

Note: This is a reference cited in *AP 42, Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

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DISTRIBUTION:

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BAY AREA
AIR QUALITY MANAGEMENT DISTRICT
939 Ellis Street
San Francisco, California 94109
(415) 771-6000

SUMMARY OF SOURCE TEST RESULTS

Report No. 9AP-42 Section 12.20

Reference

Report Sect. 4Reference 73Test Date: 12

Test Times:

Run A: 1018-1114Run B: 1143-1239Run C: 1331-1427

SOURCE INFORMATION

BAAQMD REPRESENTATIVES

Firm Name and Address

FRANCIS PLATING OF OAKLAND
785 7th Street
Oakland, CA 94607

Firm Representative and Title

R
No process, AOC
description: emissions

Phone No. (4)

Source:

c below detection limit

Source Test Engineers

C. McClure/G. Cowell/H. Doi

Services Division / Enforcement Division

N. Lew

Permit Conditions

None

Plant No.

CANNOT USE

Operates 8

quested by:

D. Hall

Operating Parameters:

Both chrome plating tanks operated continually during testing.

Applicable Regulations: 11-8-301.1

VN Recommended: NO

Source Test Results and Comments:

METHOD: TEST	RUN A	RUN B	RUN C	AVERAGE	LIMIT
ST-17 Volume Flowrate, SDCFM				12600	
Stack Temperature, °F.....	55	58	58	57	
* Water Content, Volume %				2	
ST-35 Total DC Amps				3000	
Hexavalent Chromium, mg/amp-hr.....	<.11	<.09	<.09	<.10	.15
Hexavalent Chromium, g/hr.....	<.33	<.26	<.27	<.29	
Hexavalent Chromium, mg/m3.....	<.016	<.012	<.012	<.013	
Isokinetic Ratio, act/theo.....	.94	.99	.96		
ARB 425 Hexavalent Chromium, mg/amp-hr.....	<.15	<.12	<.12	<.13	<.15
Isokinetic Ratio, act/theo.....	.98	.98	.98		

* Determined psychrometrically

NO COMMERCIAL USE OF THESE RESULTS IS AUTHORIZED

Source Test Team Leader

Date

C. McClure
C. McClure

Senior Source Test Engineer

Date

K. Kunaniec
K. Kunaniec

Approved by Source Test Manager

Date

G. Karels
G. Karels

DISTRIBUTION:

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Source Inventory
Planning
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BAY AREA
AIR QUALITY MANAGEMENT DISTRICT
939 Ellis Street
San Francisco, California 94109
(415) 771-6000

SUMMARY OF SOURCE TEST RESULTS

Report No. 9 AP-42 Section 12.2a
Reference
Report Sect. 4
Reference 73
Test Date: 12
Test Times:
Run A: 1018-1114
Run B: 1143-1239
Run C: 1331-1427

SOURCE INFORMATION		BAAQMD REPRESENTATIVES
Firm Name and Address FRANCIS PLATING OF OAKLAND 785 7th Street Oakland, CA 94607	Firm Representative and Title Randall Lewis, General Manager Phone No. (415) 444-5535	Source Test Engineers C. McClure/G. Cowell/H. Doi
Permit Conditions None	Source: Chrome Plating Tanks S-2, S-3 Plant No. 2324 Operates 8 hr/day & 260 days/yr	Permit Services Division / Enforcement Division N. Lew Test Requested by: D. Hall
Operating Parameters: Both chrome plating tanks operated continually during testing.		
Applicable Regulations: 11-8-301.1		VN Recommended: NO

Source Test Results and Comments:

METHOD: TEST	RUN A	RUN B	RUN C	AVERAGE	LIMIT
ST-17 Volume Flowrate, SDCFM				12600	
Stack Temperature, °F.....	55	58	58	57	
* Water Content, Volume %				2	
ST-35 Total DC Amps				3000	
Hexavalent Chromium, mg/amp-hr.....	<.11	<.09	<.09	<.10	.15
Hexavalent Chromium, g/hr.....	<.33	<.26	<.27	<.29	
Hexavalent Chromium, mg/m3.....	<.016	<.012	<.012	<.013	
Isokinetic Ratio, act/theo.....	.94	.99	.96		
ARB 425 Hexavalent Chromium, mg/amp-hr.....	<.15	<.12	<.12	<.13	<.15
Isokinetic Ratio, act/theo.....	.98	.98	.98		

* Determined psychrometrically

NO COMMERCIAL USE OF THESE RESULTS IS AUTHORIZED

Source Test Team Leader <i>C. McClure</i> C. McClure	Date <i>2/7/90</i>	Senior Source Test Engineer <i>K. Kuraniec</i> K. Kuraniec	Date <i>2/7/90</i>	Approved by Source Test Manager <i>G. Karels</i> G. Karels	Date <i>2/15/90</i>
--	-----------------------	--	-----------------------	--	------------------------

No process, APCD
description & emissions
below detection limit.

CANNOT USE

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATINA OF OAKLAND

LAB # 89-1342

ADDRESS 785- 7TH ST.

S.T. #

CITY OAKLAND

ZIP 94607

BOOK

TEL (415) 444-5535

PLT # 2324

PAGE

SOURCE OPERATION S-2, S-3 CHROME TANKS

FILM

SAMPLE DESCRIPTION 1 BOTTLE LABELED PROBE AND NOZZLE RINSE

MANUFACTURER'S NAME

J. McCURE

PERSON SUBMITTING SAMPLE

12 / 22 / 89

DATE

W. Dymy

SIGNATURE OF ANALYST

1-2-90

DATE

LAB REF # 09010-50

LAB METHOD MOP-34

Probe and Nozzle Rinse

Total Cr +6 < 8 ug

Total Cr < 10 ug

V.N. RECOMMENDED

☐ YES

REG. _____ RULE _____

SEC. _____

☐ NO

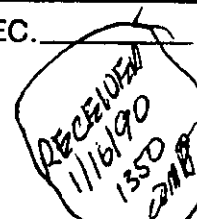
☐ FURTHER EVALUATION NEEDED

[Signature]

LABORATORY DIRECTOR SIGNATURE

1-5-89

DATE



B. A. A. Q. M. D.

939 ELLIS ST.

SAN FRANCISCO

CA

94109

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATING OF OAKLANDLAB # 89-1342ADDRESS 4785 - 17TH ST.

S.T. #

CITY OAKLAND ZIP 94607

BOOK

TEL. 4151 444-5535 PLT # 2324

PAGE

SOURCE OPERATION S-2, S-3 CHROME TANKS

FILM

SAMPLE DESCRIPTION 1 BOTTLE LABELED PROBE AND NOZZLE RINSE

MANUFACTURER'S NAME

C. McCURE

PERSON SUBMITTING SAMPLE

12 1 22 199

DATE

W. Dwyer

SIGNATURE OF ANALYST

1-2-90

DATE

LAB REF # 09010-50LAB METHOD MOP-34Probe and Nozzle RinseTotal Cr +6. < 8 ugTotal Cr < 10 ug

V.N. RECOMMENDED

☐ YES

REG. _____

RULE _____

SEC. _____

☐ NO☐ FURTHER EVALUATION NEEDED[Signature]
LABORATORY DIRECTOR SIGNATURE1-5-90
DATERECEIVED
1/16/90
1350
2088

REQUEST FOR LABORATORY ANALYSIS

FL-1

NAME FRANCIS PLATING OF OAKLAND LAB # 89-1342
 ADDRESS 785-7TH ST. S.T. #
 CITY OAKLAND CA ZIP 94607 BOOK
 TEL. (415) 444-5535 PLT # 2324 PAGE
 SOURCE OPERATION (S#) S-2, S-3 CHROME TANKS FILM

