet the chromium emission concentration is presented in units of milligrams per dry standard cubic meter (mg/dscm) and in parts per million (ppm). The chromium emission rate is presented in units of pounds per hour (lb/hr) and the chromium emission factor is presented in units of milligrams per ampere-hour (mg/amp-hr). Airflow is also presented in units of dry standard cubic feet per minute (dscf/min). The corresponding computer printout and field data sheet follow each summary sheet for each run. The run numbers indicate the run number on a particular test day, not the total number of runs of a particular sample type. The appropriate process information data sheet follows the field data sheet, where applicable. Laboratory analysis results are included in Appendix A of this report. Example calculations of results are included in Appendix B of this report.

In some cases the hexavalent chromium concentrations were greater than the total chromium concentrations. The explanation is that there is a difference in accuracy between the 2 analytical methods (colorimetric versus ICP). The ICP method would likely be the more accurate method.

A brief summary of the chromium tests conducted without parts in the bath and with parts in the bath is presented in Table 1 below.

Table 1. Summary of results from chromium emissions tests conducted at the #5A chromic acid anodizing tank at Boeing Plant 2.

Date	Time	Condition	Hex. Chromium mg/dscm	Hex. Chromium mg/amp-hr	Total Chromium mg/dscm	Total Chromium mg/amp-hr
8/22	1521-1722	Bath Off	0.109	--	0.094	--
8/29	1008-1209	Bath Off	0.105	--	0.096	--
9/19	1022-1223	Bath Off	0.008	--	0.018	--
8/22	1045-1148	Bath On	1.235	23.8	0.805	15.5
8/22	1401-1503	Bath On	1.389	25.7	0.759	14.0
8/31	1215-1315	Bath On	0.438	33.6	0.431	33.1
8/31	1558-1659	Bath On	0.457	36.4	0.418	33.3
9/19	1300-1401	Bath On	0.947	17.7	0.916	17.1
11/30	0803-0904	Bath On	0.261	6.19	0.280	6.64
11/30	1546-1649	Bath On	0.416	8.23	0.423	8.36
12/1	0729-0831	Bath On	0.263	5.94	0.273	6.17
Average			0.512	19.7	0.410	16.8

SECTION 1
"BLANK" RUNS
(NO PARTS IN BATH)

Sample Type:	Bath Off - Blank Run 1
Sampling Date:	August 22, 1989
Sample Times:	15:21-17:22 (120 minutes)
Run Number:	3
Lab Numbers:	915092, 915095, 915099, 915102
Airflow:	20,499 dscf/min
Hexavalent Chromium	
Concentration:	0.109 mg/dscm
Concentration:	0.05 ppm
Emission Rate:	3802.6 mg/hr
Total Chromium	
Concentration:	0.094 mg/dscm
Concentration:	0.04 ppm
Emission Rate:	3261.7 mg/hr

METHOD 1-5 - HEXAVALENT AND TOTAL CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 18045CR3
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: AUGUST 22, 1989
RUN #: 3-CR+6
OPERATORS: GUENTHER/MACKAY
CONTACT: P. STEBENALER

LAB #: 915092
START TIME: 15:21 O'CLOCK
STOP TIME: 17:22 O'CLOCK
SAMPLE TIME: 120.0 MINUTES
CONDITION: BATH OFF

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

905.2 890.8 14.4
381.9 379.7 2.2
473.1 470.2 2.9
745.0 725.6 19.4
TOTAL H2O GAIN: 38.9
TOTAL VOLUME (SCF) 1.83
PERCENT MOISTURE: 1.68
BWS: 0.0168

INIT. METER VOL.: 475.221
FINAL METER VOL.: 587.407
VOLUME SAMPLED: 112.186
STD VOLUME (DSCF): 107.545
STD VOLUME (DSCM): 3.046
Y FACTOR: 0.993

PITOT Cpi: 0.84
NOZZLE DIA. INCHES: 0.313
NOZZLE AREA FT^2: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT^2: 12.566
METER TEMP. DEG F: 90.9
BAROM. PRES. "HG: 30.04
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.04
ORIFICE PRES "H2O: 1.322
METER PRES. "HG: 30.14

AVERAGE X CO2: 0.1
AVERAGE X O2: 20.9
AVERAGE X CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.67

HEXAVALENT CHROMIUM

HEXAVALENT CHROMIUM CONCENTRATION (ug): 332.5
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.109
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.05
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 3802.6

TOTAL CHROMIUM

TOTAL CHROMIUM CONCENTRATION (ug): 285.2
TOTAL CHROMIUM CONC. ON FILTER (ug): < 0.15
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.094
TOTAL CHROMIUM IN SAMPLE (ppm): 0.04
TOTAL CHROMIUM EMISSION RATE (mg/hr): 3261.7

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NE 1	0.19	87	NW 1	0.18	87
2	0.27	87	2	0.28	88
3	0.28	87	3	0.32	89
4	0.28	88	4	0.28	88
5	0.26	87	5	0.27	88
6	0.19	87	6	0.21	88

PERCENT ISOKINETICS: 103 X
STACK TEMPERATURE: 87.6 DEG. F.
AVERAGE VELOCITY HEAD: 0.249 " OF H2O
STACK GAS VELOCITY: 28.56 FT/SEC.
STACK GAS AIR FLOW: 20498.8 DSCF/MIN.

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TRAVERSE SAMPLING DATA

Page 1 of 1

Client Quin-Plant 2
 Date 8-22-89
 Sample Location Blk 2-10
Tank #5 Exhaust
 Operators JAG - KSW
 Sample Box 1
 Run# 3

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm 0.011Leak Test Vac 19

Pitots, Pretest

Pitots, Posttest

Orsat Sampling System

Tedlar Bag

Thermocouple @ _____ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter _____

Distance Upstream _____

Distance Downstream _____

Filter #	tare	mg
	Final Wt.	Initial Wt.
#1 Bubbler	905.2	890.8
#2 Impinger	381.9	379.7
#3 Bubbler	473.1	470.2
#4 Silica Gel	745.0	775.6

TOTAL WATER VOLUME _____

Start Time 1521
 Stop Time 1700
 Barometric Pressure "Hg 763
 Static Pres "H₂O _____
 Production Rate 3AT HOF

NOMOGRAPH SETUP

% Moisture _____

Meter Temp. _____

Stack Temp. _____

ΔH @ _____ Y

Pitot# _____ Side# _____

Cp _____

Nozzle Diameter _____

K Factor _____

Reference ΔP _____

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
NW 1	0	475.22	.19	.59	.99	81	88	5.4		68	87
2	10		.27	1.41	1.41	92	88	6.4	NB	67	87
3	20		.28	1.48	1.48	95	88	7.4		64	87
4	30		.28	1.48	1.48	92	88	7.4		65	88
5	40		.24	1.37	1.37	92	89	6.4		66	87
6	50		.19	1.02	1.02	92	89	6		67	87
NE 1	60		.18	.95	.95	94	89	6		68	87
2	70		.28	1.48	1.48	95	89	7.4		66	88
3	80		.32	1.69	1.69	96	89	8.4		66	87
4	90		.28	1.48	1.48	95	87	7.4		67	88
5	100		.27	1.42	1.42	92	85	7		67	88
6	110		.21	1.11	1.11	91	82	6.4		68	88
EMP	120	587.407									

1.322 90.9

Sample Type:	Bath Off - Blank Run 2
Sampling Date:	August 29, 1989
Sample Times:	10:08-12:09 (120 minutes)
Run Number:	1
Lab Numbers:	915486, 915487, 915489, 915491
Airflow:	21,403 dscf/min
Hexavalent Chromium	
Concentration:	0.105 mg/dscm
Concentration:	0.05 ppm
Emission Rate:	3818.9 mg/hr
Total Chromium	
Concentration:	0.096 mg/dscm
Concentration:	0.04 ppm
Emission Rate:	3492.5 mg/hr

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 2805CR1
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: AUGUST 29, 1989
RUN #: 1-CR+6
OPERATORS: GUENTHER/MACKAY
CONTACT: P. SIEBENALER

LAB #: 915486
START TIME: 10:08 O'CLOCK
STOP TIME: 12:09 O'CLOCK
SAMPLE TIME: 120.0 MINUTES
CONDITION: BATH OFF

HEXAVALENT CHROMIUM
HEXAVALENT CHROMIUM CONCENTRATION (ug): 331.5
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.105
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.05
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 3818.9

TOTAL CHROMIUM
TOTAL CHROMIUM CONCENTRATION (ug): 301.2
TOTAL CHROMIUM CONC. ON FILTER (ug): 2.0
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.096
TOTAL CHROMIUM IN SAMPLE (ppm): 0.04
TOTAL CHROMIUM EMISSION RATE (mg/hr): 3492.5

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

1263.4 1251.8 11.6
634.2 631.9 2.3
478.7 473.9 4.8
772.8 746.1 26.7

TOTAL H2O GAIN: 45.4 ✓
TOTAL VOLUME (SCF) 2.14 ✓
PERCENT MOISTURE: 1.88 ✓
BWS: 0.0188

PITOT Cp: 0.840
NOZZLE DIA INCHES: 0.313 ✓
NOZZLE AREA FT²: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT²: 12.566
METER TEMP. DEG F: 78.5 ✓
BAROM. PRES. "HG: 30.04
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.04
ORIFICE PRES "H2O: 1.403 ✓
METER PRES. "HG: 30.14

INIT. METER VOL.: 734.162
FINAL METER VOL.: 847.802 ✓
VOLUME SAMPLED: 113.640 ✓
STD VOLUME (DSCF): 111.469 ✓
STD VOLUME (OSCM): 3.157
Y FACTOR: 0.993

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.65

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NW 1	0.22	81	NE 1	0.23	82
2	0.27	82	2	0.29	82
3	0.30	82	3	0.33	83
4	0.29	82	4	0.29	84
5	0.26	82	5	0.29	84
6	0.21	83	6	0.27	85

PERCENT ISOXINETICS: 102 %
STACK TEMPERATURE: 82.7 DEG. F.
AVERAGE VELOCITY HEAD: 0.270 " OF H2O
STACK GAS VELOCITY: 29.62 FT/SEC.
STACK GAS AIR FLOW: 22330.2 ACF/MIN.

2.2189

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2 B. H. CR 1 9/15/88 TRAVERSE SAMPLING DATA

Page 1 of 1

Client PREMCO
 Date 9-29-89
 Sample Location BLDG 2-10
LAKE C. POND
 Operators CR 1
 Sample Box # A
 Run# -

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm 4.01 / .007Leak Test Vac 16 / 17

Pitots, Pretest

Pitots, Postest

Orsat Sampling System

Tedlar Bag

Thermocouple @ - °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter -Distance Upstream -Distance Downstream -

Filter # - care - mgs
 Final Initial Net
 Wt. Wt. Wt.

#1 Bubbler 1263.4 - 1251.7 -#2 Impinger 634.2 - 631.9 -#3 Bubbler 478.7 - 473.9 -#4 Silica Gel 772.8 - 746.1 -TOTAL WATER VOLUME -

Start Time 1009
 Stop Time 1207
 Barometric
 Pressure "Hg 763
 Static Pres "H₂O -
 Production Rate
BATH OFF

NOMOGRAPH SETUP

X Moisture 2Meter Temp. 75Stack Temp. 80ΔH_e .942 γ .993Pitot # 46 Side # ACp 240Nozzle Diameter .313K Factor 5.18Reference ΔP -

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
NE 1	0	734.162	.22	1.14	1.14	66	64	6 1/2		67	81
2	10		.27	1.40	1.40	74	64	7	N/A	61	82
3	20		.30	1.56	1.56	80	70	8 1/2		61	87
4	30		.29	1.50	1.50	83	72	8 1/2		58	82
5	40		.26	1.35	1.35	85	74	7		59	87
6	50		.21	1.04	1.09	85	76	6		59	83
WE 1	60		.23	1.19	1.19	83	76	6 1/2		60	82
2	70		.29	1.50	1.50	85	77	9		61	82
3	80		.33	1.71	1.71	88	79	9 1/2		63	83
4	90		.29	1.50	1.50	89	79	8		64	84
5	100		.29	1.50	1.50	88	80	8		60	84
6	110		.27	1.40	1.40	88	80	7 1/2		65	85
EXP	120	847.802									

1.403 78.5

Sample Type:	Bath Off - Blank Run 3
Sampling Date:	September 19, 1989
Sample Times:	10:22-12:23 (120 minutes)
Run Number:	1
Lab Numbers:	917067, 917068, 917069, 917077
Airflow:	20,651 dscf/min
Hexavalent Chromium	
Concentration:	0.008 mg/dscm
Concentration:	0.004 ppm
Emission Rate:	272.0 mg/hr
Total Chromium	
Concentration:	0.018 mg/dscm
Concentration:	0.009 ppm
Emission Rate:	645.4 mg/hr

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 4805CR1
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
RUN #1: SEPTEMBER 19, 1989
OPERATORS: GUENTHER/MACKEY
CONTACT: P. SIEBENALER

LAB #: 917077
START TIME: 10:22 O'CLOCK
STOP TIME: 12:23 O'CLOCK
SAMPLE TIME: 120.0 MINUTES
CONDITION: BATH OFF

FINAL WT. INIT. WT. NET WT. OF H2O G. OF H2O G.
448.5 454.9 -6.4
480.5 464.0 16.5
434.9 429.1 5.8
320.1 318.0 2.1
686.0 669.9 16.1
TOTAL H2O GAIN: 34.1
TOTAL VOLUME (SCF) 1.61
PERCENT MOISTURE: 1.51
BWS: 0.0151

PITOT Cp: 0.84
NOZZLE DIA. INCHES: 0.313
NOZZLE AREA FT²: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT²: 12.566
METER TEMP. DEG F: 74.6
BAROM. PRES. "HG: 30.24
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.24
ORIFICE PRES "H2O: 1.255
METER PRES. "HG: 30.33

INIT. METER VOL.: 444.328
FINAL METER VOL.: 550.148
VOLUME SAMPLED: 105.820
STD VOLUME (DSCF): 105.212
STD VOLUME (DSCM): 2.980
Y FACTOR: 0.993

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.69

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NW 1	0.20	78	NE 1	0.21	78
2	0.23	78	2	0.27	79
3	0.27	78	3	0.31	80
4	0.27	79	4	0.27	81
5	0.24	79	5	0.26	82
6	0.20	79	6	0.24	83

PERCENT ISOKINETICS: 100 %
STACK TEMPERATURE: 79.5 DEG. F.
AVERAGE VELOCITY HEAD: 0.246 " OF H2O
STACK GAS VELOCITY: 28.11 FT/SEC.
STACK GAS AIR FLOW: 21196.6 ACF/MIN.
20651.0 DSCF/MIN.

HEXAVALENT CHROMIUM

HEXAVALENT CHROMIUM CONCENTRATION (ug): 23.1
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.008
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.004
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 272.0

TOTAL CHROMIUM

TOTAL CHROMIUM CONCENTRATION (ug): 56.6
TOTAL CHROMIUM CONC. ON FILTER (ug): 0.175
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.018
TOTAL CHROMIUM IN SAMPLE (ppm): 0.009
TOTAL CHROMIUM EMISSION RATE (mg/hr): 645.4

AT

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TRAVERSE SAMPLING DATA

Page 1 of 1

4804-501 917077

Client Bayer Plant
 Date 9/10/89
 Sample Location Task #5 Vent Stack
(Booth Off Test)
 Operators RSM/JAG
 Sample Box # 6144
 Run# 1 - Chromatography/Heavy Metals Suppl.

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm .005 / .004Leak Test Vac 20 / 15"☒ Pitots, Pretest☒ Pitots, Posttest☒ Orsat Sampling System☒ Tedlar Bag☒ Thermocouple @ _____ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter _____
 Distance Upstream _____
 Distance Downstream _____

Filter #	N/A	tare	mgs
	Final Wt.	Initial Wt.	Net Wt.
#1	NaOH	448.5	454.9
#2	NaOH	420.5	464.0
#3	H ₂ O ₂	434.9	429.1
#4	MT Silica	320.1	318.0
	Gel	686.0	669.9

TOTAL WATER VOLUME _____

Start Time 10:22
 Stop Time 12:23
 Barometric 30.24"
 Pressure "Hg 768 mm.
 Static Pres "H₂O -0.02
 Production Rate
1.74 - 0.15

NOMOGRAPH SETUP

X Moisture _____
 Meter Temp. _____
 Stack Temp. 78
 ΔH 943 y. 992
 Pitot# 4' Glass Side# 1
 Cp .84
 Nozzle Diameter .313
 K Factor _____
 Reference ΔP _____

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
1	0	444.328	.20	1.01	1.01	60	59	2.8	—	55	78
2	10		.23	1.17	1.17	66	60	3.1	—	49	78
3	20		.27	1.37	1.37	70	62	3.1	—	51	78
4	30		.27	1.37	1.37	75	65	3.1	—	50	79
5	40		.24	1.22	1.22	79	68	3.0	—	51	79
6	50		.20	1.01	1.01	81	71	2.9	—	52	79
7	60		.21	1.06	1.06	81	73	2.9	—	55	78
8	70		.27	1.37	1.37	84	75	3.1	—	53	79
9	80		.31	1.57	1.57	86	77	3.4	—	54	80
10	90		.27	1.37	1.37	88	78	3.3	—	52	81
11	100		.26	1.32	1.32	87	79	3.2	—	53	82
12	110		.24	1.22	1.22	87	79	3.1	—	50	83
13	120	550.148									

1.255 / 74.6

SECTION 2
BATH ON RUNS
(PARTS IN BATH)

Sample Type:	Bath On - Run 1
Sampling Date:	August 22, 1989
Sample Times:	10:45-11:48 (60 minutes)
Run Number:	1
Lab Numbers:	915089, 915090, 915093, 915097 915100
Airflow:	20,704 dscf/min
Process Data	
Bath Amperage:	1825 amp-hr/hr
Voltage:	22.7 volts D.C.
Bath Temperature:	94° F
Ambient Temperature:	74° F
Hexavalent Chromium	
Concentration:	1.235 mg/dscm
Concentration:	0.57 ppm
Emission Rate:	43,457 mg/hr
Emission Factor:	23.8 mg/amp-hr
Total Chromium	
Concentration:	0.805 mg/dscm
Concentration:	0.37 ppm
Emission Rate:	28,305 mg/hr
Emission Factor:	15.5 mg/amp-hr

FILE NAME: 18045CR1
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: AUGUST 22, 1989
RUN #: 1-CR+6
OPERATORS: GUENTHER/MACEY
CONTACT: P. SIEBENALER

LAB #: 915090
START TIME: 10:45 O'CLOCK
STOP TIME: 11:48 O'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

HEXAVALENT CHROMIUM CONCENTRATION (ug):	1874.0
HEXAVALENT CHROMIUM CONCENTRATION (mg/ds.cm):	1.235
HEXAVALENT CHROMIUM IN SAMPLE (ppm):	0.57
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr):	43456.5
ASPHENRE HOURS PER HOUR (amp-hr/hr):	1825
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr):	23.8

PITOT CP:	0.84
NOZZLE DIA INCHES:	0.313
NOZZLE AREA FT^2:	0.0005
STACK DIA. INCHES:	48.00
STACK AREA FT^2:	12.566
BAROM. TEMP. DEG F:	75.0
BAROM. PRES. "HG:	30.04
STATIC PRES. "H2O:	0.00
STACK PRES. "HG:	30.04
ORIFICE PRES "H2O:	1.278
METER PRES. "HG:	30.13

[illegible]

1252.3	1246.9	5.4
628.3	628.2	0.1
472.5	471.4	1.1
722.0	711.9	10.1
TOTAL H2O GAIN:		16.7
TOTAL VOLUME (SCF)		0.79
PERCENT MOISTURE:		1.45
BWS:		0.0145

AVERAGE % CO2:	0.1
AVERAGE % O2:	20.9
AVERAGE % CO:	0
STACK GAS MW. DRY:	28.85
STACK GAS MW. WET:	28.69

INIT. METER VOL.:	364.682
FINAL METER VOL.:	418.957
VOLUME SAMPLED:	54.275
STD VOLUME (DSCF):	53.570
STD VOLUME (DSCM):	1.517
Y FACTOR:	0.993

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NW 1	0.20	78	NE 1	0.19	78
2	0.25	79	2	0.27	78
3	0.27	78	3	0.31	78
4	0.28	78	4	0.28	79
5	0.26	78	5	0.26	80
6	0.20	78	6	0.23	80

PERCENT ISOKINETICS:	101 X	538.5 DEG. R.
STACK TEMPERATURE:	78.5 DEG. F.	
AVERAGE VELOCITY HEAD:	0.249 " OF H2O	
STACK GAS VELOCITY:		28.30 FT/SEC.
STACK GAS AIR FLOW:	21340.7 ACF/MIN.	20704.1 DSCF/MIN.

August 22, 1989

<u>Time</u>	<u>Amps</u>
<u>Run 1:</u> 1050	200
1055	1820
1100	1820
1105	1820
1110	1820
1115	1820
1120	1820
1145	1820

Cycle is 55 minutes

1825 $\frac{\text{amp-hr}}{\text{hr}}$

ambient temp 74°F

<u>Time</u>	<u>Amps</u>
<u>Run 2:</u> 1400	0
1401	200
1403	1820
1405	1820
1407	1820
1410	1820
1415	1820
1420	1820
1425	1820
1430	1820
1435	1820
1440	1820
1445	1820
1450	1820
1455	1820

3 min.

52 min.

Sample Calculation - Run 2

$$(3 \text{ min} * 200 \text{ amps} + 52 \text{ min} * 1820 \text{ amps}) / 55 \text{ min} * \frac{60 \text{ min/hr}}{55 \text{ min}}$$

$$= \underline{\underline{1889}} \text{ amp-hr/hr}$$

Sample Type:	Bath On - Run 2
Sampling Date:	August 22, 1989
Sample Times:	14:01-15:03 (60 minutes)
Run Number:	2
Lab Numbers:	915091, 915094, 915098, 915101
Airflow:	20,555 dscf/min
Process Data	
Bath Amperage:	1889 amp-hr/hr
Voltage:	22.7 volts D.C.
Bath Temperature:	94° F
Ambient Temperature:	83° F
Hexavalent Chromium	
Concentration:	1.389 mg/dscm
Concentration:	0.64 ppm
Emission Rate:	48,503 mg/hr
Emission Factor:	25.7 mg/amp-hr
Total Chromium	
Concentration:	0.759 mg/dscm
Concentration:	0.35 ppm
Emission Rate:	26,494 mg/hr
Emission Factor:	14.0 mg/amp-hr

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 1B05CR2
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: AUGUST 22, 1989
RUN #: 2-CR+6
OPERATORS: GUENTHER/MACKAY
CONTACT: P. STEBENALER

LAB #: 915091
START TIME: 14:01 O'CLOCK
STOP TIME: 15:03 O'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

HEXAVALENT CHROMIUM
HEXAVALENT CHROMIUM CONCENTRATION (ug): 2092.5
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 1.389
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.642
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 48502.8
AMPHERE HOURS PER HOUR (amp-hr/hr): 1889
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr): 25.7

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

1253.0 1247.1 5.9
624.8 623.6 1.2
532.0 531.5 0.5
833.0 822.8 10.2
TOTAL H2O GAIN: 17.8
TOTAL VOLUME (SCF) 0.84
PERCENT MOISTURE: 1.55
BWS: 0.0155

PITOT CP: 0.84
NOZZLE DIA INCHES: 0.313
NOZZLE AREA FT^2: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT^2: 12.566
METER TEMP. DEG F: 85.9
BAROM. PRES. "HG: 30.04
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.04
ORIFICE PRES "H2O: 1.294
METER PRES. "HG: 30.14

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.68

TOTAL CHROMIUM
TOTAL CHROMIUM CONCENTRATION (ug): 1143.0
TOTAL CHROMIUM CONC. ON FILTER (ug): < 0.18
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.759
TOTAL CHROMIUM IN SAMPLE (ppm): 0.35
TOTAL CHROMIUM EMISSION RATE (mg/hr): 26494.0
AMPHERE HOURS PER HOUR (amp-hr/hr): 1889
TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr): 14.0

INIT. METER VOL.: 419.767
FINAL METER VOL.: 474.771
VOLUME SAMPLED: 55.004
STD VOLUME (DSCF): 53.208
STD VOLUME (DSCM): 1.507
Y FACTOR: 0.993

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NE 1	0.21	86	NW 1	0.18	86
2	0.27	86	2	0.26	86
3	0.31	85	3	0.28	86
4	0.27	85	4	0.27	85
5	0.26	85	5	0.25	85
6	0.24	86	6	0.20	84
PERCENT ISOINETICS:					
STACK TEMPERATURE:		102 X	545.4 DEG. F.		
AVERAGE VELOCITY HEAD:		85.4 DEG. F.	28.49 FT/SEC.		
STACK GAS VELOCITY:		0.249 " OF H2O	20555.4 DSCF/MIN.		
STACK GAS AIR FLOW:		21482.2 ACF/MIN.			



TRAVERSE SAMPLING DATA

Page 1 of

18045012

915091

Client BOEING-PLNT #2
 Date 8-22-89
 Sample Location BLDG 2-10
TANK #5 Exhaust
 Operators JAG-RSM
 Sample Box 1
 Run# 2-Gr

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm 1.04/1.01Leak Test Vac 12/18

Pitots, Pretest

Pitots, Posttest

Orsat Sampling System

Tedlar Bag

Thermocouple @ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter

Distance Upstream

Distance Downstream

Filter # 1 tare 1253.0 mgs
 Final Initial Net
 Wt. Wt. Wt.

#1 Bubbler 1253.0 - 1247.1 -#2 Impinger 624.8 - 623.6 -#3 Bubbler 532.0 - 531.5 -#4 Silica Gel 833.0 - 822.8 -

TOTAL WATER VOLUME

Start Time 14:01
 Stop Time 15:05
 Barometric
 Pressure "Hg 763
 Static Pres "H₂O
 Production Rate
13.44 - 0.00

NOMOGRAPH SETUP

X Moisture 2

Meter Temp.

Stack Temp.

ΔH .972 Y .993Pitot# 1 Side#Cp .84Nozzle Diameter .313

K Factor

Reference ΔP

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F	
				Ideal	Actual	In	Out					
ME1	0	419.767	.21	1.08	1.08	77	77	7		60	86	
2	5		.27	1.39	1.37	81	78	8		55	86	
3	10		.31	1.59	1.59	83	79	10		57	85	
4	15		.27	1.39	1.39	87	79	8	N/A	59	85	
5	20		.26	1.34	1.34	90	81	8		59	85	
6	25		.24	1.23	1.23	91	82	7.5		59	86	
UW1	30		.18	.92	.92	87	83	5.5		64	86	
2	35		.26	1.36	1.36	91	84	8		62	86	
3	40		.28	1.46	1.46	94	82	9		60	86	
4	45		.27	1.41	1.41	96	86	8.5		62	85	
5	50		.25	1.31	1.31	97	87	8		62	85	
6	55		.20	1.05	1.05	98	88	7		63	84	
END	60	474.321										

1294 85.9

August 22, 1989

<u>Time</u>	<u>Amps</u>
<u>Run 1:</u> 1050	200
1055	1820
1100	1820
1105	1820
1110	1820
1115	1820
1120	1820
1145	1820

Cycle is 55 minutes

$$\frac{1825 \text{ amp-hr}}{\text{hr}}$$

<u>Time</u>	<u>Amps</u>
<u>Run 2:</u> 1400	0
1401	200
1403	1820
1405	1820
1407	1820
1410	1820
1415	1820
1420	1820
1425	1820
1430	1820
1435	1820
1440	1820
1445	1820
1450	1820
1455	1820

3 min.

52 min

Sample Calculation - Run 2

$$(3 \text{ min} * 200 \text{ amps} + 52 \text{ min} * 1820 \text{ amps}) / 55 \text{ min} * \frac{60 \text{ min/hr}}{55 \text{ min}}$$

$$= \underline{\underline{1889}} \text{ amp-hr/hr}$$

ambient temp 83°F
 bath temp 94°F
 temp. above tank 86°F

Sample Type:	Bath On - Run 3
Sampling Date:	August 31, 1989
Sample Times:	12:15-13:15 (60 minutes)
Run Number:	2
Lab Numbers:	915750, 915751, 915752, 915753 915754
Airflow:	21,546 dscf/min
Process Data	
Bath Amperage:	477 amp-hr/hr (small load)
Voltage:	21.8 volts D.C.
Bath Temperature:	94° F
Ambient Temperature:	82° F
Hexavalent Chromium	
Concentration:	0.438 mg/dscm
Concentration:	0.20 ppm
Emission Rate:	16,029 mg/hr
Emission Factor:	33.6 mg/amp-hr
Total Chromium	
Concentration:	0.431 mg/dscm
Concentration:	0.20 ppm
Emission Rate:	15,775 mg/hr
Emission Factor:	33.1 mg/amp-hr

METHOD 1-5 - HEXAVALENT AND TOTAL CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 380W5CR2
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: AUGUST 31, 1989
RUN #: 2-CR-6
OPERATORS: GUENTHER/MIDMEYER
CONTACT: P. SIEBENWALDER

LAB #: 915750
START TIME: 12:15 O'CLOCK
STOP TIME: 13:15 O'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

HEXAVALENT CHROMIUM
HEXAVALENT CHROMIUM CONCENTRATION (ug): 689.3
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.438
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.203
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 16028.6
AMPHERE HOURS PER HOUR (amp-hr/hr): 477
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr): 33.6

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

1188.1 1180.1 8.0
570.0 570.2 -0.2
480.6 480.8 -0.2
736.2 724.7 11.5
TOTAL H2O GAIN: 19.1
TOTAL VOLUME (SCF) 0.90
PERCENT MOISTURE: 1.59
BWS: 0.0159

PITOT CP: 0.840
NOZZLE DIA INCHES: 0.313
STACK AREA FT^2: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT^2: 12.566
METER TEMP. DEG F: 77.5
BAROM. PRES. "HG: 30.04
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.04
ORIFICE PRES "H2O: 1.418
METER PRES. "HG: 30.14

TOTAL CHROMIUM
TOTAL CHROMIUM CONCENTRATION (ug): 678.4
TOTAL CHROMIUM CONC. ON FILTER (ug): < 0.30
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.431
TOTAL CHROMIUM IN SAMPLE (ppm): 0.20
TOTAL CHROMIUM EMISSION RATE (mg/hr): 15774.9
AMPHERE HOURS PER HOUR (amp-hr/hr): 477
TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr): 33.1

INIT. METER VOL.: 2.263
FINAL METER VOL.: 58.832
VOLUME SAMPLED: 56.569
STD VOLUME (DSCF): 55.594
STD VOLUME (DSCM): 1.574
Y FACTOR: 0.993

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.68

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NW 1	0.24	80	NE 1	0.24	83
2	0.30	80	2	0.28	83
3	0.33	81	3	0.31	83
4	0.29	82	4	0.31	84
5	0.26	83	5	0.29	84
6	0.24	83	6	0.19	83

PERCENT ISOKINETICS: 101 %
STACK TEMPERATURE: 82.4 DEG. F.
AVERAGE VELOCITY HEAD: 0.272 " OF H2O
STACK GAS VELOCITY: 29.71 FT/SEC.
STACK GAS AIR FLOW: 22402.8 ACF/MIN. 21545.8 DSCF/MIN.