SAMPLE DESCRIPTION 1-BOTTLE - LABELED - PROBE
NOZZLE RINSE - H₂O + WASH MATERIAL + ACETONE

☐ TWO PART SAMPLE

☐ SPEC. SHEET ATTACHED

SAMPLE TAKEN FROM FRANCIS PLATING CO
 SAMPLE TAKEN BY C. MECLURE DATE 12/21/89 TIME 1:30
 SAMPLE TAKEN IN PRESENCE OF B.A.A.Q.M.D. ☒ YES ☐ NO BY DS.
 SAMPLE RCVD. BY W. O. YUN DATE 12-22-89 TIME 10:30
 SAMPLE RCVD. BY DATE TIME
 CONTAINER DESCRIPTION AND MARKINGS: 1-BOTTLE - LABELED
PLT# 2324 - 12/21/89 - PROBE WASH - H₂O +
ACETONE.

MFG. NAME

MFG. CODE(S)

☒ COPY FL-2 TO: C. MECLURE (OTHER THAN REQUESTOR)

☐ PRIORITY SAMPLE BY:
 REASON DATE REQD.

ANALYSIS REQUESTED: ST-35-HEXCHROME LAB METHOD 34
☐ OPERATION UNDER PERMIT CONDITION # FOR APPLIC. #
☐ OPERATION UNDER "BUBBLE"
☐ FOR INFORMATION ONLY
 COMPLIANCE WITH REG. 11 RULE 8 SECTION(S) 310.1,
 SUBSECTIONS
C. Meclure PERSON SUBMITTING SAMPLE W. O. Yun PERSON RECEIVING SAMPLE FOR ANALYSIS
 DATE 12 / 22 / 89 DATE 12 / 22 / 89

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATING OF OAKLAND INC.
 ADDRESS 785 7TH ST.
 CITY OAKLAND CA ZIP 94607
 TEL. (415) 444-5535 PLT # 2324
 SOURCE OPERATION S-2, S-3 CHROME TANKS

LAB # 89-1341
 S.T. #
 BOOK
 PAGE
 FILM

SAMPLE DESCRIPTION 3 FILTERS C-1, 12 RUN A C-4 RUN B
C-6 RUN C 7 JARS IMPINGERS (S-1, S-2) (S-3, S-4) (S-5, S-6)
AND BLANK

MANUFACTURER'S NAME

C. McCURE

PERSON SUBMITTING SAMPLE

121 21 189

DATE

W. Ouying
 SIGNATURE OF ANALYST

1-2-89
 DATE

LAB REF #

CVD 2-24

09010-51

LAB METHOD

mop-34

		Cr ⁶⁺ Total ug Total Chrome, ug	
*Impinger	9-1	< 4 ug/imp	< 10 ug/imp
	S-3	< 4 ug "	< 10 " "
	S-4	< 4 ug "	< 10 " "
	LOD	4 ug	10 ug
Filter	C-1 Run A	< 8 ug/filter	< 10 ug/filter
	C-4 Run B	< 8 " "	< 10 " "
	C-6 Run C	< 8 " "	< 10 " "
	LOD	8 ug	10 ug
* Analyzed 16 impingers only.			

V.N. RECOMMENDED

☐ YES

REG. _____

RULE _____

SEC. _____

☐ NO

☐ FURTHER EVALUATION NEEDED

[Signature]
 LABORATORY DIRECTOR SIGNATURE

1-5-89
 DATE

RECEIVED
 1/16/90
 1350
 C2M8

B. A. A. Q. M. D. 939 ELLIS ST. SAN FRANCISCO CA 94109

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATING OF OAKLAND INC.
 ADDRESS 785 7TH ST.
 CITY OAKLAND CA ZIP 94607
 TEL. (415) 444-5535 PLT# 2324
 SOURCE OPERATION S-2, S-3 CILCONC TANKS

LAB # 89-1341
 S.T. #
 BOOK
 PAGE
 FILM

SAMPLE DESCRIPTION 3 FILTERS C-1, C-4 RUN B
-6 -100C 7 JAZZ IMPINGERS (S-1, S-2) (S-3, S-4) (S-5, S-6)
AND TANK

MANUFACTURER'S NAME

C. McCLURE

PERSON SUBMITTING SAMPLE

12-1-89

DATE

[Signature]

SIGNATURE OF ANALYST

1-2-89

DATE

LAB REF #

CV02-24

09010-51

LAB METHOD

mcp-34

	Cr ⁺ Total ug	Total Chrome, ug
*Impinger S-1	<4 ug/imp	<10 ug/imp
S-2	<4 ug	<10 "
S-4	<4 ug	<10 "
LOD	1 ug	10 ug
Filter C-1 Run A	<8 ug/filter	<10 ug/filter
C-4 Run B	<8 "	<10 "
C-6 Run C	<8 "	<10 "
LOD	8 ug	10 ug
* Analyzed 14T impingers only.		

V.N. RECOMMENDED

☐ YES

REG.

RULE

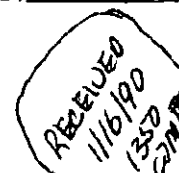
SEC.

☐ NO

☐ FURTHER EVALUATION NEEDED

[Signature]
 LABORATORY DIRECTOR SIGNATURE

1-5-90
 DATE



REQUEST FOR LABORATORY ANALYSIS

FL-1

NAME FRANCIS PLATING OF OAKLAND INC
 ADDRESS 785 8TH ST.
 CITY OAKLAND CH ZIP 94607
 TEL. 415-444-5535 PLT # 2324
 SOURCE OPERATION (S#) S-2, S-3 CHROME TANKS

LAB # 84-1341
 S.T. #
 BOOK
 PAGE
 FILM

SAMPLE DESCRIPTION 3 FILTERS EXPOSED TO CHROME GAS,
7 BOTTLES 6-CONDENSATE + H₂O + 1 BLANK
H₂O.

☐ TWO PART SAMPLE

☐ SPEC. SHEET ATTACHED

SAMPLE TAKEN FROM FRANCIS PLATING OF OAKLAND

SAMPLE TAKEN BY C. MECLURE DATE 12/21/89 TIME 1300

SAMPLE TAKEN IN PRESENCE OF B.A.A.Q.M.D. ☒ YES ☐ NO BY US.

SAMPLE RCVD. BY C. MECLURE DATE 12-21-89 TIME 4:45 PM

SAMPLE RCVD. BY _____ DATE _____ TIME _____

CONTAINER DESCRIPTION AND MARKINGS: 3 PLASTIC CASES W/FILTERS
LABELLED C-1-RUN A, C-4-RUN B, C-6 RUN C.
7 JARS - IMPINGER CONDENSATE + H₂O - IMPINGERS
S-1 & 2 RUN A, S-3 & 4 RUN B, S-5 & 6 RUN C. + 1-BLANK H₂O.