415750 TRAVERSE SAMPLING DATA

Page 1 of

Client P. Boering
 Date 8.31.87
 Sample Location T312-2-10
 Tank & Exhaust
 Operators JAG - Jmm
 Sample Box # 1
 Run# 2 Cr 1/ Cr

EQUIPMENT CHECKS

Initial/Final
 Leak Rate Cfm 1.07 .008
 Leak Test Vac 15 120
☒ Pitots, Pretest
☒ Pitots, Posttest
☒ Orsat Sampling System
☒ Tedlar Bag
 Thermocouple @ _____ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter _____
 Distance Upstream _____
 Distance Downstream _____

Filter # _____ tare _____ mgs
 Final Initial Net
 Wt. Wt. Wt.
 #1 Bubblers 585.3 602.8 - 1180.1 = _____
 #2 Impinger 570.0 570.2 = _____
 #3 Bubblers 480.6 480.8 = _____
 #4 Silica Gel 736.2 724.7 = _____
 TOTAL WATER VOLUME _____

Start Time 1215
 Stop Time 1315
 Barometric Pressure "Hg 71.6
 Static Pres "H₂O _____
 Production Rate _____

NOMOGRAPH SETUP

% Moisture _____
 Meter Temp. _____
 Stack Temp. _____
 ΔH@ 942.5 °F _____
 Pitot# _____ Side# _____
 Cp 340
 Nozzle Diameter 3.3
 K Factor _____
 Reference ΔP _____

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
UV 1	0	2.263	.24	1.20	1.28	70	70	6		64	80
2	5		.30	1.54	1.54	74	70	7		58	80
3	10		.33	1.69	1.69	79	71	7.4	NA	63	81
4	15		.27	1.07	1.49	80	72	7		62	82
5	20		.26	1.34	1.34	82	73	6.4		60	82
6	25		.24	1.25	1.25	83	74	6		60	83
UV 1	30		.24	1.25	1.25	82	75	6		61	83
2	35		.28	1.46	1.46	84	76	7		58	83
3	40		.31	1.61	1.61	85	76	7.4		58	83
4	45		.21	1.61	1.61	86	77	7.4		60	84
5	50		.29	1.51	1.51	86	77	7		60	84
6	55		.19	1.99	1.99	85	77	6		60	83
END	60	8.932									

1.418 77.7

8/31/84

27

Run Z
(Method 5/4)

Parts into bath @ 12:13
Parts removed @ 1:09

Material from bath is 3 runs of 2 3/8" tubing (6 ft tubes total)

Called Relatively Small Load & this is why Amps

55 min cycle

1 min	12:14	5 V
1 min	12:15	5 V
3 min	12:16	12.0 V
7 min	12:19	22.3 V
	12:26	22.9 V
	12:33	23.2 V
43 min	12:43	23.2 V
	12:53	23.2 V
	1:02	23.3 V
	1:09	0.5 V (OFF)

200 Amps
250 Amps
550 Amps
680 Amps
450 Amps
400 Amps
400 Amps
400 Amps
400 Amps
0 Amps

($1 \times 200 + 1 \times 250 + 3 \times 550 + 7 \times 680 + 43 \times 400$) / 55 = 60

94.0°F Bath Temp
82°F Ambient

477 Amp-hr / hr
21.8 volts

$1 \times 5 + 1 \times 5 + 3 \times 12 + 7 \times 22.3 + 7 \times 22.9 + 36 \times 23.2$

Third Amalgam Bath without stack testing:

Parts into bath @ 1:19
Parts removed from bath @ 2:14

23.2V 1820 Amps
94.0°F 87°F Ambient

Velocity Measurements after & before Run #1

Opening #

1
6
11
16
21
26
31
36
41
46
51
56

ft/min

2380
2230
1650
1700
1610
1390
720
1000
820
790
720
280

Measured 2:40 - 2:50 PM

Sample Type:	Bath On - Run 4
Sampling Date:	August 31, 1989
Sample Times:	15:58-16:59 (60 minutes)
Run Number:	3
Lab Numbers:	915755, 915756, 915757, 915758 915759
Airflow:	21,746 dscf/min
Process Data	
Bath Amperage:	464 amp-hr/hr (small load)
Voltage:	21.5 volts D.C.
Bath Temperature:	94° F
Ambient Temperature:	82° F
Hexavalent Chromium	
Concentration:	0.457 mg/dscm
Concentration:	0.21 ppm
Emission Rate:	16,873 mg/hr
Emission Factor:	36.4 mg/amp-hr
Total Chromium	
Concentration:	0.418 mg/dscm
Concentration:	0.19 ppm
Emission Rate:	15,463 mg/hr
Emission Factor:	33.3 mg/amp-hr

METHOD 1-5 - HEXAVALENT AND TOTAL CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME:	380W5CR3	LAB #:	915755	HEXAVALENT CHROMIUM
CLIENT:	BOEING PLANT #2	START TIME:	15:58 O'CLOCK	HEXAVALENT CHROMIUM CONCENTRATION (ug):
LOCATION:	SEATTLE, WASHINGTON	STOP TIME:	16:59 O'CLOCK	HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm):
SAMPLE SITE:	BLDG 210/TANK #5	SAMPLE TIME:	60.0 MINUTES	HEXAVALENT CHROMIUM IN SAMPLE (ppm):
SAMPLE DATE:	AUGUST 31, 1989	CONDITION:	BATH ON	HEXAVALENT CHROMIUM EMISSION RATE (mg/hr):
RUN #:	3-CR+6			AMPHERE HOURS PER HOUR (amp-hr/hr):
OPERATORS:	GUENTHER/WIDMEYER			HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr):
CONTACT:	P. STEBENALER			

FINAL WT. OF H2O G.	INIT. WT. OF H2O G.	NET WT. OF H2O G.	PITOT Cp:	TOTAL CHROMIUM
641.6	636.9	4.7	0.840	TOTAL CHROMIUM CONCENTRATION (ug):
627.4	624.5	2.9	0.313	TOTAL CHROMIUM CONC. ON FILTER (ug):
1162.8	1161.6	1.2	0.0005	TOTAL CHROMIUM CONCENTRATION (mg/dscm):
812.5	798.6	13.9	48.00	TOTAL CHROMIUM IN SAMPLE (ppm):
TOTAL H2O GAIN:	22.7	"HG:	12.566	TOTAL CHROMIUM EMISSION RATE (mg/hr):
TOTAL VOLUME (SCF)	1.07	"H2O:	79.6	AMPHERE HOURS PER HOUR (amp-hr/hr):
PERCENT MOISTURE:	1.85	STATIC PRES. "H2O:	30.20	TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr):
BWS:	0.0185	STACK PRES. "HG:	30.20	
		ORIFICE PRES "H2O:	1.447	
		METER PRES. "HG:	30.31	

INIT. METER VOL.:	59.658	AVERAGE % CO2:	0.1
FINAL METER VOL.:	117.286	AVERAGE % O2:	20.9
VOLUME SAMPLED:	57.628	AVERAGE % CO:	0
STD VOLUME (DSCF):	56.717	STACK GAS MW. DRY:	28.85
STD VOLUME (DSCM):	1.606	STACK GAS MW. WET:	28.65
Y FACTOR:	0.993		

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NW 1	0.22	83	NE 1	0.24	83
2	0.29	83	2	0.31	84
3	0.31	81	3	0.33	84
4	0.31	82	4	0.31	84
5	0.26	83	5	0.29	84
6	0.22	83	6	0.25	84

PERCENT ISOKINETICS:	102 %
STACK TEMPERATURE:	83.2 DEG. F.
AVERAGE VELOCITY HEAD:	0.277 " OF H2O
STACK GAS VELOCITY:	29.95 FT/SEC.
STACK GAS AIR FLOW:	22581.0 ACF/MIN.
	21745.4 DSCF/MIN.

AT

30

3 B0#5 W 3

915755 TRAVERSE SAMPLING DATA

Page 1 of 1

Client Boeing
 Date 8-31-89
 Sample Location Blade 2-10
TANK 5 Exhaust Stack
 Operators _____
 Sample Box # B
 Run# 3

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm 1.01 / 1.004Leak Test Vac 20 / 19

Pitots, Pretest

Pitots, Posttest

Orsat Sampling System

Tedlar Bag

Thermocouple @ _____ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter _____

Distance Upstream _____

Distance Downstream _____

Filter # 3 tare _____ mgs
 Final Initial Net
 Wt. Wt. Wt.

#1 Bubbler 641.6 - 636.9 = _____#2 Impinger 627.4 - 624.5 = _____#3 Bubbler 626.2 - 626.1 = _____#4 Silica 536.6 - 535.5 = _____Gel 812.5 - 799.6 = _____

TOTAL WATER VOLUME _____

Start Time 1558
 Stop Time 1659
 Barometric 30.20
 Pressure "Hg 767
 Static Pres "H₂O _____
 Production Rate _____
Bath on

NOMOGRAPH SETUP

X Moisture _____

Meter Temp. _____

Stack Temp. _____

ΔH_e .943 Y .593

Pitot# _____ Side# _____

Cp 840Nozzle Diameter .313

K Factor _____

Reference ΔP _____

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O	Gas Meter Temp °F	Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out		
NW1	0	54.658	.27	1.14	1.14	78	71	6.4	65
2	5		.29	1.51	1.51	74	72	9	64
3	10		.31	1.61	1.61	77	72	9.2	62
4	15		.31	1.61	1.61	79	74	9.6	57
5	20		.26	1.35	1.35	84	76	8	59
6	25		.22	1.14	1.14	86	77	7	60
7	30		.24	1.25	1.25	85	77	7.4	60
8	35		.31	1.61	1.61	86	78	9.4	58
9	40		.33	1.72	1.72	88	79	10	57
10	45		.31	1.61	1.61	89	79	9.4	61
11	50		.29	1.51	1.51	88	80	9	61
12	55		.25	1.70	1.70	88	80	8	61
END	60	112.286							

1.447 79.6

1.20

8/31/51. Boon Plan 2

31

Run 3
(Method 6/8)

Parts Dipped into Tank 5 @ 3:58 PM

Parts Removal from Tank @ 4:54 PM

Same Exact Type Used in Run 2 tank
(No tubes: 3X22 @ 23)

3 min 3:59	5.1 V	180 amps
4 min 4:00	17.0 V	750 amps
4 min 4:02	22.5 V	500 amps
2 min 4:06	22.7 V	420 amps
20 4:08	22.8 V	420 amps
4:10	23.0 V	400 amps
4:34	23.0 V	400 amps
6 min 4:37	23.0 V	390 amps
7 min 4:40	23.0 V	400 amps
8 min 4:47		
4:54		

93.5°F Bath

82°F Ambient

$$(3 \times 180 + 4 \times 750 + 2 \times 500 + 20 \times 420 + 13 \times 400 + 7 \times 390) / 55 \times 60 / 55 = 464$$

$$(3 \times 5.1 + 4 \times 17 + 2 \times 22.5 + 2 \times 22.7 + 24 \times 22.8 + 20 \times 23) / 55 = 21.5 \text{ V}$$

Sample Type:	Bath On - Run 5
Sampling Date:	September 19, 1989
Sample Times:	13:00-14:01 (60 minutes)
Run Number:	2
Lab Numbers:	917070, 917071, 917072, 917078
Airflow:	20,402 dscf/min
Process Data	
Bath Amperage:	1853 amp-hr/hr
Voltage:	21.1 volts D.C.
Bath Temperature:	94° F
Ambient Temperature:	80° F
Hexavalent Chromium	
Concentration:	0.947 mg/dscm
Concentration:	0.44 ppm
Emission Rate:	32,818 mg/hr
Emission Factor:	17.7 mg/amp-hr
Total Chromium	
Concentration:	0.916 mg/dscm
Concentration:	0.42 ppm
Emission Rate:	31,747 mg/hr
Emission Factor:	17.1 mg/amp-hr

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 480F5CR2
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: SEPTEMBER 19, 1989
RUN #: 2-CR+6
OPERATORS: GUENTHER/MACKAY
CONTACT: P. SIEBEMALER

LAB #: 917078
START TIME: 13:00 O'CLOCK
STOP TIME: 14:01 O'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

HEXAVALENT CHROMIUM
HEXAVALENT CHROMIUM CONCENTRATION (ug): 1458.5
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.947
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.44
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 32818.1
AMPHERE HOURS PER HOUR (amp-hr/hr): 1853
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr): 17.7

TOTAL CHROMIUM
TOTAL CHROMIUM CONCENTRATION (ug): 1410.6
TOTAL CHROMIUM CONC. ON FILTER (ug): 0.275
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.916
TOTAL CHROMIUM IN SAMPLE (ppm): 0.42
TOTAL CHROMIUM EMISSION RATE (mg/hr): 31747.4
AMPHERE HOURS PER HOUR (amp-hr/hr): 1853
TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr): 17.1

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.
477.8 445.5 32.3
447.4 442.6 4.8
380.4 378.7 1.7
320.4 319.1 1.3
608.3 599.6 8.7
TOTAL H2O GAIN: 48.8
TOTAL VOLUME (SCF) 2.30
PERCENT MOISTURE: 4.06
BUS: 0.0406

INIT. METER VOL.: 550.430
FINAL METER VOL.: 605.566
VOLUME SAMPLED: 55.136
STD VOLUME (DSCF): 54.602
STD VOLUME (DSCM): 1.541
Y FACTOR: 0.993

PITOT CP: 0.84
NOZZLE DIA INCHES: 0.313
NOZZLE AREA FT^2: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT^2: 12.566
METER TEMP. DEG F: 78.8
BAROM. PRES. "HG: 30.24
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.24
ORIFICE PRES "H2O: 1.336
METER PRES. "HG: 30.34

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.41

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NE 1	0.20	82	NW 1	0.22	84
2	0.25	82	2	0.26	84
3	0.31	83	3	0.29	84
4	0.28	83	4	0.29	84
5	0.25	83	5	0.28	84
6	0.20	84	6	0.22	84
PERCENT ISOKINETICS: 105 X					
STACK TEMPERATURE: 83.4 DEG. F.			543.4 DEG. R.		
AVERAGE VELOCITY HEAD: 0.253 " OF H2O			28.72 FT/SEC.		
STACK GAS VELOCITY: 21654.3 ACF/MIN.			20402.1 DSCF/MIN.		



480#5Cr2917078 TRAVERSE SAMPLING DATA

Client P... (Plant #2)
 Date 7.19.81
 Sample Location Side 1
#5 Chrome Tank Exhaust
 Operators SMO - RSM
 Sample Box # Blue 2
 Run# 2.12

EQUIPMENT CHECKS

Initial/Final
 Leak Rate Cfm 1.05 / 1.02
 Leak Test Vac 15 / 15
☒ Pitots, Pretest
☒ Pitots, Posttest
☒ Orsat Sampling System
☒ Tedlar Bag
 Thermocouple @ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter
 Distance Upstream
 Distance Downstream

Filter # tare mgs
 Final Initial Net
 Wt. Wt. Wt.
N/A
 #1 Bubbler 447.8 - 445.5 -
N/A
 #2 Impinger 447.4 - 442.6 -
HN03
 #3 Bubbler 380.4 - 378.7 -
MI
 #4 Silica 320.4 - 319.1 -
Gel 608.3 - 599.6 -

TOTAL WATER VOLUME

Start Time 1300
 Stop Time 1401
 Barometric 30.24"
 Pressure "Hg 46.8mm
 Static Pres "H₂O 0.02
 Production Rate

NOMOGRAPH SETUP

X Moisture
 Meter Temp.
 Stack Temp.
 ΔHe 943 y 993
 Pitot# Side#
 Cp 0.84
 Nozzle Diameter 0.312
 K Factor
 Reference ΔP

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F	
				Ideal	Actual	In	Out					
N 1	0	550.430	.20	1.04	1.04	73	44	4	N/A	63	82	
2	5		.25	1.30	1.30	76	74	5		58	82	
3	10		.31	1.62	1.62	80	75	6		57	83	
4	15		.28	1.46	1.46	83	76	5 1/2		57	83	
5	20		.25	1.30	1.30	84	76	5		61	83	
6	25		.20	1.15	1.15	84	77	4 1/2		57	84	
W 1	30		.22	1.15	1.15	83	78	4 1/2		56	84	
2	35		.26	1.36	1.36	86	76	5		56	84	
3	40		.29	1.52	1.52	86	79	5 1/2		56	84	
4	45		.29	1.52	1.52	87	79	5 1/2		56	84	
5	50		.28	1.46	1.46	88	80	5 1/2		58	84	
6	55		.22	1.15	1.15	88	80	4 1/2		57	84	
END	60	555.66										

1.336 79.8

Boeing Plans Z

9/19/89

Chronic Acid Bath Tank #5 (Amperage)

Ambient Temp = 22.1

Pours into tank @ 12:59:30

Bath Temp 22.5

Once 1:55 PM

(55 minute Amperage Cycle)

Time	Voltage	Amperage	Time	Voltage	Amperage
12:58	0.3	0	1:16	22.4	1800
12:59	0.3	0	1:17	22.9	1800
1:00 PM	1.6	150	1:18	22.9	1800
1:01	6.8	1420	1:19	22.9	1810
1:02	11.8	1820	1:20	22.9	1810
1:03	17.0	1820	1:21	23.0	1810
1:04	21.3	1820	1:22	23.0	1810
1:05	22.4	1800	1:23	23.0	1810
1:06	22.5	1760	1:24	23.0	1810
1:07	22.6	1750	1:25	23.0	1820
1:08	22.7	1740	1:26	23.0	1810
1:09	22.7	1740	1:27	23.0	1810
1:10	22.7	1740	1:28	23.1	1810
1:11	22.7	1740	1:29	23.1	1820
1:12	22.8	1760	1:30	23.1	1810
1:13	22.8	1800	↓	↓	↓
1:14	22.8	1800	1:52	23.1	1810
1:15	22.8	1800			

22

1853 amp-hr / hr
21.1 volts

Sample Type:	Bath On - Run 6
Sampling Date:	November 30, 1989
Sample Times:	08:03-09:04 (60 minutes)
Run Number:	1
Lab Numbers:	922093, 922094, 922095
Airflow:	22,740 dscf/min
Process Data	
Bath Amperage:	1628 amp-hr/hr
Voltage:	21.6 volts D.C.
Bath Temperature:	94° F
Ambient Temperature:	76° F
Hexavalent Chromium	
Concentration:	0.261 mg/dscm
Concentration:	0.12 ppm
Emission Rate:	10,077 mg/hr
Emission Factor:	6.19 mg/amp-hr
Total Chromium	
Concentration:	0.280 mg/dscm
Concentration:	0.13 ppm
Emission Rate:	10,813 mg/hr
Emission Factor:	6.64 mg/amp-hr

Note: Acid in bath was drained and replaced prior to this test.

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 5BOW5CR1
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: NOVEMBER 30, 1989
RUN #: 1-CR-6
OPERATORS: S. WACKAY
CONTACT: P. STEBENALER

LAB #:
START TIME: 8:03 O'CLOCK
STOP TIME: 9:04 O'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

922094

HEXAVALENT CHROMIUM

HEXAVALENT CHROMIUM CONCENTRATION (ug): 438
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.261
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.121
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 10077.1
AMPHERE HOURS PER HOUR (amp-hr/hr): 1628
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr): 6.19

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

1225.8 1245.5 -19.7
558.2 565.3 -7.1
515.4 514.0 1.4
756.9 749.7 7.2

TOTAL H2O GAIN: -18.2
TOTAL VOLUME (SCF) -0.86
PERCENT MOISTURE: -1.47
BWS: -0.0147

PITOT Cp: 0.842
NOZZLE DIA INCHES: 0.313
NOZZLE AREA FT²: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT²: 12.566
METER TEMP. DEG F: 74.5
BAROM. PRES. "HG: 30.40
STATIC PRES. "H2O: -0.02
STACK PRES. "HG: 30.40
ORIFICE PRES "H2O: 3.078
METER PRES. "HG: 30.63

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 29.01

TOTAL CHROMIUM
TOTAL CHROMIUM CONCENTRATION (ug): 470
TOTAL CHROMIUM CONC. ON FILTER (ug): < 0.15
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.280
TOTAL CHROMIUM IN SAMPLE (ppm): 0.129
TOTAL CHROMIUM EMISSION RATE (mg/hr): 10813.3
AMPHERE HOURS PER HOUR (amp-hr/hr): 1628
TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr): 6.64

INIT. METER VOL.: 843.639
FINAL METER VOL.: 901.593
VOLUME SAMPLED: 57.954
STD VOLUME (DSCF): 59.304
STD VOLUME (DSCM): 1.680
Y FACTOR: 1.012

SAMPLE VELOCITY TEMPERATURE
POINT " OF H2O DEGREES F.

NW	1	2	3	4	5	6
VELOCITY	0.20	0.28	0.31	0.31	0.30	0.25
TEMPERATURE	72	73	73	73	73	73
POINT	NE 1	2	3	4	5	6
" OF H2O	0.18	0.30	0.35	0.31	0.30	0.28
TEMPERATURE	73	74	74	74	74	74

PERCENT ISOXINETICS:

102 %

STACK TEMPERATURE:

73.3 DEG. F.

AVERAGE VELOCITY HEAD:

0.279 " OF H2O

STACK GAS VELOCITY:

29.55 FT/SEC.

STACK GAS AIR FLOW:

22281.0 ACF/MIN.

22740.0 DSCF/MIN.

533.3 DEG. R.



922094 TRAVERSE SAMPLING DATA

Page 1 of

Client <u> </u> Date <u> </u> Sample Location <u> </u> Operators <u> </u> Sample Box # <u> </u> Run# <u>1 Cr# / Tcr</u>	SCHEMATIC TRAVERSE LAYOUT Stack Diameter <u>48"</u> Distance Upstream <u> </u> Distance Downstream <u> </u> Filter # <u> </u> tare <u> </u> mgs Final Initial Net Wt. Wt. Wt. <u>2-Nx04</u> #1 Bubbler <u>1225.8 - 1245.5 -</u> <u>1-HNO3</u> #2 Impinger <u>557.2 - 565.3 -</u> #3 Bubbler <u>5154 - 514.0 -</u> #4 Silica Gel <u>756.9 - 749.7 -</u> TOTAL WATER VOLUME <u> </u>	Start Time <u> </u> Stop Time <u> </u> Barometric Pressure "Hg <u> </u> Static Pres "H2O <u> </u> Production Rate <u> </u> NOMOGRAPH SETUP X Moisture <u> </u> Meter Temp. <u> </u> Stack Temp. <u> </u> ΔH@ <u>1.0</u> Y <u>1.0</u> Pitot# <u> </u> Side# <u> </u> Cp <u> </u> Nozzle Diameter <u> </u> K Factor <u> </u> Reference ΔP <u> </u>
EQUIPMENT CHECKS Initial/Final Leak Rate Cfm <u> </u> Leak Test Vac <u> </u> Pitots, Pretest <u> </u> Pitots, Postest <u> </u> Orsat Sampling System <u> </u> Tedlar Bag <u> </u> Thermocouple @ <u> </u> °F		

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H2O	Orifice Setting (ΔH), In H2O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
1	0	843.629	.20	2.12	2.12	51	47	11	111	36	72
2	5		.29	3.22	3.22	76	47	12		38	73
3	10		.31	3.22	3.22	85	42	12		37	72
4	15		.31	3.22	3.22	95	42	12		41	73
5	20		.20	2.74	2.74	100	52	11		41	73
6	25		.18	1.97	1.97	100	51	10		41	73
7	30		.30	3.29	3.29	109	56	12		42	74
8	35		.35	3.80	3.80	109	57	13		43	74
9	40		.31	3.40	3.40	111	60	12		42	74
10	45		.29	3.29	3.29	113	60	12		42	74
11	50		.29	3.07	3.07	116	60	12		41	73
12	55	901.593									

3078 74.5

✓

Prime version - 6-25-1944
Arthur Holm

39

11/30/89

Chronic Acid Bath 5

Bath Temp. - 94°

Parts descrip. - Clips, fittings, filters, Angles, ch...

Bath ON

Run 1

	Time	Volts	Amperes	(Fpm) Velocity	(Velometer)
ants in -	0806	23.0	100 1 min	Pt. 5	2400
	0807	21.6	500 1	10	1790
	08	21.0	950 1	15	1690
	0809	13.0	1000 1	20	1560
	0810	17.5	1800 3	25	960
	11	20.0	1500	30	1010
	12	22.0	1800	35	970
	13	22.0	1500 2	40	750
	14	22.5	1500	45	950
	10 8 15	22.5	1550 1	50	460
	16	22.5	1500 9	55	260
	17	22.5	1500	60	600
	18				
	19	22.5	1500		
	08 20	22.5	1500		
	25	22.5	1550 1/2		
	08 30	22.5	1550		
	35	22.6	1550		
	40	22.6	1550		
	45				
	50	22.6	1575 11		
	55	22.9	1575		
	09 00	22.9	1575		
Bath OFF	0901			Amb. Temp. 76°	
				1100 - 1105	
ants out	0902	21.6 v	1628 amp-hr/hr	(over time)	

Sample Type:	Bath On - Run 7
Sampling Date:	November 30, 1989
Sample Times:	15:46-16:49 (60 minutes)
Run Number:	4
Lab Numbers:	922106, 922107, 922108
Airflow:	22,084 dscf/min
Process Data	
Bath Amperage:	1896 amp-hr/hr
Voltage:	21.2 volts D.C.
Bath Temperature:	94.5° F
Ambient Temperature:	77° F
Hexavalent Chromium	
Concentration:	0.416 mg/dscm
Concentration:	0.19 ppm
Emission Rate:	15,601 mg/hr
Emission Factor:	8.23 mg/amp-hr
Total Chromium	
Concentration:	0.423 mg/dscm
Concentration:	0.20 ppm
Emission Rate:	15,853 mg/hr
Emission Factor:	8.36 mg/amp-hr

Note: Acid in bath was drained and replaced prior to this test.

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 580W5CR4
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: NOVEMBER 30, 1989
RUN #: 4-CR-6
OPERATORS: S. MACKEY
CONTACT: P. STEBEMALER

LAB #: 922107
START TIME: 15:46 0'CLOCK
STOP TIME: 16:49 0'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

HEXAVALENT CHROMIUM
HEXAVALENT CHROMIUM CONCENTRATION (ug): 679
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.416
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.193
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 15600.6
AMPHERE HOURS PER HOUR (amp-hr/hr): 1896
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr): 8.23

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

1255.5 1254.3 1.2
570.1 568.3 1.8
518.0 517.6 0.4
771.7 762.6 9.1
TOTAL H2O GAIN: 12.5
TOTAL VOLUME (SCF) 0.59
PERCENT MOISTURE: 1.01
BWS: 0.0101

PITOT CP: 0.842
NOZZLE DIA INCHES: 0.313
NOZZLE AREA FT^2: 0.0005
STACK DIA. INCHES: 48.000
STACK AREA FT^2: 12.566
METER TEMP. DEG F: 91.8
BAROM. PRES. "HG: 30.48
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.48
ORIFICE PRES "H2O: 2.977
METER PRES. "HG: 30.70

TOTAL CHROMIUM
TOTAL CHROMIUM CONCENTRATION (ug): 690
TOTAL CHROMIUM CONC. ON FILTER (ug): < 0.15
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.423
TOTAL CHROMIUM IN SAMPLE (ppm): 0.196
TOTAL CHROMIUM EMISSION RATE (mg/hr): 15853.3
AMPHERE HOURS PER HOUR (amp-hr/hr): 1896
TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr): 8.36

INIT. METER VOL.: 946.900
FINAL METER VOL.: 1004.859
VOLUME SAMPLED: 57.959
STD VOLUME (DSCF): 57.586
STD VOLUME (DSCM): 1.631
Y FACTOR: 1.012

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MW. DRY: 28.85
STACK GAS MW. WET: 28.74

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NE 1	0.21	79	NW 1	0.22	78
2	0.30	79	2	0.27	78
3	0.33	80	3	0.30	78
4	0.30	79	4	0.30	78
5	0.29	79	5	0.29	78
6	0.26	78	6	0.24	77

PERCENT ISOKINETICS: 102 %
STACK TEMPERATURE: 78.4 DEG. F.
AVERAGE VELOCITY HEAD: 0.275 " OF H2O
STACK GAS VELOCITY: 29.58 FT/SEC.
STACK GAS AIR FLOW: 22051.3 DSCF/MIN.