MFG. NAME _____

MFG. CODE(S) _____

☒ COPY FL-2 TO: C. MECLURE S-507

(OTHER THAN REQUESTOR)

☐ PRIORITY SAMPLE BY: _____

REASON _____ DATE REQD. _____

ANALYSIS REQUESTED: ST-35 BARDINO HEX CHROME + TOTAL

☐ OPERATION UNDER PERMIT CONDITION # _____ FOR APPLIC. # CHROME METHOD

☐ OPERATION UNDER "BUBBLE"

☐ FOR INFORMATION ONLY

COMPLIANCE WITH REG. 11 RULE 8 SECTION(S) 310.1

SUBSECTIONS _____

C. MECLURE
 PERSON SUBMITTING SAMPLE

C. MECLURE
 PERSON RECEIVING SAMPLE FOR ANALYSIS

DATE 12 / 21 / 89

DATE 12 / 21 / 89

L. J. JAMES PLATING CO.

12/21/89 & 12/27/89

PLEASE NOTE: THIS SERIES OF TESTS WERE
PERFORMED SIDE BY SIDE ST-35//ARB-425
WITH A SPECIAL DUAL PROBE AND
MATCHING 6.9MM GLASS NOZZLES.

C. MCCLURE

S-507

C = HEX CHROME CONCENTRATION
IN (mg/m^3)

$$C_{\text{mg}/\text{m}^3} = \frac{(35.31) (\overset{\text{TOTAL}}{\text{HEX}} \text{ Wt}_{\text{mg}})}{1000 \cdot V_0}$$

$$C_{\text{mg}/\text{m}^3}_A = \frac{(35.31) (48 \mu\text{g})}{1000 \times 18.06}$$

$$C_{\text{mg}/\text{m}^3}_A = 0.0156 \text{ mg}/\text{m}^3$$

$$C_{\text{mg}/\text{m}^3}_B = \frac{(35.31) (48 \mu\text{g})}{1000 \times 23.53}$$

$$C_{\text{mg}/\text{m}^3}_B = 0.012 \text{ mg}/\text{m}^3$$

$$C_{\text{mg}/\text{m}^3}_C = \frac{(35.31) (48 \mu\text{g})}{1000 \times 22.69}$$

$$C_{\text{mg}/\text{m}^3}_C = 0.012 \text{ mg}/\text{m}^3$$

FRANCIS PLATING CO

12/21/89

ST-35

OUTLET

$$V_{0A} = \frac{(530) (V_m) (P_b)}{29.92 (T_m)}$$

$$V_{0A} = \frac{(530) (17.42) (30.14)}{29.92 (575)}$$

$$V_{0A} = 18.06 \text{ SCF}$$

$$V_{0B} = \frac{(530) (22.83) (30.14)}{29.92 (578)}$$

$$V_{0B} = 23.53 \text{ SCF}$$

$$V_{0C} = \frac{(530) (22.02) (30.14)}{29.92 \cdot 578}$$

$$V_{0C} = 22.69 \text{ SCF}$$

HEX CHROME EMISSION RATE

FRITZTULIS PLATING CO.

12/21/89

ST-35

OUTLET

$$M = g/hr.$$

$$M = \frac{(60)(1 \times 10^{-6})(\overset{\text{TOTAL HEX WEIGHT}}{W_p})(Q_{stack})}{V_o}$$

RUN A)

$$M_A \text{ g/hr} = \frac{(60)(1 \times 10^{-6})(48 \mu g)(12600)}{18.06 \text{ scf}}$$

$$M_{g/hr_A} = < 0.335 \text{ g/hr}$$

RUN B)

$$M_B \text{ g/hr} = \frac{(60)(1 \times 10^{-6})(48 \mu g)(12600)}{23.53 \text{ scf}}$$

$$M_B \text{ g/hr} = < 0.257 \text{ g/hr.}$$

RUN C)

$$M_C \text{ g/hr} = \frac{(60)(1 \times 10^{-6})(48 \mu g)(12600)}{22.69 \text{ scf}}$$

$$M_C \text{ g/hr} = < 0.266 \text{ g/hr}$$

HEX CHROME PLATING EMISSION FACTOR

HEX CHROME PLATING

12/21/89

ST-35

$$M_A = \frac{1000 \cdot M}{I}$$

I = AMPERAGE AT BATH

$$M_A = \text{mg/A-HR}$$

M = HEX CHROME EMISSION RATE

RUN A)

$$M_A \text{ mg/A-HR} = \frac{1000 \cdot (0.335 \text{ g/hr})}{3000 \text{ AMPS}}$$

$$M_A = 0.112 \text{ mg/A-HR}$$

RUN B)

$$M_A \text{ mg/A-HR} = \frac{1000 \cdot (0.257 \text{ g/hr})}{3000 \text{ AMPS}}$$

$$M_A \text{ mg/A-HR} = 0.086 \text{ mg/A-HR}$$

RUN C)

$$M_A \text{ mg/A-HR} = \frac{1000 \cdot (0.266 \text{ g/hr})}{3000 \text{ A}}$$

$$M_A \text{ mg/A-HR} = 0.088 \text{ mg/A-HR}$$



BAY AREA AIR QUALITY MANAGEMENT DISTRICT
939 Ellis Street, San Francisco, Calif. 94109 • 771-6000

Source Test Data Sheet

Source Operation CHROME SWINGEN
Plant FRANCIS PLATING - DALLAS
Sample Type ST-35 / ARB
Process Cycle CHROME - HARD
Duct Size 5.3 x 37" - RECTANGULAR
Duct Pressure -0.6" - IMPINGERS
ASSUMED H₂O

Source Test No. RUN A - PORT A
Run No. 12/20/89
Date: 21

6.7mm Nozzle Diameter
0.85 Pilot Tube
AMBIENT AIR Gas Collector No.
→ 30.4 Barometric Pressure
0.001818 Leak Rate @ 15" Hg
TIME EACH RUN
7 # OF POINTS
8 TIME EACH RUN

SAMPLING IMPINGERS S-1, S-2
SOME CONDITIONS
12/13/89

Sampling Train: ARB-125 - T₁-T₂ IMPINGERS

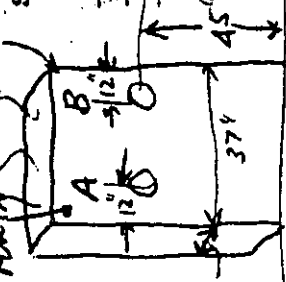
Traverse Point	Dist. From	INITIAL TRAVERSE		h	V _g FPS	Time 1018	Rate CFM	METER Temp of	Volume Ft ³	Vac "Hg	Sat'd Gas Temp of	T _g Duct Temp of	REMARKS
		Duct Temp of	H ₂ O										
P ₁ .02 .01				P ₁	.01 5.6		0.1323	56	310.42	1.5	35	65	8.0 0.86
P ₂ .04 .02				P ₂	.02 7.9		0.1874	56	311.25	2.2	40	65	11.2 1.4
P ₃ .10 .03				P ₃	.03 9.7		0.2292	56	314.15	2.0	36	55	13.8 1.47
P ₄ .12 .07				P ₄	.07 14.8		0.3501	56	316.75	5.0	30	55	21.0 2.6
P ₅ .13 .09				P ₅	.09 16.8		0.3969	56	319.25	5.0	35	55	23.8 2.5
P ₆ .12 .11				P ₆	.11 17.5		0.4388	56	322.46	5.2	36	55	26.3 3.21
P ₇ .12 .10				P ₇	.10 17.9		0.4185	56	325.54	6.5	30	55	25.1 3.01
P ₈ .12 .09				P ₈	.09 16.8	1114	0.3969	56	322.84	10.4	36	55	23.8 2.37
					13.74				17.42				17.42

$$D_h = 13.7 \left[\frac{T_g + 460}{V_g (100 - H_2O)} \right]^{1/2} = 1/2, 0.0001 \text{ Condensate } V_o \text{ gm.}$$

$$Q_m = \left[(2.638 \times 10^{-3}) (100 - H_2O) \right]^{1/2} \left[\frac{V_g}{T_g} \right]^{1/2}$$

$$V_s = 2.9 (K_p) \sqrt{T_s \cdot h_{H_2O}}$$

PORT A



Source Test Team
CHUCK McCLOURE
Source Test Engineer I

CHUCK McCLOURE
Source Test Engineer I

CHUCK McCLOURE
Source Test Engineer I

CHUCK McCLOURE
Source Test Engineer I

CHUCK McCLOURE
Source Test Engineer I

AMBIENT 0930
WINDS 44/10.8-47

STARK 0935
W.S. 52/55.5
20.2

STARK 0935
W.S. 52/55.5
20.2

STARK 0935
W.S. 52/55.5
20.2

STARK 0935
W.S. 52/55.5
20.2



BAY AREA AIR QUALITY MANAGEMENT DISTRICT
939 Ellis Street, San Francisco, Calif. 94109 • 771-6000

Source Test Data Sheet

Source Operation SCRUBBED
Plant FRANCIS PAVING CO
Sample Type CHRYSLER
Process Cycle
Duct Size 37X53
Duct Pressure -0.6
ASSUMED H₂O 300

Source Test No. 12/20/89

Run No. 2

Date: 12/20/89

Nozzle Diameter 6.9
Pilot Tube D.P.E.
Gas Collector No. N/A
Barometric Pressure 30.14
Leak Rate @ 15" Hg 0.00177
TIME EACH POINT 7
OF POINTS 8
TIME EACH RUN 56

FILTER C-6 IMPINGERS S-5, S-6

Sampling Train:

Traverse Point	Dist. From inches	INITIAL TRAVERSE		h H ₂ O	V _s FPS	Time	METER				SAMPLING			REMARKS
		Duct Temp °F	h				Rate CFM	Temp of	Volume Ft ³ 350.17	Vac "Hg	Sat'd Gas Temp °F	T _s Duct Temp °F		

$$Q_n = 13.7 \left[\frac{T_a + 460}{V_s (100 - H_2O)} \right]^{1/2}$$

$$Q_m = \left[(2.638 \times 10^{-3}) (100 - H_2O) D^2 \right] \frac{V_s}{T_s}$$

$$V_s = 2.9 (K_p) \sqrt{T_s \cdot h_{H_2O}}$$

Condensate gm.

H₂O
CO₂
CO
H₂

Source Test Team

CHUCK McCURE

Source Test Engineer I

FRANCIS PLATING CO.
12/21/89

RUNA - ST-35

OVERALL ISOKINETIC RATIO

$$R = \frac{(T_s)(V_o)(29.92)(100)}{(V_s)(T)(A_n)(P_s)(530)(60)(100-H_2O\%)}$$

$$R = \frac{(5150R)(17.42 \text{ ft}^3)(29.92)(100)}{(13.474)(56)(0.0004025 \text{ ft}^2)(29.92 - \frac{0.6}{13.6})(530)(60)(99)}$$

ft/sec min

$$R = \frac{26842130}{25499400} = 1.05$$

29.87

$$R_{ST} = \frac{26842130}{28559329} = 0.94$$

T_s = TEMP STACK OR

V_o = METERED VOLUME ft^3

V_s = AVERAGE STACK VOLUME

T = TIME OF RUN

A_n = AREA NOZZLE ft^2

P_s = PRESSURE IN STREAM

$H_2O\%$ = 1%

RUNA ARB425

$$R = \frac{(515)(18.089)(29.92)(100)}{(13.474)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_A = 0.9759 = 0.98$$

ARB

RUN B - ST-35

$$R = \frac{(518)(22.83)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_{ST} = 0.99$$

-FRANCIS PLATING CO
12/21/89

RUN B ARB-425

$$R = \frac{(518)(22.51)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_B = 0.98$$

ARB

RUN C ST-35

$$R = \frac{(518)(22.02)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_C = 0.96$$

ST

RUN C ARB-425

$$R = \frac{(518)(22.59)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_C = 0.98$$

ARB

SOURCE TEST LAB DATA SHEET

PLANT: FRANCIS PLATING CO
 SOURCE OPERATION: CHROME TANK SURFER-OUTLET
 SOURCE TEST NO.: ST-35
 FILTER MEDIA: FILTER FOLLOWED BY H₂O IMPINGERS

PAGE 1 OF 1
 INITIAL GMR
 TEST DATE 12/21/89
 Drying Procedure: 105 °C 24 hours before
 and after test, desiccated

FILTER DATA

RUN NO.	FILTER NO.	TARE WEIGHT (g)	FINAL WEIGHT (g)	SAMPLE WEIGHT (µg)

PROBE & NOZZLE RINSE DATA (total all runs)

RUN NO.	FILTER NO.	JAR TARE WEIGHT (g)	JAR FINAL WEIGHT (g)	SAMPLE WEIGHT (µg)

PROBE & NOZZLE RINSE DATA (each run)

RUN NO.	PROBE/NOZZLE NOS.	JAR TARE WEIGHT (g)	JAR FINAL WEIGHT (g)	SAMPLE WEIGHT (µg)

IMPINGER DATA

SOLUTION	NO.	A TARE WEIGHT	B FILLED WEIGHT	C FINAL WEIGHT	C-A SAMPLE WEIGHT	C-B CONDENSATE WEIGHT
H ₂ O	S-1	499.3	599.7	601.8	102.5	2.1
	S-2	495.0	597.0	597.1	101.7	- <0.3>
	S-3	482.3	583.1	584.4	104.1	3.3
	S-4	479.2	580.0	580.0	99.7	- <1.1>
	S-5	515.8	616.6	616.6	103.3	2.5
	S-6	488.4	588.2	588.2	98.9	- <0.9>

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATING OF OAKLAND LAB # 89-1343
 ADDRESS 785 7TH ST. ST. #
 CITY OAKLAND ZIP 94607 BOOK
 TEL. (415) 444-5535 PLT # 2324 PAGE
 SOURCE OPERATION S-2, S-3 CHROME TANK FILM

SAMPLE DESCRIPTION 1 BOTTLE CONTAINING PROBE WASH
OF 0.1 N NaOH.