TRAVERSE SAMPLING DATA

Page 1 of 1

422107

Client Corning
 Date 11/30/89
 Sample Location Test 2 Vent Stack
Exit Point 2 7-19
 Operators MSH
 Sample Box # C
 Run# 4 - Cr¹⁶ / Tail Chimney

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm 0.15 / 0.14Leak Test Vac 15" / 15"☒ Pitots, Pretest☒ Pitots, Posttest☒ Orsat Sampling System☒ Tedlar Bag☒ Thermocouple @ _____ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter 48" d
 Distance Upstream _____
 Distance Downstream _____

Filter # 7 tare _____ mgs
 Final Initial Net
 Wt. Wt. Wt.

#1 Bubblers 1255.5 - 1254.3 = _____#2 Impinger 570.1 - 568.3 = _____#3 Bubblers 518.0 - 517.6 = _____#4 Silica Gel 771.7 - 762.6 = _____

TOTAL WATER VOLUME _____

Start Time 15:46
 Stop Time 16:49
 Barometric Pressure "Hg 30.45
 Static Pres "H₂O _____
 Production Rate 8.0

NOMOGRAPH SETUP

X Moisture 2Meter Temp. 93Stack Temp. 78ΔH 1.92 y 10.0Pitot # 4 Side # ACp .842Nozzle Diameter .313K Factor 10.2Reference ΔP N/A

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
1	0	946.900	.21	2.27	2.27	79	60	10	N/A	43	79
2	5		.30	3.24	3.24	99	61	12		42	79
3	10		.33	3.56	3.56	108	62	12		42	80
4	15		.30	3.24	3.24	113	64	12		42	79
5	20		.29	3.13	3.13	117	70	12		41	79
6	25		.26	2.80	2.80	120	72	11		43	78
7	30		.22	2.37	2.37	113	72	10		46	78
8	35		.27	2.91	2.91	120	73	11		47	78
9	40		.30	3.24	3.24	121	73	11		45	78
10	45		.30	3.24	3.24	122	74	12		44	78
11	50		.29	3.13	3.13	123	75	12		46	78
12	55		.24	2.59	2.59	122	77	11		48	77
13	60	1004.857									

2.977 91.8

11/30/89
Chromic Acid Bath 5

R4-MS Bath Temp. 94.5

Ambient Temp. 77°

Parts descrip. -

	Time	Volts	Amperes F-O-	Amperes
Parts In	15 54	0.5		
Parts In	15 53	2.1	100	
Bath on	15 56	2.1	100	
	15 57	5.0	100	
	15 58	8.5	1800	53
	15 59	13.5	1800	
	16 00	17.7	1800	
	16 01	21.5	1800	
	16 02	22.0	1800	
	16 03	22.0	1800	
	16 04	22.0		
	16 05			
	16 06			
	16 07			
	16 08			
	16 09			
	16 10			
	16 15	22.5	1800	
Parts In	16 18			
Parts out	16 20			
	16 25			
	16 30			
	16 40			
	16 50	22.5	1800	
Bath OFF	16 57			
Parts out	16 53	21.2 V		1896 amp-hr/hr

Sample Type:	Bath On - Run 8
Sampling Date:	December 1, 1989
Sample Times:	07:29-08:31 (60 minutes)
Run Number:	5
Lab Numbers:	922114, 922115, 922116
Airflow:	22,318 dscf/min
Process Data	
Bath Amperage:	1680 amp-hr/hr
Voltage:	20.8 volts D.C.
Bath Temperature:	93.5° F
Ambient Temperature:	70° F
Hexavalent Chromium	
Concentration:	0.263 mg/dscm
Concentration:	0.12 ppm
Emission Rate:	9,982 mg/hr
Emission Factor:	5.94 mg/amp-hr
Total Chromium	
Concentration:	0.273 mg/dscm
Concentration:	0.13 ppm
Emission Rate:	10,371 mg/hr
Emission Factor:	6.17 mg/amp-hr

Note: Acid in bath was drained and replaced prior to this test.

METHOD 1-5 - HEXAVALENT CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME: 58045CR5
CLIENT: BOEING PLANT #2
LOCATION: SEATTLE, WASHINGTON
SAMPLE SITE: BLDG 210/TANK #5
SAMPLE DATE: DECEMBER 1, 1989
RUN #: 5-CR+6
OPERATORS: S. MACKEY
CONTACT: P. STEBEMALER

LAB #: 922114
START TIME: 07:29 O'CLOCK
STOP TIME: 08:31 O'CLOCK
SAMPLE TIME: 60.0 MINUTES
CONDITION: BATH ON

HEXAVALENT CHROMIUM

HEXAVALENT CHROMIUM CONCENTRATION (ug): 442.00
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm): 0.263
HEXAVALENT CHROMIUM IN SAMPLE (ppm): 0.122
HEXAVALENT CHROMIUM EMISSION RATE (mg/hr): 9982.0
AMPHERE HOURS PER HOUR (amp-hr/hr): 1680
HEX. CHROMIUM EMISSION FACTOR (mg/amp-hr): 5.94

FINAL WT. INIT. WT. NET WT.
OF H2O G. OF H2O G. OF H2O G.

1176.0 1175.6 0.4
628.0 627.2 0.8
479.2 478.5 0.7
801.9 791.0 10.9
TOTAL H2O GAIN: 12.8
TOTAL VOLUME (SCF) 0.60
PERCENT MOISTURE: 1.01
BWS: 0.0101

PITOT Cp: 0.842
NOZZLE DIA. INCHES: 0.313
NOZZLE AREA FT^2: 0.0005
STACK DIA. INCHES: 48.00
STACK AREA FT^2: 12.566
METER TEMP. DEG F: 72.5
BAROM. PRES. "HG: 30.50
STATIC PRES. "H2O: 0.00
STACK PRES. "HG: 30.50
ORIFICE PRES "H2O: 3.045
METER PRES. "HG: 30.72

TOTAL CHROMIUM

TOTAL CHROMIUM CONCENTRATION (ug): 459.2
TOTAL CHROMIUM CONC. ON FILTER (ug): < 0.15
TOTAL CHROMIUM CONCENTRATION (mg/dscm): 0.273
TOTAL CHROMIUM IN SAMPLE (ppm): 0.127
TOTAL CHROMIUM EMISSION RATE (mg/hr): 10370.5
AMPHERE HOURS PER HOUR (amp-hr/hr): 1680
TOTAL CHROMIUM EMISSION FACTOR (mg/amp-hr): 6.17

INIT. METER VOL.: 5.482
FINAL METER VOL.: 63.027
VOLUME SAMPLED: 57.545
STD VOLUME (DSCF): 59.295
STD VOLUME (DSCM): 1.679
Y FACTOR: 1.012

AVERAGE % CO2: 0.1
AVERAGE % O2: 20.9
AVERAGE % CO: 0
STACK GAS MJ. DRY: 28.85
STACK GAS MJ. WET: 28.74

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE DEGREES F.
NW 1	0.20	71	WE 1	0.22	71
2	0.27	72	2	0.30	72
3	0.30	73	3	0.34	72
4	0.31	72	4	0.30	72
5	0.30	72	5	0.30	72
6	0.24	71	6	0.27	72

PERCENT ISO KINETICS: 104 %
STACK TEMPERATURE: 71.8 DEG. F. 531.8 DEG. R.
AVERAGE VELOCITY HEAD: 0.278 " OF H2O
STACK GAS VELOCITY: 29.55 FT/SEC.
STACK GAS AIR FLOW: 22277.4 ACF/MIN. 22318.3 DSCF/MIN.



922114 TRAVERSE SAMPLING DATA

[illegible]

3.045 72.5

12-1-89

Chromic Acid Bath #5

Bath Temp. 93.5°F Ambient - 70°F

Parts Descrip. - supports, stiffeners

Run 5

Wts In -
~~Wts Out~~
 Wts ON →

Time

Volts

Amperes

0730

0.6

0

0731

0.4

0

0732

3.0

100

0733

6.0

100

0734

11.5

800

0735

15.5

1800

0736

21.8

1800

0737

21.8

1600

0738

22.0

1600

0739

22.0

1600

0740

22.0

1600

0741

22.2

1600

0742

22.5

1600

0743

22.6

1600

0744

22.6

1600

0745

22.6

1600

0750

~~0751~~

0753

22.7

1600

0800

0805

22.6

1600

0810

0820

0825

22.6

1650

2

0827

1.0

0

0828

0.8

0

Bath off

Wts out

Set B 11760
 Run 5 → 6280
 final 479.2
 8019

$$(2 \times 100 + 800 + 48 \times 1600 + 2 \times 1650)$$

20.8 V.

1680 amp-hr/hr

SECTION 3
PARTICLE SIZE DISTRIBUTION RUNS
BATH ON

The results of the particle size distribution tests are presented graphically in Figures 1 and 2. Figure 1 is a plot of the distribution based on chemical analysis for total chromium of each filter which lined each stage of the Cascade impactor. Figure 2 is a plot of the distribution based on the gravimetric weight gain on each filter prior to chemical analysis. The detailed results are presented on computer printouts titled "Pilat (U of W) Mark 5 Particle Size Analysis Results" and computer printouts titled "Particle Size Analysis" which are included for each run. The column titled "aerodynamic diameter" (in micrometers) was plotted against the column titled "percent less than" on semi-log versus probability graph paper to illustrate the particle size distributions.

The aerodynamic mass mean diameter (MMD) value represents the size (in micrometers - microns) at which fifty percent of the particulate matter collected was composed of particles of less than that value. The average MMD based on chemical analysis of the progressive size ranges was 4.9 microns. The average MMD based on gravimetric (weight) analysis of the cut sizes was 3.0 microns.

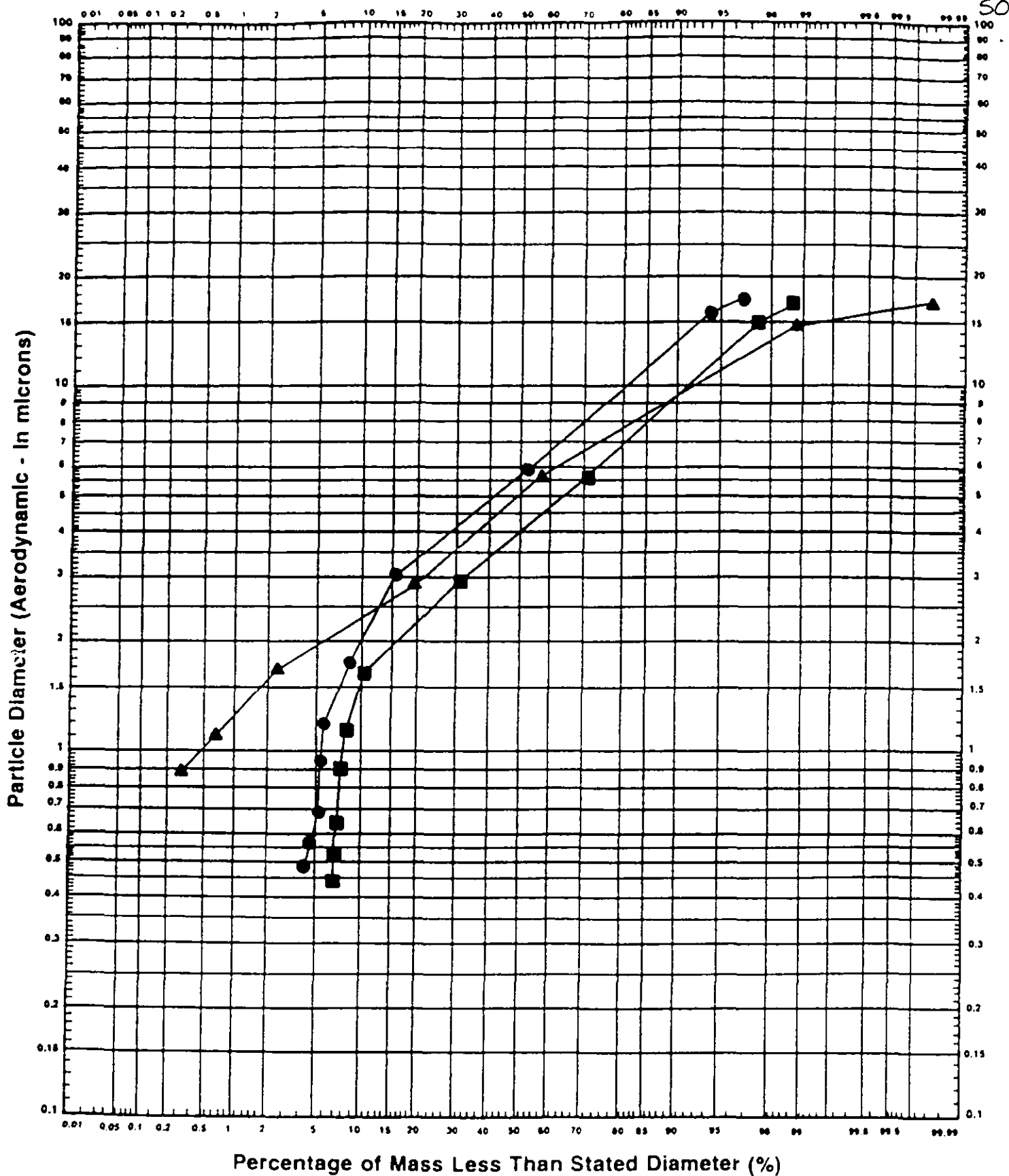


Figure 1.

Plot of particle size versus chromium concentration at the #5A chromic acid anodizing tank at Boeing Plant 2 in Seattle, Washington.



Symbol	Run #	MMD microns
●	1	5.7
■	2	4.0
▲	3	5.1
	Average	4.9

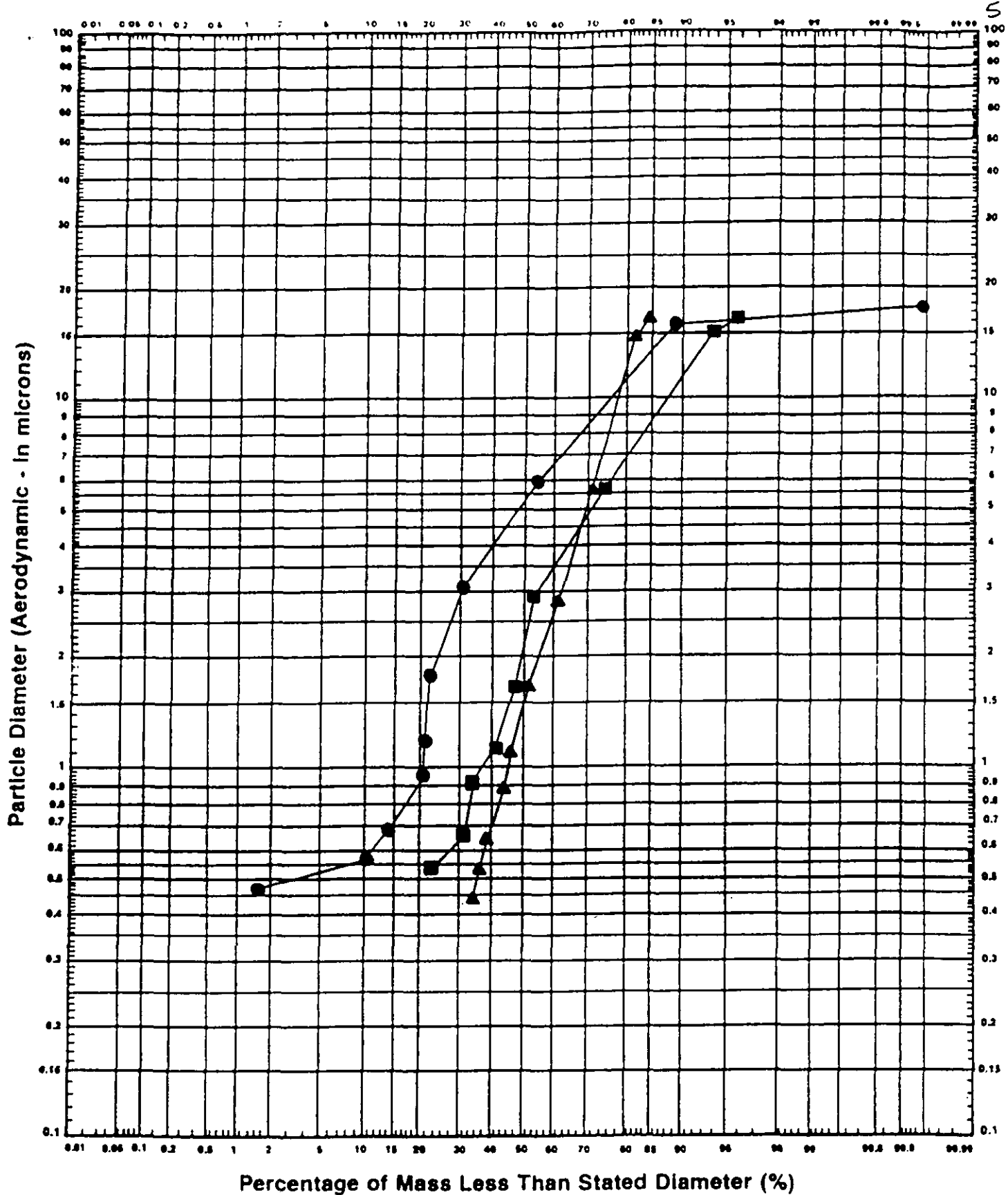


Figure 2.

Plot of particle size versus particle mass (weight) at the #5A chromic acid anodizing tank at Boeing Plant 2 in Seattle, Washington.



Symbol	Run #	MMD microns
●	1	5.3
■	2	2.2
▲	3	1.5
	Average	3.0

Sample Type:		Particle Size Distribution Bath On - Run 1	
Sampling Date:		August 31, 1989	
Sample Times:		10:26-11:26 (60 minutes)	
Run Number:		1	
Lab Numbers:		915962-915976, 915744-915749	
Airflow:		21,696 dscf/min	
Process Data			
Bath Amperage:		1892 amp-hr/hr	
Voltage:		22.4 volts D.C.	
Bath Temperature:		94.5° F	
Ambient Temperature:		78° F	
Stage #	Distribution d50's (microns)	Cumulative Cr % Less Than Stated Diameter	Chromium Concentration (mg/dscm)
1	17.54	96.85	0.009
2	15.87	94.54	0.007
3	5.90	52.90	0.117
4	3.07	15.90	0.104
5	1.76	7.93	0.022
6	1.18	5.74	0.006
7	0.95	5.29	0.001
8	0.68	5.01	0.001
9	0.56	4.72	0.001
10	0.47	4.45	0.001
final	<0.47	0.00	0.013
TOTAL			0.280

PILAT (U of W) MARK 5 PARTICLE SIZE ANALYSIS RESULTS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME	A:BOEMKSR1	Y FACTOR	0.993		AVERAGE % CO2	0.1	Percent
CLIENT	BOEING AEROSPACE	VOLUME H2O	N/A	mls	AVERAGE % O2	20.9	Percent
LOCATION	SEATTLE, WASHINGTON	VOLUME H2O	N/A	SCF	AVERAGE CO	0.0	Percent
SAMPLE SITE	PLANT NO. 2, BLDG. 2-10	PERCENT MOISTURE	0.81	%	STACK GAS MW DRY	28.85	gm/mole
SAMPLE DATE	CHROMIC TANK #5 EXHAUST	METER TEMP	67.5	Deg. F.	STACK GAS MW WET	28.76	gm/mole
OPERATOR	AUGUST 31, 1989	MARK 5 TEMP	80.2	Deg. F.	PITOT CP	0.84	
	GUENTHER/WIDMEYER	BAROMETRIC PRESSURE	30.16	" of Hg	NOZZLE DIAMETER	0.188	inches
RUN #	1-MK5/Cr	STATIC PRESSURE	30.16	" of Hg	STACK AREA	0.0002	sq. ft.
LAB #	915744	ORIFICE PRESSURE	0.17	" of Hg	STACK DIAMETER	48.00	inches
START TIME	10:26	MKS OUTLET ABS. PRES	29.16	" of Hg	VELOCITY HEAD	12.566	sq. ft.
STOP TIME	11:26	MKS DELTA P	1.00	" of Hg	STACK GAS VELOCITY	0.27	" H2O
SAMPLE TIME	60.00	AVG. MARK 5 PRESSURE	30.06	" of Hg	STACK GAS VELOCITY	1766.7	ft/sec
INITIAL METER VOL	982.904	SAMPLE RATE	0.32	cf/min	STACK GAS AIRFLOW	22201.2	scf/min
FINAL METER VOL	1001.815	SAMPLE RATE (STD)	0.32	dscf/min	STACK GAS AIRFLOW	21695.7	dscf/min
VOLUME SAMPLED	18.911	SAMPLE RATE (MK5)	0.32	liter/min	% ISO KINETICS	95	%
VOLUME (STACK COND)	19.40	ACTUAL CU FT	8.95				
STD VOLUME	18.955	DSCF					
STD VOLUME	0.537	DSCM					

CONFIGURATION: 1,6,12,90,110,110,105,105,78,FINAL FILTER

SET ID # MK-QUARTZ B-TIRI-TANK 5,OUT

STAGE #	TARE WT. Gms.	FINAL WT. Gms.	NET WT. Gms.	FRACTIONAL %	PERCENT < than	AERO DYN. DIAMETER μm	PARTIC. CONC. Gr/DSCF	PARTIC. CONC. @ STACK CONDITIONS Gr/ACF	PARTIC. CONC. Mg/DSCM
NOZZLE:	0.00000	0.00000	0.00000						
1	0.85776	0.85777	0.00001	0.00	99.44	17.54	8.1E-06	8.0E-06	0.02
2	1.28918	1.28940	0.00022	0.22	87.08	15.87	1.8E-04	1.8E-04	0.41
3	1.29306	1.29363	0.00057	0.57	55.06	5.90	4.6E-04	4.5E-04	1.06
4	1.30218	1.30261	0.00043	0.43	30.90	3.07	3.5E-04	3.4E-04	0.80
5	1.29894	1.29908	0.00014	0.14	23.03	1.76	1.1E-04	1.1E-04	0.26
6	1.29425	1.29428	0.00003	0.03	21.35	1.18	2.4E-05	2.4E-05	0.06
7	1.29866	1.29867	0.00001	0.01	20.79	0.95	8.1E-06	8.0E-06	0.02
8	1.26340	1.26352	0.00012	0.12	14.04	0.68	9.8E-05	9.5E-05	0.22
9	1.16257	1.16263	0.00006	0.06	10.67	0.56	4.9E-05	4.8E-05	0.11
10	1.17138	1.17154	0.00016	0.16	1.69	0.47	1.3E-04	1.3E-04	0.30
FILTER	0.35832	0.35835	0.00003	0.03	0.00	<.47	2.4E-05	2.4E-05	0.06
TOTAL:			1.78	100.00			0.0014	0.0014	3.31

BLANK #1: 1.30168 1.30168 0.00000 PH10: 74.6 % < 10 Microns
BLANK #2: 1.28174 1.28176 0.00002 MHD: 3.08 Microns @ 50%
AVERAGE BLANK: 0.00001

P A R T I C L E S I Z E A N A L Y S I S

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INPUT DATA

DATE OF TEST AUGUST 31, 1989
TIME OF TEST 10:26-11:26
LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
TEST NUMBER 1
PART. DIAM. TYPE C (I=IMP AERO, C=CLAS AERO, P=PHYSICAL)
TEST TYPE OUTLET
RUN NUMBER 1-TANK 5 - FILE NAME: B:T1R1-TANK 5.OUT
REMARKS:
IMPACTOR TYPE MK5-78
 MARK 5 IMPACTOR WITH A 78 HOLE FINAL STAGE

WATER VAPOR 0.81%
CARBON DIOXIDE 0.10%
CARBON MONOXIDE 0.00%
OXYGEN 20.90%
SULFUR DIOXIDE 0.00%
NITROGEN 79.00%
PARTICLE DENSITY 1.00 GRAMS/CM^3

GAS METER VOLUME 18.911 CUBIC FEET
IMPACTOR DELTA P 1.00 INCHES HG
ORIFICE DELTA P 0.17 INCHES H2O
STACK PRESS. (BELOW ATMOS.) 0.00 INCHES H2O
BAROMETRIC PRESS. 30.16 INCHES HG
STACK TEMPERATURE 80.2 DEGREES F
METER TEMPERATURE 67.5 DEGREES F
IMPACTOR TEMPERATURE 80.2 DEGREES F
SAMPLE TIME 60.00 MINUTES
AV. VELOCITY OF STACK GAS 1766.70 FEET/MINUTE
GAS METER PRESSURE 0.00 IN HG
NOZZLE DIAMETER 0.19 INCHES
MAXIMUM AERODYN. DIAMETER 60.00 MICRONS

MASS GAIN ON STAGE 1 0.01 MG
MASS GAIN ON STAGE 2 0.22 MG
MASS GAIN ON STAGE 3 0.57 MG
MASS GAIN ON STAGE 4 0.43 MG
MASS GAIN ON STAGE 5 0.14 MG
MASS GAIN ON STAGE 6 0.03 MG
MASS GAIN ON STAGE 7 0.01 MG
MASS GAIN ON STAGE 8 0.12 MG
MASS GAIN ON STAGE 9 0.06 MG
MASS GAIN ON STAGE 10 0.16 MG
MASS GAIN ON FINAL FILTER 0.03 MG

RESULTS

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DATE OF TEST AUGUST 31, 1989
 TIME OF TEST 10:26-11:26
 LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
 TEST NUMBER 1 -
 RUN NUMBER 1-TANK 5
 ACTUAL FLOW RATE (STACK CONDITIONS) 0.325 CFM
 FLOW RATE (STANDARD CONDITIONS) 0.318 CFM
 PERCENT ISOKINETIC SAMPLING 95.55 %

VISCOSITY OF GAS STREAM		0.0001830 GRAMS/CM-SEC		
STAGE	CCF	DP(CLAS AERO)	DP(IMP AERO)	CUM FRACTION
1	1.009	17.538	17.621	0.9944
2	1.010	15.865	15.948	0.8708
3	1.028	5.896	5.978	0.5506
4	1.054	3.074	3.155	0.3090
5	1.094	1.758	1.839	0.2303
6	1.140	1.179	1.259	0.2135
7	1.175	0.952	1.031	0.2079
8	1.248	0.676	0.755	0.1404
9	1.306	0.555	0.634	0.1067
10	1.372	0.467	0.547	0.0169

TOTAL MASS PER DRY NORMAL CUBIC METER 3.2958 MG/CUBIC METER

PART. DIAM.	TYPE	C	(I=IMP AERO, C=CLAS AERO, P=PHYSICAL)		
ARTICLE SIZE	CUMFR		CUMFR	CUM.MASS	dM/dLOG D
(MICRONS)	(STD. DEV.)		(PER CENT)	(MG/DRY N.CU.METER)	
0.20	-9.8170		0.0	0.000	0.000
0.25	-7.7933		0.0	0.000	0.000
0.40	-3.5306		0.0	0.001	0.054
0.50	-1.7166		4.3	0.142	3.926
0.75	-1.0210		15.4	0.506	1.489
1.00	-0.7962		21.3	0.702	0.488
1.50	-0.7653		22.2	0.732	0.444
2.00	-0.7254		23.4	0.772	0.213
2.50	-0.6733		25.0	0.825	1.127
4.00	-0.0619		47.5	1.567	5.061
5.00	0.1987		57.9	1.908	1.252
7.50	0.3707		64.5	2.124	2.865
10.00	0.6623		74.6	2.459	2.464
15.00	1.0733		85.8	2.829	1.725
20.00	4.9321		100.0	3.296	0.000
25.00	9.0597		100.0	3.296	0.000
40.00	19.2740		100.0	3.296	0.000
50.00	29.4350		100.0	3.296	0.000

TIME OF TEST 10:26-11:26
 LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
 TEST NUMBER 1
 RUN NUMBER 1-TANK 5

*** INHALABLE PARTICULATE MATTER ***

AM. MASS LESS THAN 2.5 MICRON:	0.83 MG/DRY NORMAL CU.METER (25.0 %)
AM. MASS LESS THAN 5.0 MICRON:	1.91 MG/DRY NORMAL CU.METER (57.9 %)
CUM. MASS LESS THAN 10.0 MICRON:	2.46 MG/DRY NORMAL CU.METER (74.6 %)
AM. MASS LESS THAN 15.0 MICRON:	2.83 MG/DRY NORMAL CU.METER (85.8 %)

NOTE: DIAMETERS FOR INHALABLE PARTICULATE ARE CLASSICAL AERODYNAMIC DIAMETERS.

LOG-NORMAL SIZE DISTRIBUTION PARAMETERS

LEAST SQUARES LINE :	Y =	-1.04 +	2.13X
MASS GEOMETRIC MEAN DIAMETER		3.077	
GEOMETRIC STANDARD DEVIATION		2.942	
CORRELATION COEFFICIENT		0.875	

TRAVERSE SAMPLING DATA

Page 1 of

Client <u>Boeing</u> Date <u>8-31-89</u> Sample Location <u>Bldg 2-10</u> <u>Chronic Toxic #5 Exhaust</u> Operators <u>JAG-JMW</u> Sample Box # <u>B</u> Run# <u>1-PS. Mark 5</u> EQUIPMENT CHECKS Initial/Final Leak Rate Cfm <u>4.04</u> Leak Test Vac <u>15.1</u> <u> </u> Pitots, Pretest <u> </u> Pitots, Posttest <u> </u> Orsat Sampling System <u> </u> Tedlar Bag <u> </u> Thermocouple @ <u> </u> °F	SCHEMATIC TRAVERSE LAYOUT Stack Diameter <u> </u> Distance Upstream <u> </u> Distance Downstream <u> </u> <table style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="text-align: left;">Filter #</th> <th style="text-align: left;">H₂ care</th> <th style="text-align: left;">mgs</th> </tr> <tr> <th></th> <th>Final Wt.</th> <th>Initial Wt.</th> </tr> </thead> <tbody> <tr> <td>#1</td> <td>NaOH <u>619.7</u></td> <td></td> </tr> <tr> <td></td> <td>Bubbler <u>634.8</u></td> <td><u>1253.7</u></td> </tr> <tr> <td></td> <td></td> <td><u>0.8</u></td> </tr> <tr> <td>#2</td> <td>HNO₃ <u>625.5</u></td> <td><u>626.1</u></td> </tr> <tr> <td></td> <td>Impinger</td> <td><u>- .6</u></td> </tr> <tr> <td>#3</td> <td>Bubbler <u>535.6</u></td> <td><u>535.7</u></td> </tr> <tr> <td></td> <td></td> <td><u>0.4</u></td> </tr> <tr> <td>#4</td> <td>Silica</td> <td></td> </tr> <tr> <td></td> <td>Gel <u>798.6</u></td> <td><u>795.9</u></td> </tr> <tr> <td></td> <td></td> <td><u>2.7</u></td> </tr> <tr> <td colspan="2">TOTAL WATER VOLUME</td> <td><u>3.3</u></td> </tr> </tbody> </table>	Filter #	H ₂ care	mgs		Final Wt.	Initial Wt.	#1	NaOH <u>619.7</u>			Bubbler <u>634.8</u>	<u>1253.7</u>			<u>0.8</u>	#2	HNO ₃ <u>625.5</u>	<u>626.1</u>		Impinger	<u>- .6</u>	#3	Bubbler <u>535.6</u>	<u>535.7</u>			<u>0.4</u>	#4	Silica			Gel <u>798.6</u>	<u>795.9</u>			<u>2.7</u>	TOTAL WATER VOLUME		<u>3.3</u>	Start Time <u>10:26</u> Stop Time <u>11:26</u> Barometric Pressure "Hg <u>30.16</u> Static Pres "H₂O <u>0.0</u> Production Rate <u> </u> <u>Bath On</u> NOMOGRAPH SETUP X Moisture <u>2</u> Meter Temp. <u> </u> Stack Temp. <u>30</u> ΔH₀ <u>.943</u> Y <u>.975</u> Pitot# <u> </u> Side# <u> </u> Cp <u>.84</u> Nozzle Diameter <u>.188</u> K Factor <u> </u> Reference ΔP <u>.27</u> <u>ΔH = .17</u>
Filter #	H ₂ care	mgs																																							
	Final Wt.	Initial Wt.																																							
#1	NaOH <u>619.7</u>																																								
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TOTAL WATER VOLUME		<u>3.3</u>																																							

[illegible]

8/31/87

Breeding Plum 2

Run 1
(Purified Sking)Started with Parts into back (2) 10:26 AM
Remaining Parts 11:21 AM

Readings during Test:

10:26	5	1	0	1
10:27	12	2	200	2
10:29	23.0		1830	
10:45	23.0	Amps 26	1830	Amps 34
10:55	23.0	Amps	1830	Amps
11:03	23.1	Amps 8	1830	Amps
11:11	23.2	Amps 18	1820	Amps 18
11:20	23.2	Volts	1820	Amps
11:21	23.2		1820	Amps
11:22	0.5 V	(Off Ony Red)		

94.5° Bath Temperature

78°F ambient

Velocity Measurements between Run 1 & 2:

Opening #

1
6
11
16
21
26
31
36
41
46
51
56
61

ft/min reading

2270
1810
1740
1640
1650
1220
730
1060
730
820
680
340
620

(Readings from 11:35 - 11:50,
(Empty Bath)

$$(1 \times 0 + 2 \times 200 + 34 \times 1830 + 18 \times 1820) / 55 \times 60 / 55$$

1892 ampmph
22.4 volt

← Woodcock Air Sampling Here; Wet Surfaces Too

Sample Type:		Particle Size Distribution Bath On - Run 2	
Sampling Date:		November 30, 1989	
Sample Times:		12:02-13:07 (65 minutes)	
Run Number:		2	
Lab Numbers:		922208-922219, 922096-922100	
Airflow:		22,576 dscf/min	
Process Data			
Bath Amperage:		1883 amp-hr/hr	
Voltage:		21.5 volts D.C.	
Bath Temperature:		93.0° F	
Ambient Temperature:		73° F	
Stage #	Distribution d50's (microns)	Cumulative Cr % Less Than Stated Diameter	Chromium Concentration (mg/dscm)
1	16.58	98.61	0.003
2	15.00	97.65	0.002
3	5.57	70.65	0.052
4	2.91	30.51	0.077
5	1.66	11.16	0.037
6	1.11	8.26	0.006
7	0.90	7.79	0.001
8	0.64	7.54	0.001
9	0.52	7.25	0.001
10	0.44	7.25	0.000
final	<0.44	0.00	0.014
TOTAL			0.196

PILAT (U of W) MARK 5 PARTICLE SIZE ANALYSIS RESULTS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME	A:80M5CR2	Y FACTOR	1.012	mls	0.1	Percent
CLIENT	BOEING AEROSPACE	VOLUME H2O	N/A	SCF	20.9	Percent
LOCATION	SEATTLE, WASHINGTON	VOLUME H2O	N/A	%	0.0	Percent
SAMPLE SITE	PLANT NO. 2, BLDG. 2-10	PERCENT MOISTURE	3.54		28.85	gm/mole
SAMPLE DATE	CHROMIC TANK #5 EXHAUST	METER TEMP	88.7	Deg. F.	28.47	gm/mole
OPERATOR	NOVEMBER 30, 1989	STACK TEMP	76.8	Deg. F.	0.842	
	MACKEY	MARK 5 TEMP	76.8	Deg. F.	0.188	inches
RUN #	2-MKS/CT	BAROMETRIC PRESSURE	30.42	" of Hg	0.0002	inches
LAB #	922208	STATIC PRESSURE	-0.02	" of H2O	48.00	inches
START TIME	12:02	STACK PRESSURE	30.42	" of Hg	12.566	sq. ft.
STOP TIME	13:07	ORIFICE PRESSURE	0.43	" of H2O	0.30	" H2O
SAMPLE TIME	65.00	MKS OUTLET ABS. PRES	29.42	" of Hg	31.0	ft/sec
INITIAL METER VOL	902.199	MKS DELTA P	1.00	" of Hg	1862.5	ft/min
FINAL METER VOL	925.122	AVG. MARK 5 PRESSURE	30.32	" of Hg	23405.2	acft/min
VOLUME SAMPLED	22.923	SAMPLE RATE	0.35	cf/min	22575.7	dscf/min
VOLUME (STACK COND)	23.55	SAMPLE RATE (STD)	0.35	dscf/min	101	%
STD VOLUME	22.719	SAMPLE RATE (MKS)	0.36	dscf/min		
STD VOLUME	0.644		9.90	liter/min		
SET ID #	RB-QUARTZ B:T1R2-TANK 5-OUT					
CONFIGURATION:	1,6,12,90,110,110,105,105,78,FINAL FILTER					

STAGE #	CHROMIUM OM FILTERS micrograms	CHROMIUM OM FILTERS milligrams	FRACTIONAL %	PERCENT < than	AERODYNAMIC DIAMETER microns	CHROMIUM CONC. mg/dscm
NOZZLE:	0.00	0.00000				
1	1.72	0.00172	1.39	98.61	16.58	0.0027
2	1.20	0.00120	0.97	97.65	15.00	0.0019
3	33.50	0.03350	27.00	70.65	5.57	0.0521
4	49.80	0.04980	40.14	30.51	2.91	0.0774
5	24.00	0.02400	19.34	11.16	1.66	0.0373
6	3.60	0.00360	2.90	8.26	1.11	0.0056
7	0.58	0.00058	0.47	7.79	0.90	0.0009
8	0.32	0.00032	0.26	7.54	0.64	0.0005
9	0.35	0.00035	0.28	7.25	0.52	0.0005
10	0.00	0.00000	0.00	7.25	0.44	0.0000
FILTER	9.00	0.00900	7.25	0.00	<.44	0.0140
TOTAL:	124.07	0.12407	100.00			0.1928

BACKHALF ANALYSIS

RINSE OF SUBSTRATES	1.62
RINSE OF IMPACTOR	24.2
IMPINGER #1 & #2	6.08
IMPINGER #3 & 4	< 0.8
FILTER	< 0.15
TOTAL:	31.9 ug

62'

85.3 X < 10 Microns
2.19 Microns @ 50%

P A R T I C L E S I Z E A N A L Y S I S

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INPUT DATA

DATE OF TEST NOVEMBER 30, 1989
TIME OF TEST 12:02-13:07
LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
TEST NUMBER 1
PART. DIAM. TYPE C (I=IMP AERO, C=CLAS AERO, P=PHYSICAL)
TEST TYPE OUTLET
RUN NUMBER 2-TANK 5 - FILE NAME: B:T1R2-TANK 5.OUT
REMARKS:
IMPACTOR TYPE MK5-78
 MARK 5 IMPACTOR WITH A 78 HOLE FINAL STAGE

WATER VAPOR 3.54%
CARBON DIOXIDE 0.10%
CARBON MONOXIDE 0.00%
OXYGEN 20.90%
SULFUR DIOXIDE 0.00%
NITROGEN 79.00%
PARTICLE DENSITY 1.00 GRAMS/CM^3

GAS METER VOLUME	22.923	CUBIC FEET
IMPACTOR DELTA P	1.00	INCHES HG
ORIFICE DELTA P	0.43	INCHES H2O
STACK PRESS. (BELOW ATMOS.)	-0.02	INCHES H2O
BAROMETRIC PRESS.	30.42	INCHES HG
STACK TEMPERATURE	76.8	DEGREES F
METER TEMPERATURE	88.7	DEGREES F
IMPACTOR TEMPERATURE	76.8	DEGREES F
SAMPLE TIME	65.00	MINUTES
AV. VELOCITY OF STACK GAS	1862.50	FEET/MINUTE
GAS METER PRESSURE	0.00	IN HG
NOZZLE DIAMETER	0.19	INCHES
MAXIMUM AERODYN. DIAMETER	60.00	MICRONS

MASS GAIN ON STAGE	1	0.04 MG
MASS GAIN ON STAGE	2	0.02 MG
MASS GAIN ON STAGE	3	0.18 MG
MASS GAIN ON STAGE	4	0.21 MG
MASS GAIN ON STAGE	5	0.06 MG
MASS GAIN ON STAGE	6	0.06 MG
MASS GAIN ON STAGE	7	0.07 MG
MASS GAIN ON STAGE	8	0.03 MG
MASS GAIN ON STAGE	9	0.08 MG
MASS GAIN ON STAGE	10	0.23 MG
MASS GAIN ON FINAL FILTER		0.00 MG

RESULTS

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DATE OF TEST NOVEMBER 30, 1989
 TIME OF TEST 12:02-13:07
 LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
 TEST NUMBER 1 -
 RUN NUMBER 2-TANK 5
 ACTUAL FLOW RATE (STACK CONDITIONS) 0.358 CFM
 FLOW RATE (STANDARD CONDITIONS) 0.345 CFM
 PERCENT ISOKINETIC SAMPLING 99.62 %

VISCOSITY OF GAS STREAM		0.0001798 GRAMS/CM-SEC		
STAGE	CCF	DP (CLAS AERO)	DP (IMP AERO)	CUM FRACTION
1	1.010	16.580	16.659	0.9592
2	1.011	14.998	15.077	0.9388
3	1.029	5.573	5.652	0.7554
4	1.055	2.905	2.983	0.5413
5	1.096	1.661	1.738	0.4801
6	1.143	1.113	1.191	0.4190
7	1.178	0.898	0.975	0.3476
8	1.253	0.638	0.714	0.3170
9	1.312	0.524	0.600	0.2355
10	1.380	0.440	0.517	0.0010

TOTAL MASS PER DRY NORMAL CUBIC METER 1.5454 MG/CUBIC METER

PART. DIAM.	TYPE	C	(I=IMP AERO, C=CLAS AERO, P=PHYSICAL)		
ARTICLE SIZE	CUMFR	CUMFR	CUM.MASS	dM/dLOG D	
(MICRONS)	(STD. DEV.)	(PER CENT)	(MG/DRY N.CU.METER)		
0.20	-23.0317	0.0	0.000	0.000	
0.25	-17.3927	0.0	0.000	0.000	
0.40	-5.5153	0.0	0.000	0.000	
0.50	-1.1746	12.0	0.186	8.410	
0.75	-0.5342	29.7	0.458	0.472	
1.00	-0.2907	38.6	0.596	1.209	
1.50	-0.0709	47.2	0.729	0.352	
2.00	-0.0213	49.2	0.760	0.225	
2.50	0.0317	51.3	0.792	0.504	
4.00	0.3785	64.7	1.001	1.324	
5.00	0.5984	72.5	1.121	1.093	
7.50	0.8744	80.9	1.250	0.519	
10.00	1.0497	85.3	1.318	0.620	
15.00	1.5456	93.9	1.451	0.790	
20.00	2.1875	98.6	1.523	0.318	
25.00	2.7887	99.7	1.541	0.089	
40.00	5.6261	100.0	1.545	0.000	
50.00	12.3087	100.0	1.545	0.000	

***** RESULTS (CONTINUED) *****

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TIME OF TEST 12:02-13:07
 LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
 TEST NUMBER 1
 RUN NUMBER 2-TANK 5

*** INHALABLE PARTICULATE MATTER ***

NUM. MASS LESS THAN 2.5 MICRON: 0.79 MG/DRY NORMAL CU.METER (51.3 %)
 NUM. MASS LESS THAN 5.0 MICRON: 1.12 MG/DRY NORMAL CU.METER (72.5 %)
 CUM. MASS LESS THAN 10.0 MICRON: 1.32 MG/DRY NORMAL CU.METER (85.3 %)
 NUM. MASS LESS THAN 15.0 MICRON: 1.45 MG/DRY NORMAL CU.METER (93.9 %)

NOTE: DIAMETERS FOR INHALABLE PARTICULATE ARE CLASSICAL AERODYNAMIC DIAMETERS.

LOG-NORMAL SIZE DISTRIBUTION PARAMETERS

LEAST SQUARES LINE : $Y = -0.69 + 2.02X$
 MASS GEOMETRIC MEAN DIAMETER 2.193
 GEOMETRIC STANDARD DEVIATION 3.124
 CORRELATION COEFFICIENT 0.763

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[illegible]

76.8

Sample Type:		Particle Size Distribution Bath On - Run 3	
Sampling Date:		November 30, 1990	
Sample Times:		14:15-15:15 (60 minutes)	
Run Number:		3	
Lab Numbers:		922220, 922223-922233, 922101-922105	
Airflow:		22,433 dscf/min	
Process Data			
Bath Amperage:		1920 amp-hr/hr	
Voltage:		20.8 volts D.C.	
Bath Temperature:		94.0° F	
Ambient Temperature:		78° F	
Stage #	Distribution d50's (microns)	Cumulative Cr % Less Than Stated Diameter	Chromium Concentration (mg/dscm)
1	16.44	99.67	0.001
2	14.87	98.84	0.002
3	5.52	56.28	0.096
4	2.88	19.09	0.084
5	1.65	2.22	0.038
6	1.10	0.69	0.004
7	0.89	0.31	0.001
8	0.63	0.00	0.001
9	0.52	0.00	0.000
10	0.44	0.00	0.000
final	<0.44	0.00	0.000
TOTAL			0.229

PILAT (U of W) MARK 5 PARTICLE SIZE ANALYSIS RESULTS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME	A:80M5CR3	Y FACTOR	1.012	AVERAGE % CO2	0.1	Percent
CLIENT	BOEING AEROSPACE	VOLUME H2O	N/A	AVERAGE % O2	20.9	Percent
LOCATION	SEATTLE, WASHINGTON	PERCENT MOISTURE	N/A	AVERAGE CO	0.0	Percent
SAMPLE SITE	PLANT NO. 2, BLDG. 2-10	METER TEMP	92.3	STACK GAS MW DRY	28.85	gm/mole
SAMPLE DATE	CHROMIC TANK #5 EXHAUST	STACK TEMP	80.0	STACK GAS MW WET	28.42	gm/mole
OPERATOR	NOVEMBER 30, 1989	MARK 5 TEMP	80.0	PITOT Cp	0.842	
	MACKAY	BAROMETRIC PRESSURE	30.45	NOZZLE DIAMETER	0.188	inches
RUN #	3-MKS/CF	STATIC PRESSURE	-0.02	NOZZLE AREA	0.0002	sq. ft.
LAB #	922223	STACK PRESSURE	30.45	STACK DIAMETER	48.00	inches
START TIME	14:15	ORIFICE PRESSURE	0.43	STACK AREA	12.566	sq. ft.
STOP TIME	15:15	MKS OUTLET ABS. PRES	29.45	VELOCITY HEAD	0.30	m H2O
SAMPLE TIME	60.00	MKS DELTA P	1.00	STACK GAS VELOCITY	31.1	ft/sec
INITIAL METER VOL	925.298	AVG. MARK 5 PRESSURE	30.35	STACK GAS VELOCITY	1868.8	ft/min
FINAL METER VOL	946.778	SAMPLE RATE	0.36	STACK GAS AIRFLOW	23483.6	scf/min
VOLUME SAMPLED	21.480	SAMPLE RATE (STD)	0.35	STACK GAS AIRFLOW	22433.1	scf/min
VOLUME (STACK CORR)	22.16	SAMPLE RATE (MKS)	0.37	% ISOXINETICS	103	%
STD VOLUME	21.171	ACTUAL CU FT	9.99			
STD VOLUME	0.600	DSCF				
SET ID # H-QUARTZ	B:T1R3-TANK 5-OUT					
CONFIGURATION:	1,6,12,90,110,110,105,105,78,FINAL FILTER					

STAGE #	CHROMIUM ON FILTERS micrograms	CHROMIUM ON FILTERS milligrams	FRACTIONAL %	PERCENT < than	AERODYNAMIC DIAMETER microns	CHROMIUM CONC. mg/dscm
NOZZLE:	0.00	0.00000				
1	0.45	0.00045	0.33	99.67	16.44	0.0008
2	1.12	0.00112	0.82	98.84	14.87	0.0019
3	57.80	0.05780	42.57	56.28	5.52	0.0964
4	50.50	0.05050	37.19	19.09	2.88	0.0842
5	22.90	0.02290	16.86	2.22	1.65	0.0382
6	2.08	0.00208	1.53	0.69	1.10	0.0035
7	0.52	0.00052	0.38	0.31	0.89	0.0009
8	0.42	0.00042	0.31	0.00	0.63	0.0007
9	0.00	0.00000	0.00	0.00	0.52	0.0000
10	0.00	0.00000	0.00	0.00	0.44	0.0000
FILTER	0.00	0.00000	0.00	0.00	<.44	0.0000
TOTAL:	135.79	0.13579	100.00			0.2264

BACKHALF ANALYSIS

RINSE OF SUBSTRATES < 0.006
RINSE OF IMPACTOR 26.1
IMPINGER #1 & #2 <2.2
IMPINGER #3 & 4 1.34
FILTER 0.15

TOTAL: 27.59 ug

P A R T I C L E S I Z E A N A L Y S I S

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INPUT DATA

DATE OF TEST NOVEMBER 30, 1989
TIME OF TEST 14:15-15:15
LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
TEST NUMBER 1
PART. DIAM. TYPE C (I=IMP AERO, C=CLAS AERO, P=PHYSICAL)
TEST TYPE OUTLET
RUN NUMBER 3-TANK 5 - FILE NAME: B:T1R3-TANK 5.OUT
REMARKS:
IMPACTOR TYPE MK5-78
 MARK 5 IMPACTOR WITH A 78 HOLE FINAL STAGE

WATER VAPOR 4.00%
CARBON DIOXIDE 0.10%
CARBON MONOXIDE 0.00%
OXYGEN 20.90%
SULFUR DIOXIDE 0.00%
NITROGEN 79.00%
PARTICLE DENSITY 1.00 GRAMS/CM^3

GAS METER VOLUME 21.480 CUBIC FEET
IMPACTOR DELTA P 1.00 INCHES HG
ORIFICE DELTA P 0.43 INCHES H2O
STACK PRESS. (BELOW ATMOS.) -0.02 INCHES H2O
BAROMETRIC PRESS. 30.45 INCHES HG
STACK TEMPERATURE 80.0 DEGREES F
METER TEMPERATURE 92.3 DEGREES F
IMPACTOR TEMPERATURE 80.0 DEGREES F
SAMPLE TIME 60.00 MINUTES
AV. VELOCITY OF STACK GAS 1868.80 FEET/MINUTE
GAS METER PRESSURE 0.00 IN HG
NOZZLE DIAMETER 0.19 INCHES
MAXIMUM AERODYN. DIAMETER 60.00 MICRONS

MASS GAIN ON STAGE 1 0.46 MG
MASS GAIN ON STAGE 2 0.09 MG
MASS GAIN ON STAGE 3 0.31 MG
MASS GAIN ON STAGE 4 0.32 MG
MASS GAIN ON STAGE 5 0.26 MG
MASS GAIN ON STAGE 6 0.14 MG
MASS GAIN ON STAGE 7 0.10 MG
MASS GAIN ON STAGE 8 0.14 MG
MASS GAIN ON STAGE 9 0.06 MG
MASS GAIN ON STAGE 10 0.05 MG
MASS GAIN ON FINAL FILTER 1.04 MG

RESULTS

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DATE OF TEST NOVEMBER 30, 1989
 TIME OF TEST 14:15-15:15
 LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
 TEST NUMBER 1 -
 RUN NUMBER 3-TANK 5
 ACTUAL FLOW RATE (STACK CONDITIONS) 0.365 CFM
 FLOW RATE (STANDARD CONDITIONS) 0.348 CFM
 PERCENT ISOKINETIC SAMPLING 101.21 %

VISCOSITY OF GAS STREAM		0.0001802 GRAMS/CM-SEC		
STAGE	CCF	DP(CLAS AERO)	DP(IMP AERO)	CUM FRACTION
1	1.010	16.438	16.518	0.8451
2	1.011	14.870	14.949	0.8148
3	1.029	5.524	5.604	0.7104
4	1.055	2.879	2.958	0.6027
5	1.097	1.646	1.724	0.5152
6	1.145	1.103	1.180	0.4680
7	1.180	0.890	0.967	0.4343
8	1.256	0.631	0.708	0.3872
9	1.316	0.518	0.595	0.3670
10	1.385	0.436	0.513	0.3502

TOTAL MASS PER DRY NORMAL CUBIC METER 5.0209 MG/CUBIC METER

PART. DIAM.	TYPE	C	(I=IMP AERO, C=CLAS AERO, P=PHYSICAL)		
PARTICLE SIZE	CUMFR	CUMFR	CUM.MASS	dM/dLOG D	
(MICRONS)	(STD. DEV.)	(PER CENT)	(MG/DRY N.CU.METER)		
0.20	-0.5795	28.1	1.412	0.977	
0.25	-0.5236	30.0	1.508	1.007	
0.40	-0.4059	34.2	1.719	1.064	
0.50	-0.3487	36.4	1.826	1.125	
0.75	-0.2300	40.9	2.054	1.596	
1.00	-0.1177	45.3	2.275	1.838	
1.50	0.0114	50.5	2.533	1.276	
2.00	0.1027	54.1	2.716	1.683	
2.50	0.1934	57.7	2.895	2.018	
4.00	0.4290	66.6	3.344	2.065	
5.00	0.5246	70.0	3.515	1.394	
7.50	0.5958	72.4	3.637	0.343	
10.00	0.6416	73.9	3.713	1.069	
15.00	0.9054	81.7	4.104	3.446	
20.00	1.3140	90.6	4.547	3.116	
25.00	1.7271	95.8	4.810	2.309	
40.00	4.1752	100.0	5.021	0.009	
50.00	10.6763	100.0	5.021	0.000	

TIME OF TEST 14:15-15:15
 LOCATION OF TEST BOEING PLANT NO. 2, BLDG. 2-10
 TEST NUMBER 1
 RUN NUMBER 3-TANK 5

*** INHALABLE PARTICULATE MATTER ***

CUM. MASS LESS THAN 2.5 MICRON:	2.90 MG/DRY NORMAL CU.METER (57.7 %)
CUM. MASS LESS THAN 5.0 MICRON:	3.51 MG/DRY NORMAL CU.METER (70.0 %)
CUM. MASS LESS THAN 10.0 MICRON:	3.71 MG/DRY NORMAL CU.METER (73.9 %)
CUM. MASS LESS THAN 15.0 MICRON:	4.10 MG/DRY NORMAL CU.METER (81.7 %)

NOTE: DIAMETERS FOR INHALABLE PARTICULATE ARE CLASSICAL AERODYNAMIC DIAMETERS.

LOG-NORMAL SIZE DISTRIBUTION PARAMETERS

LEAST SQUARES LINE :	$Y = -0.11 + 0.88X$
MASS GEOMETRIC MEAN DIAMETER	1.329
GEOMETRIC STANDARD DEVIATION	13.741
CORRELATION COEFFICIENT	0.996



TRAVERSE SAMPLING DATA

Page 1 of 1

922223

Client Battelle
Date 11/30/93
Sample Location Tank #31 Van. Sp.
Operators KSH
Sample Box # A
Run# 3 - Particle Size 11/30/93

EQUIPMENT CHECKS

Initial/Final

Leak Rate Cfm 0.001 / 0.001Leak Test Vac 0.1 / 0.1

Pitots, Pretest

Pitots, Posttest

Orsat Sampling System

Tedlar Bag

Thermocouple @ 0 °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter 4.5
Distance Upstream 0
Distance Downstream 0

Filter # 1 Acid mgs
Final Initial Net
Wt. Wt. Wt.
#1 Bubbler 1267.3 1259.1 - 0.001
#2 Impinger 619.7 617.5 - 0.001
#3 Bubbler 495.8 - 493.9 - 0.001
#4 Silica Gel 725.2 - 718.8 - 0.001
TOTAL WATER VOLUME 18.7

Start Time 14:15
Stop Time 15:15
Barometric Pressure "Hg 30.45
Static Pres "H₂O 0
Production Rate Battelle

NOMOGRAPH SETUP

% Moisture 0
Meter Temp. 15
Stack Temp. 75
ΔH₀ 1.72 Y 1.010
Pitot # 4 Side # 4
Cp 0.87
Nozzle Diameter .188
K Factor 0
Reference ΔP 0

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F
				Ideal	Actual	In	Out				
3	0	925.298	.30	.43	.43	63	64	9	N/A	45	80
"	5		.30	.43	.43	86	64	9		45	80
"	10		.30	.43	.43	105	64	9		45	80
"	15		.30	.43	.43	115	65	9		46	80
"	20		.30	.43	.43	121	66	9		46	80
"	25		.30	.43	.43	124	68	9		45	80
"	30		.30	.43	.43	127	69	9		46	80
"	35		.30	.43	.43	129	70	9		45	80
"	40		.30	.43	.43	130	71	9		45	80
"	45		.30	.43	.43	131	72				80
"	50		.30	.43	.43	132	73				80
"	55		.30	.43	.43	133	74				80
	60	946.778									

430

92.3

80

Particle Sizing

11/30/89

Chromic Acid Bath 5

Temp. (Bath) 94°F Part Descrip. - Supports

Run 3 PS	Time	Voltage	Amperes
Bath on	1415	2.0	200
	1416	3.5	1200
	1417	8.0	1800
	1418	13.5	1800
	1419	18.0	1800
	1420	21.5	
	1421	21.8	
	1422	22.0	
	1423	22.0	
	1424	22.0	
	1425	22.0	
	1430	22.0	
	1435	↓	
	1440	22.0	
	1445	↓	
	1450	22.0	
	1455	↓	
	1500	22.0	
	1505	22.0	1800
Bath off	1510		

(200 + 1200 + 53 * 1800)

1920 amp-hr/l

20.8 volts

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SECTION 4
AMBIENT WORKROOM SAMPLE

Sample Type:	Ambient Workroom Sample 1
Sampling Date:	August 31, 1989
Sample Times:	10:22-16:52 (390 minutes)
Run Number:	1
Lab Numbers:	915740, 915741, 915743
Process Data	
Bath Status:	Bath On and Off During Test
Ambient Temperature:	80° F
Hexavalent Chromium	
Concentration:	0.295 mg/dscm
Concentration:	0.14 ppm
Total Chromium	
Concentration:	0.249 mg/dscm
Concentration:	0.12 ppm
Description of sampling set-up: Sampling site was 53 inches above ground, 7 feet from the north end on tank #5, 10 inches away from east edge of tank, adjacent to ventilation opening #56. The top edge of the tank is 43 inches above the ground. The acid liquid level was approximately 6 inches below the top of the tank.	