MANUFACTURER'S NAME

C. Mc CLURE

PERSON SUBMITTING SAMPLE

12-1-22-89

DATE

W. Dymy

SIGNATURE OF ANALYST

1-2-90

DATE

LAB REF #

CV03-24

09010-51

LAB METHOD

ARB 425

Total sample volume = 100 ml
 1) $Cr^{+6} < 5.7 \mu g/100 \text{ ml wash}$
 2) Total $Cr < 10 \mu g/100 \text{ ml wash}$
 LOD $Cr^{+6} = 5.7 \mu g/100 \text{ ml wash}$
 $Cr^{+3} = 10 \mu g/100 \text{ ml wash}$

V.N. RECOMMENDED

☐ YES

REG. _____

RULE _____

SEC. _____

☐ NO

☐ FURTHER EVALUATION NEEDED

[Signature]

LABORATORY DIRECTOR SIGNATURE

1-16-90

DATE

REPORT OF LABORATORY ANALYSIS

12/21/89
FL-2

NAME FRANCIS PLATING CO OAKLAND LAB # 89-1343
 ADDRESS 785 7TH ST. OUTSIDE
 CITY OAKLAND ZIP 94607 BOOK
 TEL. (415) 444-5535 PLT # 2324 PAGE
 SOURCE OPERATION D-2 S-3 CILICOME TANK FILM
 SAMPLE DESCRIPTION 1 BOTTLE CONTAINING PROBE WASH
0.1 N NaOH.
 MANUFACTURER'S NAME

C. Mc CLURE
 PERSON SUBMITTING SAMPLE

12-1-22-189
 DATE

W. Dymally
 SIGNATURE OF ANALYST

1-2-90
 DATE

LAB REF # Q9010-51
 LAB METHOD ARB 425

Total sample volume = 100 ml
 1) $C_r^{+6} < 5.7 \mu g/100 \text{ ml wash}$
 2) Total $C_r < 10 \mu g/100 \text{ ml wash}$
 LOD $C_r^{+6} = 5.7 \mu g/100 \text{ ml wash}$
 $C_r^{+3} = 10 \mu g/100 \text{ ml wash}$

V.N. RECOMMENDED ☐ YES REG. _____ RULE _____ SEC. _____
☐ NO
☐ FURTHER EVALUATION NEEDED
1-16-90
 LABORATORY DIRECTOR SIGNATURE DATE

REQUEST FOR LABORATORY ANALYSIS

FL-1

NAME FRANCIS PLATING OF OAKLAND LAB # 89-1343
 ADDRESS 785-7TH ST. S.T. # _____
 CITY OAKLAND CA ZIP 94607 BOOK _____
 TEL. (415) 444-5535 PLT # 2324 PAGE _____
 SOURCE OPERATION (S#) S-2, S-3 CHROME TANKS FILM _____

SAMPLE DESCRIPTION 1-BOTTLE OF PROBE WASH - 0.1N
NAOH ≥ 100 ML.

☐ TWO PART SAMPLE

☐ SPEC. SHEET ATTACHED

SAMPLE TAKEN FROM FRANCIS PLATING OF OAKLAND
 SAMPLE TAKEN BY C. MECLURE DATE 12/21/89 TIME 1300
 SAMPLE TAKEN IN PRESENCE OF B.A.A.Q.M.D. ☒ YES ☐ NO BYUS
 SAMPLE RCVD. BY W. Dymy DATE 12-22-89 TIME 10:30
 SAMPLE RCVD. BY _____ DATE _____ TIME _____
 CONTAINER DESCRIPTION AND MARKINGS: 1-BOTTLE CONTAINING
PROBE WASH OF 0.1N NAOH ≥ 100 ML. MARKED
ARB 425.
 MFG. NAME _____
 MFG. CODE(S) _____

☒ COPY FL-2 TO: C. MECLURE (OTHER THAN REQUESTOR)

☐ PRIORITY SAMPLE BY: _____
 REASON _____ DATE REQD. _____

ANALYSIS REQUESTED: ARB 425 METHOD
☐ OPERATION UNDER PERMIT CONDITION # _____ FOR APPLIC. # _____
☐ OPERATION UNDER "BUBBLE"
☐ FOR INFORMATION ONLY
 COMPLIANCE WITH REG. 11 RULE 8 SECTION(S) 310.1,
 SUBSECTIONS _____
C. Meclure PERSON SUBMITTING SAMPLE W. Dymy PERSON RECEIVING SAMPLE FOR ANALYSIS
 DATE 12 / 22 / 89 DATE 12 / 22 / 89

B. A. A. Q. M. D. 939 ELLIS ST. SAN FRANCISCO CA 94109

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATING OF OAKLAND INC.
 ADDRESS 785 7TH ST.
 CITY OAKLAND CA ZIP 94607
 TEL. (415) 444-5535 PLT# 2324
 SOURCE OPERATION S-2, S-3 CHROME TANKS

LAB # 89-1340
 ST. #
 BOOK
 PAGE
 FILM

SAMPLE DESCRIPTION 3 FILTERS C-2, C-3, C-5
7 JARS IMP(T-1, T-2) (T-3, T-4) (T-5-T6) BLANK

MANUFACTURER'S NAME

C. McCURE
 PERSON SUBMITTING SAMPLE

12 / 21 / 89
 DATE

W. D. Young
 SIGNATURE OF ANALYST

1-2-90
 DATE

LAB REF #

CVD-2-24

09010-51

LAB METHOD

ARB 425

Sample	Hexavalent Chrome	Total Chrome
1) Run A + C ₂	< 11.4 µg/sample	< 20 µg/sample
2) Run B + C ₃	< 11.4 " "	< 20 " "
3) Run C + C-5	< 11.4 " "	< 20 " "
LOD	2 µg/25ml Sample	20 µg/sample
	11.4 µg/sample	

V.N. RECOMMENDED

☐ YES

REG. _____ RULE _____

SEC. _____

☐ NO

☐ FURTHER EVALUATION NEEDED

R. J. Jaramado
 LABORATORY DIRECTOR SIGNATURE

1-16-90
 DATE

REPORT OF LABORATORY ANALYSIS

FL-2

NAME FRANCIS PLATING OF OAKLAND INC.
 ADDRESS 715 7TH ST.
 CITY OAKLAND ZIP 94607
 TEL. (415) 444-5535 PLT# 2324
 SOURCE OPERATION 2-2, 2-3 ENHANCE TANKS

LAB # 89-1340
 S.T. #
 BOOK
 PAGE
 FILM

SAMPLE DESCRIPTION 3 FILTERS C-2, C-3, C-5
7 JARS IMP T-2 (T-3, T-4) (T-5-T6) BLANK

MANUFACTURER'S NAME

C. McCLURE
 PERSON SUBMITTING SAMPLE

12 / 21 / 89
 DATE

W. O'Quinn
 SIGNATURE OF ANALYST

1-2-90
 DATE

LAB REF # CUD-2-24
09010-51
 LAB METHOD ARB 4.5

Sample	Hexavalent Chromium	Total Chromium
1) Run A + C-2	< 11.4 µg/sample	< 20 µg/sample
2) Run B + C-4	< 11.4 " "	< 20 " "
3) Run C + C-5	< 11.4 " "	< 20 " "
LOD	2 µg/95ml Sample	20 µg/sample
	11.4 µg/sample	

V.N. RECOMMENDED

☐ YES REG. _____ RULE _____ SEC. _____
☐ NO
☐ FURTHER EVALUATION NEEDED

[Signature]
 LABORATORY DIRECTOR SIGNATURE

1-16-90
 DATE

REQUEST FOR LABORATORY ANALYSIS

FL-1

NAME <u>FRANCIS PLATING OF OAKLAND, INC</u>	LAB # <u>89-1340</u>
ADDRESS <u>785-7TH ST.</u>	S.T. #
CITY <u>OAKLAND CA</u> ZIP <u>94607</u>	BOOK
TEL. <u>415/444-5535</u> PLT # <u>2324</u>	PAGE
SOURCE OPERATION (S#) <u>S-2, S-3 CHROME TANKS</u>	FILM
SAMPLE DESCRIPTION <u>3 FILTERS EXPOSED TO CHROME 6 AS.</u> <u>7 BOTTLES - 6 SAMPLES 1 BLANK</u> <u>D. IN NaOH + CONDENSATE</u>	
☐ TWO PART SAMPLE ☐ SPEC. SHEET ATTACHED	