METHOD 1-5 - HEXAVALENT AND TOTAL CHROMIUM EMISSIONS
AM TEST, INC. - AIR QUALITY DIVISION

FILE NAME:	3BO#5WR1	LAB #:	915740
CLIENT:	BOEING PLANT #2	START TIME:	10:22 O'CLOCK
LOCATION:	SEATTLE, WASHINGTON	STOP TIME:	16:52 O'CLOCK
SAMPLE SITE:	BLDG 210/TANK #5	SAMPLE TIME:	390.0 MINUTES
SAMPLE DATE:	AUGUST 31, 1989	CONDITION:	BATH ON & OFF
RUN #:	1-CR+6		
OPERATORS:	S. MACKEY		
CONTACT:	P. SIEBENALER		

FINAL WT. OF H2O G.	INIT. WT. OF H2O G.	NET WT. OF H2O G.	PITOT Cp:	NA
			NOZZLE DIA INCHES:	NA
			NOZZLE AREA FT^2:	NA
1274.9	1250.4	24.5	STACK DIA. INCHES:	NA
629.0	629.9	-0.9	STACK AREA FT^2:	NA
475.3	473.4	1.9	METER TEMP. DEG F:	147.1
817.4	783.8	33.6	BAROM. PRES. "HG:	30.20
TOTAL H2O GAIN:		59.1	STATIC PRES. "H2O:	0.00
TOTAL VOLUME (SCF)		2.79	STACK PRES. "HG:	30.20
PERCENT MOISTURE:		1.45	ORIFICE PRES "H2O:	1.0
Bws:		0.0145	METER PRES. "HG:	30.27

INIT. METER VOL.:	830.633	AVERAGE % CO2:	0.1
FINAL METER VOL.:	1045.964	AVERAGE % O2:	20.9
VOLUME SAMPLED:	215.331	AVERAGE % CO:	0
STD VOLUME (DSCF):	189.298	STACK GAS MW. DRY:	28.85
STD VOLUME (DSCM):	5.361	STACK GAS MW. WET:	28.69
Y FACTOR:	0.999		

HEXAVALENT CHROMIUM

HEXAVALENT CHROMIUM CONCENTRATION (ug):	1581.3
HEXAVALENT CHROMIUM CONCENTRATION (mg/dscm):	0.295
HEXAVALENT CHROMIUM IN SAMPLE (ppm):	0.14

TOTAL CHROMIUM

TOTAL CHROMIUM CONCENTRATION (ug):	1336.7
TOTAL CHROMIUM CONC. ON FILTER (ug):	< 0.30
TOTAL CHROMIUM CONCENTRATION (mg/dscm):	0.249
TOTAL CHROMIUM IN SAMPLE (ppm):	0.12

*Note: Slight differences in total chromium and hexavalent chromium concentrations are due to different methods of analysis.

AT

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TRAVERSE SAMPLING DATA

Page 1 of 1

3B045WR

Client Boeing Plant 2
 Date 8/31/87
 Sample Location Workroom Air
NE of Bath #5
 Operators KSM
 Sample Box A
 Run# 1 Workroom (Total Cr/Cr + 6 Sample Train)

EQUIPMENT CHECKS

Initial/Final
 Leak Rate Cfm <.02 / .005
 Leak Test Vac 15" / 11"
 Pitots, Pretest
 Pitots, Posttest
 Orsat Sampling System
 Tedlar Bag
 Thermocouple @ _____ °F

SCHEMATIC TRAVERSE LAYOUT

Stack Diameter N/A
 Distance Upstream _____
 Distance Downstream _____

Filter # Tellur tare _____ mgs
 Final Initial Net
 Wt. Wt. Wt.

#1 Bubbler 625.7 - 1250.4 = _____#2 Impinger 629.0 - 629.9 = _____#3 Bubbler 475.3 - 473.4 = _____#4 Silica Gel 817.4 - 783.8 = _____

TOTAL WATER VOLUME _____

Start Time 10:22
 Stop Time 16:52
 Barometric
 Pressure "Hg 30.20"
 Static Pres "H₂O 0
 Production Rate _____

NOMOGRAPH SETUP

% Moisture N/A
 Meter Temp. N/A
 Stack Temp. _____
 ΔH@ _____ Y _____
 Pitot# _____ Side# _____
 Cp _____
 Nozzle Diameter _____
 K Factor _____
 Reference ΔP _____

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading (ΔP), In. H ₂ O	Orifice Setting (ΔH), In H ₂ O		Gas Meter Temp °F		Pump Vacuum In. Hg Gauge	Filter Box Temp °F	Imp. Exit Temp °F	Imp. Stack Temp °F
				Ideal	Actual	In	Out				
Start	0	830.633	N/A	7.00	1.00	106	72	7.5		65	77
15	15				1.0	155	78	7.4		54	77
30	30				1.0	170	80	7.3		60	77
45	45				1.0	178	94	7.0		60	78
1 Hour	60				1.0	183	100	7.0		60	78
75	75				1.0	183	103	7.0		60	78
90	90				1.0	183	103	7.0		61	78
105	105				1.0	187	103	7.0		61	78
2 Hours	120				1.0	190	103	7.0		61	79
135	135				1.0	190	110	7.0		63	79
150	150				1.0	192	111	7.0		61	81
165	165				1.0	192	113	7.0		60	82
3 Hours	180				1.0	193	113	7.0		58	81
195	195				1.0	194	114	7.0		58	80
210	210				1.0	194	114	7.0		59	80
225	225				1.0	194	114	7.0		58	81
4 Hours	240				1.0	194	115	7.0		57	81
255	255				1.0	195	115	7.0		57	82
270	270				1.0	195	116	7.0		57	81
285	285				1.0	195	116	7.0		57	81
5 Hours	300				1.0	195	116	7.0		57	80
315	315				1.0	195	116	7.0		57	78
330	330				1.0	195	116	7.0		61	81
345	345				1.0	196	116	7.0		61	81
6 Hours	360				1.0	196	116	7.0		63	85

3/31/31 Breeding Plant 2

Run 1 - Started with Parts into bath @ 10:26 AM
 (Purified Sludge) Removed Parts 11:21 AM

Readings during Test:

10:45	23.0 Volts	1830 Amps	94.5° Bath Temperature
10:55	23.0 Volts	1830 Amps	
11:03	23.1 Amps	1830 Amp	78°F ambient
11:11	23.2 Amps	1820 Amp	
11:20	23.2 Volts	1820 Amp	
11:22	0.5 V (off Amps)	0 Amps	

Velocity Measurements between Run 1 & 2:
 Opening # ft/min reading

1	2270
2	1810
3	1740
4	1640
5	1650
6	1270
7	730
8	1060
9	730
10	820
11	690
12	340
13	620

(Readings from 11:35 - 11:50)
 (Empty Bath)

← Workroom Air Samples: Hot; Wet Surfaces Too

SECTION 5
VENTILATION SYSTEM EVALUATION

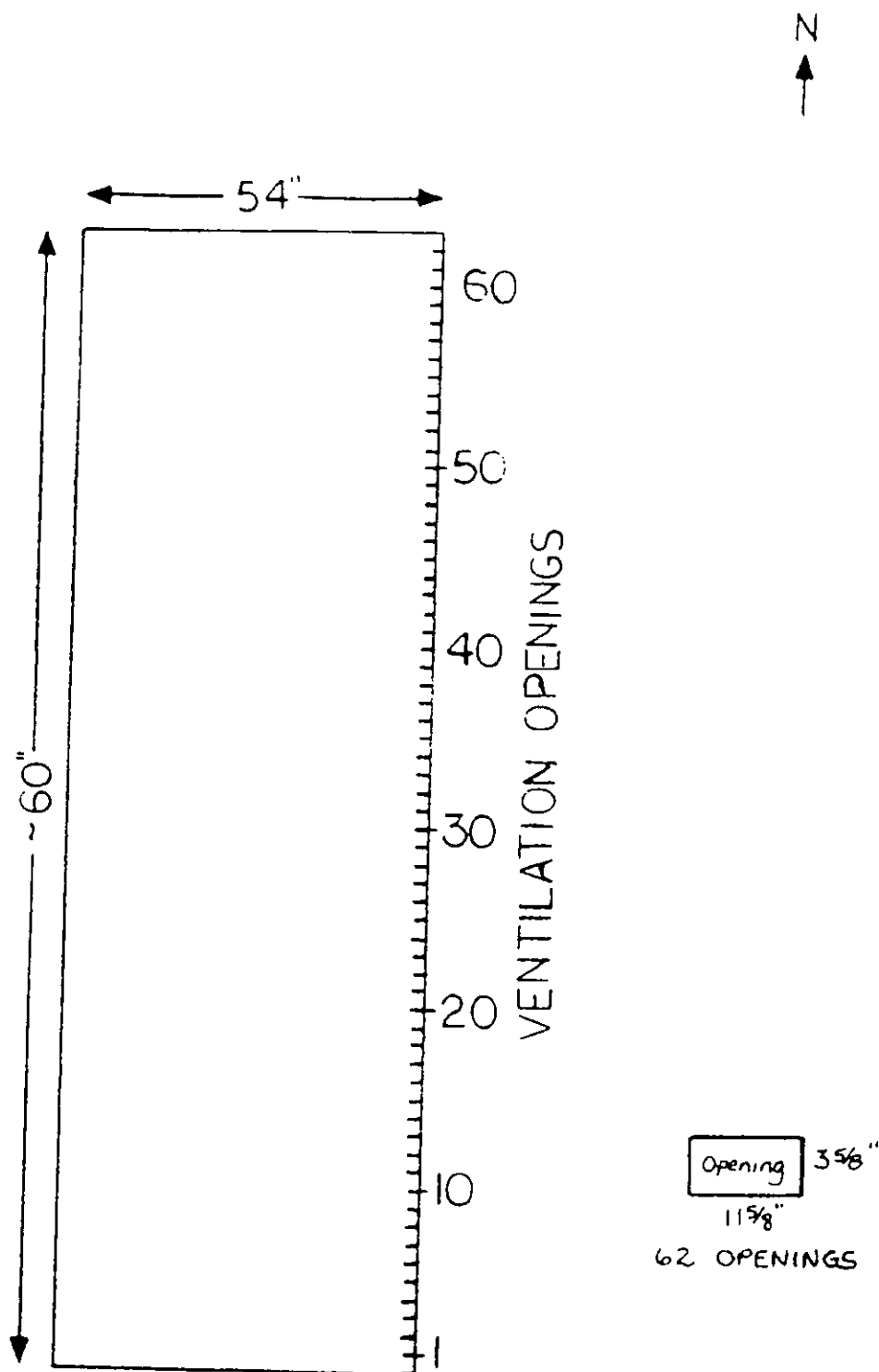


Figure 3. Diagram of the #5A chromic acid anodizing tank and ventilation system at Boeing Plant 2 in Seattle, Washington.

Sample Type:	Ventilation Opening Velocities
Sampling Dates:	August 22, August 31, November 30, 1989
Description of sampling procedure: A Kurz velometer with a digital readout in feet per minute was used to measure the velocity at each vent opening along the east side of the #5 chromic anodize aluminum tank. There are 62 openings along the east side. The measurement was taken at the centroid of each 3.375 by 11.625 inch opening. The velocity was lowest from openings 54-58, therefore, the ambient workroom sample was taken at that point.	

Velocity Traverse

.84 .

Boring - Plant 2 - #5A Chromic Anodize Aluminum Tank

Date: August 22, 1989

Time: 11:55-12:15

<u>Vent #</u>	<u>Velocity (ft/min)</u>	<u>Vent #</u>	<u>Velocity ft/min</u>
1	2050	32	
2		33	1330
3	2060	34	
4		35	980
5	2450	36	970
6		37	
7	1830	38	460
8		39	
9	1810	40	730
10		41	
11	1720	42	790
12		43	
13	1800	44	690
14		45	
15	1700	46	630
16		47	
17	1740	48	810
18		49	
19	1570	50	1090
20		51	
21	1560	52	1020
22		53	
23	1560	54	240
24		55	
25	930	56	340
26		57	
27	1110	58	620
28		59	
29	960	60	
30		61	470
31	810	62	

Ambient Temperature 74°F

Velocity Traverse

Boeing - Plant 2 - #5A Chromic Anodize Aluminum Tank
 Date: August 22, 1989
 Time: 16:55 - 17:15

<u>Vent #</u>	<u>Velocity (ft/min)</u>	<u>Vent #</u>	<u>Velocity ft/min</u>
1		32	1060
2	1780	33	
3		34	1090
4	1690	35	
5		36	1080
6	2140	37	
7		38	1130
8	1800	39	
9		40	830
10	1690	41	
11		42	860
12	1640	43	
13		44	780
14	1770	45	
15		46	820
16	1560	47	
17		48	730
18	1420	49	
19		50	690
20	1500	51	
21		52	690
22	1470	53	
23		54	700
24	1220	55	
25		56	320
26	1390	57	
27		58	620
28	1190	59	
29		60	550
30	990	61	
31		62	510

Ambient Temperature 83°F

Velocity Traverse

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Boeing - Plant 2 - #5A Chromic Anodize Aluminum Tank
Date: August 31, 1989
Time: 11:35 - 11:50

<u>Vent #</u>	<u>Velocity (ft/min)</u>	<u>Vent #</u>	<u>Velocity ft/min</u>
1	2270	32	
2		33	
3		34	
4		35	
5		36	1060
6	1810	37	
7		38	
8		39	
9		40	
10		41	730
11		42	
12	1740	43	
13		44	
14		45	
15		46	820
16	1640	47	
17		48	
18		49	
19		50	
20		51	680
21	1650	52	
22		53	
23		54	
24		55	
25		56	340
26	1220	57	
27		58	
28		59	
29		60	
30		61	620
31	730	62	

Ambient Temperature 78°F

Velocity Traverse

87

Boeing - Plant 2 - #5A Chromic Anodize Aluminum Tank
Date: August 31, 1989
Time: 14:20-14:50

<u>Vent #</u>	<u>Velocity (ft/min)</u>	<u>Vent #</u>	<u>Velocity ft/min</u>
1	2380	32	
2		33	
3		34	
4		35	
5		36	1060
6	2230	37	
7		38	
8		39	
9		40	
10		41	820
11	1650	42	
12		43	
13		44	
14		45	
15		46	
16	1700	47	
17		48	
18		49	
19		50	
20		51	720
21	1610	52	
22		53	
23		54	
24		55	
25		56	280
26	1390	57	
27		58	
28		59	
29		60	
30		61	500
31	720	62	

Ambient Temperature 81°F

Velocity Traverse

88

Boeing - Plant 2 - #5A Chromic Anodize Aluminum Tank
Date: November 30, 1989
Time: 11:00 - 11:05

<u>Vent #</u>	<u>Velocity (ft/min)</u>	<u>Vent #</u>	<u>Velocity ft/min</u>
1		32	
2		33	
3		34	
4		35	970
5	2400	36	
6		37	
7		38	
8		39	
9		40	750
10	1790	41	
11		42	
12		43	
13		44	
14		45	950
15	1690	46	
16		47	
17		48	
18		49	
19		50	460
20	1560	51	
21		52	
22		53	
23		54	
24		55	260
25	960	56	
26		57	
27		58	
28		59	
29		60	600
30	1010	61	
31		62	

Ambient Temperature 76°F

SECTION 6
CHROMIC ACID SAMPLES
TAKEN FROM THE BATH

Sample Type:	Acid Sample
Sampling Date:	September 19, 1989
Lab Number:	917176
Hexavalent Chromium:	5,292,000 micrograms/sample 5.3 grams/sample
Total Chromium:	48,470,000 micrograms/sample 48.5 grams/sample

Sample Type:	Acid Sample
Sampling Date:	November 30, 1989
Lab Number:	922117
Hexavalent Chromium:	6,578,000 micrograms/sample 6.6 grams/sample
Total Chromium:	6,325,000 micrograms/sample 6.3 grams/sample

METHODOLOGY REFERENCES

Hexavalent chromium testing and analysis methods detailed in a paper titled "Method Development and Testing for Measurement of Source Levels of Hexavalent and Total Chromium" by Daniel G. Bivens and W.G. DeWees were utilized for this project. Sampling procedures specified in the July 1, 1988 Title 40 Code of Federal Regulations, Part 60 (40 CFR 60), Appendix A, Methods 1-5 were also used. Quality assurance procedures for source testing are outlined in the EPA's reference manual titled Quality Assurance Handbook for Air Pollution Measurement Systems, Volume 3, EPA-600/4-77-027b.

The specific sampling and analysis procedures utilized for the particle size distribution tests detailed in the operations manual for the Mark 3-5 impactor by Dr. Michael J. Pilat and the "Procedures for Cascade Impactor Calibration and Operation in Process Streams - Revised 1983 to Include PM₁₀ Source Sampling" prepared by Mr. D. Bruce Harris for the U.S. EPA, February 1984. Methodology and procedures specified in "Recommended Methodology for the Determination of Particle Size Distributions in Ducted Sources" prepared by the Southern Research Institute for the California Air Resources Board, ARB Contract A3-092-32, May 1986 were also referenced for sampling strategies.

SAMPLING AND ANALYSIS PROCEDURES

CHROMIUM SAMPLES

The sample train used for collecting chromium (Cr) and hexavalent chromium (Cr^{+6}) samples was an EPA Method 5 design with modifications. A typical Method 5 sample train schematic is included as Figure 5 in Appendix B of this report. The sample train did not include a heated front-half filter. A quartz probe with an attached nozzle^{what type nozzle, S.S.?} was inserted into the stack to draw a sample into the impingers. The probe was equipped with "S" type pitot tubes to measure velocity and a thermocouple sensor to measure the stack temperature at each sample point. The thermocouple probe was connected to a Fluke^R digital thermocouple indicator which is accurate to within ± 1 degree Fahrenheit. The condenser section of the sample train consisted of a modified Greenburg-Smith bubbler containing 100 milliliters (ml) of 0.1 N sodium hydroxide (NaOH) absorbing solution, a second impinger containing 100 ml of NaOH absorbing solution, a third impinger containing 100 ml of 0.1 N nitric acid (HNO_3), an empty bubbler, a back-half filter holder containing a Teflon^R filter, and a final bubbler containing indicating silica gel desiccant. The condenser section was maintained at a temperature below 68° F by adding ice to the condenser section throughout sampling.

The sample train was connected to a control box by means of an umbilical cord which contains a vacuum hose, pitot lines, thermocouple wires and a 4-wire electrical cord. The control box (meter box) was used to monitor stack conditions and to facilitate isokinetic sampling. The control box consists of a diaphragm pump which is used to pull the stack gas through the sample train, fine and coarse metering valves to control the sampling rate, a vacuum gauge which measures the pressure drop from the sampling nozzle to the metering valves, and a calibrated

dry gas meter. At the outlet of the dry gas meter is a calibrated orifice which is used to isokinetically control the flow of gas through the metering system. The pressure drop across the orifice was monitored with both low and high range magnehelic gauges. The pitot tubes utilized to measure stack gas velocity are connected to the control box via the umbilical cord. The control box contains low and high range magnehelic gauges which are used for the velocity measurement.

The chrome anodizing tank ventilation system exhausts to a 48 inch diameter circular stack with two (2) sampling ports available 90 degrees apart. The sampling ports are located 2 feet upstream and 6 feet downstream from the nearest flow disturbance. Two (2) traverses of six (6) points each were performed during each chromium test. Figure 4 located in Appendix B of this report is a schematic of the stack and the point locations selected. The sample probe was marked with felt pen to indicate the proper point location. ^{should have used an extension}

Stack condition measurements were made prior to collecting a sample. The appropriate sample nozzle was selected and isokinetic operating parameters were established utilizing a Hewlett-Packard programmable calculator. The sampling nozzle, probe and condenser section glassware were all cleaned and rinsed with dilute nitric acid and deionized water prior to assembling the sample train. The sample train was assembled and leak tested following the procedures outlined in Method 5. Before each test, a final check was made to assure that the process was operating at the desired production rate and the desired operating parameters (e.g. whether parts were in the bath or not). Ice was added to the condenser section. The sample nozzle was positioned in the stack at the first sample point. The sample pump was then turned on and the gas sampling rate was adjusted for isokinetic sampling. Sampling proceeded isokinetically at each of the 6 traverse

points. The pump was then turned off and the sample train was carefully moved to the second port. The nozzle was then positioned at the first point of the second traverse, the pump was again turned on and isokinetic sampling was performed at the 6 traverse points. Upon completion of the second traverse the sample probe was removed from the stack and a post-test leak check was performed according to Method 5 procedures.

Following sample collection, the samples were transferred to a temporary laboratory area for clean up. The back-half filters were transferred to glass petri dishes labeled with the sample date, client name and run number. These filters, along with an unused filter to be analyzed as a blank, were submitted to the Am Test laboratory for washing and chromium analysis.

The contents of the nozzle, probe liner and connective glassware were quantitatively transferred to a glass storage container using 0.1 N NaOH. An iodine flask with a female ball joint end was attached to the male ball joint end of the probe to assure that no material collected inside the glassware was lost during the rinsing and brushing of the probe. The contents of the iodine flask were quantitatively transferred to the storage container labeled with sample date, client name, and run number. A portion of the reagent was placed in a glass storage container and submitted with the samples to the Am Test laboratory for analysis for chromium.

The bubblers and impingers utilized for the condenser section of the sample train were weighed with a readability of ± 0.1 grams before and after sampling on a Mettler PE3000 electronic top loading balance. The difference between the initial and final weights of the condenser section constitute the amount of moisture gain

during the run. The contents of the first two (2) impingers containing sodium hydroxide absorbing solution were then transferred to a 1000 ml glass graduated cylinder. The bubblers and impingers were rinsed with 0.1 N NaOH reagent into the graduated cylinder, and the liquid level was recorded. The liquid was then transferred to the corresponding sample container for each run which contained the probe wash. For some samples, the solutions from the probe rinse, 1st impinger and 2nd impinger were analyzed separately to evaluate collection efficiency for this method. The liquid in the 3rd and 4th impingers containing HNO_3 were transferred to a 1000 ml glass graduated cylinder. The bubblers and impingers were rinsed with 0.1 N HNO_3 reagent into the graduated cylinder, and the liquid level was recorded. The sample was then transferred to a labeled sample container. The samples were then transferred to the Am Test, Inc. laboratory for chromium analysis. The sodium hydroxide solutions were analyzed for both hexavalent chromium and total chromium. The nitric acid solutions and the teflon filter were analyzed for total chromium only. Hexavalent chromium (Cr^{+6}) analysis was performed using EPA Method 7176, which is a diphenylcarbazide colorimetric method. The hexavalent chromium analyses were completed within 48 hours after collecting the samples. Total chromium (Cr) analysis was performed using EPA Method 200.7, which utilizes inductively coupled plasma (ICP) spectroscopy. The laboratory results were presented in units of micrograms per sample. Those units were converted to concentrations in air.

PARTICLE SIZE DISTRIBUTION SAMPLES

Cascade impactors separate the sampled aerosol particles into size increments by inertial impaction of the particles onto a collection surface, in this case ultrapure microfiber quartz filter paper. This occurs at successive stages through the impactor, hence the name "Cascade". The resulting index of the measured particle

size is traditionally separated by the particle diameter collected with 50% collection efficiency by each jet stage, and this diameter is usually called the "cut diameter" and is characterized by the symbol " D_{50} ". The aerodynamic cut diameter is the diameter of a unit density sphere which will be collected with 50% efficiency by the specific impactor jet stage.

For these tests Am Test used straight nozzles at the inlet to a Pilat (U of W) Mark 5 source test Cascade impactor. A photograph of the Mark 5 impactor with a straight nozzle is illustrated in Figure 6 in Appendix B of this report. A complete Cascade impactor sampling train is illustrated in Figure 7 in Appendix B of this report. The Mark 5 impactors were assembled using the jet stages listed for configuration #78, which has a 78 hole final stage. In this configuration, the Mark 5 impactor provided eleven (11) aerodynamic particle size cut diameters, of sizes between 0.25 and 20 microns in diameter.

Whatman QM-A ultra-pure quartz microfiber filter substrates which were mounted on stainless steel inserts and baked at 500° F for 2 hours, desiccated for 24 hours and weighed to a constant weight on a Mettler AE163 electronic balance (set to a time integrating mode with a readability of 0.01 milligrams) were used to line each collection plate. Following the impactor run, the substrates were desiccated for 24 hours and weighed again.

The impactor probe was connected to a condenser section identical to the chromium samples. The sample train was connected to a control box as described for the chromium samples. The impingers were weighed before and after each test for calculating the percent moisture in the gas stream. The condenser section was leak-checked prior to starting a test. The flow rate through the impactor was pre-

set to the rate necessary for isokinetic sampling at the specific velocity point. The particle size distribution samples were collected at a point of average velocity in the stack. The impactor was positioned in the stack at the sample point with the nozzle facing away from the gas stream and was allowed to come to stack temperature for approximately one-half (1/2) hour. Ice was added to the condenser section of the particle size sample train. The nozzle was then turned to face the flow and the sample pump was turned on and the sampling rate was maintained at the pre-set rate necessary for isokinetic sampling.

After a sample had been collected the pump was then turned off and the impactor was removed from the stack, taking care that the nozzle tip did not touch the port nipple. The impactor was detached from the probe and carefully transferred to Am Test's temporary laboratory. The impactor was allowed to cool and was carefully disassembled and inspected. Each stage was inspected for proper particulate matter deposition. All interior impactor surfaces were checked for wall loss. The stainless steel inserts, onto which the substrates are mounted, were transferred to individual petri dishes and placed into a Labconco^R desiccator. Two (2) substrate blanks for each sample set were kept with the sample to audit desiccation and weighing techniques. To obtain final weights, the substrates were desiccated for 24 hours and weighed. The weights were recorded on the data sheet and in a bound laboratory notebook. The total weight of each substrate and stainless steel insert is approximately 1 gram. The low tare weight and simple donut-like shape of the Pilat (U of W) Mark 5 Cascade impactor insert-substrate combination allows for a great degree of weighing accuracy.

Once final weights were obtained for the substrates, each filter which lines the substrates was placed in a separate glass jar for submittal to the laboratory for

chromium analysis. The substrates were collectively rinsed into a separate jar. The appropriate blank filters and reagents were submitted along with the samples.

QUALITY ASSURANCE

A strict quality assurance program was followed throughout preparation, sampling, analysis, data reduction, and report preparation. This program includes recent equipment calibrations, careful chain-of-custody procedures, the use of ACS quality or better reagents, analysis of control samples, and "by-hand" calculation checks of computerized results.

The sample nozzles used to collect isokinetic samples were calibrated on-site before sampling using inside calipers readable to 0.001 inches. The dry gas meters used to accurately measure sample volumes have been recently calibrated using a spirometer at the Washington State Department of Ecology (DOE) laboratory in Redmond, Washington. The "S" type pitot tubes used for velocity measurement have also been recently calibrated utilizing a wind tunnel and a standard "P" type pitot tube and are inspected regularly for proper alignment. The digital Fluke thermocouple indicators used to obtain temperature measurements are accurate to ± 1 degree Fahrenheit. Each thermocouple probe used to monitor temperatures is checked quarterly at 3 different temperature settings.

In addition to quantitative clean-up and analysis procedures, filter and reagent blanks, as well as standard solutions were carried throughout the analysis procedures. The samples were weighed to constant weights of ± 0.05 milligrams following desiccation in a Labconco "Autodry Cabinet" desiccator. This desiccator is an electronic dehumidifier which automatically maintains the humidity inside the desiccator. The dehumidifier automatically recharges the internal desiccant every 5.5 hours with a small heater and fan inside the chamber. An Airguide humidity indicator accurate to $\pm 1\%$ is used to check the humidity inside the

desiccator when obtaining tare and final weights. Indicating silica gel desiccant is added to the chamber as necessary to maintain the humidity at the desired level. The Mettler AE163 electronic balance used to obtain weights was set to a time integrating mode (100,000 readings per minute) with a readability of 0.01 milligrams. The balance was calibrated prior to every use. The calibration of Am Test's Mettler balances is checked by the manufacturer on a yearly basis.

Each stage of the Mark 5 cascade impactor has been calibrated using a polydispersed aerosol and optical size analyzer at a range of temperatures and flow rates to determine the inertial impaction parameter for that particular stage. This parameter is intrinsic for each stage being controlled by the physical configuration of the stage and other pertinent aerodynamic values. The stages are periodically examined for wear and corrosion and are replaced as necessary.

Additional information with respect to Am Test, Inc.'s laboratory quality assurance and quality control protocol programs is included in the Appendix of this report.

CALCULATION OF RESULTS

The velocity, airflow, molecular weight, and moisture results were calculated using EPA 40 CFR 60 Method 1-5 criteria. Copies of pertinent equations are included in the Appendix of this report. Final results calculations were performed using spreadsheet programs run on a Hewlett-Packard Vectra computer system. Sample "by-hand" calculations of results were completed using a Hewlett-Packard calculator, and may be found in the Appendix of this report.

The particle size distribution results were computer generated in the Am Test Air Quality Division office utilizing a Cascade impactor data reduction program. This computer program is considered as "state-of-the-art" by particle sizing experts throughout the United States. Computer printouts of the particle size distribution results are included in Appendix A. To plot the particle size distribution, the column titled "aerodynamic diameter" (in micrometers) was plotted against the column titled "percent less than" on semi-log versus probability graph paper to illustrate the particle size distribution for each Mark 5 run.

ANALYSIS REPORT

CLIENT: Am Test, Inc. - Air Quality Division

DATE RECEIVED: 8/23/89

REPORT TO: Kris Hansen

DATE REPORTED: 9/11/89

BOEING PLANT #2

Laboratory Sample Numbers	Client Identification	Total Chromium (ug)	Hexavalent Chromium (ug)	Volume (ml)
915088	Blank (0.1N NaOH)	<0.60	0.036	100.
915089	Run 1 Probe	1,040.	1,670. 1,650.] 1060	215.
915090	Run 1 First Impinger	78.6	94.	130.
915091	Run 2 Probe & First Impinger	1,030.	1,970.	285.
915092	Run 3 Probe & First Impinger	283.	321.	329.
915093	Run 1 Second Impinger	102.	120.	130.
915094	Run 2 Second Impinger	113.	122. 123.] 122.5	139.
915095	Run 3 Second Impinger	2.2	11.5	144.
DETECTION LIMIT*		0.006	0.025	-
EPA Method used		200.7	7176	-

< = less than

*Values reported in mg/l.

JTD/pb

REPORTED BY


John T. Dailey

ANALYSIS REPORTCLIENT: Am Test, Inc. - Air Quality
Division

DATE RECEIVED: 8/23/89

REPORT TO: Kris Hansen

DATE REPORTED: 8/31/89

BOEING PLANT #2

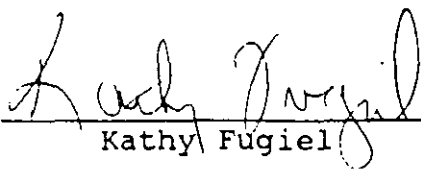
Laboratory Sample Nos.	Client Identification	Total Chromium	Volume (ml)
915096	Blank Filter	<0.15*	-
915097	Run 1 Filter	<0.15*	-
915098	Run 2 Filter	<0.18*	-
915099	Run 3 Filter	<0.15*	-
915100	Run 1 HNO ₃ Solution	<1.00**	166.
915101	Run 2 HNO ₃ Solution	<1.02**	170.
915102	Run 3 HNO ₃ Solution	<0.894**	149.
915103	Blank Solution (0.1 N HNO ₃)	<0.600**	100.
DETECTION LIMIT		0.006	-
EPA Method used		200.7 (ICP)	-

*Reported in ug/filter.

**Reported in ug.

KF/pb

REPORTED BY


Kathy Fugiel

ANALYSIS REPORT

CLIENT: Air Quality Division

DATE RECEIVED: 8/30/89

REPORT TO: Kris Hansen

DATE REPORTED: 9/7/89

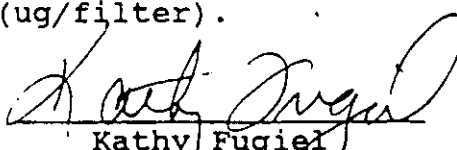
BOEING PLANT 2

Laboratory Sample No.	Client Identification	Total Chromium (ug)*	Hexavalent Chromium (ug)
915486	Run 1 1st Impinger & Probe	297.	324.
915487	Run 1 2nd Impinger	4.16	7.49
915488	0.1 N NaOH Blank	0.70	2.2
915489	Run 1 HNO ₃ 3rd & 4th Imp	<1.06	-
915490	0.1 N HNO ₃ Blank	<0.60	-
915491	Run 1 Teflon Filter*	2.0	-
915492	Teflon Filter Blank*	<0.30	-
EPA Method		200.7	7196

*Filter samples reported as (ug/filter).

KF/ja

REPORTED BY:


Kathy Fugiel

ANALYSIS REPORT

CLIENT: Am Test, Inc. - Air Quality Division

DATE RECEIVED: 9/1/89

REPORT TO: Kris Hansen

DATE REPORTED: 9/12/89

BOEING PLANT #2 - BUILDING 210

Laboratory Sample Numbers	Client Identification	Hexavalent Chromium	Total Chromium	Volume (ml)
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Blank

915737	0.1N NaOH	<2.5	<0.60	100.
915738	0.1N HNO ₃	-	<0.60	100.
915739	Teflon Filter	-	<0.30	-

R1 Workroom (Bath #5)

915740	NaOH Imp #1	1,553.	1,311.	187.
915741	NaOH Imp #2	28.3	25.7	133.
915742	NH ₄ PO ₃ Imp #3 & #4	-	<0.87	145.
915743	Filter	-	<0.30	-

R1-PS

915744	Impactor Rinse NaOH	156.	162.	183.
915745	Probe Rinse	1.16	4.45	89.
915746	Imp #1	<3.7	<0.90	150.
915747	Imp #2	<3.3	<0.79	132.
915748	Imp #3 & #4	-	<0.83	138.
915749	Filter	-	<0.30	-

R2

915750	Probe Rinse	620.	603.	178.
915751	Imp #1	26.1	32.0	130.
915752	Imp #2	43.2	42.3	137.
915753	Imp #3 & #4	-	1.09	121.
915754	Filter	-	<0.30	-

-2-

CLIENT: Am Test, Inc. - Air Quality Division

DATE RECEIVED: 9/1/89

REPORT TO: Kris Hansen

DATE REPORTED: 9/12/89

BOEING PLANT #2 - BUILDING 210

Laboratory Sample Numbers	Client Identification	Hexavalent Chromium	Total Chromium	Volume (ml)
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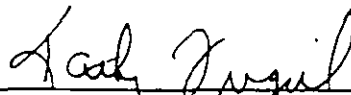
R3

915755	Probe	654.	596.	143.
915756	Imp #1	35.3	34.0	141.
915757	Imp #2	44.2	42.2	141.
915758	Imp #3 & #4	-	<0.96	160.
915759	Filter	-	<0.30	-

Liquid Solutions are reported in units of ug.
Filter Samples are reported in units of ug/filter.
Volumes are reported in units of ml.

KF/pb

REPORTED BY


Kathy Fugiel

ANALYSIS REPORTCLIENT: Am Test, Inc. - Air Quality
Division

DATE RECEIVED: 9/6/89

DATE REPORTED: 9/18/89

REPORT TO: Kris Hansen

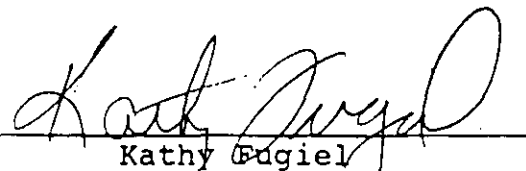
BOEING PLANT #2

Laboratory Sample Nos.	Client Identification	Total Chromium	Volume (ml)
915962	Rinse of Substrates	0.75 ug	50.
915963	Blank Solution	<0.60 ug	100.
915965	Blank Filters	0.65*	-
915966	1	4.75*	-
915967	2	3.47*	-
915968	3	62.7*	-
915969	4	55.7*	-
915970	5	12.*	-
915971	6	3.3*	-
915972	7	0.67*	-
915973	8	0.42*	-
915974	9	0.45*	-
915975	10	0.40*	-
915976	12	6.7*	-
Filter Reagent Blank		0.20 ug	-
EPA Method No.		200.7 (ICP)	-

*Reported in ug/filter.

KF/pb

REPORTED BY


Kathy Eugiel

ANALYSIS REPORT

CLIENT: Air Quality Division

DATE RECEIVED: 9/20/89

REPORT TO: Kris Hansen

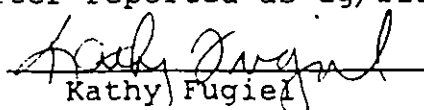
DATE REPORTED: 9/29/89

BOEING PLANT 2

Laboratory Sample No.	Client Identification	Chromic Acid Bath #5	
		Total Chromium	Hexavalent Chromium
917067	Run 1 Imp 1 & 2 (.1N NaOH)	24.5	<6.25
917068	Run 1 Imp 3 & 4 (0.1 HNO ₃)	1.33	-
917069	Run 1 Teflon Filter	0.175	-
917070	Run 2 Imp 1 & 2 (0.1 NaOH)	268.	247. 246.]
917071	Run 2 Imp 3 & 4 (0.1 HNO ₃)	1.64	-
917072	Run 2 Teflon Filter	0.275	-
917073	Blank Solution (0.1 NaOH)	<1.00	<2.5
917074	Blank Solution (0.1 NHCO ₃)	1.2	-
917075	Blank Teflon Filter	0.175	-
917076	Bath #5 Sample 48,470,000.	5,292,000.	131
917077	Run 1 Probe (0.1 N NaOH)	28.8	23.1
917078	Run 2 Probe (0.1 N NaOH)	1,141.	1,212.
Detection Limit (ug/ml)		0.006	0.025
EPA Method		200.7 (ICP)	7196

Solutions reported in ug, filter reported as ug/filter.

REPORTED BY:


Kathy Fugiel

KF/ja

ANALYSIS REPORT

CLIENT: AM TEST, INC. - AIR QUALITY
DIVISION

DATE RECEIVED: 12/4/89
DATE REPORTED: 12/11/89
DATE REVISED: 1/30/90

REPORT TO: Kris Hansen

BOEING PLANT #2

Laboratory Sample Nos.	Client Identification	Chromium (ug/filter)	Volume (ml)
922208	Stage RB-1	1.72	-
922209	Stage RB-2	1.20	-
922210	Stage RB-3	33.5	-
922211	Stage RB-4	49.8	-
922212	Stage RB-5	24.0	-
922213	Stage RB-6	3.60	-
922214	Stage RB-7	0.58	-
922215	Stage RB-8	0.32	-
922216	Stage RB-9	0.35	-
922217	Stage RB-10	<0.15	-
922218	Stage RB-12	9.0	-
922219	Rinse of RB Substrates	1.62*	116.
922220	Rinse of H Substrates	<0.6*	100.
922221	Blank Sol.	<0.6*	130.
922222	Blank Filters**	<0.15	-
922223	Stage H-1	0.45	-
922224	Stage H-2	1.12	-
922225	Stage H-3	57.8	-

*Results are in ug.

**4 Mark 5 filters.

Filters were brought to 25 mls. total volume.

-2-

CLIENT: AM TEST, INC. - AIR QUALITY
DIVISION

DATE RECEIVED: 12/4/89
DATE REPORTED: 12/11/89
DATE REVISED: 1/30/90

REPORT TO: Kris Hansen

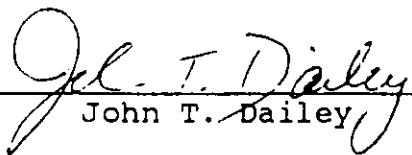
BOEING PLANT #2

Laboratory Sample Nos.	Client Identification	Chromium (ug/filter)	Volume (ml)
922226	Stage H-4	50.5	-
922227	Stage H-5	22.9	-
922228	Stage H-6	2.08	-
922229	Stage H-7	0.52	-
922230	Stage H-8	0.42	-
922231	Stage H-9	<0.15	-
922232	Stage H-10	<0.15	-
922233	Stage H-12	<0.15	-
BLANK 1		<0.15	-
BLANK 2		<0.15	-

Filters were brought to 25 mls. total volume.

JTD/pb

REPORTED BY


John T. Dailey

ANALYSIS REPORT

CLIENT: AM TEST, INC. - AIR QUALITY
DIVISION

DATE RECEIVED: 11/30/89
DATE REPORTED: 12/1/89
DATE REVISED: 1/30/90

REPORT TO: Kris Hansen

BOEING PLANT #2

Laboratory Sample Numbers	Client Identification	Chromium (ug)	Hexavalent Chromium (ug)	Volume (ml)
922090	Blank 0.1 N NaOH	<0.6	<25.	100.
922091	Blank 0.1 N HNO ₃	<0.6	-	100.
922092	Blank Teflon Filter	<0.15*	-	-
922093	Run 1 Teflon Filter	<0.15*	-	-
922094	Run 1 NaOH Imp 1 & 2	470.	438.	395.
922095	Run 1 HNO ₃ Imp 3 & 4	<1.1	-	177.
922096	R2-PS Teflon Filter	<0.15*	-	-
922097	R2-PS NaOH Imp 1 & 2	6.08	<70.	380.
922098	R2-PS HNO ₃ Imp 3 & 4	<0.8	-	133.
922099	R2 Impactor Stages Rinse	24.2	<53.	212.

*Results are in ug/filter.

Filters were brought to 25 mls. total volume.

-2-

CLIENT: AM TEST, INC. - AIR QUALITY
DIVISIONDATE RECEIVED: 11/30/89
DATE REPORTED: 12/1/89
DATE REVISED: 1/30/90

REPORT TO: Kris Hansen

BOEING PLANT #2

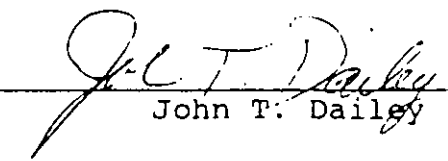
Laboratory Sample Numbers	Client Identification	Chromium (ug)	Hexavalent Chromium (ug)	Volume (ml)
922100	R2 Straight Nozzle Rinse	86.6	68.4	38.
922101	R3-PS Teflon Filter	-	-	-
922102	R3-PS NaOH Imp 1 & 2	<2.16	<90.	360.
922103	R3-PS HNO ₃ Imp 3 & 4	1.34	-	168.
922104	R3-PS Impactor Rinse	26.1	<32.	128.
922105	R3-PS Nozzle Rinse	30.2	22.6	20.
922106	R4 Teflon Filter	<0.15*	-	-
922107	R4 NaOH Imp 1 & 2	6.90	679.	350.
922108	R4 HNO ₃ Imp 3 & 4	<0.9	-	157.

*Results are in ug/filter.

Filters were brought to 25 mls. total volume.

JTD/pb

REPORTED BY


John T. Dailey

ANALYSIS REPORTCLIENT: AM TEST, INC. - AIR QUALITY
DIVISIONDATE RECEIVED: 12/1/89
DATE REPORTED: 12/11/89
DATE REVISED: 1/30/90

REPORT TO: Kris Hansen

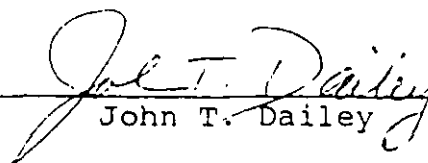
BOEING PLANT #2

Laboratory Sample Nos.	Client Identification	Chromium (ug)	Hexavalent Chromium (ug)	Volume (ml)
922111	Blank 0.1 N NaOH	<0.6	25.	100.
922112	Blank 0.1 N HNO ₃	<0.6	-	100.
922113	Blank Teflon Filter	<0.15*	-	-
922114	Run 5 NaOH Imp 1 & 2	458.	442.	409.
922115	Run 5 HNO ₃ Imp 3 & 4	1.20	-	171.
922116	Run 5 Teflon Filter	<0.15*	-	-
922117	Chromic Acid Bath #5	6,325,000.	6,578,000.	253.

*Results are in ug/filter.
Filters were brought to 25 mls. total volume.

JTD/pb

REPORTED BY


John T. Dailey

APPENDIX B

SAMPLE CALCULATION SHEET
METHODS 1-5

CLIENT: BOEING PLANT #2

DATE OF TEST: 8-22-89

LOCATION: Seattle, Wash.

RUN #: 1 B0#5CR1

Lab #
915090

1- Bath On

Particulate Matter Emission Concentration - Equation 5-1

$$V_{mstd} = 17.647^\circ R / ^\circ Hg * 54.27 ft^3 * .993 * (30.04 ^\circ Hg + (1.278 ^\circ H_2O / 13.6)) / (460 + 75.0 ^\circ F)$$

$$= 53.570 \text{ dscf}$$

$$dscm = 53.570 \text{ dscf} / 35.31 \text{ ft}^3/\text{m}^3$$

$$= 1.517 \text{ dscm}$$

Substitution of Equation 5-4 into 5-5

$$W_a = \text{_____ mg} * \text{_____ ml} / \text{_____ ml}$$

$$= \text{_____ mg}$$

$$M_n = (\text{net weight filter catch}) + (\text{net weight "B" section}) - W_a + \text{Back-half}$$

$$= \text{_____ mg} = \text{_____ mg} + \text{_____ mg} - \text{_____ mg} + \text{_____ mg}$$

$$C_s = (0.001 \text{ g/mg}) * (15.43 \text{ grains/gram}) * \text{_____ mg} / \text{_____ dscf}$$

$$= \text{_____ gr/dscf} \quad (\text{Equation 5-6})$$

$$\text{gr/dscf @ 7\% O}_2 = \text{_____ gr/dscf} * (20.9\% - 7\% \text{O}_2) / (20.9\% - \text{_____ \%O}_2)$$

$$= \text{_____ gr/dscf @ 7\% O}_2$$

$$\text{gr/dscf @ 12\% CO}_2 = \text{_____ gr/dscf} * 12\% / \text{_____ \%CO}_2$$

$$= \text{_____ gr/dscf @ 12\% CO}_2$$

$$\text{mg/dscm} = \text{_____ mg} / \text{_____ dscm}$$

$$= \text{_____ mg/dscm}$$

Particulate Matter Emission Rate

$$\text{pounds/hour} = \text{_____ gr/dscf} * \text{_____ dscf/min} * 60 \text{ min/hr} * 1 \text{ lb/7000 grains}$$

$$= \text{_____ lb/hr}$$

Moisture - Equation 5-2 and 5-3

$$V_{wstd} = 0.04715 \text{ ft}^3/\text{g} * 16.7 \text{ grams of H}_2\text{O collected in impingers}$$

$$= 0.787 \text{ scf}$$

SAMPLE CALCULATION SHEET (continued)
METHODS 1-5

$$B_{ws} = (.787 \text{ scf}) / (.787 \text{ scf} + 53.570 \text{ dscf})$$

$$= 0.0145$$

$$\% \text{Moisture} = 0.0145 \cdot 100$$

$$= 1.45 \%$$

Molecular weight - Equation 3-2

$$M_d = 0.440 \cdot (0.1 \% \text{CO}_2) + 0.320 \cdot (20.9 \% \text{O}_2) + 0.280 \cdot (79.0 \% \text{CO} + \% \text{N}_2)$$

$$M_d = 28.85 \text{ g/g-mole (dry)}$$

$$M_s = 28.85 \text{ g/g-mole} \cdot (1 - 0.0145) + 18.0 \text{ g/g-mole} \cdot 0.0145$$

$$M_s = 28.69 \text{ g/g-mole (wet)}$$

Stack gas velocity and volumetric flow rate - Equation 2-9 and 2-10

$$V_s = 85.49 \cdot .84 \cdot .4987 \cdot \sqrt{538.5^\circ \text{R} / 28.69 \text{ g/g-mole} / 30.04 \text{ "Hg}}$$

$$V_s = 28.30 \text{ ft/sec (std)}$$

$$Q_{sd} = 3600 \cdot (1 - 0.0145) \cdot 28.30 \text{ ft/sec} \cdot 12.566 \text{ ft}^2 \cdot (528^\circ \text{R} / 538.5^\circ \text{R}) \cdot (30.04 \text{ "Hg} / 29.92 \text{ "Hg})$$

$$= 1242249 \text{ dscf/hr} / 60 \text{ min/hr}$$

$$= 20704.1 \text{ dscf/min (dry standard cubic feet per minute)}$$

$$\text{acfm} = 28.30 \text{ ft/sec} \cdot 12.566 \text{ ft}^2 \cdot 60 \text{ sec/min}$$

$$= 21340.7 \text{ acfm (actual cubic feet per minute)}$$

Isokinetic variation - Equation 5-8

$$I = 0.09450 \cdot 53.570 \text{ dscf} \cdot 538.5^\circ \text{R} / (30.04 \text{ "Hg} \cdot 28.30 \text{ ft/sec} \cdot 60 \text{ min} \cdot 0.005343 \text{ ft}^2 \cdot (1 - 0.0145))$$

$$I = 10.1 \%$$

All of the above numbered equations are from the 40 CFR 60 and assume English units.

AMTEST

Example Calculations for Chromium
Boeing Plant 2, #5A Chrome Anodizing Tank
August 22, 1989, Run 1 with Bath On
Lab #'s 915089, 915090, 915093, 915097, 915100

Sample	$\mu\text{g Cr}$	$\mu\text{g Cr}^{+6}$
Probe	1040	1660
1 st Impinge	78.6	94.
2 nd Impinge	102	120
3 rd Impinge	<1.00	—
Filter	<0.15	—
TOTAL	1220.6	1874.0

Hexavalent Chromium (Cr^{+6})

$$\frac{1874 \mu\text{g in sample} \times \frac{1 \text{ mg}}{1000 \mu\text{g}}}{1.517 \text{ dscm of gas sampled}} = \boxed{1.235 \text{ mg/dscm}}$$

$$1.235 \text{ mg/dscm} \times \frac{22.414 \text{ ml/m-mole}}{51.996 \text{ mg/m-mole}} \times \frac{293.15^\circ\text{K}}{273.15^\circ\text{K}} \times \frac{1 \text{ m}^3}{10^6 \text{ ml}} \times 10^6 \text{ ppm} = \boxed{0.57 \text{ ppm}}$$

$$= \boxed{0.57 \text{ ppm}}$$

$$1.235 \text{ mg/dscm} \times 60 \text{ min/hr} \times \frac{1 \text{ m}^3}{35.32 \text{ ft}^3} \times 20704.1 \text{ dscf/m}^3 = \boxed{43,456.5 \frac{\text{mg}}{\text{hr}}}$$

$$43456.5 \text{ mg/hr} / 1825 \text{ amp-hr/hr} = \boxed{23.8 \text{ mg/amp-hr}}$$

Total Chromium

$$\frac{1220.6 \mu\text{g}}{1.517 \text{ dscm}} \times \frac{1 \text{ mg}}{1000 \mu\text{g}} = \boxed{0.805 \text{ mg/dscm}}$$

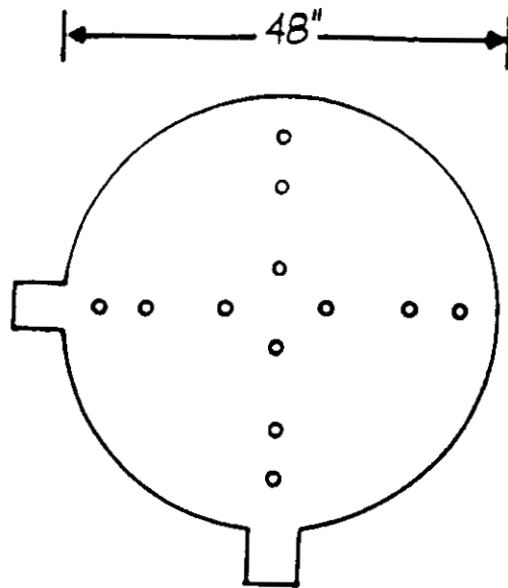
$$0.805 \text{ mg/dscm} \times \frac{22.414 \text{ ml/m-mole}}{51.996 \text{ mg/m-mole}} \times \frac{293.15^\circ\text{K}}{273.15^\circ\text{K}} \times \frac{1 \text{ m}^3}{10^6 \text{ ml}} \times 10^6 \text{ ppm} = \boxed{0.37 \text{ ppm}}$$

$$0.805 \text{ mg/dscm} \times 60 \text{ min/hr} \times \frac{1 \text{ m}^3}{35.32 \text{ ft}^3} \times 20704.1 \text{ dscf/m}^3 = \boxed{28304.7 \text{ mg/hr}}$$

$$\frac{28304.7 \text{ mg/hr}}{1825 \text{ amp-hr/hr}} = \boxed{15.5 \text{ mg/amp-hr}}$$

CROSS SECTIONAL AREA

Traverse Point	Distance (inches)
1	2.11
2	7.01
3	14.21
4	33.79
5	40.99
6	45.89



STACK DIMENSIONS

48 inch diameter circular stack

2 ports

A = 2 feet

B = 6 feet

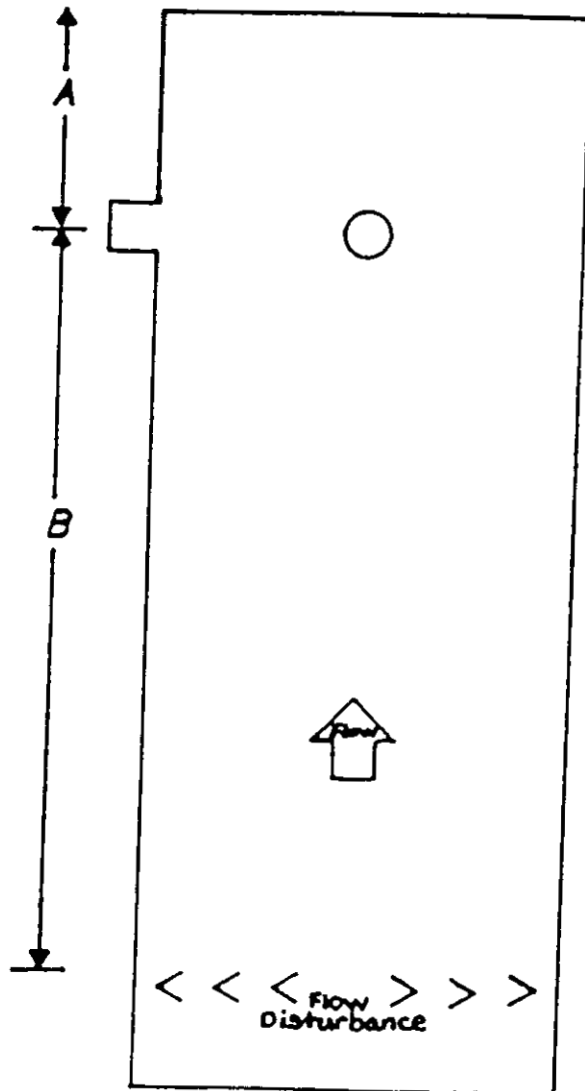


Figure 4. Location of sampling ports and traverse points.

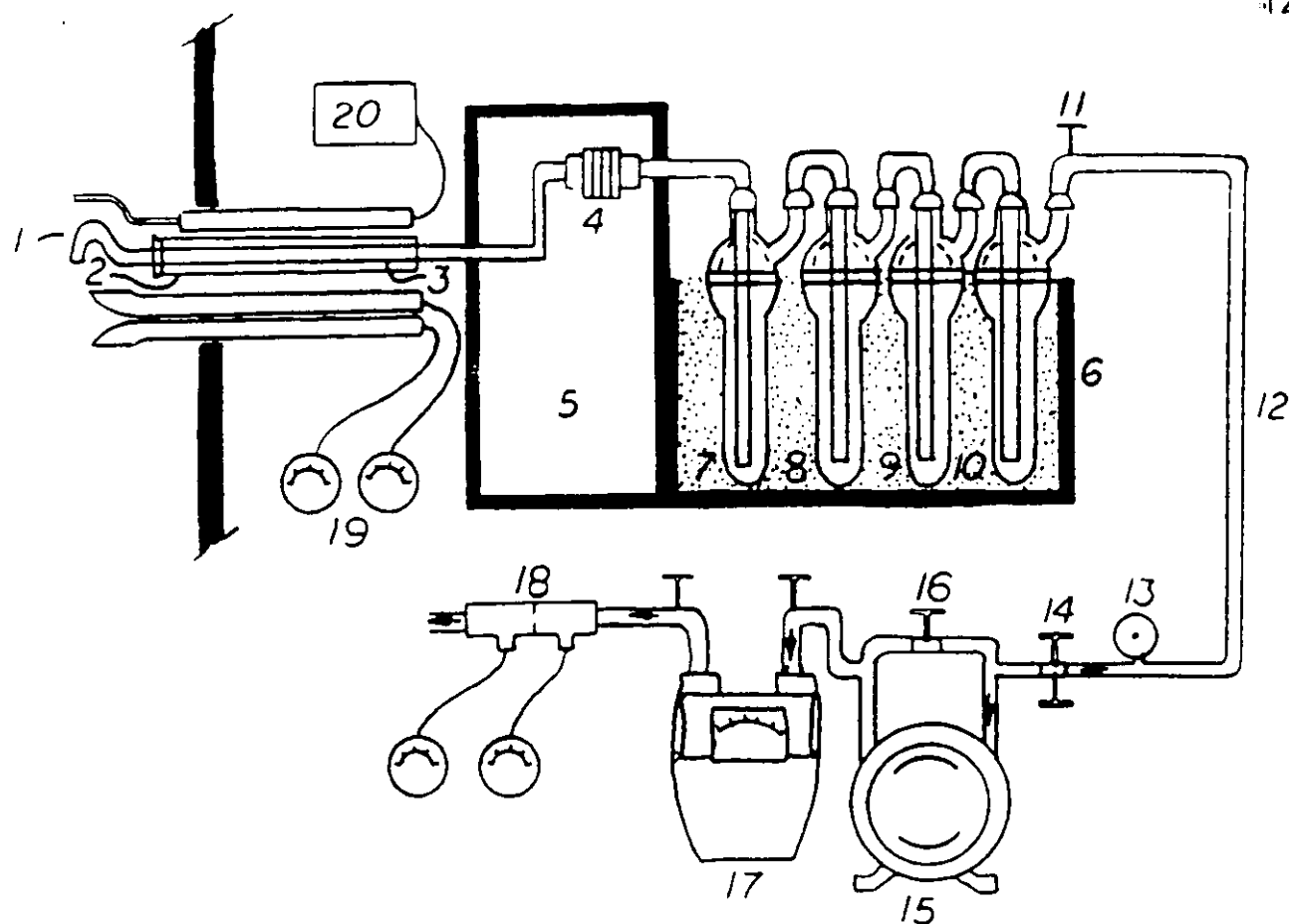
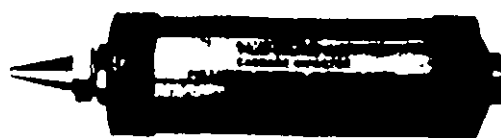


Figure 5. Method 5 Sample Train.