SAMPLE TAKEN FROM <u>FRANCIS PLATING STACK</u>	
SAMPLE TAKEN BY <u>CHUCK MECLURE</u>	DATE <u>12/21/89</u> TIME <u>1300</u>
SAMPLE TAKEN IN PRESENCE OF B.A.A.Q.M.D.	☒ YES ☐ NO <u>BY DS.</u>
SAMPLE RCVD. BY <u>Chelavit</u>	DATE <u>12-21-89</u> TIME <u>4:45 PM</u>
SAMPLE RCVD. BY	DATE TIME
CONTAINER DESCRIPTION AND MARKINGS: <u>3 PLASTIC CASES WITH</u> <u>EXPOSED FILTERS - LABELED C-2, C-3, C-5</u> <u>7 JARS - 1 IMPINGER T-1, T-2, T-3, T-4, T-5, T-6, + 1-BLANK</u> <u>C-2 RUN A, C-3 RUN B, C-5 RUN C</u> <u>RUNA RUNB RUNC</u>	
MFG. NAME	
MFG. CODE(S)	

☒ COPY FL-2 TO: C. MECLURE - S-507 (OTHER THAN REQUESTOR)☐ PRIORITY SAMPLE BY: _____
REASON _____ DATE REQD. _____

ANALYSIS REQUESTED: <u>2ARB METHOD 425</u>	
☐ OPERATION UNDER PERMIT CONDITION # _____	FOR APPLIC. # _____
☐ OPERATION UNDER "BUBBLE"	
☐ FOR INFORMATION ONLY	
COMPLIANCE WITH REG. <u>11</u>	RULE <u>8</u> SECTION(S) <u>310.1</u>
<u>C.2 Meclure</u> PERSON SUBMITTING SAMPLE	<u>Chelavit</u> PERSON RECEIVING SAMPLE FOR ANALYSIS
DATE <u>12 / 21 / 89</u>	DATE <u>12 / 21 / 89</u>

C = HEX CHROME CONCENTRATION (mg/m^3)

FRANCIS PLATTING CO
12/21/89

ARB-425

OUTLET

TOTAL HEX

$\mu\text{g} = < 11.4 \mu\text{g} / \text{SAMPLE}$

$$C_{\text{mg}/\text{m}^3} = \frac{(35.31) \left(\frac{\text{TOTAL HEX}}{\text{W}} \right)}{1000 \cdot V_0}$$

RUN A)

$$C_A \text{ mg}/\text{m}^3 = \frac{(35.31) (< 11.4 \mu\text{g})}{1000 \cdot (18.72)}$$

$$V_{0A} = \frac{(530 \times V_m)(P_b)}{29.92 (T_m)}$$

$$V_{0A} = \frac{(530)(18.09)(30.14)}{29.92 (516)}$$

$$C_A \text{ mg}/\text{m}^3 = < 0.021 \text{ mg}/\text{m}^3$$

$$V_{0A} = 18.72 \text{ SCF}$$

RUN B)

$$C_B \text{ mg}/\text{m}^3 = \frac{(35.31) (< 11.4 \mu\text{g})}{1000 \cdot (23.29)}$$

$$V_{0B} = \frac{(530)(22.51)(30.14)}{29.92 (516)}$$

$$V_{0B} = 23.29 \text{ SCF}$$

$$C_B \text{ mg}/\text{m}^3 = 0.017 \text{ mg}/\text{m}^3$$

$$V_{0C} = \frac{(530)(22.59)(30.14)}{29.92 (516)}$$

$$V_{0C} = 23.37 \text{ SCF}$$

RUN C)

$$C_C \text{ mg}/\text{m}^3 = \frac{(35.31) (< 11.4 \mu\text{g})}{1000 \cdot (23.37)}$$

$$C_C \text{ mg}/\text{m}^3 = 0.017 \text{ mg}/\text{m}^3$$

HEX CHROME EMISSION RATES

FRANCIS PLATINCO

12/21/89

ARIS-425

OUTLET

$$M = g/hr$$

$$M = \frac{(60)(1 \times 10^{-6}) \left(\overset{\text{TOTAL HEX CHROME WEIGHT}}{Wp \mu g} \right) (Q_{STACH})}{V_D}$$

RUN A)

$$M_A = \frac{(60)(1 \times 10^{-6})(<11.4 \mu g)(12600)}{18.72 \text{ SCF}}$$

$$V_{QA} = 18.72 \text{ SCF}$$

$$V_{DB} = 23.29 \text{ SCF}$$

$$V_{DC} = 23.37 \text{ SCF}$$

$$M_A = <0.460 \text{ g/hr}$$

RUN B)

$$M_B = \frac{(60)(1 \times 10^{-6})(<11.4 \mu g)(12600)}{23.29 \text{ SCF}}$$

$$M_B = <0.37 \text{ g/hr}$$

RUN C)

$$M_C = \frac{(60)(1 \times 10^{-6})(<11.4 \mu g)(12600)}{23.37}$$

$$M_C = <0.369 \text{ g/hr}$$

HEX CHROME PLATING EMISSION FACTOR

FRANCIS PLATING CO.

12/21/89

OUTLET

ARB-425

$$M_A = \text{mg/A-HR.}$$

$$M_A = \frac{1000 \cdot M}{I}$$

RUN A)

$$M_A = \frac{1000 (< 0.460 \text{ g/HR})}{3000 \text{ A}}$$

$$M_A = < 0.153 \text{ mg/A-HR}$$

RUN B)

$$M_B = \frac{1000 (< 0.37 \text{ g/HR})}{3000 \text{ A}}$$

$$M_B = < 0.123 \text{ mg/A-HR.}$$

RUN C)

$$M_C = \frac{1000 (< 0.369 \text{ g/HR})}{3000}$$

$$M_C = < 0.123 \text{ mg/A-HR.}$$



BAY AREA AIR QUALITY MANAGEMENT DISTRICT
939 Ellis Street, San Francisco, Calif. 94109 • 771-6000

Source Test Data Sheet

Source Operation CHARMIE SCRAPERS
Plant FRANKS PLEATING-DANMANN
Sample Type HEX-CHARMIE
Process Cycle CHARMIE-HARD
Duct Size 53X37
Duct Pressure 7.778 - 8.6
30%
ASSUMED H₂O

Source Test No. ARM A - PORT A

Run No. 12/20/79

Date: 21

6.9mm Nozzle Diameter
0.85 Pitot Tube
AMBIENT Gas Collector No.
3014 Barometric Pressure
10.30422 Leak Rate @ 15" Hg
7 TIME EACH POINT
8 # OF POINTS
56 TIME EACH RUN

Sampling Train:

Traverse Point	Dist. From inches	INITIAL TRAVERSE		Traverse Point	h H ₂ O	V _S FPS	Time	Rate CFM	METER Temp OF	Volume Ft ³	Vac "Hg	Sat'd Gas Temp OF	T _g Duct Temp OF	REMARKS
		Duct Temp OF	H H ₂ O											
				P ₁	.01	5.6	7mm	0.138	56	161.34	1.5	35	55	8 0.937
				P ₂	.02	7.9	7mm	.18714	56	161.347	2.0	36	55	11.2 1.318
				P ₃	.03	9.7	7mm	.2292	56	162.665	2.5	36	55	13.8 1.613
				P ₄	.07	14.8	7mm	.35011	56	163.278	7.0	37	55	21.0 2.785-2.462
				P ₅	.09	16.8	7mm	.39699	56	164.74	9.0	35	55	23.8 3.195-2.785
				P ₆	.11	18.5	7mm	.43887	56	165.525	10	36	55	26.3 3.195
				P ₇	.10	17.7	7mm	.4185	56	172.72	9.0	36	55	25.1 2.937
				P ₈	.09	16.2	11.7mm	.39699	56	175.57	9.0	36	55	23.8 2.842
						13.474				18.089				18.089

IMPAIRMENTS T₁, T₂

ARB-425

ALTER C-2

$$D_n = 13.7 \left[\frac{T_g + 460}{V_s (100 - H_2O)} \right]^{1/2}$$

Condensate V_o gm.