1. Sampling nozzle
2. Sampling probe sheath
3. Heated sample probe liner
4. Out of stack filter assembly
5. Heated filter compartment maintained at $248^{\circ}\text{F} \pm 25^{\circ}\text{F}$
(or temperature specified in 40 CFR subpart)
6. Impinger case - contains ice during sampling
7. First impinger containing 100 ml H_2O
8. Modified Greenburg-Smith impinger containing 100 ml H_2O
9. Third impinger - empty
10. Fourth impinger containing indicating silica gel desiccant
11. Impinger exit gas temperature sensor
12. Umbilical cord - vacuum line
13. Vacuum gauge
14. Fine and coarse adjustment valves
15. Leak free pump
16. By-pass valve
17. Dry gas meter with inlet and outlet temperature sensors
18. Orifice meter with magnehelic gauges
19. S-type pitot tube with magnehelic gauges
20. Fluke multi-channel digital thermocouple indicator



Mark 5 with sampling nozzle



PCSC Precutter with Nozzle on Mark 3

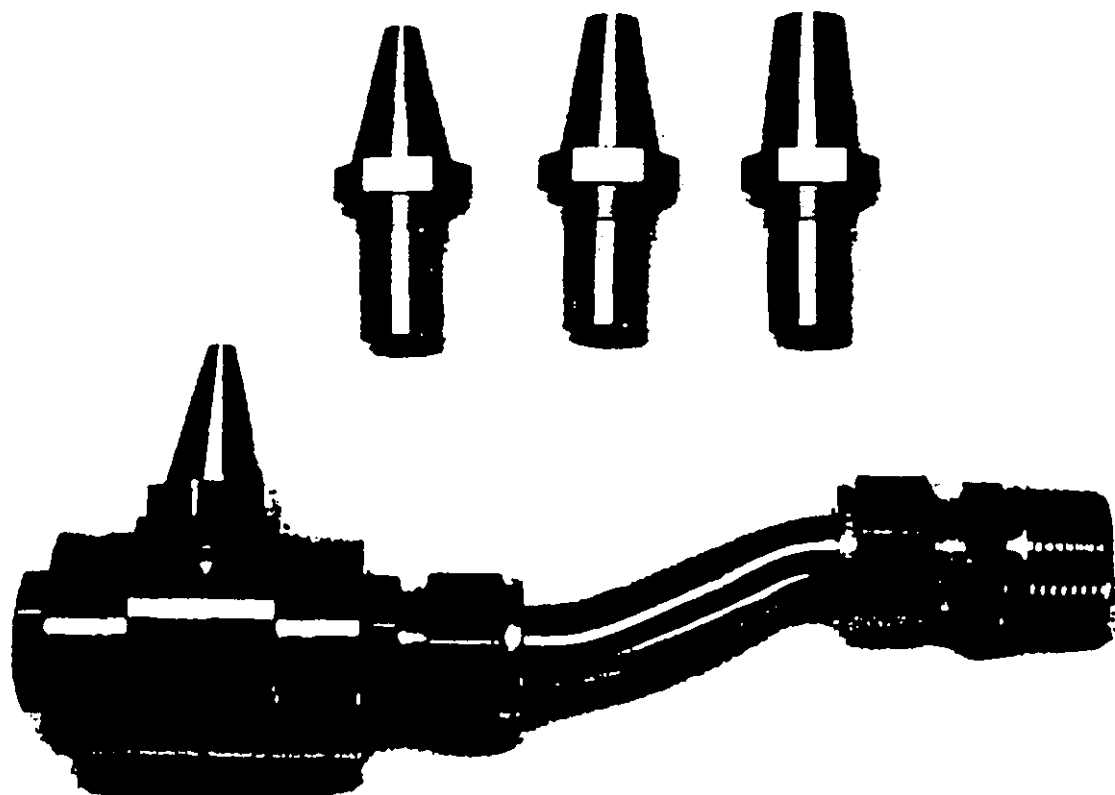


Figure 6. Illustration of Pilat (U of W) Mark 5 Cascade impactor, Mark 3 impactor equipped with a PCSC precutter attachment, and an enlargement of the PCSC precutter attachment.

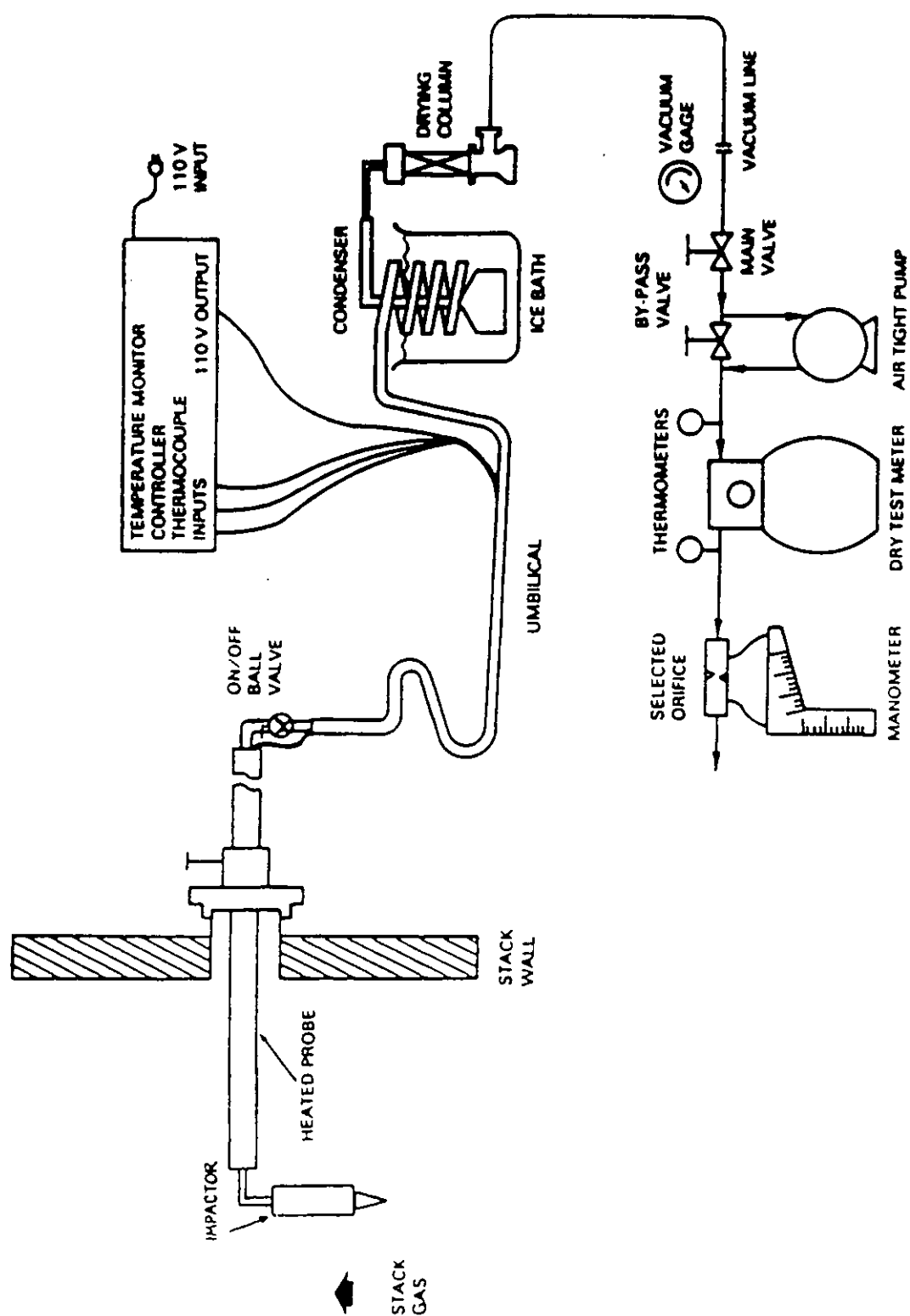


Figure 7. Cascade Impactor particulate sampling train.

METHOD 1 - LOCATION OF TRAVERSE POINTS

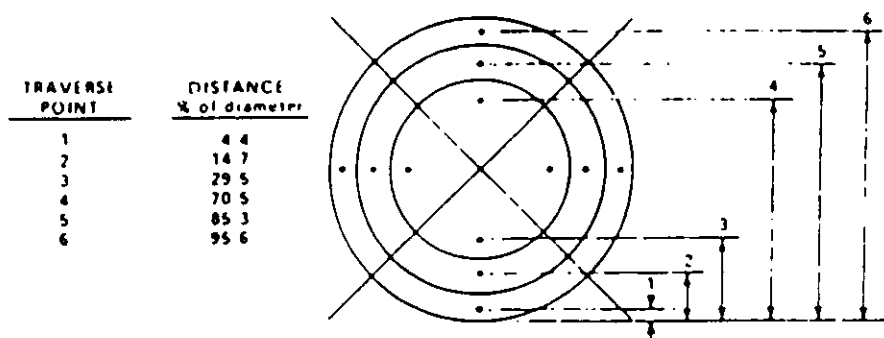
Circular Stacks

Figure 1-3. Example showing circular stack cross section divided into 12 equal areas, with location of traverse points indicated.

TABLE 1-2. LOCATION OF TRAVERSE POINTS IN CIRCULAR STACKS

(Percent of stack diameter from inside wall to traverse point)

Traverse point number on a diameter	Number of traverse points on a diameter—											
	2	4	6	8	10	12	14	16	18	20	22	24
1	14.6	6.7	4.4	3.2	2.6	2.1	1.8	1.6	1.4	1.3	1.1	1.1
2	85.4	25.0	14.6	10.5	8.2	6.7	5.7	4.9	4.4	3.9	3.5	3.2
3		75.0	29.6	19.4	14.6	11.8	9.9	8.5	7.5	6.7	6.0	5.5
4		93.3	70.4	32.3	22.6	17.7	14.6	12.5	10.9	9.7	8.7	7.9
5			85.4	67.7	34.2	25.0	20.1	16.9	14.6	12.9	11.6	10.5
6				95.6	80.6	65.8	55.6	46.9	39.8	34.1	29.5	26.1
7					89.5	77.4	64.4	53.8	45.3	38.6	33.0	29.4
8					96.8	85.4	75.0	63.4	53.5	45.0	38.4	33.6
9						91.8	82.3	73.1	62.5	54.2	46.2	40.2
10						97.4	88.2	79.9	71.7	61.8	53.5	47.2
11							93.3	85.4	78.0	70.4	61.2	53.3
12							97.9	90.1	83.1	78.4	69.4	60.7
13								94.3	87.5	81.2	75.0	68.5
14								98.2	91.5	85.4	79.6	73.8
15									95.1	89.1	83.5	78.2
16									98.4	92.5	87.1	82.0
17										95.8	90.3	85.4
18										98.6	93.3	88.4

Rectangular Stacks

For a rectangular cross section, an equivalent diameter (D_e) shall be calculated from the following equation, to determine the upstream and downstream distances:

$$D_e = \frac{2LW}{L+W}$$

where L —length and W —width.

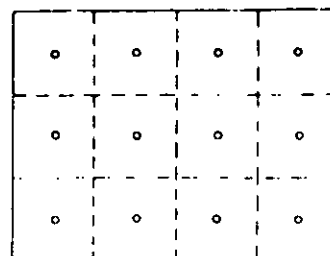
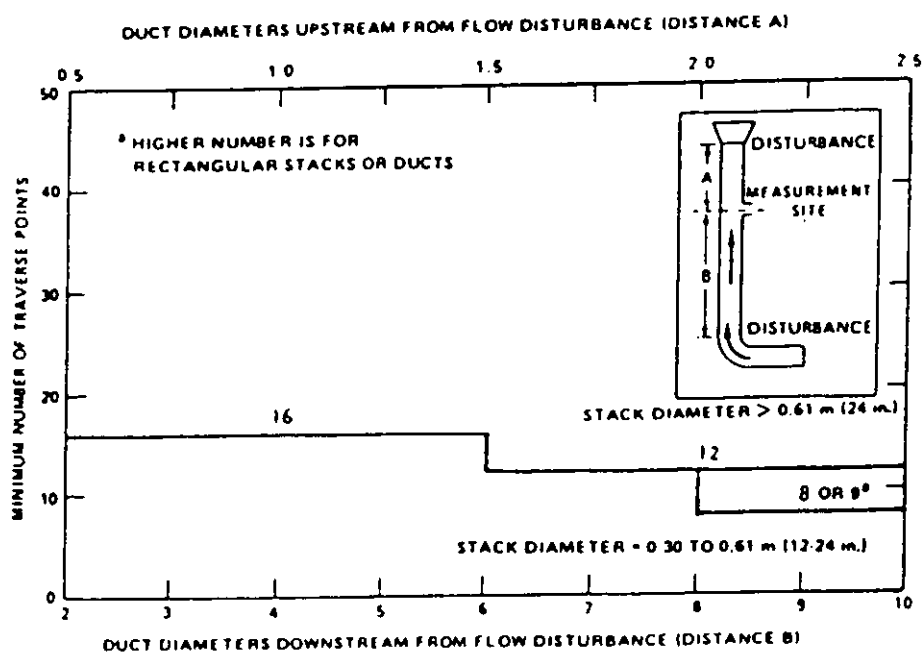
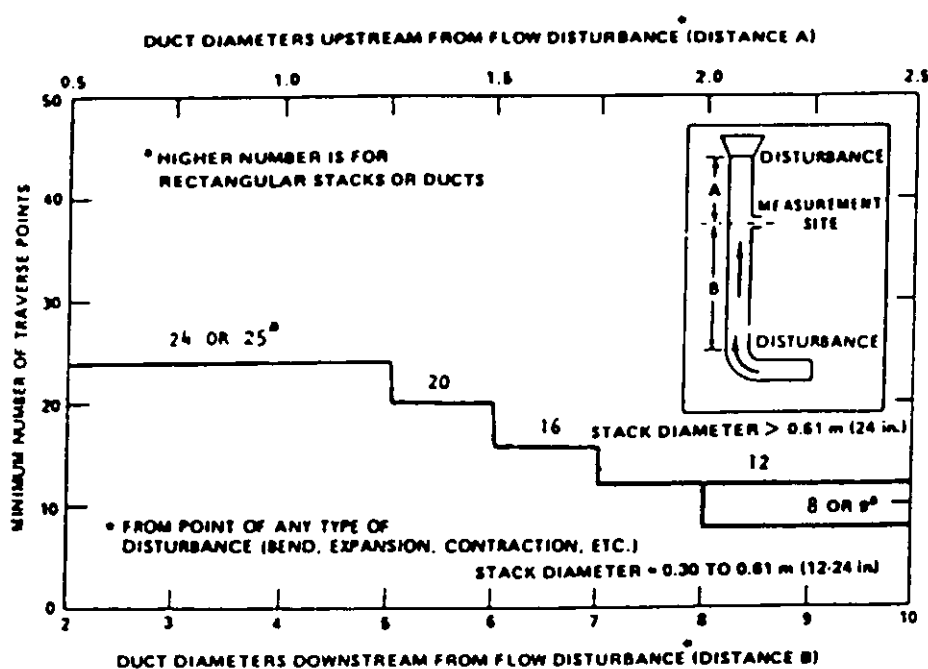


Figure 1-4. Example showing rectangular stack cross section divided into 12 equal areas, with a traverse point at centroid of each area.

METHOD 1 - MINIMUM NUMBER OF TRAVERSE POINTS



METHOD 2 - STACK GAS VELOCITY AND VOLUMETRIC FLOW CALCULATIONS

5.1 Nomenclature.

A = Cross-sectional area of stack, m^2 (ft^2).
 B_w = Water vapor in the gas stream (from Method 3 or Reference Method 4), proportion by volume.
 C_p = Pitot tube coefficient, dimensionless
 K_p = Pitot tube constant.

$$34.97 \frac{m}{sec} \left[\frac{(g/g\text{-mole})(mm\ Hg)}{(^{\circ}K)(mm\ H_2O)} \right]^{1/2}$$

for the metric system and

$$85.49 \frac{ft}{sec} \left[\frac{(lb/lb\text{-mole})(in.\ Hg)}{(^{\circ}R)(in.\ H_2O)} \right]^{1/2}$$

for the English system.

M_d = Molecular weight of stack gas, dry basis (see Section 3.6) $g/g\text{-mole}$ ($lb/lb\text{-mole}$).
 M_w = Molecular weight of stack gas, wet basis, $g/g\text{-mole}$ ($lb/lb\text{-mole}$).
 $= M_d(1 - B_w) + 18.0 B_w$

Eq. 2-5

P_{atm} = Barometric pressure at measurement site, $mm\ Hg$ ($in.\ Hg$).

P_s = Stack static pressure, $mm\ Hg$ ($in.\ Hg$).

P_t = Absolute stack gas pressure, $mm\ Hg$ ($in.\ Hg$).

$= P_{atm} + P_s$

Eq. 2-6

Eq. 2-6

P_{std} = Standard absolute pressure, $760\ mm\ Hg$ ($29.92\ in.\ Hg$).

Q_{std} = Dry volumetric stack gas flow rate corrected to standard conditions, $dscm/hr$ ($dscf/hr$).

L = Stack temperature, $^{\circ}C$ ($^{\circ}F$).

T_s = Absolute stack temperature, $^{\circ}K$, ($^{\circ}R$).

$= 273 + L$ for metric.

Eq. 2-7

$= 460 + L$ for English.

Eq. 2-8

T_{std} = Standard absolute temperature, $293\ ^{\circ}K$ ($528\ ^{\circ}R$).

u = Average stack gas velocity, m/sec (ft/sec).

Δh = Velocity head of stack gas, $mm\ H_2O$ ($in.\ H_2O$).

$3,600$ = Conversion factor, sec/hr .

18.0 = Molecular weight of water, $g/g\text{-mole}$ ($lb/lb\text{-mole}$).

5.2 Average Stack Gas Velocity.

$$u_s = K_p C_p (\sqrt{\Delta p})_{std} \sqrt{\frac{T_{std}}{P_s M_d}}$$

Equation 2-9

5.3 Average Stack Gas Dry Volumetric Flow Rate

$$Q_{std} = 3,600(1 - B_w)u_s \left(\frac{T_{std}}{T_s} \right) \left(\frac{P_s}{P_{std}} \right)$$

Eq. 2-10

METHOD 3 - MOLECULAR WEIGHT AND EXCESS AIR CALCULATIONS

6.1 Nomenclature.

M_d = Dry molecular weight, $g/g\text{-mole}$ ($lb/lb\text{-mole}$).

%EA = Percent excess air.

%CO₂ = Percent CO₂ by volume (dry basis).

%O₂ = Percent O₂ by volume (dry basis).

%CO = Percent CO by volume (dry basis).

%N₂ = Percent N₂ by volume (dry basis).

0.264 = Ratio of O₂ to N₂ in air, v/v.

0.280 = Molecular weight of N₂ or CO, divided by 100.

0.320 = Molecular weight of O₂ divided by 100.

0.440 = Molecular weight of CO, divided by 100.

6.2 Percent Excess Air. Calculate the percent excess air (if applicable), by substituting the appropriate values of percent O₂, CO, and N₂ (obtained from Section 4.1.3 or 4.2.4) into Equation 3-1.

% EA =

$$\frac{\%O_2 - 0.5\% CO}{0.264\% N_2 (\%O_2 - 0.5\% CO)} \times 100$$

Eq. 3-1

NOTE: The equation above assumes that ambient air is used as the source of O₂ and that the fuel does not contain appreciable amounts of N₂ (as do coke oven or blast furnace gases). For those cases when appreciable amounts of N₂ are present (coal, oil, and natural gas do not contain appreciable amounts of N₂) or when oxygen enrichment is used, alternate methods, subject to approval of the Administrator, are required.

6.3 Dry Molecular Weight. Use Equation 3-2 to calculate the dry molecular weight of the stack gas

$$M_d = 0.440(\%CO_2) + 0.320(\%O_2) + 0.280(\%N_2 + \%CO)$$

Eq. 3-2

METHOD 4 - STACK GAS MOISTURE CALCULATIONS

2.3.1 Nomenclature.

B_{ws} = Proportion of water vapor, by volume, in the gas stream.

M_w = Molecular weight of water, 18.0 g/g-mole (18.0 lb/lb-mole).

P_m = Absolute pressure (for this method, same as barometric pressure) at the dry gas meter, mm Hg (in. Hg).

P_{std} = Standard absolute pressure, 760 mm Hg (29.92 in. Hg).

R = Ideal gas constant, 0.06236 (mm Hg) (m³)/(g-mole) (°K) for metric units and 21.85 (in. Hg) (ft³)/(lb-mole) (°R) for English units.

T_m = Absolute temperature at meter, °K (°R).

T_{std} = Standard absolute temperature, 293°K (528°R).

V_m = Dry gas volume measured by dry gas meter, dcm (dcf).

ΔV_m = Incremental dry gas volume measured by dry gas meter at each traverse point, dcm (dcf).

$V_{m(d)} = V_m Y$ = Dry gas volume measured by the dry gas meter, corrected to standard conditions, dscm (dscf).

$V_{w(d)} = V_m Y$ = Volume of water vapor condensed corrected to standard conditions, scm (scf).

$V_{w(s)} = V_m Y$ = Volume of water vapor collected in silica gel corrected to standard conditions, scm (scf).

V_f = Final volume of condenser water, ml.

V_i = Initial volume, if any, of condenser water, ml.

W_f = Final weight of silica gel or silica gel plus impinger, g.

W_i = Initial weight of silica gel or silica gel plus impinger, g.

Y = Dry gas meter calibration factor.

ρ = Density of water, 0.9982 g/ml (0.002201 lb/ml).

2.3.2 Volume of Water Vapor Condensed.

$$V_{w(d)} = \frac{(V_f - V_i) \rho R T_{std}}{P_{std} M_w}$$

$$= K_1 (V_f - V_i)$$

Eq. 4-1

$K_1 = 0.001333 \text{ m}^3/\text{ml}$ for metric units

$= 0.04707 \text{ ft}^3/\text{ml}$ for English units

2.3.3 Volume of Water Vapor Collected in Silica Gel

$$V_{w(s)} = \frac{(W_f - W_i) R T_{std}}{P_{std} M_w}$$

$$= K_2 (W_f - W_i)$$

Eq. 4-2

Where:

$K_2 = 0.001335 \text{ m}^3/\text{g}$ for metric units

$= 0.04715 \text{ ft}^3/\text{g}$ for English units

2.3.4 Sample Gas Volume.

$$V_{m(d)} = V_m Y \frac{(P_m)(T_{std})}{(P_{std})(T_m)}$$

$$= K_3 Y \frac{V_m P_m}{T_m}$$

Eq. 4-3

Where:

$K_3 = 0.3858 \text{ } ^\circ\text{K}/\text{mm Hg}$ for metric units

$= 17.64 \text{ } ^\circ\text{R}/\text{in. Hg}$ for English units

NOTE: If the post-test leak rate (Section 2.2.6) exceeds the allowable rate, correct the value of V_m in Equation 4-3, as described in Section 6.3 of Method 5.

2.3.5 Moisture Content.

$$B_{ws} = \frac{V_{w(d)} + V_{w(s)}}{V_{w(d)} + V_{w(s)} + V_{m(d)}}$$

Eq. 4-4

NOTE: In saturated or moisture droplet-laden gas streams, two calculations of the moisture content of the stack gas shall be made, one using a value based upon the saturated conditions (see Section 1.2), and another based upon the results of the impinger analysis. The lower of these two values of B_{ws} shall be considered correct.

METHOD 5 - PARTICULATE EMISSION CALCULATIONS - (1)

6.1 Nomenclature.

- A_n = Cross-sectional area of nozzle, m^2 (ft^2).
 B_{wv} = Water vapor in the gas stream, proportion by volume.
 C_n = Acetone blank residue concentration, mg/g .
 c = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, $g/dscm$ ($g/dscf$).
 I = Percent of isokinetic sampling.
 L_m = Maximum acceptable leakage rate for either a pretest leak check or for a leak check following a component change: equal to $0.0057 m^3/min$ ($0.02 cfm$) or 4 percent of the average sampling rate, whichever is less.
 L_i = Individual leakage rate observed during the leak check conducted prior to the i^{th} component change ($i=1, 2, 3, \dots, n$), m^3/min (cfm).
 L_p = Leakage rate observed during the post-test leak check, m^3/min (cfm).
 m_w = Mass of residue of acetone after evaporation, mg .
 m_p = Total amount of particulate matter collected, mg .
 M_w = Molecular weight of water, $18.0 g/g\text{-mole}$ ($18.0 lb/lb\text{-mole}$).
 P_a = Barometric pressure at the sampling site, $mm\ Hg$ ($in.\ Hg$).
 P_s = Absolute stack gas pressure, $mm\ Hg$ ($in.\ Hg$).
 P_{std} = Standard absolute pressure, $760\ mm\ Hg$ ($29.92\ in.\ Hg$).
 R = Ideal gas constant, $0.06236\ mm\ Hg\text{-}m^3/K\text{-}g\text{-mole}$ ($21.85\ in.\ Hg\text{-}ft^3/R\text{-}lb\text{-mole}$).
 T_m = Absolute average dry gas meter temperature (see Figure 5-2), $^{\circ}K$ ($^{\circ}R$).
 T_s = Absolute average stack gas temperature (see Figure 5-2), $^{\circ}K$ ($^{\circ}R$).
 T_{std} = Standard absolute temperature, $293^{\circ}\ K$ ($528^{\circ}\ R$).
 V_n = Volume of acetone blank, ml .
 V_{wv} = Volume of acetone used in wash, ml .
 V_t = Total volume of liquid collected in impingers and silica gel (see Figure 5-3), ml .
 V_s = Volume of gas sample as measured by dry gas meter, dcm ($dscf$).
 $V_{s,cor}$ = Volume of gas sample measured by the dry gas meter, corrected to standard conditions, $dscm$ ($dscf$).
 $V_{wv,cor}$ = Volume of water vapor in the gas sample, corrected to standard conditions, scm (scf).
 v_s = Stack gas velocity, calculated by Method 2, Equation 2-8, using data obtained from Method 5, m/sec (ft/sec).
 W_w = Weight of residue in acetone wash, mg .
 Y = Dry gas meter calibration factor.
 ΔH = Average pressure differential across the orifice meter (see Figure 5-2), $mm\ H_2O$ ($in.\ H_2O$).
 ρ_w = Density of acetone, mg/ml (see label on bottle).

ρ_w = Density of water, $0.9982\ g/ml$ ($0.002201\ lb/ml$).

θ = Total sampling time, min .

θ_1 = Sampling time interval, from the beginning of a run until the first component change, min .

θ_2 = Sampling time interval, between two successive component changes, beginning with the interval between the first and second changes, min .

θ_n = Sampling time interval, from the final (n^{th}) component change until the end of the sampling run, min .

13.6 = Specific gravity of mercury.

60 = Sec/min.

100 = Conversion to percent.

6.2 Average Dry Gas Meter Temperature and Average Orifice Pressure Drop. See data sheet (Figure 5-2).

6.3 Dry Gas Volume. Correct the sample volume measured by the dry gas meter to standard conditions ($20^{\circ}\ C$, $760\ mm\ Hg$ or $68^{\circ}\ F$, $29.92\ in.\ Hg$) by using Equation 5-1.

$$V_{m(scd)} = V_m Y \left(\frac{T_{std}}{T_m} \right) \left[\frac{P_{wv} + \frac{\Delta H}{13.6}}{P_{std}} \right]$$

$$= K_1 V_m Y \frac{P_{wv} + (\Delta H/13.6)}{T_m}$$

Equation 5-1

Where:

$K_1 = 0.3858\ ^{\circ}K/mm\ Hg$ for metric units
 $= 17.64\ ^{\circ}R/in.\ Hg$ for English units

NOTE: Equation 5-1 can be used as written unless the leakage rate observed during any of the mandatory leak checks (i.e., the post-test leak check or leak checks conducted prior to component changes) exceeds L_m . If L_p or L_i exceeds L_m , Equation 5-1 must be modified as follows:

(a) Case I. No component changes made during sampling run. In this case, replace V_m in Equation 5-1 with the expression:

$$[V_m - (L_p - L_m)\theta]$$

(b) Case II. One or more component changes made during the sampling run. In this case, replace V_m in Equation 5-1 by the expression:

$$\left[V_m - (L_p - L_m)\theta_1 - \sum_{i=2}^n (L_i - L_m)\theta_i - (L_p - L_m)\theta_n \right]$$

and substitute only for those leakage rates (L_i or L_p) which exceed L_m .

6.4 Volume of Water Vapor.

METHOD 5 - PARTICULATE EMISSION CALCULATIONS - (2)

$$V_{s (std)} = V_s \left(\frac{P_s}{P_{std}} \right) \left(\frac{RT_{std}}{P_{std}} \right) = K_1 V_s \quad \text{Equation 5-2}$$

Where:

$K_1 = 0.001333 \text{ m}^3/\text{ml}$ for metric units
 $= 0.04707 \text{ ft}^3/\text{ml}$ for English units.

6.5 Moisture Content.

$$B_w = \frac{V_{w (std)}}{V_{m (std)} + V_{w (std)}}$$

Eq. 5-3

NOTE: In saturated or water droplet-laden gas streams, two calculations of the moisture content of the stack gas shall be made, one from the impinger analysis (Equation 5-3), and a second from the assumption of saturated conditions. The lower of the two values of B_w shall be considered correct. The procedure for determining the moisture content based upon assumption of saturated conditions is given in the Note of Section 1.2 of Method 4. For the purposes of this method, the average stack gas temperature from Figure 5-2 may be used to make this determination, provided that the accuracy of the in-stack temperature sensor is $\pm 1^\circ \text{C}$ (2°F).

6.6 Acetone Blank Concentration.

$$C_a = \frac{m_a}{V_a P_a} \quad \text{Eq. 5-4}$$

6.7 Acetone Wash Blank.

$$W_a = C_a V_{wsp} \quad \text{Eq. 5-5}$$

6.8 Total Particulate Weight. Determine the total particulate catch from the sum of the weights obtained from Containers 1 and 2 less the acetone blank (see Figure 5-3).

NOTE: Refer to Section 4.1.5 to assist in calculation of results involving two or more filter assemblies or two or more sampling trains.

6.9 Particulate Concentration.

$$C_p = (0.001 \text{ g/mg}) (m_p / V_{m (std)})$$

Eq. 5-6

6.10 Conversion Factors:

	From	To	Multiply by
acv		m ³	0.02832
g/n ³		g/n ³	15.43
g/n ³		kg/n ³	2.205 × 10 ⁻³
g/n ³		g/m ³	35.31

6.11 Isokinetic Variation.

6.11.1 Calculation From Raw Data

I

$$100 T_s [K_1 V_s + (P_m / T_m) (P_{std} + 511 / 13.6)] \\ 60 W_v, P, A_s$$

Eq. 5-7

Where:

$K_1 = 0.003454 \text{ mm Hg} \cdot \text{m}^3/\text{ml} \cdot ^\circ \text{K}$ for metric units.
 $= 0.002689 \text{ in. Hg} \cdot \text{ft}^3/\text{ml} \cdot ^\circ \text{R}$ for English units.

6.11.2 Calculation From Intermediate Values.

$$I = \frac{T_s V_{m (std)} P_{std} 100}{T_{std} V_s \theta A_s P_s 60 (1 - B_{ws})} \\ = K_2 \frac{T_s V_{m (std)}}{P_s V_s A_s \theta (1 - B_{ws})}$$

Equation 5-8

where:

$K_2 = 4.320$ for metric units
 $= 0.09450$ for English units.