H₂O _____
CO₂ _____
CO _____
O₂ _____

$$Q_m = \left[(2.638 \times 10^{-3}) (100 - H_2O) D_n^2 \right] \frac{V_s}{T_s}$$

$$V_s = 2.9 (k_p) \sqrt{T_s \cdot h_{H_2O}}$$

Source Test Team
CHUCK McCURE
Source Test Engineer I
ALAN COLLIER
HROSHE DOE



Source Operation HARD CHROME
Plant FRANCIS PLATING CO.
Sample Type CHROME
Process Cycle CHROME BURST
Duct Size 53X37
Duct Pressure -26"
ASSUMED H₂O 300

12/29/50

6.9mm Nozzle Diameter
0.85 Pitot Tube
AMS Gas Collector M
30.12 Barometric Press
001278 Leak Rate @ 15"
7 TIME EACH POINT
8 # OF POINTS
576 TIME EACH RUN

[illegible]

$$D_n = 13.7 \left[\frac{T_B + 460}{V_S (100 - RH_2O)} \right]^{1/2}$$

$$Q_m = (2.638 \times 10^{-3}) (100 - w_{H_2O}) D_{\text{Hun}}^2 \frac{V_s}{T_s}$$

$$V_S = 2.9(kT_p) \sqrt{T_S \cdot h_{H_2O}}$$

Condensate

V_0
 H_2O
 CO_2
 CO
 NO_2

Source Test Team

CHUCK MCCLURE

Source Test Engineer 1

66TH CONGRESS
H. R. 1



BAY AREA AIR QUALITY MANAGEMENT DISTRICT
939 Ellis Street, San Francisco, Calif. 94109 • 771-6000

Source Test Data Sheet

Source Operation CHROME PLATING CO.
Plant FORNUS PLATING CO.
Sample Type CHROME - ARB 425
Process Cycle SEPARABLE
Duct Size 37X53
Duct Pressure -0.6"
ASSUMED H₂O 390

Source Test No. RUN C-425

Run No. 12/20/79

Date: 21

6.9mm Nozzle Diameter
0.85 Pitot Tube
ATMB Gas Collector No.
30.14 Barometric Pressure
20014717 Leak Rate @ 15"Hg
7 TIME EACH POINT
8 # OF POINTS
56 TIME EACH RUN

Sampling Train:

Traverse Point Dist. From inches

INITIAL TRAVERSE Duct Temp OF H₂O in

Traverse Point

h H₂O

V_S FPS

Time 1331

Rate CFM

Temp OF

Volume Ft³

Vac "Hg

Sat'd Gas Temp OF

T_g Duct Temp OF

REMARKS

APPROX CEH

FILTER C-5 IMPROVERS T-5, T-6

ARB-425

METER

Temp OF

Volume Ft³

Vac "Hg

Sat'd Gas Temp OF

T_g Duct Temp OF

REMARKS

APPROX CEH

$$Q_n = 13.7 \left[\frac{T_g + 460}{V_s (100 - H_2O)} \right]^{1/2}$$

$$Q_m = \left[(2.638 \times 10^{-3}) (100 - H_2O) D_{nan}^2 \right] \frac{V_s}{T_g}$$

$$V_s = 2.9 (k_p) \sqrt{T_s \cdot h_{H_2O}}$$

Condensate V_O gm.

H₂O
CO₂
CO
O₂

PAN C

A O
B O
C O

Source Test Team

CHUCK MCCLURE

Source Test Engineer I

KEVIN COWELL

HIROSHI DOI

FIG V-59

77a

SOURCE TEST LAB DATA SHEET

PLANT: FRANCIS PLATING CO
 SOURCE OPERATION: CHROME TANK SCRUBBER - OUTLET
 SOURCE TEST NO.: ARR-425
 FILTER MEDIA: FILTER FOLDED BY
0.1N NaOH IMPINGERS FOLLOWED BY FILTER

PAGE 1 OF 1
 INITIAL CMR
 TEST DATE 12/31/89
 Drying Procedure: 105 °C 24 hours before
and after test, desiccated

FILTER DATA

RUN NO.	FILTER NO.	TARE WEIGHT (g)	FINAL WEIGHT (g)	SAMPLE WEIGHT (µg)

PROBE & NOZZLE RINSE DATA (total all runs)

RUN NO.	FILTER NO.	JAR TARE WEIGHT (g)	JAR FINAL WEIGHT (g)	SAMPLE WEIGHT (µg)

PROBE & NOZZLE RINSE DATA (each run)

RUN NO.	PROBE/NOZZLE NOS.	JAR TARE WEIGHT (g)	JAR FINAL WEIGHT (g)	SAMPLE WEIGHT (µg)

IMPINGER DATA

SOLUTION	NO.	A TARE WEIGHT	B FILLED WEIGHT	C FINAL WEIGHT	C-A SAMPLE WEIGHT	C-B CONDENSATE WEIGHT
0.1N NaOH	T-1	514.4	615.2	671.5	157.1	56.3
}	T-2	503.2	605.3	671.5	17.7	- (54.4)
	T-3	473.0	574.0	671.5	102.5	1.5
	T-4	503.3	607.6	671.5	104.2	- (0.17)
	T-5	496.3	598.0	671.5	103.5	1.8
	T-6	470.1	572.0	671.5	101.8	- (0.17)

455
460
515R
50R

$$R_i = \frac{(T_{s1})(V_{m1})}{(v_{s1})(t_1)(A_n)(60)(100 - \%H_2O)} \times 100$$

where:

R_i = isokinetic ratio at point 1

V_{m1} = metered volume at point 1, ft^3 - 17.42

v_{s1} = stack gas velocity at point 1, fps - 13.474

t_1 = sample time at point 1, minutes 50

A_n = cross-sectional area of nozzle, ft^2 $\frac{0.0004025}{13.6}$ $A_n =$

H_2O = water vapor in gas stream, percent 10%

100 = conversion from percent to mole fraction

60 = conversion from seconds to minutes

10.9.1 The overall isokinetic ratio:

$$R = \frac{(T_s)(V_o)(29.92)(100)}{(V_s)(t)(A_n)(P_s)(530)(60)(100 - H_2O)}$$

11. Reporting

11.1 Results shall be reported as shown in Figure IV-55.

$$R_i = \frac{(515^{\circ}R)(17.44 \text{ ft}^3)}{(13.474)(50)(0.0004025)(60)(99)} \times 100$$

$$R_i = 557?$$

$$R = \frac{(515^{\circ}R)(17.42)(29.92)(100)}{(13.474)(50)(0.0004025)(29.92)(530)(60)(99)}$$

$$R = \frac{26842130}{25542084} = 1.05$$

OK

$$1 \text{ cm}^2 = 0.15499 \text{ in}^2$$

$$0.3739 \text{ cm}^2 = 0.15499 \text{ in}^2 (0.3739)$$

$$0.3739 \text{ cm}^2 = 0.05795 \text{ in}^2 = \text{AREA}_N$$

$$A_N = 0.05795 \text{ in}^2$$

$$A_n = 0.05795 \text{ in}^2 \cdot \frac{1 \text{ ft}^2}{144 \text{ in}^2}$$

$$A_n = 0.0004025 \text{ ft}^2$$

12
12

$$144 \text{ in}^2 = 1 \text{ ft}^2$$

$$\frac{1 \text{ cm}^2}{1.1 \text{ m}^2} = \frac{0.3937 \text{ in}^2}{1.1 \text{ m}^2} = 0.15499 \text{ in}^2$$

$$\odot D = 6.9 \text{ mm}$$

$$D = .69 \text{ cm}$$

$$r = \frac{.69}{2} = 0.345$$

$$A = \pi (0.345 \text{ cm})^2$$

$$A = 0.3739 \text{ cm}^2$$

$$\downarrow$$

$$\text{TO } \text{ft}^2$$

$$D = 6.9 \text{ mm} = D$$

$$r = \frac{D}{2} = \frac{6.9}{2} = 3.45$$

$$A = \pi r^2$$

$$A = \pi (3.45)^2 \text{ mm}^2$$

$$A = 37.393 \text{ mm}^2$$

$$A = 3.7393 \text{ cm}^2$$

$$6.9 \text{ mm} = 0.69 \text{ cm}$$

$$r = \frac{0.69}{2} = 0.345$$

$$A = \pi (0.345)^2$$

$$A = 0.3739 \text{ cm}^2$$

$$A = 1.496 \text{ cm}^2$$

FRANCIS PLATING CO.

12/21/89

(OUTLET)

RUNA - ST-35

OVERALL ISO KINETIC RATIO

$$R = \frac{(T_s)(V_o)(29.92)(100)}{(V_s)(T)(A_n)(P_s)(530)(60)(100-H_2O\%)}$$

$$R = \frac{(5150R)(17.42 \text{ ft}^3)(29.92)(100)}{(13.474)(56)(0.0004025 \text{ ft}^2)(29.92 - \frac{0.6}{13.6})(530)(60)(99)}$$

$$R = \frac{26842130}{25499400} = 1.05$$

$$R_{ST} = 0.94$$

T_s = TEMP STACK OR

V_o = METERED VOLUME FT³

V_s = AVERAGE STACK VELOCITY

T = TIME OF RUN

A_n = AREA NOZZLE FT²

P_s = PRESSURE IN STACK

$H_2O\%$ = 1%

RUNA ARB 425

$$R = \frac{(515)(18.089)(29.92)(100)}{(13.474)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_A = 0.9759 = 0.98$$

ARB

RUN B - ST-35

$$R = \frac{(518)(22.83)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_{ST} = 0.99$$

FRANCIS PLATING CO
12/21/89

RUN B ARB-425

$$R = \frac{(518)(22.51)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_B = 0.98$$

ARB

RUN C ST-35

$$R = \frac{(518)(22.02)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_C = 0.96$$

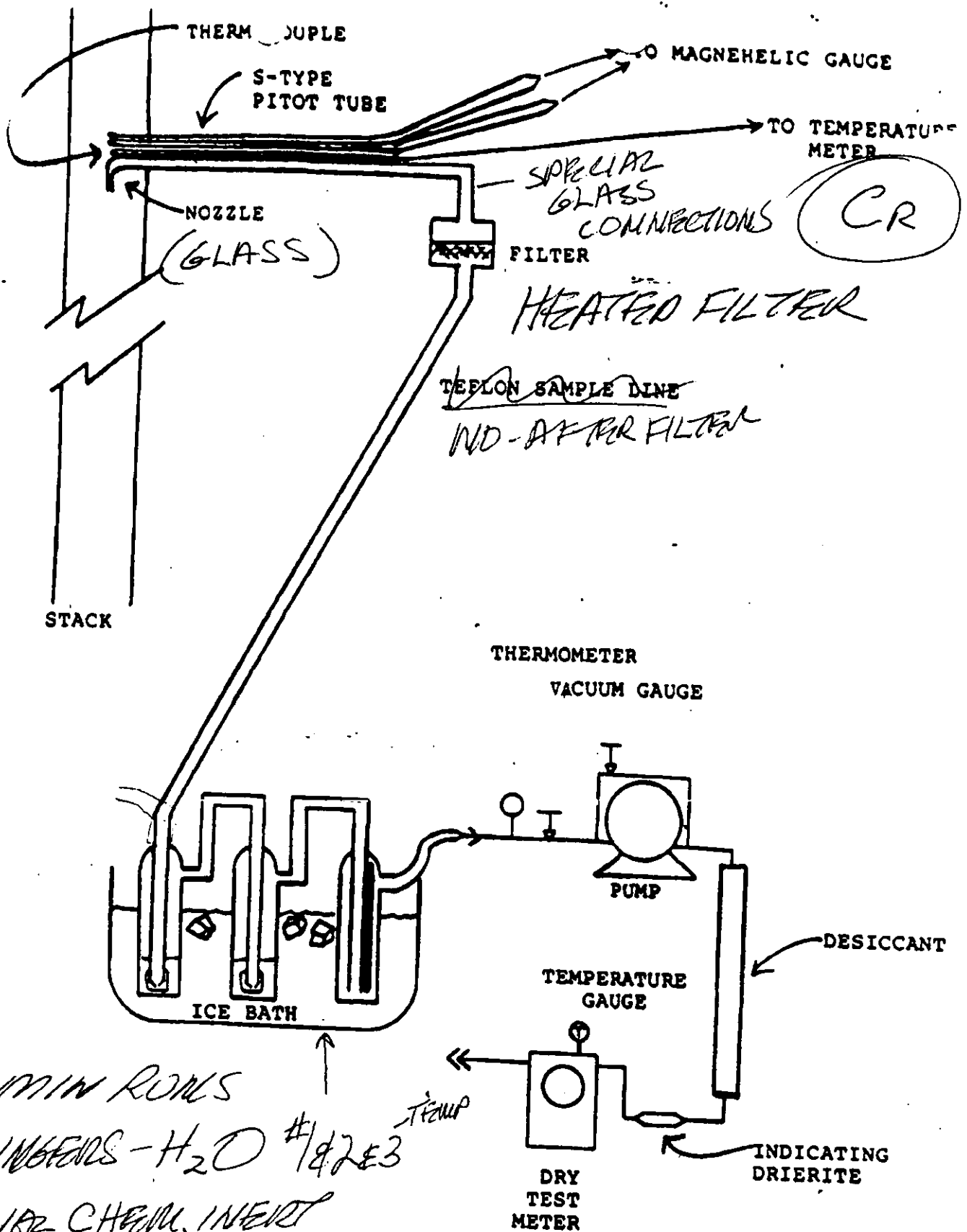
ST

RUN C ARB-425

$$R = \frac{(518)(22.59)(29.92)(100)}{(16.825)(56)(0.0004025)(29.87)(530)(60)(99)}$$

$$R_C = 0.98$$

ARB



50 MIN RUNS
 IMPINGERS - H_2O #1 & 2 & 3
 SPECIAL CHEM. INERT
 STORAGE BOTTLES FOR
 THE WASH. PLASTIC GLASS
 PARFLEX FILTERS

FIGURE IV-95
 PROBE AND SAMPLE TRAIN