6.12 Acceptable Results. If 90 percent < I < 110 percent, the results are acceptable. If the particulate results are low in comparison to the standard, and I is over 110 percent or less than 90 percent, the Administrator may accept the results.

NOMENCLATURE METHOD 5 CALCULATIONS

- $V_{m_{std}}$ = Volume of gas sample measured by the dry gas meter, corrected to standard conditions, dscm (dscf).
- Y = Dry gas meter calibration factor
- P_b = Barometric pressure at the sampling site, mm Hg (in. Hg)
- H = Average pressure differential across the orifice meter, mm H₂O (in. H₂O)
- T_m = Absolute average dry gas meter temperature, ° K (° R)
- dscm = Dry standard cubic meters
- dscf = Dry standard cubic feet
- W_a = Weight of residue in acetone wash
- M_a = Mass of residue of acetone after evaporation, mg
- C_a = Acetone blank residue concentration, mg/g
- V_a = Volume of acetone blank
- V_{aw} = Volume of acetone used in wash, ml
- M_n = Total amount of particulate matter collected, mg
- C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, mg/dscm (gr/dscf)
- gr/dscf = grains per dry standard cubic foot
- $V_{w_{std}}$ = Volume of water vapor in the gas sample, corrected to standard conditions, scm (scf)
- B_{w_s} = Water vapor in the gas stream, proportion by volume
- M_d = Molecular weight of stack gas, g/g-mole on dry basis
- M_w = Molecular weight of stack gas, g/g-mole on wet basis
- V_s = Stack gas velocity, calculated by Method 2, Equation 2-9, using data obtained from Method 5, m/sec (ft/sec)
- C_p = Pitot tube coefficient, dimensionless
- Δp = Velocity head of stack gas, mm H₂O (in. H₂O)
- P_s = Absolute stack gas pressure, mm Hg (in. Hg)

NOMENCLATURE (continued)
METHOD 5 CALCULATIONS

- Q_{std} = Dry volumetric stack gas flow rate corrected to standard conditions, dscm/hr (dscf/hr)
- dscf/min = dry standard cubic feet per minute (also identified as dcfm or scfm)
- acfm = actual cubic feet per minute
- I = Percent of isokinetic sampling
- A_n = Cross-sectional area of nozzle, m^2 (ft^2)

**RESUME OF
KRIS A. HANSEN****DIRECTOR - AIR QUALITY DIVISION****EDUCATION**

- B.S., Chemistry, Central Washington University, 1973
- Coursework and 2.5 years research completed towards M.S., Chemistry, Central Washington University
- Professional training courses and specialty conferences

PROFESSIONAL MEMBERSHIPS

- Air and Waste Management Association (AWMA)
- American Association for Aerosol Research (AAAR)
- Pacific Northwest International Section of AWMA (PNWIS)
- Source Evaluation Society (SES)
- American Chemical Society (ACS)

PROFESSIONAL EXPERIENCE

Am Test, Inc.'s Air Quality Division, developed and directed by Mr. Hansen, is in its eighth year of existence. Mr. Hansen conducts full-service source testing activities including particulate matter emissions testing, particle size distribution, testing for sulfur and nitrogen oxides, hydrogen sulfide, semi-volatile and volatile organic compounds (including dioxins and furans), toxic air pollutant characterization, opacity measurement, continuous emission monitor certifications and other EPA methodology. He has developed sampling and analysis techniques for many sources for which EPA guidelines are not available. Mr. Hansen has worked at all types of industrial facilities, including oil and gas refineries, gas, oil and coal-fired power plants, nuclear plants, aluminum plants, wood products industries, smelters, incinerators, and other industrial sources throughout the lower 48 states, Alaska and Canada. Mr. Hansen manages an experienced field testing and laboratory analysis staff and maintains a strict quality assurance program. Mr. Hansen manages all phases of project development, including cost estimation, scheduling, sample collection, analysis and report preparation.

Prior to joining Am Test, Inc., Mr. Hansen's professional experience included 4.5 years as the Manager of Laboratory services for an environmental consulting firm which specialized in air quality studies, and 2 years as a laboratory instructor while attending graduate school at Central Washington University, where his research emphasis was in gas chemistry. Mr. Hansen has 14 years of professional experience.

Mr. Hansen assists in the instruction of the EPA 450 "Source Sampling for Particulate Pollutants" and the EPA 468 "Source Sampling and Analysis of Gaseous Pollutants" courses offered yearly by the EPA in cooperation with the University of Washington. Students are represented by industry and governmental agency personnel from across the U.S. and other nations. Mr. Hansen was the recipient of the 1987 PNWIS/APCA "Hardhat Award" which was presented in recognition of his contributions to the advancement of source sampling technology in the Pacific Northwest.

**RESUME OF
JAMES A. GUENTHOER**

SENIOR PROJECT ENGINEER

EDUCATION

- B.S., Geology, Juniata College, Huntington, Pennsylvania, 1972
- M.S.E., Environmental Engineering Division, Department of Civil Engineering, University of Washington, Seattle, Washington, 1985
- Professional training courses and specialty conferences

PROFESSIONAL MEMBERSHIPS

- Air and Waste Management Association (AWMA)
- Pacific Northwest International Section of AWMA (PNWIS)
- Source Evaluation Society (SES)

PROFESSIONAL EXPERIENCE

Mr. Guenthoer has been a Project Engineer for Am Test, Inc. for the past 6 years and has had 12 years of professional experience. Mr. Guenthoer specializes in performing field sampling for source emission evaluations. He is experienced in the collection of samples to be analyzed for particulate matter, particle size distribution, sulfur and nitrogen oxides, hydrogen sulfide, semi-volatile and volatile organic compounds (including dioxins and furans), toxic air pollutants, and other chemical species. Mr. Guenthoer was the recipient of the 1989 PNWIS/APCA "Hardhat Award" which is presented each year to an individual to recognize his contribution to the advancement of source sampling technology in the Pacific Northwest.

Mr. Guenthoer was formerly the Operations Manager for Pollution Control Systems, Inc. of Seattle, Washington for 4 years and handled the design, technical services and sales of in-stack source test Cascade impactors. He assists in the instruction of EPA 450 and EPA 468 Source Sampling Short Courses for particulate and gaseous pollutants which are offered each year by the EPA in cooperation with the University of Washington. He has also assisted in presenting workshops to demonstrate the use of in-stack source test Cascade impactors.

Mr. Guenthoer was formerly associated with Air Pollution Systems of Kent, Washington conducting research and development studies on novel particulate control technology. Prior to his association with Air Pollution Systems, Mr. Guenthoer was the Testing Manager for Rossnagel and Associates of Medford, New Jersey and was in charge of all water quality, industrial hygiene, and source testing for 3 branch offices. During his 5 years with this environmental testing firm, Mr. Guenthoer conducted compliance source tests for local, state and federal government in over 20 states and served as a professional expert witness in environmental legal cases. Other professional experience with the State of Washington's Department of Ecology involved surface water quality studies. Mr. Guenthoer was also involved in statewide limnological investigations and bathymetric mapping over a period of 2 years with the United States Geological Survey in Tacoma, Washington.

**RESUME OF
K. STEVEN MACKEY**

SENIOR AIR QUALITY SPECIALIST

EDUCATION

- A.S., Mathematics, North Idaho College, Coeur d'Alene, Idaho, 1978
- Professional training courses and specialty conferences

PROFESSIONAL MEMBERSHIPS

- Source Evaluation Society (SES)

PROFESSIONAL EXPERIENCE

Mr. Mackey is a Senior Air Quality Specialist for Am Test Inc.'s Air Quality Division, and has ten (10) years experience in the field of air quality and industrial source emissions testing. Mr. Mackey conducts source testing and activities related to source emission testing, including field sampling, test equipment maintenance and calibration, test planning and preparation, and data reduction and evaluation. He has performed sampling for particulates, particle size distribution, sulfur and nitrogen oxides, reduced sulfur compounds, semi-volatile and volatile organic compounds, toxic air pollutants, opacity, and other EPA methodology.

Prior to joining Am Test, Inc., Mr. Mackey was the Source Testing Coordinator for the State of Oregon Department of Environmental Quality (ODEQ) for 1.5 years. This position was responsible for coordinating a statewide source sampling program and source quality assurance program for industrial air pollution sources. His work with ODEQ's Air Quality Division involved evaluating source tests, plant site source test observations, approving source test methods, and evaluating source self-monitoring practices.

Prior to his position with the State of Oregon, Mr. Mackey was a Field Supervisor for Omni Environmental Services in Beaverton, Oregon for 4 years. Experience with this firm was primarily in industrial source emission testing and field and laboratory woodstove emissions testing. In addition to source testing at a wide variety of industrial facilities, Mr. Mackey ran an extensive field study on woodstove performance in Vermont and New York involving specially designed sampling equipment. Prior to working with Omni, Mr. Mackey worked 4.5 years as a Project Leader and Field Technician for an environmental consulting firm which specialized in air quality studies. Experience with this firm involved extensive source testing and other emissions monitoring projects.

**RESUME OF
ANGELA F. BLAISDELL**

ENVIRONMENTAL CHEMIST/TECHNICAL WRITER

EDUCATION

- B.S., Marine Resources, Western Washington University, Bellingham, Washington, 1980
- Minors in Chemistry and Biology
- Professional training courses and specialty conferences

PROFESSIONAL MEMBERSHIPS

- Air and Waste Management Association (AWMA)
- Pacific Northwest International Section of AWMA (PNWIS)

PROFESSIONAL EXPERIENCE

Ms. Blaisdell has worked with Am Test, Inc.'s Air Quality Division for the past 5 years. Her main responsibility is to assure that the administrative functions of the Air Quality Division are in order. She communicates with clients by telephone, drafts all letters, proposals, test plans, reports, and invoices for review by the Director. She reviews current literature for each test method and incorporates the methodology into our testing and reporting protocol.

In addition to her administrative responsibilities, Ms. Blaisdell oversees the field technicians in pre-field preparation activities for Air Quality Division projects, and assists in clean-up and analysis of samples collected by the test crew. On larger projects she travels with the test team to assist in sample collection, in-field analysis and data reduction in Am Test's mobile laboratory. Ms. Blaisdell has performed analysis of particulate matter and particle size distribution samples, sulfur oxides (SO_2 and SO_3), nitrogen oxides (NO_x), hydrogen sulfide (H_2S), lead, metals, urea, ammonia, formaldehyde, arsenic, semi-volatile and volatile organic compounds, and other chemical species collected in air. The data collected is then transferred to computerized data reduction programs, which Ms. Blaisdell helps write and tailors for each project.

Prior to joining Am Test, Ms. Blaisdell acted as a Project Leader and Office Manager for an environmental engineering consulting firm in the Seattle area for 2 years. Experience with that firm involved sample collection, analysis and report preparation for source and ambient air, water and industrial hygiene studies. Other professional experience includes a year long project with the Institute of Freshwater Studies in Bellingham, Washington which involved weekly sampling by boat and subsequent laboratory analysis of Lake Whatcom's water for various water quality parameters. She was also an Office and Property Manager for a property management firm, where responsibilities included word processing, computer program development, accounting duties and various executive secretarial duties. She has 9 years of professional experience. Ms. Blaisdell also worked on various research projects in the Chemistry department while attending Western Washington University.

**RESUME OF
JAN M. WIDMEYER**

PROJECT ASSISTANT

EDUCATION

- B.S., Marine Resources, Western Washington University, Bellingham, Washington, 1980
- Minor in Biology
- Professional training courses and specialty conferences

PROFESSIONAL EXPERIENCE

Ms. Widmeyer has been with Am Test, Inc.'s Air Quality Division for the past 3 years. She is responsible for pre-field preparation for Air Quality Division projects and prepares the sample trains, glassware, labware and sampling hardware to be used on each specific project. In the field, Ms. Widmeyer acts as a Project Assistant, assisting the Project Manager, Project Engineer, or Project Leader in setting up the equipment at each sample site, performing the tests, recovering the samples after each test, and in demobilizing. She is responsible for properly labeling and identifying each sample, and initiates chain-of-custody procedures.

In the laboratory, Ms. Widmeyer is experienced in gravimetric analysis of particulate matter samples and particle size distribution samples. She assists in preparation and analysis of samples for quantifying sulfur oxides, nitrogen oxides, hydrogen sulfide, lead, toxic metals (including hexavalent chromium), semi-volatile compounds (including dioxins and furans), and volatile organic compounds.

In the office, Ms. Widmeyer reduces the field data and inputs values into data reduction programs and performs by-hand calculations to verify computer program integrity. She assists in technical writing and word processing of proposals, test plans, and reports. Ms. Widmeyer also performs many administrative duties to keep the Air Quality Division office organized and operating efficiently.

AM TEST LABORATORIES, INC.

AIR QUALITY DIVISION

Am Test, Inc.'s Air Quality Division has full-service source testing capabilities. An experienced staff, utilizing state-of-the-art equipment conducts field testing and laboratory analyses. Some of the types of tests we perform include measurements of particulate matter, particle size distribution (including PM_{10}), acid gas emissions, metals emissions, sulfur and nitrogen oxides, hydrogen sulfide and total reduced sulfur, volatile organic compounds (VOC), and semi-volatile organic compounds (including dioxins, furans, PCB, and PAH). We perform continuous emission monitoring system (CEMS) performance evaluations and relative accuracy test audits (RATA) for our clients using CEMS to demonstrate compliance. The Air Quality Division of Am Test provides testing services to a variety of industrial clients including: oil and gas refineries; gas, oil, coal, and nuclear-fired power plants; chemical plants; wood products industries; smelters; cement plants; landfills; and municipal waste incinerators.

Services provided by the Air Quality Division of Am Test include:

- Source Industrial Atmospheric Emission Evaluations
- Toxic Pollutant Emissions Studies
- Particle Size Distribution Studies (PM_{10})
- Continuous Emission Monitoring System (CEMS) Evaluations
- Vapor Recovery System Efficiency Evaluations
- Pilot Plant Studies
- Engineering and Control Technology Evaluations
- Ambient Air Quality Evaluations
- Industrial Hygiene Studies
- SARA Title III Emissions Measurements
- Mobile Source Test Laboratory includes
"real time" gaseous constituent analyzers,
capability of in-field analysis and data reduction

Phone Mr. Kris A. Hansen, Director, Air Quality Division, Am Test,
at (206)885-1664 or (206)333-4054 for more details, or a firm price quotation.

AM TEST LABORATORY

Am Test, Inc. is a full service analytical testing laboratory located in Redmond, Washington. It was organized to provide the highest caliber laboratory testing of environmental and industrial samples. Experienced environmental chemists and microbiologists have the best available equipment at their disposal. Experienced personnel, top-notch instrumentation, and a personal interest in your testing requirements assures:

- Quick turnaround time for sample analysis
- Maximum quality control on projects
- Direct dialogue with the analytical staff
- Reasonable rates

Am Test's laboratory is departmentalized into the following disciplines:

**AIR QUALITY
ENVIRONMENTAL
INDUSTRIAL**

OIL AND FUELS

TRACE ORGANICS

MICROBIOLOGY

**Source and Ambient Testing
Water, Wastes, Tissue, Vegetation
Food, Materials Testing, Industrial
Hygiene, Special Projects
Lubricating Oil, Fuel Analysis,
Hydraulics, Contamination Analysis
PCB's, Pesticides, Hazardous Wastes,
Priority Pollutants
Water, Wells, Food, Product Evaluation**

Am Test's laboratory utilizes the following instrumentation:

Finnigan Inco 50 GC/MS
Jarrell Ash ICP Plasma Emission Spectrometer
Gas Chromatographs: FID, EC, TC, N₂, P
Jarrell Ash Arc, Spark Emission Spectrometer
Xertex TOX Analyser
Source Test Equipment-EPA Approved

Atomic Absorption Spectrometer-Flame
Atomic Absorption Spectrometer-Graphite Furnace
Atomic Absorption Spectrometer-Hydride Generation
UV/Visible Spectrometer
Infrared Spectrometer
High Performance Liquid Chromatograph

Am Test is one laboratory in a group of laboratories serving the West. Other facilities include:

Amtest, Inc., Portland, Oregon
Amtest of Arizona, Phoenix, Arizona
Can Test Ltd., Vancouver, B.C.
Loring Laboratories, Calgary, Alberta
Metropolitan Clinical Laboratories, Ltd., Vancouver, B.C.

Phone Mr. Shawn Moore, General Manager or Mr. Mark Fugiel, Technical Director, at (206)885-1664 for more details or a price quotation.

AM TEST, INC. - AIR QUALITY DIVISION

QUALITY ASSURANCE PLAN

Introduction

The purpose of the quality assurance plan is to provide guidelines for achieving quality control in air pollution measurements. The detailed procedures to be utilized are included in the Environmental Protection Agency's (EPA's) reference manual titled Quality Assurance Handbook for Air Pollution Measurements Systems, Volume 3, EPA-600/4-77-027b, along with current updates. The accuracy of emissions test data depends upon proper equipment performance and on the proficiency and conscientiousness of the operators. All aspects of testing, analysis, data reduction and report preparation should be performed by experienced and well-trained personnel.

Methods

Am Test, Inc.'s Air Quality Division typically uses published test methods; e.g., New Source Performance Standards (NSPS) test methods published in the current edition of Title 40 Code of Federal Regulations, Part 60 (40 CFR 60), and the methods adopted for state and local code enforcement. Specialized test work, such as particle size testing, requires special test methods. Am Test uses those methodologies and sampling hardware which are considered by the experts in the field to be state-of-the-art. Procedures and protocol are updated regularly as research groups refine the testing methodology, analytical procedures and calibration techniques. A high level of quality control is maintained in both our standard NSPS testing and specialty testing areas.

Chain of Custody

Samples are recovered from the sampling train after each test. Sample recovery is carried out in a suitable area sheltered from wind and dust to prevent contamination of samples. Sample recovery procedures outlined in the applicable test method are followed in detail (usually 40 CFR 60). All sample containers are identified using a sample label. After recovery, all sample containers are sealed. Most sample recovery and analysis is performed in the field to prevent sample loss or contamination during shipment and to provide the client with preliminary results. The data is then input into the field computer and hard copy, disc copy and back-up are produced. Data relative to samples, collected during each test, are immediately inspected for completeness and placed under the custody of the Project Manager or Project Engineer until custody is transferred when the samples are turned over to the laboratory. Sample storage boxes with locks are used to store samples in the field prior to in-field analysis. A chain of custody sheet follows each sample through the above procedures. The samples are then turned over to the Am Test Laboratory or subcontract laboratory if additional analysis is to be performed. The chain of custody form must be signed each time the samples change hands to couriers or laboratory personnel.

Quality Control

Reliability in stationary source testing is maintained through strict adherence to quality control procedures. Am Test's procedures are designed to control both the accuracy and precision of the test results.

Accuracy is maintained through rigorous calibration procedures using the standards specified in the methods and/or Volume III of the Quality Assurance Handbook for Air Pollution Measurement Systems, use of control samples where appropriate, and performance and systems audits. The accuracy of source test capabilities is monitored through voluntary participation in EPA's stationary source audits for coal and Methods 5, 6 and 7.

Performance audits are conducted during the sampling and analytical phases of the test program. The volumetric flow metering device is audited during the sampling phase using a standard dry gas meter. This audit is conducted once on each metering device during each set of field tests, using procedures outlined in Volume III of the Quality Assurance Handbook. Barometers, thermocouple indicators, nozzles, and other equipment used during the field test phase are audited periodically to ensure the collection of acceptable data.

Blanks of reagents used to collect, recover, and analyze samples are collected to check the quality and ensure that reagents meet criteria established in the test methods. Filter blanks and substrate blanks are also carried through analysis procedures. When analyses are conducted in the field, blank samples are analyzed to ensure the accuracy of results. Samples analyzed in the laboratory are subject to the stringent quality controls exercised in the laboratory (see Am Test, Inc.'s Laboratory Quality Assurance/Quality Control Plan).

A systems audit is performed, consisting of an on-site qualitative inspection and review of the total measurement system. This inspection is conducted on a daily basis by the quality assurance coordinator. During the systems audit, the auditor observes the procedures and techniques of the field team in the following general areas:

- Setting up and leak testing the sampling train
- Isokinetic sampling check of the sampling train
- Final leak check of train
- Sample recovery

Results of the systems audit are summarized on the field data sheets and discussed in the final report.

Acceptable precision is maintained through strict adherence to acceptable limits of difference in replicate measurements at each step of the procedure from initial calibration of sampling equipment to the final analytical determinations. These limits are specified in the methods and in Volume III of the Quality Assurance Handbook.

Data Validation

Data validations are accomplished by using internal quality control checks and performing internal systems, performance, and data audits. For each major measurement parameter, the frequency and type of quality control checks with control limits and corrective action are established. An example of an internal quality control check would be to analyze a standard solution after every tenth analysis for projects requiring a large number of repetitious analyses. Audit samples are used in projects where appropriate. This applies most specifically to chemical analyses (i.e., Methods 6, 7). If the control sample analytical results are not within the control limits, corrective action is taken to identify and resolve the problem before continuing with analyses.

Field calibration checks are performed on the control box (using a standard dry gas meter), thermocouples, and digital readouts. Visual inspections of pitot tubes, glassware, and other equipment are also made. The main purpose of a systems audit is to ensure that the measurement system will generate valid data if operated properly. By performing pre-test, on-site, and post-test calibrations of the measurement systems, data validation checks of the performance of the test equipment can be easily performed.

Data reduction and reporting have been shown to be significant potential sources of system error. Most of Am Test's calculations are performed by a validated computer program to minimize error. Data entry is performed by individuals familiar with testing procedures. The data printouts are then validated by comparison with the field and analytical data sheets. In addition, hand calculation checks are made to validate the computer output.

All data generated by each phase of a laboratory or field sampling program are reviewed by a senior reviewer. The data must be signed off by the senior reviewer prior to releasing the data for report preparation.

Calibration

All sample train components requiring calibration are calibrated at the recommended intervals using approved methods. Standardized calibration procedures are in accordance with calibration criteria outlined in the test method and Volume III of the Quality Assurance Handbook. The following sample train components are calibrated using standardized procedures:

- Pitot tubes
- Differential pressure gauge and magnelic gauges
- Barometer
- Rotameter (rate meter)
- Digital indicators
- Thermocouples
- Meter box dry gas meters
- Sampling nozzles

Calibration procedures for special equipment and components such as the analytical system used in particle size determinations are calibrated using procedures tailored to the specific system based on accepted methods published in the literature, or manufacturer's specifications.

The general calibration program consists of conducting a full-scale laboratory calibration procedure on each new or repaired component requiring calibration, prior to its use in the field. At the completion of a field test, each component requiring calibration receives a post-test calibration check. Any component not meeting the calibration criteria is thoroughly checked and calibrated using the full-scale laboratory procedure.

For particle size impactor calibration, each impactor stage has been calibrated by the manufacturer using a polydispersed aerosol and optical size analyzer at a range of temperatures and flow rates to determine the inertial impaction parameter for that particular stage. This parameter is intrinsic for each stage, being controlled by the physical configuration of the stage and other pertinent aerodynamic values. The stages are periodically examined for wear and corrosion and replaced as necessary.

Preventative Maintenance

All equipment used in emission measuring systems is maintained in good operating order. In order to achieve this objective, an effective preventative maintenance program is necessary. Am Test's preventative maintenance program consists of 3 major components: (1) short interval inspection; (2) replacement of obsolete or damaged components; and (3) scheduled disassembly and overhaul. Procedures used in this program follow those outlined in Volume III of the Quality Assurance Handbook.

Training

The level of training of the personnel performing their various tasks on a source testing project is as important as any other aspect of a quality assurance program. Only those persons who have had extensive experience are able to judge the credibility of a source test or to recognize problems as they arise. Only personnel who are sufficiently trained to perform the source tests are utilized.

AM TEST, INC. LABORATORY QUALITY ASSURANCE PROGRAM

INTRODUCTION

Am Test, Inc. provides a comprehensive program of laboratory services in the areas of environmental, industrial, microbiological, oils, and air quality testing. Conclusions based on data generated by Am Test may have a significant impact on the assessment of environmental quality and workplace safety, as well as on the economic burden placed on industry and the public. Therefore, the establishment of a continuing program to insure the reliability and validity of results is one of the fundamental responsibilities of the laboratory.

This Quality Assurance (QA) Program outlines the policies, organization and operations established for assuring the integrity of the analytical data Am Test produces and uses.

ANALYTICAL METHODS

Sample analyses are performed in accordance with widely accepted procedures such as those of the U.S. Environmental Protection Agency (EPA), the U.S. Geological Survey (USGS), the American Society for Testing and Materials (ASTM), the Association of Official Analytical Chemists (AOAC), and the U.S. Food and Drug Administration (FDA).

DOCUMENTATION

Field sampling performed by laboratory personnel is documented with specific sample identification, date and initials of the individual performing the sampling. This information (or any information provided by the client when they perform their own sampling) accompanies the sample and is eventually included with the project record.

Upon receipt at the laboratory, the sample is given a unique (specific) identifying number from the log book and the date, client identification, and parameters required are recorded next to the number and initialed by the person receiving the sample. A project record is initiated during the log-in procedure and includes the date received, laboratory sample numbers, client identification and analysis parameters required. The sample is then preserved, split and stored as required by the client to maintain the viability of the parameters being analyzed. Special storage facilities are available for samples requiring security storage.

As analyses are performed, all data is recorded and initialed in bound, dated, and specifically labeled laboratory notebooks which are kept on permanent file. Notebooks include quality control data, analytical calculations, and results for the specific analyses. Sample analysis records are cross-referenced between books where necessary. The results are also recorded on the project record as they are obtained.

When a project is completed, a report is generated. This report includes methods summaries, results, raw data, graphs and/or quality control information, where applicable. Copies are kept on file with the project record. Information is filed in three separate filing systems: a quick-reference final report file, the project record and final report file, and raw data laboratory notebooks. The samples are retained for 30 days after the final report has been received by the client. Longer storage times may be obtained upon request.

SUPPLIES

Reagents - All reagents, chemicals, and gases used are of the purity required for the specific analyses. Each chemical is dated when received, and when opened. Chemicals are stored compatibly and are isolated where needed to insure stability and to avoid contamination and hazardous environments.

Standards - All primary standards are obtained from reliable sources (i.e., EPA, FDA, chemical suppliers), handled, prepared, and stored appropriately. Standards are prepared and standardized against primary standards from alternate sources. Primary, stock, and working standards are checked initially and regularly for accuracy. These solutions are restandardized or prepared fresh as required by their stability, or when inaccuracies fall outside set control limits. All standardized solutions are labeled with the name of the solution, concentration, solvent, preservative, date, and initials of the individual preparing the solution.

Water - Deionized or distilled water is used in the laboratory for dilution, preparation of reagent solutions, and final rinsing of glassware. Conductivity of the deionized water is monitored daily to insure good quality. Possible contaminants in the deionized water are analyzed for on a monthly basis.

Glassware - Volumetric glassware, Class A, is used in all precise measurements of volume. Glassware for general use in the laboratory is washed with detergent, rinsed with tap water, and finally rinsed twice with deionized water. Any acid or solvent rinsing required prior to analysis is performed. Glassware for individual analyses are identified and stored separately.

INSTRUMENTATION

All instruments are adjusted and calibrated to the manufacturer's specifications. Once the instrument is optimized, and is performing at the desired sensitivity, the sample set is analyzed. The standard curve serves as an additional check on instrument function.

Maintenance is performed at intervals specified by the manufacturer, or more frequently when necessary. The majority of Am Test's instruments are covered under maintenance service contracts with the manufacturer. Performance and maintenance records are maintained for each instrument.

AM TEST, INC. LABORATORY QUALITY CONTROL PROGRAM

INTRODUCTION

Quality Control (QC) consists of specific activities and procedures designed to measure and control the quality of the data being produced. Am Test uses a systematic attempt to assure the precision and accuracy of analyses by detecting errors and preventing their recurrence or by measuring the degree of error inherent in the methods applied. Confidence in the accuracy of analytical results and improvements in analytical precision is established by the identification of the determinate sources of error. Precision is governed by the indeterminate errors inherent in the procedure and can be monitored by statistical methods.

Am Test uses any or all of the following quality control techniques to assure accurate, precise, and unbiased analytical data.

PURITY

Blanks are carried with the samples through the entire analytical procedure to serve as a check on reagent purity and possible laboratory contamination. If a blank is found to be contaminated, steps are taken to identify the source and eliminate or minimize the contamination which was detected. Sample values are not quantitated until the blank value is reduced to an acceptable level. The magnitude of the blank is taken into account when calculating the results of the analysis.

ACCURACY

Accuracy refers to the relationship of the analytical value to the actual value for a standard or well-defined control sample, expressed as a percentage. In applicable tests, standards obtained from reliable sources (i.e., NBS, EPA) are run through the analytical procedure to serve as a check on the accuracy of the final analysis. Where standards are not available, laboratory blanks are "spiked" with a known quantity of the parameter being analyzed for. If the values obtained for the standard or spiked sample are not within a statistically acceptable range of the actual value, the procedure is reworked to identify and correct the problem.

PRECISION

Precision is the degree of agreement of repeated measurements of the same property, expressed in terms of dispersion of test results about the mean result obtained by repetitive testing of a homogeneous sample under specific conditions. Replicate analyses of samples and standards are performed in conjunction with normal sample procedures to monitor the precision obtainable for that particular analytical method. If the percentage variation is not within the acceptable limits determined for the specific test, the procedure is reworked to make the precision acceptable.

RECOVERY

"Spiked" samples or NBS certified matrix standards are run with the sample analyses to determine recovery data and matrix effects.

STANDARD CURVES

With each analysis, a minimum of three standards are run in conjunction with the samples to provide a working standard curve. The samples are quantitated using values obtained from their concurrently developed standard curve. The standard curve also functions as a check on the proper operation of the analytical instrument, the viability of the procedure for that parameter, and as a method for determining the linear range of the analysis. Samples are diluted or concentrated when necessary to fall within the limits of the standard curve.